

SI 1: Scree plot for PCA of DrugBank dataset

Below is presented the accompanying scree plot for the PCA detailed in the main text. The large number of PCs required to capture the variance is a reflection of the rich information content of the feature set; indeed, it would reflect poorly on it if it could be easily reduced to a small number of dimensions.

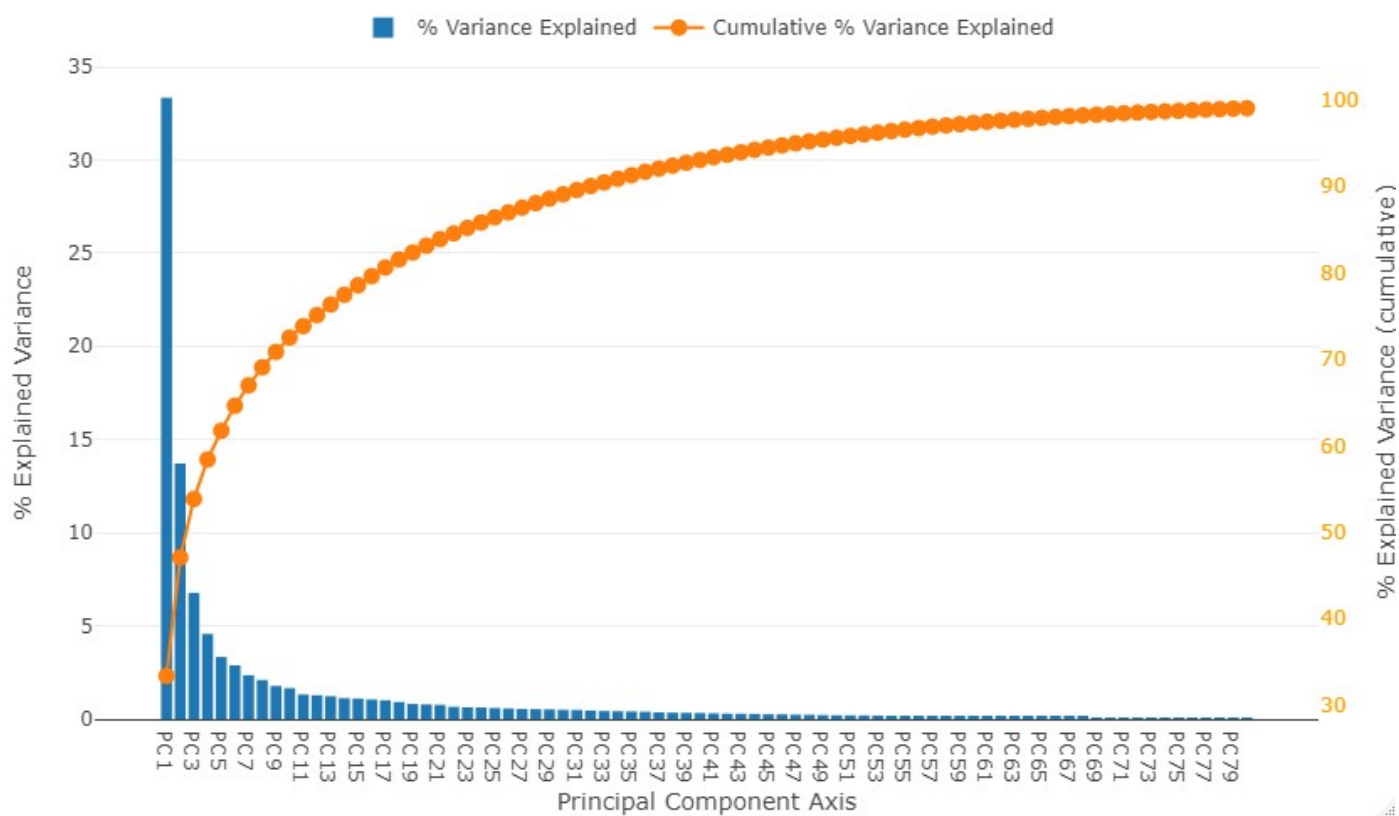


Figure SI-1. Accompanying scree plot for the PCA detailed in the main text.

SI 2: High error results

Readers may be interested in examining individual systems poorly predicted by COSMO-RS and our RF-based models. Below are listed i) the 50 highest-error COSMO-RS predictions for our dataset and ii) the 50 highest-error RF-hybrid (10-fold CV) predictions for our dataset. All solubility values are in $\log_{10}(\text{g}/100\text{g})$.

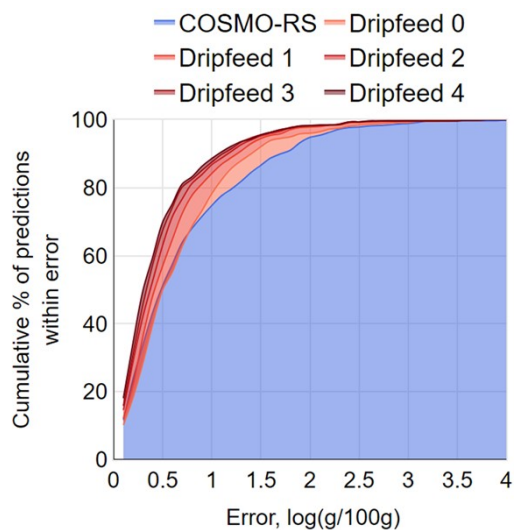
These data are included in the repository at https://github.com/AntonyVass/cmac_solpred_cosmo_rf.

Solvent Name	Solute Name	Exp. Solubility	COSMO Solubility	COSMO abs. error	RF Solubility	RF Abs. Error
Tetrahydrofuran	hydrocortisone	-1.9615	1.9971	3.9586	-0.1635	1.7980
2-Propanol	Sulfapyridine	-0.6518	2.5258	3.1777	-0.0349	0.6169
Acetonitrile	glibenclamide	-2.9626	0.2058	3.1683	-1.5324	1.4302
Methanol	2-aminoacetic acid	-3.3979	-0.2381	3.1599	-0.7616	2.6363
Ethyl Acetate	ascorbic acid	-1.5850	1.5090	3.0940	-0.4038	1.1812
Tetrahydrofuran	paracetamol	-1.4318	1.4971	2.9289	1.1767	2.6085
Ethanol	2-aminoacetic acid	-3.5229	-0.6261	2.8967	-1.1155	2.4073
Formamide	ibuprofen	-0.1825	2.6561	2.8386	1.6192	1.8017
Acetonitrile	ascorbic acid	-0.9872	1.8183	2.8055	-0.2224	0.7647
Acetonitrile	3-pyridinecarboxylic acid	-1.2558	1.5327	2.7885	0.5552	1.8110
1-Octanol	Sulfapyridine	-1.8214	0.9644	2.7859	-0.2375	1.5839
Methanol	glibenclamide	-2.7258	-0.0181	2.7077	-1.7453	0.9806
Acetone	glibenclamide	-1.9780	0.7219	2.6998	-0.8647	1.1133
Propionic Acid	2,2-dichloro-n-((1s,2r)-1-(fluoromethyl)-2-hydroxy-2-(4-(methylsulfonyl)phenyl)ethyl)acetamide	-0.5655	2.0158	2.5813	0.5046	1.0701
Chloroform	hydrocortisone	-0.5653	1.9364	2.5017	-0.3387	0.2266
Acetone	311-03-5	-0.5925	1.6738	2.2663	0.0880	0.6805
Methanol	haloperidol	0.2547	2.5144	2.2597	0.8929	0.6382
1,2-Dichloroethane	311-03-5	-0.5659	1.6921	2.2580	-0.0758	0.4901
2-Propanol	ascorbic acid	-0.7338	1.4295	2.1633	-0.3955	0.3383
Ethyl Acetate	hydrocortisone	-0.5061	1.6545	2.1606	-0.4556	0.0504
Benzene	hydrocortisone	-1.8296	0.3029	2.1325	-1.1609	0.6687
Ethanol	glibenclamide	-2.6345	-0.5781	2.0564	-2.0751	0.5594
Acetone	(2R,3R,4R,5S)-hexane-1,2,3,4,5,6-hexol	-0.9213	1.1125	2.0338	-0.3484	0.5730
Acetone	3-((6-O-(6-deoxy-??-L-mannopyranosyl)-??-D-glucopyranosyl)oxy)-2-(3,4-dihydroxyphenol)-5,7-dihydroxy-4H-1-benzopyran-4-one	-0.5452	1.4576	2.0029	0.3576	0.9028
Ethyl Acetate	lamivudine	-2.2007	-0.2210	1.9797	-0.5668	1.6338
Isopropyl Acetate	hydrocortisone	-0.7947	1.1719	1.9666	-0.6089	0.1858
Tetrahydrofuran	sulfamethoxypyridazine	-1.8083	0.1313	1.9395	-0.2467	1.5616
Dichloromethane	311-03-5	-0.0228	1.8862	1.9090	-0.2027	0.1799
Methyl Acetate	hydrocortisone	-0.2017	1.7041	1.9058	-0.3011	0.0994
Benzene	ibuprofen	1.3882	3.2879	1.8997	1.1579	0.2302
Tetrahydrofuran	311-03-5	-0.1482	1.7505	1.8987	-0.0762	0.0720
Benzene	piroxicam	-1.5139	0.3769	1.8908	-0.9281	0.5858
Chloroform	311-03-5	-0.0835	1.7866	1.8700	0.1051	0.1886
acetophenone	ibuprofen	-0.2772	1.5601	1.8372	1.4695	1.7467
Ethanol	ascorbic acid	0.0202	1.8523	1.8322	0.0480	0.0278
Dichloromethane	n-(1-oxopentyl)-n-((2-(1h-tetrazol-5-yl)(1,1-biphenyl)-4-yl)methyl)-l-valine	0.2925	2.1225	1.8300	0.1268	0.1657
2-Propanol	glibenclamide	-2.7055	-0.9017	1.8038	-2.0492	0.6564
Ethyl Acetate	glibenclamide	-1.4915	0.3093	1.8008	-1.6030	0.1115
Chloroform	n-(1-oxopentyl)-n-((2-(1h-tetrazol-5-yl)(1,1-biphenyl)-4-yl)methyl)-l-valine	0.1913	1.9882	1.7969	0.4732	0.2819
Acetone	haloperidol	0.3023	2.0944	1.7921	0.9224	0.6201

