

Supporting information

**Curcumin based ionic Pt(II) complexes:  
antioxidant and antimicrobial activity**

Rossella Caligiuri,<sup>a</sup> Giuseppe Di Maio,<sup>a,b</sup> Nicolas Godbert,<sup>a</sup> Francesca Scarpelli,<sup>a</sup> Angela Candreva,<sup>a</sup> Isabella Rimoldi,<sup>c</sup> Giorgio Facchetti,<sup>c</sup> Maria Giovanna Lupo,<sup>d</sup> Emilia Sicilia,<sup>e</sup> Gloria Mazzone,<sup>e</sup> Fortuna Ponte,<sup>e</sup> Isabella Romeo,<sup>f,g</sup> Massimo La Deda,<sup>a,b</sup> Alessandra Crispini,<sup>a</sup> Renata De Rose<sup>h</sup> and Iolinda Aiello<sup>\*a,b</sup>

<sup>a</sup> MAT-INLAB, LASCAMM CR-INSTM, Unità INSTM della Calabria, Dipartimento di Chimica e Tecnologie Chimiche, Università della Calabria, Ponte Pietro Bucci Cubo 14C, Arcavacata di Rende (CS), 87036, Italy, E-mail: [iolinda.aiello@unical.it](mailto:iolinda.aiello@unical.it)

<sup>b</sup> CNR NANOTEC-Istituto di Nanotecnologia UOS Cosenza, Arcavacata di Rende (CS), 87036, Italy

<sup>c</sup> Dipartimento di Scienze Farmaceutiche, Università degli Studi di Milano, Via Venezian 21, Milan, 20133, Italy

<sup>d</sup> *Dipartimento di Medicina, Università degli Studi di Padova, 35128 Padova, Italy*

<sup>e</sup> *Dipartimento di Chimica e Tecnologie Chimiche, Ponte Pietro Bucci Cubo 14C, Arcavacata di Rende (CS), 87036, Italy*

<sup>f</sup> Dipartimento di Scienze della Salute, Università degli Studi "Magna Græcia" di Catanzaro, Campus "S. Venuta", Viale Europa, 88100 Catanzaro, Italy

<sup>g</sup> Net4Science Academic Spin-Off, Università degli Studi "Magna Græcia" di Catanzaro, Campus "S. Venuta", Viale Europa, 88100 Catanzaro, Italy

<sup>h</sup> LAB CF-INABEC Dipartimento di Chimica e Tecnologie Chimiche, Ponte Pietro Bucci Cubo 15C, Arcavacata di Rende (CS), 87036, Italy

## Table of Contents

**Figure S1:** Absorption spectra of DPPH in MeOH dilute solution (100  $\mu\text{M}$ ) varying time

**Figure S2:** FT-IR spectrum of complex [(pn)Pt(curc)][CF<sub>3</sub>SO<sub>3</sub>], **1**

**Figure S3:** <sup>1</sup>H-NMR spectrum of complex [(pn)Pt(curc)][CF<sub>3</sub>SO<sub>3</sub>], **1**

**Figure S4:** <sup>13</sup>C-NMR spectrum of complex [(pn)Pt(curc)][CF<sub>3</sub>SO<sub>3</sub>], **1**

**Figure S5:** ESI-MS spectrum of complex [(pn)Pt(curc)][CF<sub>3</sub>SO<sub>3</sub>], **1**

**Figure S6:** FT-IR spectrum of complex [(pic)Pt(curc)][CF<sub>3</sub>SO<sub>3</sub>], **2**

**Figure S7:** <sup>1</sup>H-NMR spectrum of complex [(pic)Pt(curc)][CF<sub>3</sub>SO<sub>3</sub>], **2**

**Figure S8:** <sup>13</sup>C-NMR spectrum of complex [(pic)Pt(curc)][CF<sub>3</sub>SO<sub>3</sub>], **2**

**Figure S9:** ESI-MS spectrum of complex [(pic)Pt(curc)][CF<sub>3</sub>SO<sub>3</sub>], **2**

**Figure S10:** FT-IR spectrum of complex [(bipy)Pt(curc)][CF<sub>3</sub>SO<sub>3</sub>], **3**

**Figure S11:** <sup>1</sup>H-NMR spectrum of complex [(bipy)Pt(curc)][CF<sub>3</sub>SO<sub>3</sub>], **3**

**Figure S12:** <sup>13</sup>C-NMR spectrum of complex [(bipy)Pt(curc)][CF<sub>3</sub>SO<sub>3</sub>], **3**

**Figure S13:** ESI-MS spectrum of complex [(bipy)Pt(curc)][CF<sub>3</sub>SO<sub>3</sub>], **3**

**Figure S14:** FT-IR spectrum of complex [(bipy-C<sub>9</sub>)Pt(curc)][CF<sub>3</sub>SO<sub>3</sub>], **4**

**Figure S15:** <sup>1</sup>H-NMR spectrum of complex [(bipy-C<sub>9</sub>)Pt(curc)][CF<sub>3</sub>SO<sub>3</sub>], **4**

**Figure S16:** <sup>13</sup>C-NMR spectrum of complex [(bipy-C<sub>9</sub>)Pt(curc)][CF<sub>3</sub>SO<sub>3</sub>], **4**

**Figure S17:** ESI-MS spectrum of complex [(bipy-C<sub>9</sub>)Pt(curc)][CF<sub>3</sub>SO<sub>3</sub>], **4**

**Figure S18:** Simulated absorption spectra of H(curc) (violet) and complexes **1** (blue), **2** (red), **3** (orange) and **4** (green) in implicit solvent methanol.

**Table S1:** Absorption wavelength ( $\lambda$ , nm), oscillator strength (f), MO contribution (%) for **H(curc)** and complexes **1-4**.

**Figure S19:** Natural Transition Orbitals (NTOs) for the main excitations of Pt(II) complexes **1-4**.

**Figure S20:** Absorption spectra of H(curc) in MeOH dilute solution (10  $\mu\text{M}$ ) varying time.

**Figure S21:** Absorption spectra of complex **1** in MeOH dilute solution (10  $\mu\text{M}$ ) varying time.

**Figure S22:** Absorption spectra of complex **2** in MeOH dilute solution (10  $\mu\text{M}$ ) varying time.

**Figure S23:** Absorption spectra of complex **3** in MeOH dilute solution (10  $\mu\text{M}$ ) varying time.

**Figure S24:** Absorption spectra of complex **4** in MeOH dilute solution (10  $\mu\text{M}$ ) varying time.

**Figure S25:** Absorption spectra of H(curc) in DMSO dilute solution (10  $\mu\text{M}$ ) varying time.

**Figure S26:** Absorption spectra of complex **1** in DMSO dilute solution (10  $\mu\text{M}$ ) varying time.

**Figure S27:** Absorption spectra of complex **2** in DMSO dilute solution (10  $\mu\text{M}$ ) varying time.

**Figure S28:** Absorption spectra of complex **3** in DMSO dilute solution (10  $\mu\text{M}$ ) varying time.

**Figure S29:** Absorption spectra of complex **4** in DMSO dilute solution (10  $\mu\text{M}$ ) varying time.

**Figure S30:** Absorption spectra of H(curc) in 10% DMSO/PBS dilute solution (10  $\mu\text{M}$ ) varying time.

**Figure S31:** Absorption spectra of complex **1** in 10% DMSO/PBS dilute solution (10  $\mu\text{M}$ ) varying time.

**Figure S32:** Absorption spectra of complex **2** in 10% DMSO/PBS dilute solution (10  $\mu\text{M}$ ) varying time.

**Figure S33:** Absorption spectra of complex **3** in 10% DMSO/PBS dilute solution (10  $\mu\text{M}$ ) varying time.

**Figure S34:** Absorption spectra of complex **4** in 10% DMSO/PBS dilute solution (10  $\mu\text{M}$ ) varying time.

**Figure S35:** Thermodynamic parameters: bond dissociation enthalpies (BDE, grey), electron transfer enthalpies (ETE, light violet), adiabatic ionization potentials (IP, red), O-H proton dissociation enthalpies (PDE, blue) and proton affinities (PA, green), at 298.15 K.

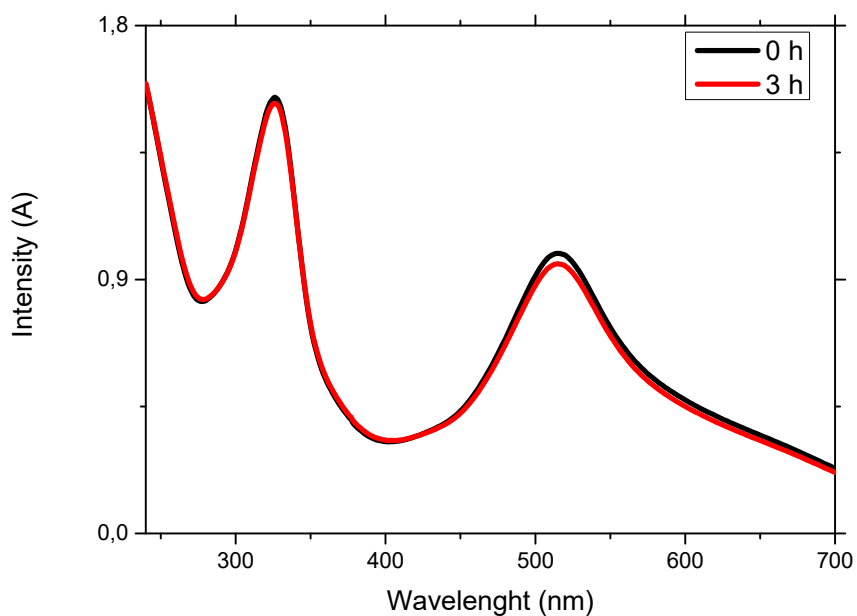
**Table S2:** ADME properties for H(curc) and Pt(II) complexes **1-4**.

**Table S3:** Theoretical and experimental  $\log P_{ow}$  values of the Pt(II) complexes **1-4**.

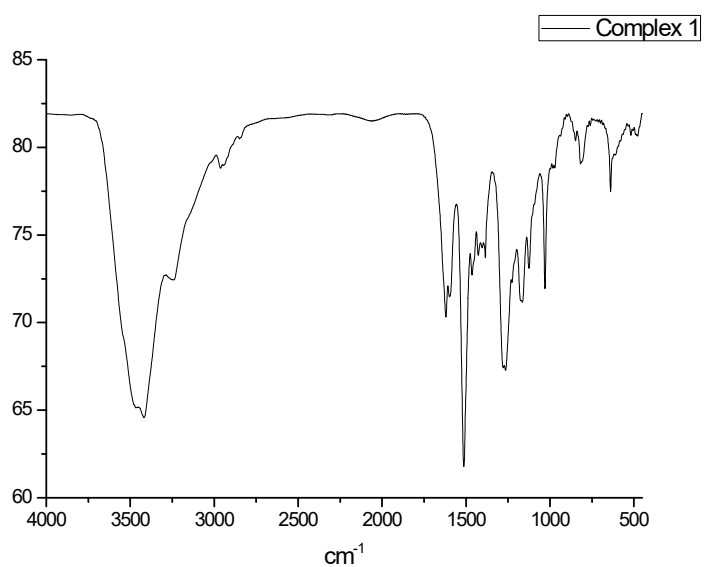
**Figure S36:** BOILED-Egg Model of complexes **1-3**.

