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SUPPORTING INFORMATION

Activator-Free Single-Component Co(I)-Catalysts for Regio- and Enantioselective Heterodimerization and Hydroacylation Reactions of 1,3-Dienes. New Reduction Procedures for Synthesis of [L]Co(I)-Complexes and Comparison to *in-situ* Generated Catalysts

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General Methods

All manipulations of moisture- and oxygen-sensitive materials were conducted using standard Schlenk technique under an argon atmosphere or in a glovebox under nitrogen atmosphere maintained at <5 ppm O₂. The Schlenk line was equipped with a vacuum pump maintained between 0.2- and 0.01-mm Hg for all evacuation and drying of air sensitive compounds. All glassware was cleaned by soaking in a base bath overnight, rinsing with water, then soaking in an acid bath (1 h), then water again and finally acetone, then dried in an oven kept at 160 °C overnight. Glass vials used for reactions were purchased from VWR (8 mL, 17 x 60 mm [O.D. x H), screw-thread sample vials, PTFE-faced silicone septa), the caps were dried in a desiccator overnight while the vials were dried in an oven at 140 °C for 24 h before transferring into the glovebox. Gas chromatographic analysis of reaction mixtures was done on an Agilent 6850 Network G.C. equipped with an HP-1 methyl siloxane column (30 m, 0.32 mm I.D, 0.25 μm), H₂ carrier gas, FID at 300 °C or an HP 5890 GC equipped with HP-5MS column (30 m, 0.32 mm I.D, 0.25 µm) and hydrogen as carrier gas with FID-detector at 250 °C. GC-MSD analysis was performed on a 6850 GC-5975 MSD equipped with an El-ionizer. Enantiomeric ratios of chiral compounds were determined by chiral gas chromatographic analyses which were performed on an Agilent 7850 A equipped with a cyclosil-B column, hydrogen carrier gas, using an FID detector at 250°C. Proton, fluorine, and phosphorous nuclear magnetic resonance spectra (¹H, ¹⁹F and ³¹P NMR) were recorded on a Bruker Avance III HD Ascend 600 MHz or 400 MHz. Solvent resonance was used as internal standard (1H NMR, CDCl₃ at 7.26 ppm, C₆D₆ at 7.16 ppm, ¹³C NMR, CDCl₃ at 77.16 ppm, C₆D₆ at 128.06 ppm). Solution state effective magnetic moment was obtained by Evans method (sealed capillary half-filled with 2% CH₂Cl₂ in CD₂Cl₂ was inserted in NMR-tube and ¹H NMR spectrum of given complex was recorded in CD₂Cl₂).

Methylene chloride was distilled over CaH₂ and stored in the glovebox over activated 4Å molecular sieves or used for reactions out of the box when freshly distilled. Et₂O, THF, and hexanes were distilled and stored over molecular sieves for prolonged used in the glovebox. Activated zinc dust was prepared by washing with dilute aqueous HCl (1 – 2 N), washing with distil water several times, and finally with copious amounts of acetone and the resulting solid transferred into a Schlenk flask and all volatiles removed under vacuum. The solid was further dried under vacuum at ca. 90 °C for at least 12 h, then allowed to cool to ambient temperature and transferred into the glovebox. Zinc dust for reactions were prepared by taking chunks of activated zinc and making a powder using a mortar and pistil inside the box. Sodium *tetrakis*-[3,5 *bis*(trifluoromethyl)phenyl]borate (NaBARF) was prepared using reported procedure.¹ All the ligands were obtained from commercial sources.

The single crystal X-ray diffraction studies were carried out on a Nonius Kappa diffractometer equipped with a Bruker APEX-II CCD and Mo K_a radiation ($\lambda = 0.71073$ Å). A desired single crystal suitable for X-ray diffraction were coated with polytoluene oil in a glovebox and was then quickly transferred and mounted on a MiTeGen Micromount with CHRISTO-LUBE MCG 1024 oil. Data were collected in a nitrogen gas stream at 100(2) K using ϕ and ϖ scans. The data were integrated using the Bruker SAINT software program and scaled using the SADABS software program. Solution by direct methods (SHELXT) produced a complete phasing model consistent with the proposed structure.

Preparation of Organic Reductant via Reductive Silylation

All reducing agents were synthesized from commercially available starting materials by reported methods with some modification.²

1,4-Bis(trimethylsilyl)-1,4-dihydropyrazine (11a)



At 0 °C, solution of Pyrazine (5 g, 62.5 mmol) in THF (20 mL) was added dropwise to a mixture of Li (3 equiv., 187.5 mmol, 1.3 g), TMSCI (3 equiv., 187.5 mmol, 24.8 mL) and THF (20 mL) over a period of 1 h. The resulting mixture allowed to come to room temperature and was stirred for 24 h at rt. Upon completion of reaction, the unreacted Li and precipitated LiCl were filtered off under inert atmosphere inside N₂-filled the glovebox. The filtrate was evacuated under high vacuum and solid yellow residue was recrystallized from diethyl ether at -35 °C to obtain the pure titled compound as yellow needles (11.6 g, 82% yield).

¹H NMR (600 MHz, C₆D₆): δ 4.70 (s, 4H), -0.03 (s, 18H); ¹³C NMR (150 MHz, C₆D₆): δ 115.5 (4xC), -1.7(6xC).

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At 0 °C, a solution of 2,3,5,6-tetramethylpyrazine (1g, 7.34 mmol) in THF (10 mL) was added dropwise to a mixture of K (0.86 g, 22 mmol), TMSCI (3 equiv., 22 mmol, 2.8 mL) and THF (10 mL) over a period of 30 min. The resulting mixture allowed to come to room temperature and was stirred for 24 h at rt. Upon completion of reaction, the unreacted K and precipitated KCI were filtered off under inert atmosphere inside N₂-filled the glovebox. The filtrate was evacuated

under high vacuum and solid residue was sublimed by using bulb-to-bulb distillation set up under reduced pressure to afford pure titled compound as a white solid (1.5 g, 72% yield).

¹H NMR (400 MHz, C₆D₆): δ 1.69 (s, 12H), 0.23 (s, 18H); ¹³C NMR (150 MHz, C₆D₆): δ 127.0 (4xC), 19.1 (4xC), 1.8 (6xC).

Preparation of Cobalt(II)-Complexes

All cobalt(II)-complexes were prepared by following our previously reported procedures:³ in the glovebox, anhydrous CoX_2 (0.95 eq) was dissolved in dry distilled THF. In another flask, the ligand (1.0 eq) was dissolved in dry distilled THF and added slowly to a stirring solution of the CoX_2 . After the addition, the resultant mixture was stirred overnight (18 - 24 h), after which the complex may be precipitated. Stirring was stopped and hexanes added to the mixture, shaken thoroughly, and allowed to sit for ~ 10 min. The supernatant was decanted, and the resultant precipitate washed with diethyl ether until the supernatant was almost colorless. Solvent was stripped and the solid dried under high vacuum (~0.1 mm Hg) for 12 to 24 h to afford desired complex.

Preparation of Cobalt(I)-Complexes

Synthesis of [(dppp)(CI)Co[µ-(dppp)]Co(CI)(dppp)] (12) by using 1,4-bis(trimethylsilyl)-1,4-dihydropyrazine



In the glovebox, a 25-mL Schlenk flask was charged with a magnetic stir bar, anhydrous $CoCl_2$ (200 mg, 1.54 mmol, 1 equiv.) and dry freshly distilled THF (0.5 M). The resulting mixture was stirred until all the $CoCl_2$ dissolved to afford a

homogenous solution. In another flask, the ligand, dppp (992 mg, 2.38 mmol, 1.55 equiv.) was dissolved in dry distilled THF (0.5 M) and added slowly to a stirring solution of the CoCl₂. After

the addition, the resultant blue mixture was stirred for 15 minutes followed by the addition of 1,4*bis*(trimethylsilyl)-1,4-dihydropyrazine (696 mg, 3.08 mmol, 2 equiv.). Upon which the color changed from blue to green. The resulting green solution stirred overnight. Stirring was stopped and solvent was stripped, washed with hexanes (1-2 mL), and the solid dried under high vacuum (~0.1 mm Hg) for 2-3 h to afford desired crude complex. The crude complex thus obtained was then dissolved in minimum amount of THF and filtered inside the glovebox. To the filtrate was then diffused hexanes (by slow evaporation of hexane into a THF solution placed in an atmosphere of hexane) to afford the green crystals (958 mg, 87% yield). The green crystal was then characterized by X-ray Crystallography (CCDC # 1873380). Solution state effective magnetic moment μ_{eff} (298 K) = 4.57 B.M.

¹H NMR (600 MHz, CD₂Cl₂): δ 17.54 ($\Delta_{1/2}$ = 94 Hz), 15.47 ($\Delta_{1/2}$ = 11 Hz), 14.93 ($\Delta_{1/2}$ = 154 Hz), 14.46 ($\Delta_{1/2}$ = 120 Hz), 11.15 ($\Delta_{1/2}$ = 75 Hz), 10.16 ($\Delta_{1/2}$ = 149 Hz), 7.49 ($\Delta_{1/2}$ = 87 Hz), 7.41 ($\Delta_{1/2}$ = 69 Hz), 3.68 ($\Delta_{1/2}$ = 22 Hz), 3.28 ($\Delta_{1/2}$ = 54 Hz), 2.79 ($\Delta_{1/2}$ = 34 Hz), 2.43 ($\Delta_{1/2}$ = 80 Hz), 1.82 ($\Delta_{1/2}$ = 22 Hz), 1.27 ($\Delta_{1/2}$ = 74 Hz), 0.13 ($\Delta_{1/2}$ = 29 Hz), 0.09 ($\Delta_{1/2}$ = 20 Hz), -3.94 ($\Delta_{1/2}$ = 111 Hz), -4.95 ($\Delta_{1/2}$ = 56 Hz), -18.70 ($\Delta_{1/2}$ = 6 Hz).

Synthesis of [(dppp)(Cl)Co[µ-(dppp)]Co(Cl)(dppp)] (12) by using Li₃N

By following the same procedure as above with modification of reducing agent, it is prepared as follows:



In the glovebox, a 25-mL Schlenk flask was charged with a magnetic stir bar, anhydrous CoCl₂ (100 mg, 0.77 mmol, 1 equiv.) and dry freshly distilled THF (0.5 M). The resulting mixture was stirred until all the CoCl₂ dissolved to

afford a homogenous solution. In another flask, the ligand, dppp (496mg, 1.19 mmol, 1.55

equiv.) was dissolved in dry distilled THF (0.5 M) and added slowly to a stirring solution of the CoCl₂. After the addition, the resultant blue mixture was stirred for 15 minutes followed by the addition of Li₃N (14 mg, 0.4 mmol, 0.52 equiv.),upon which the color changed from blue to green. The resulting green solution stirred overnight. Stirring was stopped and mixture was filtered out with fret funnel over celite. The resulting filtrate was stripped, washed with hexanes (1-2 mL), and the solid dried under high vacuum (~0.1 mm Hg) for 2-3 h to afford desired crude complex (545 mg, 77% yield). Some portion of the crude complex was then dissolved in minimum amount of THF and filtered inside the glovebox. To the filtrate was then diffused hexanes (by slow evaporation of hexane into a THF solution placed in an atmosphere of hexane) to afford the green crystals. The green crystal was then characterized by X-ray Crystallography. This crystal structure has been previously reported by our group.³

The crystal was an olive-green rectangular rod. All work was done at 150 K. The initial unit cell was determined to be primitive triclinic with cell constants of a = 21.25 Ang., b = 21.49, c = 23.69, alpha = 66.38 deg., beta = 65.92 deg., and gamma = 71.77 deg. Since this cell matches that for our previously determined structure of the same compound (CCDC # 1873380), no data collection was done.

Synthesis of [(dppe)(CI)Co[µ-(dppe)]Co(CI)(dppe)] (13)

In the glovebox, a 20-mL scintillation vial was charged with a magnetic stir bar, anhydrous CoCl₂



(31 mg, 0.24 mmol, 1 equiv.) and dry freshly distilled THF (1 mL). The resulting mixture was stirred until all the $CoCl_2$ dissolved to afford a homogenous solution. In another vial, the ligand, dppe (147 mg, 0.37 mmol, 1.55 equiv.) was dissolved in dry distilled THF (3 mL) and

added slowly to a stirring solution of the CoCl₂. After the addition, the resultant mixture was stirred for 15 minutes followed by the addition of 1,4-*bis*(trimethylsilyl)-1,4-dihydropyrazine (107

mg, 0.48 mmol, 2 equiv.) upon which the color changed from green to brownish yellow. The resulting brownish yellow solution stirred for 6 h. Stirring was stopped, solvent was stripped, washed with hexanes (1-2 mL), and the solid dried under high vacuum (~0.1 mm Hg) for 2-3 h to afford desired crude complex (crude 80% yield). The crude complex thus obtained was then dissolved in minimum amount of THF and filtered inside the glovebox. To the filtrate was then diffused hexanes (by slow evaporation of hexane into a THF solution placed in an atmosphere of hexane) to afford the yellowish green crystals (79 mg, 48% yield). The green crystal was then characterized by X-ray Crystallography (CCDC # 1966496). Solution state effective magnetic moment μ_{eff} (301 K) = 3.46 B.M.

¹H NMR (600 MHz, CD₂Cl₂): δ 13.76 ($\Delta_{1/2}$ = 122 Hz), 11.75 ($\Delta_{1/2}$ = 141 Hz), 10.53 ($\Delta_{1/2}$ = 260 Hz), 3.27 ($\Delta_{1/2}$ = 145 Hz), 2.20 ($\Delta_{1/2}$ = 164 Hz), 1.41 ($\Delta_{1/2}$ = 65 Hz), 0.91 ($\Delta_{1/2}$ = 130 Hz), -4.22 ($\Delta_{1/2}$ = 187 Hz), -5.61 ($\Delta_{1/2}$ = 11 Hz).