Solvent Name	Solute Name	Exp. Solubility	COSMO Solubility	COSMO abs. Error	RF Solubility	RF Abs. Error
Water	2-aminoacetic acid	1.3813	0.1143	-1.2670	-1.5556	2.9369
Methanol	2-aminoacetic acid	-3.3979	-0.2381	3.1599	-0.7616	2.6363
Tetrahydrofuran	paracetamol	-1.4318	1.4971	2.9289	1.1767	2.6085
Water	(2S)-5,7-dihydroxy-2-(3-hydroxy-4-methoxyphenyl)-2,3-dihydrochromen-4-one	-3.8539	-2.7924	1.0615	-1.4012	2.4527
Water	indomethacin	-4.0000	-4.1367	-0.1367	-1.5902	2.4098
Ethanol	2-aminoacetic acid	-3.5229	-0.6261	2.8967	-1.1155	2.4073
Water	gsk-B	-3.0969	-4.9208	-1.8239	-0.8189	2.2781
Hexane	4-Hydroxybenzoic acid	-3.3038	-3.8327	-0.5289	-1.4269	1.8769
Acetonitrile	3-pyridinecarboxylic acid	-1.2558	1.5327	2.7885	0.5552	1.8110
Formamide	ibuprofen	-0.1825	2.6561	2.8386	1.6192	1.8017
Tetrahydrofuran	hydrocortisone	-1.9615	1.9971	3.9586	-0.1635	1.7980
Cyclohexane	hydrocortisone	-3.9700	-3.6882	0.2818	-2.2090	1.7610
acetophenone	ibuprofen	-0.2772	1.5601	1.8372	1.4695	1.7467
Chloroform	2-hydroxypropane-1,2,3-tricarboxylic acid	-2.1549	-4.0555	-1.9006	-0.4637	1.6912
Tetrahydrofuran	2,2-dichloro-n-((1s,2r)-1-(fluoromethyl)-2-hydroxy-2-(4-(methylsulfonyl)phenyl)ethyl)acetamide	1.3137	0.7614	-0.5523	-0.3557	1.6694
Formamide	sclareol	-0.9996	-0.1116	0.8880	0.6458	1.6454
Ethyl Acetate	lamivudine	-2.2007	-0.2210	1.9797	-0.5668	1.6338
1-Octanol	Sulfapyridine	-1.8214	0.9644	2.7859	-0.2375	1.5839
Tetrahydrofuran	sulfadiazine	-3.1808	-3.0555	0.1253	-1.6037	1.5771
Tetrahydrofuran	sulfamethoxy-pyridazine	-1.8083	0.1313	1.9395	-0.2467	1.5616
N,N-Dimethylformamide	sulfadiazine	1.4178	0.5734	-0.8444	-0.1277	1.5455
Water	naproxen	-3.1549	-2.8136	0.3413	-1.6369	1.5181
Carbon Tetrachloride	hydrocortisone	-3.1899	-2.5923	0.5977	-1.6920	1.4979
Water	d-(-)fructose	2.4680	1.6708	-0.7972	1.0009	1.4672
Acetonitrile	glibenclamide	-2.9626	0.2058	3.1683	-1.5324	1.4302
Water	lamivudine	1.9289	2.4048	0.4759	0.5723	1.3566
Acetonitrile	n-(1-oxopentyl)-n-((2-(1h-tetrazol-5-yl)(1,1-biphenyl)-4-yl)methyl)-l-valine	1.1949	1.6570	0.4621	-0.1415	1.3364
Heptane	gsk-Z	0.8732	-0.7103	-1.5834	-0.4214	1.2946
Ethanol	(benzyl-disulfanyl)methyl benzene	-0.1614	0.8082	0.9695	1.1326	1.2939
Chloroform	haloperidol	1.7776	1.7329	-0.0447	0.5169	1.2607
Heptane	gsk-B	1.6317	-0.7109	-2.3426	0.3758	1.2559
Methanol	piroxicam	-1.6716	-0.2128	1.4588	-0.4305	1.2411
Hexane	mefenamic acid	-0.4384	-1.7544	-1.3161	-1.6592	1.2208
Ethyl Formate	salicylic acid	0.1536	1.6827	1.5290	1.3462	1.1926
2-Methoxy-2-Methylpropane	gsk-A	-2.1249	-0.6711	1.4539	-0.9407	1.1843
Ethyl Acetate	ascorbic acid	-1.5850	1.5090	3.0940	-0.4038	1.1812
Diethylether	diclofenac	-0.0741	1.5651	1.6392	1.1048	1.1789
Hexane	acetanilide	-2.6939	-1.1767	1.5173	-1.5264	1.1675
2-Propanol	3-((6-O-(6-deoxy-??-L-mannopyranosyl)-??-D-glucopyranosyl)oxy)-2-(3,4-dihydroxyphenol)-5,7-dihydroxy-4H-1-benzopyran-4-one	0.2034	-0.7320	-0.9354	-0.9518	1.1552
Diethylether	naproxen	-0.0306	1.5269	1.5576	1.1239	1.1545

SI 2: Drip-feeding CV of RF-pure

Drip-feeding CV was also performed using the RF-pure method (i.e. excluding COSMO-RS predictions). The results were as expected, adding little to the interpretation and conclusions of an already dense presentation, and so are excluded from the main text; for completeness, they are presented here.



Model	RMSE	MAE	FI
COSMO-RS	0.97	0.71	NA
Dripfeed 0	0.89	0.67	0.47
Dripfeed 1	0.76	0.56	0.54
Dripfeed 2	0.71	0.51	0.61
Dripfeed 3	0.68	0.48	0.63
Dripfeed 4	0.67	0.46	0.64

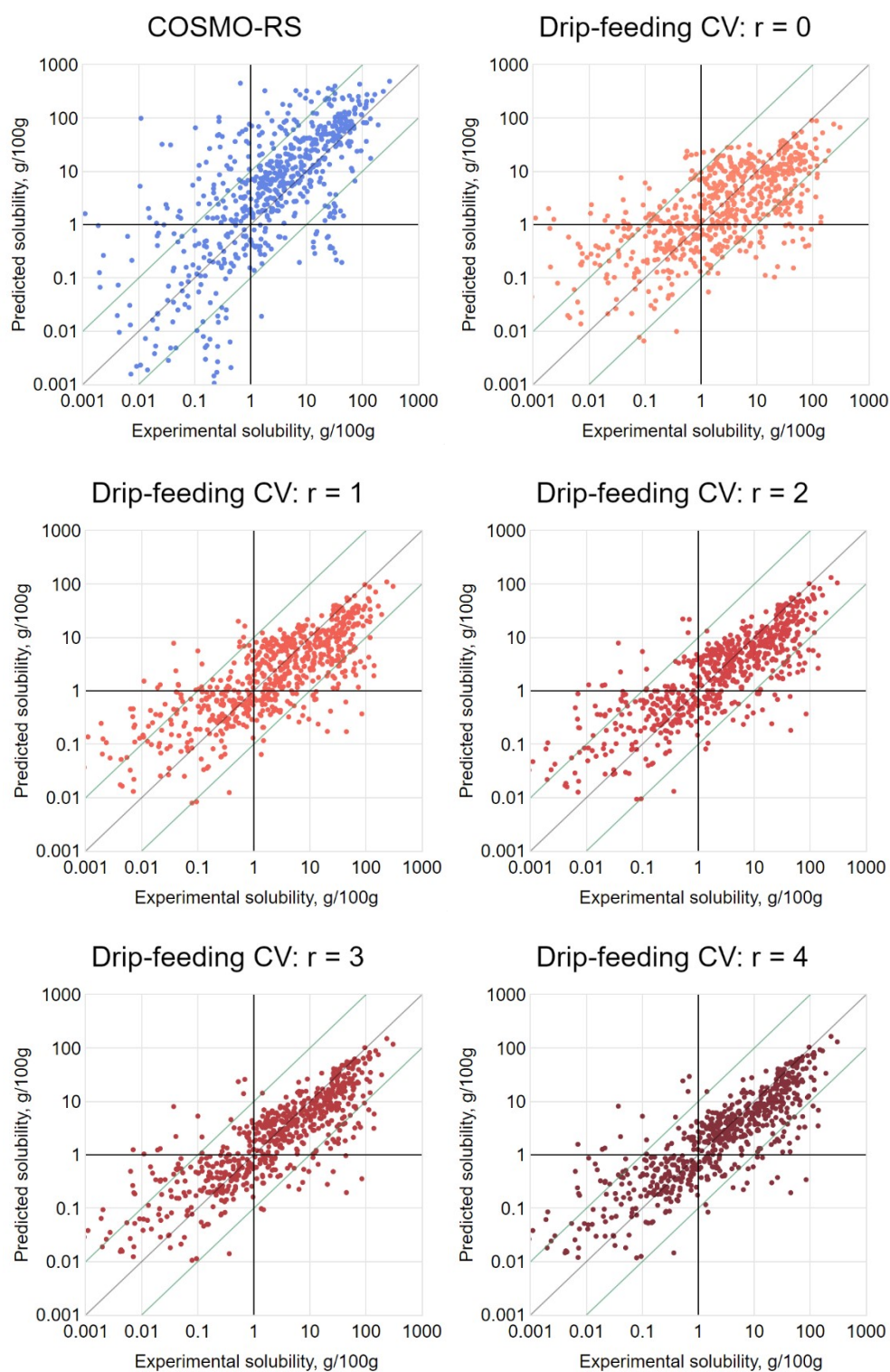


Figure SI-2. The results of drip-feeding CV for RF-pure, for r values of 0 to 4. Predictions were only made for solutes appearing in the dataset 5+ times to keep the plots comparable (658 of 714 instances). COSMO-RS prediction errors were recalculated for this subset for reference.

SI 3: Solvent list

Solvent ID	Solvent Name
1	1,2-dichloroethane
2	1,4-dioxane
3	1-butanol
4	1-heptanol
5	1-hexanol
6	1-octanol
7	1-pentanol
8	1-propanol
9	2-butanol
10	2-butanone
11	2-methoxy-2-methylpropane
12	2-methyltetrahydrofuran
13	2-propanol
14	4-methyl-2-pentanone
15	acetone
16	acetonitrile
17	acetophenone
18	anisole
19	benzene
20	carbon tetrachloride
21	chlorobenzene
22	chloroform
23	cyclohexane
24	cyclohexanone
25	dichloromethane
26	dimethylformamide
27	diethyl ether
28	dodecane
29	ethanol
30	ethyl acetate
31	ethyl formate
32	ethylene glycol
33	formamide
34	hexane
35	isobutanol
36	isopentanol
37	isopropyl acetate
38	methanol
39	methyl acetate
40	methylcyclohexane
41	n-butyl acetate
42	n-heptane
43	n-propyl acetate
44	octane
45	propionic acid
46	propylene glycol
47	tetrahydrofuran
48	toluene
49	water

