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Synthesis of diphenyl-(2-thienyl)phosphine, its chalcogenide derivatives and a series of novel complexes of lanthanide nitrates and triflates

Supplementary Information

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I. X-Ray Crystallography

For all figures in this section thermal ellipsoids are shown at the 50% probability level. Standard CPK colors are used throughout, and all non-polar hydrogen atoms have been omitted for clarity.

- For the free ligands 4-6, the numbering scheme for all atoms is shown on the thermal ellipsoid plot.
- For the Ln-ligand complexes, the numbering scheme for all atoms but the pendant phenyl rings are shown with a ball and stick model.

A. Ph₂thienylP=O (4)

In this structure, the electron density corresponding to the thienyl ring was disordered. This disorder was modelled over two positions on the trigonal pyramid of the phosphorus atom with a 0.80:0.20 occupancy ratio. Only one part is shown below for clarity. All hydrogen atoms have been omitted for clarity, and atoms are depicted using standard CPK colors. SADI commands were used to give bond lengths and angles that are consistent with literature values, and EADP commands were used to give reasonable thermal values.



Experimental

Single crystals of $C_{16}H_{13}OPS$ (4) were grown by slow evaporation from a solution in CDCl₃. A suitable crystal was selected and mounted on a nylon loop using a very small amount of paratone oil on a XtaLAB Synergy, Dualflex, HyPix diffractometer. The crystal was kept at 99.98(13) K during data collection. Using Olex2 [1], the structure was solved with the the structure was solved with the ShelXS [2] structure solution program using Direct Methods and refined with the ShelXL [3] refinement package using Least Squares minimization.

- 1. Dolomanov, O.V., Bourhis, L.J., Gildea, R.J, Howard, J.A.K. & Puschmann, H. (2009), J. Appl. Cryst. 42, 339-341.
- 2. Sheldrick, G.M. (2008). Acta Cryst. A64, 112-122.
- 3. Sheldrick, G.M. (2015). Acta Cryst. C71, 3-8.

Crystal structure determination of 4

Crystal Data for C₁₆H₁₃OPS (M =284.29 g/mol): monoclinic, space group P2₁/n (no. 14), a = 9.6867(2) Å, b = 9.9115(2) Å, c = 14.6285(3) Å, β = 96.773(2)°, V = 1394.67(5) Å³, Z = 4, T = 99.98(13) K, μ (CuK α) = 3.040 mm⁻¹, Dcalc = 1.354 g/cm³, 10173 reflections measured (10.414° ≤ 2 Θ ≤ 153.192°), 2697 unique (R_{int} = 0.0393, R_{sigma} = 0.0359) which were used in all calculations. The final R_1 was 0.0713 (I > 2 σ (I)) and wR_2 was 0.2002 (all data).

Refinement model description

Number of restraints - 2, number of constraints - unknown. Details: 1. Fixed Uiso At 1.2 times of: All C(H) groups 2. Restrained distances $C5-S1A \approx C5-C6$ with sigma of 0.02 $C7-S1A \approx C7-C6$ with sigma of 0.02 3. Uiso/Uaniso restraints and constraints Uanis(C3A) = Uanis(C8)Uanis(C6) = Uanis(C10)Uanis(S1A) = Uanis(S1)Uanis(C2A) = Uanis(C9)Uanis(C10A) = Uanis(C7A)Uanis(C2) = Uanis(C6A)Uanis(C4A) = Uanis(C7)Uanis(C10A) = Uanis(C4)Uanis(C9A) = Uanis(C2)Uanis(C8A) = Uanis(C6A)Uanis(C6A) = Uanis(C2)4. Others Fixed Sof: S1(0.8) C6A(0.2) H6A(0.2) C7A(0.2) H7A(0.2) C8A(0.2) H8A(0.2) C9A(0.2) H9A(0.2) C10A(0.2) H10A(0.2) C2(0.8) H2(0.8) C3(0.8) H3(0.8) C4(0.8) H4(0.8) C6(0.8) H6(0.8) C7(0.8) H7(0.8) C8(0.8) H8(0.8) C9(0.8) H9(0.8) C10(0.8) H10(0.8) S1A(0.2) C2A(0.2) H2A(0.2) C3A(0.2) H3A(0.2) C4A(0.2) H4A(0.2) 5.a Aromatic/amide H refined with riding coordinates: C6A(H6A), C7A(H7A), C8A(H8A), C9A(H9A), C10A(H10A), C2(H2), C3(H3), C4(H4), C6(H6), C7(H7), C8(H8), C9(H9), C10(H10), C12(H12), C13(H13), C14(H14), C15(H15), C16(H16), C2A(H2A), C3A(H3A), C4A(H4A) 5.b Fitted hexagon refined as free rotating group: C1(C6A,C7A,C8A,C9A,C10A), C5(C6,C7,C8,C9,C10)

Table 1 Crystal data and structure refinement for 4.

CCDC Number	2096359
Empirical formula	$C_{16}H_{13}OPS$
Formula weight	284.29
Temperature/K	99.98(13)
Crystal system	monoclinic
Space group	P21/n
a/Å	9.6867(2)
b/Å	9.9115(2)
c/Å	14.6285(3)
α/°	90
β/°	96.773(2)
γ/°	90
Volume/Å ³	1394.67(5)
Z	4
$\rho_{calc}g/cm^3$	1.354
μ/mm^{-1}	3.040
F(000)	592.0
Crystal size/mm ³	$0.262 \times 0.136 \times 0.049$
Radiation	$CuK\alpha \ (\lambda = 1.54184)$
2Θ range for data collection/ ^c	10.414 to 153.192
Index ranges	$\text{-}11 \leq h \leq 12, \text{-}8 \leq k \leq 12, \text{-}18 \leq l \leq 16$
Reflections collected	10173
Independent reflections	2697 [$R_{int} = 0.0393$, $R_{sigma} = 0.0359$]
Data/restraints/parameters	2697/2/169
Goodness-of-fit on F ²	1.088
Final R indexes $[I \ge 2\sigma(I)]$	$R_1 = 0.0713, wR_2 = 0.1953$
Final R indexes [all data]	$R_1 = 0.0761, wR_2 = 0.2002$
Largest diff. peak/hole / e Å-3	1.50/-0.77

B. Ph₂thienylP=S (5)



Experimental

Single crystals of $C_{16}H_{13}PS_2$ were grown by slow evaporation of a CDCl₃ solution. A suitable crystal was selected and mounted using a small amount of paratone oil on a nylon loop on a Bruker APEX-II CCD diffractometer. The crystal was kept at 296(2) K during data collection. Using Olex2 [1], the structure was solved with the ShelXS [2] structure solution program using Direct Methods and refined with the ShelXL [3] refinement package using Least Squares minimization.

- 4. Dolomanov, O.V., Bourhis, L.J., Gildea, R.J, Howard, J.A.K. & Puschmann, H. (2009), J. Appl. Cryst. 42, 339-341.
- 5. Sheldrick, G.M. (2008). Acta Cryst. A64, 112-122.
- 6. Sheldrick, G.M. (2015). Acta Cryst. C71, 3-8.

