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Light-induced O<sub>2</sub>-dependent aliphatic carbon-carbon (C-C) bond cleavage in bipyridine-ligated Co(II) chlorodiketonate complexes

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	8 • O(CH <sub>2</sub> CH <sub>3</sub> ) <sub>2</sub>	9 • 0.5 CH <sub>3</sub> CN • 0.5 O(CH <sub>2</sub> CH <sub>3</sub> ) <sub>2</sub>	10	12
Empirical formula	C <sub>39</sub> H <sub>36</sub> Cl <sub>2</sub> CoN <sub>4</sub> O <sub>7</sub>	$C_{40}H_{36.5}CI_2CoN_{4.5}O_{6.5}$	$C_{37}H_{30}Cl_2CoN_4O_8$	$C_{24}H_{22}CI_2C0N_6O_8$
Formula weight	802.55	814.06	788.48	652.30
Crystal system	Triclinic	Orthorhombic	Monoclinic	Triclinic
Space group	P-1	Pna2₁	P2₁/n	P-1
a (Å)	11.3963(6)	21.4753(12)	13.8215(7)	10.0272(8)
b (Å)	12.6001(7)	17.7824(9)	18.0740(10)	10.2270(6)
c (Å)	14.2773(7)	19.6340(12)	14.7453(7)	15.4354(6)
α (deg)	73.724(2)	90	90	83.304(4)
β (deg)	86.205(2)	90	110.426(2)	83.219(4)
γ (deg)	69.356(2)	90	90	61.217(7)
V (Å <sup>3</sup> )	1840.40(17)	7497.9(7)	3451.9(3)	1372.19(17)
Z	2	8	4	2
$\rho_{calc}$ (g/cm <sup>3</sup> )	1.448	1.442	1.517	1.579
Temp (K)	100	100	100	100
Crystal size	0.28 × 0.226 ×	0.24 × 0.21 × 0.20	0.26 × 0.25 ×	0.24 × 0.19
(mm <sup>-1</sup> )	0.11		0.02	x0.12
Diffractometer	Bruker D8	Bruker D8	Bruker D8	Rigaku XtaLAB
	Venture	Venture	Venture	Mini II
				Diffractometer
Abs. coeff. (mm <sup>-1</sup> )	0.668	0.656	0.712	0.879
20 (deg)	54.298	55.162	55.1	60.234
Reflections collected	87249	96371	54870	25701
Indep. Reflections	8151	17244	7949	8973
<i>R</i> 1 / w <i>R</i> 2 <sup>b</sup>	0.0411 / 0.1019	0.0356 / 0.0837	0.0352 / 0.0736	0.0363 / 0.0885
GOF ( <i>F</i> <sup>2</sup> )	1.047	1.032	1.027	1.042
$\Delta \rho_{max} / \Delta \rho_{min}$ (e Å <sup>-3</sup> )	1.0 / -0.7	0.61 / -0.26	0.58 / -0.41	0.47 / -0.49

Table S1. Summary of X-ray data collection and refinement.<sup>a</sup>

<sup>*a*</sup> Radiation used: MoKα ( $\lambda = 0.71073$  Å). <sup>*b*</sup> R1 =  $\sum ||F_0| - |F_c|| / \sum |F_0|$ ; wR2 =  $[\sum [w(F_0^2 - F_c^2)^2] / [\sum (F_0^2)^2]]^{1/2}$  where  $w = 1/[\sigma^2(F_0^2) + (aP)^2 + bP]$ .

Table 52. Selected bond distances (A) for 8-10				
	8	9A	9B	10
Co(1)-O(1)	2.0596(14)	2.034(3)	2.024(3)	2.0577(13)
Co(1)-O(2)	2.0437(15)	2.030(2)	2.027(2)	2.0286(13)
Co(1)-N(1)	2.1333(17)	2.146(3)	2.134(3)	2.1544(16)
Co(1)-N(2)	2.1251(17)	2.152(3)	2.139(3)	2.1040(15)
Co(1)-N(3)	2.1136(17)	2.114(3)	2.110(3)	2.1307(16)
Co(1)-N(4)	2.1321(18)	2.109(3)	2.106(3)	2.1269(16)
O(1)-C(22)	1.267(2)	1.260(4)	1.280(4)	1.272(2)
O(2)-C(21)	1.265(3)	1.262(4)	1.262(4)	1.267(2)
C(21)-C(23)	1.404(3)	1.412(5)	1.420(5)	1.413(3)
C(22)-C(23)	1.414(3)	1.406(5)	1.411(5)	1.412(3)

Table S2. Selected bond distances (Å) for 8-10

Table S3. Selected bond angles (deg) for 8-10

	8	9A	9B	10
O(1)-Co(1)-N(1)	171.12(6)	168.96(10)	86.70(10)	162.39(6)
O(1)-Co(1)-N(2)	96.60(6)	92.73(11)	170.22(11)	93.02(6)
O(1)-Co(1)-N(3)	92.08(6)	97.44(11)	94.03(12)	91.67(6)
O(1)-Co(1)-N(4)	87.37(6)	89.61(11)	93.29(12)	94.73(5)
O(2)-Co(1)-O(1)	86.02(6)	86.70(10)	92.88(11)	85.80(5)
O(2)-Co(1)-N(1)	87.80(6)	89.36(10)	89.47(11)	81.29(6)
O(2)-Co(1)-N(2)	89.04(6)	87.42(11)	89.19(11)	97.13(6)
O(2)-Co(1)-N(3)	95.97(6)	93.09(11)	94.63(11)	90.26(6)
O(2)-Co(1)-N(4)	170.14(6)	168.98(11)	171.70(11)	167.23(6)
N(2)-Co(1)-N(1)	76.91(7)	76.79(12)	76.90(12)	76.85(6)
N(2)-Co(1)-N(4)	98.99(7)	103.14(12)	95.99(12)	95.58(6)
N(3)-Co(1)-N(1)	94.90(7)	93.06(12)	92.19(11)	100.28(6)
N(3)-Co(1)-N(2)	170.26(7)	169.83(13)	171.93(12)	171.51(6)
N(3)-Co(1)-N(4)	76.96(7)	77.09(12)	77.11(12)	76.98(6)
N(4)-Co(1)-N(1)	99.54(7)	96.06(11)	99.11(12)	100.51(6)
C(22)-O(1)-Co(1)	128.50(13)	127.9(2)	129.4(2)	128.22(12)
C(21)-O(2)Co(1)	127.21(14)	129.3(2)	130.7(2)	131.73(12)
O(1)-C(22)-C(23)	123.45(19)	124.5(3)	123.8(3)	124.29(17)
O(2)-C(21)-C(23)	124.3(2)	123.2(3)	122.9(3)	121.59(16)
O(1)-C(22)-C(25)	115.39(18)	114.3(3)	114.1(3)	115.70(15)
O(2)-C(21)-C(24)	114.95(19)	114.1(3)	114.6(3)	114.49(16)
C(21)-C(23)-C(22)	124.82(19)	124.7(3)	124.8(3)	125.15(17)
C(23)-C(21)-C(24)	120.75(19)	122.6(3)	122.5(3)	123.86(16)
C(23)-C(22)-C(25)	121.14(19)	121.1(3)	122.1(3)	119.96(16)

