

## Supporting Information

### Design of “turn-off” luminescent Ln-MOFs for sensitive detection of cyanide anion

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## Crystallographic Data Collection and Structure Determination

Single-crystal X-ray diffraction data of Ln-MOFs **1** and **2** were collected at 150(10) K using a Rigaku HyPix diffractometer with graphite-monochromatic Mo K $\alpha$  radiation ( $\lambda = 0.71073 \text{ \AA}$ ). An empirical absorption correction was applied using the SADABS program. The program Olex2 was used to solve the structures of Ln-MOFs **1** and **2**. All the structures were solved with ShelXT using intrinsic phasing and refined with full-matrix least-squares method on  $F^2$  by using the ShelXL refinement package. All non-hydrogen atoms were refined and treated anisotropically. The diffraction contribution of disordered solvent molecules in the structure were removed by the PLATON/SQUEEZE. The Ln-MOFs **1** and **2** topological type was explored from the crystallographic data with TOPOS Pro software.

**Table S1.** Crystallographic data and structure refinement for **1** and **2**

<b>Compound</b>	<b>1</b>	<b>2</b>
CCDC number	2106266	2106267
Formula	C <sub>21</sub> H <sub>24</sub> Br <sub>3</sub> EuN <sub>3</sub> O <sub>9</sub>	C <sub>21</sub> H <sub>24</sub> Br <sub>3</sub> N <sub>3</sub> O <sub>9</sub> Tb
<i>D</i> <sub>calc.</sub> / g cm <sup>-3</sup>	2.057	2.091
$\mu$ /mm <sup>-1</sup>	6.673	7.011
Formula Weight	854.12	861.08
Size/mm <sup>3</sup>	0.12×0.11×0.10	0.12 × 0.11 × 0.10
<i>T</i> /K	149.98(10)	150.04(10)
Crystal System	triclinic	triclinic
Space Group	<i>P</i> -1	<i>P</i> -1
<i>a</i> /Å	11.2941(2)	11.2415(3)
<i>b</i> /Å	11.5981(2)	11.5520(3)
<i>c</i> /Å	12.9541(3)	12.9279(4)
$\alpha$ /°	108.813(2)	109.079(3)
$\beta$ /°	114.558(2)	114.551(3)
$\gamma$ /°	98.648(2)	97.782(2)
<i>V</i> /Å <sup>3</sup>	1378.90(5)	1369.46(8)
<i>Z</i>	2	2
Wavelength/Å	0.71073	0.71073
Radiation type	Mo K $\alpha$	Mo K $\alpha$
$\theta_{min}$ /°	1.912	1.913
$\theta_{max}$ /°	29.481	29.482
Measured Refl's.	30072	31328
Refl's $I \geq 2\sigma(I)$	5641	5384
Indep't Refl's	6452	6425
<i>R</i> <sub>int</sub>	0.0364	0.0588
Parameters	388	388
Restraints	240	120
Goof	1.051	1.065
Final <i>R</i> indexes [ $I \geq 2\sigma(I)$ ]	<i>R</i> <sub>1</sub> =0.0233 <i>wR</i> <sub>2</sub> =0.0464	<i>R</i> <sub>1</sub> =0.0414 <i>wR</i> <sub>2</sub> =0.1041
Final <i>R</i> indexes [all data]	<i>R</i> <sub>1</sub> =0.0303 <i>wR</i> <sub>2</sub> =0.0481	<i>R</i> <sub>1</sub> =0.0530 <i>wR</i> <sub>2</sub> =0.1094

**Table S2.** Selected Bond Lengths (Å) and Angles (°) for **1**.

Eu1-O1	2.3645(17)	N2-C16	1.320(4)
Eu1-O1 <sup>i</sup>	2.6528(17)	N2-C17	1.462(5)
Eu1-O2 <sup>i</sup>	2.5144(19)	N2-C18	1.447(5)
Eu1-O3 <sup>ii</sup>	2.4555(19)	C1-C2	1.385(4)
Eu1-O4 <sup>ii</sup>	2.4925(19)	C1-C6	1.394(4)
Eu1-O5	2.3781(17)	C1-C7	1.501(4)
Eu1-O6 <sup>i</sup>	2.4161(18)	C2-C3	1.372(4)
Eu1-O7	2.4238(17)	C3-C4	1.401(4)
Eu1-O8	2.4417(19)	C4-C5	1.399(4)
Br1-C2	1.904(3)	C4-C8	1.510(4)
Br2-C5	1.900(3)	C5-C6	1.384(4)
Br3-C10	1.902(3)	C9-C10	1.388(3)
O1-C7	1.259(3)	C9-C11 <sup>iii</sup>	1.394(4)
O2-C7	1.248(3)	C9-C12	1.513(3)
O3-C8	1.257(3)	C10-C11	1.384(4)
O4-C8	1.255(3)	O9-C19	1.177(7)
O5-C12	1.258(3)	N3-C19	1.351(7)
O6-C12	1.257(3)	N3-C20	1.458(7)
O7-C13	1.229(3)	N3-C21	1.452(7)
O8-C16	1.236(3)	N3A-C19A	1.381(8)
N1-C13	1.325(4)	N3A-C20A	1.469(8)
N1-C14	1.443(5)	N3A-C21A	1.435(8)
N1-C15	1.440(5)	O9A-C19A	1.191(8)
O1-Eu1-O1 <sup>i</sup>	71.05(6)	C16-N2-C17	120.8(3)
O1-Eu1-O2 <sup>i</sup>	121.13(6)	C16-N2-C18	121.0(3)
O1-Eu1-O3 <sup>ii</sup>	84.46(6)	C18-N2-C17	118.1(3)
O1-Eu1-O4 <sup>ii</sup>	75.86(6)	C2-C1-C6	117.7(2)
O1-Eu1-O5	74.85(6)	C2-C1-C7	123.0(2)
O1-Eu1-O6 <sup>i</sup>	75.14(6)	C6-C1-C7	119.3(2)
O1-Eu1-O7	144.58(6)	C1-C2-Br1	119.6(2)
O1-Eu1-O8	140.49(6)	C3-C2-Br1	118.8(2)
O2 <sup>i</sup> -Eu1-O1 <sup>i</sup>	50.09(5)	C3-C2-C1	121.6(2)
O3 <sup>ii</sup> -Eu1-O1 <sup>i</sup>	148.65(6)	C2-C3-C4	121.2(2)
O3 <sup>ii</sup> -Eu1-O2 <sup>i</sup>	147.99(6)	C3-C4-C8	116.1(2)
O3 <sup>ii</sup> -Eu1-O4 <sup>ii</sup>	52.55(6)	C5-C4-C3	117.3(2)
O4 <sup>ii</sup> -Eu1-O1 <sup>i</sup>	133.71(6)	C5-C4-C8	126.6(2)
O4 <sup>ii</sup> -Eu1-O2 <sup>i</sup>	145.93(6)	C4-C5-Br2	124.1(2)
O5-Eu1-O1 <sup>i</sup>	67.29(6)	C6-C5-Br2	115.0(2)
O5-Eu1-O2 <sup>i</sup>	82.60(6)	C6-C5-C4	120.9(2)

