

Electronic Supplementary Information for

Which is the real oxidant in the competitive ligand self-hydroxylation and substrate oxidations, a biomimetic iron(II)-hydroperoxo species or an oxo-iron(IV)-hydroxy one?

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Scheme S1 Possible reaction pathways of (a) sulfoxidation of thioanisole and (b) hydroxylation of cyclohexane mediated by Fe^{II}-OOH species (**1**).

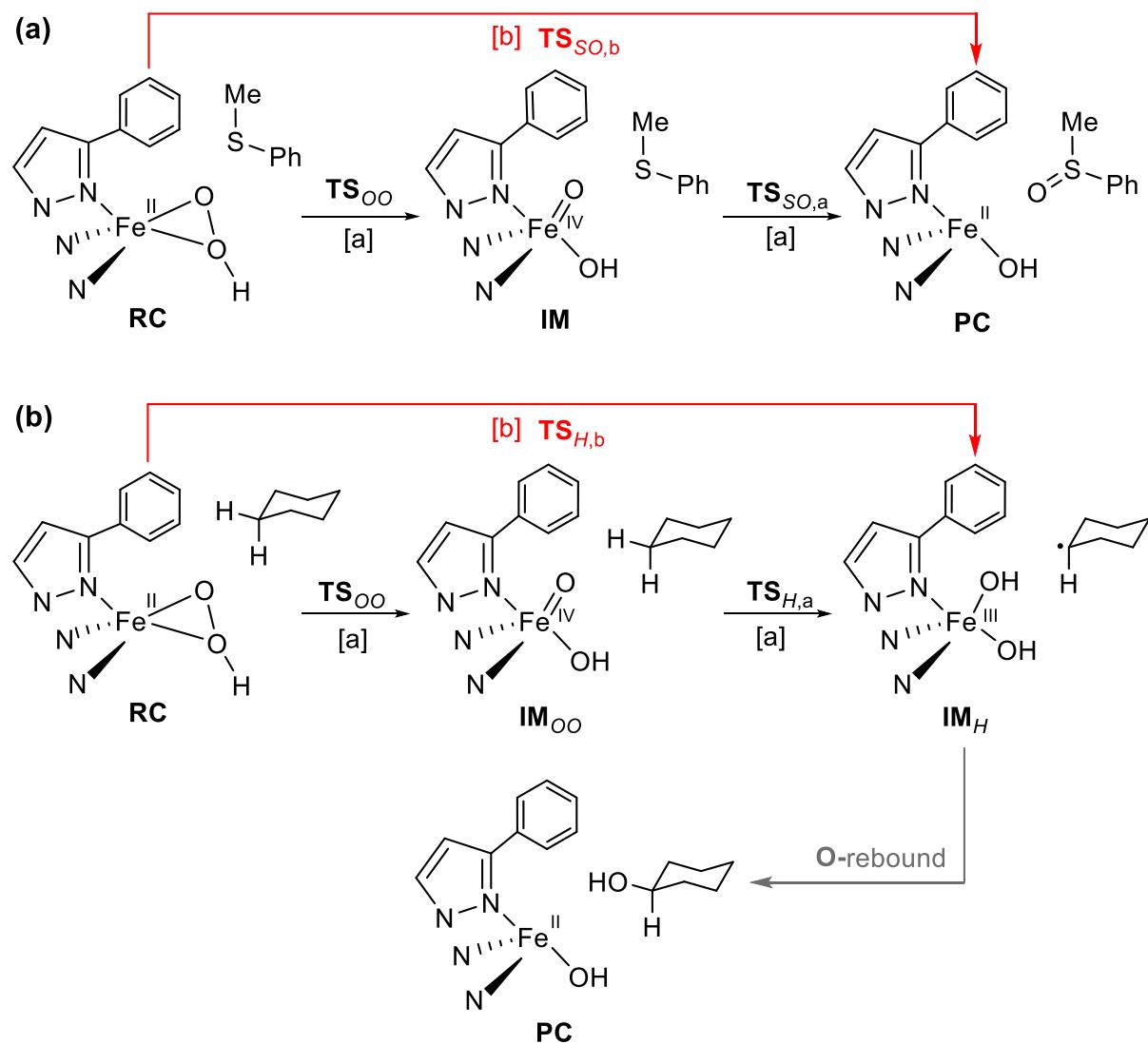


Table S1 Relative energies (in kcal/mol) of the ligand self-hydroxylation of **1** in the absence of any external substrate calculated using UB3LYP functional in solvent