SI 4: Solute list

Solute ID	Chemical Name	ΔH_{fus} (kJ/mol)	Melting point (°C)	Reference
1	2-(2-bromoethoxy)-1,3,5-trichlorobenzene	24.05	50.9	[1]
2	2-[(6-oxobenzo[c][2,1]benzoxaphosphinin-6-yl)methyl]butanedioic acid	43.63	188	[2]
3	2-aminopyridine	15.31	58.15	[3]
4	2-hydroxy-1,2-diphenylethanone	40.3	135	[4]
5	3,4-dimethoxybenzoic acid	29.62	179.97	[3]
6	3-[bis(3-hydroxypropyl)phosphoryl]propan-1-ol	34.21	114.61	[5]
7	4-aminobenzoic acid	20.93	186	[3]
8	4-hydroxy-3-methoxybenzaldehyde	21.36	81.75	[3]
9	4-hydroxybenzoic acid	32.5	216.25	[6]
10	acetanilide	21.65	114.3	[3]
11	ascorbic acid	19.72	192	[3]
12	benzoic acid	17.452	122.4	[3]
13	benzyl disulfide	44.69	68.57	[7]
14	butylparaben	24.57	67.57	[3]
15	carbamazepine	26.33	190.2	[8]
16	citric acid	40.32	153.85	[9]
17	D-fructose	36.06	104	[3]
18	D-sorbitol	35	92.7	[10]
19	diclofenac	40.4	179.45	[11]
20	diphenylamine	19.5	325.6	[12]
21	florfenicol	18.98	152	[13]
22	frambinone	22.75	84.43	[14]
23	fumaric acid	33.3	287	[15]
24	glibenclamide	52.9	173.33	[16]
25	glycine	28.4	262	[15]
26	haloperidol	22.44	151.5	[15]
27	hesperetin	35.9	226.07	[17]
28	hexanedioic acid	34.85	153.4	[3]
29	hydrocortisone	33.9	86.1	[18]
30	hydroquinone	27.11	171.95	[19]
31	ibuprofen	25.5	74	[20]
32	indomethacin	36.85	160.1	[21]
33	iodopropynyl butylcarbamate	22.69	68.26	[3]
34	isoniazid	28.15	172.89	[3]
35	ketoprofen	28.25	94.5	[3]
36	lamivudine	20.84	178.6	[22]
37	lovastatin	43.14	172.35	[23]
38	maltol	21.23	160.25	[24]
39	mefenamic acid	38.7	230.5	[15]
40	metacetamol	28.8	146.85	[25]
41	m-phthalic acid	43.2	344.26	[26]
42	N-(pyridin-3-ylmethylideneamino)pyridin-2-amine	34	175.85	[27]
43	naproxen	31.75	155.85	[3]
44	nicotinic acid	13.01	236.01	[28]
45	octadecanoic acid	56.4	69.6	[29]
46	oxolane-2,5-dione	20.4	119	[3]
47	paracetamol	28.1	168.6	[30]
48	phenacetin	31.27	135	[3]
49	phenylphosphinic acid	12.8	356	[31]
50	piroxicam	35.52	202.1	[32]
51	propylparaben	28.01	96.05	[3]
52	pyridoxine	31.15	160.4	[33]
53	pyrrolidine-2,5-dione	15.17	128.7	[34]
54	quercetin	32.9	318	[35]
55	rutin	82.3	177	[17]
56	salicylic acid	23.2	158.85	[36]
57	sclareol	28.7	103.3	[35]
58	sulfadiazine	44.3	255.5	[15]
59	sulfamethoxazole	33.76	167.5	[37]
60	sulfamethoxypyridazine	22.3	182.5	[15]
61	sulfapyridine	5.39	192	[38]

62	testosterone	28.2	426.45	[39]
63	testosterone propionate	22.14	120	[15]
64	theophylline	30.91	273.1	[40]
65	triclosan	17.76	57.95	[3]
66	trimethoprim	49.8	472.85	[39]
67	tris(1,2-dioxyphenyl)cyclotriphosphazene	19.54	249.05	[41]
68	valsartan	11.9	358.14	[42]
69	gsk-Q	58.42	242	*
70	gsk-A	18.59	245.7	*
71	gsk-H	33.14	147	*
72	gsk-S	18.98	52.7	*
73	gsk-Z	27.99	161.4	*
74	gsk-E	33.04	202	*
75	gsk-B	32.06	225	*

* Melting temperature and fusion enthalpy were obtained from Differential Scanning Calorimetry with a TA Instruments Q100 DSC, using a 10°C/min ramp and a maximum temperature of 300°C per default for a DSC scan. Solute identity withheld due to commercial sensitivity.

SI 5: Solubility Dataset

The following table is also available as a .csv file.

Solute id	Solute Name	Solvent id	Solvent Name	Solubility (g/100g)	Reference
1	2-(2-bromoethoxy)-1,3,5-trichlorobenzene	8	1-propanol	6.6716	[43]
		13	2-propanol	6.0484	
		15	acetone	89.6024	
		16	acetonitrile	23.2484	
		29	ethanol	7.2141	
		38	methanol	8.5503	
		42	n-heptane	27.822	
2	2-[(6-oxobenzo[c][2,1]benzoxaphosphinin-6-yl)methyl]butanedioic acid	2	1,4-dioxane	0.7695**	[44]
		8	1-propanol	1.5078	
		19	benzene	0.1463	
		23	cyclohexane	0.0947	
		29	ethanol	3.1778	
		30	ethyl acetate	0.4564	
		38	methanol	6.2608	
		47	tetrahydrofuran	1.7834	
3	2-aminopyridine	15	acetone	85.515	[19]
		22	chloroform	32.761	
		29	ethanol	121.1282	
		30	ethyl acetate	48.0374	
		41	n-butyl acetate	25.0209	
4	2-hydroxy-1,2-diphenylethanone	3	1-butanol	0.6699	[45]
		8	1-propanol	0.828	
		13	2-propanol	0.5614	
		15	acetone	4.6894	
		16	acetonitrile	3.4742	
		23	cyclohexane	0.3429	
		29	ethanol	1.1093	
		30	ethyl acetate	3.2595	
		35	isobutanol	0.552	
		38	methanol	1.6148	
		41	n-butyl acetate	1.9893	
5	3,4-dimethoxybenzoic acid	3	1-butanol	1.4089	[46]
		8	1-propanol	2.014	
		10	2-butanone	3.0685	
		13	2-propanol	1.9219	
		29	ethanol	2.948	
		30	ethyl acetate	1.4995	
		35	isobutanol	1.0118	
39	methyl acetate	1.9832			

6	3-[bis(3-hydroxypropyl)phosphoryl]propan-1-ol	3	1-butanol	3.3616	[5]
		8	1-propanol	14.2959	
		13	2-propanol	5.8822	
		15	acetone	0.3903	
		29	ethanol	11.2647	
		35	isobutanol	4.716	
		38	methanol	38.9079	
7	4-aminobenzoic acid	2	1,4-dioxane	11.712	[47]
		3	1-butanol	5.9958	
		4	1-heptanol	2.7498	
		5	1-hexanol	3.6734	
		6	1-octanol	2.2456	
		7	1-pentanol	4.2021	
		9	2-butanol	5.3453	
		13	2-propanol	7.5876	[48]
		15	acetone	12.433	
		19	benzene	0.0702	[49]
		29	ethanol	15.8719	[47]
		35	isobutanol	3.2973	
		38	methanol	21.5602	[50]
47	tetrahydrofuran	32.5448			
49	water	0.5409			
8	4-hydroxy-3-methoxybenzaldehyde	8	1-propanol	36.7084	[51]
		29	ethanol	274.41	[52]
		49	water	0.9917	[53]
9	4-hydroxybenzoic acid	6	1-octanol	17.1225	[54]
		13	2-propanol	34.4479	[20]
		14	4-methyl-2-pentanone	15.2063	
		22	chloroform	0.0191	[55]
		29	ethanol	40.0603	[56]
		30	ethyl acetate	12.4719	
		34	hexane	0.0005	[54]
		38	methanol	54.2084	[56]
		48	toluene	0.1651	[20]
49	water	0.744	[57]		
10	acetanilide	6	1-octanol	17.8864	[54]
		11	2-methoxy-2-methylpropane	3.025	#
		13	2-propanol	19.57	
		14	4-methyl-2-pentanone	13.1	
		16	acetonitrile	21.435	
		18	anisole	3	
		22	chloroform	16.8501	[58]
		23	cyclohexane	0.0339	[59]
		28	dodecane	0.19	#
		29	ethanol	33.2195	
		34	hexane	0.002	[54]
		36	isopentanol	16.5567	#
		37	isopropyl acetate	8.955	
		42	n-heptane	0.06	
		48	toluene	0.845	
49	water	0.3313			
11	ascorbic acid	13	2-propanol	0.1846	[60]
		16	acetonitrile	0.103	
		29	ethanol	1.0475	
		30	ethyl acetate	0.026	
		49	water	32.3572	
12	benzoic acid	1	1,2-dichloroethane	9.2173	[61]
		3	1-butanol	41.9647	[62]
		5	1-hexanol	30.0675	
		6	1-octanol	13.8886	
		7	1-pentanol	40.4525	
		9	2-butanol	39.9084	[63]
		13	2-propanol	48.8187	[61]
		16	acetonitrile	16.9477	[64]
		19	benzene	11.569	[61]
		20	carbon tetrachloride	4.1257	
21	chlorobenzene	10.2347			

		22	chloroform	15.0566	
		23	cyclohexane	1.4953	
		27	diethyl ether	37.0771	
		29	ethanol	57.7567	
		32	ethylene glycol	19.0798	
		33	formamide	48.7885	
		34	hexane	1.3592	
		35	isobutanol	29.6238	
		38	methanol	74.3314	
		41	n-butyl acetate	21.5181	
		42	n-heptane	1.4428	
		45	propionic acid	38.3433	
13	benzyl disulfide	22	chloroform	54.3717	[65]
		29	ethanol	0.6897**	
		30	ethyl acetate	29.7846	
		34	hexane	2.1299	
		48	toluene	41.324**	
14	butylparaben	3	1-butanol	149.9717	[66]
		5	1-hexanol	111.1637	
		6	1-octanol	73.7917	
		29	ethanol	237.1531	
		38	methanol	306.7349	
15	carbamazepine	3	1-butanol	1.7086	[67]
		10	2-butanone	3.4286	
		13	2-propanol	1.2675	
		15	acetone	1.5171	
		16	acetonitrile	4.1141	
		22	chloroform	11.3271	[68]
		25	dichloromethane	11.9245	
		29	ethanol	3.0925	
		30	ethyl acetate	1.1863	
		38	methanol	9.7092	
16	citric acid	2	1,4-dioxane	30.8055	[69]
		3	1-butanol	23.4759	[70]
		7	1-pentanol	14.0859	[69]
		22	chloroform	0.007	[71]
		29	ethanol	73.7287	[72]
		30	ethyl acetate	1.3475	
		32	ethylene glycol	66.575	
		33	formamide	61.623	[69]
		38	methanol	94.459	
		46	propylene glycol	11.3393	
		49	water	140.9947	[73]
17	D-fructose	29	ethanol	1.7238	[74]
		38	methanol	13.4464	
		49	water	293.7709	[75]
18	D-Sorbitol	3	1-butanol	0.2202	
		8	1-propanol	0.3758	
		15	acetone	0.1199	[76]
		16	acetonitrile	0.0923	
		38	methanol	3.3472	
19	diclofenac	1	1,2-dichloroethane	1.2109	
		2	1,4-dioxane	39.6463	
		6	1-octanol	3.4726	
		7	1-pentanol	4.3796	
		15	acetone	15.8773	
		17	acetophenone	11.5429	
		19	benzene	1.0342	
		21	chlorobenzene	0.9717	
		22	chloroform	2.3518	[48]
		23	cyclohexane	0.0216	
		27	diethyl ether	0.8432	
		29	ethanol	5.6614	
		30	ethyl acetate	7.8747	
		31	ethyl formate	1.6176	
		32	ethylene glycol	0.811	
		33	formamide	1.1586	

