

*Supporting Information*

**Titanium complexes with unsymmetrical imidazolin-2-iminato ligands**

Marvin Koneczny,<sup>a</sup> Arife Büsra Erol,<sup>a</sup> Marc Mauduit,<sup>b</sup> Moris Eisen,<sup>c</sup> and Matthias Tamm<sup>\*a</sup>

a. Institut für Anorganische und Analytische Chemie, Technische Universität Braunschweig, Hagenring 30, 38106 Braunschweig, Germany.

E-mail: [m.tamm@tu-bs.de](mailto:m.tamm@tu-bs.de)

b. Univ Rennes; Ecole Nationale Supérieure de Chimie de Rennes, CNRS, ISCR-UMR 6226, F-35000 Rennes, France

c. Schulich Faculty of Chemistry, Technion – Israel Institute of Technology, Technion City, 32000 Haifa, Israel

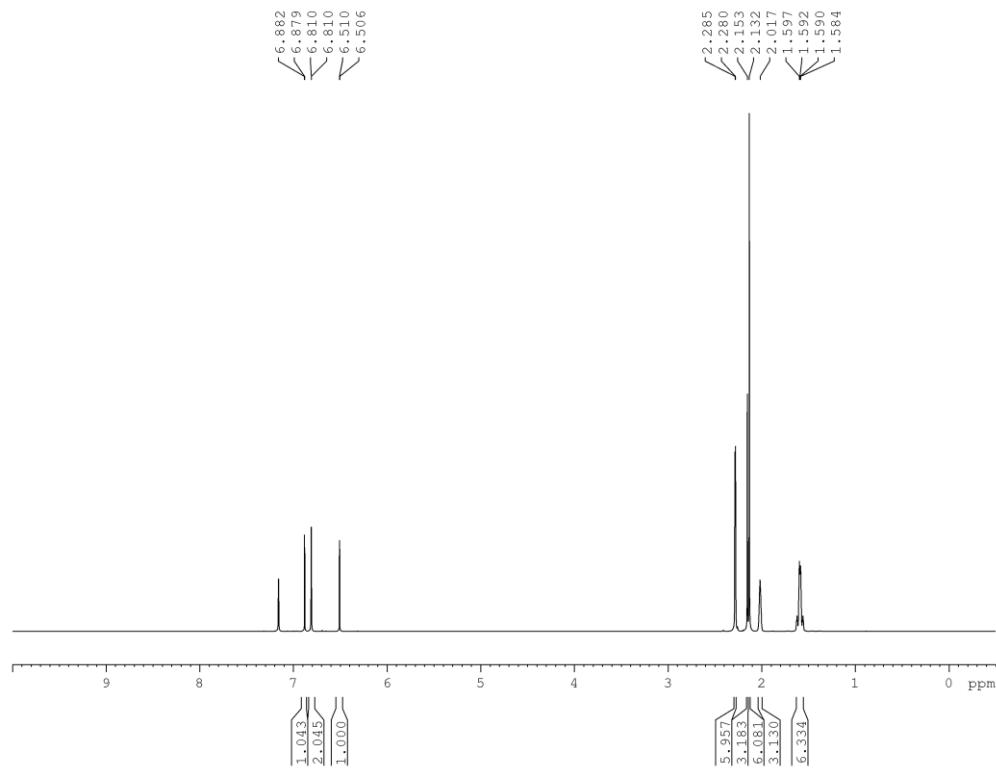
## Table of Contents

1. Spectra .....	3
1.1 1-(Adamantyl)-3-(2,4,6-trimethylphenyl)-imidazolin-2-ylidene (1a).....	3
1.2 1-(Adamantyl)-3-(2,6-diisopropylphenyl)-imidazolin-2-ylidene (1b) .....	4
1.3 [ $\text{Im}^{\text{AdMes}}\text{NSiMe}_3$ ] (2a) .....	5
1.4 [ $\text{Im}^{\text{AdDipp}}\text{NSiMe}_3$ ] (2b) .....	6
1.5 [ $\text{Im}^{\text{AdMes}}\text{NH}$ ] (3a) .....	7
1.6 [ $\text{Im}^{\text{AdDipp}}\text{NH}$ ] (3b) .....	8
1.7 [ $(\text{Im}^{\text{AdMes}}\text{N})\text{TiCl}_3$ ] (4a) .....	9
1.8 [ $(\text{Im}^{\text{AdDipp}}\text{N})\text{TiCl}_3$ ] (4b) .....	10
1.9 [ $\text{Cp}(\text{Im}^{\text{AdMes}}\text{N})\text{TiCl}_2$ ] (5a) .....	11
1.10 [ $\text{Cp}(\text{Im}^{\text{AdDipp}}\text{N})\text{TiCl}_2$ ] (5b) .....	12
1.11 [ $(\text{Im}^{\text{AdMes}}\text{N})_2\text{TiCl}_2$ ] (6a).....	13
1.12 [ $(\text{Im}^{\text{AdDipp}}\text{N})_2\text{TiCl}_2$ ] (6b) .....	14
2. X-Ray Crystal Structure Determinations .....	16
2.1 1-(Adamantyl)-3-(2,4,6-trimethylphenyl)-imidazolin-2-ylidene (1a).....	17
2.2 1-(Adamantyl)-3-(2,6-diisopropylphenyl)-imidazolin-2-ylidene (1b) .....	19
2.3 [ $(\text{Im}^{\text{AdMes}}\text{NSiMe}_3)$ ] (2a) .....	21
2.4 [ $\text{Im}^{\text{AdMes}}\text{NH}$ ] (3a) .....	23
2.5 [ $\text{Im}^{\text{AdDipp}}\text{NH}$ ] (3b) .....	25
2.6 [ $(\text{Im}^{\text{AdMes}}\text{NH})\text{TiCl}_3$ ] (4a) .....	27
2.7 [ $(\text{Im}^{\text{AdDipp}}\text{N})\text{TiCl}_3$ ] $\text{CH}_2\text{Cl}_2\cdot\text{Solv}$ (4b) .....	29
2.8 [ $\text{Cp}(\text{Im}^{\text{AdMes}}\text{N})\text{TiCl}_2$ ] (5a) .....	31
2.9 [ $\text{Cp}(\text{Im}^{\text{AdDipp}}\text{N})\text{TiCl}_2$ ] (5b) .....	33
2.10 [ $(\text{Im}^{\text{AdMes}}\text{N})_2\text{TiCl}_2\cdot 0.5\text{CH}_2\text{Cl}_2$ ] (6a) .....	35
2.11 [ $(\text{Im}^{\text{AdDipp}}\text{N})_2\text{TiCl}_2\cdot \text{CH}_2\text{Cl}_2$ ] (6b) .....	37

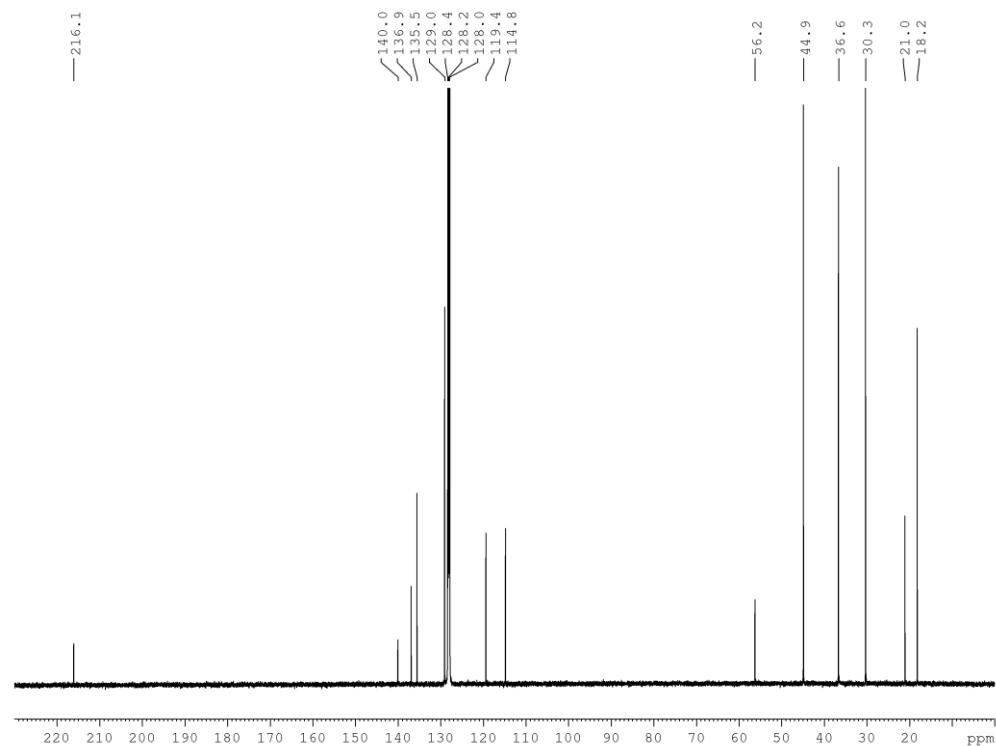
# 1. Spectra

## 1.1 1-(Adamantyl)-3-(2,4,6-trimethylphenyl)-imidazolin-2-ylidene (**1a**)

<sup>1</sup>H NMR (500 MHz, C<sub>6</sub>D<sub>6</sub>) of **1a**:

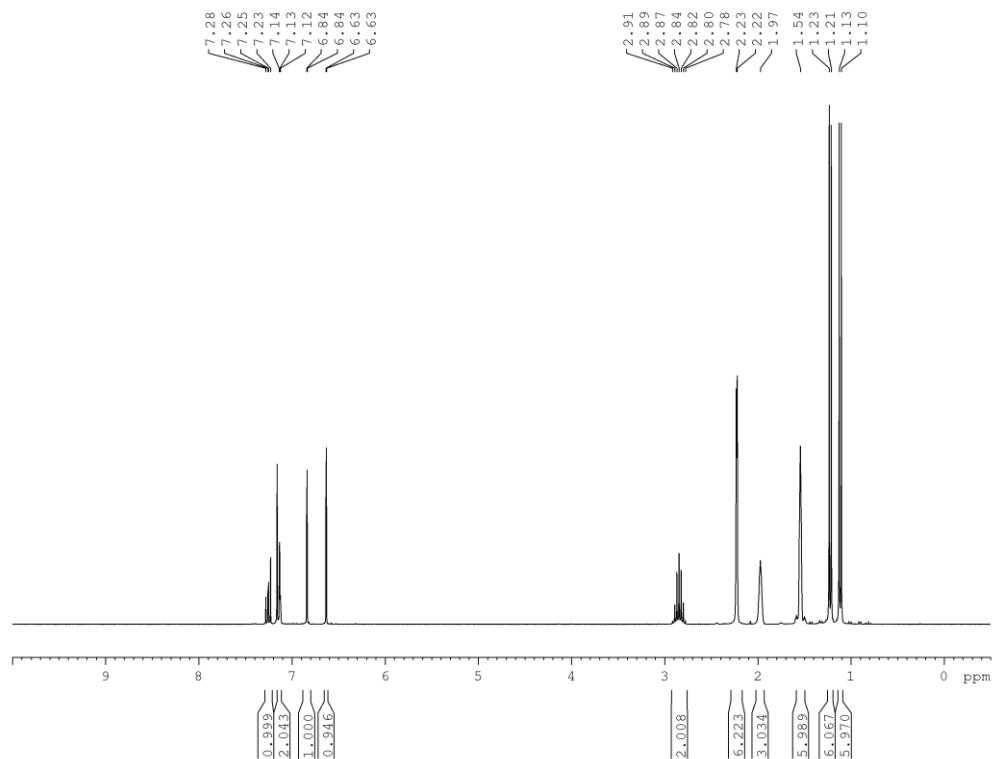


<sup>13</sup>C NMR (126 MHz, C<sub>6</sub>D<sub>6</sub>) of **1a**:

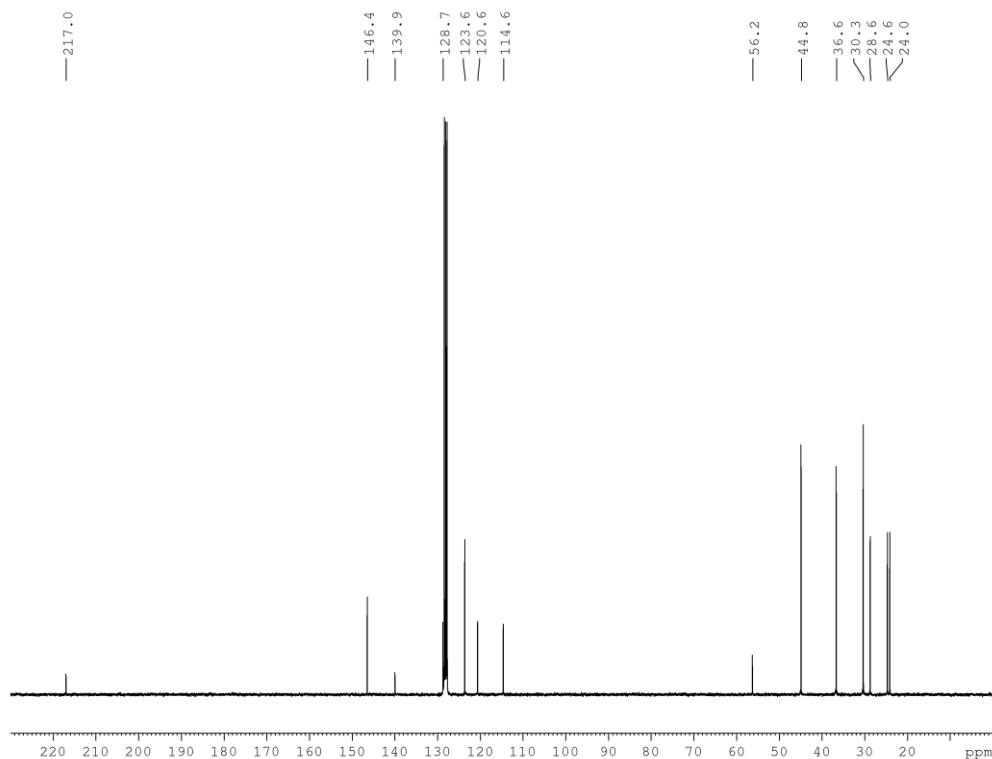


## 1.2 1-(Adamantyl)-3-(2,6-diisopropylphenyl)-imidazolin-2-ylidene (**1b**)

<sup>1</sup>H NMR (300 MHz, C<sub>6</sub>D<sub>6</sub>) of **1b**:

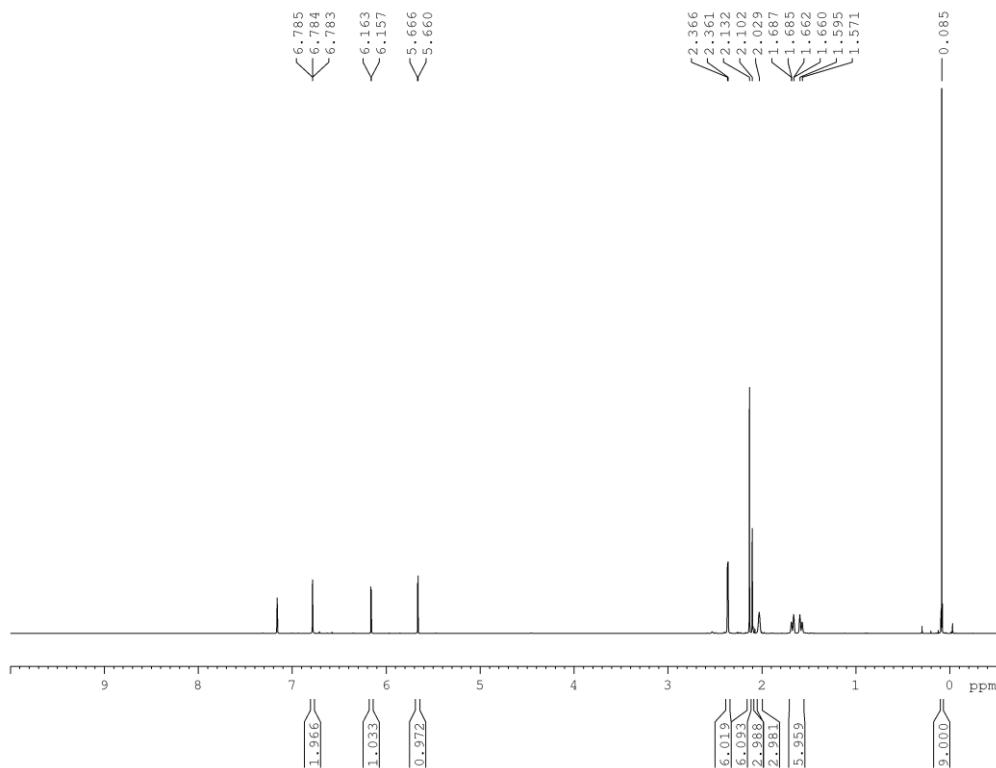


<sup>13</sup>C NMR (75 MHz, C<sub>6</sub>D<sub>6</sub>) of **1b**:

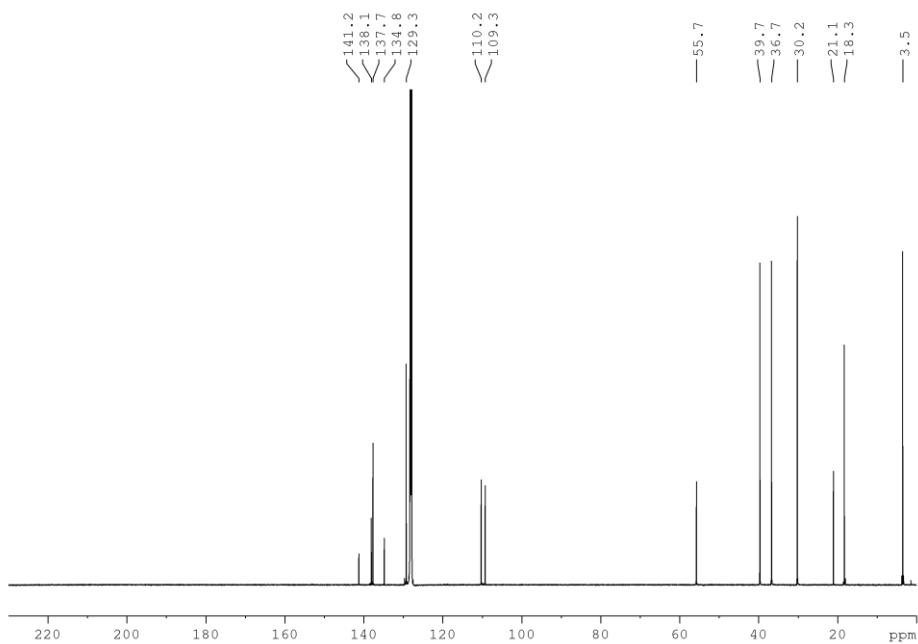


### 1.3 [Im<sup>AdMes</sup>NSiMe<sub>3</sub>] (2a)

<sup>1</sup>H NMR (500 MHz, C<sub>6</sub>D<sub>6</sub>) of 2a:

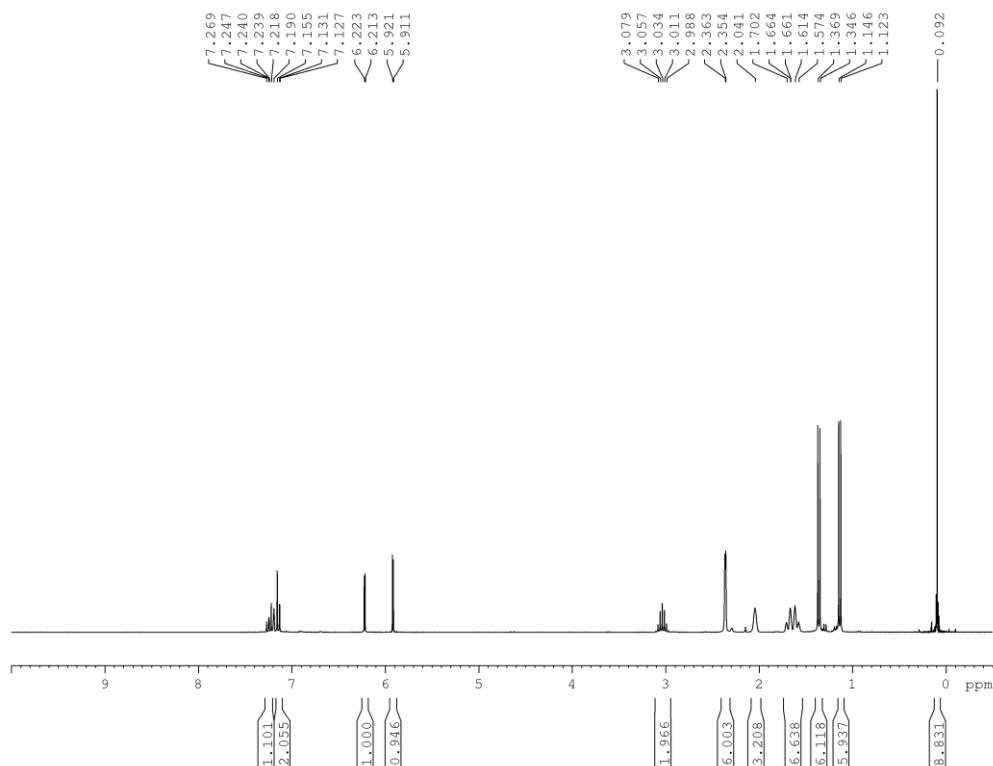


<sup>13</sup>C NMR (126 MHz, C<sub>6</sub>D<sub>6</sub>) of 2a:

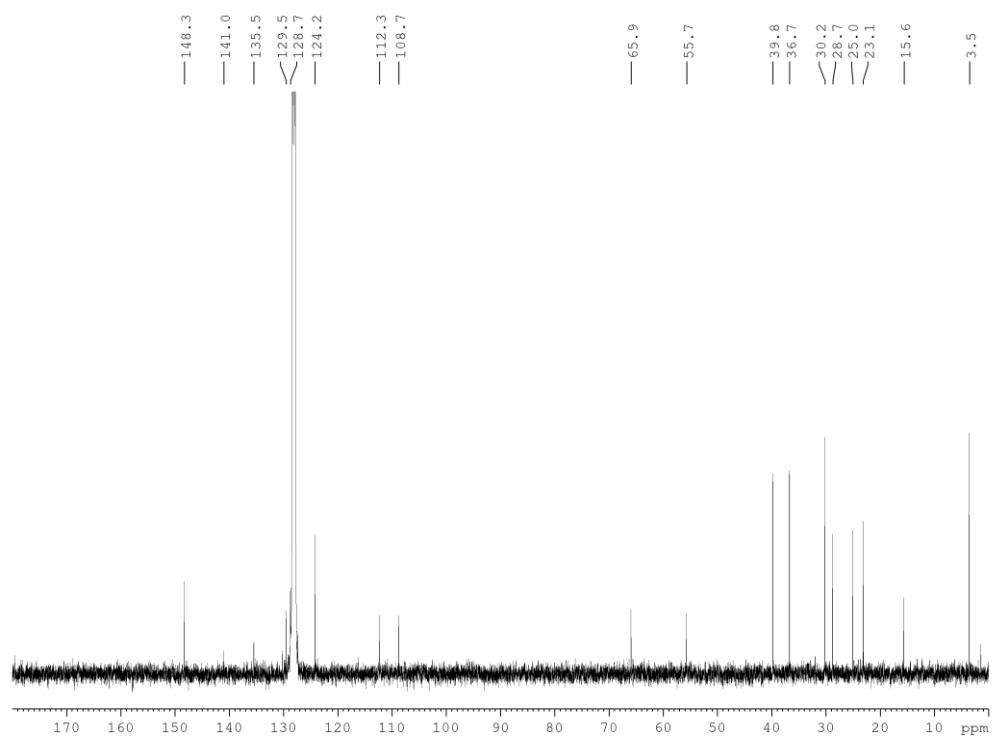


### 1.4 [Im<sup>AdDipp</sup>NSiMe<sub>3</sub>] (2b)

<sup>1</sup>H NMR (400 MHz, C<sub>6</sub>D<sub>6</sub>) of 2b:

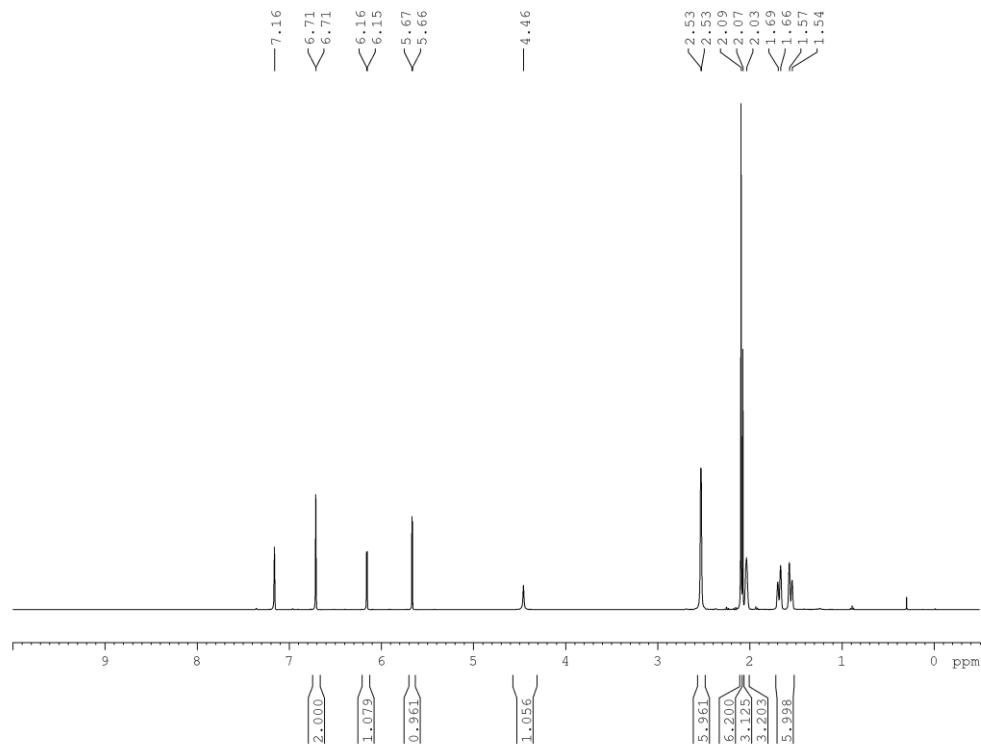


<sup>13</sup>C NMR (100 MHz, C<sub>6</sub>D<sub>6</sub>) of 2b:

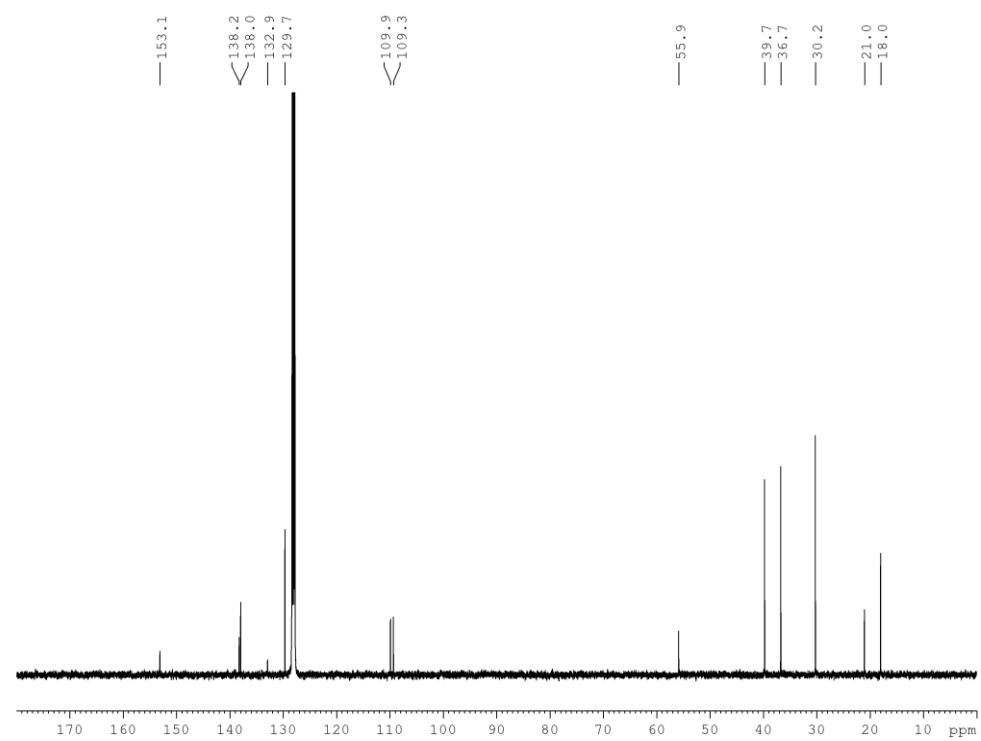


### 1.5 [Im<sup>AdMes</sup>NH] (**3a**)

<sup>1</sup>H NMR (400 MHz, C<sub>6</sub>D<sub>6</sub>) of **3a**:

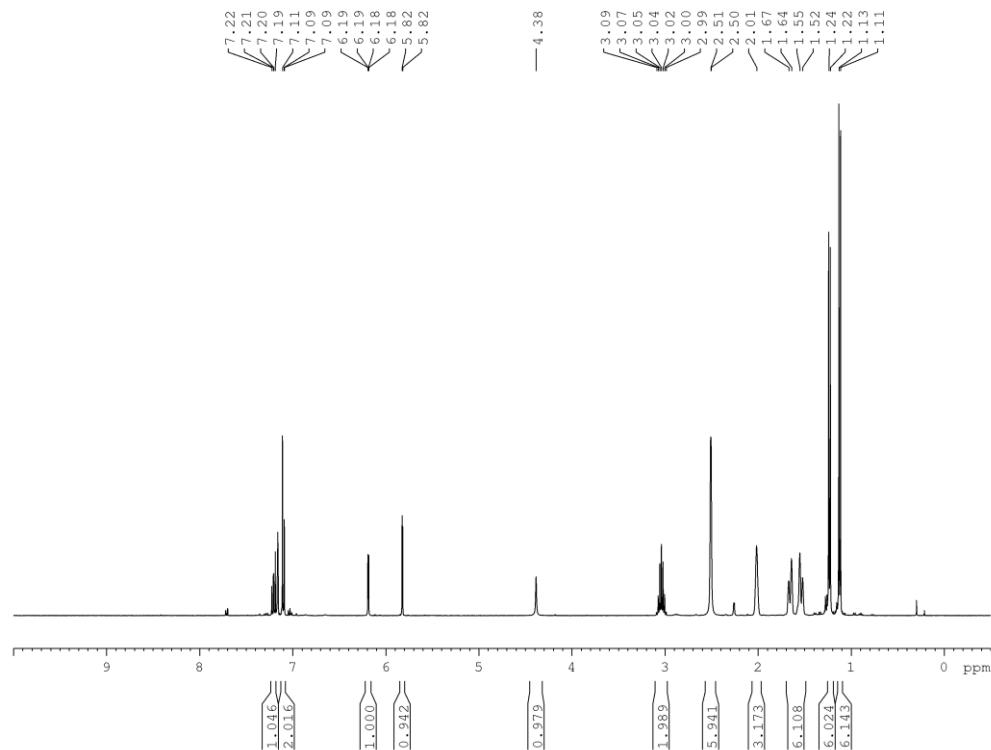


<sup>13</sup>C NMR (100 MHz, C<sub>6</sub>D<sub>6</sub>) of **3a**:

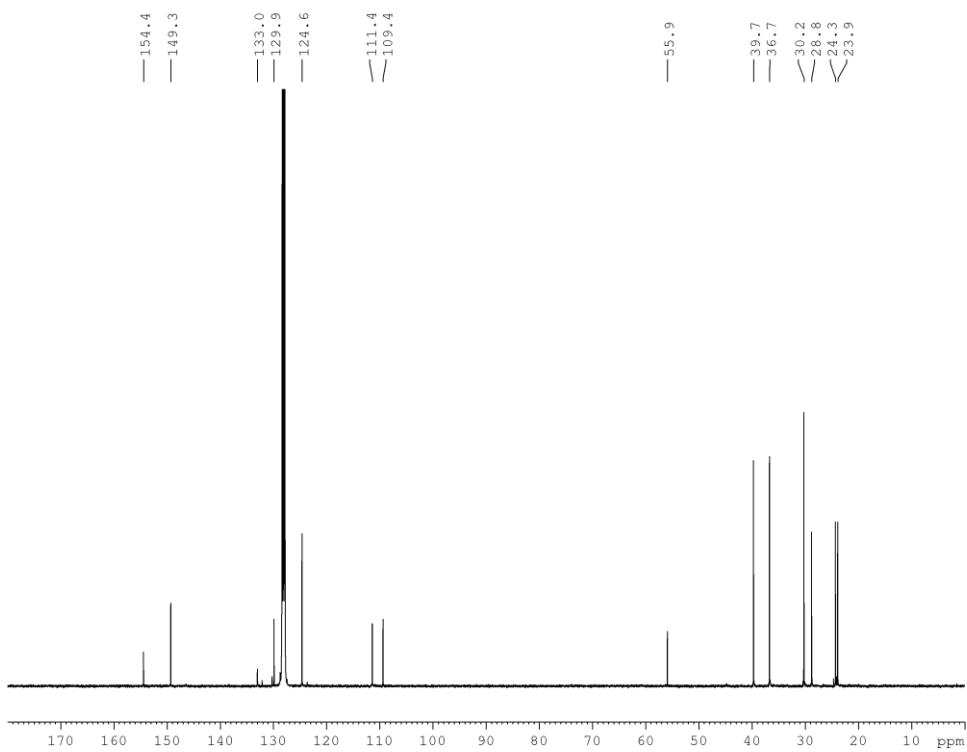


## 1.6 [Im<sup>AdDipp</sup>NH] (3b)

<sup>1</sup>H NMR (400 MHz, C<sub>6</sub>D<sub>6</sub>) of 3b:

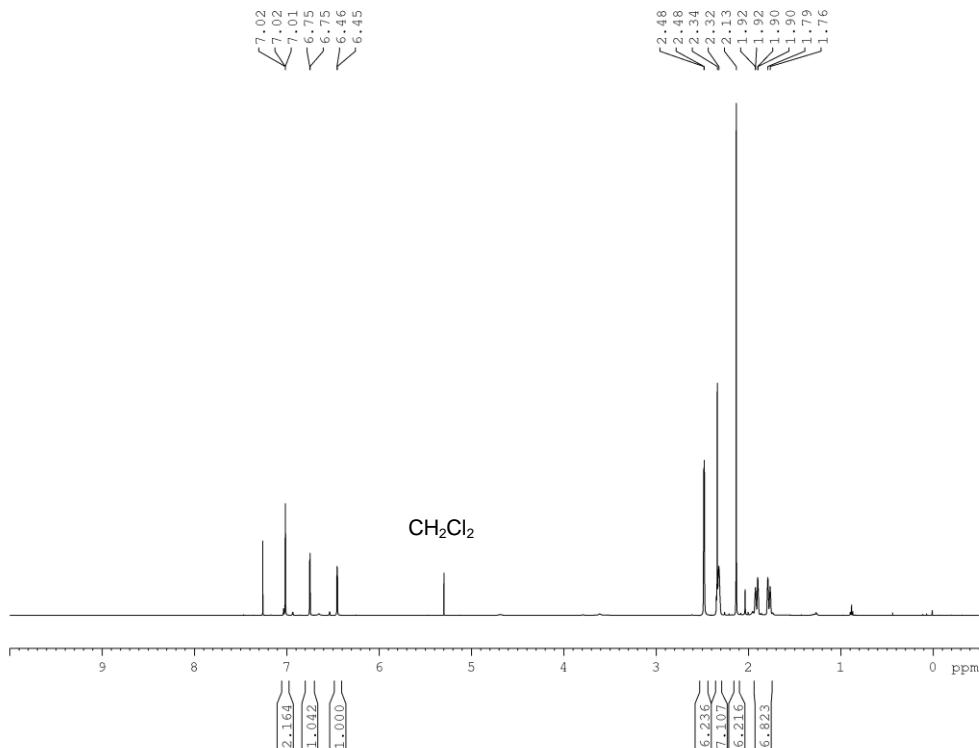


<sup>13</sup>C NMR (100 MHz, C<sub>6</sub>D<sub>6</sub>) of 3b:

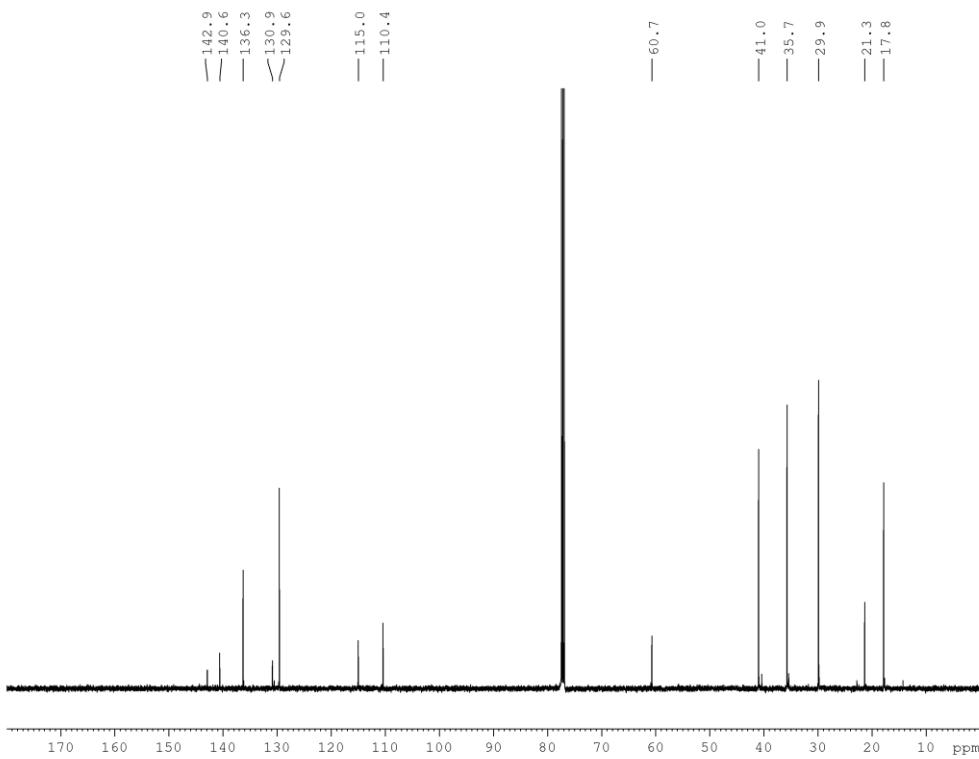


### 1.7 [(Im<sup>AdMes</sup>N)TiCl<sub>3</sub>] (4a)

<sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) of 4a:

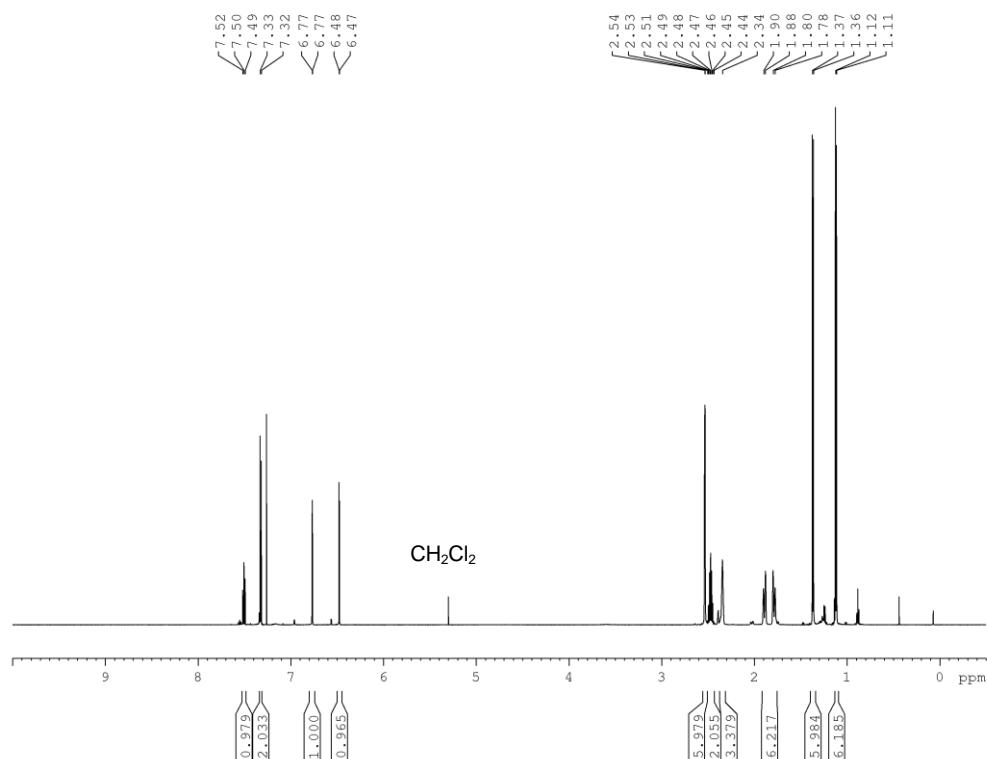


<sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>) of 4a:

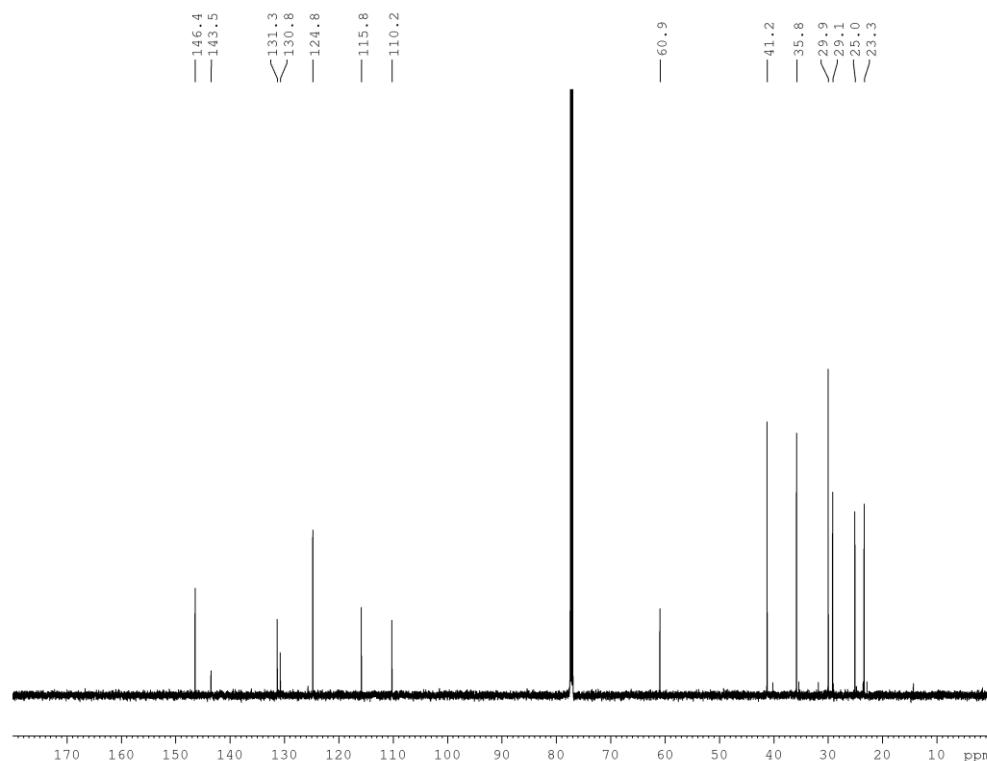


**1.8 [(Im<sup>AdDipp</sup>N)TiCl<sub>3</sub>] (4b)**

<sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>) of **4b**:

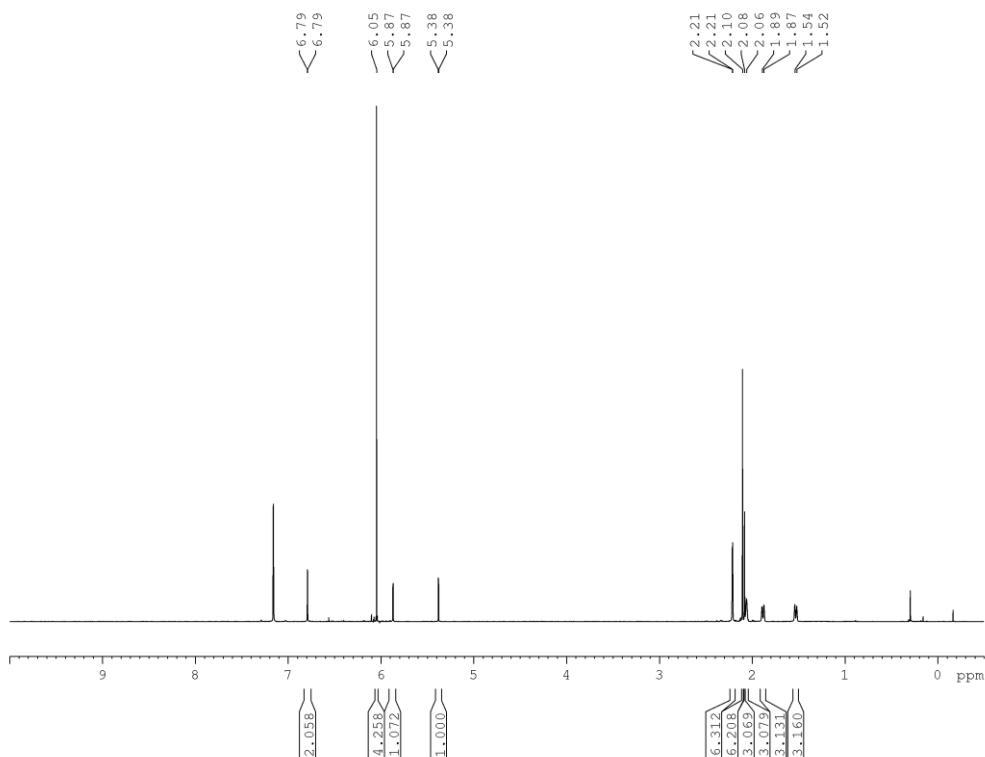


<sup>13</sup>C NMR (151 MHz, CDCl<sub>3</sub>) of **4b**:

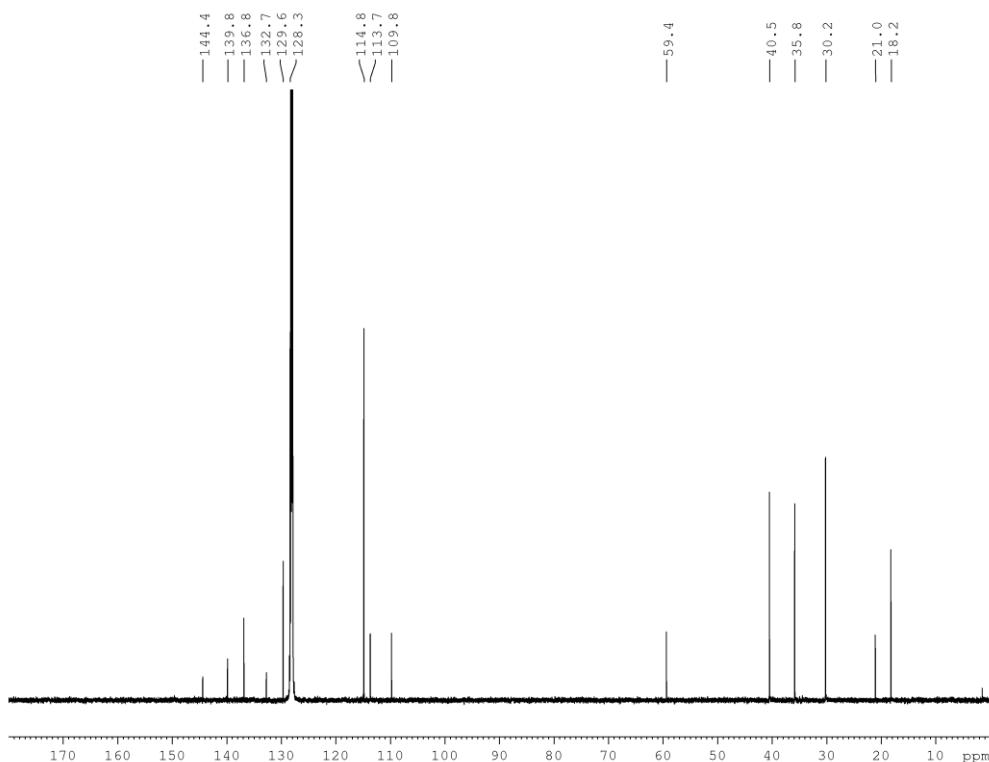


**1.9 [Cp(Im<sup>AdMes</sup>N)TiCl<sub>2</sub>] (5a)**

<sup>1</sup>H NMR (600 MHz, C<sub>6</sub>D<sub>6</sub>) of **5a**:

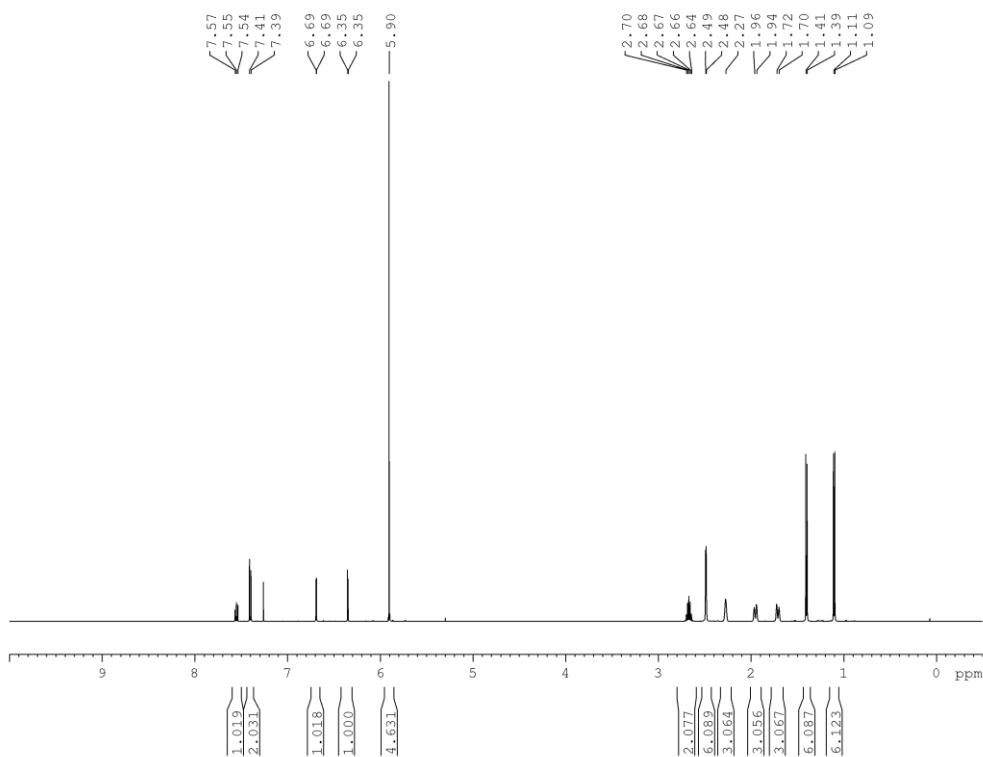


<sup>13</sup>C NMR (151 MHz, C<sub>6</sub>D<sub>6</sub>) of **5a**:

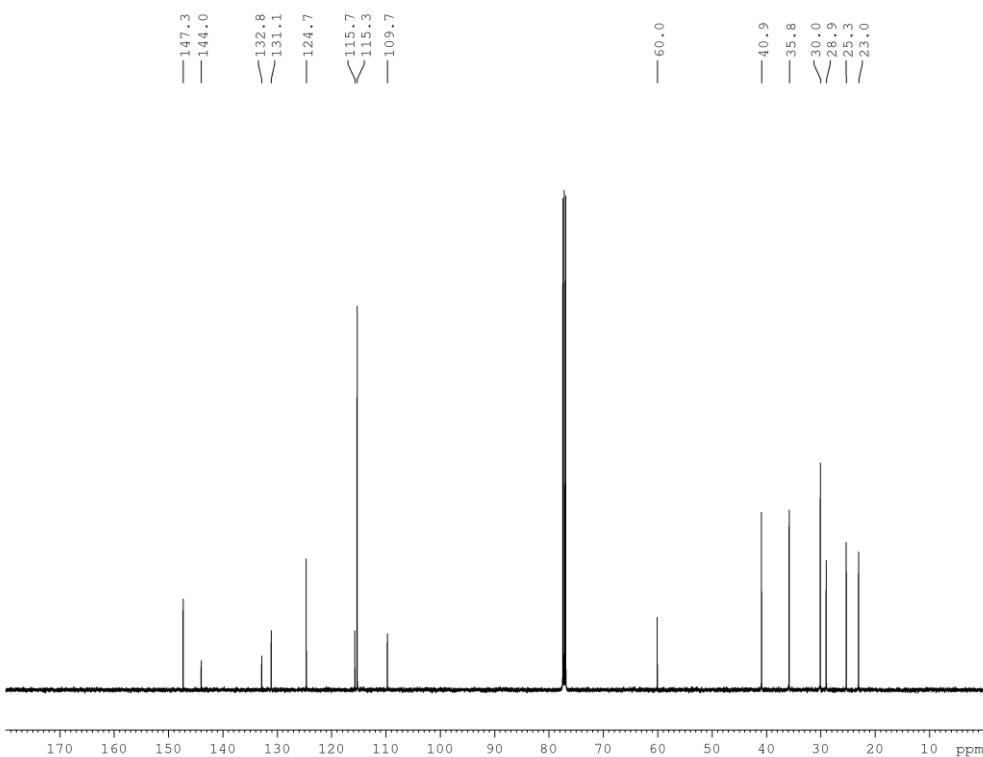


### 1.10 [ $\text{Cp}(\text{Im}^{\text{AdDippN}})\text{TiCl}_2$ ] (**5b**)

$^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ ) of **5b**:

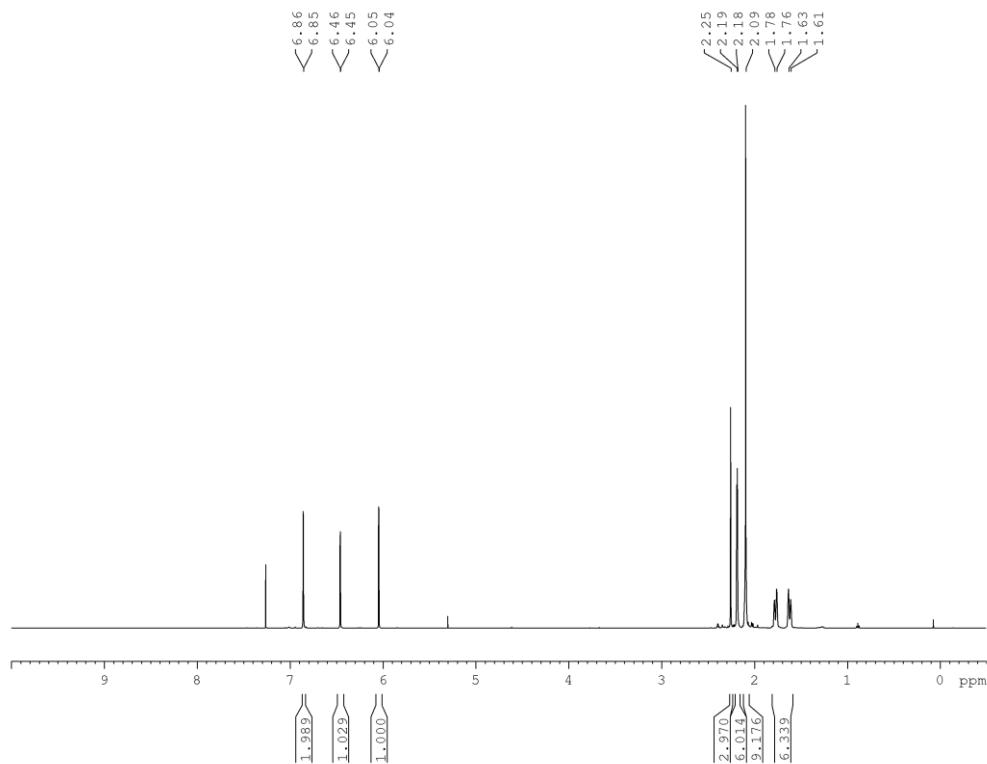


$^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ ) of **5b**:

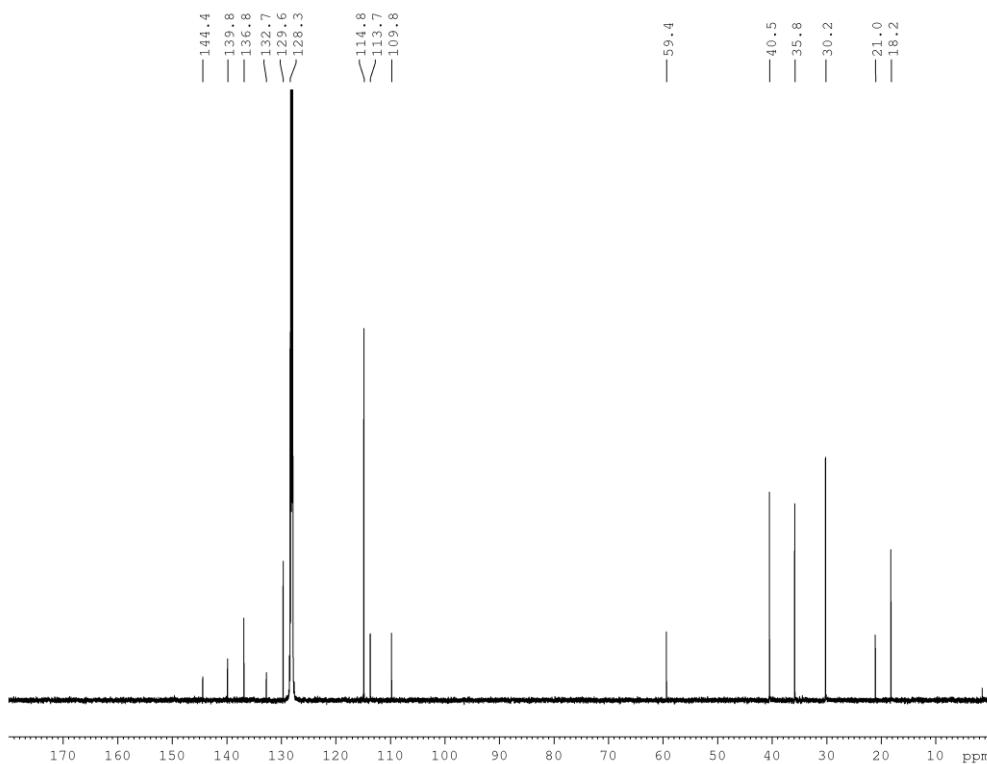


### 1.11 [(Im<sup>AdMes</sup>N)<sub>2</sub>TiCl<sub>2</sub>] (6a)

<sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) of 6a:

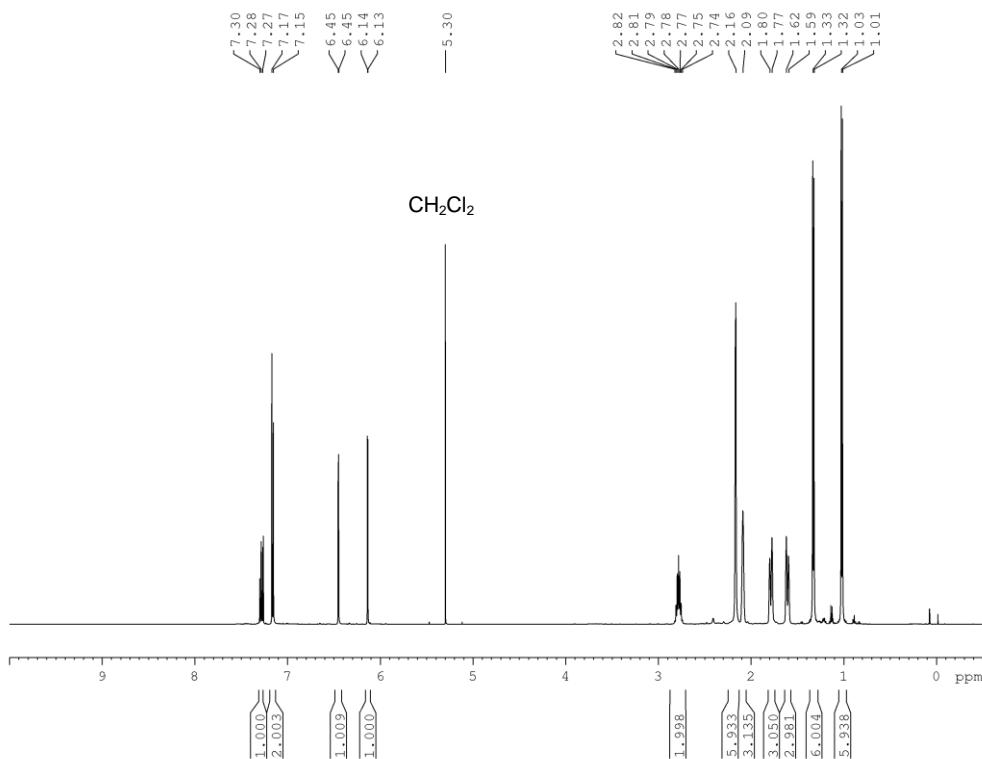


<sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>) of 6a:

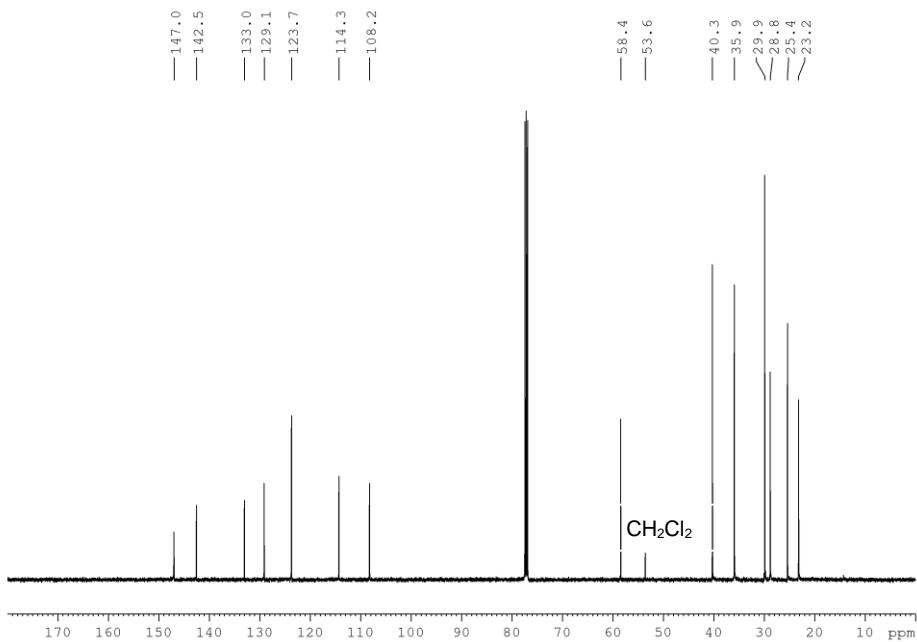


### 1.12 [(Im<sup>AdDipp</sup>N)<sub>2</sub>TiCl<sub>2</sub>] (**6b**)

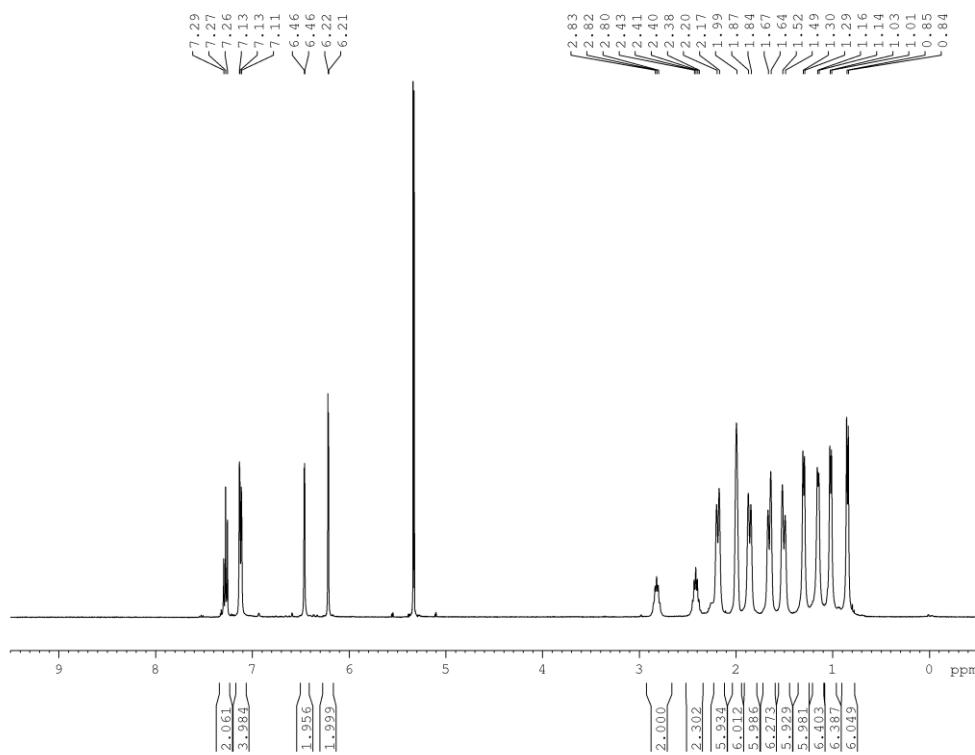
<sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) of **6b**:



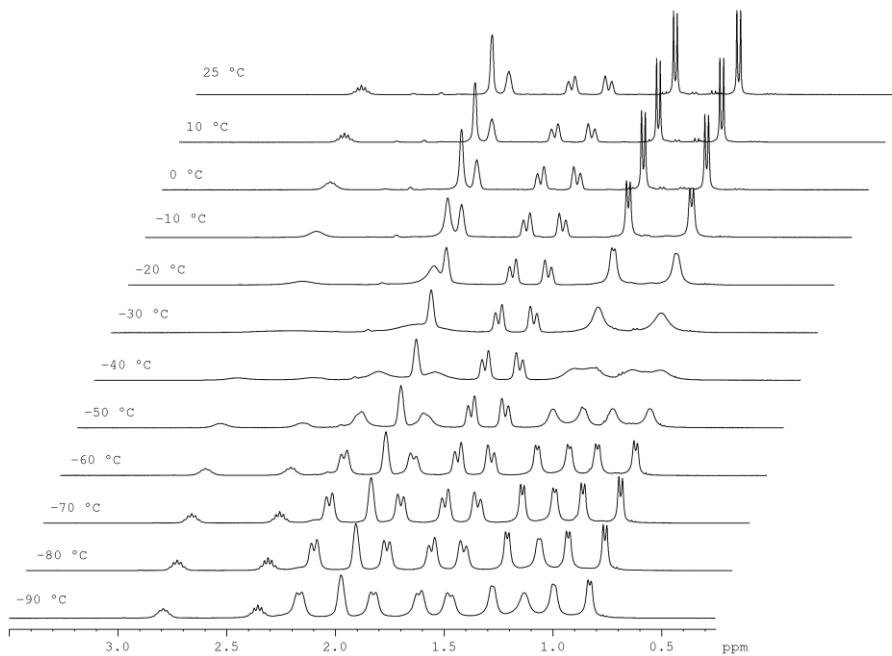
<sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>) of **6b**:



<sup>1</sup>H NMR (400 MHz, CD<sub>2</sub>Cl<sub>2</sub>) of **6b** at -70 °C:



<sup>1</sup>H NMR (400 MHz, CD<sub>2</sub>Cl<sub>2</sub>) of **6b** at variable temperatures; zoom-in to the aliphatic region:



## 2. X-Ray Crystal Structure Determinations

For a summary of crystal data, see Tables S4.1 - 4.6. Crystals were mounted on either MiTiGen or Hampton mounts in perfluorinated inert oil. Intensity measurements were performed at 100 K using a Rigaku XtaLAB Synergy S Single Source with mirror-focussed CuK $\alpha$  radiation or a Rigaku XtaLAB Synergy S Single Source with mirror-focused MoK $\alpha$  radiation. Additional measurements were performed using an Oxford Diffraction Xcalibur Eos diffractometer with MoK $\alpha$  radiation. Data reduction was performed with the CrysAlisPRO software.[A] Absorption corrections were based on multi-scans, analytical methods or face-indexation using a gaussian grid. The structures solved using either direct methods in SHELXS[B] or intrinsic phasing in SHELXT[C] and were refined anisotropically on F<sub>2</sub> using the program SHELXL[D] in OLEX2.[E] The hydrogen atoms were, unless otherwise noted, included either as constituents of idealized rigid methyl groups allowed to rotate but not tip, or using a riding model starting from calculated positions. The implementation of BYPASS[F] in OLEX2 had to be used for compound **4b**. One figure was created using the program Mercury[G].

CCDC 2174864-2174874 contain the supplementary crystallographic data for this paper. These data are provided free of charge by the Cambridge Crystallographic Data Centre.

### References:

- [A] Rigaku Oxford Diffraction, CrysAlisPRO Software System, Rigaku Corporation, Oxford, UK.
- [B] G. M. Sheldrick, *Acta Cryst.* **2007**, A64, 112-122.
- [C] G. M. Sheldrick, *Acta Cryst.* **2015**, A71, 3–8.
- [D] G. M. Sheldrick, *Acta Cryst.* **2015**, C71, 3–8.
- [E] O. V. Dolomanov, L. J. Bourhis, R. J. Gildea, J. A. K. Howard, H. Puschmann, *J. Appl. Cryst.* **2009**, 42, 339–341.
- [F] P. van der Sluis, A. L. Spek, *Acta Cryst.* **1990**, A46, 194-201.
- [G] C. F. Macrae, I. Sovago, S. J. Cottrell, P. T. A. Galek, P. McCabe, E. Pidcock, M. Platings, G. P. Shields, J. S. Stevens, M. Towler et al., *J. Appl. Crystallogr.* **2020**, 53, 226.

## 2.1 1-(Adamantyl)-3-(2,4,6-trimethylphenyl)-imidazolin-2-ylidene (**1a**)

Identification code	mk31mk
CCDC Number:	2174864
Empirical formula	C <sub>22</sub> H <sub>28</sub> N <sub>2</sub>
Formula weight	320.46
Temperature	100(2) K
Wavelength	0.71073 Å
Instrument (scan mode)	XtaLAB Synergy, Single source HyPix (ω scan)
Crystal system	Monoclinic
Space group	P2 <sub>1</sub> /c
Unit cell dimensions	a = 13.5173(4) Å      α= 90° b = 10.3836(2) Å      β= 95.869(2)° c = 12.8450(4) Å      γ = 90°
Volume	1793.45(8) Å <sup>3</sup>
Z	4
Density (calculated)	1.187 Mg/m <sup>3</sup>
Absorption coefficient	0.069 mm <sup>-1</sup>
F(000)	696
Crystal habitus	plate (colourless)
Crystal size	0.279 x 0.211 x 0.143 mm <sup>3</sup>
Theta range for data collection	2.478 to 44.855°
Index ranges	-26<=h<=26, -20<=k<=20, -25<=l<=25
Reflections collected	134137
Independent reflections	14658 [R(int) = 0.0485]
Completeness to theta = 25.242°	99.9 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	1.00000 and 0.88245
Refinement method	Full-matrix least-squares on F <sup>2</sup>
Data / restraints / parameters	14658 / 0 / 220
Goodness-of-fit on F <sup>2</sup>	1.066
Final R indices [I>2sigma(I)]	R1 = 0.0441, wR2 = 0.1296
R indices (all data)	R1 = 0.0624, wR2 = 0.1381
Largest diff. peak and hole	0.606 and -0.314 e.Å <sup>-3</sup>
Crystallisation Details:	Toluol/n-Hexan -40°C
Solution:	SHELXT-2014/5 (G. M. Sheldrick, Acta Cryst., 2015, A71, 3-8)
Refinement:	SHELXL-2018/3 (G. M. Sheldrick, Acta Cryst. (2008), A64, 112-122)
Interface:	OLEX2 v1.2 (O. V. Dolomanov et al., J. Appl. Cryst., 2009, 42, 339-341)
Measurement and Refinement Details:	-

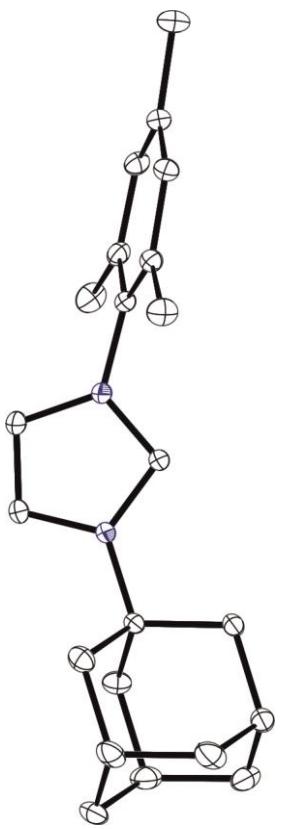


Figure S1: Molecular structure of **1a** with thermal displacement parameters drawn at 50% probability. All hydrogens are omitted for clarity. Selected bond lengths [Å] and angles [°]: C1-N1 1.3739(5), C1-N2 1.3651(5), N1-C4 1.4313(5), N2-C13 1.4711(6), N1-C1-N2 102.05(3).