	$\Delta E(B1)$	$\Delta\Delta BS$	$\Delta E(B2)$	$\Delta Z_0(B1)$	$\Delta E(\text{Thermal})^{[a]}$	$-T\Delta S^{[a]}$	$\Delta \text{Disp}^{[b]}$	$\Delta G(B2)^{[c]}$
5¹	0.00	+0.00	0.00	+0.00	+0.00	+0.00	+0.00	0.00
3¹	17.82	+1.16	18.98	+0.50	-0.37	+0.99	-1.30	18.80
5^{TS₁₂}	18.97	+0.07	19.04	-1.24	-0.24	+0.35	-1.09	16.83
3^{TS₁₂}	25.84	+0.78	26.62	-0.50	-0.51	+1.36	-1.40	25.57
5²	-16.17	-0.53	-16.70	+0.44	-0.40	+0.56	-1.09	-17.20
3²	-3.37	-0.24	-3.61	+0.69	-0.55	+1.52	-2.57	-4.51
5^{TS₂₃}	1.16	-1.83	-0.66	-0.88	-0.74	+2.42	-2.88	-2.75
3^{TS₂₃}	4.89	-1.52	3.36	-0.92	-0.69	+2.64	-3.17	1.23
5³	-8.70	-5.27	-13.97	-0.37	-0.70	+2.47	-2.82	-15.39
3³	-8.85	-5.45	-14.30	-0.36	-0.72	+2.81	-2.83	-15.39
5^{TS₃₄}	5.89	-10.62	-4.73	-3.93	-0.19	+0.65	-0.33	-8.52
3^{TS₃₄}	19.66	-10.82	8.84	-3.15	-0.75	+2.43	-1.75	5.62
5^{TS_{N-proton}}	25.13	-7.81	17.32	-4.88	+0.13	+0.69	-3.95	9.05
3^{TS_{N-proton}}	53.68	-7.24	46.44	-4.47	-0.93	+3.48	-4.45	40.08
5⁴	-37.90	-6.85	-44.75	-1.07	+0.18	-0.30	+0.17	-45.78
3⁴	-22.79	-6.73	-29.52	-0.86	-0.51	+2.07	-1.92	-30.74
5^{TS_{4'5}}	-29.59	-7.44	-37.03	-4.25	-1.09	+3.40	-1.35	-40.31
3^{TS_{4'5}}	-12.92	-6.60	-19.52	-3.45	-1.45	+4.61	-3.02	-22.81
5⁵	-66.69	-9.02	-75.72	+0.73	-0.41	+1.44	-1.81	-75.76
3⁵	-53.83	-8.28	-62.11	+1.59	-0.77	+3.00	-3.72	-62.02

Note: B1 represents a small basis set of LACVP(Fe), 6-31+G*(O), 6-31G*(B,N), 6-31G(C,H); B2 represents a larger basis set of LACV3P(Fe), 6-311+G** (others). ^[a] Temperature is 298.15 K. ^[b] Grimme's D3-BJ corrections on the optimized structures. ^[c] Sum of the previous five columns.

Table S2 Mulliken spin densities of various species involved the ligand self-hydroxylation of **1** in the absence of any external substrate calculated at the UB3LYP/LACVP(Fe), 6-31+G*(O), 6-31G*(B,N), 6-31G(C,H) level in solvent

	Fe	O _p	O _{dH}	H	C	Tp ^{Ph,Me}		
⁵ 1	3.78	0.12	0.01	0.00	0.00	0.10		
³ 1	2.02	-0.01	-0.03	0.00	0.00	0.01		
⁵ TS _{I2}	3.51	0.32	0.07	0.00	0.00	0.11		
³ TS _{I2}	2.22	0.04	-0.28	0.00	0.00	0.02		
⁵ 2	3.15	0.62	0.10	0.00	0.00	0.13		
³ 2	1.40	0.61	0.03	0.00	0.00	-0.03		
⁵ TS ₂₃	2.90	0.47	0.05	0.03	-0.15	0.69		
³ TS ₂₃	2.79	-0.46	0.03	-0.03	0.15	-0.48		
⁵ 3	2.85	0.07	0.05	0.06	-0.08	1.05		
³ 3	2.76	0.05	0.05	-0.05	0.06	-0.87		
⁵ TS ₃₄	3.78	0.03	0.14	0.01	0.00	0.03		
³ TS ₃₄	2.18	0.00	0.00	-0.03	0.01	-0.15		
⁵ TS _{N-proton}	3.73	0.04	0.13	0.00	0.00	0.09		
³ TS _{N-proton}	1.98	-0.02	-0.01	0.00	0.00	0.04		
⁵ 4	3.79	-0.01	0.14	0.00	-0.01	0.09		
³ 4	2.50	0.06	0.04	-0.04	-0.23	-0.33		
	Fe	O _p	O _{dH}	H	C	Tp ^{Ph,Me}		
						H(H ₂ O)	OH(H ₂ O)	
⁵ TS _{4'5}	3.80	0.02	0.05	0.00	0.00	0.12	0.00	0.01
³ TS _{4'5}	2.00	-0.03	0.01	0.00	0.00	0.01	0.00	0.00
⁵ 5	3.78	0.06	0.01	0.00	-0.01	0.16	0.00	0.00
³ 5	1.97	0.03	-0.02	0.00	-0.01	0.04	0.00	0.00