O5-Eu1-O3 <sup>ii</sup>	125.52(6)	C5-C6-C1	121.2(2)
O5-Eu1-O4 <sup>ii</sup>	73.49(6)	O1-C7-Eu1 <sup>i</sup>	64.09(13)
O5-Eu1-O6 <sup>i</sup>	133.16(6)	O1-C7-C1	118.5(2)
O5-Eu1-O7	140.50(6)	O2-C7-Eu1 <sup>i</sup>	57.71(13)
O5-Eu1-O8	73.05(6)	O2-C7-O1	121.8(2)
O6 <sup>i</sup> -Eu1-O1 <sup>i</sup>	69.36(6)	O2-C7-C1	119.7(2)
O6 <sup>i</sup> -Eu1-O2 <sup>i</sup>	83.13(6)	C1-C7-Eu1 <sup>i</sup>	177.36(17)
O6 <sup>i</sup> -Eu1-O3 <sup>ii</sup>	85.95(6)	O3-C8-Eu1 <sup>ii</sup>	59.84(13)
O6 <sup>i</sup> -Eu1-O4 <sup>ii</sup>	130.94(6)	O3-C8-C4	118.4(2)
O6 <sup>i</sup> -Eu1-O7	75.07(6)	O4-C8-Eu1 <sup>ii</sup>	61.52(14)
O6 <sup>i</sup> -Eu1-O8	144.37(6)	O4-C8-O3	121.4(2)
O7-Eu1-O1 <sup>i</sup>	114.83(6)	O4-C8-C4	120.2(2)
O7-Eu1-O2 <sup>i</sup>	73.37(6)	C4-C8-Eu1 <sup>ii</sup>	178.18(18)
O7-Eu1-O3 <sup>ii</sup>	74.76(6)	C10-C9-C11 <sup>iii</sup>	117.9(2)
O7-Eu1-O4 <sup>ii</sup>	110.97(6)	C10-C9-C12	124.5(2)
O7-Eu1-O8	71.20(6)	C11 <sup>iii</sup> -C9-C12	117.6(2)
O8-Eu1-O1 <sup>i</sup>	115.46(6)	C9-C10-Br3	120.22(19)
O8-Eu1-O2 <sup>i</sup>	76.65(7)	C11-C10-Br3	118.05(19)
O8-Eu1-O3 <sup>ii</sup>	95.87(7)	C11-C10-C9	121.7(2)
O8-Eu1-O4 <sup>ii</sup>	73.40(7)	C10-C11-C9 <sup>iii</sup>	120.5(2)
Eu1-O1-Eu1 <sup>i</sup>	108.95(6)	O5-C12-C9	114.5(2)
C7-O1-Eu1 <sup>i</sup>	90.64(15)	O6-C12-O5	127.3(2)
C7-O1-Eu1	160.36(17)	O6-C12-C9	118.2(2)
C7-O2-Eu1 <sup>i</sup>	97.48(15)	O7-C13-N1	124.8(3)
C8-O3-Eu1 <sup>ii</sup>	93.88(15)	O8-C16-N2	124.9(3)
C8-O4-Eu1 <sup>ii</sup>	92.21(16)	C19-N3-C20	118.9(7)
C12-O5-Eu1	138.50(16)	C19-N3-C21	121.4(6)
C12-O6-Eu1 <sup>i</sup>	135.86(15)	C21-N3-C20	119.5(6)
C13-O7-Eu1	125.31(18)	O9-C19-N3	123.5(7)
C16-O8-Eu1	123.83(18)	C19A-N3A-C20A	120.7(8)
C13-N1-C14	121.2(3)	C19A-N3A-C21A	119.7(8)
C13-N1-C15	121.6(3)	C21A-N3A-C20A	119.5(8)
C15-N1-C14	117.1(3)	O9A-C19A-N3A	124.4(8)

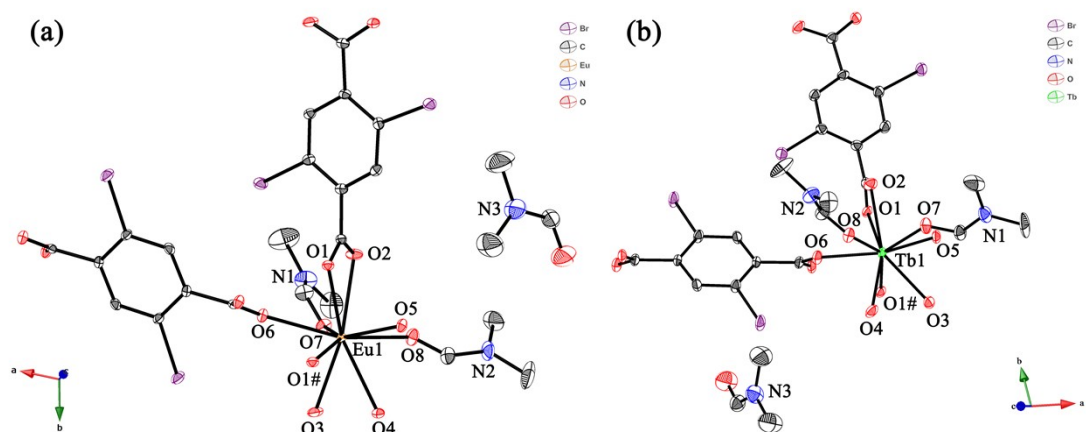
Symmetry codes: (i) 1-X, -Y, 1-Z; (ii) 1-X, 1-Y, 1-Z; (iii) -X, -Y, 1-Z