		38	methanol	5.4569	
		42	n-heptane	0.0177	
		45	propionic acid	3.9621	
		46	propylene glycol	1.9213	
20	diphenylamine	3	1-butanol	43.0394	[63]
		9	2-butanol	41.6423	
		36	isopentanol	28.781	
21	Florfenicol	3	1-butanol	0.5419	[13]
		8	1-propanol	1.2855	
		13	2-propanol	0.7956	
		29	ethanol	2.4125	
		35	isobutanol	0.2963	
		36	isopentanol	0.3901	
		38	methanol	12.9488	[77]
		45	propionic acid	0.272	[13]
		47	tetrahydrofuran	20.5918	[78]
		49	water	0.1092	
22	frambinone	3	1-butanol	57.8619	[14]
		8	1-propanol	60.1412	
		9	2-butanol	37.6299	
		13	2-propanol	67.3298	
		15	acetone	116.2637	
		29	ethanol	96.8699	
		30	ethyl acetate	39.7248	
		39	methyl acetate	49.3838	
23	fumaric acid	19	benzene	0.0045	[79]
		22	chloroform	0.02	
		49	water	0.6998	
24	glibenclamide	3	1-butanol	0.0043	[80]
		13	2-propanol	0.002	
		14	4-methyl-2-pentanone	0.0501	
		15	acetone	0.0105	
		16	acetonitrile	0.0011	
		29	ethanol	0.0023	
		30	ethyl acetate	0.0323	
		38	methanol	0.0019	
25	glycine	26	dimethylformamide	0.0464	[81]
		29	ethanol	0.0003	
		38	methanol	0.0004	[82]
		49	water	24.0616	
26	haloperidol	2	1,4-dioxane	2.4154	
		3	1-butanol	3.971	
		7	1-pentanol	1.3946	
		13	2-propanol	7.3404	
		15	acetone	2.0059	
		19	benzene	5.2437	
		22	chloroform	59.9185	
		26	dimethylformamide	21.6883	
		29	ethanol	4.6772	
		30	ethyl acetate	4.3397	
		38	methanol	1.7975	
		41	n-butyl acetate	1.6084	
		46	propylene glycol	0.445	
27	Hesperetin	3	1-butanol	1.3808	[84]
		15	acetone	12.9931	[85]
		16	acetonitrile	1.5202	
		29	ethanol	2.9156	
		30	ethyl acetate	2.3881	
		38	methanol	3.5678	
		49	water	0.0001	
		49	water	0.0001	[84]
28	hexanedioic acid	3	1-butanol	5.9939	[86]
		13	2-propanol	7.3003	[87]
		15	acetone	4.2448	[88]
		19	benzene	0.02	[87]
		22	chloroform	0.2565	[88]
		30	ethyl acetate	0.8233	[89]
		48	toluene	0.0072	[88]

		49	water	2.3701	[90]
29	hydrocortisone	6	1-octanol	0.4115	[91]
		19	benzene	0.0148	
		20	carbon tetrachloride	0.0006	
		22	chloroform	0.2721	
		23	cyclohexane	0.0001	
		30	ethyl acetate	0.3118	
		34	hexane	0.0005	
		37	isopropyl acetate	0.1604	
		39	methyl acetate	0.6285	
		41	n-butyl acetate	0.2175	
		46	propylene glycol	1.6094	
47	tetrahydrofuran	0.0109			
30	hydroquinone	29	ethanol	56.5762	[92]
		41	n-butyl acetate	12.5969	
31	ibuprofen	2	1,4-dioxane	9.0429	[93]
		7	1-pentanol	38.6346	[94]
		9	2-butanol	71.3233	
		13	2-propanol	104.5087	
		14	4-methyl-2-pentanone	62.75	[95]
		15	acetone	191.9609	[93]
		17	acetophenone	0.5282	
		19	benzene	24.4442	
		21	chlorobenzene	3.7644	
		22	chloroform	58.7729	
		23	cyclohexane	43.9589	
		26	dimethylformamide	41.3183	
		27	diethyl ether	6.3261	
		29	ethanol	41.0192	
		30	ethyl acetate	117.8434	
		31	ethyl formate	26.1583	
		32	ethylene glycol	6.4508	
		33	formamide	0.6569	
		34	hexane	9.61	[95]
		35	isobutanol	70.0541	[94]
		38	methanol	16.2917	[93]
41	n-butyl acetate	58.89	[95]		
42	n-heptane	12.2085	[93]		
45	propionic acid	73.9792			
46	propylene glycol	25.4709			
47	tetrahydrofuran	57.446**	[20]		
48	toluene	65.17	[95]		
32	indomethacin	2	1,4-dioxane	11.7761	[96]
		35	isobutanol	1.1973	[97]
		49	water	0.0001	[98]
33	iodopropynyl butylcarbamate	3	1-butanol	90.1139	[99]
		4	1-heptanol	44.3722	
		5	1-hexanol	56.7472	
		7	1-pentanol	71.9053	
		8	1-propanol	119.1368	
		29	ethanol	167.1137	
34	isoniazid	15	acetone	1.8612**	[100]
		29	ethanol	1.4307**	
		30	ethyl acetate	0.2228**	
		38	methanol	4.1301**	
35	ketoprofen	3	1-butanol	32.6079	[101]
		4	1-heptanol	15.8149	[101]
		5	1-hexanol	22.053	[102]
		6	1-octanol	14.494	
		7	1-pentanol	24.2685	
		8	1-propanol	35.8786	
		9	2-butanol	59.5923	
		13	2-propanol	61.4996	
		27	diethyl ether	38.149	
		29	ethanol	41.6094	
		30	ethyl acetate	52.1339	
35	isobutanol	38.4992			

		38	methanol	32.7219	[101]	
36	lamivudine	3	1-butanol	0.6667	[103]	
		8	1-propanol	0.8966		
		9	2-butanol	0.495		
		13	2-propanol	0.6242		
		15	acetone	0.1188		
		16	acetonitrile	0.218	[104]	
		26	dimethylformamide	20.258	[103]	
		29	ethanol	1.4449		
		30	ethyl acetate	0.0063		
		38	methanol	3.5777		
		47	tetrahydrofuran	0.4303	[104]	
49	water	84.9	[103]			
37	lovastatin	5	1-hexanol	1.8094	[105]	
		6	1-octanol	1.2906**	[23]	
		10	2-butanone	6.5327	[105]	
		15	acetone	8.6955	[106]	
		29	ethanol	3.1099		
		30	ethyl acetate	2.6248	[105]	
		37	isopropyl acetate	1.9784	[106]	
		38	methanol	3.9857	[105]	
		39	methyl acetate	2.452	[106]	
		41	n-butyl acetate	1.8483	[105]	
		43	n-propyl acetate	2.3388	[105]	
38	maltol	1	1,2-dichloroethane	2.2567	[24]	
		3	1-butanol	1.9829		
		8	1-propanol	2.5101		
		9	2-butanol	1.9132		
		13	2-propanol	2.1925		
		15	acetone	2.9581		
		16	acetonitrile	3.3163		
		29	ethanol	3.4651		
		30	ethyl acetate	2.2255		
		35	isobutanol	1.8524		
		38	methanol	7.4185		
		41	n-butyl acetate	1.4032		
		43	n-propyl acetate	1.6238		
39	mefenamic acid	8	1-propanol	0.75	[63]	
		13	2-propanol	0.8046	[107]	
		15	acetone	1.7271	[108]	
		23	cyclohexane	0.2583	[107]	
		26	dimethylformamide	5.743		
		29	ethanol	0.997		
		30	ethyl acetate	1.0722		
		34	hexane	0.3645		
		42	n-heptane	0.2652	[109]	
46	propylene glycol	0.1187				
40	metacetamol	11	2-methoxy-2-methylpropane	0.85	#	
		13	2-propanol	19.48		
		14	4-methyl-2-pentanone	6.94		
		16	acetonitrile	8.05		
		18	anisole	0.24		
		28	dodecane	0.115		
		37	isopropyl acetate	2.455		
		42	n-heptane	0.03		
41	m-phthalic acid	48	toluene	0.11	[110]	
		3	1-butanol	0.7574		
		4	1-heptanol	0.4209		
		5	1-hexanol	0.4958		
		6	1-octanol	0.357		
		7	1-pentanol	0.5984		
		8	1-propanol	0.9983		
		9	2-butanol	0.8224		
		13	2-propanol	1.1557		
		15	acetone	5.9599		[111]
		30	ethyl acetate	1.1083**		[110]
35	isobutanol	0.537				

		36	isopentanol	0.5519			
		38	methanol	1.996			
42	N-(pyridin-3-ylmethylideneamino)pyridin-2-amine	6	1-octanol	1.7241	[27]		
		34	hexane	0.0068			
		49	water	0.0139			
43	naproxen	1	1,2-dichloroethane	3.1347	[112]		
		2	1,4-dioxane	30.3348	[113]		
		3	1-butanol	4.462			
		4	1-heptanol	3.8564			
		5	1-hexanol	3.8111			
		6	1-octanol	2.8823			
		7	1-pentanol	3.4459			
		9	2-butanol	4.4684			
		13	2-propanol	5.1804			
				15	acetone	29.4826	[112]
				16	acetonitrile	2.5355	[114]
				17	acetophenone	80.3036	[112]
				19	benzene	2.1507	
				21	chlorobenzene	1.8352	
				22	chloroform	6.0239	
				23	cyclohexane	0.0328	
				27	diethyl ether	0.9319	
				29	ethanol	7.6115	[114]
				30	ethyl acetate	6.9764	
				31	ethyl formate	4.1876	[112]
				32	ethylene glycol	1.4335	
				33	formamide	3.3655	[113]
				35	isobutanol	2.7074	
				38	methanol	10.6326	[112]
				39	methyl acetate	8.7764	[113]
				41	n-butyl acetate	4.7539	
				42	n-heptane	0.0069	[114]
				45	propionic acid	9.1888	[112]
		46	propylene glycol	2.3389			
		47	tetrahydrofuran	52.7637	[113]		
		49	water	0.0007	[115]		
44	nicotinic acid	16	acetonitrile	0.0555	[116]		
		27	diethyl ether	0.1352			
		49	water	1.6619	[117]		
45	octadecanoic acid	15	acetone	3.8008	[118]		
		34	hexane	2.4276			
		42	n-heptane	1.5127			
46	oxolane-2,5-dione	13	2-propanol	14.3424	[119]		
		15	acetone	28.6563			
		16	acetonitrile	41.7409			
		30	ethyl acetate	9.4886			
47	paracetamol	2	1,4-dioxane	5.562	[69]		
		3	1-butanol	8.327	[120]		
		6	1-octanol	2.9397	[54]		
		7	1-pentanol	11.932	[120]		
		8	1-propanol	11.932	[121]		
		13	2-propanol	12.115	[120]		
		14	4-methyl-2-pentanone	1.66	[121]		
		15	acetone	9.938	[120]		
		16	acetonitrile	2.754			
		18	anisole	0.05	[121]		
		19	benzene	0.0213	[69]		
		22	chloroform	0.0405			
		23	cyclohexane	0.0054	[58]		
		28	dodecane	0.065	[121]		
		29	ethanol	20.991	[120]		
		30	ethyl acetate	0.945			
		33	formamide	63.6745	[69]		
		34	hexane	0.0007	[54]		
		36	isopentanol	5.4872	[121]		
37	isopropyl acetate	0.75					
38	methanol	33.211	[120]				