Synthesis of [(S,S)-(BDPP)(CI)Co[µ-[(S,S)-(BDPP)]]Co(CI)(S,S)-(BDPP)] (14)



In the glovebox, a 20-mL scintillation vial was charged with a magnetic stir bar, anhydrous $CoCl_2$ (50 mg, 0.38 mmol, 1 equiv) and dry freshly distilled THF (0.5 M). The resulting mixture was stirred until all the $CoCl_2$ dissolved to

afford a homogenous solution. In another vial, the ligand, (*S*,*S*)-BDPP (263 mg, 0.6 mmol, 1.55 equiv.) was dissolved in dry distilled THF (0.5 M) and added slowly to a stirring solution of the CoCl₂. After the addition, the resultant blue mixture was stirred for 15 minutes followed by the addition of 1,4-*bis*(trimethylsilyl)-1,4-dihydropyrazine (174 mg, 0.77 mmol, 2 equiv.). Upon which the color changed from blue to green. The resulting green solution stirred overnight. Stirring

was stopped, solvent was stripped, washed with hexanes (1-2 mL), and the solid dried under high vacuum (~0.1 mm Hg) for 2-3 h to afford desired crude complex (244 mg, 84% yield). The crude complex thus obtained was then dissolved in minimum amount of THF and filtered inside the glovebox. To the filtrate was then diffused hexanes (by slow evaporation of hexane into a THF solution placed in an atmosphere of hexane) to afford the green crystals (160 mg, 55% yield). The green crystal was then characterized by X-ray Crystallography. This crystal structure has been previously reported by our group.³ Solution state effective magnetic moment μ_{eff} (301 K) = 5.25 B.M.

¹H NMR (600 MHz, C₆D₆): δ 17.55 ($\Delta_{1/2} = 62$ Hz), 14.13 ($\Delta_{1/2} = 84$ Hz), 10.74 ($\Delta_{1/2} = 6$ Hz), 9.93 ($\Delta_{1/2} = 31$ Hz), 7.82 ($\Delta_{1/2} = 73$ Hz), 7.76 ($\Delta_{1/2} = 27$ Hz), 7.47 ($\Delta_{1/2} = 25$ Hz), 7.39 ($\Delta_{1/2} = 44$ Hz), 6.72 ($\Delta_{1/2} = 67$ Hz), 6.40 ($\Delta_{1/2} = 27$ Hz), 6.21 ($\Delta_{1/2} = 62$ Hz), 3.57 ($\Delta_{1/2} = 24$ Hz), 3.42 ($\Delta_{1/2} = 80$ Hz), 3.08 ($\Delta_{1/2} = 53$ Hz), 2.99 ($\Delta_{1/2} = 36$ Hz), 2.91 ($\Delta_{1/2} = 13$ Hz), 1.91 ($\Delta_{1/2} = 24$ Hz), 1.41 ($\Delta_{1/2} = 25$ Hz), 1.22 ($\Delta_{1/2} = 78$ Hz), 1.03 ($\Delta_{1/2} = 76$ Hz), 0.95 ($\Delta_{1/2} = 40$ Hz), 0.89 ($\Delta_{1/2} = 51$ Hz), 0.12 ($\Delta_{1/2} = 51$ Hz), -2.10 ($\Delta_{1/2} = 4$ Hz), -4.23 ($\Delta_{1/2} = 6$ Hz), -7.60 ($\Delta_{1/2} = 4$ Hz), -11.84 ($\Delta_{1/2} = 4$ Hz), -17.07 ($\Delta_{1/2} = 9$ Hz), -17.29 ($\Delta_{1/2} = 4$ Hz).

Synthesis of $\{[(R,R)-QuinoxP^*]cobalt(\mu-Br)\}_2$ (18)



In the glovebox, a 20 mL scintillation vial was charged with a magnetic stir bar, anhydrous CoBr₂ (50 mg, 0.145 mmol, 1 equiv.) and dry freshly distilled THF (3 mL). The resulting mixture was stirred until all the CoBr₂ dissolved to afford a

homogenous solution. In another vial, the ligand, R-QuinoxP* (50 mg, 0.145 mmol, 1.0 equiv.) was dissolved in dry distilled THF (2 mL) and added slowly to a stirring solution of the CoBr₂.

After the addition, the resultant green mixture was stirred for 3 h followed by the addition of 1,4*bis*(trimethylsilyl)-1,4-dihydropyrazine (33 mg, 0.145 mmol, 1 equiv.). Upon which the color changed from green to deep green. The resulting green solution stirred overnight. Stirring was stopped and solvent was stripped and the solid dried under high vacuum (~0.1 mm Hg) for 2 to 3 h to afford desired crude complex (65 mg, crude yield 95%). The crude complex thus obtained was then dissolved in minimum amount of hexanes and filtered inside the glovebox. The filtrate was kept at -25 °C overnight to afford the green crystals (55 mg, 80% yield). The green crystal was then characterized by X-ray Crystallography (CCDC # 1966498). Solution state effective magnetic moment μ_{eff} (301 K) = 4.70 B.M.

¹H NMR (600 MHz, C₆D₆): δ 26.25 ($\Delta_{1/2}$ = 73 Hz), 9.71 ($\Delta_{1/2}$ = 13 Hz), 8.84 ($\Delta_{1/2}$ = 11Hz), 3.04 ($\Delta_{1/2}$ = 67 Hz), 0. 29 ($\Delta_{1/2}$ = 3 Hz), 0.12 ($\Delta_{1/2}$ = 33 Hz), 0.04 ($\Delta_{1/2}$ = 4 Hz).

Synthesis of {[(S,S)-Ph-BPE]cobalt(µ-Br)}₂ (19)

Method A, in situ generation of the precursor Co(II) complex: In the glovebox, a 20 mL



scintillation vial was charged with a magnetic stir bar, anhydrous $CoBr_2$ (44 mg, 0.2 mmol, 1 equiv.) and dry freshly distilled THF (2 mL). The resulting mixture was stirred until all the $CoBr_2$ dissolved to afford a homogenous solution. In another vial, the ligand, (*S*,*S*)-Ph-BPE(106 mg, 0.21 mmol, 1.05

equiv.) was dissolved in dry distilled THF (3 mL) and added slowly to a stirring solution of the CoBr₂. After the addition, the resultant purple mixture was stirred for 2 h followed by the addition of 1,4-*bis*(trimethylsilyl)-1,4-dihydropyrazine (45.3 mg, 0.2 mmol, 1 equiv.). Upon which the color changed from purple to green. The resulting bluish green solution stirred overnight.

Stirring was stopped and solvent was stripped, washed with hexanes (3-4 mL), and the solid dried under high vacuum (~0.1 mm Hg) for 2 to 3 h to afford desired crude complex (132 mg, yield 94%). The crude complex thus obtained was then dissolved in minimum amount of THF and filtered inside the glovebox. To the filtrate was then diffused hexanes (by slow evaporation of hexane into a THF solution placed in an atmosphere of hexane) to afford the green crystals. The green crystal was then characterized by X-ray Crystallography (CCDC # 1966499).

Method B, Use of isolated Co(II) precursor: In the glovebox, a 20 mL scintillation vial was charged with a magnetic stir bar, isolated (Ph-BPE)CoBr₂ (15 mg, 0.021 mmol, 1 equiv.) and dry freshly distilled THF (1 mL). To the resultant solution, 1,4-*bis*(trimethylsilyl)-1,4-dihydropyrazine (13 mg, 0.058 mmol, 2 equiv.) was added, upon which the color changed from purple to green. The resulting bluish green solution stirred overnight. Stirring was stopped and solvent was stripped, washed with hexanes (1 mL), and the solid dried under high vacuum (~0.1 mm Hg) for 2 to 3 h to afford desired crude complex (crude yield 91%). The crude complex thus obtained was then dissolved in minimum amount of THF and filtered inside the glovebox. To the filtrate was then diffused hexanes (by slow evaporation of hexane into a THF solution placed in an atmosphere of hexane) to afford the green crystals (7.9 mg, 52% yield). Solution state effective magnetic moment μ_{eff} (301 K) = 3.85 B.M.

¹H NMR (600 MHz, C₆D₆): δ 67.86($\Delta_{1/2}$ = 33 Hz), 51.61($\Delta_{1/2}$ = 6 Hz), 10.83($\Delta_{1/2}$ = 107 Hz), 8.99 ($\Delta_{1/2}$ = 4 Hz), 7.35 ($\Delta_{1/2}$ = 22 Hz), 7.25($\Delta_{1/2}$ = 36 Hz), 7.10 ($\Delta_{1/2}$ = 36 Hz), 7.00 ($\Delta_{1/2}$ = 47 Hz), 6.91 ($\Delta_{1/2}$ = 49Hz), 4.71 ($\Delta_{1/2}$ = 113Hz), 4.45 ($\Delta_{1/2}$ = 6 Hz), 3.67 ($\Delta_{1/2}$ = 33Hz), 3.57 ($\Delta_{1/2}$ = 24 Hz), 3.33 ($\Delta_{1/2}$ = 24 Hz), 3.32 ($\Delta_{1/2}$ = 7 Hz), 3.23 ($\Delta_{1/2}$ = 58 Hz), 3.00 ($\Delta_{1/2}$ = 13 Hz), 2.99 ($\Delta_{1/2}$ = 9 Hz), 2.98 ($\Delta_{1/2}$ = 22Hz), 2.69 ($\Delta_{1/2}$ = 49 Hz), 2.10 ($\Delta_{1/2}$ = 60Hz), 1.90 ($\Delta_{1/2}$ = 67Hz), 1.76 ($\Delta_{1/2}$ = 44 Hz), 1.66 ($\Delta_{1/2}$ = 7 Hz), 1.65 ($\Delta_{1/2}$ = 9 Hz), 1.64 ($\Delta_{1/2}$ = 7 Hz), 1.54 ($\Delta_{1/2}$ = 82 Hz), 1.42 ($\Delta_{1/2}$ = 29 Hz), 1.26 ($\Delta_{1/2}$ = 7 Hz). Synthesis of $\{[(R,R)-iPr-DuPhos]cobalt(\mu-Cl)\}_2$ (20)⁴



In the glovebox, a 20 mL scintillation vial was charged with a magnetic stir bar, anhydrous CoCl₂ (75 mg, 0.58 mmol, 1 equiv.) and dry freshly distilled THF (5 mL). The resulting mixture was stirred until all the CoCl₂ dissolved to afford a homogenous solution. In another vial, the ligand, ^{*i*}Pr-DuPhos

(250 mg, 0.58 mmol, 1.0 equiv.) was dissolved in dry distilled THF (3 mL) and added slowly to a stirring solution of the CoCl₂. After the addition, the resultant green mixture was stirred for 2 h followed by the addition of 1,4-*bis*(trimethylsilyl)-1,4-dihydropyrazine (131 mg, 0.58 mmol, 1 equiv.), upon which the color changed from green to brownish orange. The resulting brown orange solution stirred overnight. Stirring was stopped, solvent was stripped, and the solid dried under high vacuum (~0.1 mm Hg) for 2 to 3 h to afford desired crude complex (280 mg, 94% yield). The crude complex thus obtained was then dissolved in minimum amount of diethyl ether and filtered inside the glovebox. The filtrate was kept at -25 °C overnight to afford the brown crystals (256 mg, 86% yield). The brown crystal was then characterized by X-ray Crystallography. Solution state effective magnetic moment μ_{eff} (301 K) = 3.93 B.M. This crystal structure has been reported in the literature by Chirik et al.⁴

¹**H NMR (600 MHz, C₆D₆):** δ 66.30 ($\Delta_{1/2}$ = 38 Hz), 57.24($\Delta_{1/2}$ = 51 Hz), 14.10 ($\Delta_{1/2}$ = 18 Hz), 9.19($\Delta_{1/2}$ = 24 Hz), 9.06($\Delta_{1/2}$ = 9 Hz), 2.35($\Delta_{1/2}$ = 31 Hz), 1.02($\Delta_{1/2}$ = 25 Hz), 0.90($\Delta_{1/2}$ = 11 Hz), 0.73 ($\Delta_{1/2}$ = 18Hz), 0.72 ($\Delta_{1/2}$ = 35 Hz), 0.59 ($\Delta_{1/2}$ = 53 Hz), -1.04 ($\Delta_{1/2}$ = 44 Hz), -1.53($\Delta_{1/2}$ = 25 Hz), Hz),, -2.69 ($\Delta_{1/2}$ = 45 Hz),-3.96 ($\Delta_{1/2}$ = 45Hz)

NMR matches with the complex reported by Chirik et al.⁴

Synthesis of [bis(N-aryliminoethyl- κ N,N')pyridine- κ N]CoCl (22)



In the glovebox, a 20 mL scintillation vial was charged with a magnetic stir bar, isolated *Bis*(imino)pyridylCoCl₂ (150 mg, 0.24 mmol, 1 equiv.) and dry freshly distilled THF (3 mL). To the resultant yellow solution, 1,4*bis*(trimethylsilyl)-1,4-dihydropyrazine (30 mg, 0.132 mmol, 0.55 equiv) was added. Upon which the color changed

from yellow to purple. The resulting purple solution stirred for 3 hr. Stirring was stopped and solvent was stripped, washed with hexanes (1-2 mL), and the solid dried under high vacuum (~0.1 mm Hg) for 2 to 3 h to afford desired crude complex (133 mg, 94% yield). The crude complex thus obtained was then dissolved in minimum amount of THF and filtered inside the glovebox. To the filtrate was then diffused hexanes (by slow evaporation of hexane into a THF solution placed in an atmosphere of hexane) and kept at -25 °C overnight to afford the purple crystals (115 mg, 81% yield). The purple crystal was then characterized by X-ray Crystallography (CCDC # 1966500).

