

Electronic Supplementary Information

Low-content SnO₂ nanodots on N-doped graphene: lattice-confinement preparation and high-performance lithium/sodium storage

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Figure S1

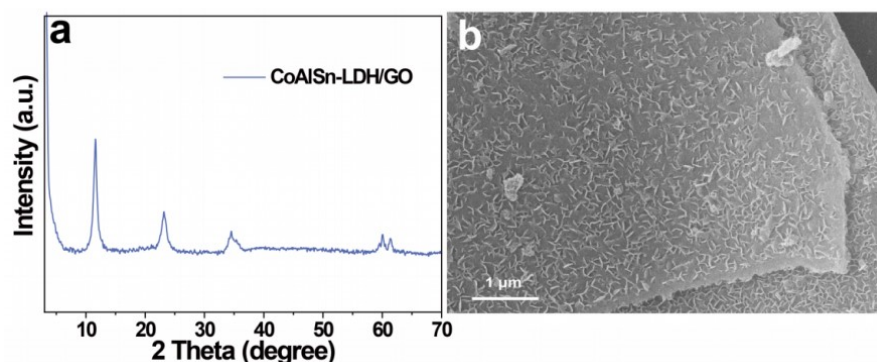


Figure S1. (a) XRD spectrum of CoAlSn-LDH/GO precursor, (b) SEM image of CoAlSn-LDH/GO precursor.

Figure S2

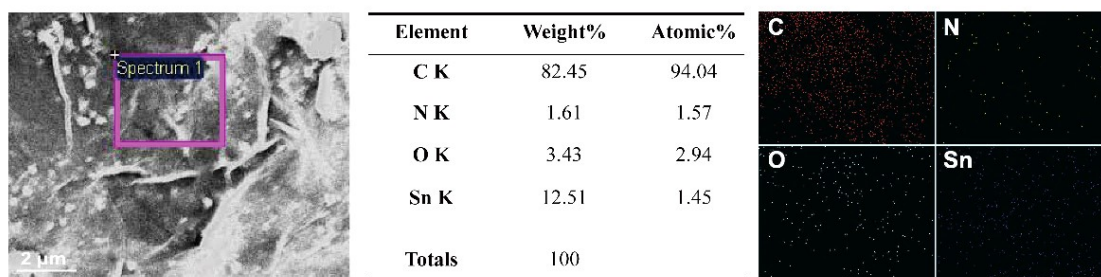


Figure S2. The SEM/EDS element mapping of SnO₂@N-rGO.

Figure S3

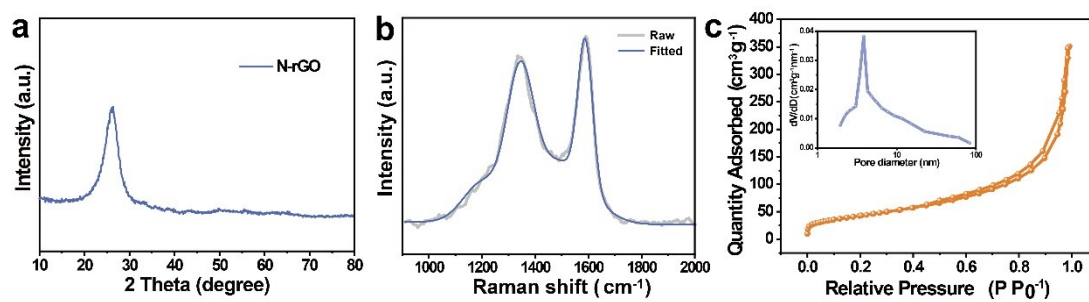


Figure S3. The N-rGO counterpart prepared for comparison: (a) XRD pattern, and (b) Raman spectrum, and (c) N₂ adsorption/desorption isotherm curve (Insert shows a mesoporous size distribution).

Figure S4

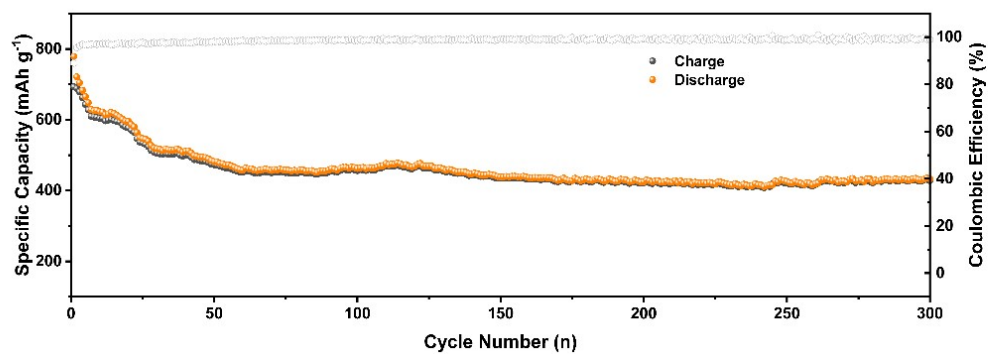


Figure S4. Long-cycling performance of SnO₂@N-rGO at 2000 mA g⁻¹ for LIBs.