**Table S3.** Selected Bond Lengths (Å) and Angles (°) for 2

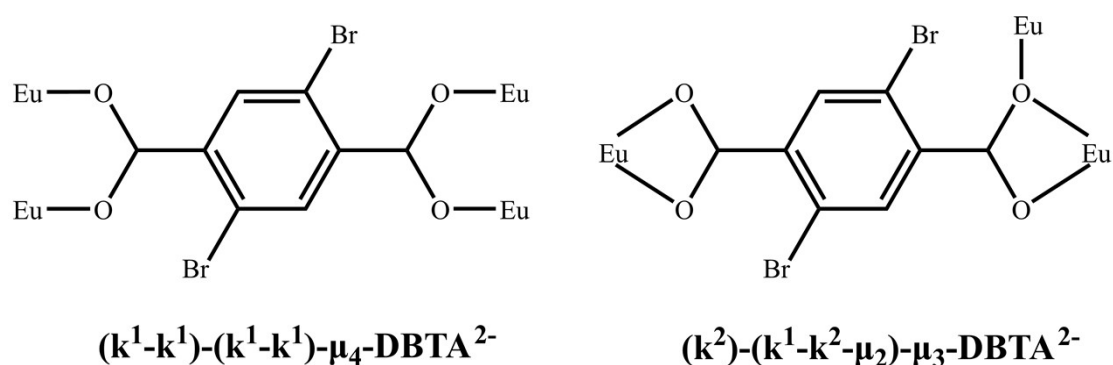
Tb1-O1	2.335(3)	N2-C16	1.313(7)
Tb1-O1 <sup>i</sup>	2.650(3)	N2-C17	1.440(7)
Tb1-O2 <sup>i</sup>	2.483(3)	N2-C18	1.455(7)
Tb1-O3 <sup>ii</sup>	2.475(3)	C1-C2	1.385(6)
Tb1-O4 <sup>ii</sup>	2.434(3)	C1-C6	1.383(6)
Tb1-O5	2.350(3)	C1-C7	1.509(6)
Tb1-O6 <sup>i</sup>	2.387(3)	C2-C3	1.390(6)
Tb1-O7	2.421(3)	C3-C4	1.391(6)
Tb1-O8	2.397(3)	C4-C5	1.405(6)
Br1-C2	1.895(5)	C4-C8	1.507(6)
Br2-C5	1.901(5)	C5-C6	1.386(6)
Br3-C12	1.902(4)	C9-C10	1.525(6)
O1-C7	1.255(6)	C10-C11	1.390(6)
O2-C7	1.245(6)	C10-C12 <sup>iii</sup>	1.391(6)
O3-C8	1.254(6)	C11-C12	1.382(6)
O4-C8	1.261(6)	O9-C19	1.175(18)
O5-C9	1.251(6)	N3-C19	1.35(2)
O6-C9	1.261(6)	N3-C20	1.42(2)
O7-C13	1.241(6)	N3-C21	1.422(19)
O8-C16	1.227(6)	O9A-C19A	1.194(17)
N1-C13	1.316(7)	C19A-N3A	1.35(2)
N1-C14	1.459(8)	N3A-C20A	1.458(18)
N1-C15	1.448(8)	N3A-C21A	1.43(3)
O1-Tb1-O1 <sup>i</sup>	71.01(11)	C16-N2-C17	121.1(6)
O1-Tb1-O2 <sup>i</sup>	121.43(11)	C16-N2-C18	121.6(5)
O1-Tb1-O3 <sup>ii</sup>	75.74(11)	C17-N2-C18	117.2(5)
O1-Tb1-O4 <sup>ii</sup>	84.06(12)	C2-C1-C7	122.0(4)
O1-Tb1-O5	75.18(11)	C6-C1-C2	118.8(4)
O1-Tb1-O6 <sup>i</sup>	75.17(11)	C6-C1-C7	119.2(4)
O1-Tb1-O7	140.86(11)	C1-C2-Br1	120.6(3)
O1-Tb1-O8	144.23(11)	C1-C2-C3	120.4(4)
O2 <sup>i</sup> -Tb1-O1 <sup>i</sup>	50.42(10)	C3-C2-Br1	119.0(3)
O3 <sup>ii</sup> -Tb1-O1 <sup>i</sup>	133.70(11)	C2-C3-C4	121.7(4)
O3 <sup>ii</sup> -Tb1-O2 <sup>i</sup>	145.82(11)	C3-C4-C5	117.1(4)
O4 <sup>ii</sup> -Tb1-O1 <sup>i</sup>	147.82(12)	C3-C4-C8	116.3(4)
O4 <sup>ii</sup> -Tb1-O2 <sup>i</sup>	147.79(11)	C5-C4-C8	126.6(4)
O4 <sup>ii</sup> -Tb1-O3 <sup>ii</sup>	53.08(11)	C4-C5-Br2	123.8(3)
O5-Tb1-O1 <sup>i</sup>	67.14(10)	C6-C5-Br2	115.3(3)
O5-Tb1-O2 <sup>i</sup>	82.13(11)	C6-C5-C4	121.0(4)