**Table S4:** Antimicrobial activity data against *E. coli* and *S. aureus*.

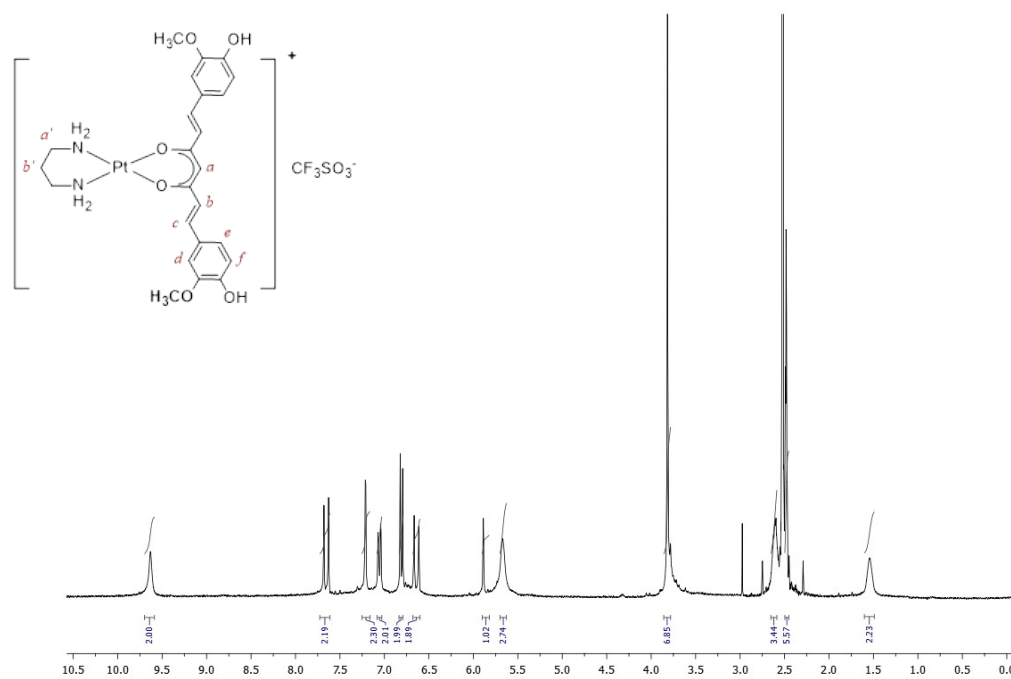
**Figure S37:** Inhibition halos on *S. aureus* (A) and *E. Coli* (B) at 0.5  $\mu\text{g}/\mu\text{l}$ . C: H(curc).



**Figure S1:** Absorption spectra of DPPH in MeOH dilute solution (100  $\mu\text{M}$ ) varying time.



**Figure S2:** FT-IR spectrum of complex  $[(pn)Pt(curc)][CF_3SO_3]$ , **1**.



**Figure S3:**  $^1H$ -NMR spectrum of complex  $[(pn)Pt(curc)][CF_3SO_3]$ , **1**.

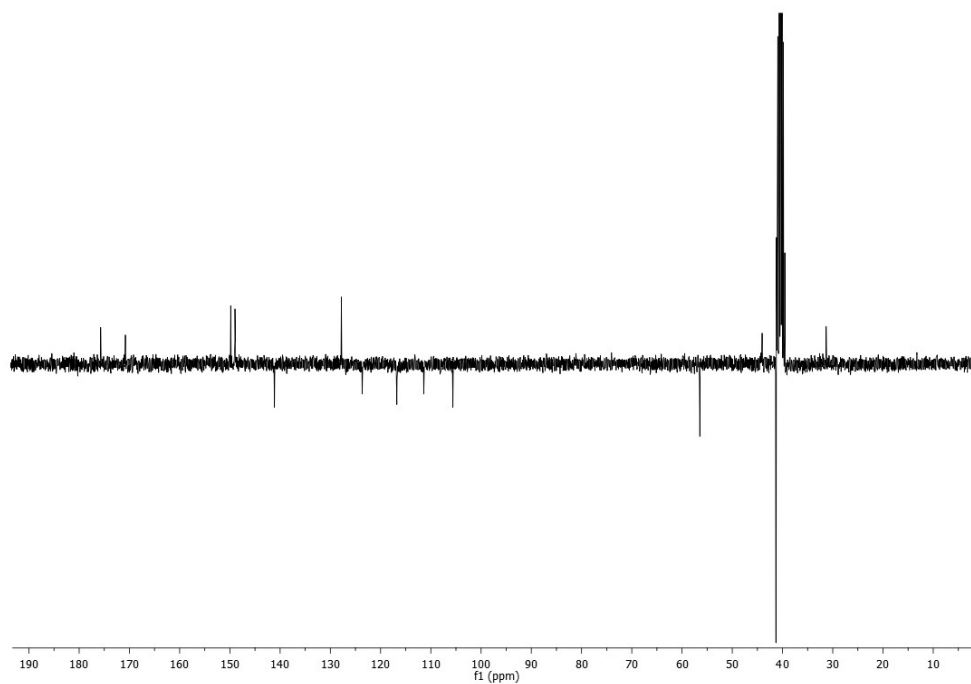


Figure S4: <sup>13</sup>C-NMR spectrum of complex [(pn)Pt(curc)][CF<sub>3</sub>SO<sub>3</sub>], **1**.

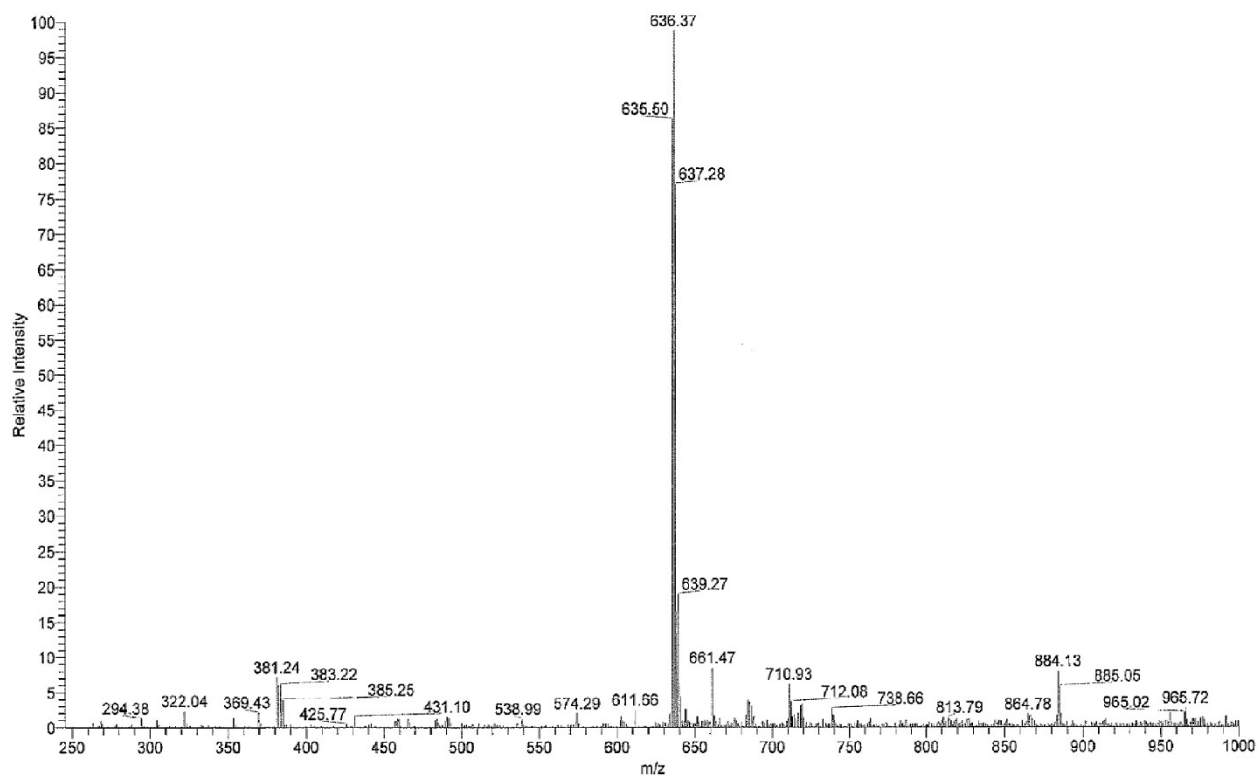


Figure S5: MS spectrum of complex [(pn)Pt(curc)][CF<sub>3</sub>SO<sub>3</sub>], **1**

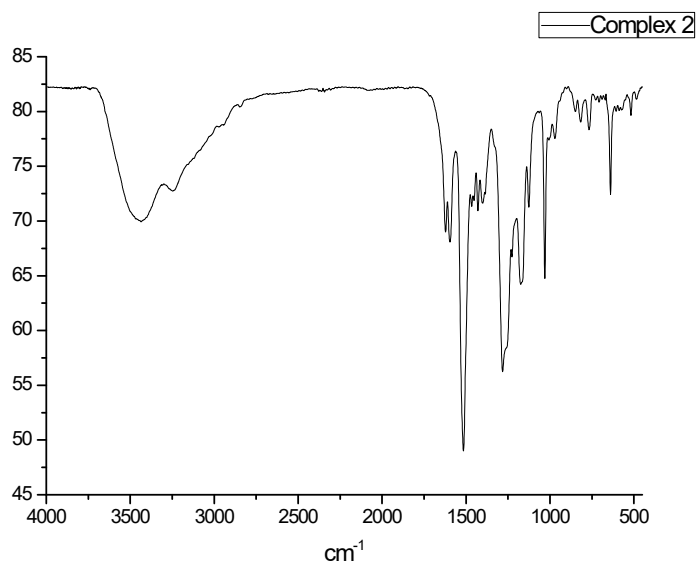


Figure S6: FT-IR spectrum of complex  $[(pic)Pt(curc)][CF_3SO_3]$ , **2**.

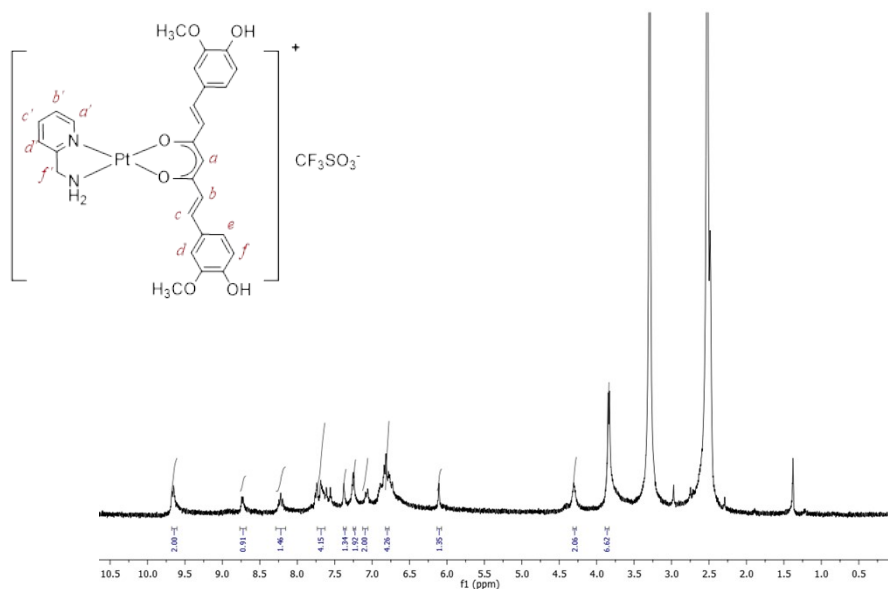


Figure S7:  $^1H$ -NMR spectrum of complex  $[(pic)Pt(curc)][CF_3SO_3]$ , **2**.