¹**H** NMR (600 MHz, C₆D₆): δ 9.56 ($\Delta_{1/2}$ = 27 Hz), 8.49 ($\Delta_{1/2}$ = 33 Hz), 8.43 ($\Delta_{1/2}$ = 49 Hz), 7.44($\Delta_{1/2}$ = 25 Hz), 7.30 ($\Delta_{1/2}$ = 22 Hz), 7.15($\Delta_{1/2}$ = 13 Hz), 6.93($\Delta_{1/2}$ = 20 Hz), 3.56($\Delta_{1/2}$ = 25Hz), 3.35($\Delta_{1/2}$ = 29 Hz), 2.91($\Delta_{1/2}$ = 34Hz), 2.27 ($\Delta_{1/2}$ = 49 Hz), 1.41 ($\Delta_{1/2}$ = 24 Hz), 1.19 ($\Delta_{1/2}$ = 20 Hz), 1.08 ($\Delta_{1/2}$ = 20 Hz), 0.88 ($\Delta_{1/2}$ = 29 Hz), 0.84 ($\Delta_{1/2}$ = 20 Hz), 0.07 ($\Delta_{1/2}$ = 11 Hz), -1.18 ($\Delta_{1/2}$ = 44 Hz), -8.93 ($\Delta_{1/2}$ = 36 Hz), -17.83 ($\Delta_{1/2}$ = 7 Hz), -18.37 ($\Delta_{1/2}$ = 34 Hz).

¹³C NMR (150 MHz, CD₂Cl₂): δ 167.86, 152.56, 150.63, 140.82, 126.39, 125.82, 123.32, 115.69, 28.65, 23.54, 23.09, 21.36.

NMR matches with the complex reported by Gal et al.⁵

Attempts towards the synthesis of Co(0)-Complexes



In the N₂-filled glovebox, a 20 mL scintillation vial was charged with a magnetic stir bar, isolated (*R*)-iPr-DuPhosCoBr₂ (20 mg, 0.031 mmol, 1 equiv.) and 1,4-*bis*(trimethylsilyl)-1,4dihydropyrazine, **11a** (36 mg, 0.16 mmol, 5 equiv.) and dry freshly distilled THF (1 mL). Upon letting it stir for ca. 2 min, 1,5-cyclooctadiene (COD) (40 μ L, 0.31 mmol, 10 equiv.) was added to the resulting solution via micro-syringe. The resulting brown solution stirred for 24 hr. Stirring was stopped, solvent was stripped, and the solid dried under high vacuum (~0.1 mm Hg) for 5 to 6 h to afford desired crude complex (15 mg, 86% yield). The ¹H-NMR recorded for crude solid in C₆D₆ and it did not show any peaks corresponding to ¹H-NMR spectrum of previously reported [(*R*)-iPr-DuPhosCo⁽⁰⁾(COD)] complex by Chirik et al.⁴ It showed the peaks analogues to [(*R*)-iPr-DuPhosCo⁽¹⁾(μ Cl)]₂ which suggests that there is only formation of [(*R*)-iPr-DuPhosCo⁽¹⁾(Br)]₂.

¹H NMR (600 MHz, C₆D₆): δ 63.41 ($\Delta_{1/2}$ = 107 Hz), 52.74($\Delta_{1/2}$ = 164 Hz), 14.04 ($\Delta_{1/2}$ = 27 Hz), 9.13 ($\Delta_{1/2}$ = 7 Hz), 8.74 ($\Delta_{1/2}$ = 27 Hz), 2.51 ($\Delta_{1/2}$ = 38 Hz), 1.40 ($\Delta_{1/2}$ = 23 Hz), 1.04 ($\Delta_{1/2}$ = 11 Hz), 0.88 ($\Delta_{1/2}$ = 0 Hz), -0.99 ($\Delta_{1/2}$ = 61 Hz), -1.38 ($\Delta_{1/2}$ = 38 Hz), -2.18 ($\Delta_{1/2}$ = 65 Hz), -4.24 ($\Delta_{1/2}$ = 137 Hz).

Using THF/MeOH solvent:



In the glovebox, a 20 mL scintillation vial was charged with a magnetic stir bar, isolated (R)-iPr-DuPhosCoBr₂ (20 mg, 0.031 mmol, 1 equiv.) and 1,4-*bis*(trimethylsilyl)-1,4dihydropyrazine, **11a** (72 mg, 0.31 mmol, 10 equiv.), dry freshly distilled THF (0.5 mL) and anhydrous MeOH (0.5 mL). Upon letting it stir for ca. 2 min, 1,5-cyclooctadiene (40 μ L, 0.31 mmol, 10 equiv.) was added to the resulting solution via micro-syringe. The resulting brown solution stirred for 48 hr. Stirring was stopped, solvent was stripped, and the solid dried under high vacuum (~0.1 mm Hg) for 5 to 6 h to afford desired crude complex (32 mg). The crude solid remains insoluble in C₆D₆ as well as THF-D₈. ¹H-NMR recorded for crude solid in C₆D₆ and THF-D₈ only showed peak correspond to free ligand [(*R*)-iPr-DuPhos]. ³¹P also showed peak corresponding to free ligand.

Preparation of Cationic Cobalt(I) Complexes

{[(*R*,*R*)-QuinoxP]cobalt(η^6 -C₆D₆)}⁺ [BARF]⁻ (23)



In N₂-filled glovebox, 20 mL scintillation vial was charged with a magnetic stir bar, {[(R,R)-QuinoxP*]cobalt(µ-Br)}₂ **18** (25 mg, 0.026 mmol, 1 equiv.), NaBARF(49 mg, 0.055 mmol, 2.1 equiv.), and 2 mL of C₆D₆. The color changed from green to purple. The resulting purple solution was stirred for 3 h at room temperature inside the glovebox. Upon letting it sit for 5 minutes, C₆D₆ was decanted. The crude complex then washed with 3 mL of hexanes and residual solvents were stripped down to obtain purple solid **23** (55 mg, 78% yield). The saturated solution of **23** in C₆D₆ leads to solid purple crystal at room temperature which was then characterized by X-ray crystallography (CCDC # 2016105).

¹H NMR (600 MHz, C₆D₆): δ 11.45 (4H), 8.82 (2H), 8.34 (8H), 8.04 (2H), 7.84 (1H), 7.64 (4H), 7.58 (1H), 7.29 (1H), 7.03 (1H), 6.41 (1H), 5.63 (1H), 5.51 (3H), 3.57 (2H), 3.46(1H), 3.27 (2H), 3.07 (6H), 1.35 (12H), 1.21(4H), 0.89 (6H), 0.77 (1H), 0.56 (1H), 0.41 (9H), 0.29 (5H), 0.12 (7H), 0.03 (7H) ; ¹³C NMR (150 MHz, C₆D₆): δ 162.56 (q, ¹*J*_{B-C} = 49.6 Hz), 135.42, 133.04, 129.84, 129.78, 126.14, 124.33, 118.08, 34.98, 26.94, 26.39, 25.64; ³¹P NMR (243 MHz, C₆D₆): δ 58.19; ¹⁹F NMR (565 MHz, C₆D₆): δ -62.05.

¹**H NMR (600 MHz, CD₂Cl₂):** δ 14.87 (bs, 2H), 8.21 (q, *J* = 3.3 Hz, 1H), 7.95 (q, *J*= 3.3 Hz, 1H), 7.69 (bs, 8H), 7.54 (bs, 4H), 4.20-3.40 (m, 9H), 1.95 (virtual coupling, t, J = 5.1 Hz, 6H), 1.00 (t, *J* = 7.4 Hz, 9H) ; ¹³**C NMR (150 MHz, CD₂Cl₂):** δ 162.12 (q, ¹*J*_{B-C} = 49.8 Hz), 155.28, 135.20, 133.26, 130.40, 129.25, (q, ²*J*_{C-F} = 29.5 Hz), 124.99 (q, ¹*J*_{C-F} = 272.4 Hz), 117.87, 37.55, 27.89, 27.24, 12.47; ³¹**P NMR (243 MHz, C₆D₆):** δ 58.24.

Synthesis, isolation, and spectral properties of {[(R,R)-'Pr-DuPhos]cobalt- η^4 -(2,3-dimethylbutadiene))}⁺ [BARF]⁻ (**25**) has been recently reported by our group.⁶

UV-Vis Spectroscopy of the Co(II) and Co(I) complexes

A background of THF was taken. All the samples were made inside the N₂-filled glovebox with appropriate concentration in THF and transferred to modified cuvette that allowed the preservation of air sensitive materials. The cuvette was placed in the in-situ UV- VIS and a wavelength scan was performed from 1100 nm to 190 nm.



Figure S1. UV-Vis Spectra of 18, 19, and 20; Each of them showing two characteristic peaks in visible region (which are the characteristic for d⁸ system).



Figure S2. UV-Vis Spectra for relative rate of reduction; This study clearly shows that 11a is much better reductant as compared to Zn.



Figure S3. Sensitivity of the Co(I) complexes to oxidation. UV-Vis spectra of 22 while expose to air

	0 min	5 min	15 min	25 min
	(% conversion)	(% conversion)	(% conversion)	(% conversion)
ZnBr ₂	0.0 ± 0.0	0.0 ± 0.0	0.0 ± 0.0	5.3 ± 3.1
(average)				
NaBArF (average)	0.0 ± 0.0	38.0 ± 7.3	70.0 ± 4.3	89.3 ± 3.3
$ZnBr_2$ (trial 1)	0.0	0.0	0.0	1.0
$ZnBr_2$ (trial 2)	0.0	0.0	0.0	7.0
$ZnBr_2$ (trial 3)	0.0	0.0	0.0	8.0
NaBArF (trial 1)	0.0	48.0	68.0	90.0
NaBArF (trial 2)	0.0	31.0	66.0	85.0
NaBArF (trial 3)	0.0	35.0	76.0	93.0

Table S1. Effect of counter ion on the rate of heterodimerization: ZnBr2 vs NaBArF



Figure S4. Graph showing effect of counter ion on heterodimerization

Adapted from Gray, M. et al. Mechanism of Cobalt-Catalyzed Heterodimerization of Acrylates and 1,3-Dienes. A Potential Role of Cationic Cobalt(I) Intermediates. ACS Catal. 10, 4337-4348 (2020).

General procedures for catalytic reactions

Procedure A1. Typical procedure for checking the viability of reducing agents via hydrovinylation of linear diene [(*E*)-1,3-nonadiene].^{7,8} (For Table 2 in the paper and Table S2)

In N₂ filled glovebox, a 25-mL Schenk flask was charged with a magnetic stir bar, $[Co(dppp)Cl_2]$ (5.5 mg, 0.01 mmol, 0.05 equiv.), reductant (X equiv., as mentioned the in table S1) and NaBARF (18 mg, 0.02 mmol, 0.1 equiv.). The flask was capped with rubber septum and taken outside of the box. It was then connected to a Schlenk line with argon and subjected to vacuum then refill (3 cycles). Freshly distilled methylene chloride (0.2M) was added via septum and the mixture stirred for 10 minutes. The flow control stopcock was closed, and an ethylene balloon inserted using a needle via the septum. A 50-mL syringe was used to evacuate the flask (3 x 25 mL) to remove residual argon. At the room temperature, (*E*)-1,3-nonadiene (25 mg, 0.2 mmol, 1 equiv.) was added neat via a microliter syringe. Progress of the reaction was monitored via GC-FID by taking an aliquot using a syringe and long needle, diluted with pentanes or hexanes and filtered through a shot pad of silica in a glass pipette. Upon completion of reaction, it was further concentrated on a rotary evaporator to afford crude product. The crude hydrovinylation product was subjected to column chromatography using pentane to obtain the product as a clear oil.

GC (methyl silicone, 80 °C, 3 min, 20 °C/min \rightarrow 250 °C): Rt major [4,1-adduct, branch], 3.1 min (80%); minor [1,4-adduct, linear], 3.8 min (20%)

Table S2: Optimization of unconventional reducing agents via Co(I)-catalyzed hydrovinylation of diene^a



Entry	Reductant (equiv)	Time (min)	Conversion (%) ^b	[4,1]:[1,4]
1	Zn (0.5)	30	100	78:22
2	11a (0.05)	75	100	79:20
3	11b (0.05)	45	100	80:20
4	11c (0.5)	24x60	4	-
5	Li ₃ N (0.10)	90	100	80:20

^a **Procedure A1** was followed with some modification in reducing agent as mentioned.

^b Conversion is based on GC analysis.

Procedure A2. Typical procedure for using lithium nitride as a reductant for Co(II) in heterodimerization of 1,3-diene and methyl acrylate.

In the N₂-filled glovebox, an 8-mL oven dried vial with a septum cap was charged with magnetic stir bar, isolated metal complex (0.05 equiv.), reducing agent (0.10 - 0.25 equiv.), activator (0.10 - 0.30 equiv.) and DCM (0.35 M). The vial was capped and while stirring the mixture, neat (E)-1,3-undecadiene (1.00 equiv.) followed by distilled methyl acrylate (1.10 equiv.) was added via micro-liter syringe. The resulting green solution was allowed to stir at room temperature and monitored via GC-FID by taking an aliquot with a glass pipette, removing from dry box atmosphere, adding 1:1 diethyl ether/hexanes, filtering through a short pad of silica in a glass pipette, and eluting with diethyl ether. Upon completion of the reaction, the mixture is diluted with 1:1 diethyl ether/hexane, filtered over a short pad of silica using a fritted glass funnel (I.D = 1 inch, height of silica pad ~1.5 inch), and concentrated.