Table S3 APT charges of various species involved the ligand self-hydroxylation of **1** in the absence of any external substrate calculated at the UB3LYP/LACVP(Fe), 6-31+G*(O), 6-31G*(B,N), 6-31G(C,H) level in solvent

	Fe	O_p	O_{dH}	H	C	Tp^{Ph,Me}		
⁵ 1	1.74	-0.65	-0.22	0.06	-0.08	-0.85		
³ 1	1.24	-0.43	-0.17	0.04	-0.06	-0.62		
⁵ TS_{I2}	0.99	0.16	-0.12	0.06	-0.11	-0.98		
³ TS_{I2}	0.71	0.30	-0.30	0.07	-0.09	-0.68		
⁵ 2	2.07	-0.70	-0.56	0.07	-0.06	-0.82		
³ 2	1.91	-0.76	-0.59	0.09	-0.07	-0.59		
⁵ TS₂₃	1.67	-0.20	-0.61	0.10	-0.23	-0.73		
³ TS₂₃	2.03	-0.54	-0.65	0.11	-0.23	-0.72		
⁵ 3	2.05	-0.90	-0.64	-0.06	0.50	-0.94		
³ 3	2.10	-0.92	-0.64	-0.06	0.48	-0.95		
⁵ TS₃₄	1.56	-0.79	-0.90	0.23	0.56	-0.66		
³ TS₃₄	1.10	-0.09	-0.92	0.58	0.00	-0.66		
⁵ TS_{N-proton}	1.72	-1.13	-0.83	0.69	0.25	-0.70		
³ TS_{N-proton}	1.66	-1.00	-0.79	0.66	0.20	-0.73		
⁵ 4	1.74	-0.87	-0.86	0.10	0.85	-0.96		
³ 4	1.29	1.02	-0.78	0.05	-0.75	-0.84		
	Fe	O_p	O_{dH}	H	C	Tp^{Ph,Me}		
⁵ TS_{4'5}	1.77	-1.17	-0.84	0.81	0.85	-1.30		
³ TS_{4'5}	1.35	-1.06	-0.77	0.76	0.82	-0.98		
⁵ 5	1.79	-1.38	-0.53	0.62	0.71	-1.16		
³ 5	1.41	-1.32	-0.46	0.62	0.68	-0.87		
	Fe	O_p	O_{dH}	H	C	Tp^{Ph,Me}	H(H₂O)	OH(H₂O)
⁵ TS_{4'5}	1.77	-1.17	-0.84	0.81	0.85	-1.30	0.77	-0.90
³ TS_{4'5}	1.35	-1.06	-0.77	0.76	0.82	-0.98	0.73	-0.85
⁵ 5	1.79	-1.38	-0.53	0.62	0.71	-1.16	0.54	-0.60
³ 5	1.41	-1.32	-0.46	0.62	0.68	-0.87	0.54	-0.59

Table S4 Relative energies (in kcal/mol) of sulfide oxidation by **1** through both routes a and b calculated using UB3LYP functional in solvent

	$\Delta E(B1)$	$\Delta\Delta BS$	$\Delta E(B2)$	$\Delta Z_0(B1)$	$\Delta E(\text{Thermal})^{[a]}$	$-T\Delta S^{[a]}$	$\Delta \text{Disp}^{[b]}$	$\Delta G(B2)^{[c]}$
⁵ RC	0.00	+0.00	0.00	+0.00	+0.00	+0.00	+0.00	0.00
³ RC	18.22	+2.01	20.23	+0.90	-0.54	+3.75	-1.86	22.48
⁵ TS _{oo}	13.94	-0.46	13.48	-1.02	-0.35	+2.49	-1.11	13.49
³ TS _{oo}	25.13	+1.84	26.97	+0.05	-0.77	+4.31	-2.09	28.48
⁵ IM	-16.32	+0.94	-15.38	+0.78	-0.62	+3.72	-3.24	-14.76
³ IM	-1.43	+0.53	-0.90	-0.03	-0.38	+3.61	-4.55	-2.26
⁵ TS _{SO,a}	-4.05	+1.22	-2.83	-1.01	-0.70	+5.88	-10.02	-8.68
³ TS _{SO,a}	2.80	+2.41	5.21	+0.16	-1.27	8.91	-12.27	0.74
⁵ TS _{SO,b}	19.52	+0.63	20.15	-0.49	-0.85	+6.50	-10.50	14.82
³ TS _{SO,b}	36.52	+1.15	37.68	+0.05	-1.21	+8.67	-12.99	32.20
⁵ PC	-36.60	+3.00	-33.60	+0.23	-0.63	+6.36	-10.07	-37.70
³ PC	-12.80	+3.22	-9.58	+0.88	-1.02	+8.05	-12.79	-14.47

Note: B1 represents a small basis set of LACVP(Fe), 6-31+G*(O), 6-31G*(B,N), 6-31G(C,H); B2 represents a larger basis set of LACV3P(Fe), 6-311+G** (others). ^[a] Temperature is 298.15 K. ^[b] Grimme's D3-BJ corrections on the optimized structures. ^[c] Sum of the previous five columns.