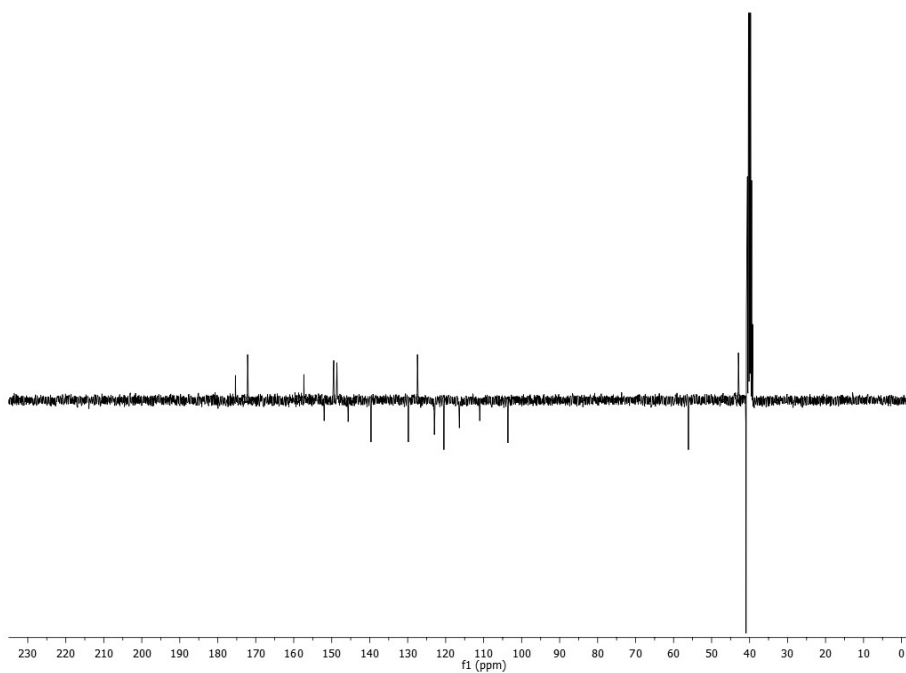


Figure S8: <sup>13</sup>C-NMR spectrum of complex [(pic)Pt(curc)][CF<sub>3</sub>SO<sub>3</sub>], **2**.

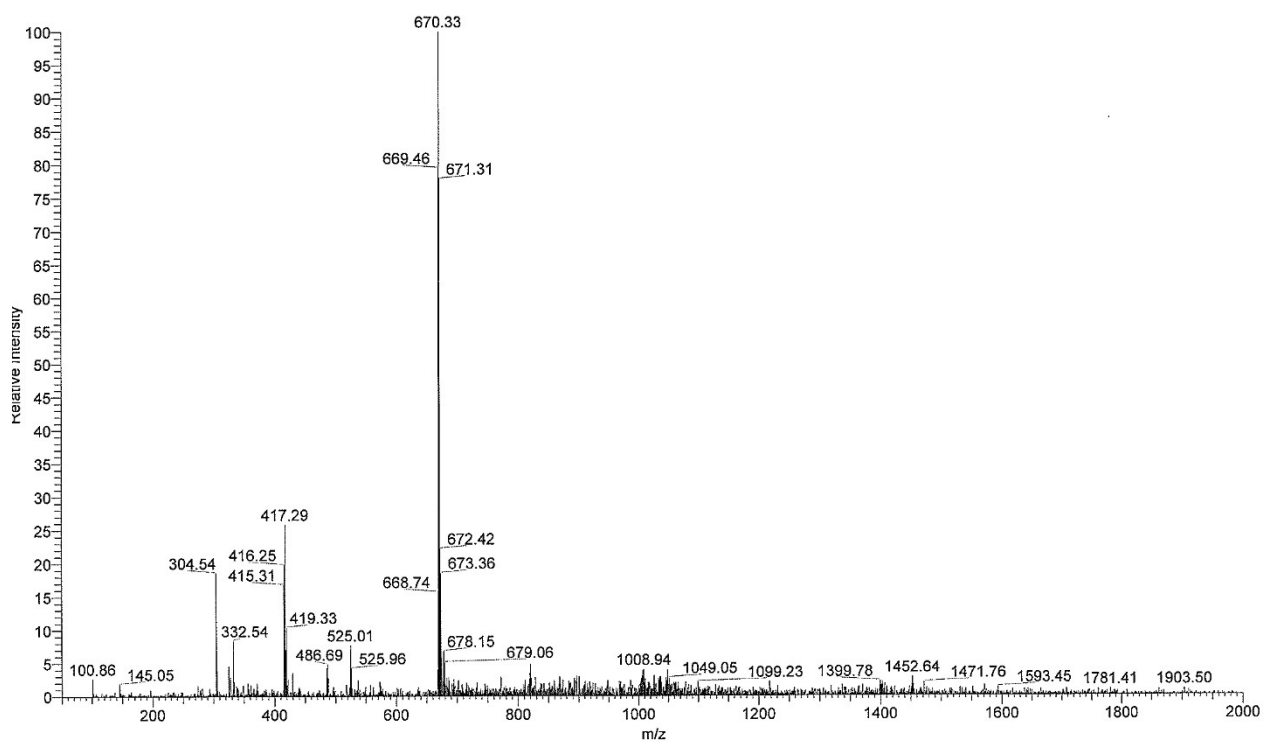
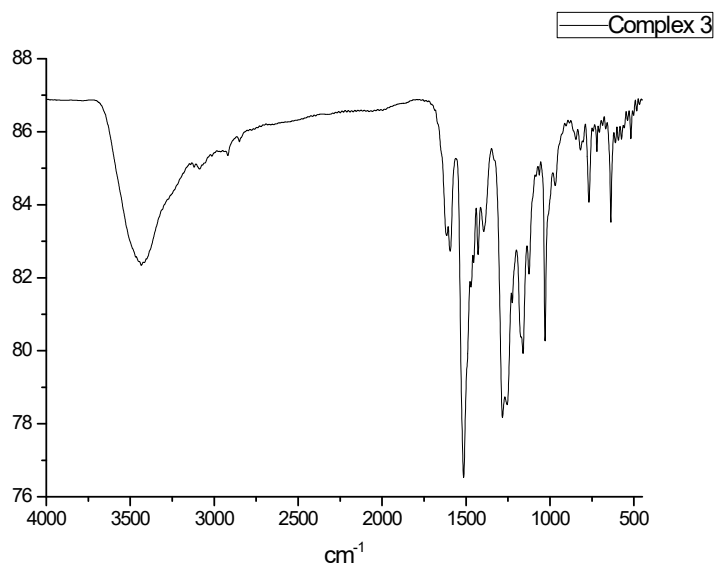
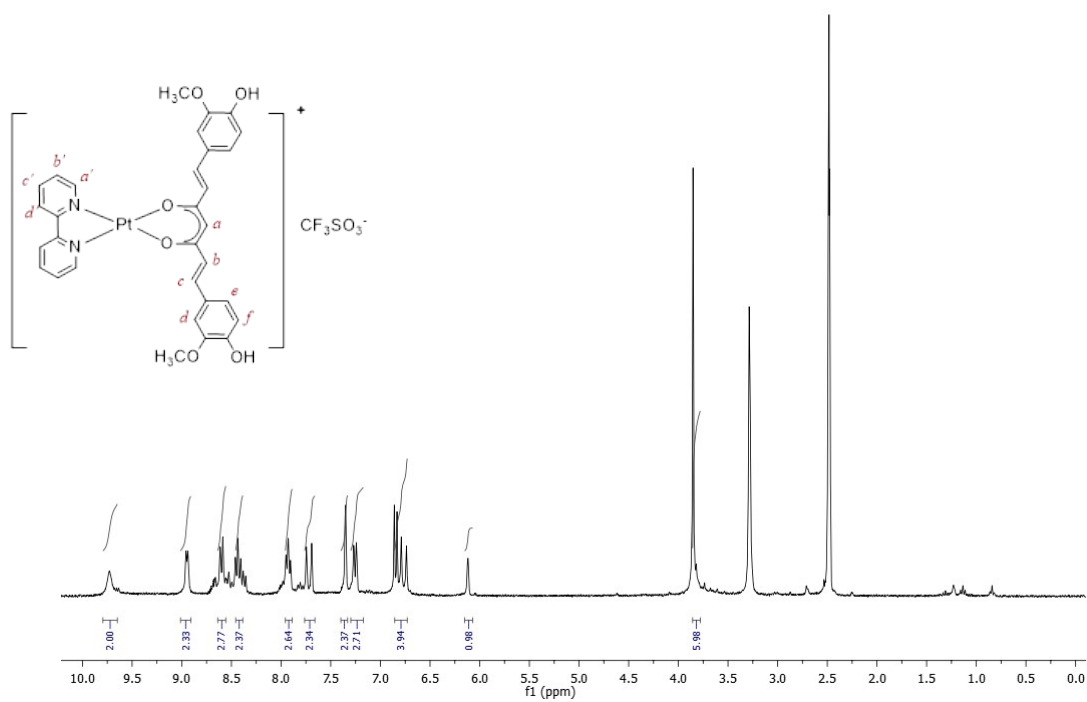


Figure S9: MS spectrum of complex [(pic)Pt(curc)][CF<sub>3</sub>SO<sub>3</sub>], **2**.



**Figure S10:** FT-IR spectrum of complex[(bipy)Pt(curc)][CF<sub>3</sub>SO<sub>3</sub>], **3**.



**Figure S11:** <sup>1</sup>H-NMR spectrum of complex [(bipy)Pt(curc)][CF<sub>3</sub>SO<sub>3</sub>], **3**.



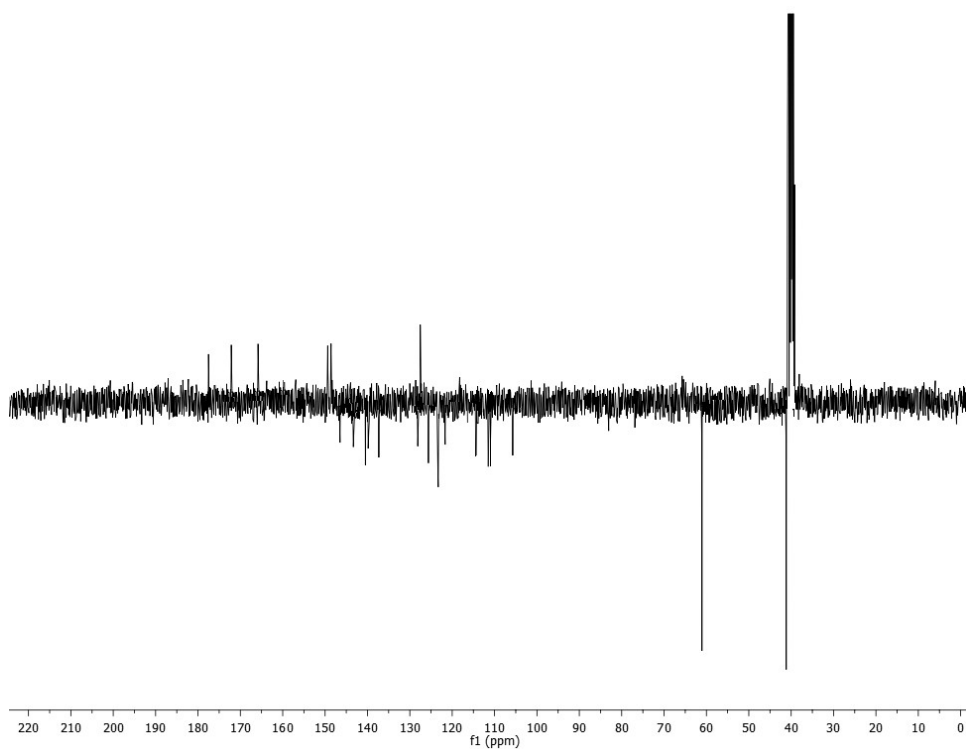


Figure S12:  $^{13}\text{C}$ -NMR spectrum of complex  $[(\text{bipy})\text{Pt}(\text{curc})][\text{CF}_3\text{SO}_3]$ , **3**.