		45	propionic acid	2.2925	[69]
		46	propylene glycol	25.8261	
		47	tetrahydrofuran	0.037	[120]
		48	toluene	0.037	[118]
		49	water	1.49	[121]
48	phenacetin	2	1,4-dioxane	3.49	[122]
		3	1-butanol	6.9872	[123]
		7	1-pentanol	4.0384	
		8	1-propanol	5.3587	
		19	benzene	0.4483	
		22	chloroform	5.2131	[58]
		23	cyclohexane	0.0218	[123]
		29	ethanol	7.1486	
		30	ethyl acetate	2.3379	[54]
		34	hexane	0.0041	[123]
		38	methanol	11.3327	[122]
47	tetrahydrofuran	8.1868			
49	phenylphosphinic acid	38	methanol	120.8774	[31]
		48	toluene	4.4432	
		49	water	7.5109	
50	piroxicam	1	1,2-dichloroethane	2.9877	[124]
		2	1,4-dioxane	1.8678	
		6	1-octanol	0.0763	
		7	1-pentanol	0.093	
		15	acetone	1.6102	
		17	acetophenone	0.0564	
		19	benzene	0.0306	
		21	chlorobenzene	0.0207	
		22	chloroform	6.4167	
		23	cyclohexane	0.007	
		26	dimethylformamide	7.7252	
		27	diethyl ether	0.0226	
		29	ethanol	0.1016	
		30	ethyl acetate	0.9215	
		31	ethyl formate	3.9912	
		32	ethylene glycol	0.0747	
		33	formamide	0.3676	
38	methanol	0.0213			
42	n-heptane	0.0045			
45	propionic acid	0.7181			
46	propylene glycol	0.1179			
51	propylparaben	3	1-butanol	63.0751	[66]
		5	1-hexanol	46.8818	
		6	1-octanol	34.5933	
		7	1-pentanol	50.47	
		29	ethanol	95.9628	
		38	methanol	116.8251	
52	pyridoxine	15	acetone	0.3704	[33]
		29	ethanol	1.3379	
		38	methanol	8.5035	
		47	tetrahydrofuran	0.5645	
53	pyrrolidine-2,5-dione	13	2-propanol	2.9882	[34]
		15	acetone	14.8157	
		29	ethanol	5.4247	
		30	ethyl acetate	3.8622	
		38	methanol	18.3454	
		47	tetrahydrofuran	11.4643	
54	quercetin	8	1-propanol	2.6694	[35]
		13	2-propanol	2.7749	
		16	acetonitrile	1.1122	
		29	ethanol	3.3058	
55	rutin	3	1-butanol	0.2851	[125]
		8	1-propanol	1.8238	
		13	2-propanol	1.5975	
		15	acetone	0.285	
		29	ethanol	5.81	

		30	ethyl acetate	0.7527	
		38	methanol	5.7985	
56	salicylic acid	1	1,2-dichloroethane	1.0663	[48]
		3	1-butanol	36.7155	[126]
		6	1-octanol	24.235	[54]
		6	1-octanol	24.235	[127]
		7	1-pentanol	28.6855	[48]
		9	2-butanol	42.8331	
		13	2-propanol	50.0766	[126]
		16	acetonitrile	10.2057	[56]
		17	acetophenone	20.7246	
		19	benzene	0.9691	[48]
		20	carbon tetrachloride	0.2999	[128]
		21	chlorobenzene	0.8274	
		22	chloroform	0.1689	[48]
		23	cyclohexane	0.1269	[129]
		27	diethyl ether	33.4357	[48]
		29	ethanol	50.8461	[130]
		31	ethyl formate	1.4244	
		32	ethylene glycol	20.6973	[48]
		33	formamide	13.6578	
		34	hexane	0.0787	[54]
		35	isobutanol	31.0935	[126]
		38	methanol	63.2867	[56]
41	n-butyl acetate	18.7647	[126]		
42	n-heptane	0.154			
45	propionic acid	13.2051	[48]		
46	propylene glycol	23.2643			
57	sclareol	8	1-propanol	13.1222	
		15	acetone	17.9245	
		16	acetonitrile	4.0583	
		23	cyclohexane	21.5067	
		26	dimethylformamide	5.042	
		29	ethanol	15.2074	[131]
		30	ethyl acetate	12.9944	
		33	formamide	0.1001	
		34	hexane	1.626	
		38	methanol	7.6426	
48	toluene	15.3403			
58	sulfadiazine	2	1,4-dioxane	0.1406	[32]
		3	1-butanol	0.0107	
		6	1-octanol	0.0027	[132]
		7	1-pentanol	0.0075	
		13	2-propanol	0.0131	
		15	acetone	0.4495	[32]
		19	benzene	0.0009	
		26	dimethylformamide	26.1715	
		29	ethanol	0.0417	[132]
		30	ethyl acetate	0.0338	
		32	ethylene glycol	0.1293	
		35	isobutanol	0.0091	
		38	methanol	0.1828	[32]
		41	n-butyl acetate	0.0126	
46	propylene glycol	0.1181			
47	tetrahydrofuran	0.0007			
59	sulfamethoxazole	6	1-octanol	0.187	[133]
		13	2-propanol	1.1293	
		19	benzene	0.0576	[134]
		22	chloroform	0.1545	
		38	methanol	11.3991	
60	sulfamethoxypyridazine	2	1,4-dioxane	7.7899	
		3	1-butanol	0.2763	
		6	1-octanol	0.0667	
		7	1-pentanol	0.2642	[135]
		13	2-propanol	0.252	
		15	acetone	4.4715	
19	benzene	0.0215			

		22	chloroform	0.3033	
		29	ethanol	0.7676	
		30	ethyl acetate	0.6535	
		32	ethylene glycol	2.2511	
		35	isobutanol	0.1627	
		38	methanol	2.5795	
		41	n-butyl acetate	0.2125	
		46	propylene glycol	1.9664	
		47	tetrahydrofuran	0.0156	
61	Sulfapyridine	6	1-octanol	0.0151	[133]
		13	2-propanol	0.2229	[134]
62	Testosterone	4	1-heptanol	13.3115	[136]
		5	1-hexanol	14.7633	
		6	1-octanol	13.3623	
		7	1-pentanol	15.418	
		42	n-heptane	0.0446	
		44	octane	0.0452	
63	Testosterone propionate	19	benzene	139.269	[137]
		20	carbon tetrachloride	42.6574	
		21	chlorobenzene	113.199	
		23	cyclohexane	4.9716	
		34	hexane	2.0088	
		42	n-heptane	2.0056	
		44	octane	2.4627	
		47	tetrahydrofuran	119.4384	
64	theophylline	3	1-butanol	0.2888	[138]
		8	1-propanol	0.3656	[50]
		13	2-propanol	0.3909	[138]
		15	acetone	0.3436	[139]
		16	acetonitrile	0.1878	[140]
		22	chloroform	0.2573	[138]
		29	ethanol	0.5343	[139]
		30	ethyl acetate	0.1519	[50]
		38	methanol	0.3542	[141]
		47	tetrahydrofuran	0.5221	[50]
		49	water	0.7421	[142]
65	triclosan	22	chloroform	169.9449	[143]
		42	n-heptane	8.5075	
66	trimethoprim	3	1-butanol	0.2488	[144]
		7	1-pentanol	0.2232	
		9	2-butanol	0.2204	
		13	2-propanol	0.1558	
		15	acetone	0.4451	
		29	ethanol	0.4293	
		38	methanol	1.5684	
		47	tetrahydrofuran	0.472	
67	tris(1,2-dioxyphenyl)cyclotriphosphazene	1	1,2-dichloroethane	0.2717	[145]
		15	acetone	0.2555	
		22	chloroform	0.8251	
		25	dichloromethane	0.949	
		29	ethanol	0.1169	
		47	tetrahydrofuran	0.7109	
		48	toluene	0.4329	
68	valsartan	16	acetonitrile	15.664	[146]
		22	chloroform	1.5534	
		25	dichloromethane	1.9612	
69	gsk-Q	10	2-butanone	2.0621	##
		11	2-methoxy-2-methylpropane	0.1486	##
		16	acetonitrile	0.1426	##
		30	ethyl acetate	0.9756	##
		37	isopropyl acetate	0.8028	##
		48	toluene	0.046	##
70	gsk-A	1	1,2-dichloroethane	1.438	##
		2	1,4-dioxane	0.3226	##
		3	1-butanol	0.0592	##
		8	1-propanol	0.0504	##
		9	2-butanol	0.0427	##

		10	2-butanone	1.4214	##
		11	2-methoxy-2-methylpropane	0.0075	##
		12	2-methyltetrahydrofuran	0.1624	##
		13	2-propanol	0.0539	##
		14	4-methyl-2-pentanone	0.3232	##
		15	acetone	2.4343	##
		16	acetonitrile	4.1959	##
		18	anisole	0.3236	##
		21	chlorobenzene	0.0638	##
		25	dichloromethane	2.6862	##
		29	ethanol	0.1442	##
		30	ethyl acetate	0.2776	##
		32	ethylene glycol	0.2643	##
		37	isopropyl acetate	0.1843	##
		38	methanol	0.5194	##
		39	methyl acetate	0.9211	##
		40	methylcyclohexane	0.0062	##
		41	n-butyl acetate	0.211	##
		42	n-heptane	0.0072	##
		43	n-propyl acetate	0.2441	##
		47	tetrahydrofuran	0.7466	##
		48	toluene	0.0281	##
		49	water	0.4711	##
71	gsk-H	1	1,2-dichloroethane	10.3363	##
		21	chlorobenzene	2.4416	##
		25	dichloromethane	3.7823	##
		40	methylcyclohexane	0.0425	##
		42	n-heptane	0.0411	##
		48	toluene	4.3255	##
72	gsk-S	3	1-butanol	25.5876	##
		13	2-propanol	31.4802	##
		18	anisole	10.0299	##
		24	cyclohexanone	24.5607	##
		30	ethyl acetate	32.4112	##
		37	isopropyl acetate	29.1601	##
		40	methylcyclohexane	45.0723	##
		49	water	0.0143	##
73	gsk-Z	12	2-methyltetrahydrofuran	8.0148	##
		15	acetone	7.9983	##
		16	acetonitrile	2.1316	##
		40	methylcyclohexane	0.3326	##
		42	n-heptane	7.4675	##
		47	tetrahydrofuran	7.7727	##
		48	toluene	1.4353	##
74	gsk-E	10	2-butanone	3.6522	##
		11	2-methoxy-2-methylpropane	1.4857	##
		16	acetonitrile	1.5377	##
		30	ethyl acetate	3.1818	##
		37	isopropyl acetate	2.7294	##
		48	toluene	2.9425	##
75	gsk-B	3	1-butanol	26.333	##
		9	2-butanol	26.8157	##
		10	2-butanone	32.4269	##
		11	2-methoxy-2-methylpropane	40.67	##
		12	2-methyltetrahydrofuran	31.0638	##
		13	2-propanol	26.6172	##
		14	4-methyl-2-pentanone	26.8182	##
		15	acetone	45.9522	##
		16	acetonitrile	24.6449	##
		18	anisole	21.4686	##
		24	cyclohexanone	24.8217	##
		29	ethanol	29.1992	##
		30	ethyl acetate	32.9241	##
		37	isopropyl acetate	31.4965	##
		38	methanol	32.2754	##
		40	methylcyclohexane	34.182	##
		42	n-heptane	42.8235	##

	47	tetrahydrofuran	33.4494	##
	48	toluene	25.8603	##
	49	water	0.0008	##

** Using all temperature points within the data source, solubility at 25°C was determined using the Van't Hoff equation for solubility.

Solubility values determined experimentally using a Technobis Crystal16 via turbidity measurements. Heating rate = 0.2°C/min, cooling rate = 0.4°C/min. Cycling four times per sample. Solubility at 25°C was determined using the Van't Hoff equation for solubility.

Solubility data generated with the automated FreeSlate CM3 platform in a 96 well format using solids in excess, operating in isothermal mode. The HPLC used for determining solute concentration is calibrated prior to sample analysis to identify the specific response factor for the given solute, using the same input material as used in the samples. Only analytical data within the linear calibration range is taken forward.

