## SUPPORTING INFORMATION

## Guest water hinders sodium diffusion in low-defect Berlin green cathode material

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**Figure S1.** Cyclic voltammograms of BG-60 at different scan rates: 0.1, 0.25, 0.5, 0.75, and 1 mV/s in the voltage range of 2.6 - 3.8 V vs. Na/Na<sup>+</sup>. The peaks at higher and lower potentials are assigned to the redox reactions of low-spin Fe<sub>C</sub><sup>3+</sup>/Fe<sub>C</sub><sup>2+</sup> and high-spin Fe<sub>N</sub><sup>3+</sup>/Fe<sub>N</sub><sup>2+</sup> couples, respectively. As demonstrated by the 0.75 mV/s scan rate, the cut-off potential of 3.1 - 3.8 V vs. Na/Na<sup>+</sup> was implemented to only focus on the low-spin Fe<sub>C</sub><sup>3+</sup>/Fe<sub>C</sub><sup>2+</sup> and avoid the characteristic cubic–rhombohedral phase conversion which introduces strain and particle cracking due to the associated volume change.<sup>1,2</sup>



**Figure S2**. Selected charge/discharge profiles of BG-100 electrode/PB full cell at the current rate of 15 mA/g (1C = 150 mA/g) in potential range of 2.6 - 3.8 V.



**Figure S3.** Pawley fits<sup>3</sup> of XRD patterns ( $\lambda = 1.5406$  Å) of (a) uncycled, (b) mid-charge, (c) charged, and (d) discharged BG-150 electrode. The observed data is plotted in black, simulated pattern for the *Fm*  $\overline{}^{3}m$  model in red, and difference in blue. Vertical markers show the Bragg positions (BG electrode-pink, and Al-green). The observed Al phase comes from the Al foil used in cell fabrication. The lattice parameters are given in Table S1.



**Figure S4.** Center shift values for BG-150 electrode (n = 0.30(5)) as a function of nominal Na content x (red dots) and from earlier studies of Na<sub>x</sub>Fe[Fe(CN)<sub>6</sub>]nH<sub>2</sub>O. n = 0.54(5) for  $x \approx 0$ , and 1.79(2) for  $x \approx 0.5$  and 1.0 (black dots).<sup>4</sup> The lines are used as eye guides.

**Table S1.** Pawley fit results for BG-150 electrode (S.G.  $Fm^3m$ )

Electrode	Lattice parameters (Å)	Volume (Å <sup>3</sup> )
Uncycled	a = b = c = 10.240(1)	1073.6(1)
Mid-charge	a = b = c = 10.248(1)	1076.4(1)
Charged	a = b = c = 10.266(1)	1081.9(1)
Discharged	a = b = c = 10.237(1)	1072.9(1)

## References

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