For Table S2, entry 2

In the N₂-filled glovebox, an 8-mL oven dried vial with a septum cap was charged with magnetic stir bar, dpppCoCl₂ (0.05 eq.), Li₃N (1.14 mg, 0.03 mmol), InBr₃ (13 mg, 0.03 mmol) DCM (0.94 mL), (E)-1,3-undecadiene (50 mg, 0.3 mmol), and distilled methyl acrylate (31 mg, 0.36 mmol). Upon completion of the reaction (2 h), the mixture is exposed to air and diluted with 1:1 diethyl ether/hexane, filtered over a short pad of silica using a fritted glass funnel, and concentrated to afford the title compound (61mg, 79% yield) as a colorless oil.

Table S3: Optimization of reaction conditions for heterodimerization of diene and methyl acrylate: Li₃N as a reducing agent ^a



Entry	Catalyst	Reductant	Activator	Conv. ^b	Isolated Yield
		(0.10 equiv.)	(0.10 equiv.)	(%),	(%)
				time (h)	
1	dpppCoBr ₂	Li₃N (0.25 eq.)	InBr₃	100, 7	65
2	dpppCoBr ₂	Li₃N	InBr₃	100, 2	79
3	dpppCoBr ₂	Li₃N	NaBARF	100, 2	52
4	dpppCoBr ₂	Li₃N (0.25 eq.)	InCl₃ (0.30 eq.)	100, 2	83
5	dpppCoCl ₂	Li₃N	InBr₃	98, 0.25	91
6	dpppCoCl ₂	Li₃N	NaBARF	100, 24	70
7 °	dpppCoCl ₂	Li₃N	InCl₃	14, 12	-
8 ^d	(S,S)-BDPP)CoBr ₂	Li ₃ N	InBr ₃	100, 3	84 (99% ee)

^a Procedure A2 followed with some modification in activators and reducing agent as mentioned.
 ^bConversion is based on GC analysis. ^c0.5M DCM is used. ^d(*S*,*S*)-BDPPCoBr₂ was synthesized in-situ as mentioned in the procedure A3.

Procedure A3. Typical procedure for in situ generation of metal complex for heterodimerization. (Table S3, Entry 8)

In N₂-filled glovebox, an 8 mL oven dried vial equipped with a septum cap was charged with a magnetic stir bar, (*S*,*S*)-BDPP (0.05 eq.), CoBr₂ (0.05 eq.) and DCM (0.35 M) and

allowed to stir for 15 min. Reducing agent (0.10 eq.) was then added and the solution was allowed to stir for another 15 minutes. $InBr_3$ (0.10 eq.), 1,3-(*E*)-undecadiene (50 mg, 0.3 mmol, 1.00 eq.), and distilled methyl acrylate (31 mg, 0.36 mmol, 1.10 eq.) were then added via microliter syringe. The mixture was stirred at room temperature and monitored by taking an aliquot using a glass pipette, diluting with mixture of diethyl ether/hexane (1:1) and filtered through a short pad of silica in a glass pipette eluting with diethyl ether and analyzed via GC. Upon completion of the reaction (0.25 h), the mixture is exposed to air and diluted with 1:1 diethyl ether/hexane, filtered over a short pad of silica using a fritted glass funnel (I.D = 1 inch, height of silica pad ~1.5 inch), and concentrated to afford the title compound as a colorless oil.

The product derived from in-situ generated catalyst [(S,S)-BDPP/CoBr₂/Li₃N/InBr₃] showed 99% ee. (See attached CSP-GC chromatogram below)



Procedure A4. Typical procedure for heterodimerization of linear diene with methyl acrylate. (For Table 3, entry 3 in paper).

In N₂ filled glovebox, an 8-mL vial equipped with a septum screw cap was charged with a magnetic stir bar, (QuinoxP*)CoBr₂ (5 mol%), Zn (100 mol%), NaBARF (10 mol%) and DCM (0.8 mL). The vial was capped and while stirring the mixture, (*E*)-1,3-nonadiene (25 mg, 0.2 mmol, 1 equiv.) was added neat using microliter syringe via the septum, followed by methyl

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acrylate (19 mg, 0.22 mmol, 1.1 equiv.). The mixture was stirred at rt and monitored by taking an aliquot using a glass pipette, diluting with a mixture of *n*-pentane/Et₂O (1: 1) and filtered through a pad of silica in a glass pipette eluting with diethyl ether and analyzed via GC-FID. Upon completion of the reaction, the vial was taken out of the box and the reaction mixture worked-up and crude product was isolated via column chromatography.

For Table 3, entry 4 in paper, Procedure C1 followed with some modification in reductant;

11a (5 mol%) was used instead of Zn (100 mol%).

B1. Typical procedure for hydrovinylation of linear diene using isolated Co(I) complex 12.^{7,8} Hydrovinylation of 1,3-diene

+ C₅H1	$\frac{ }{ } - \frac{\text{catalyst, Nat}}{CH_2C}$	aBARF (0.1 e	equiv)	+ C ₅ H ₁₁ + C ₅ H ₁₁ 1]-HV [1,4]-HV
entry	catalyst	time (min)	conver. (%) % yield (% <i>rr*</i>)
1. (dp Zn	pp)CoCl ₂ (0.05 equ (1 equiv.) (in situ)	iiv) 30	100	95 (78:22)
2. (dp (12	pp) ₃ Co ₂ Cl ₂ * , 0.02 equiv)	15	100	97 (77:23)

*isolated Co(I) complex. +rr regioisomeric ratio.

Figure S4: Hydrovinylation of linear diene using isolated Co(I) complex 12

In N₂ filled glovebox, a 25-mL Schlenk flask was charged with a magnetic stir bar, [Co₂(dppp)₃Cl₂] **(12)** (0.004 mmol, 0.02 equiv.), and NaBARF (0.008 mmol, 0.04equiv.). The flask was capped with rubber septum and taken outside of the box. It was then connected to a Schlenk line with argon and subjected to vacuum then refill (3 cycles). Freshly distilled methylene chloride (DCM) (0.25 M) was added via septum and the mixture stirred for a few minutes. The flow control stopcock was closed, and an ethylene balloon inserted using a needle via the septum. A 50-mL syringe was used to evacuate the flask (3 x 25 mL) to remove residual argon. At the room temperature, (*E*)-1,3-nonadiene (25 mg, 0.2 mmol) was added neat via a microliter syringe. Progress of the reaction was monitored via GC-FID by taking an aliquot using a syringe and long needle, diluted with pentanes or hexanes, and filtered through a shot pad of silica in a glass pipette. Upon completion of reaction, it was further concentrated on a rotary evaporator to afford crude product. The crude hydrovinylation product was subjected to column chromatography using pentane to obtain the product as a clear oil.

GC (methyl silicone, 80 °C, 3 min, 20 °C/min \rightarrow 250 °C): Rt major [4,1-adduct/branch], 3.1 min (80%); minor [1,4-adduct, linear], 3.8 min (20%)

For entry 1, i.e., in situ generation of the Co(I) complex, $[Co_2(dppp)_3Cl_2]$ (**12**) was replaced by Co(dppp)Cl₂ (0.01 mmol, 0.05 equiv.), Zn (0.2 mmol, 1 equiv.).

Procedure B2. Use of isolated Co(I) complex (18) for heterodimerization of linear diene with methyl acrylate. (For Table 3, entry 2 in paper)

In N₂ filled glovebox, an 8-mL vial equipped with a septum screw cap was charged with a magnetic stir bar, Isolated {[(R,R)-QuinoxP*]cobalt(µ-Br)}₂ (**18**) (3.8 mg, 0.02 equiv.) , NaBARF (0.05 equiv., 8.9 mg) and DCM (0.8 mL). The vial was capped and while stirring the mixture, (*E*)-1,3-nonadiene (25 mg, 0.2 mmol, 1 equiv.) was added neat using microliter syringe via the septum, followed by methyl acrylate (19 mg, 0.22 mmol, 1.1 equiv.). The mixture was stirred at rt and monitored by taking an aliquot using a glass pipette, diluting with a mixture of *n*-pentane/Et₂O (1: 1) and filtered through a pad of silica in a glass pipette eluting with diethyl ether and analyzed via GC-FID. Upon completion of the reaction, the vial was taken out of the box and the reaction mixture worked-up and crude product was isolated via column chromatography.

Procedure B3. Use of isolated {[(R,R)-QuinoxP]Co(I)(η^6 -C₆D₆)}+[BARF]-(23) for heterodimerization of linear diene with methyl acrylate. (For Table 3, entry 1 in paper)

In N₂ filled glovebox, an 8-mL vial equipped with a septum screw cap was charged with a magnetic stir bar, Isolated {[(R,R)-QuinoxP*]cobalt($\eta^6 - C_6D_6$)}+ [BARF]⁻ (**23**) (0.01 equiv.,) and DCM (0.8 mL). The vial was capped and while stirring the mixture, (E)-1,3-nonadiene (25 mg, 0.2 mmol, 1 equiv.) was added neat using microliter syringe via the septum, followed by methyl acrylate (19 mg, 0.22 mmol, 1.1 equiv.). The mixture was stirred at rt and monitored by taking an aliquot using a glass pipette, diluting with a mixture of *n*-pentane/Et₂O (1: 1) and filtered through a pad of silica in a glass pipette eluting with diethyl ether and analyzed via GC-FID. Upon completion of the reaction, the vial was taken out of the box and the reaction mixture worked-up and crude product was isolated via column chromatography.

NMR data of compound 25 matched with the previously reported compound.⁸

GC (methyl silicone, 80 °C, 3 min, 20 °C/min \rightarrow 250 °C): Rt major [4,1-adduct/branch], 7.24 min (>95%)

CSP-GC (cyclosil B, 125 °C, 60 min, 5°C→ 175 °C): Rt from dpppCoBr₂/Zn/NaBARF; [1,4 - adduct] 28.8 min (49.7%), 29.8 min (50.3%).

The product derived from (QuinoxP*)CoBr₂ / Zn/NaBARF; 28.8 min (96.8%), 29.8 min (3.2%); **93%ee**.

The product derived from (QuinoxP*)CoBr₂/**11a**/NaBARF; 28.7 min (96.6%), 29.7 min (3.4%); **93% ee.**

The product derived from Isolated {[(R,R)-QuinoxP*]cobalt(μ -Br)}₂ (**18**)/ NaBARF; 28.7 min (96.2%), 29.7 min (3.8%); **92% ee.**

The product derived from Isolated {[(R,R)-QuinoxP*]cobalt($\eta^6 - C_6D_6$)}+ [BARF]⁻ (**23**) in DCM; 28.7 min (97%), 29.7 min (3%); **94% ee.**

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The product derived from Isolated {[(R,R)-QuinoxP*]cobalt($\eta^6 - C_6D_6$)}+ [BARF]⁻ (23) in Benzene-D₆; 28.7 min (98%), 29.7 min (2%); **96% ee.**

Procedure C. Use of isolated {[(R,R)-iPr-DuPhos]Co(I)- η^4 -(2,3-dimethylbutadiene))}+ [BARF]– (25) for hydroacylation of isoprene/2-siloxy 1,3-diene (for Table 4, entry 1).

In N₂ filled glovebox, an 8-mL vial equipped with a septum screw cap was charged with a magnetic stir bar, Isolated {[(*R*,*R*)-^{*i*}Pr-DuPhos]cobalt- η^4 -(2,3-dimethylbutadiene))}⁺ [BARF]⁻ (**30**) (0.01 equiv.) and diethyl diethyl ether (0.8 mL). The vial was capped and while stirring the mixture, Benzaldehyde (0.2 mmol, 1.0 equiv.) was added neat using microliter syringe via the septum, followed by isoprene (0.6 mmol, 3 equiv.). The mixture was stirred at rt and monitored by taking an aliquot using a glass pipette, diluting with diethyl ether, and filtering through a pad of silica in a glass pipette eluting with diethyl ether. The mixture was analyzed via GC-FID. Upon completion of the reaction the vial was taken out of the box, exposed to air, and quenched with diethyl ether (5 mL) and the resulting mixture was filtered over a short pad of silica (using a fritted glass funnel, 1.D = 1 inch, height of silica pad ~1.5 inch) eluting diethyl ether. It was further concentrated on a rotary evaporator to afford crude product. The crude hydroacylation product was subjected to column chromatography using 5% ethyl acetate-hexanes to obtain the desired products as a clear oil. NMR and CSP-GC data matched with previously reported compounds by our group.⁶

Table S4. Activator-free, single-component (entries 1 and 5) and in situ generated [Co^l]⁺-catalysts for hydroacylation of isoprene and 2-trimethylsilyloxy-1,3-butadiene^a



no.	Co ^l -source (equiv.)	reductant (equiv.)	activator (equiv.)	1,2:1,4 adduct	time (h)	Conversion, (% ee) ^c
		Me		(is	soprene)	
1. b	{[(<i>R,R</i>)- <i>i</i> -Pr-DUPHOS] Co ^l [2,3-DMBD]} ⁺ [BARF] ⁻ (25 , 0.01)	0	0	60:40	40	60 (96, <i>R</i>)
2.	{[(<i>R</i> , <i>R</i>)- <i>i</i> -Pr-DUPHOS] Co ^I Cl} ₂ (20 , 0.025)	0	NaBARF (0.075)	60:40	24	100 (96, <i>R</i>)
3.	{[(<i>R</i> , <i>R</i>)- <i>i</i> -Pr-DUPHOS] Co ^I Cl} ₂ (20 , 0.025)	0	0	N/A	24	0 (N/A)
4.	[(<i>R,R</i>)- <i>i</i> -Pr-DUPHOS] Co ^{ll} Br ₂ (0.05)	Zn (0.5)	NaBARF (0.075)	60:40	24	93 (96, <i>R</i>)
5.	[(<i>R</i> , <i>R</i>)- <i>i</i> -Pr-DUPHOS] Co ^{II} Br ₂ (0.05)	Zn (0.5)	0	N/A	24	0 (N/A)
	отмs (2-trimethylsilyloxy-1,3-butadiene)					
6 ^b	{[(<i>R,R</i>)- <i>i</i> -Pr-DUPHOS] Co ^l [2,3-DMBD]} ⁺ [BARF] [−] (25 , 0.05)	0	0	35:65	30	90 (98, <i>R</i>)
7.	[(<i>R</i> , <i>R</i>)- <i>i</i> -Pr-DUPHOS] Co ^{ll} Br ₂ 20 (0.05)	Zn (0.5)	NaBARF (0.075)	35:65	30	87 (98, <i>R</i>)
8.	[(R,R)-i-Pr-DUPHOS] Co ^{II} Br ₂ 20 (0.05)	Zn (0.5)	0	N/A	30	0 (N/A)

^a Single-component catalyst (shaded entries). DMBD: 2,3-dimethyl-1,3-butadiene ^bConversion is based on the consumption of benzaldehyde. ^c Regioselectivities and enantioselectivities were determined by GC and CSP-GC respectively. **Procedure D1.** Use of isolated {[(R,R)-iPr-DuPhos]Co(I)-η⁴ -(2,3-dimethylbutadiene))}+ [BARF]– (25) for hydroboration of 2,3-dimethylbuta-1,3-diene (for Table 5, entry 1 in the paper).