Crystal structure determination of [5]

Crystal Data for C₁₆H₁₃PS₂ (M = 300.35 g/mol): monoclinic, space group P2₁/n (no. 14), a = 10.18080(10) Å, b = 9.50350(10) Å, c = 14.9030(2) Å, β = 94.0376(5)°, V = 1438.33(3) Å³, Z = 4, T = 296(2) K, μ (CuK α) = 4.246 mm⁻¹, *Dcalc* = 1.387 g/cm³, 17457 reflections measured (10.196° ≤ 2 Θ ≤ 144.3°), 2831 unique (R_{int} = 0.0344, R_{sigma} = 0.0237) which were used in all calculations. The final R_1 was 0.0548 (I > 2 σ (I)) and wR_2 was 0.1564 (all data).

Refinement model description

Number of restraints - 0, number of constraints - unknown. Details: 1. Fixed Uiso At 1.2 times of: All C(H) groups 2.a Aromatic/amide H refined with riding coordinates: C2(H2), C3(H3), C4(H4), C6(H6), C7(H7), C8(H8), C9(H9), C10(H10), C12(H12), C13(H13), C14(H14), C15(H15), C16(H16)

Table 1 Crystal data	and structure refinement for 5
Identification code	5
CCDC Number	1844568
Empirical formula	$C_{16}H_{13}PS_2$
Formula weight	300.35
Temperature/K	296(2)
Crystal system	monoclinic
Space group	$P2_1/n$
a/Å	10.18080(10)
b/Å	9.50350(10)
c/Å	14.9030(2)
α/°	90
β/°	94.0376(5)
γ/°	90
Volume/Å ³	1438.33(3)
Ζ	4
$\rho_{calc}g/cm^3$	1.387
μ/mm^{-1}	4.246
F(000)	624.0
Crystal size/mm ³	$0.388 \times 0.221 \times 0.053$
Radiation	$CuK\alpha (\lambda = 1.54178)$
2Θ range for data collection/°	10.196 to 144.3
Index ranges	$11 \le h \le 12, 11 \le k \le 11, 18 \le l \le 18$
Reflections collected	17457
Independent reflections	$2831 \ [R_{int} = 0.0344, R_{sigma} = 0.0237]$
Data/restraints/parameters	2831/0/172
Goodness-of-fit on F ²	1.058
Final R indexes $[I \ge 2\sigma(I)]$	$R_1 = 0.0548, wR_2 = 0.1545$
Final R indexes [all data]	$R_1 = 0.0569, wR_2 = 0.1564$
Largest diff. peak/hole / e Å $^{\text{-3}}$	0.90/-0.88

C. Ph₂thienylP=Se (6)

In this structure the electron density corresponding to the thienyl ring was disordered. This disorder was modeled as two rotational isomers (rotated 180 ° around the P1-C1 bond) with a 0.893:0.107 ratio.



Experimental

Single crystals of $C_{16}H_{13}PSSe$ were grown by slow evaporation of a CDCl₃ solution. A suitable crystal was selected and mounted on a nylon loop using a small amount of paratone oil on a Bruker APEX-II CCD diffractometer. The crystal was kept at 173(2) K during data collection. Using Olex2 [1], the structure was solved with the ShelXT [2] structure solution program using Direct Methods and refined with the ShelXL [3] refinement package using Least Squares minimization.

- 1. Dolomanov, O.V., Bourhis, L.J., Gildea, R.J, Howard, J.A.K. & Puschmann, H. (2009), J. Appl. Cryst. 42, 339-341.
- 2. Sheldrick, G.M. (2015). Acta Cryst. A71, 3-8.
- 3. Sheldrick, G.M. (2015). Acta Cryst. C71, 3-8.

Crystal structure determination of [6]

Crystal Data for C₁₆H₁₃PSSe (M = 347.25 g/mol): monoclinic, space group P2₁/n (no. 14), a = 10.398(3) Å, b = 9.531(3) Å, c = 14.811(5) Å, $\beta = 93.489(4)^{\circ}$, V = 1465.2(8) Å³, Z = 4, T = 173(2) K, μ (MoK α) = 2.797 mm⁻¹, Dcalc = 1.574 g/cm³, 11733 reflections measured ($4.656^{\circ} \le 2\Theta \le 50.702^{\circ}$), 2683 unique ($R_{int} = 0.0445$, $R_{sigma} = 0.0386$) which were used in all calculations. The final R_1 was 0.0465 (I > 2 σ (I)) and wR_2 was 0.1324 (all data).

Refinement model description

Number of restraints - 0, number of constraints - unknown.

Details: 1. Fixed Uiso At 1.2 times of: All C(H) groups 2.a Aromatic/amide H refined with riding coordinates: C2(H2), C3(H3), C4(H4), C6(H6), C7(H7), C8(H8), C9(H9), C10(H10), C12(H12), C13(H13), C14(H14), C15(H15), C16(H16)

and structure refinement for 6.
6
1844765
$C_{16}H_{13}PSSe$
347.25
173(2)
monoclinic
$P2_1/n$
10.398(3)
9.531(3)
14.811(5)
90
93.489(4)
90
1465.2(8)
4
1.574
2.797
696.0
$0.368 \times 0.219 \times 0.034$
MoKa ($\lambda = 0.71073$)
4.656 to 50.702
$\text{-}12 \leq h \leq 12, \text{-}11 \leq k \leq 11, \text{-}17 \leq l \leq 17$
11733
$2683 [R_{int} = 0.0445, R_{sigma} = 0.0386]$
2683/0/172
1.065
$R_1 = 0.0465, wR_2 = 0.1233$
$R_1 = 0.0607, wR_2 = 0.1324$
0.73/-0.70

Details regarding modeling of the complete molecular disorder present in structures $Ln(4)_3(NO_3)_3$ (Ln = La, Pr, Sm)

The electron density of each of these structures was disordered for each atom except for the Ln metal. The asymmetric unit of these structures contain the Ln metal, one phosphine ligand, and one bidentate nitrate anion. Due to the high symmetry of the space group the data was solved in (R-3), when the asymmetric unit is grown the full structure depicts six ligands and six bidentate nitrates occupying the same positions around the metal center. Since the La metal carries a 3+ charge, and there are no outer sphere nitrate anions present, there must be three bidentate nitrates bound directly to the metal center. This means that there is complete molecular disorder in this structure. For ease of discussion, if the bidentate nitrate groups are considered to be monodentate ligands, then the La³⁺ center would be six-coordinate with pseudo-octahedral geometry (see figure below).



The metal coordination sphere of the $[Ln(4)_3(NO_3)_3]$ crystal structures. Left: all nine oxygen atoms (red) that coordinate to the metal (nitrogen and non-coordinating oxygen atom of the nitrate groups are depicted with lighter blue and red colors) with a tricapped trigonal prism polyhedral schematic (magenta) generated using CrystalMaker; Right: depiction of the bidentate nitrates as monodentate ligands through the nitrogen atom to give a six-coordinate metal (octahedral geometry).

If the complex's geometry is thought of as an octahedron, it cannot be determined directly from the electron density data whether the *facial* or *meridional* isomer of this octahedron exists. However, when we built the model of the *meridional* isomer by hand, steric clashing existed between aromatic rings of neighboring phosphine oxide ligands. When we built the model of the *facial* isomer by hand, there was no steric clashing between ligands. Based on this analysis we concluded that the *facial* isomer is present in the solid state. The figures here, and in the body of the manuscript, were prepared by building this model by hand (by manually deleting the atoms not present in the *facial* isomer).