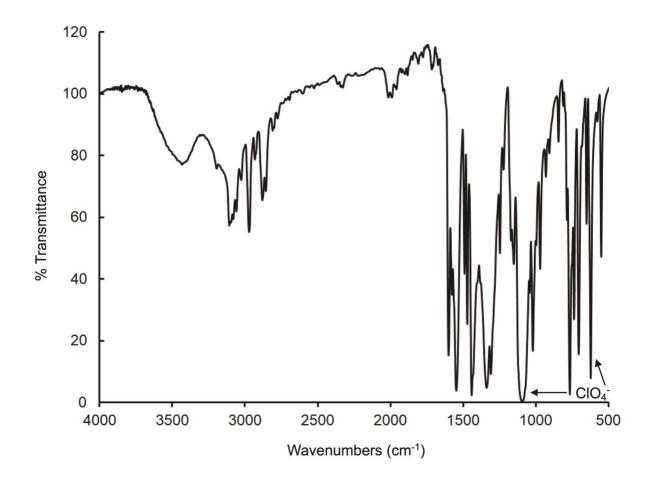


Fig S1. FTIR spectrum for 8 collected as a KBr pellet.

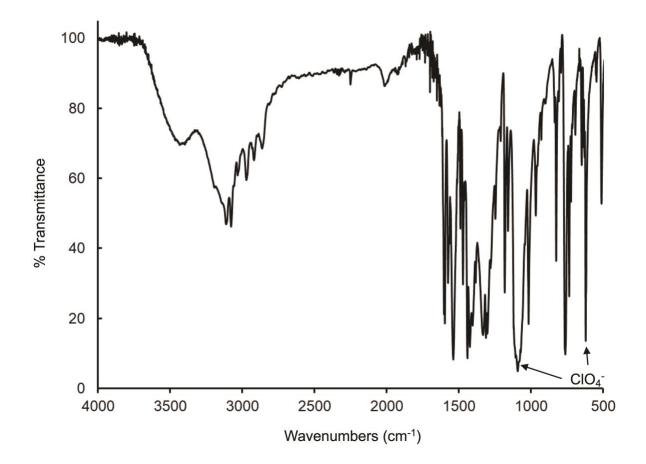


Fig S2. FTIR spectrum for 9 collected as a KBr pellet.

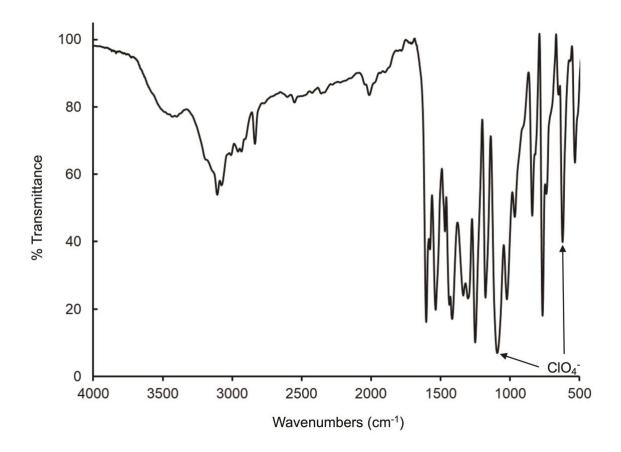


Fig S3. IR spectrum for 10 collected as a KBr pellet.

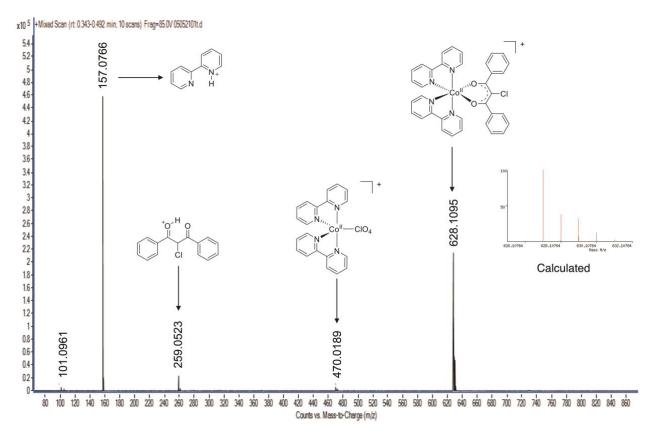


Fig S4. ESI-MS of 8 in CH<sub>3</sub>CN.

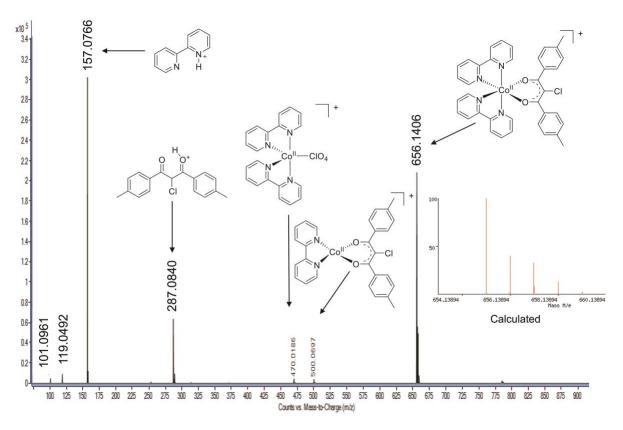


Fig S5. ESI-MS of 9 in CH<sub>3</sub>CN.