In N₂ filled glovebox, an 8-mL vial equipped with a septum screw cap was charged with a magnetic stir bar, Isolated {[[(R,R)-^{*i*}Pr-DuPhos]cobalt- η^4 -(2,3-dimethylbutadiene))}⁺ [BARF]⁻ (**25**) (0.02 equiv.) and DCM (0.8 mL). The vial was capped and while stirring the mixture, HBPin (0.22 mmol, 1.05 equiv.) was added neat using microliter syringe via the septum, followed by 2,3-dimethylbuta-1,3-diene (0.2 mmol, 1 equiv.). The mixture was stirred at rt and monitored by taking an aliquot using a glass pipette, diluting with diethyl ether, and filtering through a pad of silica in a glass pipette eluting with diethyl ether. The mixture was analyzed via GC-FID. Upon completion of the reaction the vial was taken out of the box, exposed to air and quenched with diethyl ether (5 mL) and the resulting mixture was filtered over a short pad of silica (using a firthed glass funnel, I.D = 1 inch, height of silica pad ~1.5 inch) eluting diethyl ether. It was further concentrated on a rotary evaporator to afford crude product. The crude hydroboration product was subjected to column chromatography using 5% diethyl ether-pentane to obtain the hydroboration products as a clear oil.

Procedure D2. Use of in-situ generated Co(I) complex for hydroboration of 2,3dimethylbuta-1,3-diene (for Table 5, entry 2 in the paper)

In N₂ filled glovebox, an 8-mL vial equipped with a septum screw cap was charged with a magnetic stir bar, $[(R,R)-Pr-DuPhosCoBr_2]$ (0.05 equiv.), Zn (0.5 equiv.), and NaBARF (0.1 equiv.) and DCM (0.8 mL). The vial was capped and while stirring the mixture, HBPin (0.22 mmol, 1.05 equiv.) was added neat using microliter syringe via the septum, followed by 2,3-dimethylbuta-1,3-diene (0.2 mmol, 1 equiv.). The mixture was stirred at rt and monitored by taking an aliquot using a glass pipette, diluting with diethyl ether, and filtering through a pad of

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silica in a glass pipette eluting with diethyl ether. The mixture was analyzed via GC-FID. Upon completion of the reaction the vial was taken out of the box, exposed to air and quenched with diethyl ether (5 mL) and the resulting mixture was filtered over a short pad of silica (using a fritted glass funnel, I.D = 1 inch, height of silica pad ~1.5 inch) eluting diethyl ether. It was further concentrated on a rotary evaporator to afford crude product. The crude hydroboration product was subjected to column chromatography using 5% diethyl ether-pentane to obtain the hydroboration product as a clear oil.

GC (methyl silicone, 50 °C, 5 min, 20 °C/min \rightarrow 250 °C): Rt from dppp, major [1,2-adduct **29**], 8.9 min (66%); minor [1,4-adduct **30**], 9.1 min (34%); from {[(*R*,*R*)-*i*Pr-DuPhos]cobalt- η^4 -(2,3dimethylbutadiene))}+ [BARF]⁻ (**25**); major [1,2-adduct **29**], 8.9 min (53%); [1,4-adduct **30**], 9.1 min (33%); [Vinyl boronate **31**], 9.2 min (14%).

NMR of **29-31** matched with the previously reported data by our group.³

¹H NMR (600 MHz, C6D6) δ 4.9 (m, 1H), 4.75 (pent, J = 1.8 Hz, 1H), 2.68 – 2.61 (m, 1H), 1.70 (s, 3H), 1.16 (d, J = 6.8 Hz, 3H), 1.06 (s, 12H), 1.03 – 1.02 (m, 2H); ¹³C NMR (150 MHz, C6D6) δ 152.16, 108.40, 82.81, 37.43, 25.02, 24.94, 22.28, 20.04. Expected 9 C; observed 8 C (missing C-B).

CSP-GC (cyclosil B, 90 °C, 60 min , 5 °C → 175 °C): Rt from dppp; [1,2 - adduct] 24.0 min (49.7%), 24.4 min (50.3%).

Using catalyst {[(R,R)-Pr-DuPhos]cobalt- η^4 -(2,3-dimethylbutadiene))}* [BARF]⁻ (**25**)): major [1,2 -adduct] 24.0 min (10.7%), 24.4 min (89.3%); **79% ee**

Using catalyst [(*R*,*R*)-^{*i*}Pr-DuPhosCoBr₂] (0.05 equiv.), Zn (0.5 equiv.), and NaBARF (0.1 equiv.): major [1,2 -adduct] 24.1 min (14.7%), 24.4 min (85.3%); **70% ee**

HRMS (ESI-MS): m/z 233.1697 ([M+Na+]); exact mass calculated for [C₁₂H₂₃BO₂Na+] 233.1686.

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Table S5: Catalytic activity of single-component catalyst {[(R,R)- i Pr-DuPhos]cobalt- η^{4} - (2,3-dimethylbutadiene))}* [BARF]⁻ (25) in an asymmetric hydroboration reaction^a



a. See procedure D1-D2 for experimental details. Conversion and relative ratios of **29**, **30** and **31** determined by GC. b. Single-component catalyst.

References

- 1. Smith, C. R., Zhang, A., Mans, D. J. and RajanBabu, T. V. (*R*)-3-Methyl-3-Phenyl-1-Pentene Via Catalytic Asymmetric Hydrovinylation. Org. Synth. 2008, **85**, 248-266.
- 2. Sulzbach, R. A., Iqbal, A. F. M., 1,4-Bis(trimethylsilyl)-1,4-dihydropyrazine by Reductive Silylation of Pyrazine. *Angew. Chem. Int. Ed. Engl.* 1971, **10**, 127.
- Duvvuri, K.; Dewese, K. R.; Parsutkar, M. M.; Jing, S. M.; Mehta, M. M.; Gallucci, J. C.; RajanBabu, T. V. Cationic Co(I)-Intermediates for Hydrofunctionalization Reactions: Regio- and Enantioselective Cobalt-Catalyzed 1,2-Hydroboration of 1,3-Dienes. J. Am. Chem. Soc. 2019, 141, 7365-7375.
- 4. Friedfeld, M. R.; Zhong, H.; Ruck, R. T.; Shevlin, M.; Chirik, P. J. Cobalt-catalyzed asymmetric hydrogenation of enamides enabled by single-electron reduction. *Science* 2018, **360**, 888-893.
- Kooistra, T. M.; Knijnenburg, Q.; Smits, J. M. M.; Horton, A. D.; Budzelaar, P. H. M.; Gal, A. W. Olefin polymerization with {*bis*(imino)pyridyl}(CoCl₂)-CI : Generation of the active species involves Co(I). *Angew. Chem. Int. Ed.* 2001, **40**, 4719-4722.
- Parsutkar, M. M.; RajanBabu, T. V. α- and β-Functionalized Ketones from 1,3-Dienes and Aldehydes: Control of Regio- and Enantioselectivity in Hydroacylation of 1,3-Dienes. *J. Am. Chem.* Soc. 2021, 143, 12825–12835.
- 7. Timsina, Y. N.; Sharma, R. K.; RajanBabu, T. V. Cobalt-catalysed asymmetric hydrovinylation of 1,3dienes, *Chem. Sci.* 2015, **6**, 3994.
- Jing, S. M.; Balasanthiran, V.; Pagar, V.; Gallucci, J. C.; RajanBabu, T. V. Catalytic Enantioselective Hetero-dimerization of Acrylates and 1,3-Dienes. *J. Am. Chem. Soc.* 2017, 139, 18034-18043.

X-Ray Crystallographic Analysis of [(dppp)(Cl)Co[μ-(dppp)]Co(Cl)(dppp)] (**12**)

Table S6. Crystallographic details for RajanBabu 2160 (12)

Formula	C81 H78 Cl2 Co2 P6		
Formula weight	1426.01		
Temperature	210(2) K		
Wavelength	0.71073 Å		
Crystal system	Triclinic		
Space group	P -1		
Unit cell dimensions	a = 21.4598(9) Å	$\alpha = 66.4390(10)^{\circ}$	
	b = 21.5413(9) Å	$\beta = 66.4010(10)^{\circ}$	
	c = 23.7245(10) Å	$\gamma = 71.7220(10)^{\circ}$	
Volume	9056.8(7) Å ³		
Z	4		
Density (calculated)	1.046 Mg/m ³		
Absorption coefficient	0.566 mm ⁻¹		
F(000)	2968		
Crystal size	0.04 x 0.15 x 0.35 mm ³		
Theta range for data collection	2.755 to 25.057°		
Index ranges	-25<=h<=25, -25<=k<=25, -2	8<=l<=28	
Reflections collected	242477		
Independent reflections	32006 [R(int) = 0.0625]		
Completeness to theta = 25.057°	99.7 %		
Refinement method	Full-matrix least-squares on F	2	
Data / restraints / parameters	32006 / 0 / 1640		
Goodness-of-fit on F ²	1.025		
Final R indices [I>2sigma(I)]	R1 = 0.0422, wR2 = 0.1056		
R indices (all data)	R1 = 0.0644, $wR2 = 0.1171$		
Largest diff. peak and hole	0.343 and -0.523 e/Å 3		






The single crystal X-ray diffraction studies were carried out on a Nonius Kappa diffractometer equipped with a Bruker APEX-II CCD and Mo K_a radiation ($\lambda = 0.71073$ Å). A 0.113 x 0.089 x 0.074 mm piece of a green block was mounted on a MiTeGen Micromount with CHRISTO-LUBE MCG 1024 oil. Data were collected in a nitrogen gas stream at 100(2) K using ϕ and ϖ scans. Crystal-to-detector distance was 40 mm and exposure time was 60 seconds per frame using a scan width of 1.0°. Data collection was 100% complete to 25.00° in θ . A total of 56953 reflections were collected covering the indices, -17 <=h<=17, -15 <=k<=15, -23 <=l<=23. 6870 reflections were found to be symmetry independent, with a R_{int} of 0.0910. Indexing and unit cell refinement indicated a primitive, monoclinic lattice. The space group was found to be $P2_1/c$. The data were integrated using the Bruker SAINT software program and scaled using the SADABS software program. Solution by direct methods (SHELXT) produced a complete phasing model consistent with the proposed structure.

All nonhydrogen atoms were refined anisotropically by full-matrix least-squares (SHELXL-2014). All hydrogen atoms were placed using a riding model. Their positions were constrained relative to their parent atom using the appropriate HFIX command in SHELXL-2014. Crystallographic data are summarized in **Table S7**.



Report date	2019-08-08					
Identification code	MP-05-076					
Empirical formula C86 H88 Cl2 Co2 O2 P6						
Molecular formula	mula C78 H72 Cl2 Co2 P6, 2(C4 H8 O)					
Formula weight	1528.14					
Temperature	100 K					
Wavelength	0.71073 Å					
Crystal system	Monoclinic					
Space group	P 1 21/c 1					
Unit cell dimensions	a = 14.7933(14) Å	<i>α</i> = 90°.				
	b = 13.2529(11) Å	$\beta = 102.074(3)^{\circ}.$				
	c = 19.5680(19) Å	$\gamma = 90^{\circ}.$				
Volume	3751.5(6) Å ³					
Z	2					
Density (calculated)	1.353 Mg/m ³					
Absorption coefficient	0.690 mm ⁻¹					
F(000)	1596					
Crystal size	0.113 x 0.089 x 0.074 mm ³					
Crystal color, habit	Green Block					
Theta range for data collection	1.408 to 25.374°.					
Index ranges	-17<=h<=17, -15<=k<=15, -23<=l<=23					
Reflections collected	56953					
Independent reflections	6870 [R(int) = 0.0910, R(sigm	aa) = 0.0726]				
Completeness to theta = 25.000°	100.0 %					
Absorption correction	Semi-empirical from equivalent	nts				
Max. and min. transmission	0.2590 and 0.2226					
Refinement method	Full-matrix least-squares on F	2				
Data / restraints / parameters	6870 / 54 / 449					
Goodness-of-fit on F ²	1.023					
Final R indices [I>2sigma(I)]	R1 = 0.0509, wR2 = 0.1155					
R indices (all data)	lices (all data) $R1 = 0.0974, wR2 = 0.1334$					
Extinction coefficient	n/a					
Largest diff. peak and hole	0.542 and -0.528 e.Å ⁻³					

Table S7. Crystal data and structure refinement for RBabu_MP-05-076.

