

Supplementary Information

A high- T_C heavy rare earth monoxide semiconductor TbO with more than half-filled 4f orbital

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Table S1 Growth parameters of TbO, Tb₂O₃, and Tb thin films.

Sample	Substrate	Target	$T_{\text{substrate}}$ [°C]	P_{O_2} [Torr]	E_{laser} [J/cm ²]	Repetition rate [Hz]
TbO	CaF ₂ (001)	Tb ₄ O ₇	375	-	0.75	15
Tb ₂ O ₃	YAlO ₃ (110)	Tb ₄ O ₇	250	5.0×10^{-6}	0.50	5
Tb	CaF ₂ (111)	Tb	R.T.	-	0.70	20

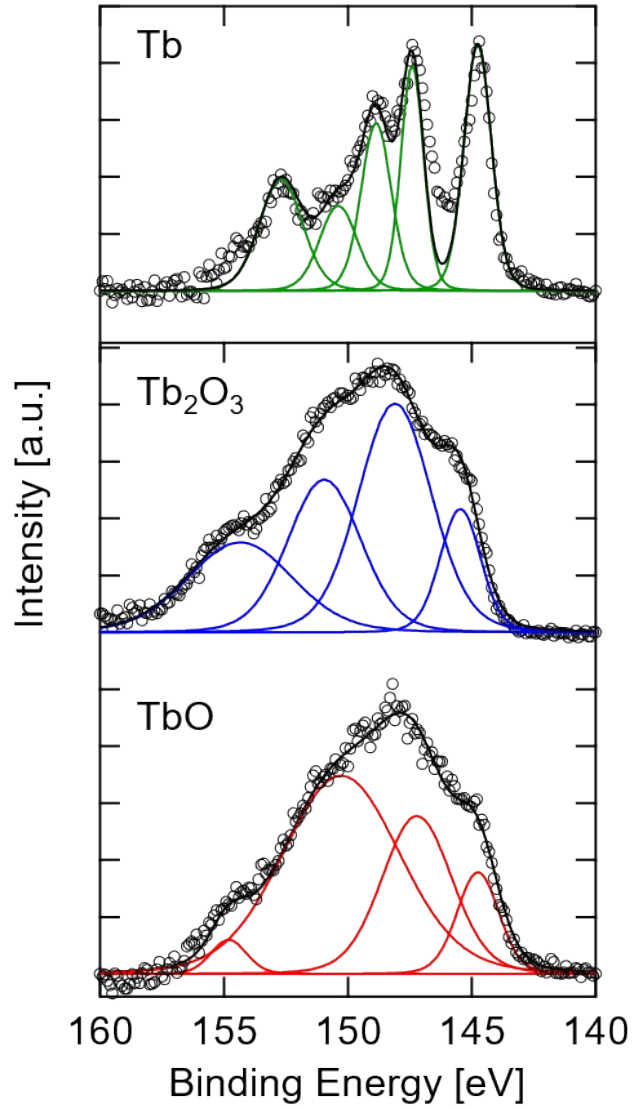


Fig. S1 Tb 4d XPS spectra of TbO, Tb₂O₃ and Tb thin films. The atomic ratio of Tb:O in TbO thin film was evaluated to be 1:0.95, as described below. Prior to the measurements, capping or surface oxidation layer was removed by in-situ Ar ion sputtering.

Quantitative analysis of TbO layer in TbO thin film by XPS

The composition of a TbO thin film was determined from the areal intensity ratio of Tb 4d and O 1s peaks by XPS measurements. The surface oxidized Tb₂O₃ layer of the TbO thin film was removed by Ar⁺ sputtering. For the quantitative composition analysis, a Tb₂O₃ thin film was used as a reference, assuming its stoichiometric composition.

The Tb:O ratio was evaluated by considering the electron inelastic mean free paths (IMFP) with the TPP-2M predictive equation.^{S1} The atomic concentration of each atom C_i ($i = \text{Tb, O}$) in TbO was obtained by the areal intensity of each atom in TbO and Tb₂O₃ ($I_i^{\text{TbO}}, I_i^{\text{Tb2O3}}$), the IMFP of each atom in TbO and Tb₂O₃ ($\lambda_i^{\text{TbO}}, \lambda_i^{\text{Tb2O3}}$), and the atomic density of each atom in TbO and Tb₂O₃ ($N_i^{\text{TbO}}, N_i^{\text{Tb2O3}}$) as follows, where K_i is relative intensity $K_i = I_i^{\text{TbO}}/I_i^{\text{Tb2O3}}$.^{S2}