Table S5 Mulliken spin densities and APT charges of various species involved in the sulfoxidation by **1** through both routes a and b calculated at the UB3LYP/LACVP(Fe), 6-31+G*(O), 6-31G*(B,N), 6-31G(C,H) level in solvent

	Spin densities						APT charges					
	Fe	O _p	O _{dH}	Tp ^{Ph,Me}	S	Sub-S	Fe	O _p	O _{dH}	Tp ^{Ph,Me}	S	Sub-S
⁵ RC	3.79	0.10	0.00	0.10	0.00	0.00	1.84	-0.68	-0.25	-0.94	-0.25	0.28
³ RC	2.02	-0.01	-0.03	0.01	0.00	0.00	1.31	-0.48	-0.18	-0.66	-0.25	0.26
⁵ TS _{OO}	3.95	0.22	-0.36	0.19	0.00	0.00	1.35	-0.13	-0.30	-0.95	-0.25	0.28
³ TS _{OO}	2.22	0.04	-0.28	0.02	0.00	0.00	0.68	0.37	-0.43	-0.65	-0.24	0.26
⁵ IM	3.16	0.61	0.11	0.12	0.00	0.00	2.09	-0.70	-0.52	-0.86	-0.22	0.22
³ IM	2.77	-0.83	0.03	0.05	-0.01	0.00	2.05	-0.51	-0.57	-0.97	-0.22	0.20
⁵ TS _{SO,a}	3.99	0.07	0.23	0.14	-0.37	-0.07	1.92	0.10	-0.63	-1.19	-0.31	0.12
³ TS _{SO,a}	2.46	-0.06	-0.03	0.01	-0.30	-0.08	0.91	0.78	-0.20	-0.83	-0.54	-0.11
⁵ TS _{SO,b}	3.79	0.10	0.04	0.08	-0.01	0.00	1.66	0.51	-1.15	-0.87	-0.22	0.07
³ TS _{SO,b}	2.06	-0.07	0.01	0.00	0.00	0.00	1.25	0.69	-1.19	-0.57	-0.21	0.04
⁵ PC	3.80	-0.01	0.13	0.07	0.01	0.00	1.76	-1.01	-0.84	-0.90	1.13	-0.14
³ PC	2.04	-0.02	-0.02	-0.01	0.00	0.00	1.38	-0.98	-0.75	-0.69	1.14	-0.10

Table S6 Relative energies (in kcal/mol) of cyclohexane hydroxylation by **1** through both routes a and b calculated using UB3LYP functional in solvent

	$\Delta E(B1)$	$\Delta\Delta BS$	$\Delta E(B2)$	$\Delta Z_0(B1)$	$\Delta E(\text{Thermal})^{[a]}$	$-T\Delta S^{[a]}$	$\Delta \text{Disp}^{[b]}$	$\Delta G(B2)^{[c]}$
⁵ RC	0.00	+0.00	0.00	+0.00	+0.00	+0.00	+0.00	0.00
³ RC	17.56	+1.42	19.06	+0.78	-1.07	+4.03	-1.69	21.12
⁵ TS_{OO}	14.05	-0.65	13.40	-1.16	-0.26	+0.44	-0.65	11.782
³ TS_{OO}	24.62	+1.34	25.96	-0.21	-0.65	+1.85	-1.95	25.00
⁵ IM_{OO}	-16.33	-0.37	-16.70	+0.66	-0.49	+0.34	-1.57	-17.76
³ IM_{OO}	-3.66	+0.30	-3.36	+0.98	-0.70	+2.92	-4.55	-4.72
⁵ TS_{H,a}	7.56	-2.09	5.47	-3.43	-1.05	+5.51	-7.13	-0.63
³ TS_{H,a}	9.74	-1.82	7.92	-3.66	-0.95	+5.36	-7.36	1.32
⁵ TS_{H,b}	44.15	-0.42	43.73	-1.93	-2.10	+8.91	-9.07	39.54
³ TS_{H,b}	30.79	-2.04	28.75	-2.73	-1.81	+7.94	-6.54	25.61
⁵ IM_H	-3.15	-4.16	-7.31	-1.56	-0.22	+2.26	-4.87	-11.69
³ IM_H	-3.04	-4.19	-7.23	-1.62	-0.18	+2.14	-4.81	-11.69
⁵ PC	-55.18	-6.34	-61.52	+0.92	-0.57	+4.44	-8.22	-64.95
³ PC	-36.96	-5.84	-42.80	+2.28	-1.22	+6.96	-0.91	-45.68

Note: B1 represents a small basis set of LACVP(Fe), 6-31+G*(O), 6-31G*(B,N), 6-31G(C,H); B2 represents a larger basis set of LACV3P(Fe), 6-311+G** (others). ^[a] Temperature is 298.15 K. ^[b] Grimme's D3-BJ corrections on the optimized structures. ^[c] Sum of the previous five columns.