D. La(4)₃(NO₃)₃

In this structure, the electron density corresponding to the thienyl rings were disordered. This disorder was modelled over two orientations with a 0.84:0.16 occupancy ratio. Only one part is shown below for clarity. All hydrogen atoms have been omitted for clarity, and atoms are depicted using standard CPK colors (La = magenta). AFIX 66 and RIGU commands were used to give bond lengths and angles that are consistent with literature values. The figure on the left depicts the thermal ellipsoids of all atoms; the figure on the right shows the atom numbering scheme for all unique atoms, with the pendant phenyl rings omitted for clarity.



Experimental

Single crystals of $C_{48}H_{39}LaN_3O_{12}P_3S_3$ were grown by slow evaporation from a solution of $CHCl_3/CH_2Cl_2$ and used as received. A suitable crystal was selected and mounted on a nylon loop using small amount of paratone oil on a Bruker APEX-II CCD diffractometer. The crystal was kept at 173(2) K during data collection. Using Olex2 [1], the structure was solved with the ShelXS [2] structure solution program using Direct Methods and refined with the ShelXL [3] refinement package using Least Squares minimization.

- 1. Dolomanov, O.V., Bourhis, L.J., Gildea, R.J, Howard, J.A.K. & Puschmann, H. (2009), J. Appl. Cryst. 42, 339-341.
- 2. Sheldrick, G.M. (2008). Acta Cryst. A64, 112-122.
- 3. Sheldrick, G.M. (2015). Acta Cryst. C71, 3-8.

Crystal structure determination of (La(4)₃(NO₃)₃)

Crystal Data for C₄₈H₃₉LaN₃O₁₂P₃S₃ (M =1177.80 g/mol): trigonal, space group R-3 (no. 148), a = 19.720(5) Å, c = 11.549(3) Å, V = 3890(2) Å³, Z = 3.00006, T = 173(2) K, μ (MoK α) = 1.099 mm⁻¹, Dcalc = 1.508 g/cm³, 9924 reflections measured (4.13° ≤ 2 Θ ≤ 50.688°), 1595 unique (R_{int} = 0.0607, R_{sigma} = 0.0381) which were used in all calculations. The final R_1 was 0.0906 (I > 2 σ (I)) and wR_2 was 0.2962 (all data).

Refinement model description

Number of restraints - 168, number of constraints - unknown.

Details: 1. Fixed Uiso At 1.2 times of: All C(H) groups 2. Shared sites {S2, C2} {S1, C17} 3. Uiso/Uaniso restraints and constraints Uanis(S2) = Uanis(C2)Uanis(S1) = Uanis(C17)4. Rigid body (RIGU) restrains O2, O3, O4, N1 with sigma for 1-2 distances of 0.004 and sigma for 1-3 distances of 0.004 P1, O1, C1, C5, C11, S1, C2, C6, C10, C12, C16, C4, H1, H2, C3, H6, C7, C9, H10, H12, C13, C15, H16, H4, H3, H7, C8, H9, H13, C14, H15, H8, H14 with sigma for 1-2 distances of 0.004 and sigma for 1-3 distances of 0.004 5. Others Fixed Sof: O2(0.5) O3(0.5) O4(0.5) N1(0.5) S1(0.42) P1(0.5) O1(0.5) C1(0.5) C2(0.42) H2(0.42) C3(0.5) H3(0.5) C4(0.5) H4(0.5) C5(0.5) C6(0.5) H6(0.5) C7(0.5) H7(0.5) C8(0.5) H8(0.5) C9(0.5) H9(0.5) C10(0.5) H10(0.5) C11(0.5) C12(0.5) H12(0.5) C13(0.5) H13(0.5) C14(0.5) H14(0.5) C15(0.5) H15(0.5) C16(0.5) H16(0.5) C17(0.08) H1(0.42) S2(0.08) 6.a Riding coordinates: C17(H1) 6.b Aromatic/amide H refined with riding coordinates: C2(H2), C3(H3), C4(H4), C6(H6), C7(H7), C8(H8), C9(H9), C10(H10), C12(H12), C13(H13), C14(H14), C15(H15), C16(H16) 6.c Fitted hexagon refined as free rotating group: C5(C6,C7,C8,C9,C10), C11(C12,C13,C14,C15,C16)

Table 1	Crystal data	and structure	refinement for	La(4) ₃ (NO3) ₃
I ADIC I	CI ystai uata	and structure	I chinement ioi	La(4)3(1103)

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CCDC Number	1846830
Empirical formula	$C_{48}H_{39}LaN_3O_{12}P_3S_3$
Formula weight	1177.80
Temperature/K	173(2)
Crystal system	trigonal
Space group	R-3
a/Å	19.720(5)
b/Å	19.720(5)
c/Å	11.549(3)
α/°	90
β/°	90
γ/°	120
Volume/Å ³	3890(2)
Z	3.00006
$\rho_{calc}g/cm^3$	1.508
μ/mm^{-1}	1.099
F(000)	1782.0
Crystal size/mm ³	$0.256 \times 0.247 \times 0.157$
Radiation	MoKa ($\lambda = 0.71073$)
2Θ range for data collection/	° 4.13 to 50.688
Index ranges	$-23 \le h \le 23, -23 \le k \le 23, -13 \le l \le 13$
Reflections collected	9924
Independent reflections	1595 [$R_{int} = 0.0607, R_{sigma} = 0.0381$]
Data/restraints/parameters	1595/168/186
Goodness-of-fit on F ²	1.202
Final R indexes $[I \ge 2\sigma(I)]$	$R_1 = 0.0906, wR_2 = 0.2957$
Final R indexes [all data]	$R_1 = 0.0910, wR_2 = 0.2962$
Largest diff. peak/hole / e Å-	3 1.26/-1.11

E. Sm(4)₃(NO₃)₃

In this structure, the electron density corresponding to the thienyl rings were disordered. This disorder was modelled over two orientations with a 0.70:0.30 occupancy ratio. Only one part is shown below for clarity. All hydrogen atoms have been omitted for clarity, and atoms are depicted using standard CPK colors (Sm = green). AFIX 66, DFIX and DANG commands were used to give bond lengths and angles that are consistent with literature values. SIMU and DELU commands were used to ensure ellipsoids with reasonable thermal values. The figure on the left depicts the thermal ellipsoids of all atoms; the figure on the right shows the atom numbering scheme for all unique atoms, with the pendant phenyl rings omitted for clarity.



Experimental

Single crystals of $C_{48}H_{39}N_3O_{12}P_3S_3Sm$ [Sm(4)₃(NO₃)₃] were grown from slow evaporation of a CH₂Cl₂/CHCl₃ solution used as received. A suitable crystal was selected and mounted on a nylon loop using a small amount of paratone oil on a Bruker APEX-II CCD diffractometer. The crystal was kept at 173 K during data collection. Using Olex2 [1], the structure was solved with the ShelXT [2] structure solution program using Intrinsic Phasing and refined with the ShelXL [3] refinement package using Least Squares minimization.