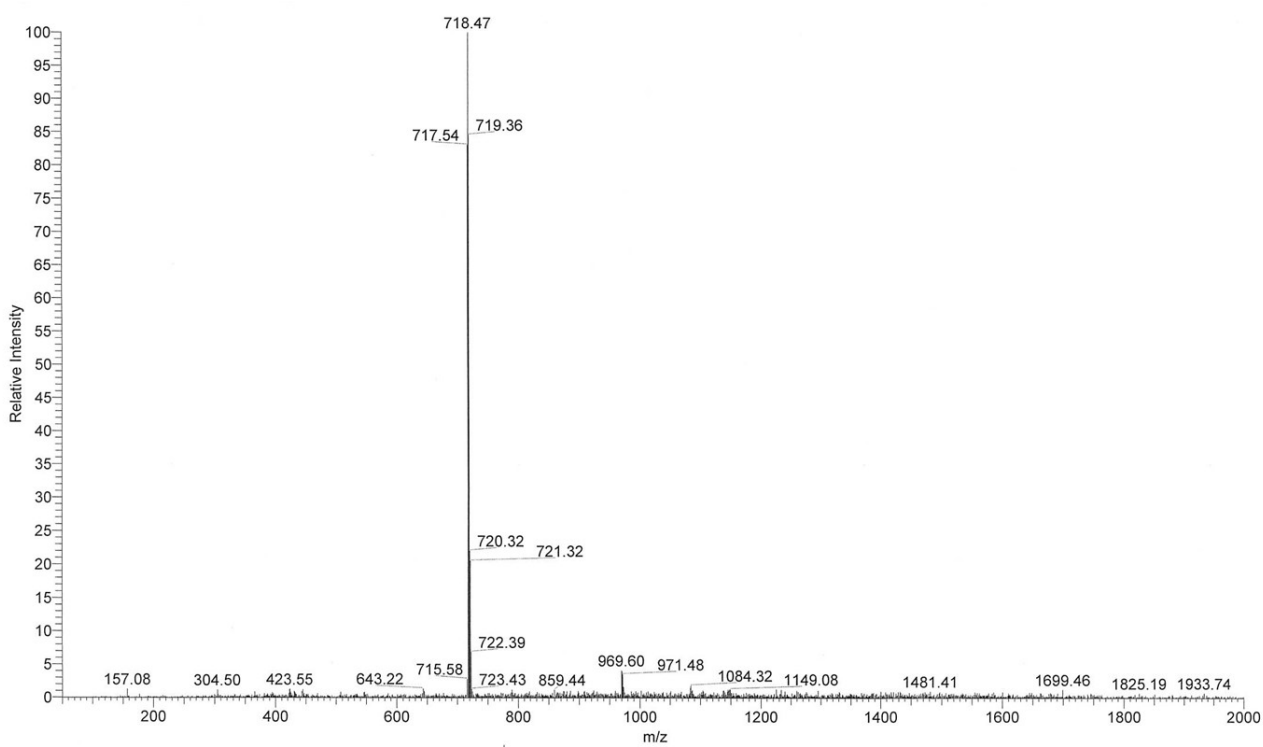


Figure S13: MS spectrum of complex  $[(\text{bipy})\text{Pt}(\text{curc})][\text{CF}_3\text{SO}_3]$ , **3**.

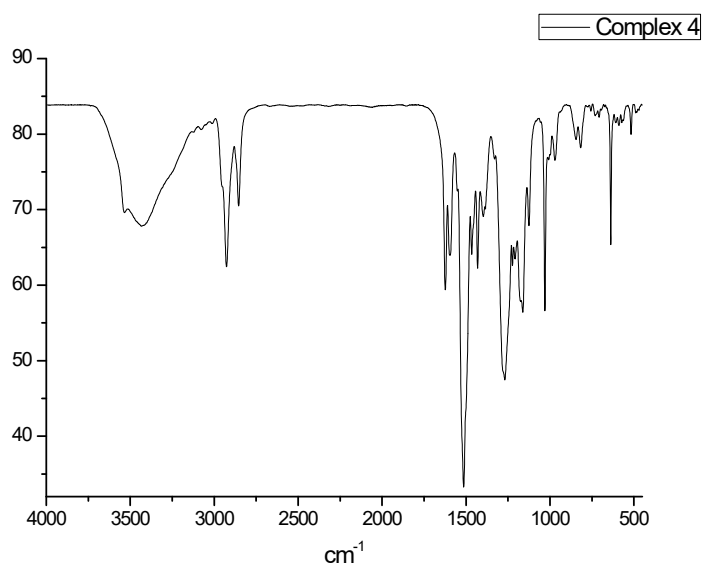


Figure S14: FT-IR spectrum of complex [(bipy-C<sub>9</sub>)Pt(curc)][CF<sub>3</sub>SO<sub>3</sub>], **4**.

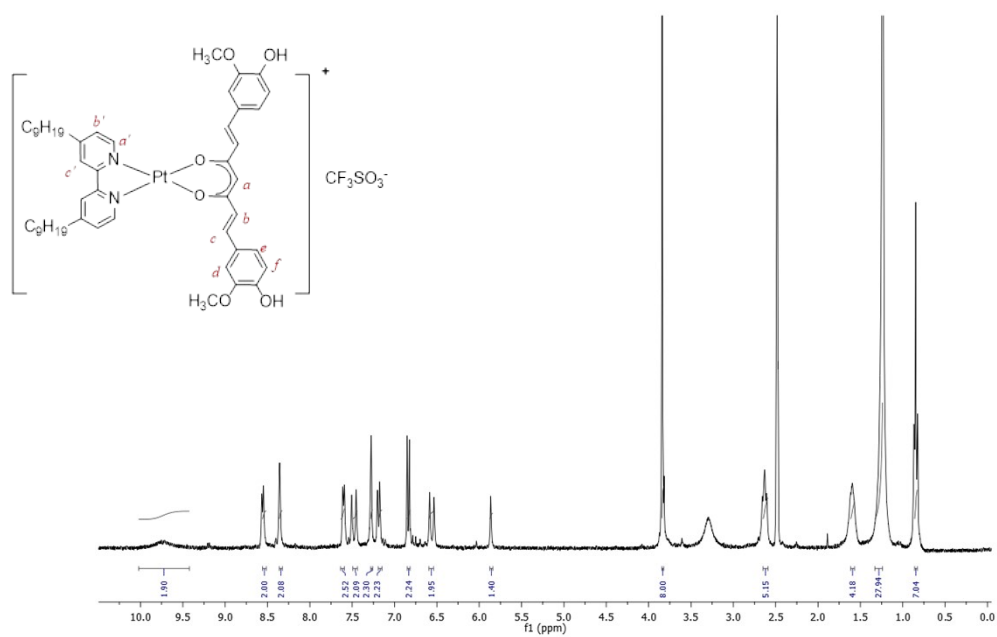
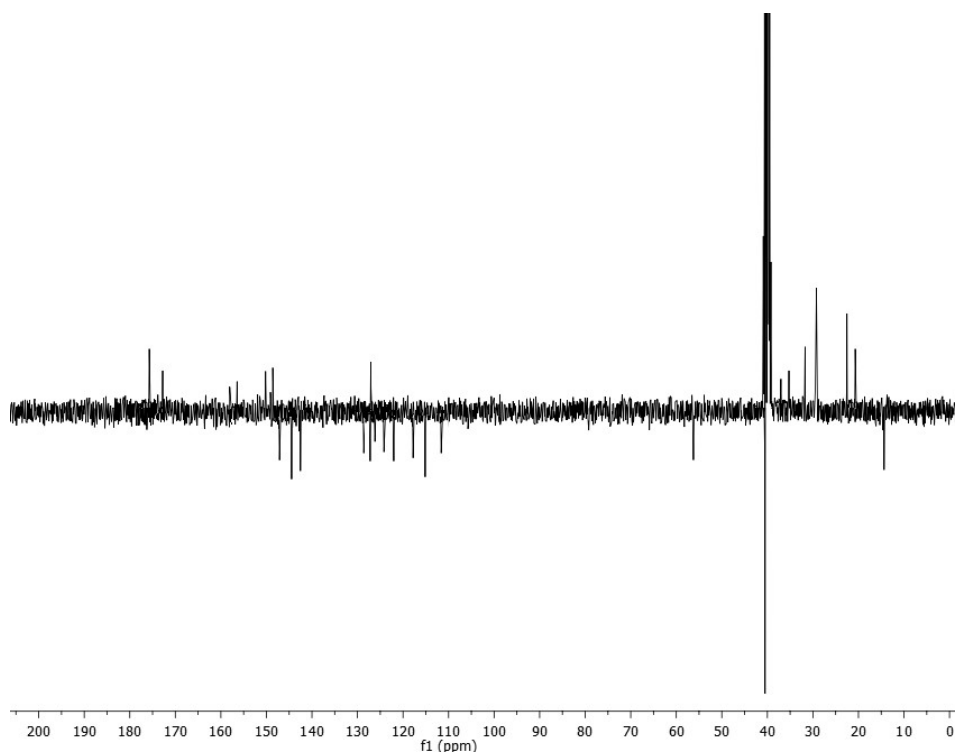
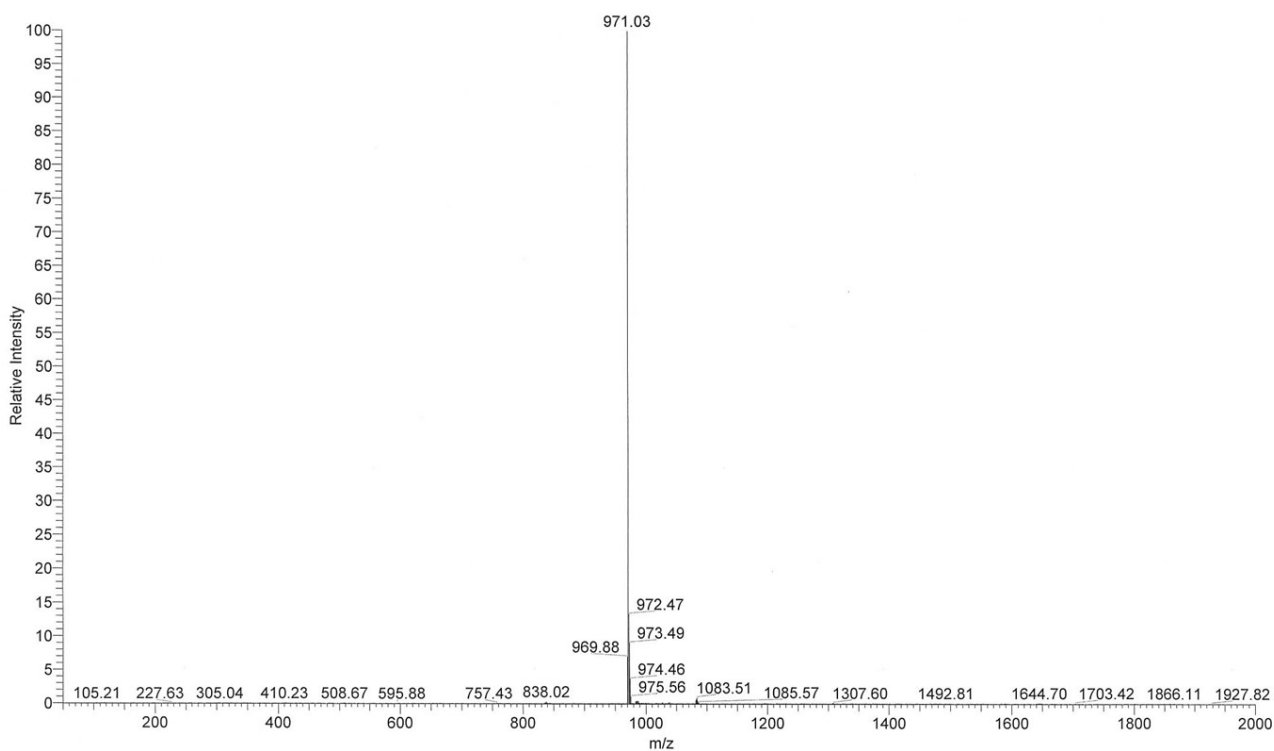