Table S7 Mulliken spin densities and APT charges of cyclohexane hydroxylation by **1** through both routes a and b calculated at the UB3LYP/LACVP(Fe), 6-31+G*(O), 6-31G*(B,N), 6-31G(C,H) level in solvent

	Spin densities						APT charges					
	Fe	O _p	O _{dH}	Tp ^{Ph,Me}	H	Sub-H	Fe	O _p	O _{dH}	Tp ^{Ph,Me}	H	Sub-H
⁵ RC	3.79	0.11	0.01	0.10	0.00	0.00	1.79	-0.68	-0.22	-0.91	-0.03	0.04
³ RC	2.03	-0.01	-0.03	0.01	0.00	0.00	1.28	-0.47	-0.16	-0.67	-0.04	0.06
⁵ TS _{oo}	3.94	0.23	-0.36	0.18	0.00	0.00	1.34	-0.09	-0.31	-0.97	-0.02	0.04
³ TS _{oo}	2.23	0.03	-0.28	0.02	0.00	0.00	0.71	0.36	-0.41	-0.68	-0.04	0.06
⁵ IM1	3.16	0.61	0.10	0.13	0.00	0.00	2.07	-0.71	-0.55	-0.82	-0.03	0.05
³ IM1	1.40	0.61	0.02	-0.03	0.00	0.00	1.91	-0.76	-0.58	-0.58	-0.04	0.06
⁵ TS _{H,a}	2.90	0.46	0.04	0.09	-0.04	0.55	1.76	-0.19	-0.53	-0.91	-0.49	0.36
³ TS _{H,a}	2.84	-0.49	0.05	0.07	0.04	-0.51	2.21	-0.53	-0.59	-1.03	-0.42	0.36
⁵ TS _{H,b}	3.79	0.07	0.06	0.09	0.00	-0.02	1.70	0.44	-0.98	-0.93	-0.51	0.27
³ TS _{H,b}	2.05	0.01	-0.07	0.00	0.00	0.02	1.06	0.72	-0.88	-0.63	-0.53	0.26
⁵ IM2	2.86	0.05	0.04	0.08	0.01	0.96	2.04	-0.94	-0.57	-0.86	0.33	0.01
³ IM2	2.84	0.01	0.04	0.08	-0.02	-0.96	2.06	-0.96	-0.59	-0.85	0.34	0.00
⁵ PC	3.80	-0.01	0.13	0.07	0.00	0.00	1.72	-0.82	-0.83	-0.90	0.29	0.54
³ PC	2.00	-0.02	0.03	-0.01	0.00	0.00	1.35	-0.77	-0.77	-0.66	0.30	0.55