- 1. Dolomanov, O.V., Bourhis, L.J., Gildea, R.J, Howard, J.A.K. & Puschmann, H. (2009), J. Appl. Cryst. 42, 339-341.
- 2. Sheldrick, G.M. (2015). Acta Cryst. A71, 3-8.
- 3. Sheldrick, G.M. (2015). Acta Cryst. C71, 3-8.

Crystal structure determination of [Sm(4)₃(NO₃)₃]

Crystal Data for C₄₈H₃₉N₃O₁₂P₃S₃Sm (M=1189.24 g/mol): trigonal, space group R-3 (no. 148), a = 19.608(9) Å, c = 11.469(5) Å, V = 3819(4) Å³, Z = 3.00006, T = 173 K, μ (MoK α) = 1.434 mm⁻¹, *Dcalc* = 1.551 g/cm³, 10213 reflections measured (4.154° $\leq 2\Theta \leq 50.91°$), 1577 unique ($R_{int} = 0.0912$, $R_{sigma} = 0.0598$) which were used in all calculations. The final R_1 was 0.0769 (I > 2 σ (I)) and wR_2 was 0.2379 (all data).

Refinement model description

Number of restraints - 191, number of constraints - unknown. Details:

1. Fixed Uiso At 1.2 times of: All C(H) groups 2. Restrained distances S1-C4 = S1-C11.7 with sigma of 0.01C3-C4 = C1-C21.34 with sigma of 0.01C2-C3 1.3 with sigma of 0.01 3. Rigid bond restraints C1, C2, C3, C4, S1 with sigma for 1-2 distances of 0.01 and sigma for 1-3 distances of 0.01 C1A, C2A, C3A, C4A, S1A with sigma for 1-2 distances of 0.01 and sigma for 1-3 distances of 0.01 C11, C12, C13, C14, C15, C16 with sigma for 1-2 distances of 0.01 and sigma for 1-3 distances of 0.01 C5, C10, C9, C8, C7, C6 with sigma for 1-2 distances of 0.01 and sigma for 1-3 distances of 0.01 4. Uiso/Uaniso restraints and constraints $C1 \approx C2 \approx C3 \approx C4 \approx S1$: within 2A with sigma of 0.02 and sigma for terminal atoms of 0.04 $C1A \approx C2A \approx C3A \approx C4A \approx S1A$: within 2A with sigma of 0.02 and sigma for terminal atoms of 0.04 $C11 \approx C12 \approx C13 \approx C14 \approx C15 \approx C16$: within 2A with sigma of 0.04 and sigma for terminal atoms of 0.08 $C5 \approx C10 \approx C9 \approx C8 \approx C7 \approx C6$: within 2A with sigma of 0.04 and sigma for terminal atoms of 0.08 Uanis(C1) = Uanis(C1A)5. Same fragment restrains {C1, C2, C3, C4, S1} sigma for 1-2: 0.01, 1-3: 0.02 as {C1A, C2A, C3A, C4A, S1A} 6. Others Fixed Sof: O2(0.5) O3(0.5) O4(0.5) N1(0.5) P1(0.5) O1(0.5) C1(0.35) C2(0.35) H2(0.35) C3(0.35) H3(0.35) C4(0.35) H4(0.35) S1(0.35) C5(0.5) C6(0.5) H6(0.5) C7(0.5) H7(0.5) C8(0.5) H8(0.5) C9(0.5) H9(0.5) C10(0.5) H10(0.5) C11(0.5) C12(0.5) H12(0.5) C13(0.5) H13(0.5) C14(0.5) H14(0.5) C15(0.5) H15(0.5) C16(0.5) H16(0.5) C1A(0.15) C2A(0.15) H2A(0.15) C3A(0.15) H3A(0.15) C4A(0.15) H4A(0.15) S1A(0.15) 7.a Aromatic/amide H refined with riding coordinates: C2(H2), C3(H3), C4(H4), C6(H6), C7(H7), C8(H8), C9(H9), C10(H10), C12(H12), C13(H13), C14(H14), C15(H15), C16(H16), C2A(H2A), C3A(H3A), C4A(H4A) 7.b Fitted hexagon refined as free rotating group: C5(C6,C7,C8,C9,C10), C11(C12,C13,C14,C15,C16)

Table 1	Crystal	data and	structure	refinement	for	Sm(4) ₃ (NO3)3.
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CCDC Number	1922734
Empirical formula	$C_{48}H_{39}N_3O_{12}P_3S_3Sm$
Formula weight	1189.24
Temperature/K	173(2)
Crystal system	trigonal
Space group	R-3
a/Å	19.608(9)
b/Å	19.608
c/Å	11.469(5)
$\alpha/^{\circ}$	90
β/°	90
γ/°	120
Volume/Å ³	3819(4)
Ζ	3.00006
$\rho_{calc}g/cm^3$	1.551
μ/mm^{-1}	1.434
F(000)	1797.0
Crystal size/mm ³	$0.114 \times 0.098 \times 0.081$
Radiation	MoKa ($\lambda = 0.71073$)
2Θ range for data collection/	° 4.154 to 50.91
Index ranges	$-23 \le h \le 23, -23 \le k \le 23, -13 \le l \le 13$
Reflections collected	10213
Independent reflections	1577 [$R_{int} = 0.0912$, $R_{sigma} = 0.0598$]
Data/restraints/parameters	1577/191/225
Goodness-of-fit on F ²	1.124
Final R indexes $[I \ge 2\sigma(I)]$	$R_1 = 0.0769, wR_2 = 0.2368$
Final R indexes [all data]	$R_1 = 0.0782, wR_2 = 0.2379$
Largest diff. peak/hole / e Å-	3 1.21/-0.95

F. Pr(4)₃(NO₃)₃

In this structure, we suspect there is some rotational disorder with the thiophene ring (as was seen with the La and Sm structures). However, this disorder seemed to be a very minor component in this case, so we did not model it. Furthermore, the data set already had a poor data/parameter ratio and we did not want to exacerbate this by adding another component. When we attempted to build a model that accounted for this disorder it resulted in a structure with statistically worse refinement values.

All hydrogen atoms have been omitted for clarity, and atoms are depicted using standard CPK colors (Pr = purple). The figure on the left depicts the thermal ellipsoids of all atoms; the figure on the right shows the atom numbering scheme for all unique atoms, with the pendant phenyl rings omitted for clarity.



Experimental

Single crystals of $C_{48}H_{39}N_3O_{12}P_3PrS_3$ [**Pr(4)**₃(**NO**₃)₃] were grown from slow evaporation of a CH₂Cl₂/CHCl₃ solution and used as received. A suitable crystal was selected and mounted on a nylon loop using a small amount of paratone oil on a Bruker APEX-II CCD diffractometer. The crystal was kept at 173 K during data collection. Using Olex2 [1], the structure was solved with the ShelXT [2] structure solution program using Intrinsic Phasing and refined with the ShelXL [3] refinement package using Least Squares minimization.

- 1. Dolomanov, O.V., Bourhis, L.J., Gildea, R.J, Howard, J.A.K. & Puschmann, H. (2009), J. Appl. Cryst. 42, 339-341.
- 2. Sheldrick, G.M. (2015). Acta Cryst. A71, 3-8.
- 3. Sheldrick, G.M. (2015). Acta Cryst. C71, 3-8.