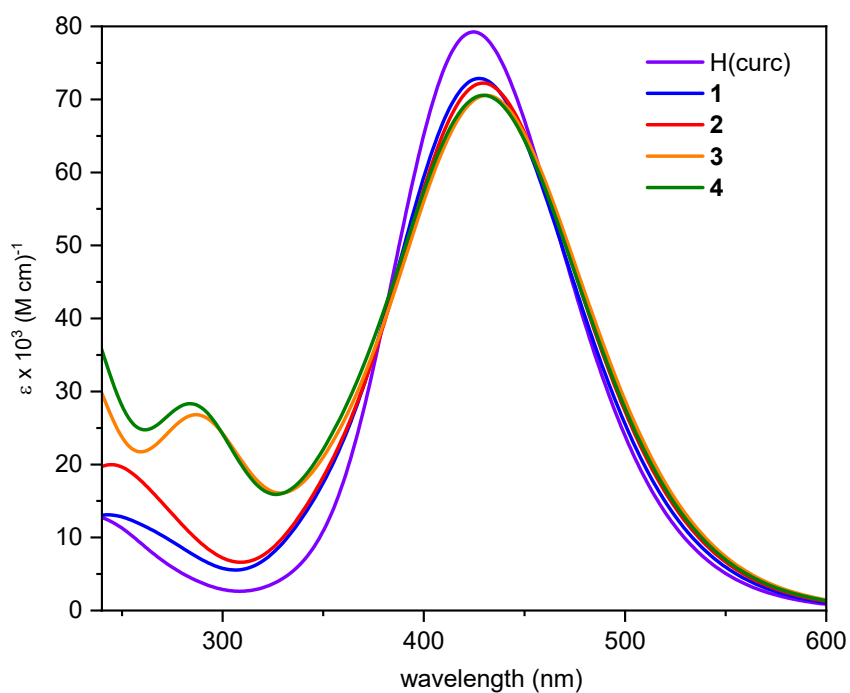
Figure S15: <sup>1</sup>H-NMR spectrum of complex [(bipy-C<sub>9</sub>)Pt(curc)][CF<sub>3</sub>SO<sub>3</sub>], **4**.



**Figure S16:**  $^{13}\text{C}$ -NMR spectrum of complex  $[(\text{bipy-C}_9)\text{Pt}(\text{curc})][\text{CF}_3\text{SO}_3]$ , **4**.



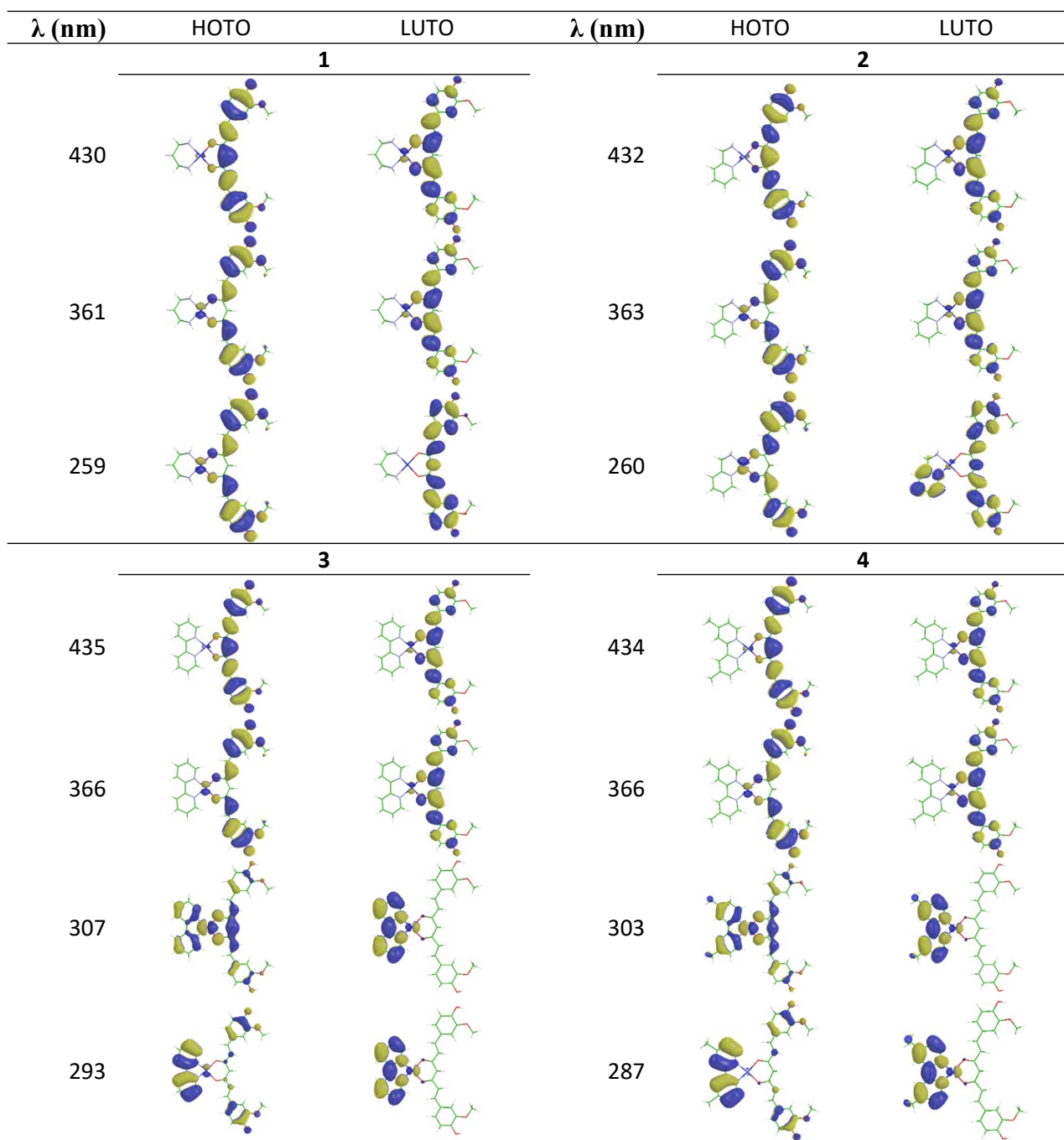
**Figure S17:** MS spectrum of complex  $[(\text{bipy-C}_9)\text{Pt}(\text{curc})][\text{CF}_3\text{SO}_3]$ , **4**.



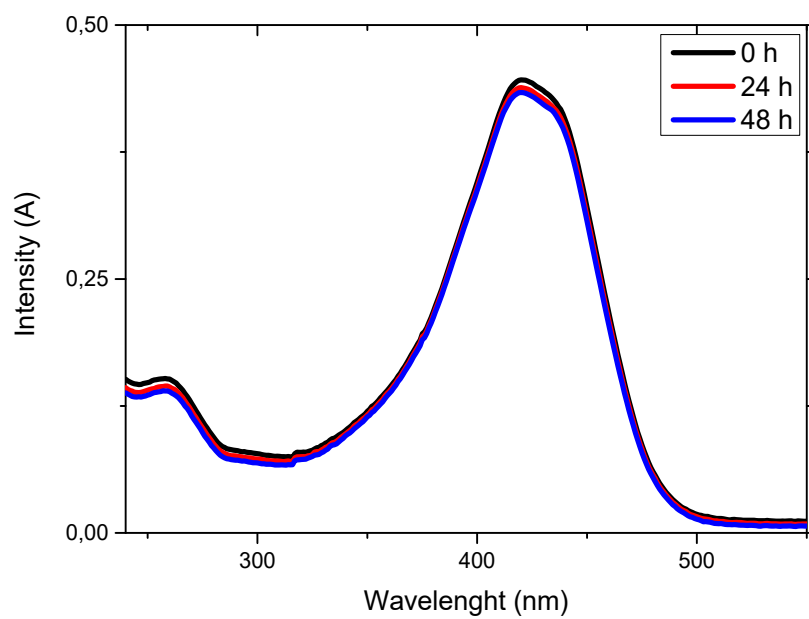
**Figure S18:** Simulated absorption spectra of **H(curc)** (violet) and complexes **1** (blue), **2** (red), **3** (orange) and **4** (green) in implicit solvent methanol.

**Table S1:** Absorption wavelength ( $\lambda$ , nm), oscillator strength ( $f$ ), MO contribution (%) for H(curc) and its Pt(II) cationic complexes **1-4**.

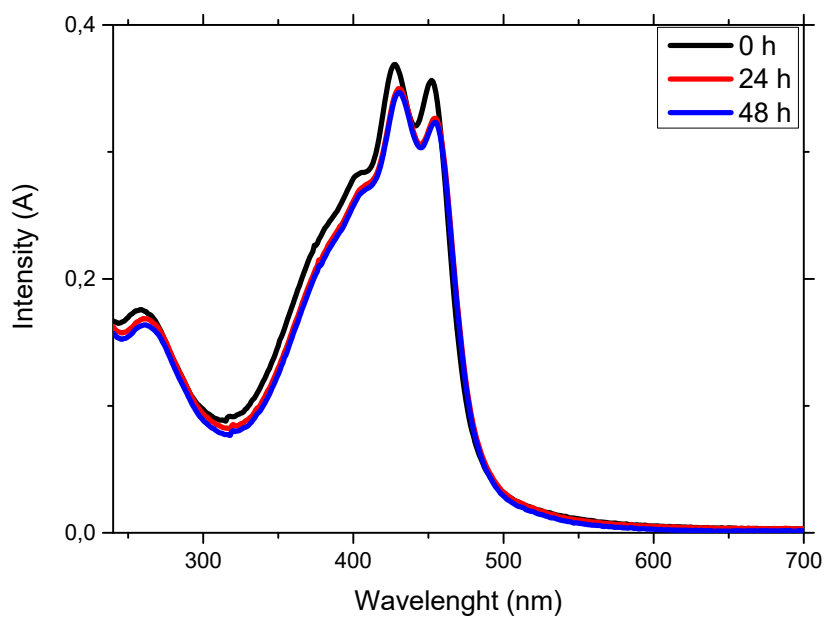
Comp	Band	$\lambda$	$f^a$	MO contribution <sup>b</sup>	Theoretical Assignment
<b>H(curc)</b>	<b>I</b>	425	1.949	H $\rightarrow$ L 94%	$\pi\pi^*$
	<b>II</b>	257	0.105	H-1 $\rightarrow$ L+1 67%	$\pi\pi^*$
		243	0.107	H $\rightarrow$ L+3 34% H-1 $\rightarrow$ L+2 18%	
<b>1</b>	<b>I</b>	430	1.749	H $\rightarrow$ L 93%	LMCT
		361	0.308	H-1 $\rightarrow$ L 93%	LC/MLCT
	<b>II</b>	259	0.122	H-1 $\rightarrow$ L+1 57% H $\rightarrow$ L+3 16%	MLCT/LC <sub>curc</sub>
<b>2</b>	<b>I</b>	432	1.725	H $\rightarrow$ L 93%	LMCT
		363	0.349	H-1 $\rightarrow$ L 92%	LC/MLCT
	<b>II</b>	260	0.130	H-1 $\rightarrow$ L+2 41%	MLCT/LC <sub>curc</sub>
<b>3</b>	<b>I</b>	435	1.665	H $\rightarrow$ L 93%	LMCT
		366	0.417	H-1 $\rightarrow$ L 89%	LC/MLCT
	<b>II</b>	307	0.131	H-2 $\rightarrow$ L+1 40%, H-4 $\rightarrow$ L+1 33%	LC <sub>bipy</sub>
		293	0.309	H-5 $\rightarrow$ L+1 70%	L <sub>curc</sub> L <sub>bipy</sub> CT
<b>4</b>	<b>I</b>	434	1.657	H $\rightarrow$ L 93%	LMCT
		366	0.441	H-1 $\rightarrow$ L 90%	LC/MLCT
	<b>II</b>	303	0.180	H-2 $\rightarrow$ L+1 36%, H-3 $\rightarrow$ L 25%, H-4 $\rightarrow$ L+1 20%	LC <sub>bipy</sub>
		287	0.344	H-5 $\rightarrow$ L+1 61%	L <sub>curc</sub> L <sub>bipy</sub> CT