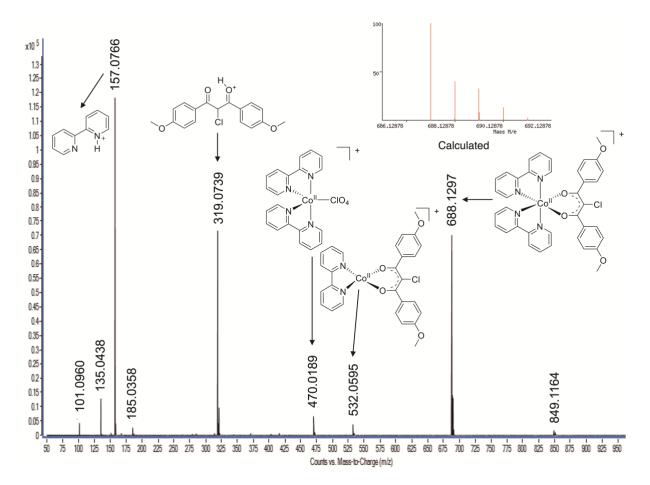
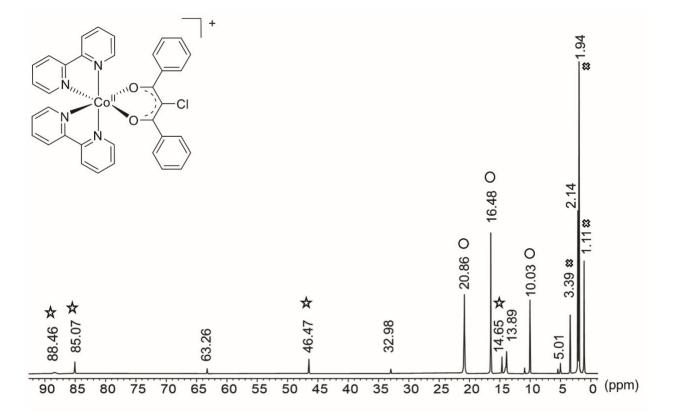
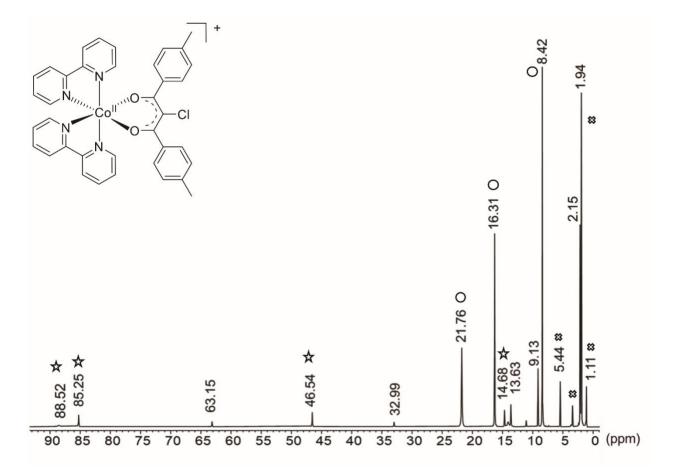


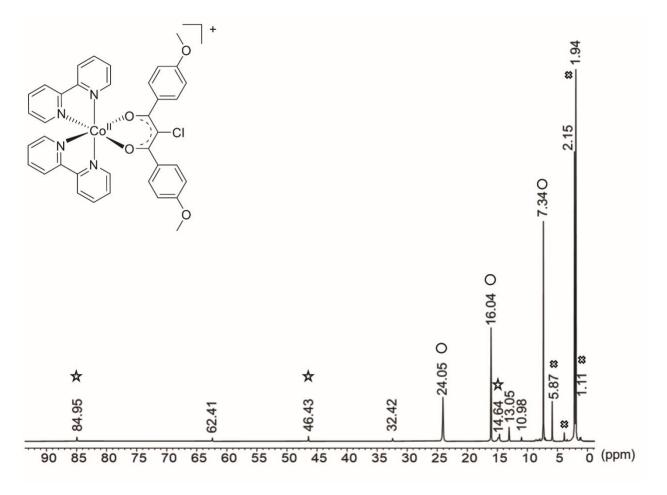
Fig S6. ESI-MS of 10 in CH<sub>3</sub>CN.



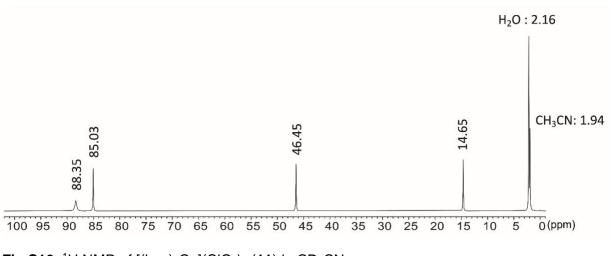
**Fig S7.** <sup>1</sup>H NMR of **8** in CD<sub>3</sub>CN. The signals marked with an open circle correspond to the coordinated diketonate ligand in **8**. Signals marked with a star indicate those from  $[(bpy)_3Co](CIO_4)_2$  (**11**). Residual solvent signals are indicated with (X).



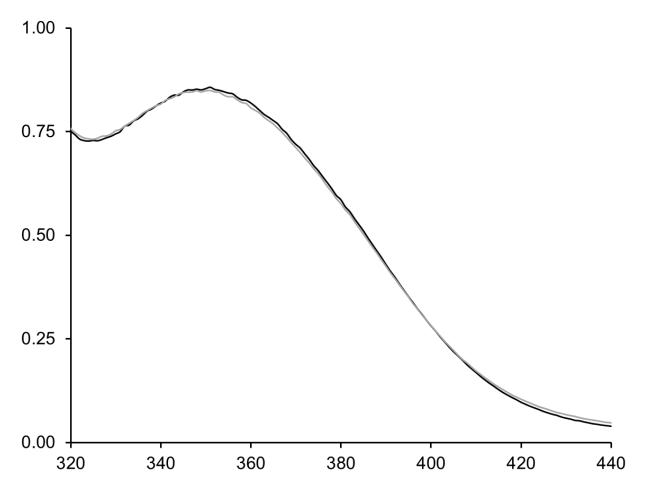
**Fig S8.** <sup>1</sup>H NMR of **9** in CD<sub>3</sub>CN. The signals marked with an open circle correspond to the coordinated diketonate ligand in **9**. Signals marked with a star indicate those from [(bpy)<sub>3</sub>Co](ClO<sub>4</sub>)<sub>2</sub> (**11**). Residual solvent signals are indicated with (X).



