## **Electronic Supplementary Information (ESI)**

Turn-on and blue-shift fluorescence sensor toward L-histidine based on stable Cd<sup>II</sup> metal-organic framework with tetranuclear cluster units

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Cd1—O1	2.4954(18)	Cd1—O2	2.2766(19)
Cd1	2.2914(18)	Cd1—O4#2	2.4338(18)
Cd1—N7 <sup>#3</sup>	2.389(2)	Cd1—O7#4	2.5287(19)
Cd1—O8#4	2.3209(19)	O1-Cd2#1	2.3097(19)
Cd2—O1#5	2.3097(19)	Cd2—O5	2.2645(19)
Cd2—O3	2.2664(18)	Cd2—O6	2.471(2)
Cd2—N6	2.356(2)	Cd2—O7#6	2.3302(18)
O4—Cd1 <sup>#5</sup>	2.2914(18)	O4—Cd1 <sup>#2</sup>	2.4338(17)
N7—Cd1 <sup>#3</sup>	2.389(2)	O7—Cd1 <sup>#7</sup>	2.5287(19)
O7—Cd2 <sup>#6</sup>	2.3302(18)	O8—Cd1 <sup>#7</sup>	2.3209(19)
O1-Cd1-07 <sup>#4</sup>	165.02(6)	O2-Cd1-O1	54.47(6)
O2-Cd1-O4#1	127.42(6)	O2-Cd1-O4 <sup>#2</sup>	81.52(7)
O2-Cd1-N7 <sup>#3</sup>	102.77(8)	O2—Cd1—O7 <sup>#4</sup>	130.77(7)
O2—Cd1—O8 <sup>#4</sup>	89.10(7)	O4#1-Cd1-O1	78.08(6)
O4#2_Cd1_O1	85.57(6)	O4#1Cd1O4#2	72.78(7)
O4#1-Cd1-N7#3	97.67(7)	O4#2-Cd1-O7#4	81.87(6)
O4#1-Cd1-O7#4	90.37(6)	O4#1-Cd1-O8#4	142.21(7)
N7#3-Cd1-O1	89.64(7)	N7#3-Cd1-O4#2	170.02(7)
N7 <sup>#3</sup> —Cd1—O7 <sup>#4</sup>	101.45(7)	O8#4—Cd1—O1	139.29(7)
O8 <sup>#4</sup> —Cd1—O4 <sup>#2</sup>	108.51(7)	O8#4-Cd1-N7#3	80.74(8)
O8 <sup>#4</sup> —Cd1—O7 <sup>#4</sup>	53.73(7)	Cd2#1-O1-Cd1	112.87(7)
O1#5-Cd2O6	155.91(7)	O1#5-Cd2-N6	84.38(8)
O1 <sup>#5</sup> —Cd2—O7 <sup>#6</sup>	85.61(7)	O1#5-Cd2-C21	129.20(8)
O5—Cd2—O1 <sup>#5</sup>	102.65(7)	O5—Cd2—O3	143.39(7)
O5—Cd2—O6	54.99(7)	O5—Cd2—N6	86.72(8)
O5—Cd2—O7 <sup>#6</sup>	95.23(7)	O5—Cd2—C21	27.88(8)
O3—Cd2—O1 <sup>#5</sup>	112.20(7)	O3—Cd2—O6	91.64(7)
O3—Cd2—N6	85.87(8)	O3—Cd2—O7 <sup>#6</sup>	98.18(7)
N6-Cd2-06	101.43(8)	O7#4—Cd2—O6	87.66(7)
O7#6-Cd2-N6	169.99(8)	Cd1#5-04-Cd1#2	107.22(7)
Cd2#6-07-Cd1#7	108.24(7)		

Table S1. Selected bond lengths (Å) and angles (°) for JXUST-14.<sup>a</sup>

<sup>a</sup>Symmetry codes: #1: *x*, *y*, *z*-1; #2: -*x*+1, -*y*, -*z*; #3: -*x*+2, -*y*+1, -*z*; #4: *x*+1, *y*-1, *z*-2; #5: *x*, *y*, *z*+1; #6: -*x*, -*y*+1, -*z*+2; #7: *x*-1, *y*+1, *z*+2.