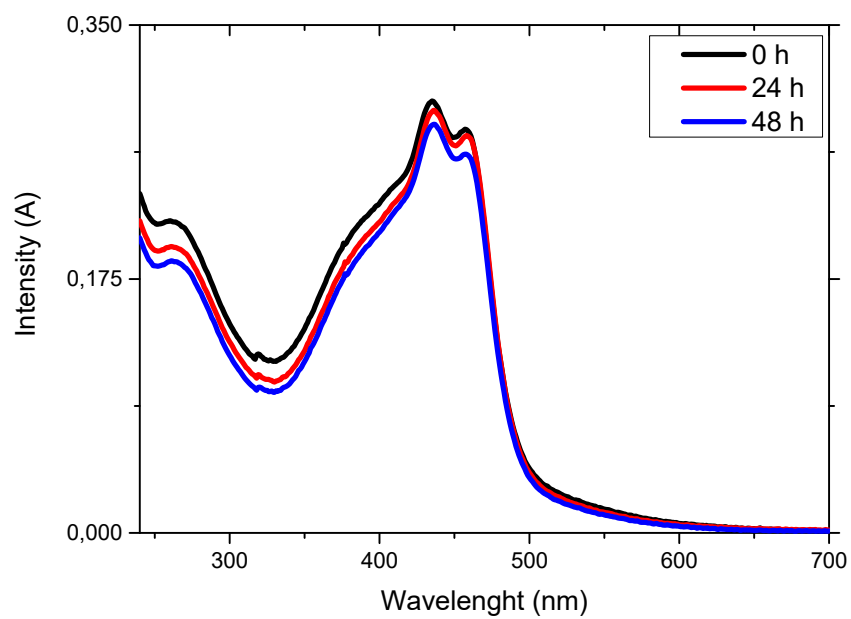
**Figure S19:** Natural Transition Orbitals (NTOs) for the main excitations of Pt(II) complexes **1-4**.



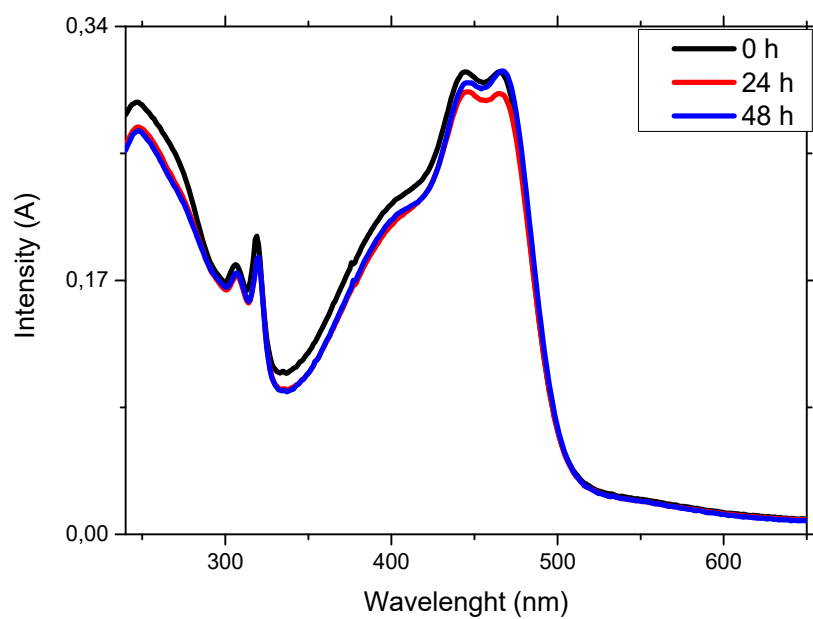
**Figure S20:** Absorption spectra of **H(curc)** in MeOH dilute solution (10 μM) varying time.



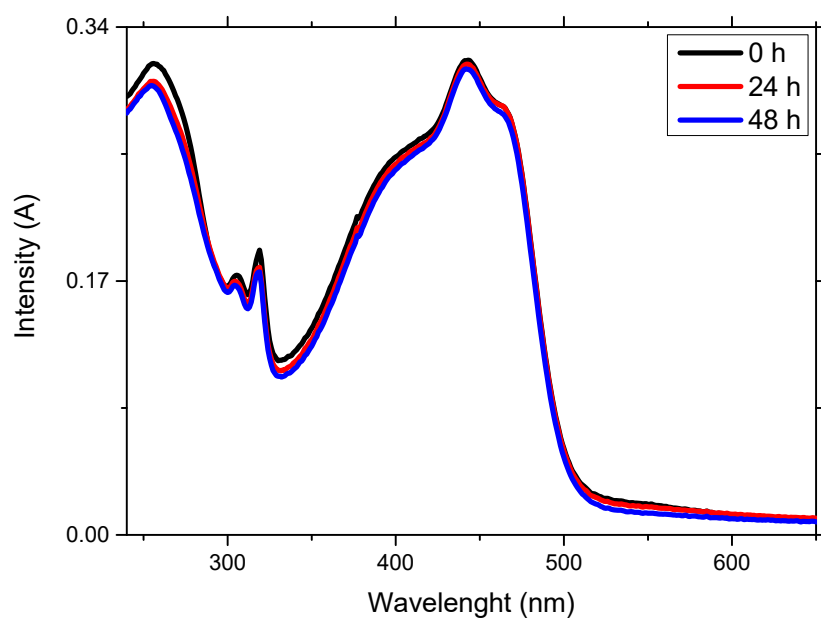
**Figure S21:** Absorption spectra of complex **1** in MeOH dilute solution (10 μM) varying time.



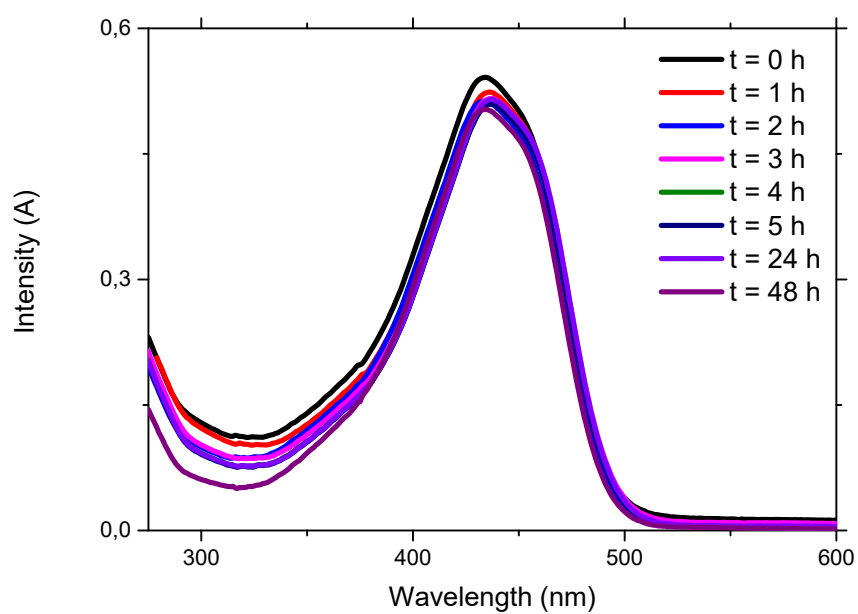
**Figure S22:** Absorption spectra of complex 2 in MeOH dilute solution (10 μM) varying time.