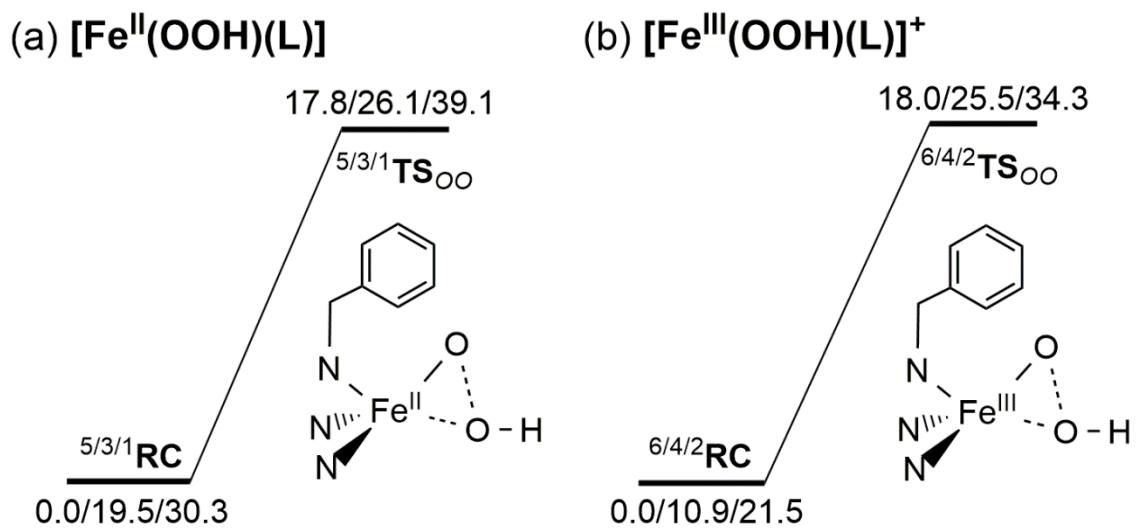


Figure S1 Energy profiles for the O–O bond cleavage of (a) $\text{Fe}^{\text{II}}\text{-OOH}$ and (b) $\text{Fe}^{\text{III}}\text{-OOH}$ species with the same ligand framework.

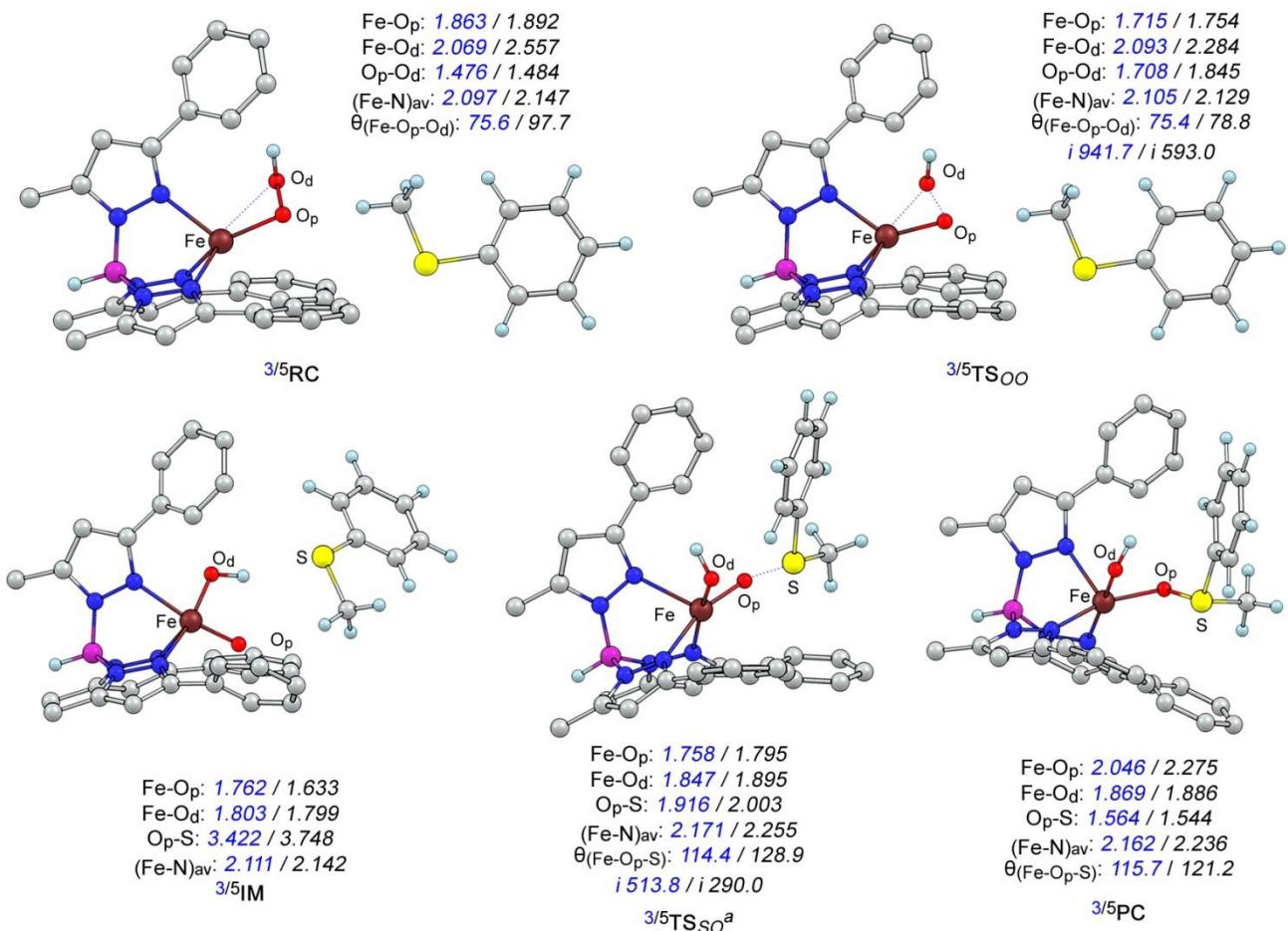


Figure S2 Key geometric information for sulfoxidation by **1** through the preferred route a as shown in Scheme S1. Bond length is in Å, bond angle is in ° and imaginary frequency is in cm⁻¹ unit. All hydrogen atoms in the ligand except that attached to B and O atoms have been omitted for clarity. Calculations were done at the UB3LYP/LACVP(Fe), 6-31+G*(O), 6-31G*(B,N), 6-31G(C,H) level in solvent.