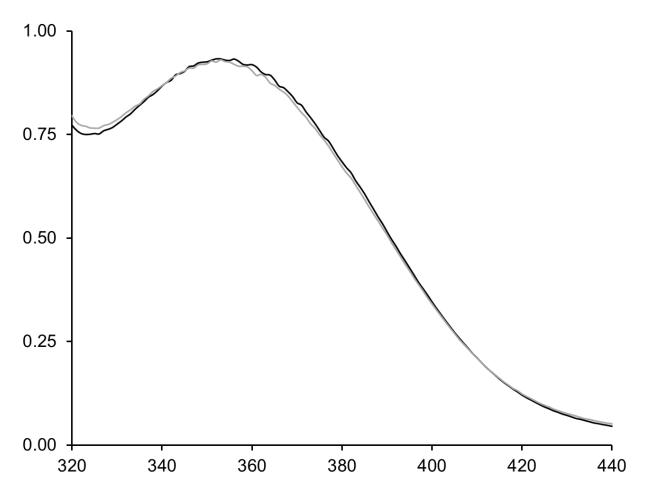
**Fig S9.** <sup>1</sup>H NMR of **10** in CD<sub>3</sub>CN. The signals marked with an open circle correspond to the coordinated diketonate ligand in **10**. Signals marked with a star indicate those from [(bpy)<sub>3</sub>Co](ClO<sub>4</sub>)<sub>2</sub> (**11**). Residual solvent signals are indicated with (X).



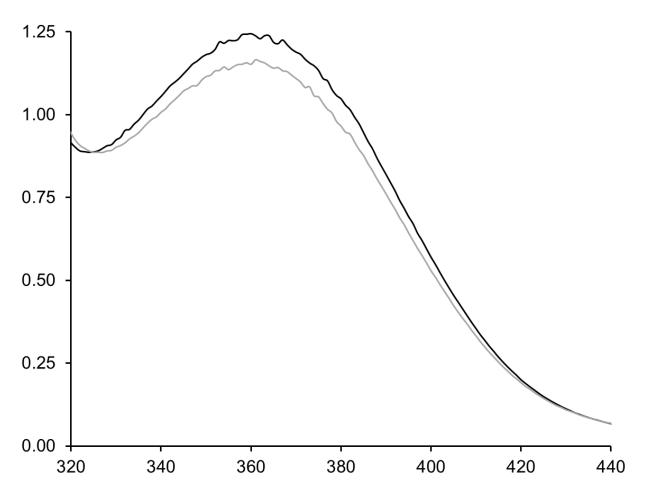
**Fig S10.** <sup>1</sup>H NMR of [(bpy)<sub>3</sub>Co](ClO<sub>4</sub>)<sub>2</sub> (**11**) in CD<sub>3</sub>CN.



**Fig S11.** UV-Vis spectrum of **8** (black) in  $CH_3CN$ . The spectrum in gray is for the same solution after exposure to air in the dark for 20 hours.



**Fig S12.** UV-Vis spectrum of **9** (black) in  $CH_3CN$ . The spectrum in gray is for the same solution after exposure to air in the dark for 20 hours.



**Fig S13.** Absorption spectrum of **10** (black) in  $CH_3CN$ . The spectrum in gray is for the same solution after exposure to air in the dark for 20 hours.

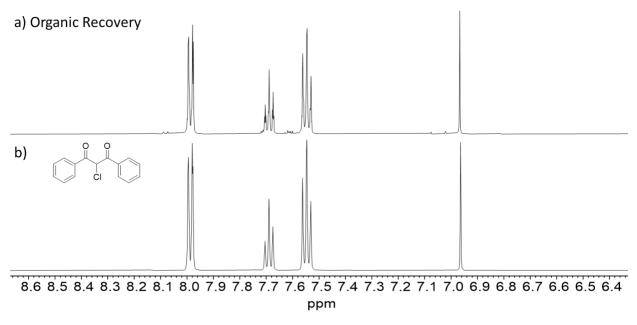
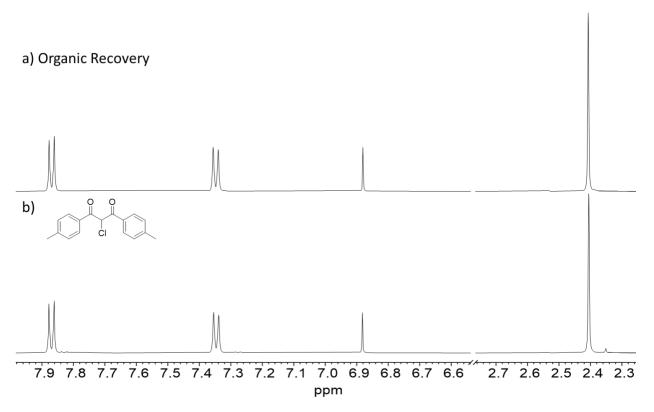
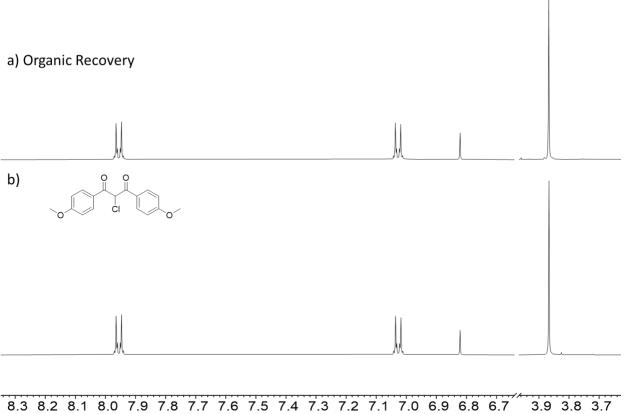


Fig S14. <sup>1</sup>H NMR spectra in CD<sub>3</sub>CN from: a) The isolated organic products from illumination of 8 at 350 nm for 20 hours under  $N_2$ . b) 2-chloro-1,3-diphenyl-1,3-propandione.