O5-Tb1-O3 <sup>ii</sup>	73.91(11)	C1-C6-C5	121.0(4)
O5-Tb1-O4 <sup>ii</sup>	126.41(11)	O1-C7-Tb1 <sup>i</sup>	65.1(2)
O5-Tb1-O6 <sup>i</sup>	133.28(11)	O1-C7-C1	118.1(4)
O5-Tb1-O7	73.33(11)	O2-C7-Tb1 <sup>i</sup>	57.4(2)
O5-Tb1-O8	140.51(12)	O2-C7-O1	122.5(4)
O6 <sup>i</sup> -Tb1-O1 <sup>i</sup>	69.46(10)	O2-C7-C1	119.4(4)
O6 <sup>i</sup> -Tb1-O2 <sup>i</sup>	83.49(11)	C1-C7-Tb1 <sup>i</sup>	176.8(3)
O6 <sup>i</sup> -Tb1-O3 <sup>ii</sup>	130.69(11)	O3-C8-Tb1 <sup>ii</sup>	61.6(2)
O6 <sup>i</sup> -Tb1-O4 <sup>ii</sup>	84.99(11)	O3-C8-O4	121.4(4)
O6 <sup>i</sup> -Tb1-O7	143.96(12)	O3-C8-C4	120.3(4)
O6 <sup>i</sup> -Tb1-O8	74.73(11)	O4-C8-Tb1 <sup>ii</sup>	59.8(2)
O7-Tb1-O1 <sup>i</sup>	115.79(11)	O4-C8-C4	118.3(4)
O7-Tb1-O2 <sup>i</sup>	76.34(12)	C4-C8-Tb1 <sup>ii</sup>	178.0(3)
O7-Tb1-O3 <sup>ii</sup>	73.67(12)	O5-C9-O6	128.0(4)
O7-Tb1-O4 <sup>ii</sup>	96.39(12)	O5-C9-C10	114.8(4)
O8-Tb1-O1 <sup>i</sup>	114.90(10)	O6-C9-C10	117.1(4)
O8-Tb1-O2 <sup>i</sup>	73.46(11)	C11-C10-C9	117.3(4)
O8-Tb1-O3 <sup>ii</sup>	110.98(12)	C11-C10-C12 <sup>iii</sup>	118.3(4)
O8-Tb1-O4 <sup>ii</sup>	74.51(12)	C12 <sup>iii</sup> -C10-C9	124.3(4)
O8-Tb1-O7	71.06(12)	C12-C11-C10	120.1(4)
Tb1-O1-Tb1 <sup>i</sup>	108.99(11)	C10 <sup>iii</sup> -C12-Br3	120.1(3)
C7-O1-Tb1	161.5(3)	C11-C12-Br3	118.3(3)
C7-O1-Tb1 <sup>i</sup>	89.5(3)	C11-C12-C10 <sup>iii</sup>	121.6(4)
C7-O2-Tb1 <sup>i</sup>	97.6(3)	O7-C13-N1	124.5(5)
C8-O3-Tb1 <sup>ii</sup>	91.9(3)	O8-C16-N2	125.8(5)
C8-O4-Tb1 <sup>ii</sup>	93.6(3)	C19-N3-C20	121.3(15)
C9-O5-Tb1	138.4(3)	C19-N3-C21	120.7(16)
C9-O6-Tb1 <sup>i</sup>	135.3(3)	C20-N3-C21	117.8(15)
C13-O7-Tb1	123.7(3)	O9-C19-N3	124.5(18)
C16-O8-Tb1	126.1(3)	O9A-C19A-N3A	126.7(16)
C13N1-C14	120.7(6)	C19A-N3A-C20A	122.0(14)
C13-N1-C15	120.8(5)	C19A-N3A-C21A	121.1(16)
C15-N1-C14	118.5(5)	C21A-N3A-C20A	116.8(16)

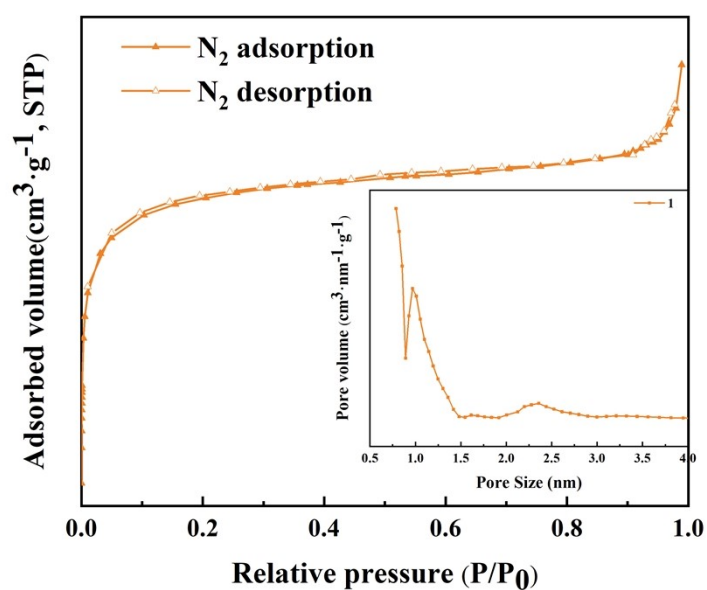
Symmetry codes: (i) 1-X, -Y, 1-Z; (ii) 1-X, 1-Y, 1-Z; (iii) -X, -Y, 1-Z



**Fig. S1** ORTEP diagrams of the structural unit **1** and **2** with partial atom numbering schemes, the thermal displacement ellipsoids are drawn at 40% probability. Hydrogen atoms have been omitted for clarity.



**Fig. S2** The coordination modes of  $\text{DBTA}^{2-}$  ligand.



**Fig. S3**  $\text{N}_2$  adsorption-desorption isotherms of **1**, insert: Pore size distribution plot.



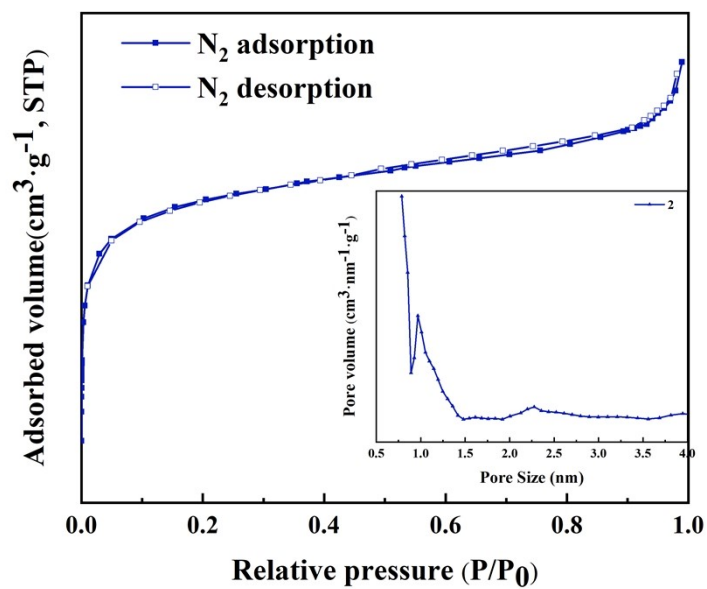


Fig. S4  $\text{N}_2$  adsorption-desorption isotherms of **2**, insert: Pore size distribution plot.