$$\begin{aligned} C_i &= (I_i^{\text{TbO}}/I_i^{\text{Tb2O3}})/\{(N_i^{\text{TbO}}/N_i^{\text{Tb2O3}}) \times (\lambda_i^{\text{TbO}}/\lambda_i^{\text{Tb2O3}})\} \\ &= (K_i)/\{(N_i^{\text{TbO}}/N_i^{\text{Tb2O3}}) \times (\lambda_i^{\text{TbO}}/\lambda_i^{\text{Tb2O3}})\} \\ &= (K_i) \times \{(N_i^{\text{Tb2O3}}) \times (\lambda_i^{\text{Tb2O3}}/\lambda_i^{\text{TbO}})\}/N_i^{\text{TbO}} \end{aligned}$$

The relative intensity (K_i) of Tb and O was calculated to be $K_{\text{Tb}} = 1.01$ and $K_{\text{O}} = 0.596$. Because of unknown N_i^{TbO} , the relative atomic concentration (C_i^{rel}) was introduced with the matrix factor, $F_i = N_i^{\text{Tb2O3}} \times (\lambda_i^{\text{Tb2O3}}/\lambda_i^{\text{TbO}})$.^{S2}

$$C_i^{\text{rel}} = C_i / \sum C_i = (K_i F_i / N_i^{\text{TbO}}) / \sum (K_i F_i / N_i^{\text{TbO}}) = (K_i F_i) / \sum (K_i F_i)$$

From the TPP-2M IMFP equation, $\lambda_{\text{Tb}}^{\text{TbO}} = 5.68 \text{ \AA}$ and $\lambda_{\text{O}}^{\text{TbO}} = 7.98 \text{ \AA}$, while $\lambda_{\text{Tb}}^{\text{Tb2O3}} = 7.65 \text{ \AA}$ and $\lambda_{\text{O}}^{\text{Tb2O3}} = 11.5 \text{ \AA}$. In the referential Tb₂O₃ thin film, $N_{\text{Tb}}^{\text{Tb2O3}} = 0.0496 \text{ mol/cm}^3$ and $N_{\text{O}}^{\text{Tb2O3}} = 0.0596 \text{ mol/cm}^3$. Thus, the matrix factor $F_{\text{Tb}} = 0.0537$ and $F_{\text{O}} = 0.0861$. Thus,

$$C_{Tb}^{rel} = (K_{Tb}/F_{Tb}) / (K_{Tb} \times F_{Tb} + K_O \times F_O) = 0.513,$$

$$C_O^{rel} = (K_O/F_O) / (K_{Tb} \times F_{Tb} + K_O \times F_O) = 0.486.$$

Therefore, the Tb:O ratio of TbO thin film, $C_{Tb}^{rel}:C_O^{rel}$, was 1:0.95.