## 2.2 1-(Adamantyl)-3-(2,6-diisopropylphenyl)-imidazolin-2-ylidene (1b)

Identification code	mk28mk
CCDC Number:	2174865
Empirical formula	C <sub>25</sub> H <sub>34</sub> N <sub>2</sub>
Formula weight	362.54
Temperature	100(2) K
Wavelength	1.54184 Å
Instrument (scan mode)	XtaLAB Synergy, Single source, HyPix ( $\omega$ scan)
Crystal system	Orthorhombic
Space group	Pnma
Unit cell dimensions	a = 10.4134(2) Å $\alpha$ = 90° b = 12.3697(2) Å $\beta$ = 90° c = 16.5809(2) Å $\gamma$ = 90°
Volume	2135.80(6) Å <sup>3</sup>
Z	4
Density (calculated)	1.127 Mg/m <sup>3</sup>
Absorption coefficient	0.490 mm <sup>-1</sup>
F(000)	792
Crystal habitus	plate (colourless)
Crystal size	0.347 x 0.100 x 0.089 mm <sup>3</sup>
Theta range for data collection	4.460 to 77.393°
Index ranges	-11≤h≤13, -15≤k≤15, -20≤l≤20
Reflections collected	39788
Independent reflections	2337 [R(int) = 0.0431]
Completeness to theta = 67.684°	100.0 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	1.00000 and 0.47134
Refinement method	Full-matrix least-squares on F <sup>2</sup>
Data / restraints / parameters	2337 / 0 / 141
Goodness-of-fit on F <sup>2</sup>	1.044
Final R indices [I>2sigma(I)]	R1 = 0.0398, wR2 = 0.0995
R indices (all data)	R1 = 0.0417, wR2 = 0.1013
Largest diff. peak and hole	0.216 and -0.194 e.Å <sup>-3</sup>
Crystallisation Details:	toluene/n-hexane -30 °C
Solution:	SHELXT-2014/5 (G. M. Sheldrick, Acta Cryst., 2015, A71, 3-8)
Refinement:	SHELXL-2018/3 (G. M. Sheldrick, Acta Cryst., 2008, A64, 112-122)
Interface:	OLEX2 v1.2 (O. V. Dolomanov et al., J. Appl. Cryst., 2009, 42, 339-341)
Measurement and Refinement Details:	-

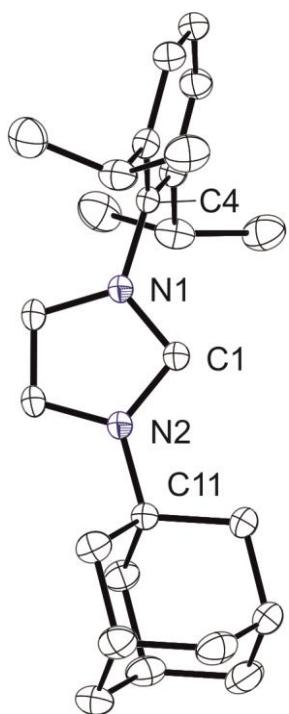


Figure S2: Molecular structure of **1b** with thermal displacement parameters drawn at 50% probability. All hydrogens are omitted for clarity. Selected bond lengths [ $\text{\AA}$ ] and angles [ $^\circ$ ]: C1-N1 1.3709(18), C1-N2 1.3627(18), N1-C4 1.4352(17), N2-C11 1.4765(17), N1-C1-N2 101.48(11).

### 2.3 [ $\text{Im}^{\text{AdMes}}\text{NSiMe}_3$ ] (2a)

Identification code	mk17mk
CCDC Number:	2174866
Empirical formula	$\text{C}_{25}\text{H}_{37}\text{N}_3\text{Si}$
Formula weight	407.66
Temperature	103(1) K
Wavelength	0.71073 Å
Instrument (scan mode)	Oxford Diffraction Xcalibur, Eos ( $\omega$ scan)
Crystal system	Monoclinic
Space group	$P2_1/n$
Unit cell dimensions	$a = 9.7011(4)$ Å $\alpha = 90^\circ$ $b = 24.1407(8)$ Å $\beta = 99.776(4)^\circ$ $c = 10.2311(4)$ Å $\gamma = 90^\circ$
Volume	2361.24(16) Å <sup>3</sup>
Z	4
Density (calculated)	1.147 Mg/m <sup>3</sup>
Absorption coefficient	0.115 mm <sup>-1</sup>
F(000)	888
Crystal habitus	irregular (colourless)
Crystal size	0.701 x 0.591 x 0.469 mm <sup>3</sup>
Theta range for data collection	2.291 to 36.316°
Index ranges	-16≤h≤16, -40≤k≤40, -17≤l≤17
Reflections collected	121520
Independent reflections	11433 [R(int) = 0.0726]
Completeness to theta = 25.242°	100.0 %
Absorption correction	Gaussian
Max. and min. transmission	0.961 and 0.942
Refinement method	Full-matrix least-squares on F <sup>2</sup>
Data / restraints / parameters	11433 / 0 / 268
Goodness-of-fit on F <sup>2</sup>	1.040
Final R indices [I>2sigma(I)]	R1 = 0.0457, wR2 = 0.1173
R indices (all data)	R1 = 0.0601, wR2 = 0.1258
Largest diff. peak and hole	0.671 and -0.236 e.Å <sup>-3</sup>
Crystallisation Details:	<i>n</i> -hexane at -27 °C
Solution:	SHELXT-2014/5 (G. M. Sheldrick, Acta Cryst., 2015, A71, 3-8)
Refinement:	SHELXL-2018/3 (G. M. Sheldrick, Acta Cryst. (2008), A64, 112-122)
Interface:	OLEX2 v1.2 (O. V. Dolomanov et al., J. Appl. Cryst., 2009, 42, 339-341)
Measurement and Refinement Details:	-

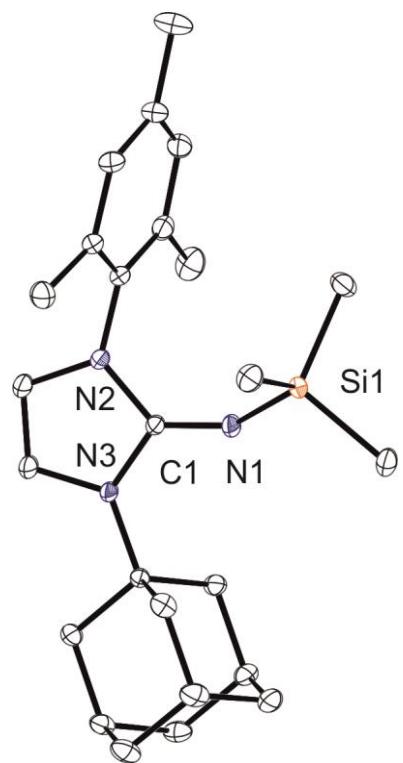


Figure S3: Molecular structure of **2a** with thermal displacement parameters drawn at 50% probability. All hydrogens are omitted for clarity. Selected bond lengths [ $\text{\AA}$ ] and angles [ $^\circ$ ]: C1-N1 1.2782(10), N1-Si1 1.6823(7), N2-C1-N3 103.99(6), C1-N1-Si1 1142.68(7).

## 2.4 [Im<sup>AdMes</sup>NH] (3a)

Identification code	mk18mk
CCDC Number:	2174867
Empirical formula	C <sub>22</sub> H <sub>29</sub> N <sub>3</sub>
Formula weight	335.48
Temperature	100(2) K
Wavelength	1.54184 Å
Instrument (scan mode)	XtaLAB Synergy, Single source, HyPix ( $\omega$ scan)
Crystal system	Orthorhombic
Space group	Pbca
Unit cell dimensions	a = 16.6873(2) Å $\alpha$ = 90° b = 11.37240(10) Å $\beta$ = 90° c = 19.4505(2) Å $\gamma$ = 90°
Volume	3691.21(7) Å <sup>3</sup>
Z	8
Density (calculated)	1.207 Mg/m <sup>3</sup>
Absorption coefficient	0.545 mm <sup>-1</sup>
F(000)	1456
Crystal habitus	fragment of trapezoid (colourless)
Crystal size	0.108 x 0.101 x 0.099 mm <sup>3</sup>
Theta range for data collection	4.547 to 77.613°
Index ranges	-21 <= h <= 21, -14 <= k <= 11, -24 <= l <= 24
Reflections collected	41369
Independent reflections	3865 [R(int) = 0.0255]
Completeness to theta = 67.684°	100.0 %
Absorption correction	Gaussian
Max. and min. transmission	1.000 and 0.793
Refinement method	Full-matrix least-squares on F <sup>2</sup>
Data / restraints / parameters	3865 / 0 / 233
Goodness-of-fit on F <sup>2</sup>	1.043
Final R indices [I>2sigma(I)]	R1 = 0.0408, wR2 = 0.1015
R indices (all data)	R1 = 0.0429, wR2 = 0.1038
Largest diff. peak and hole	0.280 and -0.283 e.Å <sup>-3</sup>
Crystallisation Details:	cooling of hot n-hexane solution to room temperature
Solution:	SHELXT-2014/5 (G. M. Sheldrick, Acta Cryst., 2015, A71, 3-8)
Refinement:	SHELXL-2018/3 (G. M. Sheldrick, Acta Cryst., 2008, A64, 112-122)
Interface:	OLEX2 v1.2 (O. V. Dolomanov et al., J. Appl. Cryst., 2009, 42, 339-341)
Measurement and Refinement Details:	N-H hydrogen was refined freely.

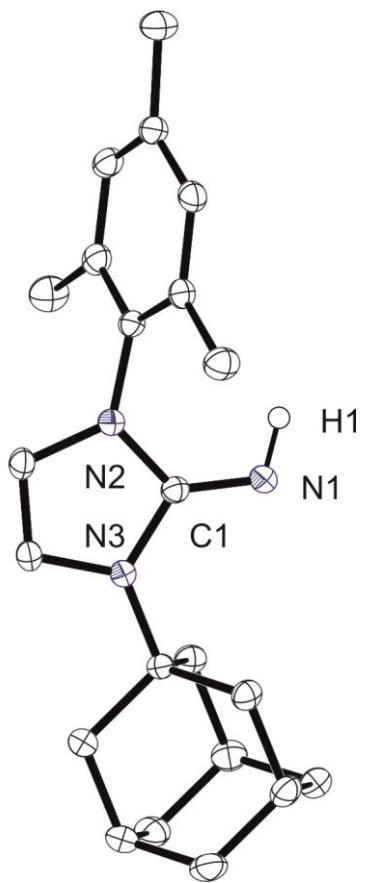


Figure S4: Molecular structure of **3a** with thermal displacement parameters drawn at 50% probability. All hydrogens are omitted for clarity. Selected bond lengths [ $\text{\AA}$ ] and angles [ $^\circ$ ]: C1-N1 1.2966(14), N2-C1-N3 104.55(9).

## 2.5 [Im<sup>AdDipp</sup>NH] (3b)

Identification code	mk15mk
CCDC Number:	2174868
Empirical formula	C <sub>25</sub> H <sub>35</sub> N <sub>3</sub>
Formula weight	377.56
Temperature	100(2) K
Wavelength	1.54184 Å
Instrument (scan mode)	XtaLAB Synergy, Single source, HyPix ( $\omega$ scan)
Crystal system	Monoclinic
Space group	P2 <sub>1</sub> /c
Unit cell dimensions	a = 13.8605(1) Å $\alpha$ = 90° b = 6.5130(1) Å $\beta$ = 98.599(1)° c = 24.0076(2) Å $\gamma$ = 90°
Volume	2142.89(4) Å <sup>3</sup>
Z	4
Density (calculated)	1.170 Mg/m <sup>3</sup>
Absorption coefficient	0.521 mm <sup>-1</sup>
F(000)	824
Crystal habitus	lath (colourless)
Crystal size	0.352 x 0.067 x 0.039 mm <sup>3</sup>
Theta range for data collection	3.225 to 77.449°
Index ranges	-17 <= h <= 17, -6 <= k <= 8, -30 <= l <= 30
Reflections collected	47042
Independent reflections	4470 [R(int) = 0.0319]
Completeness to theta = 67.684°	100.0 %
Absorption correction	Gaussian
Max. and min. transmission	1.000 and 0.784
Refinement method	Full-matrix least-squares on F <sup>2</sup>
Data / restraints / parameters	4470 / 0 / 261
Goodness-of-fit on F <sup>2</sup>	1.038
Final R indices [I>2sigma(I)]	R1 = 0.0378, wR2 = 0.0967
R indices (all data)	R1 = 0.0401, wR2 = 0.0998
Largest diff. peak and hole	0.265 and -0.212 e.Å <sup>-3</sup>
Crystallisation Details:	n-hexane -40 °C
Solution:	SHELXT-2014/5 (G. M. Sheldrick, Acta Cryst., 2015, A71, 3-8)
Refinement:	SHELXL-2018/3 (G. M. Sheldrick, Acta Cryst., 2008, A64, 112-122)
Interface:	OLEX2 v1.2 (O. V. Dolomanov et al., J. Appl. Cryst., 2009, 42, 339-341)
Measurement and Refinement Details:	N-H hydrogen was refined freely.

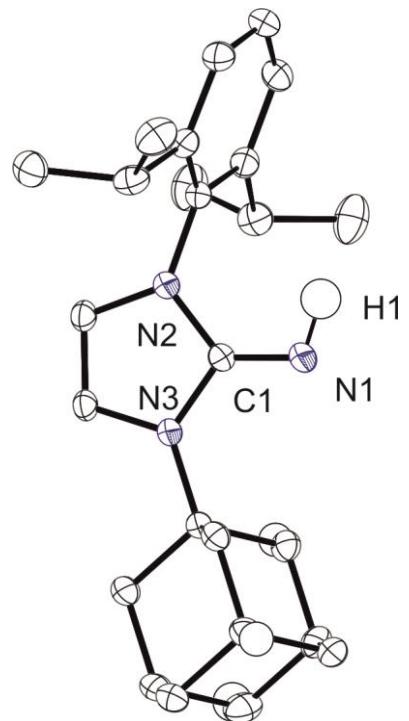


Figure S5: Molecular structure of **3b** with thermal displacement parameters drawn at 50% probability. All hydrogens are omitted for clarity. Selected bond lengths [ $\text{\AA}$ ] and angles [°]: C1-N1 1.2891(13), N2-C1-N3 104.78(8).