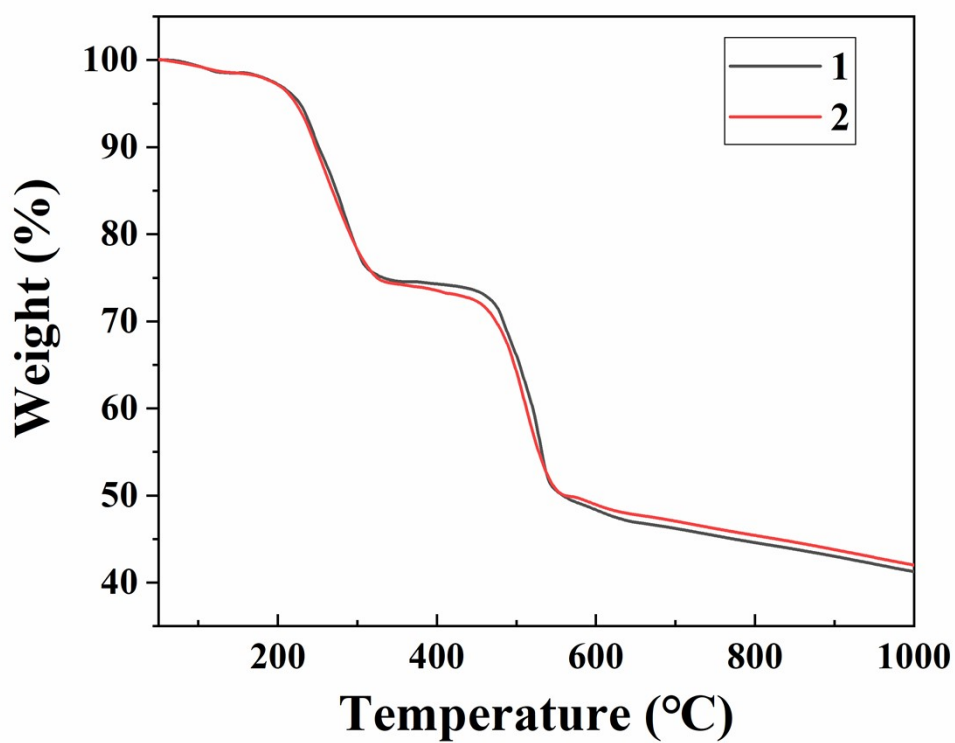
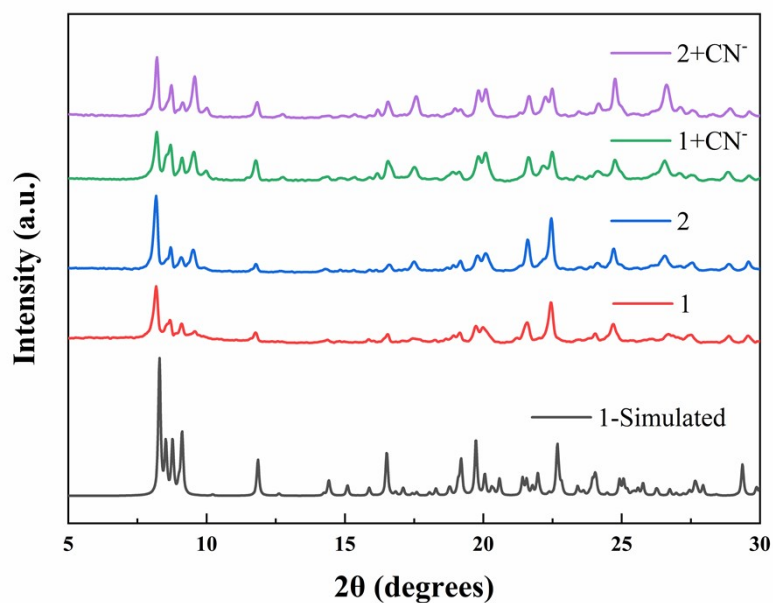
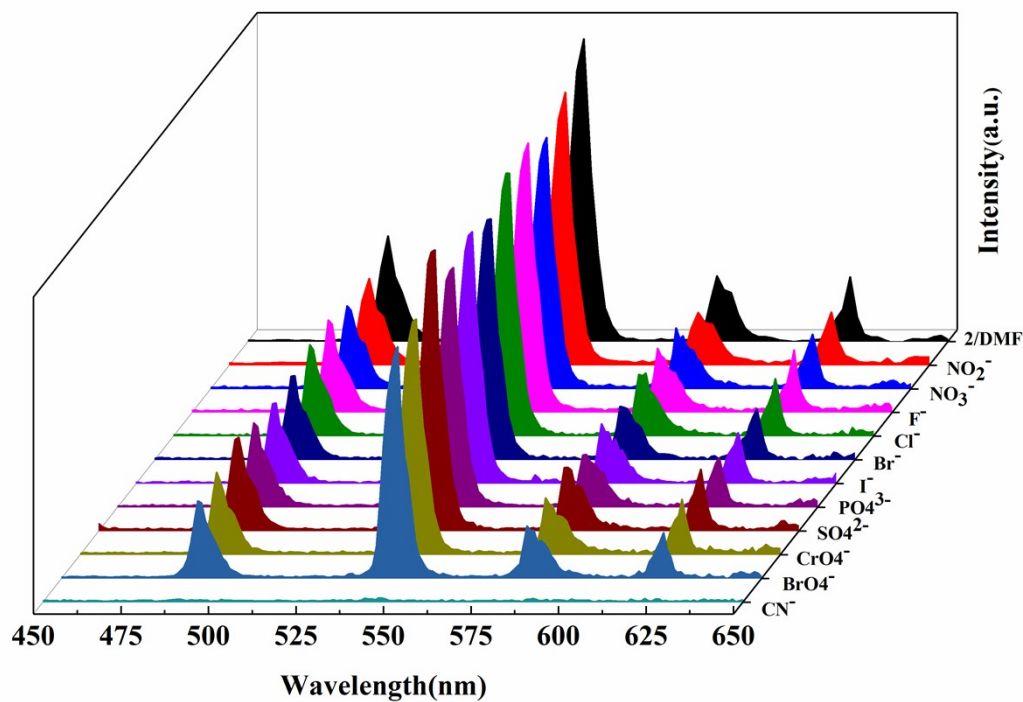