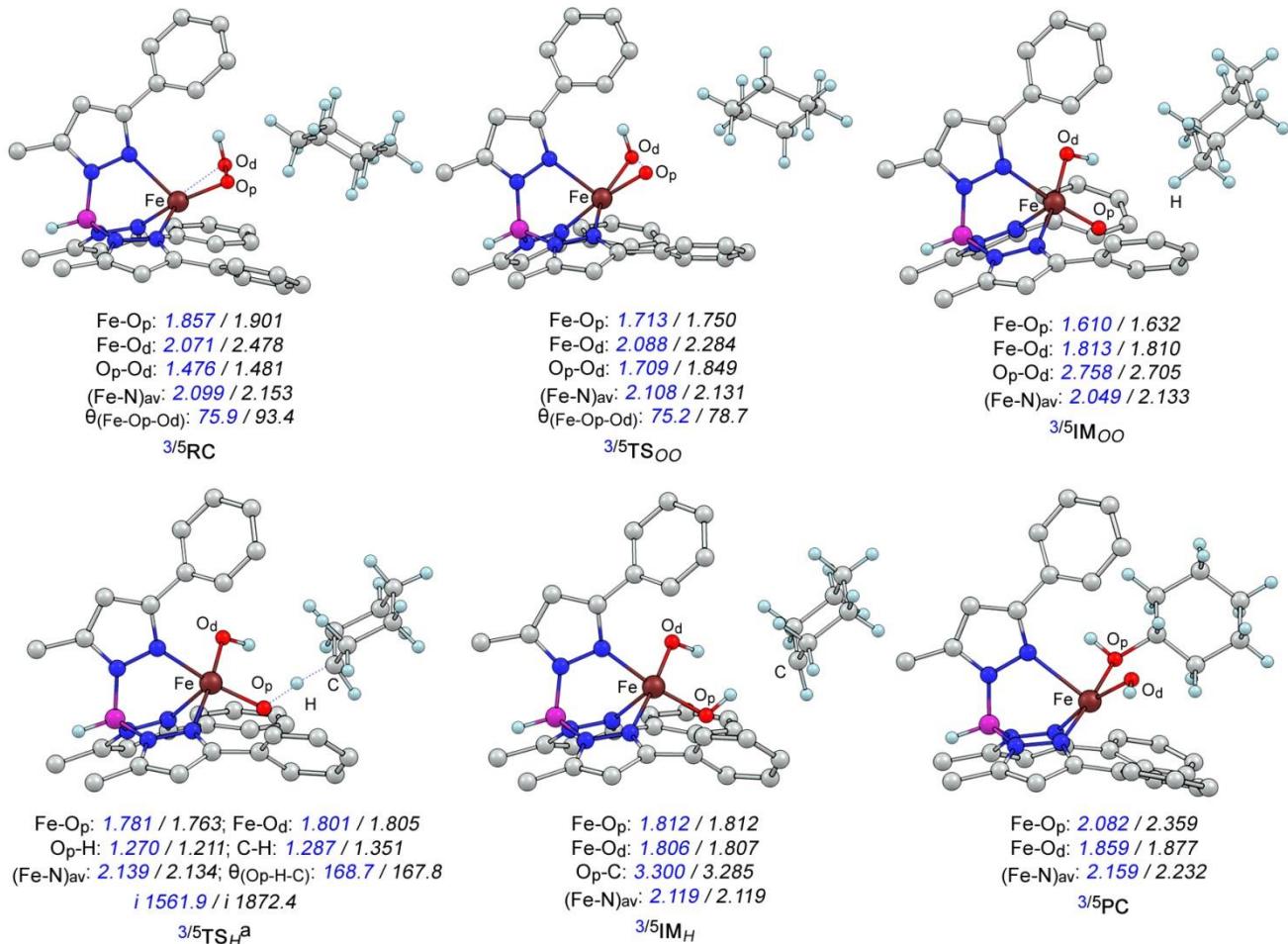


Figure S3 Key geometric information for cyclohexane oxidation by **1** through the preferred route a as shown in Scheme S1. Bond length is in Å, bond angle is in ° and imaginary frequency is in cm⁻¹ unit. All hydrogen atoms in the ligand except that attached to B and O atoms have been omitted for clarity. Calculations were done at the UB3LYP/LACVP(Fe), 6-31+G*(O), 6-31G*(B,N), 6-31G(C,H) level in solvent.