**Fig S15.** <sup>1</sup>H NMR spectra in CD<sub>3</sub>CN from: a) The isolated organic products from illumination of **9** at 350 nm for 20 hours under  $N_2$ . b) 2-chloro-1,3-bis(4-methylphenyl)-1,3-propanedione.



ppm

**Fig S16.** <sup>1</sup>H NMR spectra in CD<sub>3</sub>CN from: a) The isolated organic products from illumination of **10** at 350 nm for 20 hours under  $N_2$ . b) 2-chloro-1,3-bis(4-methoxyphenyl)-1,3-propandione.

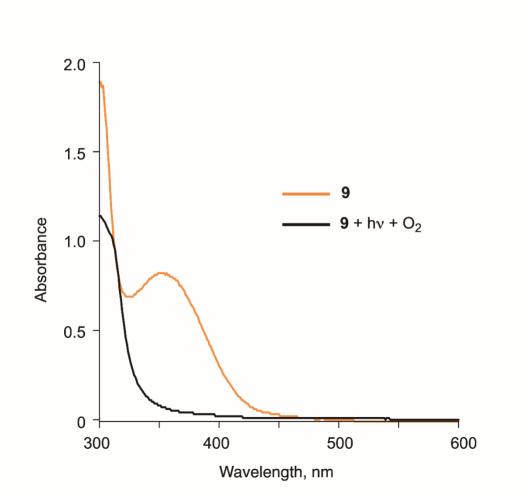


Fig S17. Absorption spectral changes upon illumination of 9 in  $O_2$ -purged CH<sub>3</sub>CN at 350 nm.

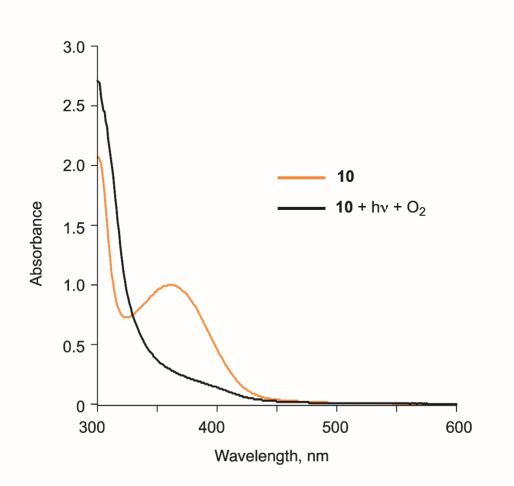
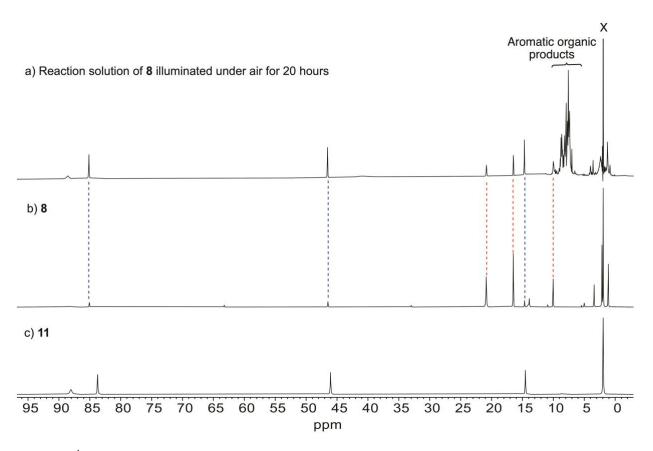


Fig S18. Absorption spectral changes upon illumination of 10 in O<sub>2</sub>-purged CH<sub>3</sub>CN at 350 nm.



**Fig S19.** <sup>1</sup>H NMR studies of the UV-light induced O<sub>2</sub> reactivity of **8** in CD<sub>3</sub>CN (residual solvent signal designated by X). a) Product mixture for an O<sub>2</sub>-purged CH<sub>3</sub>CN solution of **8** after illumination with 350 nm lamps for 20 hours; b) <sup>1</sup>H NMR spectrum of **8**. The change in the relative intensity of the signals for **8** and [(bpy)<sub>3</sub>Co](ClO<sub>4</sub>)<sub>2</sub> (**11**), designated by red and blue dashed lines, respectively, and the appearance of signals for free aromatic organic products, provide evidence for the light-driven diketonate reactivity. c) <sup>1</sup>H NMR of [(bpy)<sub>3</sub>Co](ClO<sub>4</sub>)<sub>2</sub> (**11**) in CD<sub>3</sub>CN. The slight shift in the signals of this compound at ~46 and ~84 ppm in the sample containing **8** (spectrum b)) is attributed to the presence paramagnetic character of the compounds.

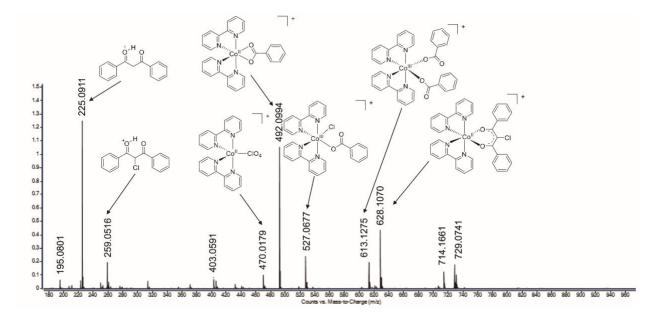


Fig S20. ESI-MS of an O<sub>2</sub>-purged CH<sub>3</sub>CN solution of 8 after illumination at 350 nm for 20 hours.

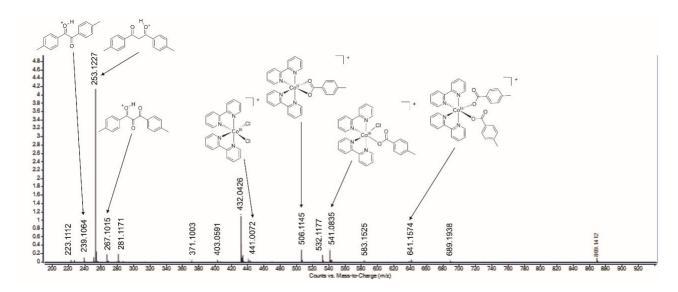


Fig S21. ESI-MS of an  $O_2$ -purged CH<sub>3</sub>CN solution of **9** after illumination at 350 nm for 20 hours.