## 2.6 [(Im<sup>AdMes</sup>NH)TiCl<sub>3</sub>] (4a)

Identification code	mk23mk
CCDC Number:	2174869
Empirical formula	C <sub>22</sub> H <sub>28</sub> Cl <sub>3</sub> N <sub>3</sub> Ti
Formula weight	488.72
Temperature	100(2) K
Wavelength	0.71073 Å
Instrument (scan mode)	XtaLAB Synergy, Single source, HyPix ( $\omega$ scan)
Crystal system	Monoclinic
Space group	P2 <sub>1</sub> /n
Unit cell dimensions	a = 9.6555(2) Å $\alpha$ = 90° b = 23.9142(3) Å $\beta$ = 98.744(2)° c = 10.2854(2) Å $\gamma$ = 90°
Volume	2347.33(7) Å <sup>3</sup>
Z	4
Density (calculated)	1.383 Mg/m <sup>3</sup>
Absorption coefficient	0.720 mm <sup>-1</sup>
F(000)	1016
Crystal habitus	irregular (orange)
Crystal size	0.370 x 0.269 x 0.100 mm <sup>3</sup>
Theta range for data collection	2.731 to 41.154°
Index ranges	-17 <= h <= 17, -44 <= k <= 44, -19 <= l <= 19
Reflections collected	209853
Independent reflections	15591 [R(int) = 0.0354]
Completeness to theta = 25.242°	99.8 %
Absorption correction	Gaussian
Max. and min. transmission	1.000 and 0.272
Refinement method	Full-matrix least-squares on F <sup>2</sup>
Data / restraints / parameters	15591 / 0 / 265
Goodness-of-fit on F <sup>2</sup>	1.051
Final R indices [I>2sigma(I)]	R1 = 0.0268, wR2 = 0.0751
R indices (all data)	R1 = 0.0329, wR2 = 0.0774
Largest diff. peak and hole	0.626 and -0.215 e.Å <sup>-3</sup>
Crystallisation Details:	DCM/n-hexane
Solution:	SHELXT-2014/5 (G. M. Sheldrick, Acta Cryst., 2015, A71, 3-8)
Refinement:	SHELXL-2018/3 (G. M. Sheldrick, Acta Cryst. (2008), A64, 112-122)
Interface:	OLEX2 v1.2 (O. V. Dolomanov et al., J. Appl. Cryst., 2009, 42, 339-341)
Measurement and Refinement Details:	-

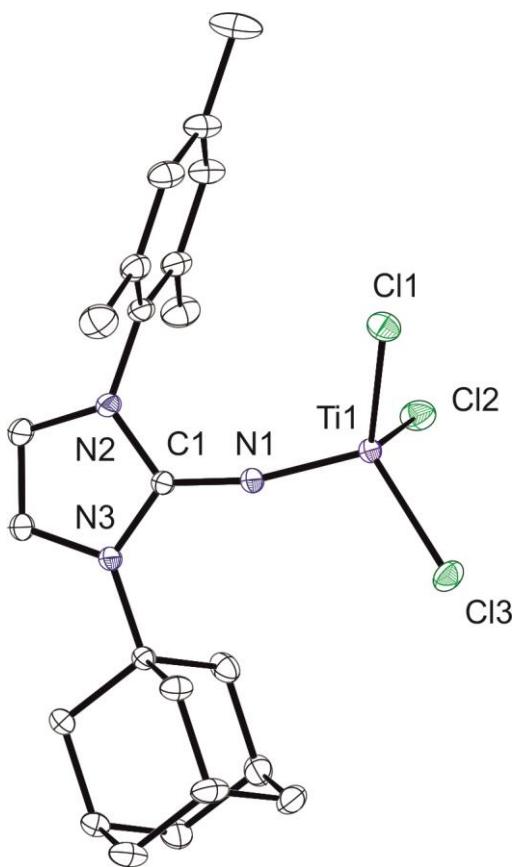


Figure S6: Molecular structure of **4a** with thermal displacement parameters drawn at 50% probability. All hydrogens are omitted for clarity. Selected bond lengths [ $\text{\AA}$ ] and angles [ $^\circ$ ]: C1-N1 1.3265(7), N1-Ti1 1.7347(5), N2-C1-N3 107.41(4), C1-N1-Ti1 162.58(5), Cl1-Ti1-Cl2 111.50(2), N1-C1-N2 124.07(5), N1-C1-N3 128.50(5).

## 2.7 [(Im<sup>AdDipp</sup>N)TiCl<sub>3</sub>] CH<sub>2</sub>Cl<sub>2</sub>·Solv (4b)

Identification code	mk26mk
CCDC Number:	2174870
Empirical formula	C <sub>26</sub> H <sub>36</sub> Cl <sub>5</sub> N <sub>3</sub> Ti
Formula weight	615.73
Temperature	100(2) K
Wavelength	0.71073 Å
Instrument (scan mode)	XtaLAB Synergy, Single source, HyPix ( $\omega$ scan)
Crystal system	Triclinic
Space group	P $\bar{1}$
Unit cell dimensions	a = 9.8202(4) Å $\alpha$ = 63.091(5) $^\circ$ b = 13.2868(6) Å $\beta$ = 81.445(4) $^\circ$ c = 14.4204(7) Å $\gamma$ = 76.358(4) $^\circ$
Volume	1628.59(15) Å <sup>3</sup>
Z	2
Density (calculated)	1.256 Mg/m <sup>3</sup>
Absorption coefficient	0.691 mm <sup>-1</sup>
F(000)	640
Crystal habitus	fragment of block (orange)
Crystal size	0.18 x 0.10 x 0.05 mm <sup>3</sup>
Theta range for data collection	2.137 to 32.385 $^\circ$
Index ranges	-14 $\leq$ h $\leq$ 14, -19 $\leq$ k $\leq$ 19, -21 $\leq$ l $\leq$ 20
Reflections collected	43682
Independent reflections	9833 [R(int) = 0.0215]
Completeness to theta = 25.242 $^\circ$	99.9 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	1.00000 and 0.88732
Refinement method	Full-matrix least-squares on F <sup>2</sup>
Data / restraints / parameters	9833 / 0 / 320
Goodness-of-fit on F <sup>2</sup>	1.030
Final R indices [I>2sigma(I)]	R1 = 0.0581, wR2 = 0.1560
R indices (all data)	R1 = 0.0667, wR2 = 0.1618
Largest diff. peak and hole	2.297 and -1.603 e.Å <sup>-3</sup>
Crystallisation Details:	dichloromethane/n-hexane -30 °C
Solution:	SHELXT-2014/5 (G. M. Sheldrick, Acta Cryst., 2015, A71, 3-8)
Refinement:	SHELXL-2018/3 (G. M. Sheldrick, Acta Cryst. (2008), A64, 112-122)
Interface:	OLEX2 v1.2 (O. V. Dolomanov et al., J. Appl. Cryst., 2009, 42, 339-341)

Measurement and Refinement Details: A solvent mask was calculated and 66 electrons were found in a volume of 231 Å<sup>3</sup> in 1 void per unit cell. This is consistent with the presence of 0.5[CH<sub>2</sub>Cl<sub>2</sub>], 0.25[C<sub>6</sub>H<sub>14</sub>] per Asymmetric Unit which account for 67 electrons per unit cell. A reason why the molecules could not be refined satisfactorily might be their disorder along a canal. Additionally modulation could be a reason, as ‘smeared’ reflexes were observed along the c\* axis.

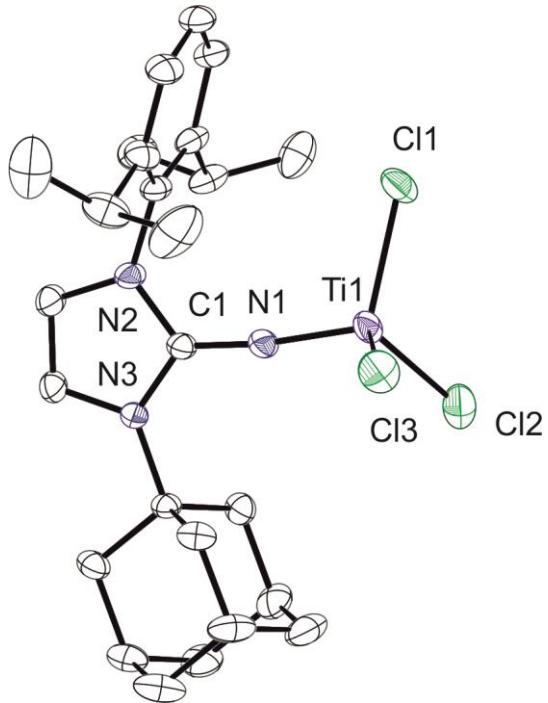


Figure S7: Molecular structure of **4b** $\cdot\text{CH}_2\text{Cl}_2\text{-Solv.}$  with thermal displacement parameters drawn at 50% probability. All hydrogens, one molecule of  $\text{CH}_2\text{Cl}_2$  and disordered solvent molecules are omitted for clarity. Selected bond lengths [Å] and angles [ $^\circ$ ]: C1-N1 1.330(3), N1-Ti 1.7316(17), N2-C1-N3 106.89(17), C1-N1-Ti1 173.04(16), Cl1-Ti1-Cl2 109.17(3), N1-C1-N2 124.33(17), 128.75(18).

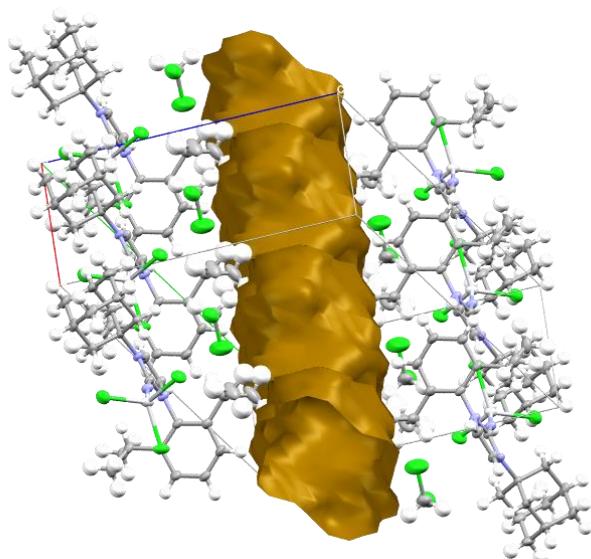


Figure S8: Depiction of the solvent accessible voids in the structure of **4b** in Mercury.

## 2.8 [Cp(Im<sup>AdMes</sup>N)TiCl<sub>2</sub>] (5a)

Identification code	mk22mk
CCDC Number:	2174871
Empirical formula	C <sub>27</sub> H <sub>33</sub> Cl <sub>2</sub> N <sub>3</sub> Ti
Formula weight	518.36
Temperature	100(2) K
Wavelength	1.54184 Å
Instrument (scan mode)	XtaLAB Synergy, Single source, HyPix ( $\omega$ scan)
Crystal system	Orthorhombic
Space group	P2 <sub>1</sub> 2 <sub>1</sub> 2 <sub>1</sub>
Unit cell dimensions	a = 9.97590(10) Å $\alpha$ = 90° b = 11.30190(10) Å $\beta$ = 90° c = 22.11580(10) Å $\gamma$ = 90°
Volume	2493.48(4) Å <sup>3</sup>
Z	4
Density (calculated)	1.381 Mg/m <sup>3</sup>
Absorption coefficient	5.029 mm <sup>-1</sup>
F(000)	1088
Crystal habitus	trapezoid (orange)
Crystal size	0.228 x 0.126 x 0.072 mm <sup>3</sup>
Theta range for data collection	3.998 to 77.473°
Index ranges	-12≤h≤10, -14≤k≤14, -27≤l≤28
Reflections collected	53498
Independent reflections	5264 [R(int) = 0.0323]
Completeness to theta = 67.684°	100.0 %
Absorption correction	Gaussian
Max. and min. transmission	1.000 and 0.670
Refinement method	Full-matrix least-squares on F <sup>2</sup>
Data / restraints / parameters	5264 / 0 / 302
Goodness-of-fit on F <sup>2</sup>	1.033
Final R indices [I>2sigma(I)]	R1 = 0.0218, wR2 = 0.0504
R indices (all data)	R1 = 0.0227, wR2 = 0.0513
Absolute structure parameter	0.434(5)
Largest diff. peak and hole	0.242 and -0.226 e.Å <sup>-3</sup>
Crystallisation Details:	toluene/n-hexane room temperature
Solution:	SHELXT-2014/5 (G. M. Sheldrick, Acta Cryst., 2015, A71, 3-8)
Refinement:	SHELXL-2018/3 (G. M. Sheldrick, Acta Cryst., 2008, A64, 112-122)
Interface:	OLEX2 v1.2 (O. V. Dolomanov et al., J. Appl. Cryst., 2009, 42, 339-341)
Measurement and Refinement Details:	Refined as a 2-component inversion twin.

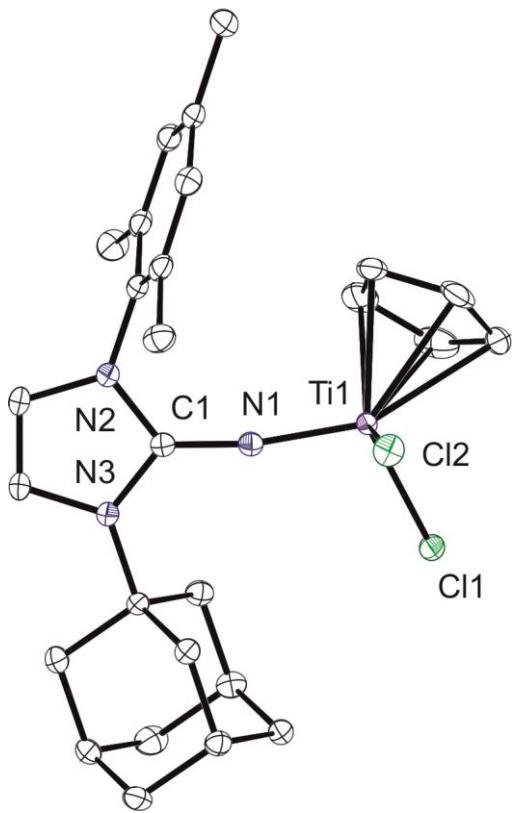


Figure S9: Molecular structure of **5a** with thermal displacement parameters drawn at 50% probability. All hydrogens are omitted for clarity. Selected bond lengths [ $\text{\AA}$ ] and angles [ $^\circ$ ]: C1-N1 1.319(2), N1-Ti1 1.7758(16), Ti1-Cl1 2.3135(5), Ti1-Cl2 2.3156(5), Ti1-C<sub>5</sub>H<sub>5</sub><sup>Centroid</sup> 2.0562(9), N2-C1-N3 105.99(15), C1-N1-Ti1 170.26(14), Cl1-Ti1-Cl2 101.87(2), N1-C1-N2 125.66(16), N1-C1-N3 128.31(16).