SI 5: References

1. Lei, Z.Y., et al., *Solubility of 2-(2,4,6-Trichlorophenoxy)ethyl Bromide in Methanol, Ethanol, Propanol, Isopropanol, Acetonitrile, n-Heptane, and Acetone*. Journal of Chemical and Engineering Data, 2011. **56**(5): p. 2714-2719.
2. Shao, X.Z., L.S. Wang, and M.Y. Li, *Measurement and Correlation of the Solubilities of 2-(6-Oxido-6H-dibenz c,e 1,2 oxaphosphorin-6-yl)methyl butanedioic Acid in Selected Solvents*. Industrial & Engineering Chemistry Research, 2012. **51**(13): p. 5082-5089.
3. Gmehling, J., *Pure compound data from the Dortmund Data Bank (DDB)*. 1983-2014.
4. Kant, S., U.S. Rai, and R.N. Rai, *Thermal and physico-chemical studies on binary organic eutectic systems 4-Aminoacetophenone with benzoin and 4-nitrophenol*. Journal of Thermal Analysis and Calorimetry, 2012. **110**(2): p. 551-557.
5. Shao, X.-Z., et al., *Determination and Correlation of Solubility of Tris(3-hydroxypropyl)phosphine Oxide in Selected Solvents*. Journal of Chemical & Engineering Data, 2013. **58**(3): p. 598-604.
6. Monte, M.J.S., et al., *Vapour pressures, enthalpies and entropies of sublimation of para substituted benzoic acids*. Journal of Thermal Analysis and Calorimetry, 2010. **100**(2): p. 465-474.
7. Wang, S.X., et al., *Heat capacity and thermodynamic properties of benzyl disulfide (C14H14S2)*. Thermochimica Acta, 2007. **463**(1-2): p. 21-25.
8. Liu, C.J., et al., *Facile Method for the Prediction of Anhydrate/Hydrate Transformation Point*. Industrial & Engineering Chemistry Research, 2013. **52**(46): p. 16506-16512.
9. Meltzer, V. and E. Pincu, *Thermodynamic study of binary mixture of citric acid and tartaric acid*. Central European Journal of Chemistry, 2012. **10**(5): p. 1584-1589.
10. Talja, R.A. and Y.H. Roos, *Phase and state transition effects on dielectric, mechanical, and thermal properties of polyols*. Thermochimica Acta, 2001. **380**(2): p. 109-121.
11. Surov, A.O., et al., *Thermodynamic and Structural Aspects of Some Fenamate Molecular Crystals*. Crystal Growth & Design, 2009. **9**(7): p. 3265-3272.
12. Chelouche, S., et al., *Experimental and modeling studies of binary organic eutectic systems to be used as stabilizers for nitrate esters-based energetic materials*. Fluid Phase Equilibria, 2019. **498**: p. 104-115.
13. Zhang, P., et al., *Measurement and Correlation of the Solubility of Florfenicol Form A in Several Pure and Binary Solvents*. Journal of Chemical & Engineering Data, 2018. **63**(6): p. 2046-2055.
14. Lu, M., et al., *Solubility and solution thermodynamics of Raspberry Ketone in pure organic solvents and binary solvent mixtures from T=(293.15 to 333.15) K*. Journal of Molecular Liquids, 2017. **246**: p. 332-341.
15. Nielsen, T.L., et al., *The CAPEC Database*. Journal of Chemical & Engineering Data, 2001. **46**(5): p. 1041-1044.
16. Paus, R., et al., *Dissolution of Crystalline Pharmaceuticals: Experimental Investigation and Thermodynamic Modeling*. Industrial & Engineering Chemistry Research, 2015. **54**(2): p. 731-742.
17. Chebil, L., et al., *Solubility of Flavonoids in Organic Solvents*. Journal of Chemical & Engineering Data, 2007. **52**(5): p. 1552-1556.
18. Moodley, K., J. Rarey, and D. Ramjugernath, *Experimental solubility data for prednisolone and hydrocortisone in various solvents between (293.2 and 328.2) K by employing combined DTA/TGA*. Journal of Molecular Liquids, 2017. **240**: p. 303-312.
19. WEI, D., F. LI, and W. ZHANG, *Determination of Solid-Liquid Equilibria Data of Dihydroxyphenols*. Chemical Industry and Engineering, 2004(3): p. 227-230.
20. Gracin, S. and Å.C. Rasmuson, *Solubility of Phenylacetic Acid, p-Hydroxyphenylacetic Acid, p-Aminophenylacetic Acid, p-Hydroxybenzoic Acid, and Ibuprofen in Pure Solvents*. Journal of Chemical & Engineering Data, 2002. **47**(6): p. 1379-1383.
21. Legendre, B. and Y. Feutelais, *Polymorphic and thermodynamic study of indomethacin*. Journal of Thermal Analysis and Calorimetry, 2004. **76**(1): p. 255-264.

22. Chadha, R., P. Arora, and S. Bhandari, *Polymorphic Forms of Lamivudine: Characterization, Estimation of Transition Temperature, and Stability Studies by Thermodynamic and Spectroscopic Studies*. ISRN Thermodynamics, 2012. **2012**: p. 671027.
23. Nti-Gyabaah, J., et al., *Solubility of lovastatin in a family of six alcohols: Ethanol, 1-propanol, 1-butanol, 1-pentanol, 1-hexanol, and 1-octanol*. International Journal of Pharmaceutics, 2008. **359**(1-2): p. 111-117.
24. Li, J., et al., *Solubility and thermodynamic properties of maltol in different pure solvents*. Journal of Molecular Liquids, 2017. **243**: p. 313-323.
25. Barrio, M., et al., *The Pressure-Temperature Phase Diagram of Metacetamol and Its Comparison to the Phase Diagram of Paracetamol*. Journal of Pharmaceutical Sciences, 2017. **106**(6): p. 1538-1544.
26. Sabbah, R. and L. Perez, *Étude thermodynamique des acides phtalique, isophtalique et téréphtalique*. Canadian Journal of Chemistry, 1999. **77**(9): p. 1508-1513.
27. Perlovich, G.L., et al., *Solubility and Transfer Processes of Some Hydrazones in Biologically Relevant Solvents*. Journal of Chemical & Engineering Data, 2013. **58**(9): p. 2659-2667.
28. Sabbah, R. and S. Ider, *Energetics of inter- and intramolecular bonds in three carboxypyridinic acids (picolinic, nicotinic, and isonicotinic acids)*. Canadian Journal of Chemistry, 1999. **77**(2): p. 249-257.
29. Danilin, V.N., et al., *Phase diagrams of binary systems formed by saturated fatty acids*. Russian Journal of Physical Chemistry, 2001. **75**(1): p. 18-20.
30. Sacchetti, M., *Thermodynamic analysis of DSC data for acetaminophen polymorphs*. Journal of Thermal Analysis and Calorimetry, 2000. **63**(2): p. 345-350.
31. Tian, N.-N., et al., *Solubilities of Phenylphosphinic Acid, Methylphenylphosphinic Acid, Hexachlorocyclotriphosphazene, and Hexaphenoxycyclotriphosphazene in Selected Solvents*. Journal of Chemical & Engineering Data, 2011. **56**(3): p. 661-670.
32. Bustamante, P., A. Martin, and M.A. Gonzalezguisandez, *Partial Solubility Parameters and Solvatochromic Parameters for Predicting the Solubility Of Single And Multiple-Drugs In Individual Solvents*. Journal of Pharmaceutical Sciences, 1993. **82**(6): p. 635-640.
33. Kikic, I., et al., *Antisolvent Precipitation of Vitamin B6: A Thermodynamic Study*. Journal of Chemical and Engineering Data, 2011. **56**(12): p. 4978-4983.
34. Wang, K., et al., *Solubilities of succinimide in different pure solvents and binary methanol plus ethyl acetate solvent mixtures*. Thermochimica Acta, 2012. **538**: p. 79-85.
35. Remil Aguda and Chau-Chyun Chen, *Solubility of Nutraceutical Compounds in Generally Recognized as Safe Solvents at 298 K*. International Journal of Chemical Engineering and Applications, 2016. **7**: p. 289-294.
36. Mota, F.L., et al., *Aqueous solubility of some natural phenolic compounds*. Industrial & Engineering Chemistry Research, 2008. **47**(15): p. 5182-5189.
37. Martínez, F., C.M. Ávila, and A. Gómez, *Thermodynamic study of the solubility of some sulfonamides in cyclohexane*. Journal of the Brazilian Chemical Society, 2003. **14**: p. 803-808.
38. Mantheni, D.R., et al., *Solid state studies of drugs and chemicals by dielectric and calorimetric analysis*. Journal of Thermal Analysis and Calorimetry, 2012. **108**(1): p. 227-233.
39. Wassvik, C.M., et al., *Contribution of solid-state properties to the aqueous solubility of drugs*. European Journal of Pharmaceutical Sciences, 2006. **29**(3-4): p. 294-305.
40. Xu, F., et al., *Low Temperature Heat Capacity and Thermal Analysis of Caffeine, Theophylline and Aminophylline*. Acta Physico-Chimica Sinica, 2010. **26**(8): p. 2096-2102.
41. Tian, N.N., L.S. Wang, and R.Y. Jiang, *Solubilities of Tris(o-phenylenedioxy)cyclotriphosphazene in Selected Solvents*. Journal of Chemical and Engineering Data, 2011. **56**(7): p. 3208-3213.
42. Kleineberg, H., *Experimental and theoretical determination of the solubility of drugs*. Universdität Oldenburg, Bachelor (undergraduate Thesis), 2009: p. 1-97.
43. Lei, Z., et al., *Solubility of 2-(2,4,6-Trichlorophenoxy)ethyl Bromide in Methanol, Ethanol, Propanol, Isopropanol, Acetonitrile, n-Heptane, and Acetone*. Journal of Chemical & Engineering Data, 2011. **56**(5): p. 2714-2719.
44. Shao, X.-Z., L.-S. Wang, and M.-Y. Li, *Measurement and Correlation of the Solubilities of 2-[(6-Oxido-6H-dibenz[c,e][1,2]oxaphosphorin-6-yl)methyl]butanedioic Acid in Selected Solvents*. Industrial & Engineering Chemistry Research, 2012. **51**(13): p. 5082-5089.
45. Zhu, Y., et al., *Solubility Modeling and Mixing Properties for Benzoin in Different Monosolvents and Solvent Mixtures at the Temperature Range from 273.15 to 313.15 K*. Journal of Chemical & Engineering Data, 2018. **63**(2): p. 341-351.
46. Li, Q., et al., *Solubility of Veratric Acid in Eight Monosolvents and Ethanol + 1-Butanol at Various Temperatures*. Journal of Chemical & Engineering Data, 2013. **58**(4): p. 1020-1028.