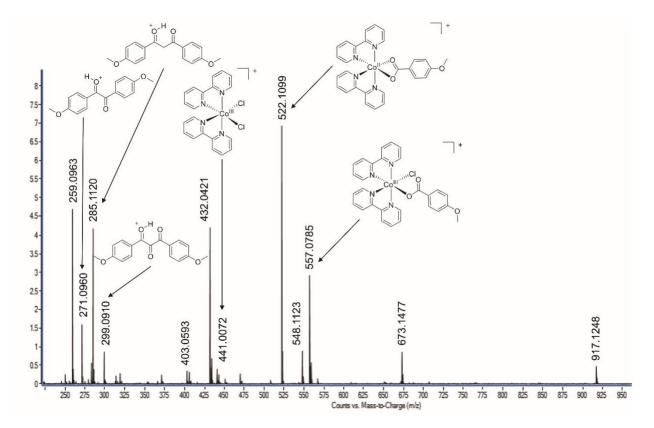
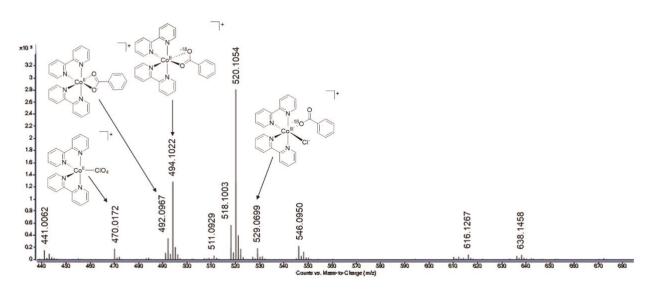
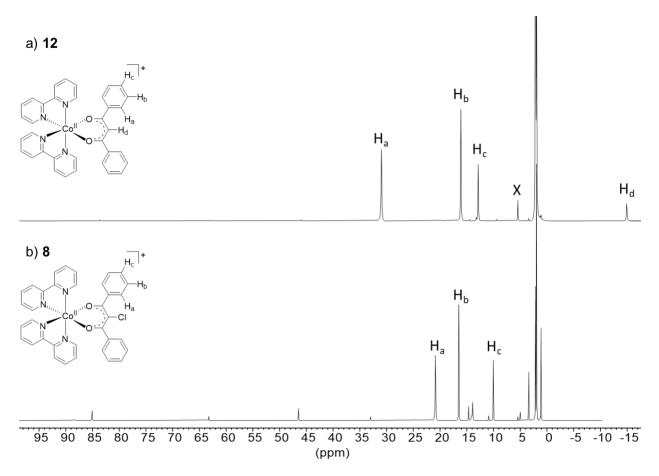


Fig S22. ESI-MS of an  $O_2$ -purged  $CH_3CN$  solution of **10** after illumination at 350 nm for 20 hours.



**Fig S23**. ESI-MS of an CH<sub>3</sub>CN solution of **8** illuminated at 350 nm under <sup>18</sup>O<sub>2</sub> for 20 hours. <sup>18</sup>O incorporation is indicated by comparison of the isotope patterns for  $[(bpy)_2Co(^{18}OOCPh)]^+$  (*m/z* 494.1022) and  $[(bpy)_2Co(O_2CPh)]^+$  (*m/z* 492.0967).



**Fig S24**. a) <sup>1</sup>H NMR of **12** in CD<sub>3</sub>CN. A residual solvent signal (CH<sub>2</sub>Cl<sub>2</sub>) is indicated with (X) in the spectrum of **12**; b) <sup>1</sup>H NMR of **8** in CD<sub>3</sub>CN.

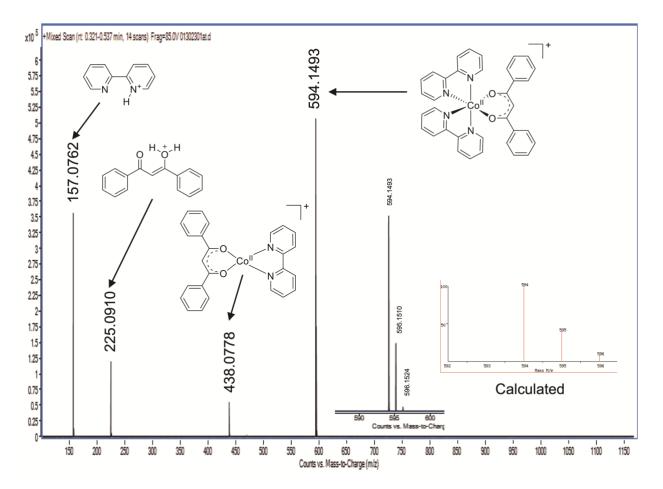


Fig S25. ESI-MS of 12 in CH<sub>3</sub>CN.

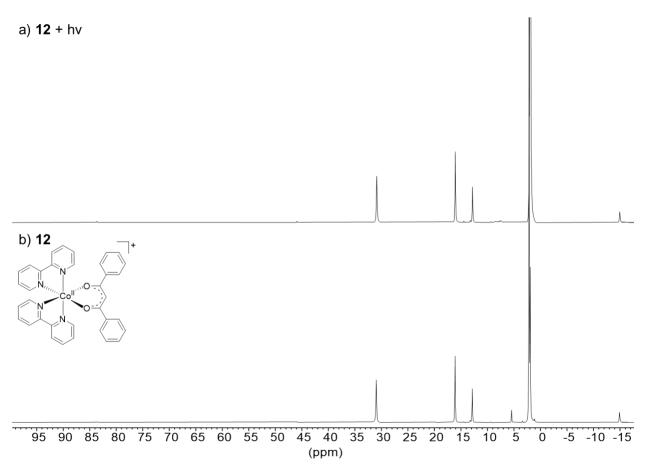


Fig S26. a) <sup>1</sup>H NMR of 12 after illumination at 350 nm in CD<sub>3</sub>CN. b) 12 before illumination.