Crystal structure determination of [Pr(4)₃(NO₃)₃]

Crystal Data for C₄₈H₃₉N₃O₁₂P₃PrS₃ (M=1179.80 g/mol): trigonal, space group R-3 (no. 148), a = 19.836(11) Å, c = 11.610(6) Å, V = 3956(5) Å³, Z = 3.00006, T = 173.15 K, μ (MoK α) = 1.194 mm⁻¹, *Dcalc* = 1.486 g/cm³, 7002 reflections measured (4.106° $\leq 2\Theta \leq 52.858°$), 1778 unique ($R_{int} = 0.0790$, $R_{sigma} = 0.0660$) which were used in all calculations. The final R_1 was 0.0947 (I > 2 σ (I)) and wR_2 was 0.2915 (all data).

Refinement model description

Number of restraints - 96, number of constraints - unknown.

Details: 1. Fixed Uiso At 1.2 times of: All C(H) groups 2. Rigid bond restraints C5, C10, C9, C8, C7, C6 with sigma for 1-2 distances of 0.01 and sigma for 1-3 distances of 0.01 C11, C12, C13, C14, C15, C16 with sigma for 1-2 distances of 0.01 and sigma for 1-3 distances of 0.01 3. Uiso/Uaniso restraints and constraints $C5 \approx C10 \approx C9 \approx C8 \approx C7 \approx C6$: within 2A with sigma of 0.04 and sigma for terminal atoms of 0.08 $C11 \approx C12 \approx C13 \approx C14 \approx C15 \approx C16$: within 2A with sigma of 0.04 and sigma for terminal atoms of 0.084. Others Fixed Sof: S1(0.5) P1(0.5) O1(0.5) O2(0.5) O3(0.5) O4(0.5) N1(0.5) C1(0.5) C2(0.5) H2(0.5) C3(0.5) H3(0.5) C4(0.5) H4(0.5) C5(0.5) C10(0.5) H10(0.5) C9(0.5) H9(0.5) C8(0.5) H8(0.5) C7(0.5) H7(0.5) C6(0.5) H6(0.5) C11(0.5) C12(0.5) H12(0.5) C13(0.5) H13(0.5) C14(0.5) H14(0.5) C15(0.5) H15(0.5) C16(0.5) H16(0.5) 5.a Aromatic/amide H refined with riding coordinates: C2(H2), C3(H3), C4(H4), C10(H10), C9(H9), C8(H8), C7(H7), C6(H6), C12(H12), C13(H13), C14(H14), C15(H15), C16(H16) 5.b Fitted hexagon refined as free rotating group: C5(C10,C9,C8,C7,C6)

Table 1 Crystal data and structure refinement for Pr(4)₃(NO₃)₃

CCDC code	1922657
Empirical formula	$C_{48}H_{39}N_3O_{12}P_3PrS_3\\$
Formula weight	1179.80
Temperature/K	173
Crystal system	trigonal
Space group	R-3
a/Å	19.836(11)
b/Å	19.836(11)
c/Å	11.610(6)
α/°	90
β/°	90
γ/°	120
Volume/Å ³	3956(5)
Ζ	3.00006
$\rho_{calc}g/cm^3$	1.486
μ/mm^{-1}	1.194
F(000)	1788.0
Crystal size/mm ³	$0.23\times0.213\times0.166$
Radiation	MoKa ($\lambda = 0.71073$)

2Θ range for data collection/ ^c	^o 4.106 to 52.858
Index ranges	$-22 \le h \le 22, -24 \le k \le 24, -12 \le l \le 14$
Reflections collected	7002
Independent reflections	1778 [$R_{int} = 0.0790, R_{sigma} = 0.0660$]
Data/restraints/parameters	1778/96/198
Goodness-of-fit on F ²	1.157
Final R indexes $[I \ge 2\sigma(I)]$	$R_1 = 0.0947, wR_2 = 0.2897$
Final R indexes [all data]	$R_1 = 0.0964, wR_2 = 0.2915$
Largest diff. peak/hole / e Å ⁻³	3 1.09/-1.15

G. Tb(4)₃(NO₃)₃(C₆H₆)

The electron density for this structure was very disordered and was modeled as both rotational and positional isomers of the thienyl ring. The carbon atoms were left as isotropic in the final structure. In the figure below all hydrogen atoms have been omitted for clarity, and atoms are depicted using standard CPK colors (Tb = purple). The figure on the left depicts the thermal ellipsoids the Tb, O, N, P, and S atoms (50% probability); the figure on the right shows the atom numbering scheme for all atoms, with the pendant phenyl rings and solvent benzene molecule omitted for clarity.



Experimental.

Single colourless irregular crystals of $[Tb(4)_3(NO_3)_3(C_6H_6)]$ were grown by vapor diffusion of benzene into a solution of acetonitrile and used as received. A suitable crystal with dimensions $0.24 \times 0.10 \times 0.08$ mm³ was selected and mounted on a nylon loop with paratone oil on a XtaLAB Synergy, Dualflex, HyPix diffractometer. The crystal was kept at a steady T = 100.00(10) K during data collection. The structure was solved with the ShelXT (Sheldrick, 2015) solution program using dual methods and by using Olex2 1.3-alpha (Dolomanov et al., 2009) as the graphical interface. The model was refined with ShelXL 2018/3 (Sheldrick, 2015) using full matrix least squares minimisation on F^2 .

A colourless irregular-shaped crystal with dimensions $0.24 \times 0.10 \times 0.08 \text{ mm}^3$ was mounted on a nylon loop with paratone oil. Data were collected using a XtaLAB Synergy, Dualflex, HyPix diffractometer equipped with an Oxford Cryosystems low-temperature device, operating at T = 100.00(10) K.

Data were measured using ω scans of 0.5° per frame for 2.0/7.9 s using Cu K_{α} radiation (micro-focus sealed X-ray tube, 50 kV, 1 mA). The total number of runs and images was based on the strategy calculation from the program CrysAlisPro (Rigaku, V1.171.40.84a, 2020). The actually achieved resolution was θ = 77.659.

Cell parameters were retrieved using the CrysAlisPro (Rigaku, V1.171.40.84a, 2020) software and refined using CrysAlisPro (Rigaku, V1.171.40.84a, 2020) on 51200 reflections, 58 % of the observed reflections. Data reduction was performed using the CrysAlisPro (Rigaku, V1.171.40.84a, 2020) software which corrects for Lorentz polarization. The final completeness is 100.00 out to 77.659 in θ CrysAlisPro 1.171.40.84a (Rigaku Oxford

Diffraction, 2020) Numerical absorption correction based on gaussian integration over a multifaceted crystal model Empirical absorption correction using spherical harmonics, implemented in SCALE3 ABSPACK scaling algorithm.

The structure was solved in the space group *Pbca* (# 61) by using dual methods using the ShelXT (Sheldrick, 2015) structure solution program. The structure was refined by Least Squares using version 2018/2 of XL (Sheldrick, 2008) incorporated in Olex2 (Dolomanov et al., 2009). All non-hydrogen atoms were refined anisotropically. Hydrogen atom positions were calculated geometrically and refined using the riding model, except for the hydrogen atom on the non-carbon atom(s) which were found by difference Fourier methods and refined isotropically when data permits.