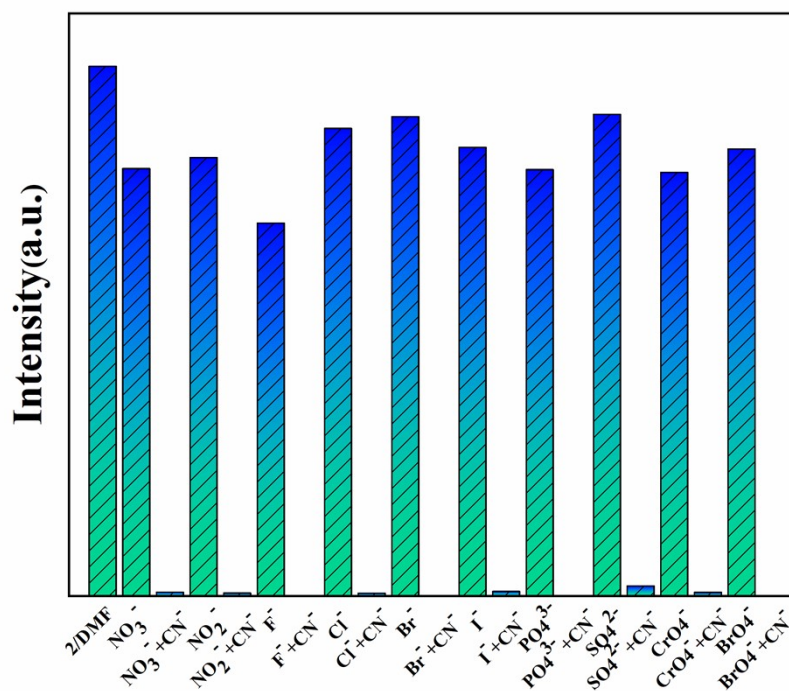
Fig. S5 Thermogravimetric curve of **1** and **2**.



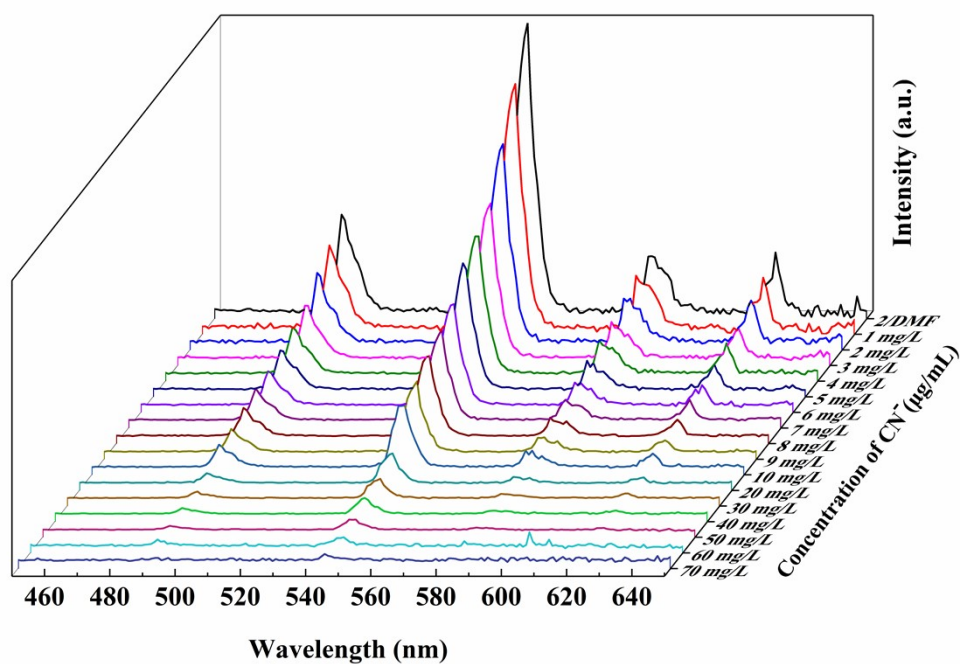
**Fig. S6** Simulated PXRD patterns of **1** and synthesized PXRD of compounds **1**, **2**, **1+CN<sup>-</sup>** and **2+CN<sup>-</sup>**.



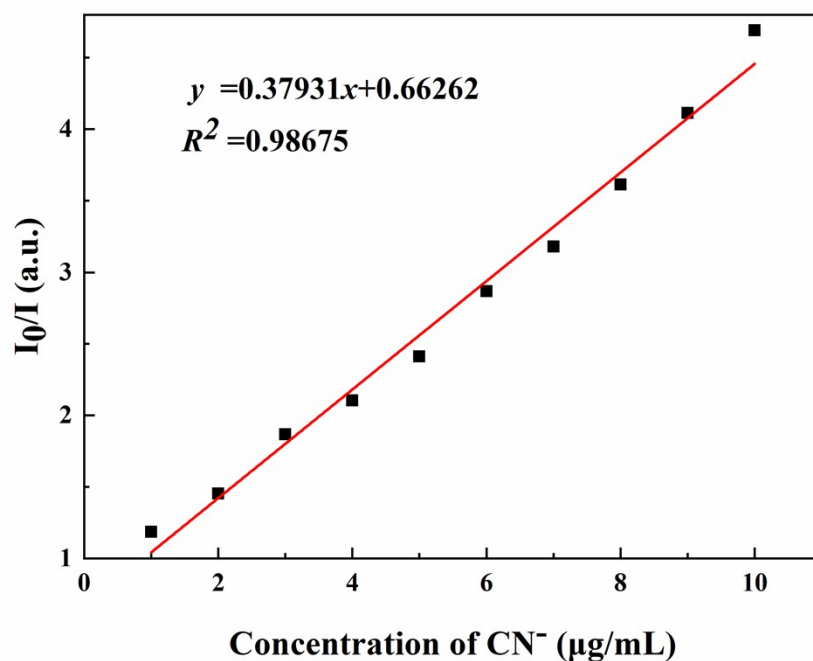
**Fig. S7** Emission spectra and luminescence intensity at 546 nm of **2/DMF** treated with different anions ( $\lambda_{\text{ex}}=303$  nm, 100  $\mu\text{g/mL}$ ).