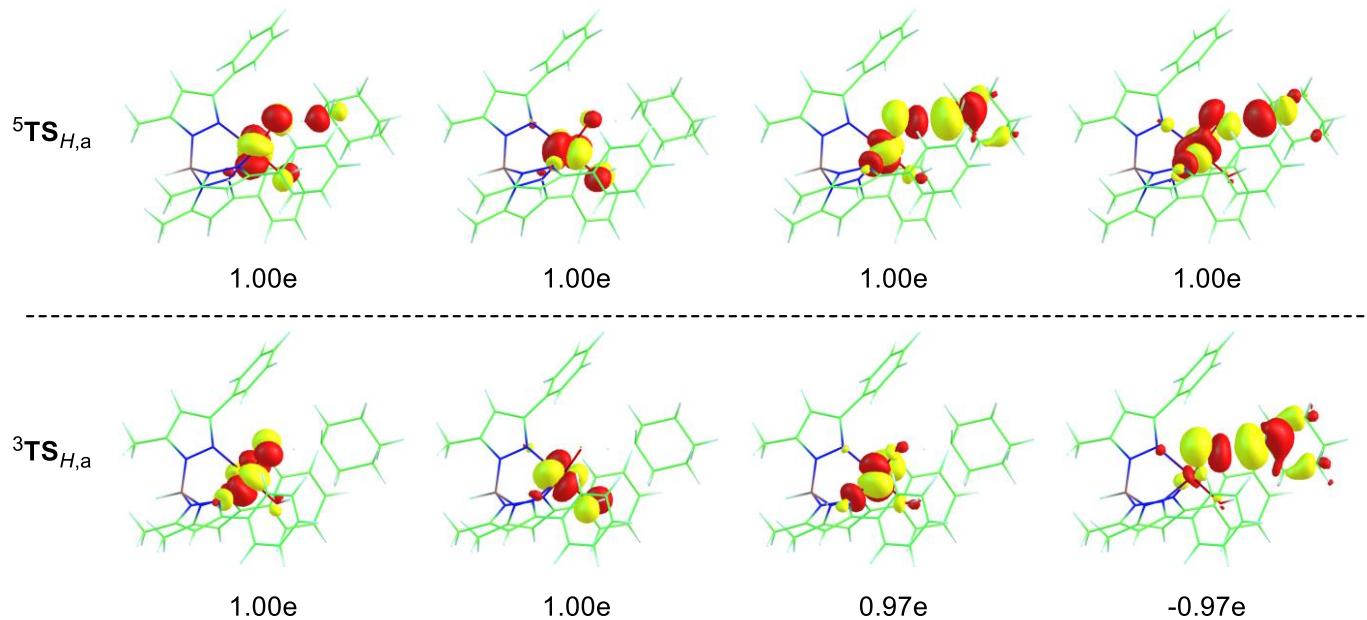


Figure S4 Singly occupied spin-natural bond orbitals (SNOs) of both $^5\text{TS}_{H,a}$ (top) and $^3\text{TS}_{H,a}$ (bottom). Positive value means a spin-up electron and negative value means a spin-down electron.

H -2.720213 2.079823 0.899157	C 3.949011 -1.475465 -1.767788	H -6.758881 -0.122296 -0.922452
H -3.640406 2.136249 -0.608722	C 4.258701 -0.593067 -0.739402	H -6.081946 -0.422655 0.679197
H -4.743183 1.351907 2.158021	H 0.049700 -0.621757 -2.572493	H -5.114689 1.689877 -1.333159
H -5.088674 2.848155 1.293718	H 3.930510 0.968180 1.550818	H -5.954649 2.006279 0.183515
H -6.957394 1.159595 1.053105	H -0.398727 -3.125004 1.143208	H -3.477471 2.380461 0.391481
H -6.254965 1.596496 -0.505752	H -2.378897 -4.612932 1.258337	H -4.031113 1.098332 1.485879
H -5.642709 -0.944713 1.110400	H -4.045601 -4.334837 3.086852	H -2.669049 0.663531 -1.231378
H -6.547273 -0.867634 -0.400189	H -3.721122 -2.546740 4.790266	H -1.424143 1.308853 0.756636
H -4.175631 -1.612407 -0.775875	H -1.747187 -1.050771 4.661915	
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C 1.759353 -2.524426 -2.616377	C -3.891338 1.363419 0.427974	
C 2.109439 -2.663676 -3.974588	C -2.911406 0.370397 -0.206866	
C 1.417677 -3.552351 -4.803213	H -2.735638 -1.724842 -0.666664	
C 0.366061 -4.320633 -4.288593	H -3.596069 -1.382091 0.842190	
C 0.013025 -4.194994 -2.938815	H -4.652125 -0.868191 -2.004270	
C 0.704710 -3.308363 -2.108838	H -5.215298 -2.125434 -0.904951	