X-Ray Crystallographic Analysis of {[(R,R)-QuinoxP]cobalt(I)(µ-Br)}2 (18)



The single crystal X-ray diffraction studies were carried out on a Nonius Kappa diffractometer equipped with a Bruker APEX-II CCD and Mo K_a radiation ($\lambda = 0.71073$ Å). A 0.273x 0.254 x 0.216 mm piece of a green block was mounted on a MiTeGen Micromount with CHRISTO-LUBE MCG 1024 oil. Data were collected in a nitrogen gas stream at 100(2) K using ϕ and $\overline{\sigma}$ scans. Crystal-to-detector distance was 80 mm and exposure time was 10 seconds per frame using a scan width of 0.75°. Data collection was 100% complete to 25.00° in θ . A total of 126335 reflections were collected covering the -11<=h<=11, -11<=k<=11, -53<=l<=58. indices, 4353 reflections were found to be symmetry independent, with a R_{int} of 0.0507. Indexing and unit cell refinement indicated a primitive, tetragonal lattice. The space group was found to be $P4_12_12_2$. The data were integrated using the Bruker SAINT software program and scaled using the SADABS software program. Solution by direct methods (SHELXT) produced a complete phasing model consistent with the proposed structure.

All nonhydrogen atoms were refined anisotropically by full-matrix least-squares (SHELXL-2014). All hydrogen atoms were placed using a riding model. Their positions were constrained relative to their appropriate HFIX parent atom using the command in SHELXL-2014. absolute stereochemistry of the molecule was established by anomalous The dispersion using the Parson's method with a Flack parameter of 0.009(2). Crystallographic data are summarized in Table S8.



Report date	2019-08-13						
Identification code	MP-05-065						
Empirical formula	C36 H56 Br2 Co2 N4 P4						
Molecular formula	C36 H56 Br2 Co2 N4 P4						
Formula weight	946.40						
Temperature	100 K						
Wavelength	0.71073 Å						
Crystal system	Tetragonal						
Space group	P41212						
Unit cell dimensions	a = 9.5542(2) Å	α= 90°.					
	b = 9.5542(2) Å	β= 90°.					
	c = 46.3014(12) Å	$\gamma = 90^{\circ}.$					
Volume	4226.5(2) Å ³						
Z	4						
Density (calculated)	1.487 Mg/m ³						
Absorption coefficient	2.857 mm ⁻¹						
F(000)	1936						
Crystal size	$0.273 \text{ x } 0.254 \text{ x } 0.216 \text{ mm}^3$						
Crystal color, habit	Green Block						
Theta range for data collection	1.759 to 26.470°.						
Index ranges	-11<=h<=11, -11<=k<=11, -53	3<=l<=58					
Reflections collected	126335						
Independent reflections	4353 [R(int) = 0.0507, R(sigma	a) = 0.0213]					
Completeness to theta = 25.000°	100.0 %						
Absorption correction	Semi-empirical from equivalents						
Max. and min. transmission	0.2602 and 0.2138						
Refinement method	Full-matrix least-squares on F ²						
Data / restraints / parameters	4353 / 0 / 226						
Goodness-of-fit on F ²	1.073						
Final R indices [I>2sigma(I)]	R1 = 0.0236, wR2 = 0.0557						
R indices (all data)	R1 = 0.0267, wR2 = 0.0563						
Absolute structure parameter	0.009(2)						
Extinction coefficient	n/a						
Largest diff. peak and hole	0.467 and -0.199 e.Å ⁻³						

Table S8. Crystal data and structure refinement for RBabu_MP-05-065 (18).





The single crystal X-ray diffraction studies were carried out on a Nonius Kappa diffractometer equipped with a Bruker APEX-II CCD and Mo K_a radiation ($\lambda = 0.71073$ Å). A 0.157 x 0.146 x 0.092 mm piece of a dark green block was mounted on a MiTeGen Micromount with CHRISTO-LUBE MCG 1024 oil. Data were collected in a nitrogen gas stream at 100(2) K using ϕ and ϖ scans. Crystal-todetector distance was 80 mm and exposure time was 30 seconds per frame using a scan width of 0.75°. Data collection was 99.8% complete to 23.291° in θ , 0.90Å. A total of 319126 reflections were collected covering the indices, -16<=h<=22, -23<=k<=23, -58<=l<=58. 17222 reflections were found to be symmetry independent, with a R_{int} of 0.1535. Indexing and unit cell refinement indicated a primitive, tetragonal lattice. The space group was found to be *P*4₃2₁2. The data were integrated using the Bruker SAINT software program and scaled using the SADABS software program. Solution by direct methods (SHELXT) produced a complete phasing model consistent with the proposed structure.

All nonhydrogen atoms were refined anisotropically by full-matrix least-squares (SHELXL-2014). All hydrogen atoms were placed using a riding model. Their positions were constrained relative to their appropriate HFIX parent atom using the command in SHELXL-2014. absolute stereochemistry of the molecule was established by anomalous The dispersion using the Parson's method with a Flack parameter of 0.006(5). Crystallographic data are summarized in Table S9.



Report date	2019-08-16
Identification code	MP-05-066
Empirical formula	C68 H72 Br2 Co2 P4
Molecular formula	C68 H72 Br2 Co2 P4
Formula weight	1290.81
Temperature	100.0 K
Wavelength	0.71073 Å
Crystal system	Tetragonal
Space group	P43212
Unit cell dimensions	$a = 21.2718(9) \text{ Å}$ $\alpha = 90^{\circ}.$
	$b = 21.2718(9) \text{ Å} \qquad \beta = 90^{\circ}.$
	$c = 52.873(2) \text{ Å}$ $\gamma = 90^{\circ}.$
Volume	23925(2) Å ³
Z	16
Density (calculated)	1.433 Mg/m^3
Absorption coefficient	2.039 mm ⁻¹
F(000)	10624
Crystal size	0.157 x 0.146 x 0.092 mm ³
Crystal color, habit	Dark Green Block
Theta range for data collection	1.032 to 23.291°.
Index ranges	-16<=h<=22, -23<=k<=23, -58<=l<=58
Reflections collected	319126
Independent reflections	17222 [R(int) = 0.1535, R(sigma) = 0.0650]
Completeness to theta = 23.291°	99.8 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	0.0243 and 0.0078
Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	17222 / 0 / 1369
Goodness-of-fit on F ²	1.078
Final R indices [I>2sigma(I)]	R1 = 0.0693, $wR2 = 0.1500$
R indices (all data)	R1 = 0.0982, $wR2 = 0.1638$
Absolute structure parameter	0.006(5)
Extinction coefficient	n/a
Largest diff. peak and hole	1.797 and -0.525 e.Å ⁻³

 Table S9. Crystal data and structure refinement for RBabu_MP-05-066 (19).

X-Ray Crystallographic Analysis of [bis(N-aryliminoethyl-kN,N')pyridine-N]CoCl (22)



The single crystal X-ray diffraction studies were carried out on a Nonius Kappa diffractometer equipped with a Bruker APEX-II CCD and Mo K_{α} radiation (λ = 0.71073 Å). A 0.286 x 0.274 x 0.209 mm piece of a purple block was mounted on a MiTeGen Micromount with CHRISTO-LUBE MCG 1024 oil. Data were collected in a nitrogen gas stream at 100(2) K using ϕ and $\overline{\sigma}$ scans. Crystal-to-detector distance was 40 mm and exposure time was 5 seconds per frame using a scan width of 0.75°. Data collection was 100% complete to 25.00° in θ . A total of 21780 reflections were collected covering the -10<=h<=10, -28<=k<=28, -11<=l<=19. indices, 6221 reflections were found to be symmetry independent, with a R_{int} of 0.0475. Indexing and unit cell refinement indicated a primitive, monoclinic lattice. The space group was found to be $P2_1/n$. The data were integrated using the Bruker SAINT software program and scaled using the SADABS software program. Solution by direct methods (SHELXT) produced a complete phasing model consistent with the proposed structure.

All nonhydrogen atoms were refined anisotropically by full-matrix least-squares (SHELXL-2014). All hydrogen atoms were placed using a riding model. Their positions were constrained relative to their parent atom using the appropriate HFIX command in SHELXL-2014. Crystallographic data are summarized in **Table S10**.



Report date	2019-08-16				
Identification code	MP-05-075				
Empirical formula	C33 H43 Cl Co N3				
Molecular formula	C33 H43 Cl Co N3				
Formula weight	576.08				
Temperature	100 K				
Wavelength	0.71073 Å				
Crystal system	Monoclinic				
Space group	P 1 21/n 1				
Unit cell dimensions	a = 8.7386(6) Å	$\alpha = 90^{\circ}$.			
	b = 22.8891(16) Å	β= 101.295(2)°.			
	c = 15.5231(11) Å	$\gamma = 90^{\circ}.$			
Volume	3044.8(4) Å ³				
Z	4				
Density (calculated)	1.257 Mg/m ³				
Absorption coefficient	0.677 mm ⁻¹				
F(000)	1224				
Crystal size	$0.286 \text{ x } 0.274 \text{ x } 0.209 \text{ mm}^3$				
Crystal color, habit	Purple Block				
Theta range for data collection	1.607 to 26.396°.				
Index ranges	-10<=h<=10, -28<=k<=28, -1	l 1<=l<=19			
Reflections collected	21780				
Independent reflections	6221 [R(int) = 0.0475, R(sign	na) = 0.0655]			
Completeness to theta = 25.242°	100.0 %				
Absorption correction	Semi-empirical from equivale	ents			
Max. and min. transmission	0.2602 and 0.2116				
Refinement method	Full-matrix least-squares on H	72			
Data / restraints / parameters	6221 / 0 / 353				
Goodness-of-fit on F ²	1.028				
Final R indices [I>2sigma(I)]	R1 = 0.0432, wR2 = 0.1027				
R indices (all data)	R1 = 0.0683, wR2 = 0.1108				
Extinction coefficient	n/a				
Largest diff. peak and hole	0.938 and -0.298 e.Å ⁻³				

 Table S10.
 Crystal data and structure refinement for RBabu_MP-05-075.





The single crystal X-ray diffraction studies were carried out on a Bruker Kappa Photon II CPAD diffractometer equipped with Mo K_{α} radiation ($\lambda = 0.71073$ Å). A 0.315 x 0.228 x 0.154 mm piece of a purple block was mounted on a MiTeGen Micromount with Paratone 24EX oil. Data were collected in a nitrogen gas stream at 100(2) K using ϕ and ϖ scans. Crystal-to-detector distance was 60 mm and exposure time was 10 seconds per frame using a scan width of 1.0°. Data collection was 99.8% complete to 25.00° in θ . A total of 124192 reflections were collected covering the indices, -17<=h<=17, -18<=k<=18, -21<=21. 23621 reflections were found to be symmetry independent, with a R_{int} of 0.0381. Indexing and unit cell refinement indicated a primitive, triclinic lattice. The space group was found to be *P*1. The data were integrated using the Bruker SAINT software program and scaled using the SADABS software program. Solution by direct methods (SHELXT) produced a complete phasing model for refinement.

nonhydrogen atoms refined anisotropically by full-matrix All were least-squares (SHELXL-2014). All hydrogen atoms were placed using a riding model. Their positions were constrained relative to using appropriate their parent atom the **HFIX** command in SHELXL-2014. absolute stereochemistry of the molecule was established by anomalous The dispersion using the Parson's method with a Flack parameter of 0.005(3). Crystallographic data are summarized in Table S11.



Report date	2020-07-13					
Identification code	MP-05-156					
Empirical formula	C56 H40 B Co D6 F24 N2	P2				
Molecular formula	C24 H28 Co D6 N2 P2, C3	2 H12 B F24				
Formula weight	1340.66					
Temperature	100.0 K					
Wavelength	0.71073 Å					
Crystal system	Triclinic					
Space group	P1					
Unit cell dimensions	a = 13.7516(7) Å	$\alpha = 106.052(2)^{\circ}.$				
	b = 14.4882(7) Å	$\beta = 102.166(2)^{\circ}.$				
	c = 17.3962(9) Å	$\gamma = 111.9630(10)^{\circ}.$				
Volume	2890.8(3) Å ³					
Z	2					
Density (calculated)	1.540 Mg/m ³					
Absorption coefficient	0.469 mm ⁻¹					
F(000)	1348					
Crystal size	0.315 x 0.228 x 0.154 mm ³					
Crystal color, habit	Purple Block					
Theta range for data collection	2.915 to 26.433°.					
Index ranges	-17<=h<=17, -18<=k<=18,	-21<=l<=21				
Reflections collected	124192					
Independent reflections	23621 [R(int) = 0.0381, R(s	sigma) = 0.0363]				
Completeness to theta = 25.000°	99.8 %					
Absorption correction	Semi-empirical from equiva	alents				
Max. and min. transmission	0.0932 and 0.0678					
Refinement method	Full-matrix least-squares or	$1 F^2$				
Data / restraints / parameters	23621 / 156 / 1662					
Goodness-of-fit on F ²	1.016					
Final R indices [I>2sigma(I)]	R1 = 0.0333, $wR2 = 0.0722$	2				
R indices (all data)	R1 = 0.0419, wR2 = 0.0763	3				
Absolute structure parameter	0.005(3)					
Extinction coefficient	n/a					
Largest diff. peak and hole	0.469 and -0.291 e.Å ⁻³					

 Table S11.
 Crystal data and structure refinement for RBabu_MP-05-156 (23).