There is a single molecule in the asymmetric unit, which is represented by the reported sum formula. In other words: Z is 8 and Z' is 1.

Citations

CrysAlisPro (Rigaku, V1.171.40.84a, 2020)

O.V. Dolomanov and L.J. Bourhis and R.J. Gildea and J.A.K. Howard and H. Puschmann, Olex2: A complete structure solution, refinement and analysis program, *J. Appl. Cryst.*, (2009), **42**, 339-341.

Sheldrick, G.M., Crystal structure refinement with ShelXL, Acta Cryst., (2015), C71, 3-8.

Sheldrick, G.M., ShelXT-Integrated space-group and crystal-structure determination, Acta Cryst., (2015), A71, 3-8.

Crystal Data. $C_{54}H_{45}N_3O_{12}P_3S_3Tb$, $M_r = 1275.94$, orthorhombic, *Pbca* (No. 61), a = 20.75550(10) Å, b = 17.56580(10) Å, c = 30.25370(10) Å, $\alpha = \beta = \gamma 90^{\circ}$, V = 11030.10(9) Å³, T = 100.00(10) K, Z = 8, Z' = 1, μ (Cu K_{α}) = 8.725, 88674 reflections measured, 11654 unique ($R_{int} = 0.0353$) which were used in all calculations. The final wR_2 was 0.1640 (all data) and R_I was 0.0608 (I $\geq 2 \sigma$ (I)).

Compound	Tb(4) ₃ (NO ₃) ₃ (C ₆ H ₆)
CCDC number	2173564
Formula	$C_{54}H_{45}N_3O_{12}P_3S_3Tb$
$D_{calc.}$ / g cm ⁻³	1.537
μ/mm^{-1}	8.725
Formula Weight	1275.94
Colour	colourless
Shape	irregular
Size/mm ³	0.24×0.10×0.08
T/K	100.00(10)
Crystal System	orthorhombic
Space Group	Pbca
a/Å	20.75550(10)
b/Å	17.56580(10)
c/Å	30.25370(10)
$\alpha/^{\circ}$	90
$\beta/^{\circ}$	90
β°	90
$V/Å^3$	11030.10(9)
Ζ	8
Ζ'	1
Wavelength/Å	1.54184
Radiation type	$Cu K_{\alpha}$
$\theta_{min}/$	3.606
$\theta_{max}/^{\circ}$	77.659
Measured Refl's.	88674
Indep't Refl's	11654
Refl's $I \ge 2$ (I)	10855
$R_{\rm int}$	0.0353
Parameters	452
Restraints	65
Largest Peak	1.730
Deepest Hole	-2.214
GooF	1.076
wR_2 (all data)	0.1640
wR_2	0.1612
R_{l} (all data)	0.0636
R_1	0.0608

H. Er(4)₃(NO₃)₃(C₆H₆)

The electron density for this structure was very disordered and was modeled as both rotational and positional isomers of the thienyl ring. The carbon atoms were left as isotropic in the final structure. In the figure below all hydrogen atoms have been omitted for clarity, and atoms are depicted using standard CPK colors (Er = mocha). The figure on the left depicts the thermal ellipsoids of the Er, O, N, P and S atoms (50% probability); the figure on the right shows the atom numbering scheme using a ball and stick model, with the pendant phenyl rings and solvent benzene molecule omitted for clarity.



Experimental.

Single colourless irregular crystals of $[Er(4)_3(NO_3)_3(C_6H_6)]$ were grown by vapor diffusion of benzene into acetonitrile and used as received. A suitable crystal with dimensions $0.19 \times 0.12 \times 0.07 \text{ mm}^3$ was selected and mounted on a nylon loop with paratone oil on a XtaLAB Synergy, Dualflex, HyPix diffractometer. The crystal was kept at a steady T = 100.00(10) K during data collection. The structure was solved with the ShelXT (Sheldrick, G.M. (2015). Acta Cryst. A71, 3-8) solution program using dual methods and by using O. V. Dolomanov, L. J. Bourhis, R. J. Gildea, J. A. K. Howard and H. Puschmann, Olex2: a complete structure solution, refinement and analysis program. J. Appl. Cryst. (2009). 42, 339-341. as the graphical interface. The model was refined with ShelXL (Sheldrick, Acta Cryst. A64 2008, 112-122) using full matrix least squares minimisation on F^2 .

Crystal Data. $C_{54}H_{45}ErN_3O_{12}P_3S_3$, $M_r = 1284.28$, orthorhombic, *Pbca* (No. 61), a = 20.75560(10) Å, b = 17.56230(10) Å, c = 30.2522(2) Å, $\alpha = \beta = \gamma = 90^{\circ}$, V = 11027.41(11) Å³, T = 100.00(10) K, Z = 8, Z' = 1, $\mu(Cu K_{\alpha}) = 5.245$, 89327 reflections measured, 11680 unique ($R_{int} = 0.0386$) which were used in all calculations. The final wR_2 was 0.1738 (all data) and R_I was 0.0686 (I $\geq 2 \sigma$ (I)).

Citations.

CrysAlisPro (Rigaku, V1.171.41.90a, 2020) O.V. Dolomanov and L.J. Bourhis and R.J. Gildea and J.A.K. Howard and H. Puschmann, Olex2: A complete structure solution, refinement and analysis program, *J. Appl. Cryst.*, (2009), **42**, 339-341.

Sheldrick, G.M., Crystal structure refinement with ShelXL, *Acta Cryst.*, (2015), **C71**, 3-8.

Sheldrick, G.M., ShelXT-Integrated space-group and crystal-structure determination, Acta Cryst., (2015), A71, 3-8.

A colourless irregular-shaped crystal with dimensions $0.19 \times 0.12 \times 0.07 \text{ mm}^3$ was mounted on a nylon loop with paratone oil. Data were collected using a XtaLAB Synergy, Dualflex, HyPix diffractometer equipped with an Oxford Cryosystems low-temperature device, operating at T = 100.00(10) K.

Data were measured using ω scans of 0.5° per frame for 1.0/4.1 s using Cu K_{α} radiation (micro-focus sealed X-ray tube, 50 kV, 1 mA). The total number of runs and images was based on the strategy calculation from the program CrysAlisPro (Rigaku, V1.171.41.90a, 2020). The actually achieved resolution was θ = 77.665.

Cell parameters were retrieved using the CrysAlisPro (Rigaku, V1.171.41.90a, 2020) software and refined using CrysAlisPro (Rigaku, V1.171.41.90a, 2020) on 43288 reflections, 48 % of the observed reflections. Data reduction was performed using the CrysAlisPro (Rigaku, V1.171.41.90a, 2020) software which corrects for Lorentz polarization. The final completeness is 100.00 out to 77.665 in θ CrysAlisPro 1.171.41.90a (Rigaku Oxford Diffraction, 2020) Numerical absorption correction based on gaussian integration over a multifaceted crystal model Empirical absorption correction using spherical harmonics, implemented in SCALE3 ABSPACK scaling algorithm.