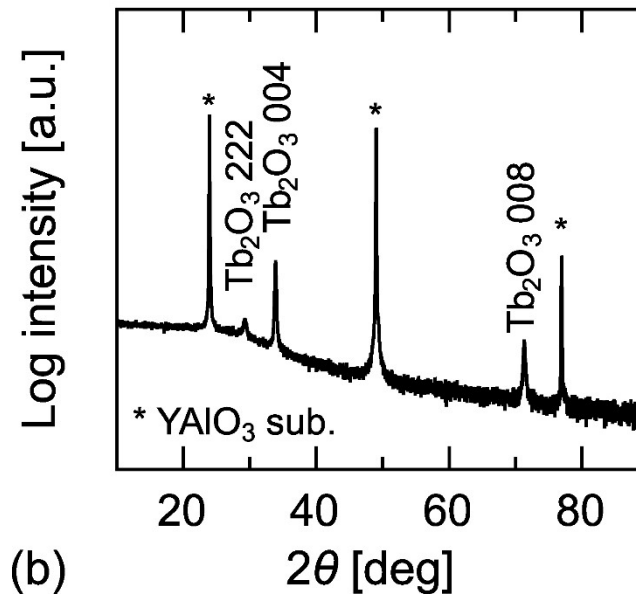
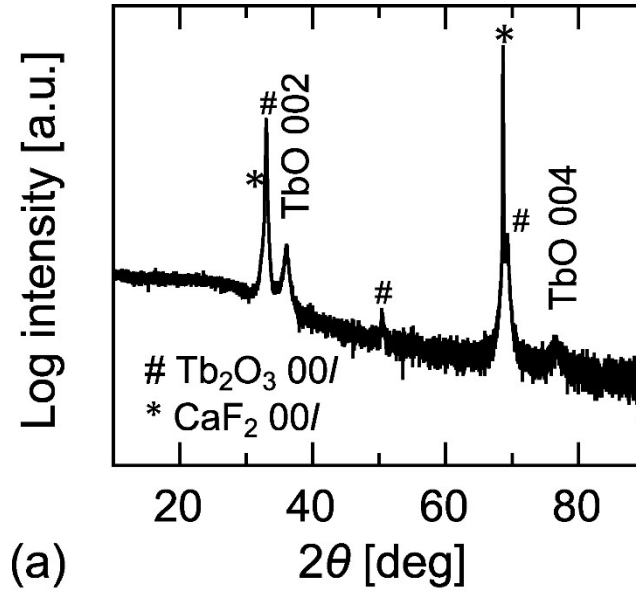
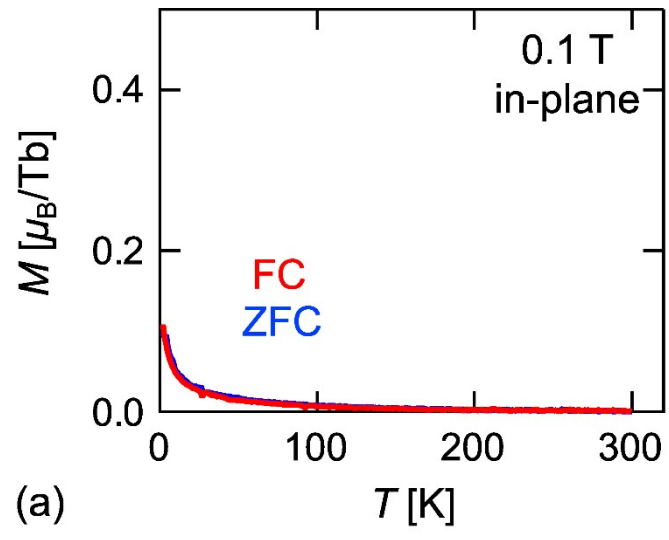
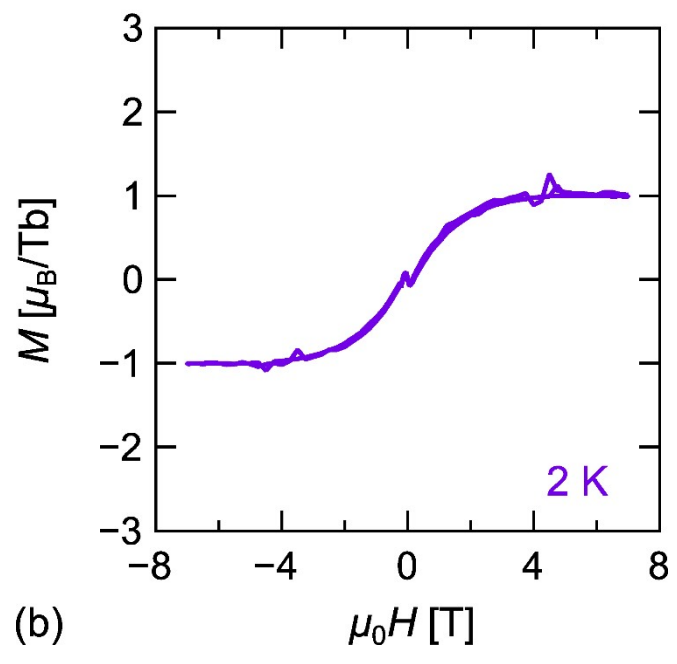


Fig. S2 Typical XRD θ - 2θ patterns of (a) TbO thin film grown on CaF_2 (001) substrate (the same data as Fig. 1a) and (b) Tb_2O_3 thin film grown on YAlO_3 (110) substrate.



(a)



(b)

Fig. S3 (a) Temperature and (b) magnetic field dependence of magnetization for Tb_2O_3 (001) epitaxial thin film under in-plane magnetic field.