S56







S58



















13C				$\bigvee_{125.82}^{126.39}$					53.50			23.10		
	iPr												Current Dat NAME MP EXPNO PROCNO F2 - Acquis Date_ Time INSTRUM PROBHD Z1 PULPROG TD SOLVENT NS DS SWH FIDRES AQ RG DW DE TE D1 D11 TD0 SF01 NUC1 P0 P1 PLW1 SF02 NUC2 CPDPRG[2 PCPD2 PLW2 PLW12 PLW12 FLW12 FLW12 FLW13 F2 - Proces SI SF WDW SSB 0 LB GB 0 PC	a Parameters -06-087-IPrPDICOC1 4 1 ition Parameters 20200916 18.58 h spect 14607_0174 (zgpg30 65536 CD2C12 1024 4 36231.883 Hz 1.105709 Hz 0.9043968 sec 189.17 13.800 usec 300.0 K 2.00000000 sec 0.03000000 sec 1 150.9304726 MHz 13C 4.00 usec 12.00 usec 81.53399658 W 600.1824007 MHz 1H waltz65 70.00 usec 28.0000000 W 0.57143003 W 0.28742000 W sing parameters 32768 150.9153719 MHz EM 1.00 Hz 1.40
190 180	170 16	0 150	140	130 120) 110	100	90 80 S68	70	60 50	0 40	30	20	ppm	

MP-05-156-QUINOXPCo(C6D6)BARF; 1H NMR in C6D6



¹H NMR in CD_2Cl_2 (600 MHz)





MP-05-156-QUINOXPCo(C6D6)BARF-31P-rerun-2048 scans



100

50

0



-100

-50

-150

-200

ppm
MP-05-156-OUINOXPCo(C	C6D6))BARF-1	9F
-----------------------	-------	---------	----

-62.05

1L23	J L CF3				SWH FIDRES AQ RG DW DE TE D1 TD0 SF01 NUC1	4 133928.578 Hz 2.043588 Hz 0.4893355 sec 189.17 3.733 usec 6.50 usec 300.2 K 1.0000000 sec 1 564.6769619 MHz 19F		
					PLW1 F2 - Process SI SF WDW SSB 0 LB GB 0 PC	48.93199921 W sing parameters 65536 564.7334352 MHz EM 0.30 Hz 1.00	-	

¹⁹F nmr in CD₂Cl₂

--62.79





Data File C:\CHEM32\1\DATA\MAHESH\MP-06-049-NONADIENE.D Sample Name: MP-06-049-NONADIENE.D

	==	
Acq. Operator	:	MAHESH
Acq. Instrument	:	Instrument 1 Location : Vial 5
Injection Date	:	7/22/2020 7:04:03 PM
		Inj Volume : 1 µl
Acq. Method	:	C:\CHEM32\1\METHODS\MP-80-RAMP.M
Last changed	:	11/9/2018 7:49:41 PM by mahesh
Analysis Method	:	C:\CHEM32\1\METHODS\MP-80-RAMP.M
Last changed	:	7/22/2020 8:44:37 PM by Jon
		(modified after loading)
Method Info	:	General Higher BP that correlates to Stambuli Group's GCMS - 30 min

```
Additional Info : Peak(s) manually integrated
```



Area Percent Report

Sort	ed By		:	Sigr	nal	
Mult	iplier:			:	-	1.0000
Dilu	ition:			:	-	1.0000
Use	Multiplier	&	Dilution	Factor	with	ISTDs

Signal 1: FID1 A,

Peak	RetTime	Туре	Width	Area	Height	Area
#	[min]		[min]	[pA*s]	[pA]	olo
1	1.864	BB	0.0152	254.16351	258.91553	1.000e2

Data File C:\CHEM32\1\DATA\MAHESH\MP-06-048-75MIN.D Sample Name: MP-06-048-75MIN.D Acq. Operator : MP Location : Vial 4 Acq. Instrument : Instrument 1 Injection Date : 7/22/2020 6:00:29 PM Inj Volume : 1 µl : C:\CHEM32\1\METHODS\MP-80-RAMP.M Acq. Method : 11/9/2018 7:49:41 PM by mahesh Last changed Analysis Method : C:\CHEM32\1\METHODS\MP-80-RAMP.M Last changed : 7/22/2020 9:08:29 PM by Jon (modified after loading) Method Info : General Higher BP that correlates to Stambuli Group's GCMS - 30 min Additional Info : Peak(s) manually integrated FID1 A, (MAHESH\MP-06-048-75MIN.D) GC method:(methyl silicone, 80 °C, 3 min, 20 °C/min → 250 °C) pA _ 700 · [4,1]-HV 600 · 500 [4,1]-HV [1,4]-HV 400 [1,4]-HV Table 2, 11a as a reductant; [4,1]: [1,4] = 79:21 300 3.854 200 100 0 3 9 min ______ Area Percent Report Sorted By : Signal Multiplier: : 1.0000 Dilution: : 1.0000 Use Multiplier & Dilution Factor with ISTDs Signal 1: FID1 A, Peak RetTime Type Width Area Height Area % # [min] [min] [pA*s] [pA] 1 3.118 BB 0.0241 1130.81628 719.54199 78.91528 3.854 BB 2 0.0202 302.13342 235.28177 21.08472

Data File C:\CHEM32\1\DATA\MAHESH\MP-06-049-LI3N-90MIN.D Sample Name: MP-06-049-LI3N-90MIN.D



0.0204 262.09161 201.19478 20.07282

1

2

3.097 BB

3.838 BB

0.0251 1043.61243 654.43933 79.92718

Data File C:\CHEM32\1\DATA\MAHESH\MP-06-049-11B.D Sample Name: MP-06-049-11B.D

Acq. Operator : MAHESH Location : Vial 4 Acq. Instrument : Instrument 1 Injection Date : 7/22/2020 8:05:15 PM Inj Volume : 1 µl : C:\CHEM32\1\METHODS\MP-80-RAMP.M Acq. Method : 7/22/2020 7:20:34 PM by MAHESH Last changed Analysis Method : C:\CHEM32\1\METHODS\MP-80-RAMP.M Last changed : 7/22/2020 9:08:29 PM by Jon (modified after loading) Method Info : General Higher BP that correlates to Stambuli Group's GCMS - 30 min

```
Additional Info : Peak(s) manually integrated
```



Реак	RetTime	Туре	Width	Area	Height	Area
#	[min]		[min]	[pA*s]	[pA]	00
1	3.134	BB	0.0243	1422.05847	905.30200	79.62317
2	3.865	BB	0.0197	363.92719	289.12943	20.37683

Data File C:\CHEM32\1\DATA\MAHESH\MP-06-049-.D Sample Name: MP-06-049-.D

```
Acq. Operator
             : MP
                                           Location : Vial 5
Acq. Instrument : Instrument 1
Injection Date : 7/22/2020 3:43:23 PM
                                         Inj Volume : 1 µl
             : C:\CHEM32\1\METHODS\MP-80-RAMP.M
Acq. Method
             : 11/9/2018 7:49:41 PM by mahesh
Last changed
Analysis Method : C:\CHEM32\1\METHODS\MP-80-RAMP.M
Last changed
             : 7/22/2020 8:40:48 PM by Jon
               (modified after loading)
Method Info
              : General Higher BP that correlates to Stambuli Group's GCMS - 30 min
```





Sorted By	:	Signal	
Multiplier:		:	1.0000
Dilution:		:	1.0000
Use Multiplier	& Dilution	Factor with	ISTDs

Signal 1: FID1 A,

Peak	RetTime	Туре	Width	Area	Height	Area
#	[min]		[min]	[pA*s]	[pA]	00
1	3.126	BB	0.0247	991.16547	626.02551	80.08844
2	3.862	BB	0.0203	246.42316	190.48459	19.91156

Data File C:\CHEM32\1\DATA\MAHESH\MP-05-082-QUINOXPCOBR2-ZN-3H.D Sample Name: MP-05-082-QUINOXPCoBr2-Zn-3H.D

	==:	
Acq. Operator	:	MP
Acq. Instrument	:	Instrument 1 Location : Vial 2
Injection Date	:	7/30/2019 9:21:08 PM
		Inj Volume : 1 µl
Acq. Method	:	C:\CHEM32\1\METHODS\MP-80-RAMP.M
Last changed	:	7/30/2019 9:17:48 PM by MP
Analysis Method	:	C:\CHEM32\1\METHODS\MP-80-RAMP.M
Last changed	:	7/18/2020 1:24:31 PM by Jon
		(modified after loading)
Method Info	:	General Higher BP that correlates to Stambuli Group's GCMS - 30 min
Sample Info	:	dcm-3h-rt





Data File C:\CHEM32\1\DATA\MAHESH\MP-05-082-QUINOXPCOBR2-ZN-3H.D Sample Name: MP-05-082-QUINOXPCoBr2-Zn-3H.D

Signal 1: FID1 A,

Peak	RetTime	Туре	Width	Area	Height	Area	
#	[min]		[min]	[pA*s]	[pA]	00	
					-		
1	5.149	BB	0.0153	224.15259	226.86758	16.62074	I
2	7.240	BB	0.0142	1124.47925	1227.58936	83.37926	
Total	s:			1348.63184	1454.45694		
=====		======					

Data File C:\CHEM32.NEW\1\DATA\MAHESH\MAHESH_CHIRAL_GC 2019-07-31 11-47-17\MP-05-082-ZN.D Sample Name: MP-05-082-Zn



Data File C:\CHEM32\1\DATA\MAHESH\MP-05-082-QUINOXPCOBR2-BTDHPY-3H.D Sample Name: MP-05-082-QUINOXPCOBR2-BTDHPY-3H.D

==================	==:	
Acq. Operator	:	MP
Acq. Instrument	:	Instrument 1 Location : Vial 3
Injection Date	:	7/30/2019 9:37:37 PM
		Inj Volume : 1 µl
Acq. Method	:	C:\CHEM32\1\METHODS\MP-80-RAMP.M
Last changed	:	7/30/2019 9:34:13 PM by MP
Analysis Method	:	C:\CHEM32\1\METHODS\MP-80-RAMP.M
Last changed	:	7/18/2020 1:26:37 PM by Jon
		(modified after loading)
Method Info	:	General Higher BP that correlates to Stambuli Group's GCMS - 30 min
Sample Info	:	dcm-3h-rt





Use Multiplier & Dilution Factor with ISTDs

Data File C:\CHEM32\1\DATA\MAHESH\MP-05-082-QUINOXPCOBR2-BTDHPY-3H.D Sample Name: MP-05-082-QUINOXPCOBR2-BTDHPY-3H.D

Signal 1: FID1 A,

Peak	RetTime	Туре	Width	Area	Height	Area	
#	[min]		[min]	[pA*s]	[pA]	00	
1	1.855	BB	0.0152	123.01768	125.06725	9.34019	
2	5.146	MM	0.0163	219.06192	224.35565	16.63241	
3	7.237	BB	0.0139	974.99933	1066.07556	74.02740	
Total	.s :			1317.07893	1415.49846		
=====	========		========				

Data File C:\CHEM32....DATA\MAHESH\MAHESH_CHIRAL_GC 2019-07-31 11-47-17\MP-05-082-BTDHPY.D Sample Name: MP-05-082-btdhpy



Data File C:\CHEM32\1\DATA\MAHESH\MP-05-082-QUINOXPCOBR-DIMER-3H.D Sample Name: MP-05-082-QUINOXPCoBr-dimer-3H

==================	==:	
Acq. Operator	:	MP
Acq. Instrument	:	Instrument 1 Location : Vial 1
Injection Date	:	7/30/2019 9:05:14 PM
		Inj Volume : 1 µl
Acq. Method	:	C:\CHEM32\1\METHODS\MP-80-RAMP.M
Last changed	:	11/9/2018 7:49:41 PM by mahesh
Analysis Method	:	C:\CHEM32\1\METHODS\MP-80-RAMP.M
Last changed	:	7/18/2020 1:26:37 PM by Jon
		(modified after loading)
Method Info	:	General Higher BP that correlates to Stambuli Group's GCMS - 30 min
Sample Info	:	dcm-3h-rt





Use Multiplier & Dilution Factor with ISTDs

Data File C:\CHEM32\1\DATA\MAHESH\MP-05-082-QUINOXPCOBR-DIMER-3H.D Sample Name: MP-05-082-QUINOXPCoBr-dimer-3H

Signal 1: FID1 A,

Peak	RetTime	Туре	Width	Area	Height	Area	
#	[min]		[min]	[pA*s]	[pA]	00	
1	1.873	BB	0.0161	308.49216	304.28009	23.78464	
2	5.166	MM	0.0167	237.52470	237.38588	18.31308	
3	7.244	BB	0.0139	751.00549	839.08966	57.90228	
Total	.s :			1297.02235	1380.75563		
=====	=========	=====	=========				

Data File C:\CHEM32....HESH\MAHESH_CHIRAL_GC 2019-07-31 11-47-17\MP-05-082-QUINOXP-DIMER.D Sample Name: MP-05-082-QUINOXP-DIMER



Data File C:\CHEM32\1\DATA\MAHESH\MP-06-QUINOXPCO(C6D6)-10H.D Sample Name: MP-06-quinoxpCo(C6D6)-10H

Acq. Operator : MP Location : Vial 1 Acq. Instrument : Instrument 1 Injection Date : 7/21/2020 6:34:34 PM Inj Volume : 1 µl Acq. Method : C:\CHEM32\1\METHODS\MP-80-RAMP.M : 11/9/2018 7:49:41 PM by mahesh Last changed Analysis Method : C:\CHEM32\1\METHODS\MP-80-RAMP.M Last changed : 7/22/2020 8:44:37 PM by Jon (modified after loading) Method Info : General Higher BP that correlates to Stambuli Group's GCMS - 30 min



Area Percent Report

Sorted By	:	Signa	1
Multiplier:		:	1.0000
Dilution:		:	1.0000
Use Multiplier	Dilution	Factor w	ith ISTDs