Figure S5

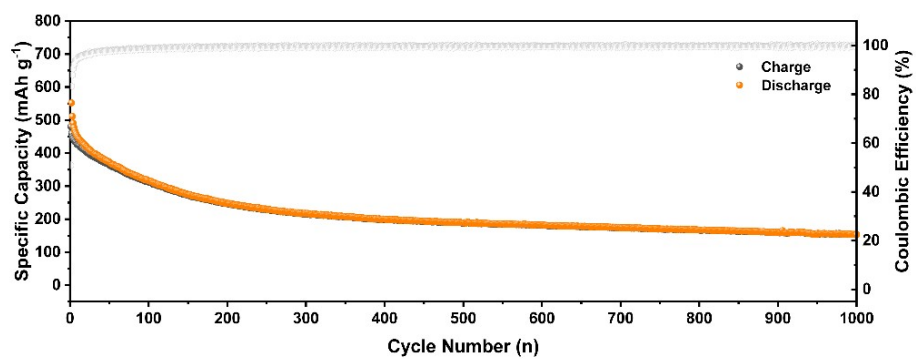


Figure S5. Long-cycling performance of SnO₂@N-rGO at 2000 mA g⁻¹ for SIBs.

Figure S6

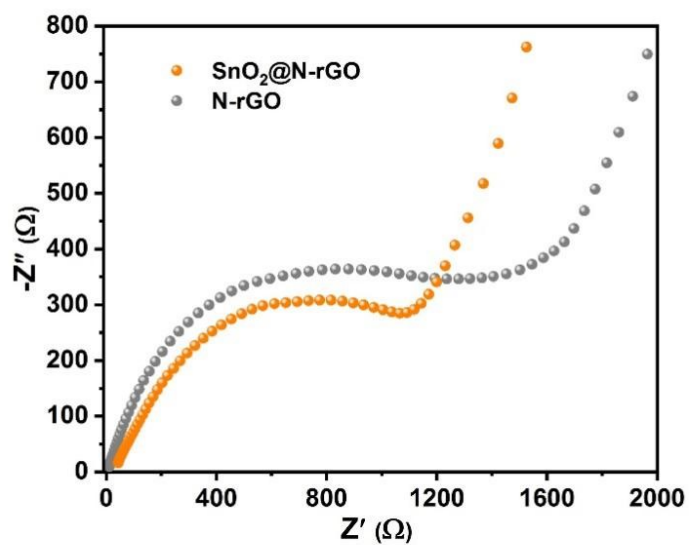


Figure S6. Comparison of EIS between the SnO₂@N-rGO composite and the N-rGO counterpart for SIBs.

Figure S7

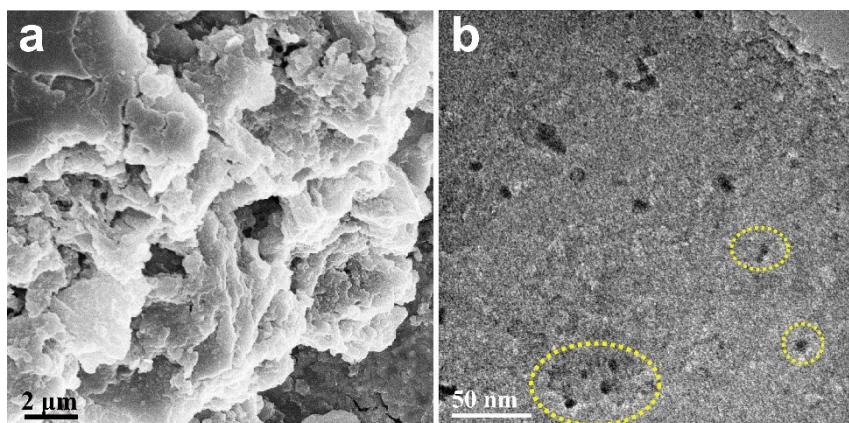


Figure S7. (a) SEM and (b) TEM image of the post-cycled $\text{SnO}_2@\text{N-rGO}$ electrode after 100 cycles at 0.1 A g^{-1} for SIBs, without significant aggregation or volume expansion, as marked by the dotted line circles.

Table S1.

Comparison of cycling performance between SnO₂@N-rGO and the reported SnO₂-based anode nanomaterials for LIBs.

SnO ₂ -based materials	Current density /mA g ⁻¹	Specific capacity /mAh g ⁻¹	Cycles	References
SnO ₂ /carbon@-void@carbon (SnO ₂ : 45.3 wt%)	500	866	200	1
SnO ₂ NPs	1000	887	1000	2
SnO ₂ /graphene (SnO ₂ : 54 wt%)	100	1420	90	3
SnO ₂ /GNP (SnO ₂ : 80 wt%)	100	745	100	4
SnO ₂ /MXenes	100	904.1	1000	5
NC@SnO ₂ (SnO ₂ : 67.81wt%)	1000	750	100	6
SnO ₂ @P@GO (SnO ₂ : 82.18 wt%)	100	550	200	7
SnO ₂ @C (SnO ₂ : 91.77 wt%)	50	725	200	8
SnO ₂ @CNFs (SnO ₂ : 18.1 wt%)	50	380.4	100	9
SnO ₂ @N-rGO (SnO ₂ : 17.9 wt%)	100	1146.2	100	This work
	2000	428.5	300	

Table S2.