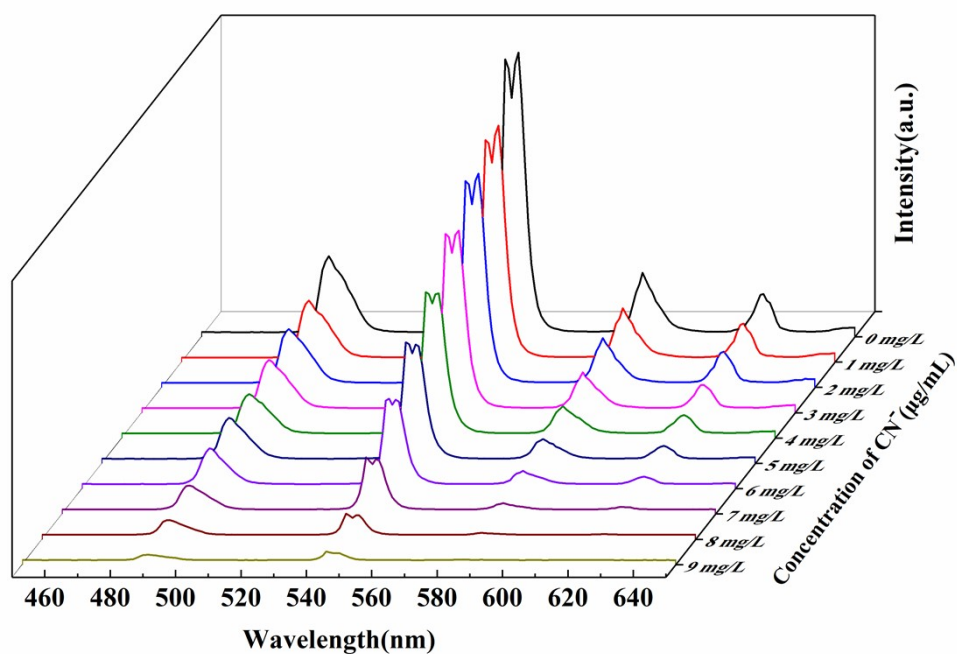
**Fig. S8** Comparison of the luminescence intensity (546 nm) of 2/DMF suspension dispersion mixed with interfering anions ( $\lambda_{\text{ex}}=303$  nm, 100  $\mu\text{g/mL}$ )



**Fig. S9** Emission spectra of 2/DMF with different concentrations of CN<sup>-</sup>(1-70  $\mu\text{g/mL}$ )



**Fig. S10** Relationship between  $I_0/I$  and different  $\text{CN}^-$  concentrations and Stern-Volmer formula fitting of **2** with concentrations of 1-10  $\mu\text{g/mL}$ .



**Fig. S11** Emission spectra of **2** with different concentrations of  $\text{CN}^-$  (1-9  $\mu\text{g/mL}$ ) in aqueous solution.

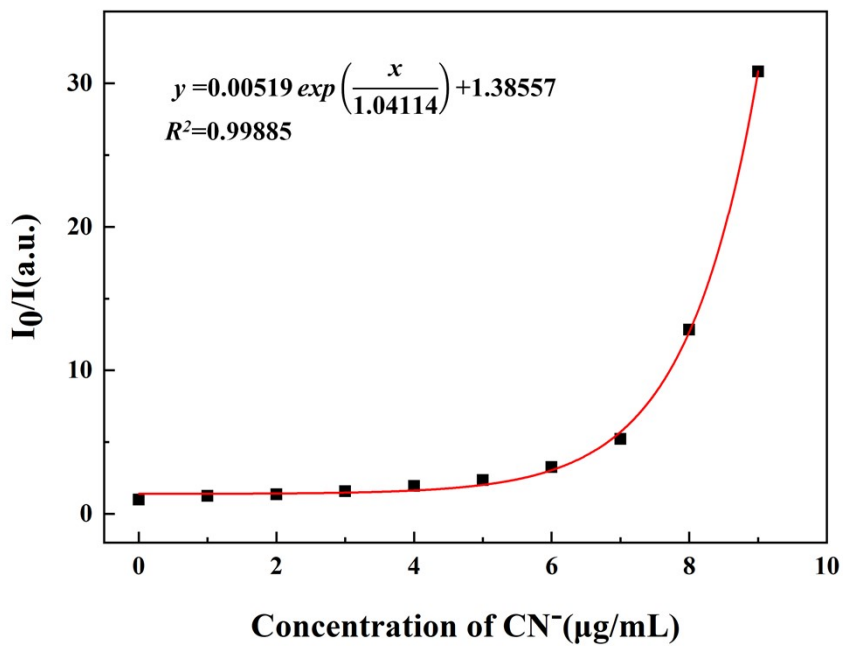


Fig. S12 Relationship between  $I_0/I$  and different  $CN^-$  concentrations of 2.

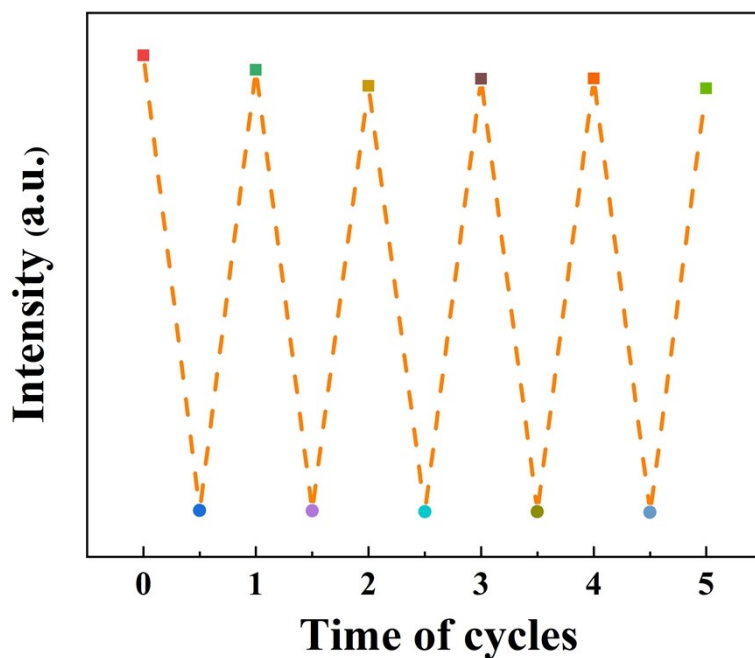


Fig. S13 Fluorescence intensity (619 nm) of 1 during five recycles.