Signal 1: FID1 A,

Peak	RetTime	Туре	Width	Area	Height	Area
#	[min]		[min]	[pA*s]	[pA]	00
1	0.885	BB S	0.0123	6.27043e4	6.80424e4	99.49537
2	1.174	BV	0.0177	121.58495	114.02174	0.19292
3	7.238	BV	0.0139	196.44603	220.20207	0.31171







Table S2, Entry 2GC Conditions: HP-5MS, 100°C, 5 min, 20°C/min to 250°C



Sor	ted By		:	Sign	al	
Mul	tiplier:			:		1.0000
Dil	ution:			:		1.0000
Use	Multiplier	5	Dilution	Factor	with	ISTDs

Signal 1: FID1 A,

Peak	RetTime	Type	Width	Area	Height	Area
+	[min]		[min]	[pA*s]	[pA]	. 8
1	8.522	vv	0.0157	22.69189	22.68320	4.16608
2	9.291	VB	0.0148	446.72842	459.56573	82.01640
3	9.679	BV	0.0160	75.26148	72.93928	13.81751

Table S2, Entry 5GC Conditions: HP-5MS, 100°C, 5 min, 20°C/min to 250°C



Signal 1: FID1 A,

Peak ‡	RetTime [min]	Туре	Width [min]	Area [pA*s]	Height [pA]	Area %
1	8.525	VV	0.0156	22.95272	23.15723	4.48849
2	9.293	vv	0.0152	411.92584	428.89218	80.55358
3	9.681	vv	0.0156	61.52719	59.30364	12.03186
4	9.790	vv	0.0187	14.96301	11.84402	2.92607
Total				511.36876	523.19708	



Data File C:\CHEM32\1\DATA\MAHESH\MP-06-128-CAT-CO-40H.D Sample Name: MP-06-128-CAT-CO-40H

Peak RetTime Type # [min]	Width [min] [Area pA*s]	Height [pA]	Area %
3 6.635 VV	0.0209	10.34493	7.13682	0.95318
4 6.728 VV	0.0143 2	36.88599	249.32899	21.82653
Totals :	10	85.31231	1049.57961	



Data File C:\CHEM32\1\DATA\MAHESH\MP-07-016-IPR-DUPHOS-COBR2-ZN-24H.D Sample Name: MP-07-016-IPR-DUPHOS-COBR2-ZN-24H Acq. Operator : MP Acq. Instrument : Instrument 1 Location : Vial 14 Injection Date : 3/30/2021 6:08:39 PM Inj Volume : 1 µl Acq. Method : C:\CHEM32\1\METHODS\MP-80-RAMP.M Last changed : 3/30/2021 6:05:04 PM by MP Analysis Method : C:\CHEM32\1\METHODS\MP-80-RAMP.M Last changed : 3/30/2021 4:15:28 PM by MP (modified after loading) Method Info : General Higher BP that correlates to Stambuli Group's GCMS - 30 min FID1 A, (MAHESH\MP-07-016-IPR-DUPHOS-COBR2-ZN-24H.D) pA Unreacted Benzaldehyde 800 5 % iPr-DUPHOSCoBr₂, 50 mol% Zn used; NO Activator: NO REACTION 600 400 200 0 4 6 8 10 12 min _____ Area Percent Report _____ Sorted By : Signal Multiplier: : 1.0000 : Dilution: 1.0000 Use Multiplier & Dilution Factor with ISTDs Signal 1: FID1 A, Peak RetTime Type Width Area Height Area [min] [pA*s] % # [min] [pA] 1 1.972 VV 0.0156 964.46014 948.07062 1.000e2 964.46014 948.07062 Totals : S97

Data File C:\CHEM32\1\DATA\MAHESH\MP-07-016-ISOLATED-IPR-DUPHOSCOCL-CRYSTA.D Sample Name: MP-07-016-ISOLATED-IPR-DUPHOSCOCL-CRYSTA Acq. Operator : MP Acq. Instrument : Instrument 1 Location : Vial 16 Injection Date : 3/30/2021 5:29:31 PM Inj Volume : 1 µl Acq. Method : C:\CHEM32\1\METHODS\MP-80-RAMP.M Last changed : 3/30/2021 4:51:34 PM by DS Analysis Method : C:\CHEM32\1\METHODS\MP-80-RAMP.M Last changed : 3/30/2021 4:15:28 PM by MP (modified after loading) Method Info : General Higher BP that correlates to Stambuli Group's GCMS - 30 min FID1 A, (MAHESH\MP-07-016-ISOLATED-IPR-DUPHOSCOCL-CRYSTA.D) pА 500 Unreacted Benzaldehyde 400 Table 4, Entry 3 300 2.5 % isolated crystalline (20)- [iPr-DUPHOSCo(I)CI]2 : NO REACTION 200 100 0 4 6 8 10 12 min _____ Area Percent Report _____ Sorted By : Signal Multiplier: : 1.0000 : Dilution: 1.0000 Use Multiplier & Dilution Factor with ISTDs Signal 1: FID1 A, Peak RetTime Type Width Area Height Area [min] [pA*s] % # [min] [pA] 1 1.996 BB 0.0183 610.35071 508.30597 1.000e2 610.35071 508.30597 Totals :

Data File C:\CHEM32.NEW\1\DATA\MAHESH\MP-05-163-DCYPE-PHCHO-OTMS-DIENE.D Sample Name: MP-05-163-DCYPE-PHCHO-OTMS-DIENE

=====	===================	==			======	==	=====	===
Acq.	Operator	:	LI					
Acq.	Instrument	:	Instrument 1	Lo	cation	:	Vial	202
Injec	tion Date	:	3/12/2020 10:06:09 PM					
				Inj	Volume	:	l µl	
Acq.	Method	:	C:\CHEM32.NEW\1\METHOD	S/MP-100-ISO	300-CHI	RA	L.M	
Last	changed	:	3/12/2020 10:04:28 PM	by LI				
			(modified after loading	ng)				
Analy	rsis Method	:	C:\CHEM32.NEW\1\METHOD	S/LI-90-ISO3	00-RAMP	5-	CYCLC	SIL.M
Last	changed	:	8/19/2020 4:47:20 PM k	by MP				
			(modified after loading	ng)				
Sampl	e Info	:	100-300-1-175-20					

Additional Info : Peak(s) manually integrated



Data File C:\CHEM32.NEW\1\DATA\MAHESH\MP-05-163-DCYPE-PHCHO-OTMS-DIENE.D Sample Name: MP-05-163-DCYPE-PHCHO-OTMS-DIENE

Signal 1: FID1 B, FID1B, Back Signal

Peak	RetTime	Туре	Width	Area	Height	Area	
#	[min]		[min]	[pA*s]	[pA]	00	
1	181.754	MF	1.3040	95.85970	1.22524	50.73448	
2	184.379	FM	1.3261	93.08420	1.16987	49.26552	
Total	s:			188.94390	2.39511		
=====	========	=====	=======				

Data File C:\CHEM32.NEW\1\DATA\MAHESH\MP-05-163-IPR-DUPHOS-PHCHO-OTMS-DIENE.D Sample Name: MP-05-163-IPR-DUPHOS-PHCHO-OTMS-DIENE.D

	==:		========		==		===
Acq. Operator	:	LI					
Acq. Instrument	:	Instrument 1	Lo	ocation	:	Vial	202
Injection Date	:	3/13/2020 12:00:37 PM					
			Inj	Volume	:	1 µl	
Acq. Method	:	C:\CHEM32.NEW\1\METHODS\M	P-100-ISC	0300-CHI	RA	AL.M	
Last changed	:	3/13/2020 4:18:32 PM by L	I				
		(modified after loading)					
Analysis Method	:	C:\CHEM32.NEW\1\METHODS\L	I-90-ISO3	300-RAMP	-5	-CYCLO	SIL.M
Last changed	:	8/19/2020 4:47:20 PM by MI	2				
		(modified after loading)					
Sample Info	:	100-300-1-175-20					

Additional Info : Peak(s) manually integrated



Data File C:\CHEM32.NEW\1\DATA\MAHESH\MP-05-163-IPR-DUPHOS-PHCHO-OTMS-DIENE.D Sample Name: MP-05-163-IPR-DUPHOS-PHCHO-OTMS-DIENE.D

Signal 1: FID1 B, FID1B, Back Signal

Peak	RetTime	Туре	Width	Area	Height	Area	
#	[min]		[min]	[pA*s]	[pA]	00	
1	181.343	MF	1.2998	512.41040	6.57039	35.55996	
2	184.202	FM	0.9961	5.75857	9.63546e-2	0.39963	
3	239.486	MM	1.7411	922.80695	8.83366	64.04041	
Total	ls :			1440.97592	15.50040		
=====		=====					



Data File C:\CHEM32....DIPSHI\MAHESH_CHIRAL_GC 2021-03-29 21-29-33\MP-06-128-CAT-CO-DMBD.D Sample Name: MP-06-128-CAT-CO-DMBD

Totals : 488.53896 6.32178

Data File C:\CHEM32\1\DATA\MAHESH\MP-06-030-CAT-CO-2H.D Sample Name: MP-06-030-CAT-CO-2H.D

=================	==:	
Acq. Operator	:	mp
Acq. Instrument	:	Instrument 1 Location : Vial 2
Injection Date	:	6/23/2020 5:28:26 PM
		Inj Volume : 1 µl
Acq. Method	:	C:\CHEM32\1\METHODS\50-RAMP.M
Last changed	:	3/5/2020 3:50:37 PM by MP
Analysis Method	:	C:\CHEM32\1\METHODS\MP-80-RAMP.M
Last changed	:	7/18/2020 2:32:42 PM by Jon
		(modified after loading)
Method Info	:	General Higher BP that correlates to Stambuli Group's GCMS - 30 min
Sample Info	:	2 MOL%

Additional Info : Peak(s) manually integrated



Data File C:\CHEM32\1\DATA\MAHESH\MP-06-030-CAT-CO-2H.D Sample Name: MP-06-030-CAT-CO-2H.D

Signal 1: FID1 A,

Peak	RetTime	Туре	Width	Area	Height	Area			
#	[min]		[min]	[pA*s]	[pA]	00			
1	8.877	FM	0.0157	431.21741	457.18283	52.94511			
2	9.072	BB	0.0133	270.53165	313.40701	33.21602			
3	9.243	BB	0.0132	112.71218	132.18829	13.83886			
Total	s:			814.46124	902.77814				
=====									

Data File C:\CHEM32.NEW\1\DATA\MAHESH\MP-06-028-CAT-CO-IPRDUPHOS-HB.D Sample Name: MP-06-028-CAT-CO-IPRDUPHOS-HB

	==:		=======	===
Acq. Operator	:	MP		
Acq. Instrument	:	Instrument 1 Location	: Vial	201
Injection Date	:	6/23/2020 8:01:07 PM		
		Inj Volume	: 1 µl	
Acq. Method	:	C:\CHEM32.NEW\1\METHODS\MP-90-30MIN-RAMP	-CYCLOSI	L.M
Last changed	:	6/19/2019 11:50:20 AM by MP		
Analysis Method	:	C:\CHEM32.NEW\1\METHODS\MP-100-ISO300-CH	IRAL.M	
Last changed	:	7/7/2020 4:22:32 PM by MP		
		(modified after loading)		
Sample Info	:	2.5HR		

Additional Info : Peak(s) manually integrated



Sample Amount:	:	1.00000	[nq/ul]	(not	used	in	calc.)
Sampie Amount:	•	1.00000	[IIG/u1]	(1100	useu	т11	Carc.)

Use Multiplier & Dilution Factor with ISTDs

Data File C:\CHEM32.NEW\1\DATA\MAHESH\MP-06-028-CAT-CO-IPRDUPHOS-HB.D Sample Name: MP-06-028-CAT-CO-IPRDUPHOS-HB

Signal 1: FID1 B, FID1B, Back Signal

Peak	RetTime	Туре	Width	Area	Height	Area			
#	[min]	1	[min]	[pA*s]	[pA]	8	I		
1	24.054	MM	0.1600	46.23942	4.81647	10.74587			
2	24.397	MM	0.1847	384.05994	34.65346	89.25413			
Tota	ls :			430.29936	39.46993				
=====									
Data File C:\CHEM32\1\DATA\MAHESH\MP-06-028-COL-1EQ.D Sample Name: MP-06-028-COL-1EQ.D

Acq. Operator : mp Location : Vial 3 Acq. Instrument : Instrument 1 Injection Date : 6/23/2020 5:53:05 PM Inj Volume : 1 µl Acq. Method : C:\CHEM32\1\METHODS\50-RAMP.M : 6/23/2020 5:47:10 PM by mp Last changed Analysis Method : C:\CHEM32\1\METHODS\MP-80-RAMP.M Last changed : 7/18/2020 2:29:03 PM by Jon (modified after loading) Method Info : General Higher BP that correlates to Stambuli Group's GCMS - 30 min

```
Additional Info : Peak(s) manually integrated
```



Area Percent Report

Sorted By		:	Sigr	nal	
Multiplier:			:	-	1.0000
Dilution:			:	-	1.0000
Use Multiplier	&	Dilution	Factor	with	ISTDs

Signal 1: FID1 A,

Peak	RetTime	Туре	Width	Area	Height	Area
#	[min]		[min]	[pA*s]	[pA]	00
1	8.876	VB	0.0151	261.94107	263.63089	42.54850
2	9.072	BV	0.0137	177.83084	203.36923	28.88602

Data File C:\CHEM32\1\DATA\MAHESH\MP-06-028-COL-1EQ.D Sample Name: MP-06-028-COL-1EQ.D

Peak RetTime Type	Width	Area	Height	Area
# [min]	[min]	[pA*s]	[pA]	%
			204.12392	
3 9.243 BB	0.0136	175.85747		28.56548
Totals :		615.62938	671.12404	

*** End of Report ***