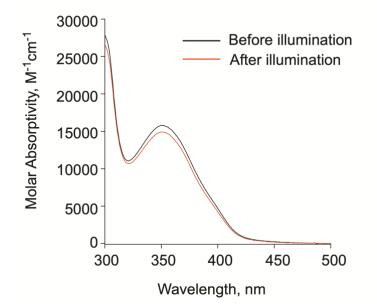
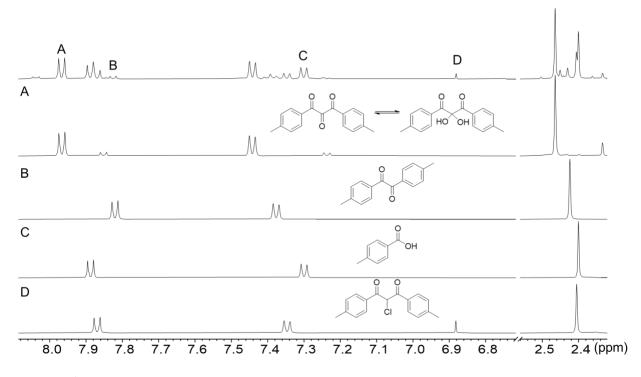
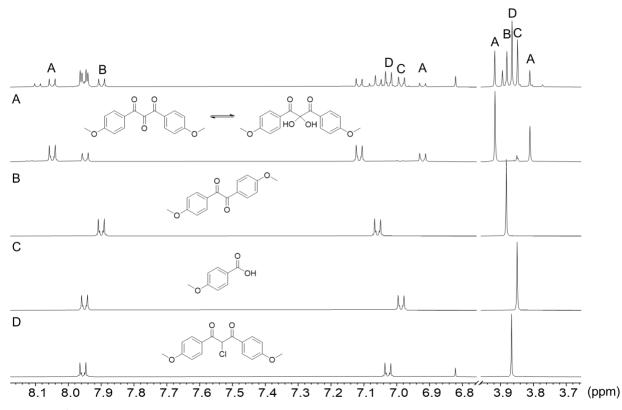


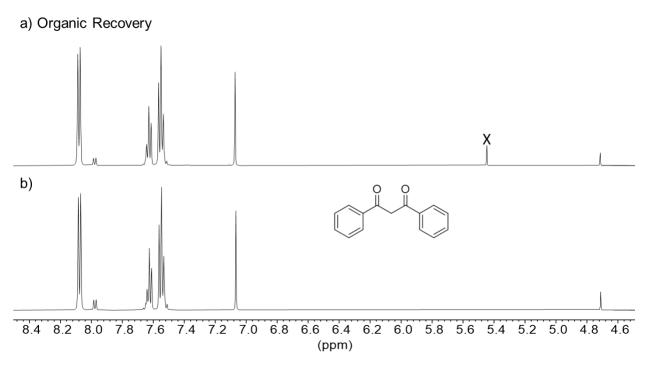
Fig S27. Absorption spectral features of 12 in acetonitrile before and after illumination of an  $O_2$  purged solution at 350 nm.



**Fig S28.** <sup>1</sup>H NMR of isolated organic products from the illumination of **9** in O<sub>2</sub>-purged CH<sub>3</sub>CN with 350 nm for 20 hours (top) compared to the <sup>1</sup>H NMR of organic products (A-D). Resonances that can be used to identify a specific product are marked with A-D. Spectra collected in CD<sub>3</sub>CN.



**Fig S29.** <sup>1</sup>H NMR of isolated organic products from the illumination of **10** in O<sub>2</sub>-purged CH<sub>3</sub>CN with 350 nm for 20 hours (top) compared to the <sup>1</sup>H NMR of organic products (A-D). Resonances that can be used to identify a specific product are marked with A-D. Spectra collected in CD<sub>3</sub>CN.



**Fig S30**. <sup>1</sup>H NMR of a) isolated organic products from the illumination of **12** in O<sub>2</sub>-purged with 350 nm for 20 hours and b) dibenzoylmethane dissolved in CD<sub>3</sub>CN. A residual solvent signal (CH<sub>2</sub>Cl<sub>2</sub>) is indicated with (X) in the spectrum of the sample derived from **12**.

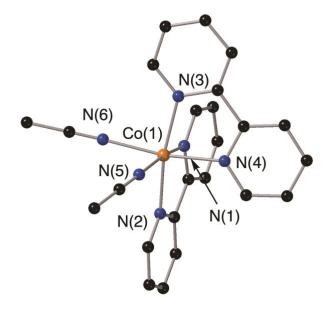


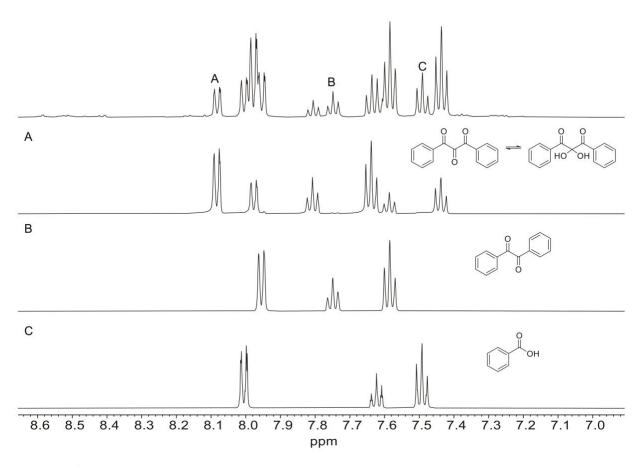
Fig S31. Representation of the cationic portion of 13.

	13	
Co(1)-N(1)	2.1245(14)	
Co(1)-N(2)	2.1228(14)	
Co(1)-N(3)	2.1198(15)	
Co(1)-N(4)	2.1246(14)	
Co(1)-N(5)	2.1358(15)	
Co(1)-N(6)	2.1356(16)	

Table S4. Selected bond distances (Å) for 13

 Table S5.
 Selected bond angles (deg) for 13

	13	
N(1)-Co(1)-N(2)	77.20(5)	
N(1)-Co(1)-N(3)	94.48(5)	
N(1)-Co(1)-N(4)	88.68(5)	
N(1)-Co(1)-N(5)	169.70(6)	
N(1)-Co(1)-N(6)	94.04(6)	
N(2)-Co(1)-N(3)	168.65(5)	
N(2)-Co(1)-N(4)	94.69(6)	
N(2)-Co(1)-N(5)	92.84(6)	
N(2)-Co(1)-N(6)	95.02(6)	
N(3)-Co(1)-N(4)	77.22(6)	
N(3)-Co(1)-N(5)	95.74(6)	
N(3)-Co(1)-N(6)	93.24(6)	
N(4)-Co(1)-N(5)	94.82(6)	
N(4)-Co(1)-N(6)	170.27(6)	



**Fig S32**. <sup>1</sup>H NMR of isolated organic products from the reaction of **13** with 1,3diphenylpropanetrione with NaOCI in CH<sub>3</sub>CN. O<sub>2</sub> was bubbled into the solution of **13** for two minutes. The isolated products (top) are compared to the <sup>1</sup>H NMR of organic standards (A-C). Spectra collected in CD<sub>3</sub>CN.