The structure was solved in the space group *Pbca* (# 61) by using dual methods using the ShelXT (Sheldrick, G.M. (2015). Acta Cryst. A71, 3-8) structure solution program. The structure was refined by Least Squares using version 2018/2 of XL (Sheldrick, 2008) incorporated in Olex2 (Dolomanov et al., 2009). All non-hydrogen atoms were refined anisotropically. Hydrogen atom positions were calculated geometrically and refined using the riding model, except for the hydrogen atom on the non-carbon atom(s) which were found by difference Fourier methods and refined isotropically when data permits.

There is a single molecule in the asymmetric unit, which is represented by the reported sum formula. In other words: Z is 8 and Z' is 1.

Compound	Er(4) ₃ (NO ₃) ₃ (C ₆ H ₆)
CCDC number	2173657
Formula	$C_{54}H_{45}ErN_3O_{12}P_3S_3$
$D_{calc.}$ / g cm ⁻³	1.547
μ/mm^{-1}	5.245
Formula Weight	1284.28
Colour	colourless
Shape	irregular
Size/mm ³	0.19×0.12×0.07
T/K	100.00(10)
Crystal System	orthorhombic
Space Group	Pbca
a/Å	20.75560(10)
<i>b</i> /Å	17.56230(10)
c/Å	30.2522(2)
$\alpha/^{\circ}$	90
$\beta/^{\circ}$	90
$\gamma \gamma^{\circ}$	90
$V/Å^3$	11027.41(11)
Ζ	8
Ζ'	1
Wavelength/Å	1.54184
Radiation type	Cu K _a
$\theta_{min}/$	2.921
$\theta_{max}/^{\circ}$	77.665
Measured Refl's.	89327
Indep't Refl's	11680
Refl's I $\geq 2 \sigma(I)$	10842
R _{int}	0.0386
Parameters	452
Restraints	65
Largest Peak	1.860
Deepest Hole	-2.279
GooF	1.038
wR_2 (all data)	0.1738
wR_2	0.1712
R_1 (all data)	0.0722
R_1	0.0686

I. Tb(4)₄(OTf)₃

In this structure, the electron density corresponding to two of the thienyl rings, one coordinated triflate and the outer sphere triflate anion were disordered. This disorder was modelled over two orientations with a 50:50 and 33:67 occupancy ratio. DFIX, DANG and SAME commands were used to ensure bond lengths consistent with accepted values, and SIMU, DELU and EADP commands were used to ensure reasonable thermal values. In the figures below, only one part is shown below for clarity. All hydrogen atoms have been omitted for clarity, and atoms are depicted using standard CPK colors (Tb = purple). The figure on the left shows all atoms as thermal ellipsoids at the 30% probability level; the figure on the right omits the pendant phenyl groups and uncoordinated triflate anion, and shows the atom numbering scheme using a ball and stick model.

We suspect that there is additional positional disorder of the thienyl rings present in this structure (meaning: some thienyl rings may be disordered with a phenyl ring of the parent phosphine ligand). This is supported by thermal values for carbon atoms on adjacent benzene rings that are smaller than their neighbors. The residual electron density present from this disorder is minimal. We have chosen not to model the positional disorder of the thienyl rings because it is minimal and because we'd rather maintain a higher data to parameter ratio.

The intensity contribution of additional disordered electron density was removed using the BYPASS¹ command as implemented in *OLEX2*.² We suspect that this electron density corresponds to one molecule of toluene that is rotationally disordered on a symmetry site. The space is located at average x,y,z coordinates of [0.500, 0.000, 0.000], has a calculated volume of 176.2 Å³ and contains approximately 44.5 electrons.



Experimental

Single crystals of $C_{67}H_{52}F_9O_{13}P_4S_7Tb$ [**Tb(4)**₄(**OTf)**₃] were grown by vapor diffusion of toluene/hexane into a solution of the complex. A suitable crystal was selected and mounted on a nylon loop using a small amount of paratone oil on a Bruker APEX-II CCD diffractometer. The crystal was kept at 173(2) K during data collection.

Using Olex2 [2], the structure was solved with the ShelXT [3] structure solution program using Intrinsic Phasing and refined with the ShelXL [4] refinement package using Least Squares minimization.

- 1. van der Sluis, P.; Spek, A. L., "BYPASS: an effective method for the refinement of crystal structures containing disordered solvent regions". *Acta Cryst.* **1990**, *A46*, 194-201.
- Dolomanov, O.V., Bourhis, L.J., Gildea, R.J, Howard, J.A.K. & Puschmann, H. (2009), J. Appl. Cryst. 42, 339-341.
- 3. Sheldrick, G.M. (2015). Acta Cryst. A71, 3-8.
- 4. Sheldrick, G.M. (2015). Acta Cryst. C71, 3-8.

Crystal structure determination of [Tb(4)₄(OTf)₃]

Crystal Data for C₆₇H₅₂F₉O₁₃P₄S₇Tb (M=1743.30 g/mol): triclinic, space group P-1 (no. 2), a = 13.7126(11) Å, b = 14.5867(12) Å, c = 19.8641(17) Å, a = 86.1680(10)°, β = 71.5830(10)°, γ = 89.4960(10)°, V = 3761.0(5) Å³, Z = 2, T = 173.15 K, μ (MoK α) = 1.300 mm⁻¹, *Dcalc* = 1.539 g/cm³, 61327 reflections measured (3.13° $\leq 2\Theta \leq 50.806^{\circ}$), 13819 unique (R_{int} = 0.0467, R_{sigma} = 0.0410) which were used in all calculations. The final R_1 was 0.0537 (I > 2 σ (I)) and wR_2 was 0.1563 (all data).

Refinement model description

Number of restraints - 349, number of constraints - unknown. Details: 1. Fixed Uiso At 1.2 times of: All C(H) groups 2. Restrained distances S1-C1 = S1-C41.7 with sigma of 0.01C4-C3 = C2-C11.34 with sigma of 0.01C3-C2 1.3 with sigma of 0.01 P3-C33A 1.78 with sigma of 0.01P3-S3A 3 with sigma of 0.01 3. Restrained planarity C17, S2, C20, C19, C18 with sigma of 0.1 C49, S4, C52, C51, C50 with sigma of 0.1 C49A, C50A, C51A, C52A, S4A with sigma of 0.1 C33A, C34A, C36A, C35A, S3A, P3 with sigma of 0.1 S3, C33, C34, C35, C36 with sigma of 0.1 4. Rigid bond restraints C33, S3, C36, C35, C34 with sigma for 1-2 distances of 0.01 and sigma for 1-3 distances of 0.01 C33A, C34A, C35A, C36A, S3A with sigma for 1-2 distances of 0.01 and sigma for 1-3 distances of 0.01 C49A, C50A, C51A, C52A, S4A

