

Supporting Information

**Co-intercalation strategy of constructing partial cation substitution
of ammonium vanadate $\{(\text{NH}_4)_2\text{V}_6\text{O}_{16}\}$ for stable zinc ion storage**

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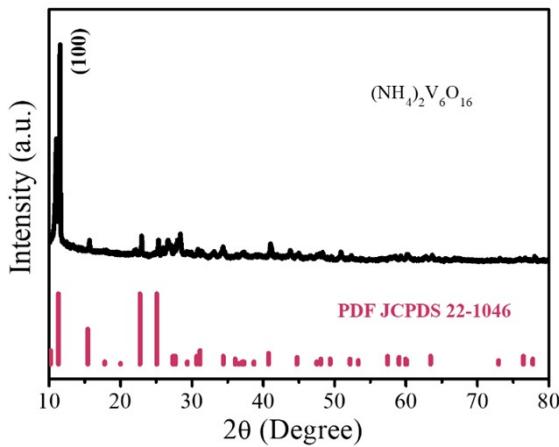


Fig. S1 XRD pattern of NVO.

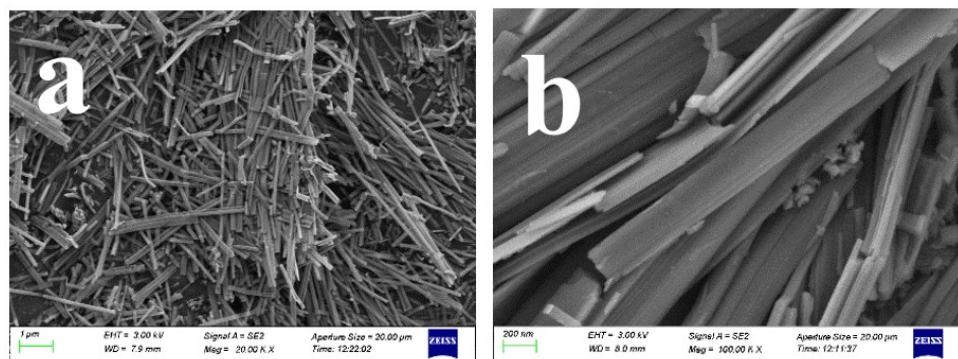


Fig. S2 (a, b) SEM images of NVO.

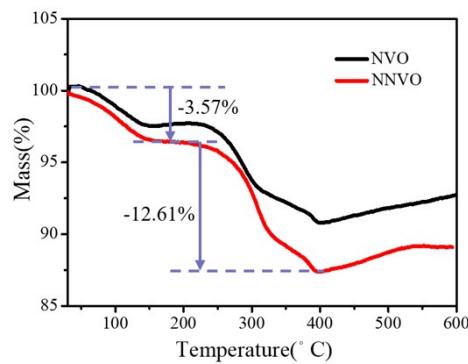


Fig. S3 The TG of NNVO and NVO. The TG analysis was tested in nitrogen.

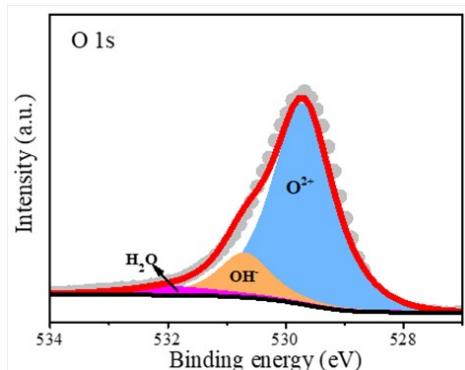


Fig. S4 O 1s regions of the XPS spectra of the NNVO.

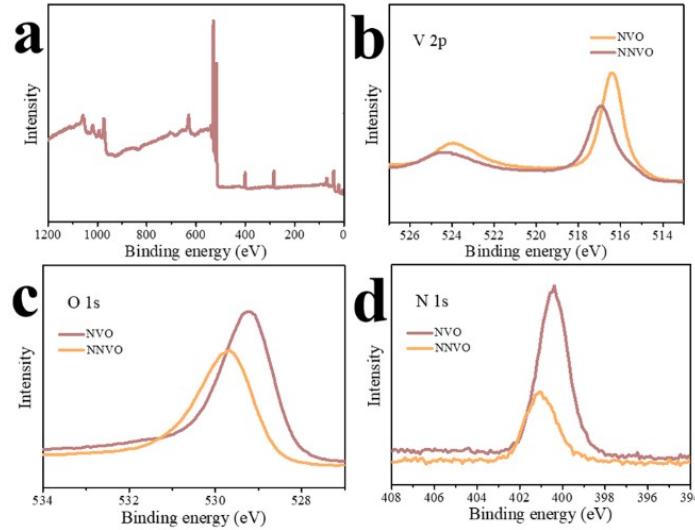


Fig. S5 (a) XPS spectra of the NVO composite. (b) V 2p, (c) N 1s, and (d) Na 1s regions of the XPS spectra of the NNVO and NVO.

