Two Birds with One Stone: High Catalytic Areas and Abundant Nitrogen Sites Inspired by Fluorine doping contributing to CO₂RR Activity and Selectivity

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Fig. S1. SEM image of CB.



Fig. S2. SEM image of N-CB.



Fig. S3. SEM image of F-CB.



Fig. S4. SEM images of N,F-CB.



Fig. S5. High-resolution XPS spectra of C 1s of (a) N,F-CB and (b) N-CB.



Fig. S6. LSV curves of N,F-CB in CO₂- or Ar-saturated 0.1 M KHCO₃ solutions.



Fig. S7. Faradaic efficiencies of products for the (a) N,F-CB, (b) N-CB, (c) F-CB, (d) CB.



Fig. S8. The CV curves of (a) N,F-CB, (b) F-CB, (c) N-CB, and (d) CB measured in 0.1 M KHCO₃ solutions saturated with CO₂ at different scan rate from 2 to 50 mV s⁻¹ in the potential ranges of 0 V ~ -0.10 V vs. RHE.



Fig. S9. The pore size distribution curves of N,F-CB, N-CB, F-CB, and CB.

	N-CB	N,F-CB
N (at. %) of XPS	0.52	3.27
Pyridinic N	34.37%	32.52%
Graphitic N	38.57%	39.34%
Pyrrolic N	21.69%	15.60%
Oxidized N	5.37%	12.54%

 Table S1. The N contents of catalysts summary from XPS analysis.

Electrocatalyst	Electrolyte (KHCO ₃)	Potentials (V vs. RHE)	FE _{CO} (%)	jco (mA cm ⁻²)	Ref.
N,F-CB	0.1 M	-0.7	93	4.6	This work
Au-N-C	0.1 M	-0.5	80	6	S1
NC(NH ₃)	0.1 M	-0.5	83	~	S2
NS-CNSs-1000	0.5 M	-0.55	85.4	2.5	S3
MNC-D	0.1 M	-0.58	92	6.8	S4
A-350-1000	0.1 M	-0.68	89	0.8	S5
NRMC-1000	0.1 M	-0.7	80	2.9	S6
NCNT-NH ₃	0.5 M NaHCO ₃	-0.8	96.5	~13	S7
NCNTs	0.1 M	-0.8	80	-0.8	S8
N-graphenes	0.5 M	-0.84	73 (Form ate)	7.0	S9
NCNT-3-700	0.5 M NaHCO ₃	-0.9	90	5.0	S10
NC-900	0.1 M	-0.93	78	0.8	S11
NCNTs-ACN- 850	0.1 M	-1.05	80	~	S12
g-C ₃ N ₄ / MWCNTs	0.1 M	-0.75	60	0.55	S13
BAX-M-950	0.1 M	-0.66	40	1.5	S14

Table S2. Summary of the reported nitrogen doped carbon electrocatalysts in CO_2RR .

Catalysts	C _{dl} (mF cm ⁻²)	R _f
СВ	1.68	1
N-CB	2.05	1.22
F-CB	5.22	3.11
N,F-CB	11.56	6.88

 Table S3. The double layer capacitance of different catalysts and the corresponding normalized roughness factor.

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