Supporting information

An efficient and eco-compatible multicomponent synthesis of 2,4,5trisubstituted imidazole derivatives via modified-silica coated cobalt ferrite nanoparticles by Tungstic acid

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General synthetic process for tri-substituted imidazole derivatives

Accurate amounts of benzil (1 mmol), benzaldehyde (1 mmol), ammonium acetate (2 mmol) and CoFe₂O₄@SiO₂@(-CH₂)₃OWO₃H (0.01 g) were stirred under neat condition at 110 °C for 15-50 min. After the reaction was supplemented (TLC), the mixture was dissolved in ethanol and the catalyst was separated by an external magnet, washed several times with acetone and dried at 100°C for 5 h in the oven for reusing in subsequent cycles. Eventually, few drops of ice-cold water were added to the remaining solution and the precipitate was washed by petroleum ether and recrystallized from ethanol or acetone-water mixture, if needed, to give pure target products. Every single product was verified by reconciling its melting point in literatures and using spectroscopic methods such as FT-IR and ¹H NMR.

FT-IR and ¹H NMR Spectra of synthesized 2,4,5-trisubstituted imidazole derivatives:

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2,4,5-triphenylimidazole (4a)	. 2
2-(4-fluorophenyl)-4,5-diphenylimidazole (4b)	. 3
2-(4-chlorophenyl)-4,5-diphenylimidazole (4c)	4
2-(2-chlorophenyl)-4,5-diphenylimidazole (4d)	5
2-(2,4-dichlorophenyl)-4,5-diphenylimidazole (4e)	6
2-(4-bromophenyl)-4,5-diphenylimidazole (4f)	7
2-(4-Hydroxyphenyl)-4,5-diphenylimidazole (4g)	8
2-(3-Hydroxyphenyl)-4,5-diphenylimidazole (4h)	9
2-(4-methoxyphenyl)-4,5-diphenylimidazole (4i) 1	10
2-(2-nitrophenyl)-4,5-diphenylimidazole (4j)	11
2-(3-nitrophenyl)-4,5-diphenylimidazole (4k)	12
2-(4-nitrophenyl)-4,5-diphenylimidazole (41)	13



FT-IR of 4a



FT-IR of 4b



FT-IR of 4c



FT-IR of 4d



¹H NMR of **4e**



FT-IR of 4e



 1 H NMR of **4f**



FT-IR of 4f



¹H NMR of 4g



FT-IR of 4g



FT-IR of 4h



¹H NMR of **4i**



FT-IR of 4i



¹H NMR of **4**j



FT-IR of 4j



 1 H NMR of **4**k



FT-IR of 4k



 1 H NMR of **4**l



FT-IR of 4l