## 2.9 [Cp(Im<sup>AdDipp</sup>N)TiCl<sub>2</sub>] (5b)

Identification code	mk27mk	
CCDC Number:	2174872	
Empirical formula	C <sub>30</sub> H <sub>39</sub> Cl <sub>2</sub> N <sub>3</sub> Ti	
Formula weight	560.44	
Temperature	100(2) K	
Wavelength	0.71073 Å	
Instrument (scan mode)	XtaLAB Synergy, Single source, HyPix ( $\omega$ scan)	
Crystal system	Monoclinic	
Space group	<i>P</i> 2 <sub>1</sub> / <i>n</i>	
Unit cell dimensions	a = 10.3898(2) Å	$\alpha$ = 90°
	b = 19.4721(2) Å	$\beta$ = 101.575(2)°
	c = 13.9922(2) Å	$\gamma$ = 90°
Volume	2773.21(7) Å <sup>3</sup>	
Z	4	
Density (calculated)	1.342 Mg/m <sup>3</sup>	
Absorption coefficient	0.526 mm <sup>-1</sup>	
F(000)	1184	
Crystal habitus	fragment of trapezoid (orange)	
Crystal size	0.221 x 0.207 x 0.148 mm <sup>3</sup>	
Theta range for data collection	2.240 to 38.315°	
Index ranges	-17≤h≤17, -33≤k≤33, -24≤l≤24	
Reflections collected	193977	
Independent reflections	14793 [R(int) = 0.0306]	
Completeness to theta = 25.242°	99.9 %	
Absorption correction	Semi-empirical from equivalents	
Max. and min. transmission	1.00000 and 0.89211	
Refinement method	Full-matrix least-squares on F <sup>2</sup>	
Data / restraints / parameters	14793 / 0 / 329	
Goodness-of-fit on F <sup>2</sup>	1.070	
Final R indices [I>2sigma(I)]	R1 = 0.0307, wR2 = 0.0836	
R indices (all data)	R1 = 0.0358, wR2 = 0.0856	
Largest diff. peak and hole	0.842 and -0.715 e.Å <sup>-3</sup>	
Crystallisation Details:	dichloromethane/ <i>n</i> -hexane room temperature	
Solution:	SHELXT-2014/5 (G. M. Sheldrick, Acta Cryst., 2015, A71, 3-8)	
Refinement:	SHELXL-2018/3 (G. M. Sheldrick, Acta Cryst. (2008), A64, 112-122)	
Interface:	OLEX2 v1.2 (O. V. Dolomanov et al., J. Appl. Cryst., 2009, 42, 339-341)	
Measurement and Refinement Details:	-	

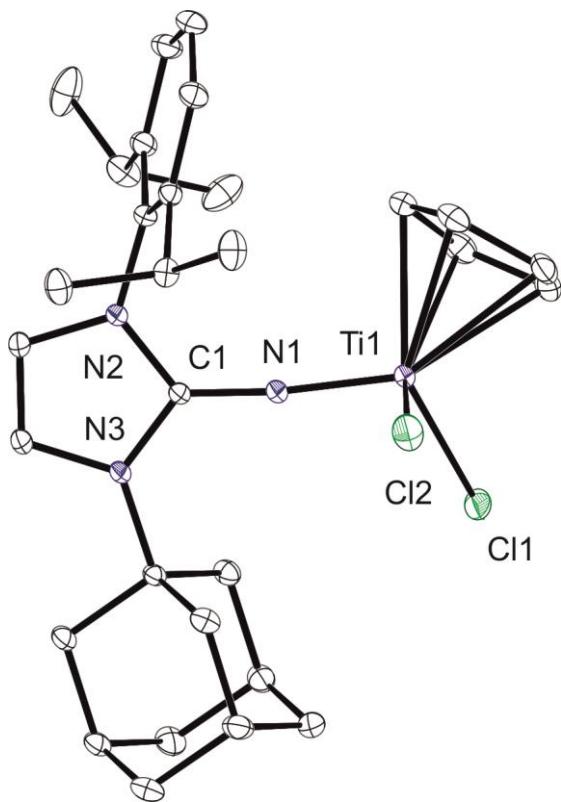


Figure S10: Molecular structure of **5b** with thermal displacement parameters drawn at 50% probability. All hydrogens are omitted for clarity. Selected bond lengths [ $\text{\AA}$ ] and angles [ $^\circ$ ]: C1-N1 1.3189(8), N1-Ti1 1.7760(5), Ti1-Cl1 2.3074(3), Ti1-Cl2 2.3117(2), Ti1-C<sub>5</sub>H<sub>5</sub><sup>Centroid</sup> 2.0689(4), N2-C1-N3 106.13(5), C1-N1-Ti1 170.86(5), Cl1-Ti1-Cl2 102.873(11), N1-C1-N2 125.75(6), N1-C1-N3 128.10(6).

## 2.10 [(Im<sup>AdMes</sup>N)<sub>2</sub>TiCl<sub>2</sub>]·0.5CH<sub>2</sub>Cl<sub>2</sub> (6a)

Identification code	mk19mk
CCDC Number:	2174873
Empirical formula	C <sub>44.5</sub> H <sub>57</sub> Cl <sub>3</sub> N <sub>6</sub> Ti
Formula weight	830.21
Temperature	100(2) K
Wavelength	1.54184 Å
Instrument (scan mode)	XtaLAB Synergy, Single source, HyPix ( $\omega$ scan)
Crystal system	Monoclinic
Space group	P2 <sub>1</sub> /n
Unit cell dimensions	a = 13.24640(4) Å $\alpha$ = 90° b = 13.47847(4) Å $\beta$ = 101.0795(3)° c = 24.79338(8) Å $\gamma$ = 90°
Volume	4344.13(2) Å <sup>3</sup>
Z	4
Density (calculated)	1.269 Mg/m <sup>3</sup>
Absorption coefficient	3.655 mm <sup>-1</sup>
F(000)	1756
Crystal habitus	irregular (orange)
Crystal size	0.253 x 0.110 x 0.098 mm <sup>3</sup>
Theta range for data collection	3.534 to 77.822°
Index ranges	-16≤h≤16, -17≤k≤17, -31≤l≤28
Reflections collected	94547
Independent reflections	9180 [R(int) = 0.0312]
Completeness to theta = 67.684°	100.0 %
Absorption correction	Gaussian
Max. and min. transmission	1.000 and 0.531
Refinement method	Full-matrix least-squares on F <sup>2</sup>
Data / restraints / parameters	9180 / 6 / 596
Goodness-of-fit on F <sup>2</sup>	1.069
Final R indices [I>2sigma(I)]	R1 = 0.0450, wR2 = 0.1290
R indices (all data)	R1 = 0.0462, wR2 = 0.1308
Largest diff. peak and hole	1.180 and -0.885 e.Å <sup>-3</sup>
Crystallisation Details:	CH <sub>2</sub> Cl <sub>2</sub> /n-hexane at room temperature

Solution: SHELXT-2014/5 (G. M. Sheldrick, Acta Cryst., 2015, A71, 3-8)

Refinement: SHELXL-2018/3 (G. M. Sheldrick, Acta Cryst., 2008, A64, 112-122)

Interface: OLEX2 v1.2 (O. V. Dolomanov et al., J. Appl. Cryst., 2009, 42, 339-341)

Measurement and Refinement Details: The CH<sub>2</sub>Cl<sub>2</sub> fragment is disordered over an inversion center and occupies a special position. Therefore, its occupancy was set at 0.5 and refined accordingly. Additionally the mesityl group on one ligand is disordered and was refined accordingly resulting in a 60/40 occupancy.

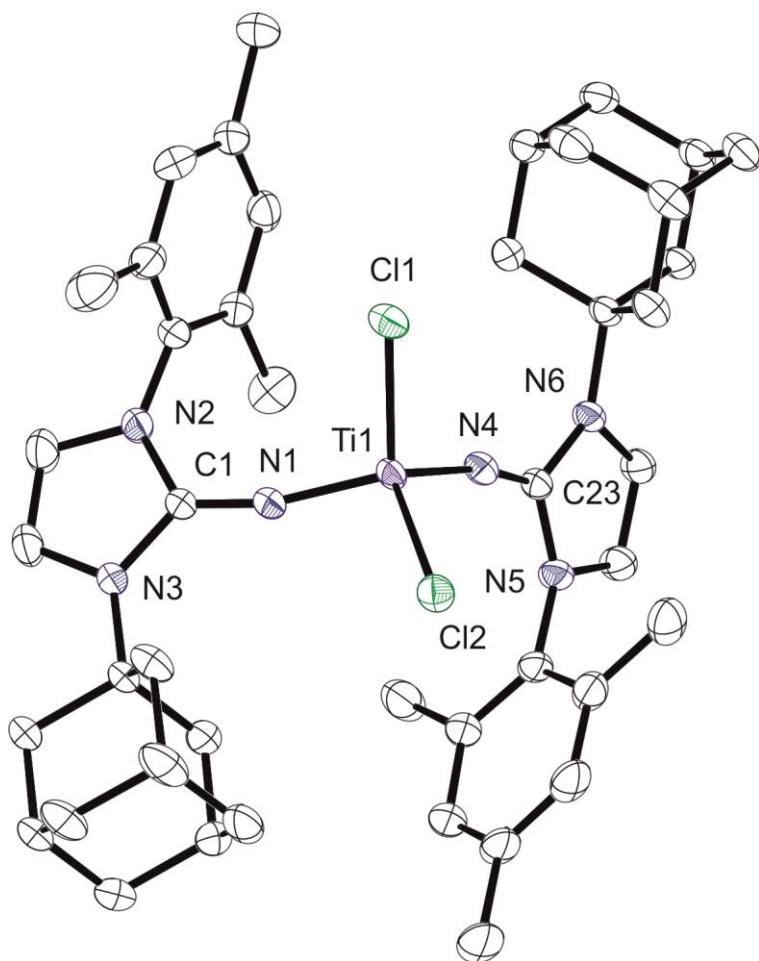


Figure S11: Molecular structure of the main component of **6a**·0.5CH<sub>2</sub>Cl<sub>2</sub> with thermal displacement parameters drawn at 50% probability. All hydrogens, 0.5 molecules of CH<sub>2</sub>Cl<sub>2</sub> and the minor component of the disordered mesityl group at N5 are omitted for clarity. Selected bond lengths [Å] and angles [°]: C1-N1 1.304(2), C23-N4 1.302(2), N1-Ti1 1.7951(15), N4-Ti1 1.7867(16), Ti1-Cl1 2.2938(5), Ti1-Cl2 2.3160(5), N2-C1-N3 105.89(14), N5-C23-N6 105.69(15), C1-N1-Ti1 166.51(13), C23-N4-Ti1 166.94(13), Cl1-Ti1-Cl2 109.49(2), N1-Ti1-N4 113.57(8).

## 2.11 [(Im<sup>AdDipp</sup>N)<sub>2</sub>TiCl<sub>2</sub>]·CH<sub>2</sub>Cl<sub>2</sub> (6b)

Identification code	mk24mk2
CCDC Number:	2174874
Empirical formula	C <sub>51</sub> H <sub>70</sub> Cl <sub>4</sub> N <sub>6</sub> Ti
Formula weight	956.83
Temperature	100(2) K
Wavelength	1.54184 Å
Instrument (scan mode)	XtaLAB Synergy, Single source, HyPix ( $\omega$ scan)
Crystal system	Monoclinic
Space group	P2 <sub>1</sub> /c
Unit cell dimensions	a = 16.9882(3) Å $\alpha$ = 90° b = 18.0462(2) Å $\beta$ = 116.512(2)° c = 18.3671(3) Å $\gamma$ = 90°
Volume	5038.71(15) Å <sup>3</sup>
Z	4
Density (calculated)	1.261 Mg/m <sup>3</sup>
Absorption coefficient	3.694 mm <sup>-1</sup>
F(000)	2032
Crystal habitus	needle (orange)
Crystal size	0.135 x 0.080 x 0.051 mm <sup>3</sup>
Theta range for data collection	2.907 to 77.812°
Index ranges	-21≤h≤21, -19≤k≤22, -23≤l≤23
Reflections collected	122767
Independent reflections	10633 [R(int) = 0.0647]
Completeness to theta = 67.684°	100.0 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	1.00000 and 0.58339
Refinement method	Full-matrix least-squares on F <sup>2</sup>
Data / restraints / parameters	10633 / 6 / 567
Goodness-of-fit on F <sup>2</sup>	1.076
Final R indices [I>2sigma(I)]	R1 = 0.0561, wR2 = 0.1599
R indices (all data)	R1 = 0.0606, wR2 = 0.1649
Largest diff. peak and hole	0.813 and -1.177 e.Å <sup>-3</sup>
Crystallisation Details:	dichloromethane/n-hexane room temperature
Solution:	SHELXT-2014/5 (G. M. Sheldrick, Acta Cryst., 2015, A71, 3-8)
Refinement:	SHELXL-2018/3 (G. M. Sheldrick, Acta Cryst., 2008, A64, 112-122)
Interface:	OLEX2 v1.2 (O. V. Dolomanov et al., J. Appl. Cryst., 2009, 42, 339-341)
Measurement and Refinement Details:	Data had to be cut due to sample decay.

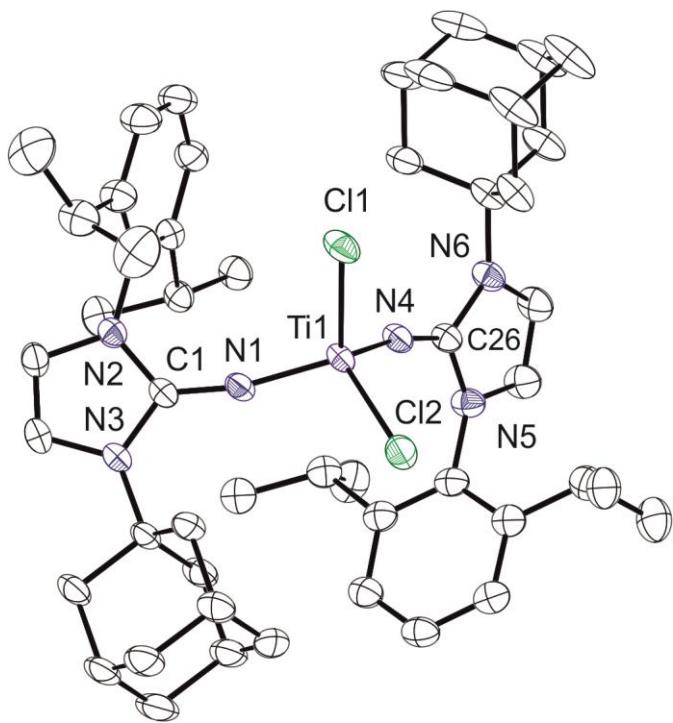


Figure S12: Molecular structure of the main component of **6b**·CH<sub>2</sub>Cl<sub>2</sub> with thermal displacement parameters drawn at 50% probability. All hydrogens and one molecule of CH<sub>2</sub>Cl<sub>2</sub> are omitted for clarity. Selected bond lengths [Å] and angles [°]: C1-N1 1.304(3), C26-N4 1.302(3), N1-Ti1 1.7994(18), N4-Ti1 1.8067(18), Ti1-Cl1 2.2836(6), Ti1-Cl2 2.3042(6), N2-C1-N3 105.69(18), N5-C26-N6 105.24(18), C1-N1-Ti1 167.14(17), C26-N4-Ti1 167.64(17), Cl1-Ti1-Cl2 105.19(3), N1-Ti1-N4 116.99(9).