Comparison of cycling performance between SnO₂@N-rGO and the reported SnO₂-based anode nanomaterials for SIBs.

SnO ₂ -based materials	Current density /mA g ⁻¹	Specific capacity /mAh g ⁻¹	Cycles	References
NBT/C@SnO ₂ NFs	200	420.7	500	10
SnO ₂ -NG (SnO ₂ : 50 wt%)	50	409.6	100	11
SnO ₂ @NC (SnO ₂ : 55.5 wt%)	1000	212.6	3000	12
SnO ₂ /rGO (SnO ₂ : 90.71 wt%)	200	204	1500	13
SnO ₂ /CNT (SnO ₂ : 72 wt%)	100	630.4	100	14
SnO _{2-x} /C nanofibers (SnO ₂ : 54 wt%)	1000	565	2000	15
N-C@SnO ₂ (SnO ₂ : 67.81 wt%)	100	270	100	6
SnO ₂ /graphene (SnO ₂ : 54 wt%)	200	650	90	3
PCNF@SnO ₂ @C (SnO ₂ : 38.5 wt%)	50	374	100	16
SnO ₂ -PC (SnO ₂ : 74.47 wt%)	100	280.1	250	17
SnO ₂ @NC-rGO (SnO ₂ : 17.9 wt%)	100 2000	387 150	100 1000	This work

References

1. Y. Li, K. Lin, X. Qin, K. Zeng, Y. Liu, Y. Xia, F. Lv, H. Zhu, F. Kang and B. Li, *Carbon*, 2021, **183**, 486-494.
2. Y. Wang, N. Jiang, D. Pan, H. Jiang, Y. Hu and C. Li, *Chem. Eng. J.*, 2022, **437**, 135422.
3. W. Chen, K. Song, L. Mi, X. Feng, J. Zhang, S. Cui and C. Liu, *J. Mater. Chem. A*, 2017, **5**, 10027-10038.
4. M. Palanisamy, C. Jamison, X. Sun, Z. Qi, H. Wang and V. G. Pol, *Carbon*, 2021, **185**, 608-618.
5. C. Zhao, Z. Wei, J. Zhang, P. He, X. Huang, X. Duan, D. Jia and Y. Zhou, *J. Alloys Compds.*, 2022, **907**, 164428.
6. J. Liang, C. Yuan, H. Li, K. Fan, Z. Wei, H. Sun and J. Ma, *Nano-Micro Lett.*, 2017,

- 10**, 21.
7. L. Zhang, K. Zhao, R. Yu, M. Yan, W. Xu, Y. Dong, W. Ren, X. Xu, C. Tang and L. Mai, *Small*, 2017, **13**, 1603973.
 8. A. A. Ambalkar, R. P. Panmand, U. V. Kawade, Y. A. Sethi, S. D. Naik, M. V. Kulkarni, P. V. Adhyapak and B. B. Kale, *New J. Chem.*, 2020, **44**, 3366-3374.
 9. Z. Huang, H. Gao, Q. Wang, Y. Zhao and G. Li, *Mate. Lett.*, 2017, **186**, 231-234.
 10. L. Wang, C. Lin, G. Yang, N. Wang and W. Yan, *Electrochim. Acta*, 2022, **411**, 140049.
 11. L. Fan, X. Song, D. Xiong and X. Li, *J. Electroanal. Chem.*, 2019, **833**, 340-348.
 12. Y. Cheng, S. Wang, L. Zhou, L. Chang, W. Liu, D. Yin, Z. Yi and L. Wang, *Small*, 2020, **16**, 2000681.
 13. Z. Kong, X. Liu, T. Wang, A. Fu, Y. Li, P. Guo, Y.-G. Guo, H. Li and X. S. Zhao, *Appl. Surface Sci.*, 2019, **479**, 198-208.
 14. J. Cui, Z.-L. Xu, S. Yao, J. Huang, J.-Q. Huang, S. Abouali, M. A. Garakani, X. Ning and J.-K. Kim, *J. Mater. Chem. A*, 2016, **4**, 10964-10973.
 15. D. Ma, Y. Li, H. Mi, S. Luo, P. Zhang, Z. Lin, J. Li and H. Zhang, *Angew. Chem. Int. Ed.*, 2018, **57**, 8901-8905.
 16. M. Dirican, Y. Lu, Y. Ge, O. Yildiz and X. Zhang, *ACS Appl. Mater. Interfaces*, 2015, **7**, 18387-18396.
 17. Z. Huang, H. Hou, G. Zou, J. Chen, Y. Zhang, H. Liao, S. Li and X. Ji, *Electrochim. Acta*, 2016, **214**, 156-164.