47. Daniels, C.R., et al., *Mathematical correlation of 4-aminobenzoic acid solubilities in organic solvents with the Abraham solvation parameter model*. *Physics and Chemistry of Liquids*, 2004. **42**(6): p. 633-641.
48. Barra, J., M.A. Pena, and P. Bustamante, *Proposition of group molar constants for sodium to calculate the partial solubility parameters of sodium salts using the van Krevelen group contribution method*. *European Journal of Pharmaceutical Sciences*, 2000. **10**(2): p. 153-161.
49. Lazzell, C.L. and J. Johnston, *Solubility Relations of Isomeric Organic Compounds. VIII. Solubility of the Aminobenzoic Acids in Various Liquids*. *The Journal of Physical Chemistry*, 1928. **32**(9): p. 1331-1341.
50. Lin, H.-M. and R.A. Nash, *An Experimental Method for Determining the Hildebrand Solubility Parameter of Organic Nonelectrolytes*. *Journal of Pharmaceutical Sciences*, 1993. **82**(10): p. 1018-1026.
51. Du, Y., et al., *The liquid-liquid phase separation and crystallization of vanillin in 1-propanol/water solution*. *Fluid Phase Equilibria*, 2016. **409**: p. 84-91.
52. Pucher, G. and W.M. Dehn, *Solubilities in mixtures of two solvents*. *Journal of the American Chemical Society*, 1921. **43**(8): p. 1753-1758.
53. Noubigh, A., M. Abderrabba, and E. Provost, *Temperature and salt addition effects on the solubility behaviour of some phenolic compounds in water*. *The Journal of Chemical Thermodynamics*, 2007. **39**(2): p. 297-303.
54. Perlovich, G.L., T.V. Volkova, and A. Bauer-Brandl, *Towards an understanding of the molecular mechanism of solvation of drug molecules: A thermodynamic approach by crystal lattice energy, sublimation, and solubility exemplified by hydroxybenzoic acids*. *Journal of Pharmaceutical Sciences*, 2006. **95**(7): p. 1448-1458.
55. Martin, A., P.L. Wu, and A. Beerbower, *Expanded Solubility Parameter Approach II: P-Hydroxybenzoic Acid and Methyl P-Hydroxybenzoate in Individual Solvents*. *Journal of Pharmaceutical Sciences*, 1984. **73**(2): p. 188-194.
56. Nordstrom, F.L. and A.C. Rasmuson, *Solubility and melting properties of salicylic acid*. *Journal of Chemical and Engineering Data*, 2006. **51**(5): p. 1668-1671.
57. Apelblat, A. and E. Manzurola, *Solubilities of L-aspartic, DL-aspartic, DL-glutamic, p-hydroxybenzoic, o-anisic, p-anisic, and itaconic acids in water from T=278 K to T=345 K*. *The Journal of Chemical Thermodynamics*, 1997. **29**(12): p. 1527-1533.
58. Baena, Y., et al., *Temperature-dependence of the solubility of some acetanilide derivatives in several organic and aqueous solvents*. *Physics and Chemistry of Liquids*, 2004. **42**(6): p. 603-613.
59. Baena, Y., et al., *Temperature-dependence of the solubility of some acetanilide derivatives in several organic and aqueous solvents*. *Physics and Chemistry of Liquids*, 2004. **42**(6): p. 603-613.
60. Shalmashi, A. and A. Eliassi, *Solubility of L-(+)-Ascorbic Acid in Water, Ethanol, Methanol, Propan-2-ol, Acetone, Acetonitrile, Ethyl Acetate, and Tetrahydrofuran from (293 to 323) K*. *Journal of Chemical & Engineering Data*, 2008. **53**(6): p. 1332-1334.
61. Beerbower, A., P.L. Wu, and A. Martin, *Expanded Solubility Parameter Approach .1. Naphthalene And Benzoic-Acid In Individual Solvents*. *Journal of Pharmaceutical Sciences*, 1984. **73**(2): p. 179-188.
62. Restaino, F.A. and A.N. Martin, *Solubility of Benzoic Acid And Related Compounds in Series of N-Alkanols*. *Journal of Pharmaceutical Sciences*, 1964. **53**(6): p. 636.
63. Febra, S.A., et al., *Extending the SAFT- γ Mie approach to model benzoic acid, diphenylamine, and mefenamic acid: solubility prediction and experimental measurement*. *Fluid Phase Equilibria*, 2021: p. 113002.
64. Perlovich, G.L. and A. Bauer-Brandl, *Thermodynamics of solutions I: Benzoic acid and acetylsalicylic acid as models for drug substances and the prediction of solubility*. *Pharmaceutical Research*, 2003. **20**(3): p. 471-478.
65. Wang, Y., et al., *Solubility of Benzyl Disulfide in Five Organic Solvents between (283.45 and 333.15) K*. *Journal of Chemical & Engineering Data*, 2013. **58**(9): p. 2483-2486.
66. Alexander, K.S., et al., *Solubility Profiles and Thermodynamics of Parabens in Aliphatic-Alcohols*. *Journal of Pharmaceutical Sciences*, 1977. **66**(1): p. 42-48.
67. Liu, W., et al., *Solubility of Carbamazepine (Form III) in Different Solvents from (275 to 343) K*. *Journal of Chemical & Engineering Data*, 2008. **53**(9): p. 2204-2206.
68. Greene, L.R., A.C. Blackburn, and J.M. Miller, *Rapid, small-scale determination of organic solvent solubility using a thermogravimetric analyzer*. *Journal of Pharmaceutical and Biomedical Analysis*, 2005. **39**(1-2): p. 344-347.
69. Barra, J., et al., *The expanded Hansen approach to solubility parameters. Paracetamol and citric acid in individual solvents*. *Journal of Pharmacy and Pharmacology*, 1997. **49**(7): p. 644-651.
70. Yang, H. and J.-h. Wang, *Solubilities of 3-Carboxy-3-hydroxypentanedioic Acid in Ethanol, Butan-1-ol, Water, Acetone, and Methylbenzene*. *Journal of Chemical & Engineering Data*, 2011. **56**(4): p. 1449-1451.

71. Seidell, A. and W.F. Linke, *Solubilities of inorganic and metal organic compounds; a compilation of quantitative solubility data from the periodical literature*. 1940-41.
72. Qin, Y., et al., *Solubility of 2-Hydroxypropane-1,2,3-tricarboxylic Acid Monohydrate in Different Binary Solvents from (278.15 to 303.15) K*. Journal of Chemical & Engineering Data, 2014. **59**(2): p. 376-381.
73. Apelblat, A. and E. Manzurola, *Solubility of oxalic, malonic, succinic, adipic, maleic, malic, citric, and tartaric acids in water from 278.15 to 338.15 K*. The Journal of Chemical Thermodynamics, 1987. **19**(3): p. 317-320.
74. Macedo, E.A. and A.M. Peres, *Thermodynamics of Ternary Mixtures Containing Sugars. SLE of d-Fructose in Pure and Mixed Solvents. Comparison between Modified UNIQUAC and Modified UNIFAC*. Industrial & Engineering Chemistry Research, 2001. **40**(21): p. 4633-4640.
75. Vasatko, J. and A. Smelik, *Solubility of anhydrous beta-d-fructose in water*. Chemické Zvesti, 1967. **21**(9-10): p. 736.
76. Zhi, W., et al., *Measurement and correlation of solubility of d-sorbitol in different solvents*. Journal of Molecular Liquids, 2013. **187**: p. 201-205.
77. Lou, Y., et al., *Thermodynamic equilibrium and cosolvency of florfenicol in binary solvent system*. Journal of Molecular Liquids, 2018. **251**: p. 83-91.
78. Wang, S., N. Chen, and Y. Qu, *Solubility of Florfenicol in Different Solvents at Temperatures from (278 to 318) K*. Journal of Chemical & Engineering Data, 2011. **56**(3): p. 638-641.
79. Weiss, J.M. and C.R. Downs, *The physical properties of maleic, fumaric and malic acids*. Journal of the American Chemical Society, 1923. **45**(4): p. 1003-1008.
80. Zhao, Y., et al., *Solubility Determination and Correlation of Glibenclamide in 11 Monosolvents and (Acetone + Acetonitrile) Binary Solvents from 283.15 K to 323.15 K*. Journal of Chemical & Engineering Data, 2019. **64**(1): p. 189-201.
81. Ramakers, L.A.I., et al., *Investigation of Metastable Zones and Induction Times in Glycine Crystallization across Three Different Antisolvents*. Crystal Growth & Design, 2020. **20**(8): p. 4935-4944.
82. Fan, J.-P., et al., *Measurement and correlation of the solubility of genistin in eleven organic solvents from T=(283.2 to 323.2)K*. The Journal of Chemical Thermodynamics, 2015. **89**: p. 142-147.
83. Subrahmanyam, C.V.S. and S. Suresh, *Solubility behaviour of haloperidol in individual solvents determination of partial solubility parameters*. European Journal of Pharmaceutics and Biopharmaceutics, 1999. **47**(3): p. 289-294.
84. Liu, L. and J. Chen, *Solubility of Hesperetin in Various Solvents from (288.2 to 323.2) K*. Journal of Chemical & Engineering Data, 2008. **53**(7): p. 1649-1650.
85. Ferreira, O. and S.P. Pinho, *Solubility of Flavonoids in Pure Solvents*. Industrial & Engineering Chemistry Research, 2012. **51**(18): p. 6586-6590.
86. Bushinsk.Vi, G.N. Freidlin, and Kolomiet.Va, *Correlation between solubility of solids in liquids and free-energy of intermolecular reaction of solvent*. Zhurnal Prikladnoi Khimii, 1974. **47**(5): p. 1054-1058.
87. Smolyan, Z.S., et al., *The solubility of carboxynitrilic and adipic acid in some solvents*. Khim. Prom-st (Moscow), 1972. **48**: p. 508.
88. Wei, D. and W. Cao, *Solubility of Adipic Acid in Acetone, Chloroform, and Toluene*. Journal of Chemical & Engineering Data, 2009. **54**(1): p. 152-153.
89. Zhang, H., et al., *An odd-even effect on solubility of dicarboxylic acids in organic solvents*. Journal of Chemical Thermodynamics, 2014: p. 91-97.
90. Machado, P.G., et al., *Phase Equilibrium Involving Xylose, Water, and Ethylene Glycol or 1,2-Propylene Glycol at Different Temperatures*. Journal of Chemical & Engineering Data, 2019. **64**(5): p. 2163-2169.
91. Hagen, T.A. and G.L. Flynn, *Solubility of hydrocortisone in organic and aqueous-media - evidence for regular solution behavior in apolar solvents*. Journal of Pharmaceutical Sciences, 1983. **72**(4): p. 409-414.
92. Li, X.N., et al., *Solubility of hydroquinone in different solvents from 276.65 K to 345.10 K*. Journal of Chemical and Engineering Data, 2006. **51**(1): p. 127-129.
93. Bustamante, P., M.A. Pena, and J. Barra, *The modified extended Hansen method to determine partial solubility parameters of drugs containing a single hydrogen bonding group and their sodium derivatives: benzoic acid/Na and ibuprofen/Na*. International Journal of Pharmaceutics, 2000. **194**(1): p. 117-124.
94. Stovall, D.M., et al., *Solubility of crystalline nonelectrolyte solutes in organic solvents: Mathematical correlation of ibuprofen solubilities with the Abraham solvation parameter model*. Physics and Chemistry of Liquids, 2005. **43**(3): p. 261-268.
95. Perciballi, F., *Formation of optimised particles for formulation and processing*, in *Strathclyde Institute of Pharmacy and Biomedical Sciences*. 2018, University of Strathclyde.
96. Ruidiaz, M.A., et al., *Solubility and preferential solvation of indomethacin in 1,4-dioxane+water solvent mixtures*. Fluid Phase Equilibria, 2010. **299**(2): p. 259-265.