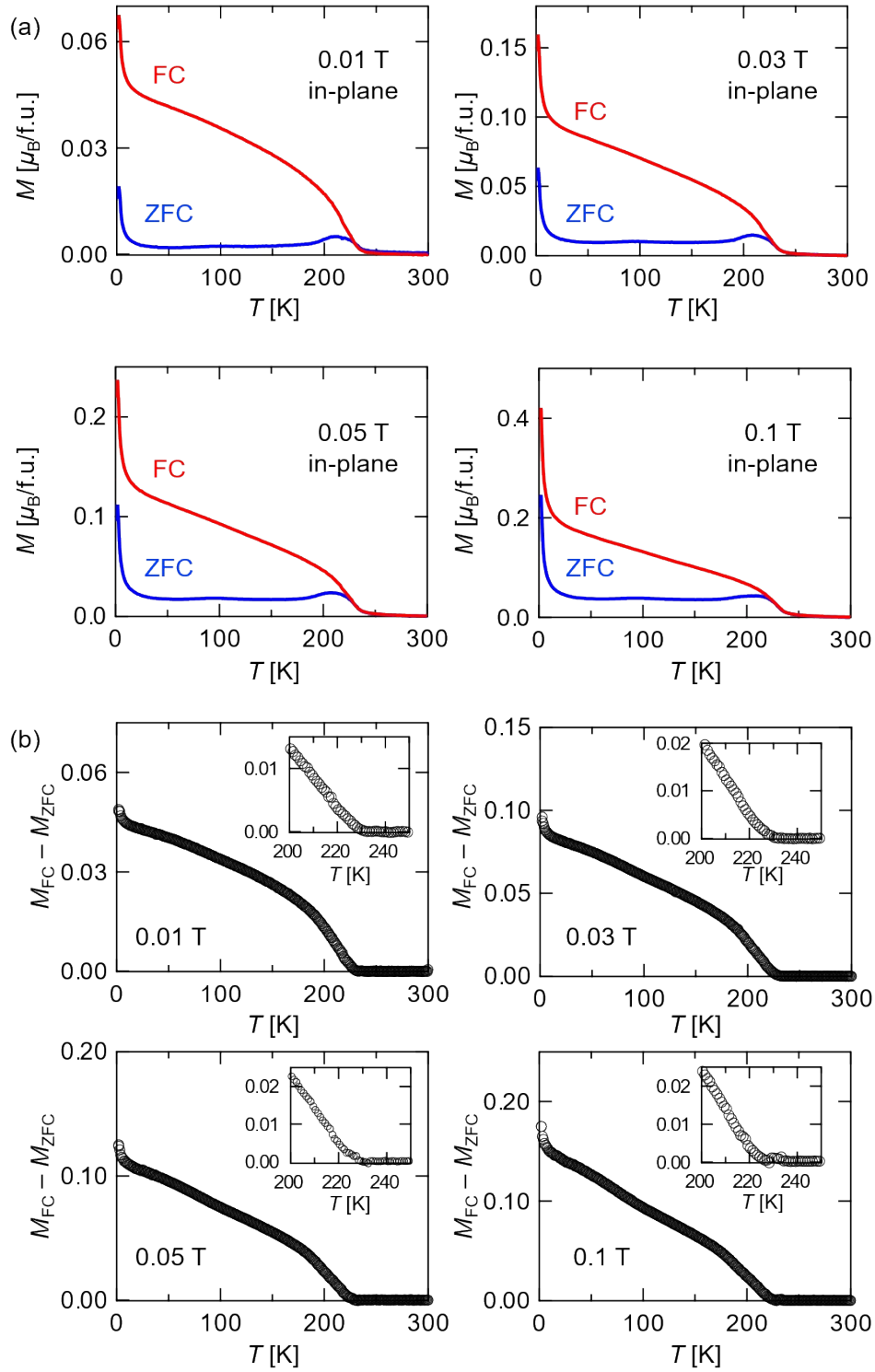


Fig. S2 (a) In-plane M - T curves during ZFC and FC for TbO epitaxial thin film under different magnetic field. (b) The difference between the FC and ZFC curves. Inset shows magnified view around T_C .

References

- ^{S1} S. Tanuma, C.J. Powell and D.R. Penn, Calculation of electron inelastic mean free paths (IMFPs) VII. Reliability of the TPP-2M IMFP predictive equation. *Surf. Interface Anal.*, 2003, **35**, 268–275.
- ^{S2} S. Tanuma, Quantitative Surface Analysis by X-ray Photoelectron Spectroscopy. *J. Surf. Anal.*, 1998, **4**, 20–34.