Revealing the electrochemical performance of manganese phosphite/RGO hybrid in acidic media

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Fig. S1 XPS survey spectra of Mn-HPO/RGO hybrid.

Element	Weight %
Mn	23.6
Р	8.9
0	45.5
С	22



Fig. S2 EDX spectra of the Mn-HPO/RGO hybrid with atomic percentage of elements.



Fig. S3 SEM images of Mn-HPO/RGO synthesized with 5 mg and 20 mg GO



Fig. S4 Charge-discharge curve for 1st and 5000th cycle along with the morphology of the active material before and after cyclic stability



Fig. S5 CV curves at 5 mV/s and GCD curves at 1 A/g for Mn-HPO/RGO-5, Mn-HPO/RGO-10 and Mn-HPO/RGO-20.



Fig. S6 EIS spectra of (a) Mn-HPO and (b) Mn-HPO/RGO-10 with fitted data.



Fig. S7 CV and GCD curves MXene (Ti₃C₂).



Fig. S8 Raman spectra of the fresh electroactive material and the electroactive material after stability

Table S1 Comparison of supercapacitor performance of Mn-HPO/RGO with previous reports

Name of the	Specific	Electrolyte	Rate	Specific	Energy	Cyclic	References
Material	Capacitance		performance	Capacitanc	Density	Stability	
	(F/g)		(in %)	e of the	of the		
				device	Device		
				(F/g)	(Wh/kg)		
$MnO_x@C@$	350	6 M KOH	34.8 %	53.4	23	94 % up	1
MnO_x						to 2000	
						cycles	
о <i>v</i> -	452.4	1 M Na ₂ SO ₄	69.9 %	90.8	40.2	92.2 %	2
MnO ₂ @Mn						up to	
O2						10000	
						cycles	
Ti ₃ C ₂ T _x	-	1 M Na ₂ SO ₄	-	23.3	8.3	-	3
(MXene)–δ							
-MnO ₂ ASC							

Mn ₃ (PO ₄) ₂ /GF	270	6 M KOH	-	28	7.6	96 % up to 10000	4
						cycles	
Mn ₃ (PO ₄) ₂	203	1 M Na ₂ SO ₄	88 %	46.8	16.64	90 % up	5
						to 10000	
						cycles	
Mn3(PO4)2	194	2 M KOH	85 %	41.9	14.89	90 % up	5
						to 10000	
						cycles	
NH4MnPO4	423	3 M KOH	65 %	65.4	29.4	93 % up	6
\cdot H ₂ O						to 100k	
						cycles	
Amorphous	912.4	1 M Na ₂ SO ₄	66.6 %	205.59	126	95 % up	7
Manganese		+ PBS				to 5000	
phosphate						cycles	
Mn-	770	1 M H ₂ SO ₄	67 %	108	34	94 % up	This work
HPO/RGO						to 12000	
hybrid						cycles	

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