97. Nandi, I., M. Bari, and H. Joshi, *Study of isopropyl myristate microemulsion systems containing cyclodextrins to improve the solubility of 2 model hydrophobic drugs*. AAPS PharmSciTech, 2003. **4**(1): p. 71-79.
98. Shakeel, F., et al., *Solubility prediction of indomethacin in PEG 400+water mixtures at various temperatures*. Journal of Molecular Liquids, 2013. **188**: p. 28-32.
99. Lim, J., et al., *Solubility of triclosan and iodopropynyl butylcarbamate in pure alkanols at several temperatures*. Fluid Phase Equilibria, 2012. **332**: p. 144-150.
100. Heryanto, R., M. Hasan, and E.C. Abdullah, *Solubility of Isoniazid in Various Organic Solvents from (301 to 313) K*. Journal of Chemical & Engineering Data, 2008. **53**(8): p. 1962-1964.
101. Perlovich, G.L., et al., *Thermodynamics of solutions IV: Solvation of ketoprofen in comparison with other NSAIDs*. Journal of Pharmaceutical Sciences, 2003. **92**(12): p. 2502-2511.
102. Daniels, C.R., et al., *Thermochemical behavior of dissolved carboxylic acid solutes: Part 2 - Mathematical correlation of ketoprofen solubilities with the Abraham general solvation model*. Physics and Chemistry of Liquids, 2004. **42**(3): p. 305-312.
103. Jozwiakowski, M.J., et al., *Solubility Behavior of Lamivudine Crystal Forms in Recrystallization Solvents*. Journal of Pharmaceutical Sciences, 1996. **85**(2): p. 193-199.
104. Ojo, E., et al., *Developing process understanding for continuous manufacturing of Lamivudine (Epivir®) Stable Form I*, in BACG Conference. 2019: London.
105. Nti-Gyabaah, J. and Y.C. Chiew, *Solubility of lovastatin in ethyl acetate, propyl acetate, isopropyl acetate, butyl acetate, sec-butyl acetate, isobutyl acetate, tert-butyl acetate, and 2-butanone, between (285 and 313) K*. Journal of Chemical and Engineering Data, 2008. **53**(9): p. 2060-2065.
106. Sun, H., J.B. Gong, and J.K. Wang, *Solubility of lovastatin in acetone, methanol, ethanol, ethyl acetate, and butyl acetate between 283 K and 323 K*. Journal of Chemical and Engineering Data, 2005. **50**(4): p. 1389-1391.
107. Abdul Mudalip, S.K., et al., *Solubility and Dissolution Thermodynamic Data of Mefenamic Acid Crystals in Different Classes of Organic Solvents*. Journal of Chemical & Engineering Data, 2013. **58**(12): p. 3447-3452.
108. Erriguible, A., et al., *Effect of pressure and non-isothermal injection on re-crystallization by CO₂ antisolvent: Solubility measurements, simulation of mixing and experiments*. The Journal of Supercritical Fluids, 2013. **76**: p. 115-125.
109. Alsenz, J., E. Meister, and E. Haenel, *Development of a partially automated solubility screening (PASS) assay for early drug development*. Journal of Pharmaceutical Sciences, 2007. **96**(7): p. 1748-1762.
110. Schmidt, A., et al., *Determination of Abraham model solute descriptors for isophthalic acid from experimental solubility data in organic solvents at 298 K*. Physics and Chemistry of Liquids, 2016. **54**(6): p. 747-757.
111. Long, B. and Z. Yang, *Measurements of the solubilities of m-phthalic acid in acetone, ethanol and acetic ether*. Fluid Phase Equilibria, 2008. **266**(1): p. 38-41.
112. Bustamante, P., A. Pena, and J. Barra, *Partial-solubility parameters of naproxen and sodium diclofenac*. Journal of Pharmacy and Pharmacology, 1998. **50**(9): p. 975-982.
113. Daniels, C.R., et al., *Mathematical correlation of naproxen solubilities in organic solvents with the Abraham solvation parameter model*. Physics and Chemistry of Liquids, 2004. **42**(5): p. 481-491.
114. Buchholz, H.K., et al., *Thermochemistry of Racemic and Enantiopure Organic Crystals for Predicting Enantiomer Separation*. Crystal Growth & Design, 2017. **17**(9): p. 4676-4686.
115. Carr, A.G., R. Mammucari, and N.R. Foster, *Solubility, Solubility Modeling, and Precipitation of Naproxen from Subcritical Water Solutions*. Industrial & Engineering Chemistry Research, 2010. **49**(19): p. 9385-9393.
116. Gonçalves, E.M. and M.E. Minas da Piedade, *Solubility of nicotinic acid in water, ethanol, acetone, diethyl ether, acetonitrile, and dimethyl sulfoxide*. The Journal of Chemical Thermodynamics, 2012. **47**: p. 362-371.
117. Wang, L.-C. and F.-A. Wang, *Solubility of Niacin in 3-Picoline + Water from (287.65 to 359.15) K*. Journal of Chemical & Engineering Data, 2004. **49**(1): p. 155-156.
118. Jouyban, A., *Handbook of Solubility Data for Pharmaceuticals*. 2009.
119. Yang, W., et al., *Solubility of succinic anhydride in different pure solvents and binary solvent mixtures with the temperature range from 278.15 to 333.15K*. Journal of Molecular Liquids, 2013. **180**: p. 7-11.
120. Granberg, R.A. and A.C. Rasmuson, *Solubility of paracetamol in pure solvents*. Journal of Chemical and Engineering Data, 1999. **44**(6): p. 1391-1395.
121. Brown, C.J., et al., *Enabling precision manufacturing of active pharmaceutical ingredients: workflow for seeded cooling continuous crystallisations*. Molecular Systems Design & Engineering, 2018. **3**(3): p. 518-549.
122. Bustamante, C. and P. Bustamante, *Nonlinear Enthalpy–Entropy Compensation for the Solubility of Phenacetin in Dioxane–Water Solvent Mixtures*. Journal of Pharmaceutical Sciences, 1996. **85**(10): p. 1109-1111.

123. Chang, Q.-L., et al., *Solubility of Phenacetinum in Methanol, Ethanol, 1-Propanol, 1-Butanol, 1-Pentanol, Tetrahydrofuran, Ethyl Acetate, and Benzene between 282.65 K and 333.70 K*. Journal of Chemical & Engineering Data, 2007. **52**(5): p. 1894-1896.
124. Bustamante, P., M.A. Peña, and J. Barra, *Partial solubility parameters of piroxicam and niflumic acid*. International Journal of Pharmaceutics, 1998. **174**(1): p. 141-150.
125. Zi, J., B. Peng, and W. Yan, *Solubilities of rutin in eight solvents at T=283.15, 298.15, 313.15, 323.15, and 333.15K*. Fluid Phase Equilibria, 2007. **261**(1): p. 111-114.
126. De Fina, K.M., et al., *Solubility of 8-hydroxybenzoic acid in select organic solvents at 298.15 K*. Journal of Chemical and Engineering Data, 1999. **44**(6): p. 1262-1264.
127. Yalkowsky, S.H., S.C. Valvani, and T.J. Roseman, *Solubility And Partitioning and Octanol Solubility And Octanol-Water Partition-Coefficients*. Journal of Pharmaceutical Sciences, 1983. **72**(8): p. 866-870.
128. Shalmashi, A. and A. Eliassi, *Solubility of salicylic acid in water, ethanol, carbon tetrachloride, ethyl acetate, and xylene*. Journal of Chemical and Engineering Data, 2008. **53**(1): p. 199-200.
129. Dearden, J.C. and J.G. O'Sullivan, *Solubility of pharmaceuticals in cyclohexane*. Journal of Pharmacy and Pharmacology, 1988. **40**: p. 77.
130. Matsuda, H., et al., *Solubilities of Salicylic Acid in Pure Solvents and Binary Mixtures Containing Cosolvent*. Journal of Chemical and Engineering Data, 2009. **54**(2): p. 480-484.
131. Alwi, H., et al., *Extraction of Antipyretic Properties from Planted AquilariaSubintegra Spp. Dried Leaves*. International Journal of Chemical Engineering and Applications, 2016. **7**(5): p. 319-322.
132. Mauger, J.W., A.N. Paruta, and R.J. Gerraughty, *Solubilities of sulfadiazine, sulfisomidine, and sulfadimethoxine in several normal alcohols*. Journal of Pharmaceutical Sciences, 1972. **61**(1): p. 94.
133. Martinez, F. and A. Gomez, *Thermodynamic study of the solubility of some sulfonamides in octanol, water, and the mutually saturated solvents*. Journal of Solution Chemistry, 2001. **30**(10): p. 909-923.
134. Regosz, A., et al., *Prediction of Solubility Of Sulfonamides in Water And Organic-Solvents Based on The Extended Regular Solution Theory*. International Journal of Pharmaceutics, 1992. **88**(1-3): p. 437-442.
135. Bustamante, P., et al., *Predicting the Solubility Of Sulfamethoxyypyridazine In Individual Solvents .1. Calculating Partial Solubility Parameters*. Journal of Pharmaceutical Sciences, 1989. **78**(7): p. 567-573.
136. Gharavi, M., K.C. James, and L.M. Sanders, *Solubilities of Mestanolone, Methandienone, Methyltestosterone, Nandrolone And Testosterone In Homologous Series of Alkanes And Alkanols*. International Journal of Pharmaceutics, 1983. **14**(2-3): p. 333-341.
137. James, K.C., C.T. Ng, and P.R. Noyce, *Solubilities of testosterone propionate and related esters in organic-solvents*. Journal of Pharmaceutical Sciences, 1976. **65**(5): p. 656-659.
138. Adjei, A., *Solubility of Xanthine Derivatives in Polar and Nonpolar Solvents*. 1980, University of Texas. p. 209.
139. Subra, P., et al., *Precipitation and phase behavior of theophylline in solvent–supercritical CO2 mixtures*. The Journal of Supercritical Fluids, 2005. **35**(2): p. 95-105.
140. Khossravi, D. and K.A. Connors, *Solvent Effects on Chemical Processes, I: Solubility of Aromatic and Heterocyclic Compounds in Binary Aqueous—Organic Solvents*. Journal of Pharmaceutical Sciences, 1992. **81**(4): p. 371-379.
141. Liu, C., et al., *Solid–Liquid Equilibrium of Theophylline in Solvent Mixtures*. Journal of Chemical & Engineering Data, 2014. **59**(2): p. 263-268.
142. Pobudkowska, A., U. Domańska, and J.A. Kryska, *The physicochemical properties and solubility of pharmaceuticals – Methyl xanthines*. The Journal of Chemical Thermodynamics, 2014. **79**: p. 41-48.
143. Aragon, D.M., et al., *Solubility of the Antimicrobial Agent Triclosan in Organic Solvents of Different Hydrogen Bonding Capabilities at Several Temperatures*. Journal of Chemical and Engineering Data, 2008. **53**(11): p. 2576-2580.
144. Li, Q.S., Z. Li, and S. Wang, *Solubility of trimethoprim (TMP) in different organic solvents from (278 to 333) K*. Journal of Chemical and Engineering Data, 2008. **53**(1): p. 286-287.
145. Tian, N.-N., L.-S. Wang, and R.-Y. Jiang, *Solubilities of Tris(o-phenylenedioxy)cyclotriphosphazene in Selected Solvents*. Journal of Chemical & Engineering Data, 2011. **56**(7): p. 3208-3213.
146. Liu, Y., et al., *Solubility of Valsartan in Different Organic Solvents and Ethanol + Water Binary Mixtures from (278.15 to 313.15) K*. Journal of Chemical & Engineering Data, 2009. **54**(3): p. 986-988.