

Supporting information

for

**Oxidative Addition or Werner Coordination Complex? Reactivity of β -
diketiminato Supported Main Group and First-row Transition Metal
Complexes towards Ammonia**

by

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Table S1. Comparison of selected bond parameters of reported nacnacM compounds.

		QAMXOR (M = Al)	XEDHOD (M = Ga)	BICXAN (M = In)	CEPHAH (M = Si)	GEMDAE (M = Ge)	XUGQOG (M = Si)	HOVGOP (M = Ge)
M-N BOND LENGTH (Å)	Experimental	1.9931	2.0793	2.3077	1.735	1.865	1.727	2.0297
		1.9882	2.0793	2.3072	1.734	1.866	1.724	2.0296
	Calculated						1.653	1.8448
		1.9576	2.0528	2.2756	1.736	1.852	1.737	2.0297
		1.9570	2.0560	2.2682	1.750	1.869	1.728	2.0296
							1.699	1.8449
N-M-N ANGLE (°)	Experimental	88.5	86.6	80.7	99.3	95.8	103.1	88.6
	Calculated	89.9	87.5	81.1	98.6	95.8	103.1	88.6
M-N-C-C ANGLE (°)	Experimental	-5.3	0.0	-2.3	-0.8	2.4	7.8	-95.4
		3.8	-0.0	2.3	-1.2	0.1	-11.0	
	Calculated	1.9	2.1	2.3	0.0	0.0	-3.9	-95.4
		1.53695	1.7	0.9	0.0	0.0	-5.5	
REFERENCE		1	2	3	4	5	6	7

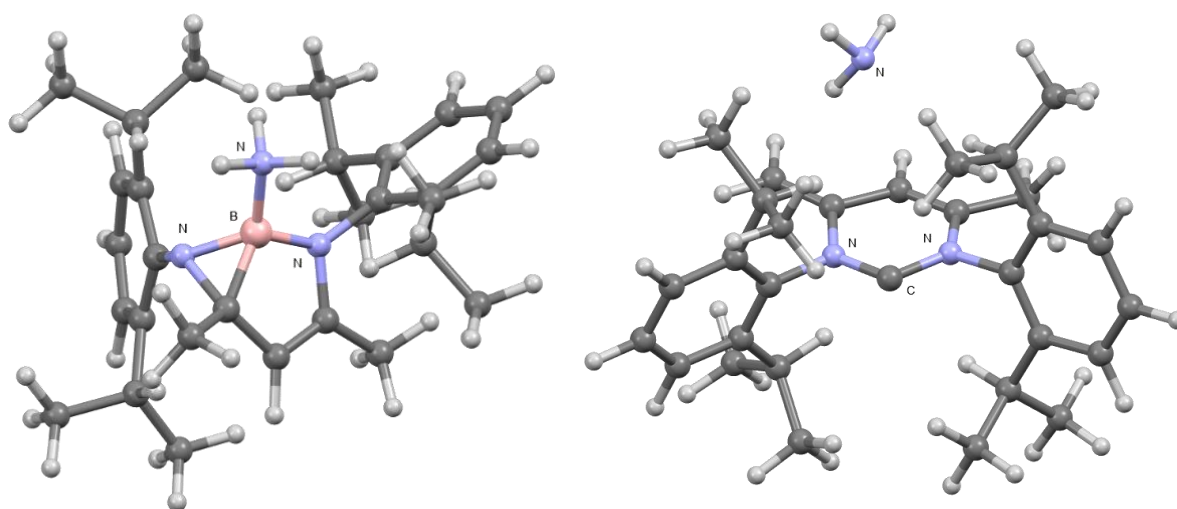


Figure S1. Optimised structure of NaCNacB·NH₃ (left) and NaCNacC·NH₃ (right).

Table S2. Comparison of group 14 NacnacM relative stabilities of 1,1- and 1,4-oxidative addition reaction products. The differences to free starting materials NacnacM and NH₃ are given in Gibbs free energies (kJ/mol).

Compound	1,1-oxidative addition (kJ/mol)	1,4-oxidative addition (kJ/mol)
NacnacC	-67.3	42.5
NacnacSi	-126.6	-48.1
NacnacGe	10.1	-46.8
NacnacSn	95.5	-70.1
NacnacPb	242.2	-66.2

Table S3. EDA-NOCV derived most important deformation densities for the interaction between NacnacM and NH₃ fragments in NacnacM·NH₃ complexes.

Compound	Orbital interaction (kJ/mol)	Percentage of all orbital interactions (%)
NacnacFe	-50.1	38
NacnacCo	-55.8	46
NacnacNi	-94.0	62
NacnacCu	-64.1	58
NacnacAl	-98.1	72
NacnacGa	-47.6	71
NacnacIn	-41.1	68
NacnacSi	-180.9	77
NacnacGe	-134.8	77
NacnacSn	-99.2	74
NacnacPb	-78.1	72

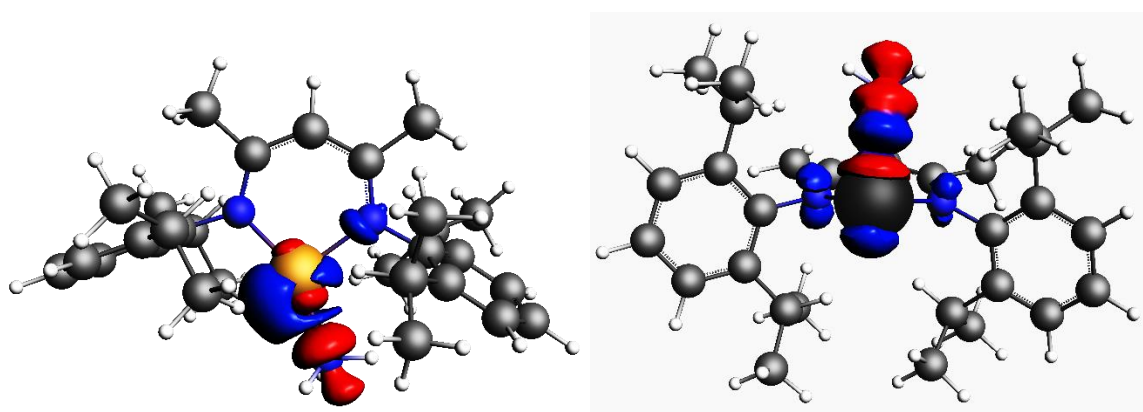


Figure S2. The most important EDA-NOCV deformation density of NacnacCu·NH₃ (left) and NacnacPb·NH₃ (right). Isovalue set at ± 0.002 a.u. and red contour corresponds to the depletion of electron density and blue contour the accumulation of electron density.

References

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