Ions	Label	Shape	Symmetry	Distortion(τ)
Cd1	HP-7	Heptagon	$D_{7\mathrm{h}}$	29.730
	HPY-7	Hexagonal pyramid	$C_{6\mathrm{v}}$	18.659
	PBPY-7	Pentagonal bipyramid	$D_{5\mathrm{h}}$	4.213
	COC-7	Capped octahedron	$C_{3\mathrm{v}}$	4.714
	CTPR-7	Capped trigonal prism	$C_{2\mathrm{v}}$	4.067
	JPBPY-7	Johnson pentagonal bipyramid J13	$D_{5\mathrm{h}}$	8.205
	JETPY-7	Johnson elongated triangular pyramid J7	$C_{3\mathrm{v}}$	20.238
Cd2	HP-6	Hexagon	$D_{6\mathrm{h}}$	32.417
	PPY-6	Pentagonal pyramid	$C_{5\mathrm{v}}$	17.674
	OC-6	Octahedron	$O_{ m h}$	4.538
	TPR-6	Trigonal prism	$D_{3\mathrm{h}}$	8.940
	JPPY-6	Johnson pentagonal pyramid J2	$C_{5\mathrm{v}}$	21.930

Table S2. SHAPE analysis of Cd<sup>II</sup> ions in JXUST-14.

**Table S3.** HOMO and LUMO energies for  $H_2BTDB$  and histidine.

	НОМО	LUMO
H <sub>2</sub> BTDB	-0.220081 ev	-0.175782 ev
histidine	-0.180861 ev	-0.105947 ev



Fig. S1 IR spectra of JXUST-14 and JXUST-14 after sensing His for 5 cycles at room temperature.



**Fig. S2** (a) View of the coordination modes of BTDB<sup>2-</sup> and 4,4-bpy in **JXUST-14**; (b) the two-fold interpenetrated structure of **JXUST-14**.



Fig. S3 The TGA curve for JXUST-14 under  $N_2$  atmosphere.



**Fig. S4** (a) The simulated and experimental PXRD patterns of **JXUST-14**; (b) The simulated and experimental PXRD patterns of **JXUST-14** after sensing histidine for 5 cycles; (c) The simulated and experimental PXRD patterns of **JXUST-14** after immersing in common organic solvents for 48 h; (d) The simulated and experimental PXRD patterns of **JXUST-14** immersed in aqueous solution with diverse pH for 48 h.



Fig. S5 Solid-state emission spectra of H<sub>2</sub>BTDB and JXUST-14.



Fig. S6 CIE chromaticity diagram displaying the color coordinate of JXUST-14.



Fig. S7 The emission spectra of JXUST-14 in some common solvents.



Fig. S8 The emission spectra of H<sub>2</sub>BTDB ligand and H<sub>2</sub>BTDB@His in EtOH solution.



Fig. S9 Competitive experiments of JXUST-14 in sensing histidine with the interference of other amino acids in EtOH solutions.



**Fig. S10** (a) Emission spectra of **JXUST-14** dispersed in EtOH suspension with various concentrations of histidine; (b) Linear relationship between fluorescence intensity and histidine concentration in a low concentration.



Fig. S11 Time-dependent emission spectra of JXUST-14 after adding His.



Fig. S12 (a) UV-vis absorption spectra of JXUST-14; (b) the optical band gap diagram of JXUST-14 calculated from the UV-vis absorption.



Fig. S13 The luminescence decay curves of JXUST-14 and JXUST-14@ histidine (1 mg JXUST-14 dispersed in 2 mL DMF solution including 5  $\mu$ L His with the concentration of 0.1 M).



Fig. S14 The UV-vis absorption spectra of H<sub>2</sub>BTDB ligand and histidine.