Fig S33. Dark red-brown solution generated upon bubbling of  $O_2$  through a CH<sub>3</sub>CN solution of 13 for 8 h.

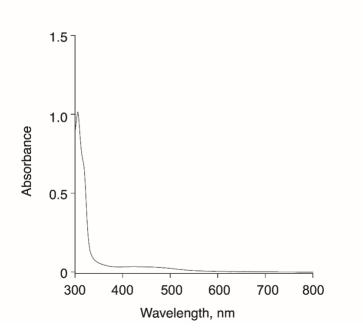
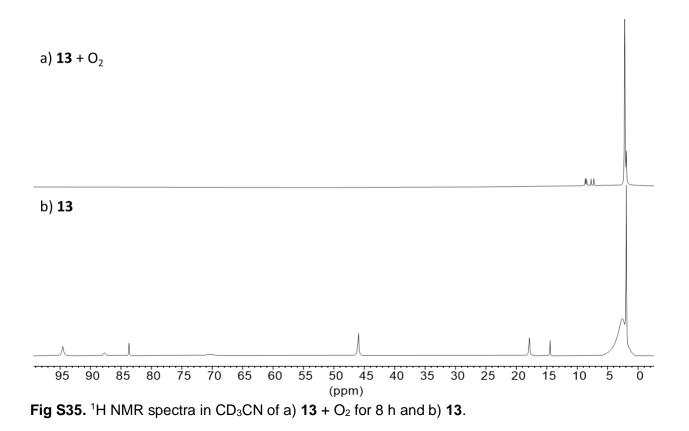


Fig S34. Absorption spectrum of 13 (2.94  $\times$  10<sup>-5</sup> M) in CH<sub>3</sub>CN after bubbling with O<sub>2</sub> for 8 h.



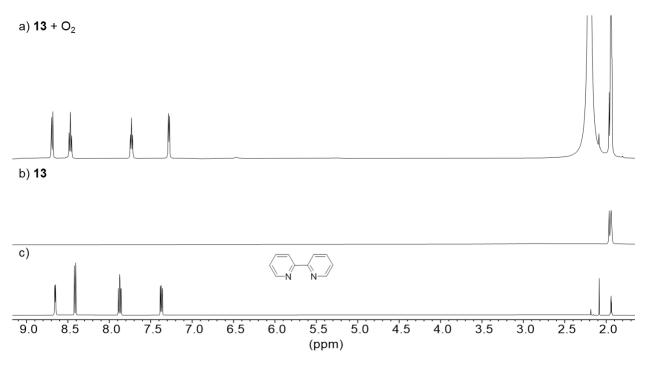
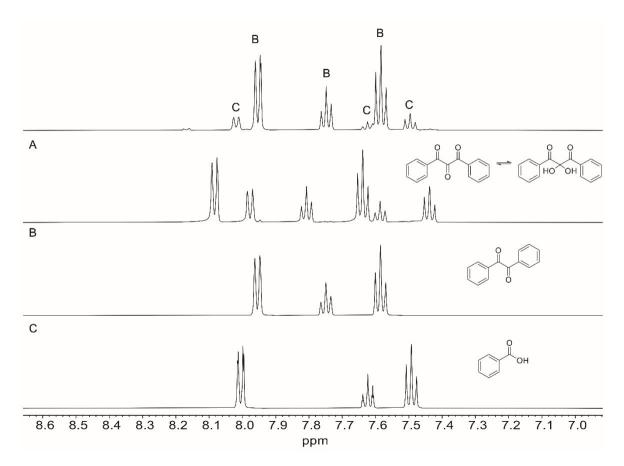
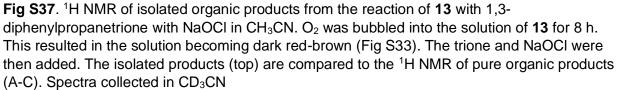
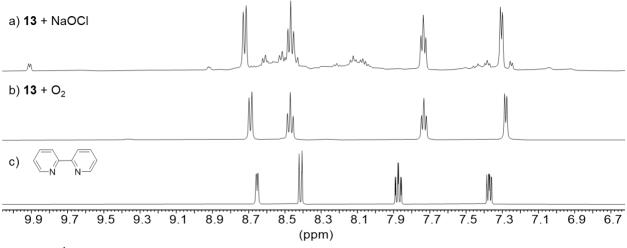


Fig S36. <sup>1</sup>H NMR spectra in CD<sub>3</sub>CN of a)  $13 + O_2$  for 8 h; b) 13 and c) 2,2'-bipyridine.







**Fig S38**. <sup>1</sup>H NMR features in the aromatic region of  $CD_3CN$  solutions of **13** upon treatment with stoichiometric NaOCI (a), or upon extended exposure to  $O_2$  (b). Distinct chemical shifts from free bipyridine c) suggest the formation of bpy-ligated Co(III) species.

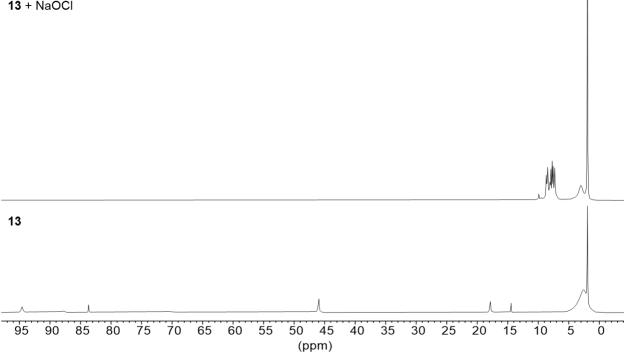


Fig S39. <sup>1</sup>H NMR spectra in  $CD_3CN$  of a) **13** + NaOCI and b) **13**.