**Figure S23:** Absorption spectra of complex 3 in MeOH dilute solution (10 μM) varying time.

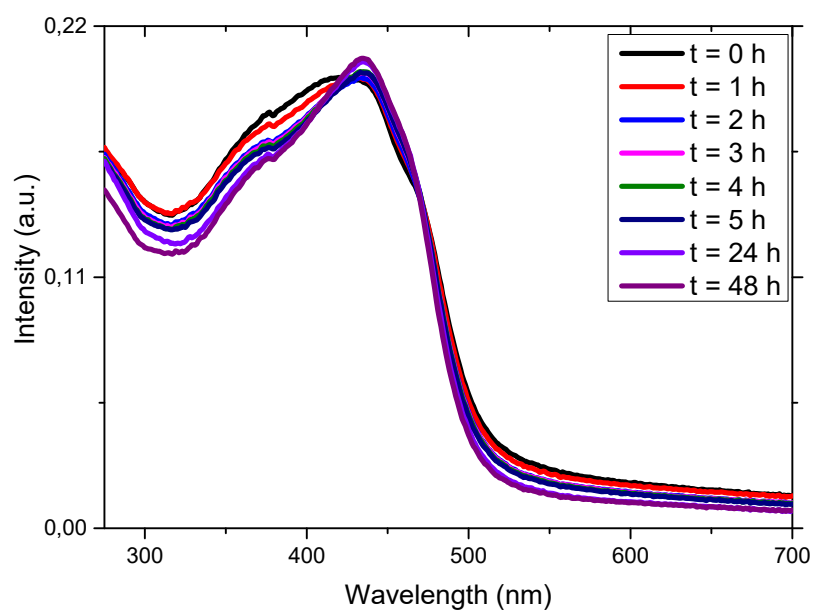


**Figure S24:** Absorption spectra of complex 4 in MeOH dilute solution (10 μM) varying time.

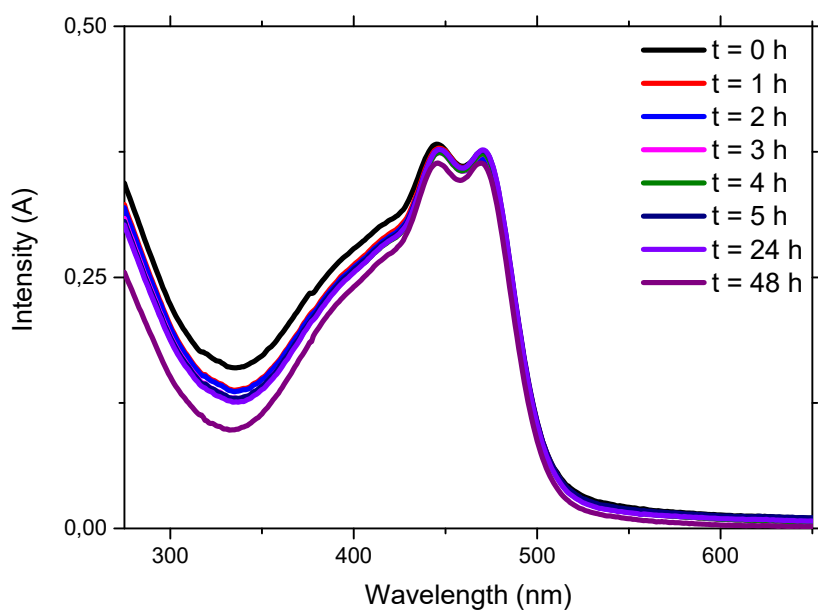


**Figure S25:** Absorption spectra of H(curc) in DMSO dilute solution (10 μM) varying time.

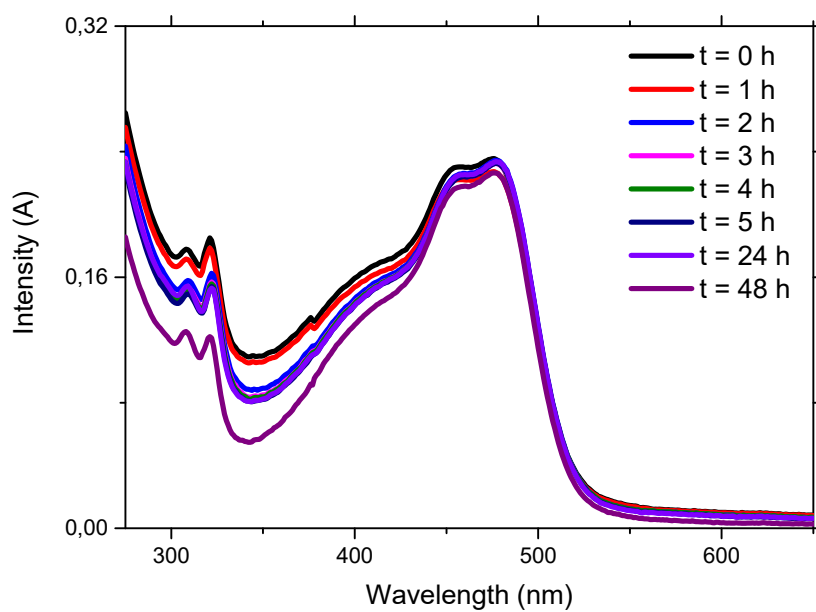




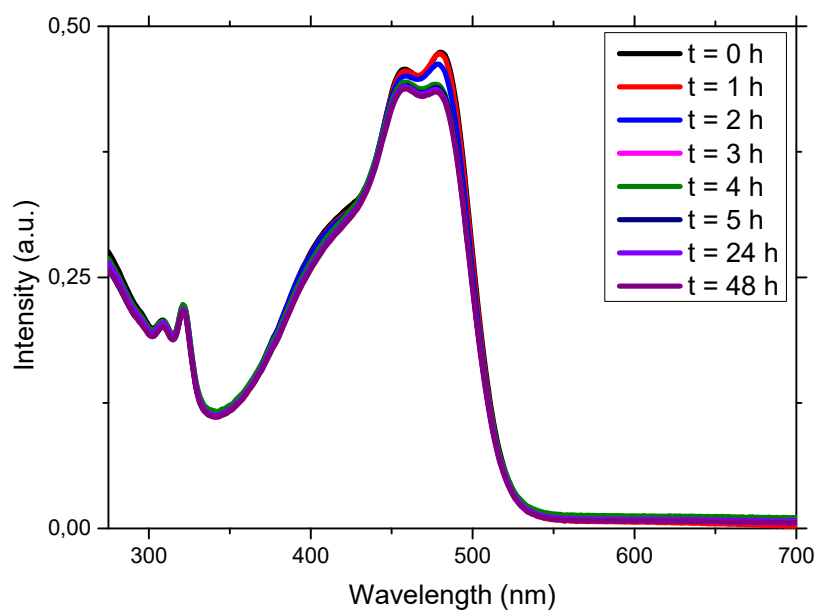
**Figure S26:** Absorption spectra of complex **1** in DMSO dilute solution (10 μM) varying time.



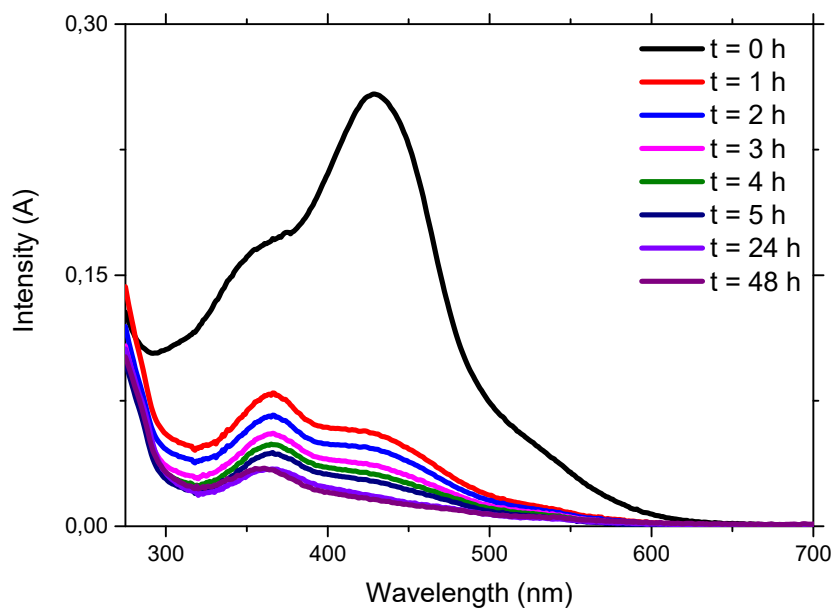
**Figure S27:** Absorption spectra of complex **2** in DMSO dilute solution (10 μM) varying time.



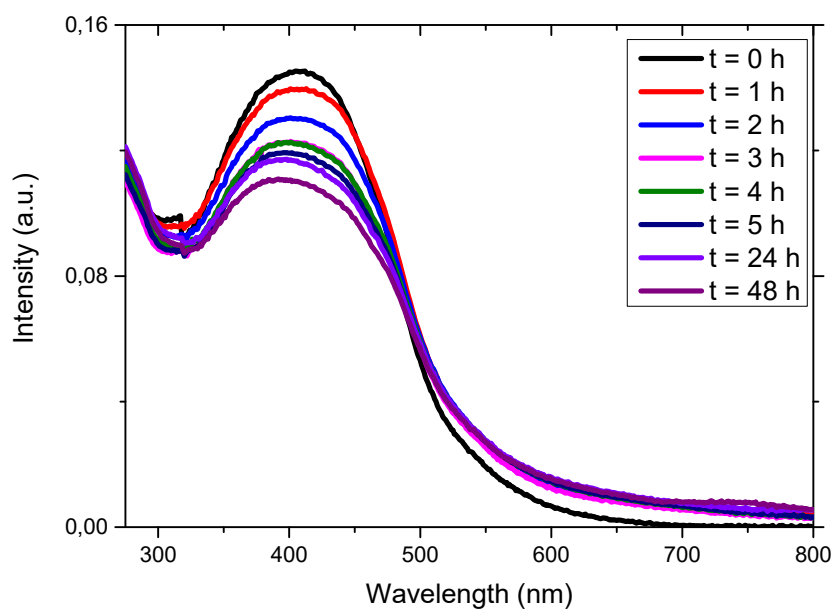
**Figure S28:** Absorption spectra of complex **3** in DMSO dilute solution (10  $\mu$ M) varying time.



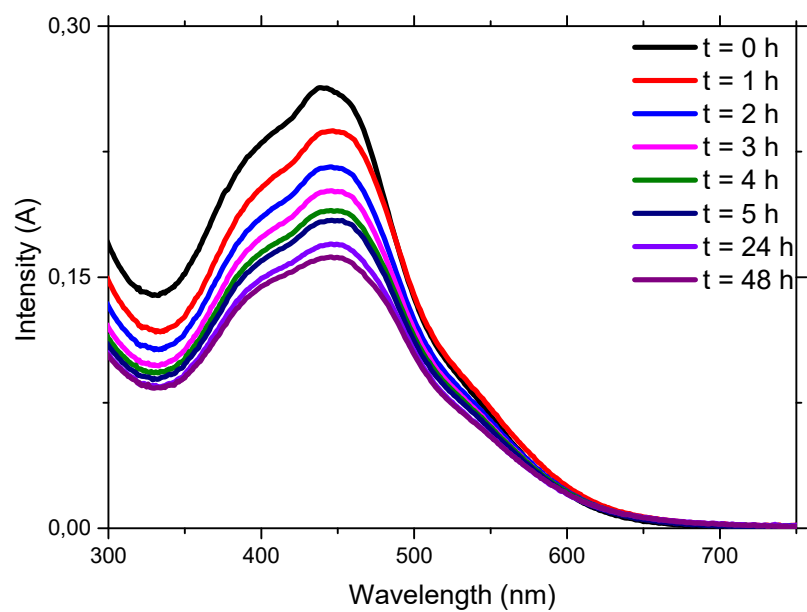
**Figure S29:** Absorption spectra of complex **4** in DMSO dilute solution (10  $\mu$ M) varying time.



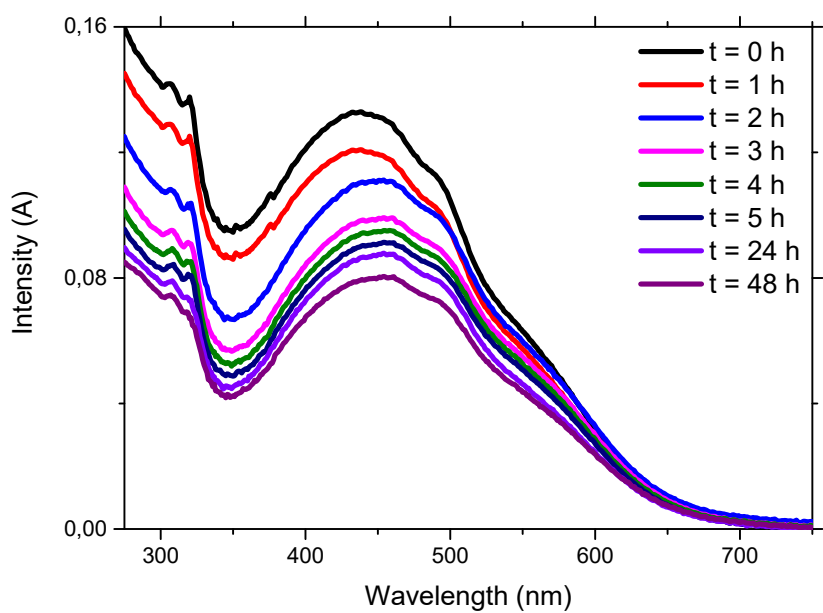
**Figure S30:** Absorption spectra of **H(curc)** in 10% DMSO/PBS dilute solution (10  $\mu$ M) varying time.



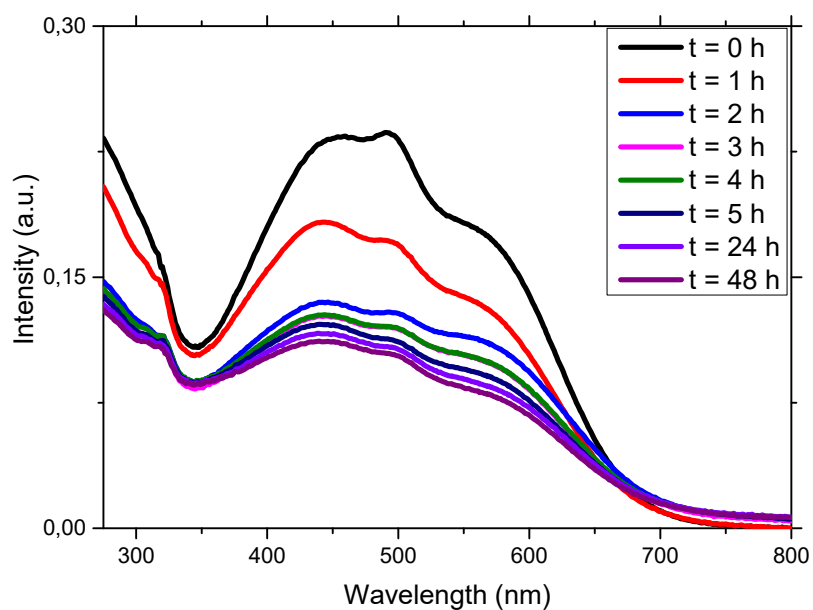
**Figure S31:** Absorption spectra of complex **1** in 10% DMSO/PBS dilute solution (10  $\mu$ M) varying time.