with sigma for 1-2 distances of 0.02 and sigma for 1-3 distances of 0.04 C49, S4, C52, C51, C50 with sigma for 1-2 distances of 0.01 and sigma for 1-3 distances of 0.01 C17, C18, C19, C20, S2 with sigma for 1-2 distances of 0.01 and sigma for 1-3 distances of 0.02 C33A, S3A, C36A, C35A, C34A with sigma for 1-2 distances of 0.01 and sigma for 1-3 distances of 0.01 C3T, F7T, F8T, F9T with sigma for 1-2 distances of 0.01 and sigma for 1-3 distances of 0.01 C2R, F4R, F5R, F6R with sigma for 1-2 distances of 0.01 and sigma for 1-3 distances of 0.01 5. Uiso/Uaniso restraints and constraints C49A \approx C50A \approx C51A \approx C52A \approx S4A: within 2A with sigma of 0.01 and sigma for terminal atoms of 0.02 $C17 \approx C18 \approx C19 \approx C20 \approx S2$: within 2A with sigma of 0.02 and sigma for terminal atoms of 0.04 $C2R \approx F4R \approx F5R \approx F6R$: within 2A with sigma of 0.04 and sigma for terminal atoms of 0.08 Uanis(C34A) = Uanis(C34)Uanis(C50) = Uanis(C50A)Uanis(C33A) = Uanis(C33)Uanis(C49) = Uanis(C49A)6. Same fragment restrains {S1, C1, C2, C3, C4} sigma for 1-2: 0.02, 1-3: 0.04 as {S2, C17, C18, C19, C20} {S1, C1, C2, C3, C4} sigma for 1-2: 0.01, 1-3: 0.02 as {S3, C33, C34, C35, C36} {S1, C1, C2, C3, C4} sigma for 1-2: 0.02, 1-3: 0.04 {S3A, C33A, C34A, C35A, C36A} {S1, C1, C2, C3, C4} sigma for 1-2: 0.02, 1-3: 0.04 as {S4, C49, C50, C51, C52} {S1, C1, C2, C3, C4} sigma for 1-2: 0.02, 1-3: 0.04 as {S4A, C49A, C50A, C51A, C52A} {S1T, O1T, O2T, O3T, C1T, F1T, F2T, F3T} sigma for 1-2: 0.01, 1-3: 0.02 as {S2T, O4T, O5T, O6T, C2T, F4T, F5T, F6T} {S1T, O1T, O2T, O3T, C1T, F1T, F2T, F3T} sigma for 1-2: 0.02, 1-3: 0.04 as {S2R, 04R, 05R, 06R, C2R, F4R, F5R, F6R} {C1T, F1T, F2T, F3T} sigma for 1-2: 0.02, 1-3: 0.04 as {C3T, F7T, F8T, F9T} {S3T, O7T, O8T, O9T, C3T, F7T, F8T, F9T} sigma for 1-2: 0.02, 1-3: 0.04 as {S3R, O7R, O8R, O9R, C3R, F7R, F8R, F9R} 7. Others

Fixed Sof: S3(0.66667) C33(0.66667) C34(0.66667) H34(0.66667) C35(0.66667) H35(0.66667) C36(0.66667) H36(0.66667) S3A(0.33333) C33A(0.3333) C34A(0.33333) H34A(0.33333) C35A(0.33333) H35A(0.33333) C36A(0.33333) H36A(0.33333) S4(0.5) C49(0.5) C50(0.5) H50(0.5) C51(0.5) H51(0.5) C52(0.5) H52(0.5) S4A(0.5) C49A(0.5) C50A(0.5) H50A(0.5) C51A(0.5) H51A(0.5) C52A(0.5) H52A(0.5) S2T(0.5) O4T(0.5) O5T(0.5) O6T(0.5) C2T(0.5) F4T(0.5) F5T(0.5) F6T(0.5) S2R(0.5) 04R(0.5) O5R(0.5) 06R(0.5) C2R(0.5) F4R(0.5) F5R(0.5) F6R(0.5) S3T(0.5) O7T(0.5) O8T(0.5) O9T(0.5) C3T(0.5) F7T(0.5) F8T(0.5) F9T(0.5) S3R(0.5) O7R(0.5) O8R(0.5) O9R(0.5) C3R(0.5) F7R(0.5) F8R(0.5) F9R(0.5)

8.Aromatic/amide H refined with riding coordinates:
C2(H2), C3(H3), C4(H4), C6(H6), C7(H7), C8(H8), C9(H9), C10(H10), C12(H12),
C13(H13), C14(H14), C15(H15), C16(H16), C18(H18), C19(H19), C20(H20), C22(H22),
C23(H23), C24(H24), C25(H25), C26(H26), C28(H28), C29(H29), C30(H30),
C31(H31), C32(H32), C34(H34), C35(H35), C36(H36), C34A(H34A), C35A(H35A),
C36A(H36A), C38(H38), C39(H39), C40(H40), C41(H41), C42(H42), C44(H44),
C45(H45), C46(H46), C47(H47), C48(H48), C50(H50), C51(H51), C52(H52),
C50A(H50A), C51A(H51A), C52A(H52A), C54(H54), C55(H55), C56(H56), C57(H57),
C58(H58), C60(H60), C61(H61), C62(H62), C63(H63), C64(H64)

Table 1 Crystal data and structure refinement for Tb(4)₄(OTf)₃.

CCDC Number	1850479
Empirical formula	$C_{67}H_{52}F_9O_{13}P_4S_7Tb$
Formula weight	1743.30
Temperature/K	173(2)
Crystal system	triclinic
Space group	P-1
a/Å	13.7126(11)
b/Å	14.5867(12)
c/Å	19.8641(17)
a/°	86.1680(10)
β/°	71.5830(10)
$\gamma/^{\circ}$	89.4960(10)
Volume/Å ³	3761.0(5)
Ζ	2
$ ho_{calc}g/cm^3$	1.539
μ/mm^{-1}	1.300
F(000)	1752.0
Crystal size/mm ³	$0.324 \times 0.193 \times 0.126$
Radiation	MoKα ($\lambda = 0.71073$)
2Θ range for data collection/	° 3.13 to 50.806
Index ranges	$\text{-16} \leq h \leq \text{16}, \text{-17} \leq k \leq \text{17}, \text{-23} \leq \text{l} \leq \text{23}$
Reflections collected	61327
Independent reflections	13819 [$R_{int} = 0.0467, R_{sigma} = 0.0410$]
Data/restraints/parameters	13819/349/1120

 $\begin{array}{ll} Goodness-of-fit \ on \ F^2 & 1.060 \\ Final \ R \ indexes \ [I>=2\sigma \ (I)] & R_1 = 0.0537, \ wR_2 = 0.1459 \\ Final \ R \ indexes \ [all \ data] & R_1 = 0.0636, \ wR_2 = 0.1563 \\ Largest \ diff. \ peak/hole \ / \ e \ Å^{-3} \ 1.68/-1.36 \end{array}$

Solvent masks information for Tb(4)₄(OTF)₃.

Number	Χ	Y	Z	Volume	Electron count Content
1	0.500	0.000	0.000	176.2	44.5 toluene (?)





Std proton Std proton File: Phosphorus

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50	64	4	
d1	1.000	n	
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c1	256		PROCESSING
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stru	161.835		DISPLAY
tof	5211.1	20	-46533.
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up	H	2	-152.
daf		2	-151.
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drn	>	U.M.	52
dpwr	1.1	ž	
dif	9200	22	579
		5	
		10	cdc ph

Phu P-(2) (pure)

226.81-

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Yealow oull Solid Test two 4-7

ALC: NO





















Thienyl-P=Se 77Se single_pulse

> Ph2(thienyl)PSe in CDCl3 77Se NMR, 76 MHz