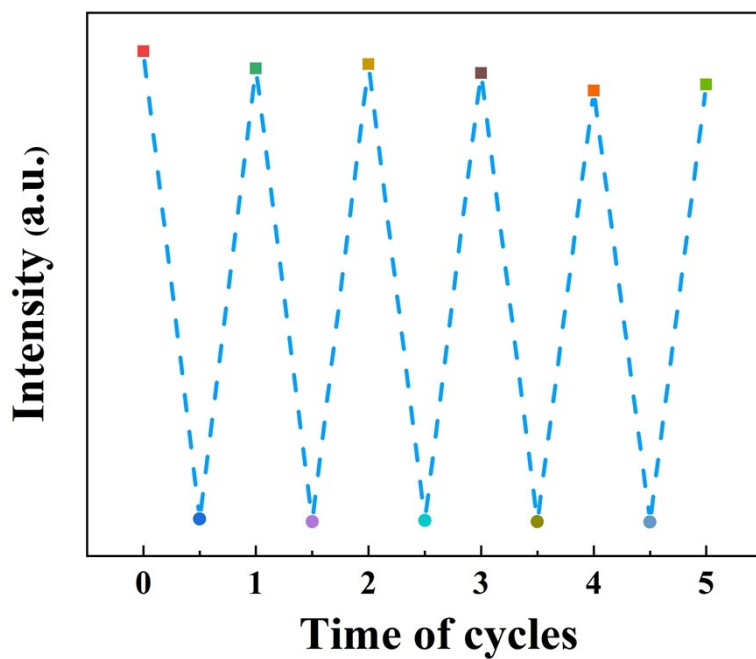


Fig. S14 Fluorescence intensity (546 nm) of 2 during five recycles.

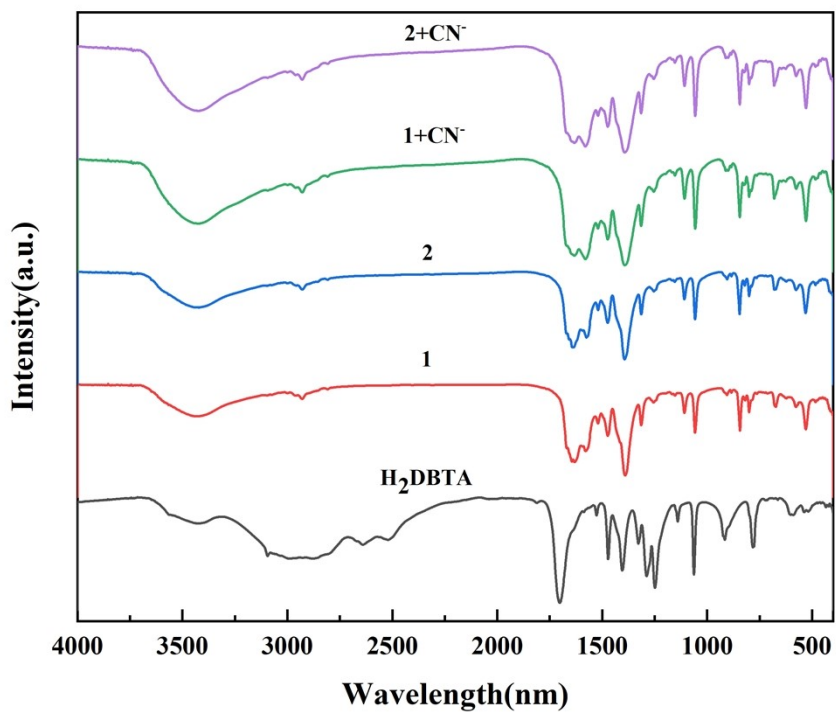


Fig. S15 FT-IR spectra of ligand, 1, 2, 1+CN<sup>-</sup> and 2+CN<sup>-</sup>.

**Table S4.** Comparison of the LOD values of CN<sup>-</sup> with other materials.

Material	Solvent	LOD	Ref.
<b>1</b>	DMF	1.691 μM	This work
<b>2</b>	DMF	8.307 μM	This work
2-HPEAPB	CH <sub>3</sub> CN-H <sub>2</sub> O	8.0 μM	1
[Hf <sub>6</sub> O <sub>4</sub> (OH) <sub>4</sub> (C <sub>8</sub> H <sub>2</sub> O <sub>4</sub> S <sub>2</sub> ) <sub>6</sub> ]·9H <sub>2</sub> O·2DMF	H <sub>2</sub> O	0.35 μM	2
4-(p-tolyl)thiazol-2-amine and 2-hydroxy-1-naphthaldehyde	Bis-tris buffer/DMSO	19.4 μM	3
6,7-dihydroxycoumarin	DMSO-water	5.77 μM	4
HBTM-HC	DMSO	1.38 μM	5
triarylborane substituted bithiazole	THF	2.1 μM	6
L	DMSO/H <sub>2</sub> O	6.4 μM	7
M-ZIF-90	H <sub>2</sub> O/DMSO	2 μM	8
bio-MOF-1⊃DAAC	H <sub>2</sub> O	5.2 ppb	9
TIP-Co	THF-H <sub>2</sub> O	89.3 nM	10
CuCl <sub>2</sub> @MOF-867	H <sub>2</sub> O	0.19 μM	11

## References

1. X. X. Ou, Y. L. Jin, X. Q. Chen, C. B. Gong, X. B. Ma, Y. S. Wang, C. F. Chow and Q. Tang, *Anal. Methods*, 2015, **7**, 5239-5244.
2. R. Dalapati, S. Nandi and S. Biswas, *Dalton Trans.*, 2020, **49**, 8684-8692.
3. S. M. Kim, M. Kang, I. Choi, J. J. Lee and C. Kim, *New J. Chem.*, 2016, **40**, 7768-7778.
4. R. Kaushik, A. Ghosh, A. Singh, P. Gupta, A. Mittal and D. A. Jose, *Acs Sensors*, 2016, **1**, 1265-1271.
5. L. Tang, L. Zhou, A. Liu, X. Yan, K. Zhong, X. Liu, X. Gao and J. Li, *Dyes Pigm.*, 2021, **186**, 109034.
6. R. Maragani, T. S. Reddy and R. Misra, *Tetrahedron Lett.*, 2016, **57**, 3853-3857.
7. Z. Li, C. Liu, S. Wang, L. Xiao and X. Jing, *Spectrochim. Acta, Part A*, 2019, **210**, 321-328.
8. A. Karmakar, N. Kumar, P. Samanta, A. V. Desai and S. K. Ghosh, *Chem. - Eur. J.*, 2016, **22**, 864-868.
9. A. Karmakar, B. Joarder, A. Mallick, P. Samanta, A. V. Desai, S. Basu and S. K. Ghosh, *Chem. Commun.*, 2017, **53**, 1253-1256.
10. Y. Liu, Q. Qiu, X. Zhang, K. Huang and D. Qin, *J. Solid State Chem.*, 2021, **300**, 122258.
11. D. Mahato, S. Fajal, P. Samanta, W. Mandal and S. K. Ghosh, *ChemPlusChem*, 2022, **87**, e202100426.