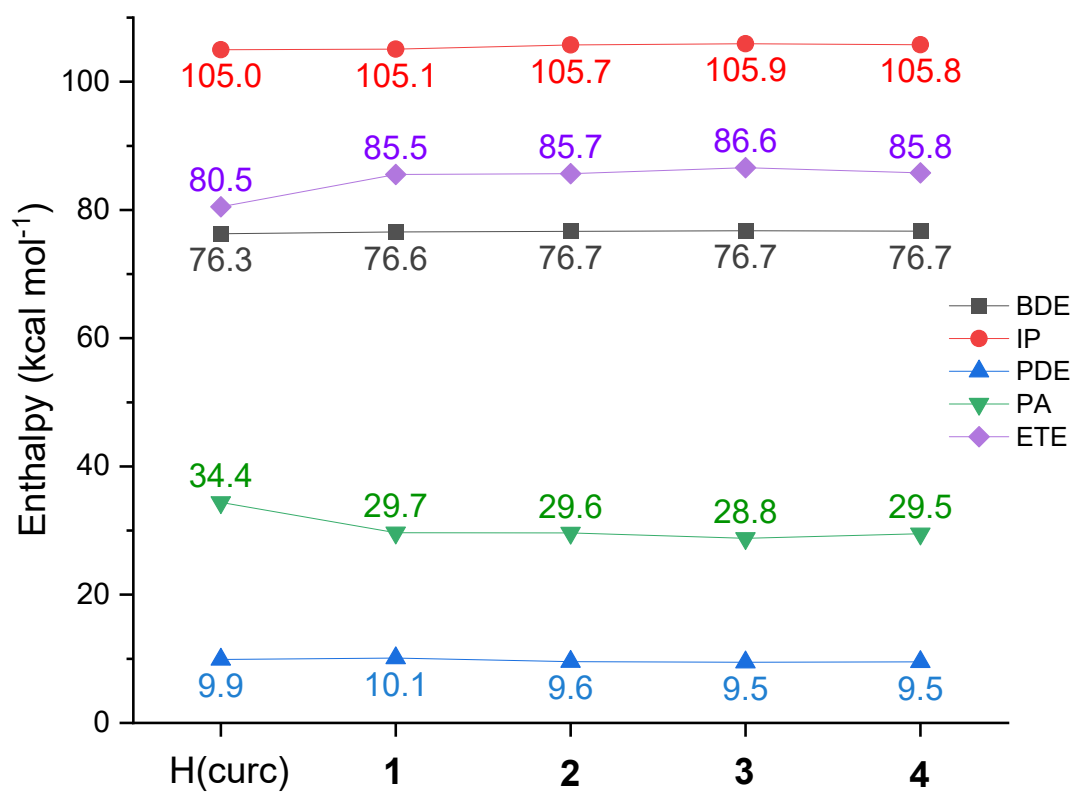
**Figure S32:** Absorption spectra of complex **2** in 10% DMSO/PBS dilute solution (10  $\mu$ M) varying time.



**Figure S33:** Absorption spectra of complex **3** in 10% DMSO/PBS dilute solution (10  $\mu$ M) varying time.



**Figure S34:** Absorption spectra of complex **4** in 10% DMSO/PBS dilute solution (10  $\mu$ M) varying time.



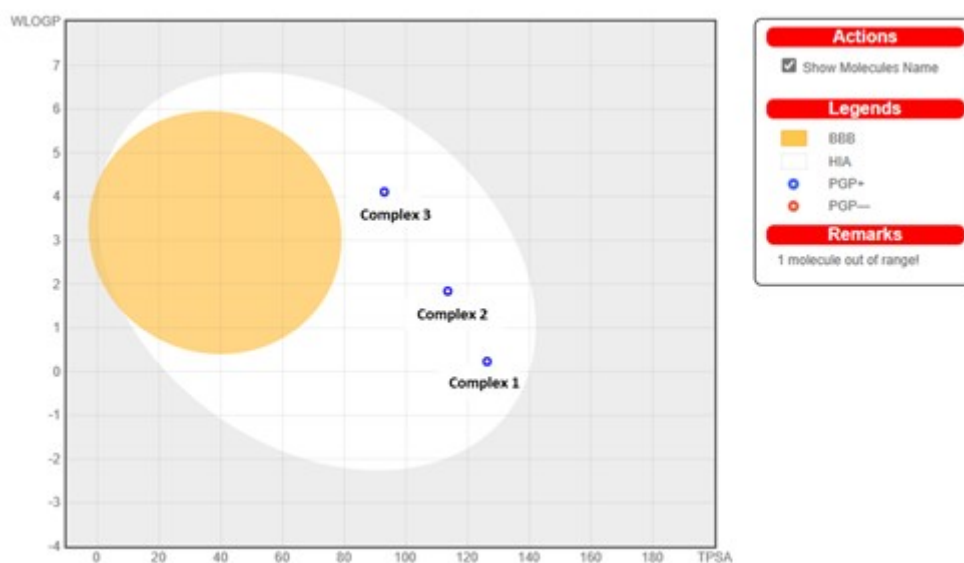
**Figure S35:** Thermodynamic parameters: bond dissociation enthalpies (BDE, grey), electron transfer enthalpies (ETE, light violet), adiabatic ionization potentials (IP, red), O–H proton dissociation enthalpies (PDE, blue) and proton affinities (PA, green), at 298.15 K.

**Table S2:** ADME properties of the Pt(II) complexes 1-4.

	1	2	3	4
MW (g/mol)	636.57	670.59	718.63	971.11
H-bond acceptors	6	6	6	6
H-bond donors	4	3	2	2
TPSA (Å)	126.28	113.55	92.98	100.82
LogS	-5.88	-6.67	-7.96	-14.11
GI absorption	High	High	High	Low
BBB permeant	No	No	No	No
Lipinski #violations	1	1	1	2
Bioavailability Score	0.56	0.56	0.56	0.56

**Table S3.** Theoretical and experimental logP<sub>ow</sub> values of the Pt(II) complexes 1-4.

	1	2	3	4
LogP <sub>ow</sub> <sup>theo</sup>	-0.32	0.89	2.69	8.01
logP <sub>ow</sub> <sup>exp</sup>	2.79	3.22	3.65	8.12

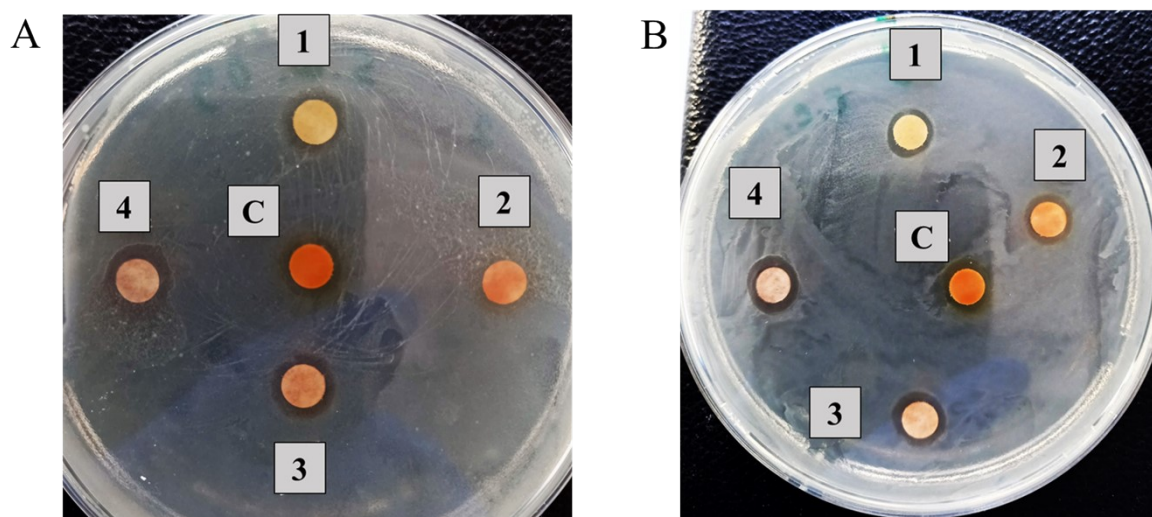


**Figure S36: BOILED-Egg Model of complexes 1-3.**

	Complex Concentration ( $\mu\text{g}/\mu\text{L}$ )					
	0.05	0.1	0.2	0.3	0.4	0.5
<b>H(curc)</b>	0	$0.20 \pm 0.5$	$0.60 \pm 0.4$	$0.80 \pm 0.1$	$1.35 \pm 0.3$	$1.99 \pm 0.1$
<b>1</b>	$0.50 \pm 0.3$	$0.90 \pm 0.1$	$1.25 \pm 0.1$	$1.78 \pm 0.1$	$2.11 \pm 0.2$	$2.31 \pm 0.2$
<b>2</b>	$0.50 \pm 0.2$	$1.08 \pm 0.1$	$1.40 \pm 0.2$	$2.10 \pm 0.1$	$2.40 \pm 0.1$	$2.60 \pm 0.1$
<b>3</b>	$0.97 \pm 0.3$	$1.50 \pm 0.2$	$1.70 \pm 0.3$	$2.45 \pm 0.2$	$2.70 \pm 0.2$	$2.88 \pm 0.1$
<b>4</b>	$1.57 \pm 0.5$	$2.00 \pm 0.3$	$2.55 \pm 0.2$	$2.67 \pm 0.1$	$2.93 \pm 0.2$	$3.12 \pm 0.1$
<b>DMSO</b>	0	0	0	0	0	0

	Complex Concentration ( $\mu\text{g}/\mu\text{L}$ )					
	0.05	0.1	0.2	0.3	0.4	0.5
<b>H(curc)</b>	0	0	0	$0.80 \pm 0.2$	$1.30 \pm 0.1$	$1.80 \pm 0.2$
<b>1</b>	0	$0.26 \pm 0.1$	$0.52 \pm 0.4$	$1.08 \pm 0.2$	$1.6 \pm 0.4$	$2.06 \pm 0.1$
<b>2</b>	0	0	$0.30 \pm 0.2$	$0.60 \pm 0.3$	$0.80 \pm 0.2$	$1.00 \pm 0.4$
<b>3</b>	0	$0.15 \pm 0.2$	$0.80 \pm 0.5$	$1.40 \pm 0.04$	$1.90 \pm 0.3$	$2.37 \pm 0.1$
<b>4</b>	0	$0.40 \pm 0.1$	$0.60 \pm 0.3$	$1.80 \pm 0.6$	$2.50 \pm 0.2$	$3.00 \pm 0.1$
<b>DMSO</b>	0	0	0	0	0	0

**Table S4: Antimicrobial activity data against *E. coli* (on the top) and *S. aureus* (on the bottom). The diameter(mm) of growth inhibition area is subtracted from the disk diameter (6mm). The values are the mean of three replicates  $\pm$  standard deviation. 0= no alone.**



**Figure S37:** Inhibition halos on *S. aureus* (A) and *E. Coli* (B) at 0.5 µg/µl. C: H(curc).