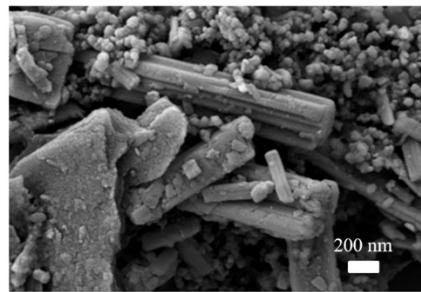


Fig. S6 The SEM image of NNVO after cycles at 0.1 A g^{-1} .

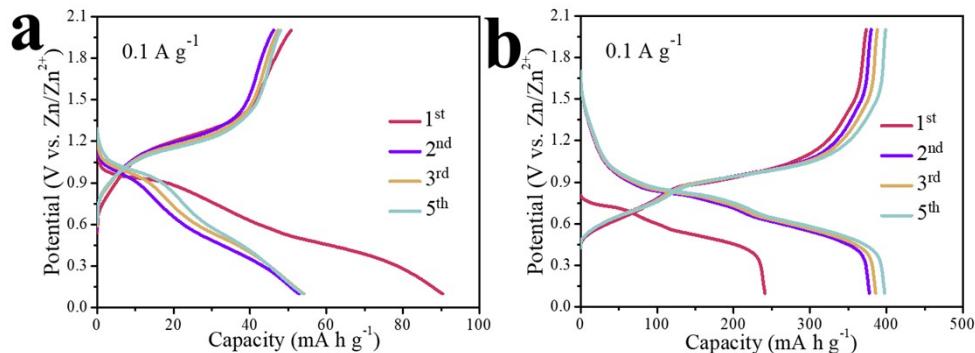


Fig. S7 Galvanostatic discharge-charge curves of V_2O_5 and NVO at 0.1 A g^{-1} , respectively.

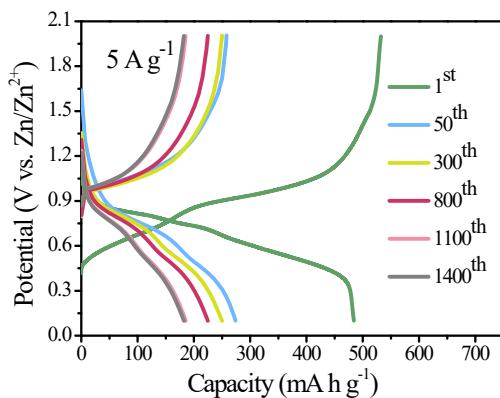


Fig. S8 Galvanostatic discharge-charge curves of NNVO at 5 A g^{-1} .

Table S1. comparison of the electrochemical performance among NNVO and other reported V-based cathode materials.

Materials	Electrolyte	Voltage window (V vs Zn/Zn ²⁺)	The energy density (Wh kg ⁻¹)	Cyclic capacity (mAh g ⁻¹ /th/A g ⁻¹)	Ref.
(NH ₄) ₂ V ₆ O ₁₆	3 M Zn(CF ₃ SO ₃) ₂	0.3-1.7	249	238.7/2000/5	1
CaV ₆ O ₁₆ ·3H ₂ O	3 M Zn(CF ₃ SO ₃) ₂	0.01-2.0	-	230/300/1	2
Na ₂ V ₆ O ₁₆ ·2.14H ₂ O	1 M ZnSO ₄ ·7H ₂ O and 0.2 M Na ₂ SO ₄	0.2-1.6	312	210/500/5	3
Na _{1+x} V ₃ O ₈	2 M ZnSO ₄	0.4-1.4	-	Ca.280/50/1	4
CrVO ₃	3 M ZnSO ₄	0.4-1.7	231.9	85.7/1000/4	5
β-AgVO ₃	1.5 M ZnSO ₄	0.4-1.3	90	95/1000/2	6
NNOD	3 M Zn(CF₃SO₃)₂	0.1-2.0	350.3	423.9/90/0.1;	This work
				182.5/1400/5	

References

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