Supplementary Information

Polymorphism and polymorph-dependent luminescence properties of the first lithium oxonitridolithosilicate Li₃SiNO₂:Eu²⁺

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Figure S1: Rietveld refinement of the X-ray diffraction data from a powder sample of β -Li₃SiNO₂. The measured pattern is shown in black, the calculated pattern in red, the difference curve in turquiose and the tick marks of the

theoretical positions for the reflections are shown in green for β -Li₃SiNO₂ and in orange for the side phase LiSi₂N₃.

Table S1: Report of the Rietveld refinements of the powder X-ray diffraction data from undoped and doped β -Li₃SiNO₂ samples.

Parameter	Value (undoped sample)	Value (doped sample)
Phase composition	17	- 1 /
β-Li ₃ SiNO ₂ /wt%	95.1(2)	93.7(6)
LiSi ₂ N ₃ /wt%	4.9(2)	6.3(6)
Powder diffraction data of β-Li ₃ SiNO ₂		
Diffractometer	STOE	e stadi p
Radiation; wavelength /Å	Мо-К	α1; 0.7093
a /Å	18.7462(5)	18.7451(7)
b/Å	11.1270(4)	11.1254(5)
c/Å	5.0897(2)	5.0916(2)
Cell volume /ų	1061.66(5)	1061.83(7)
2θ range / °	2 – 40	2-42
2θ step width / °	0.015	0.015
Rexp /%	1.78	5.31
Rwp /%	6.14	8.40
R _p /%	3.91	6.25
R _{Braaa} /%	2.10	2.733



Figure S2: Comparison of the powder X-ray diffraction patterns for α - (negative values/ dotted line) and β -Li₃SiNO₂ (positive values/ solid line). The inset shows similarities and significant differences between the polymorphs in the lower diffraction angles.

Table S2: Wyckoff positions, atomic coordinates, and equivalent isotropic displacement parameters U_{eq} (Å²) of orthorhombic β -Li₃SiNO₂ (standard deviations in parentheses).

Atom	Wyckoff-Position	x	У	Z	Ueq	occ.
Si1	8c	0.98497(2)	0.67981(5)	0.77906(9)	0.0141(2)	1
Si2	8c	0.76186(2)	0.67932(4)	0.17254(9)	0.0143(2)	1
01	8c	0.93834(6)	0.5624(2)	0.6621(2)	0.0162(3)	1
O2	8c	0.80929(6)	0.5616(2)	0.2823(2)	0.0174(4)	1
O3	8c	0.92995(6)	0.3576(2)	0.2544(3)	0.0273(5)	1
O4	8c	0.67788(7)	0.6406(2)	0.2236(3)	0.0295(5)	1
N1	8c	0.96028(8)	0.6952(2)	0.1019(3)	0.0186(4)	1
N2	8c	0.78621(8)	0.6951(2)	0.8473(3)	0.0183(4)	1
Li1	8c	0.8728(2)	0.4583(2)	0.4744(4)	0.034(2)	0.9854(7)
Eu1	8c	0.8728(2)	0.4583(2)	0.4744(4)	0.000026(2)	0.0146(7)
Li2	8c	0.8783(2)	0.2303(3)	0.4774(5)	0.052(3)	0.9888(8)
Eu2	8c	0.8783(2)	0.2303(3)	0.4774(5)	0.000031(3)	0.0112(8)
Li3	8c	0.8737(2)	0.6836(3)	0.4728(7)	0.029(2)	1
Li4	8c	0.8716(2)	0.5747(5)	0.9658(9)	0.0279(9)	0.78
Li4a	8c	0.8733(8)	0.518(2)	0.964(3)	0.028(3)	0.22
Li5	8c	0.0144(2)	0.4493(3)	0.6988(7)	0.035(2)	1
Li6	8c	0.7488(2)	0.4395(3)	0.1271(7)	0.032(2)	1

Table S3: Anisotropic displacement parameters U_{ij} (Å²) of orthorhombic β -Li₃SiNO₂ (standard deviations in parentheses).

Atom	U 11	U22	U 33	U 12	U 13	U 23
Si1	0.0158(2)	0.0133(2)	0.0132(2)	-0.0002(2)	-0.0007(2)	-0.0008(2)
Si2	0.0164(2)	0.0129(2)	0.0135(2)	-0.0006(2)	0.0003(2)	-0.0001(2)
01	0.0175(6)	0.0137(6)	0.0173(6)	0.0002(4)	0.0002(5)	-0.0017(4)
02	0.0208(6)	0.0140(6)	0.0173(6)	-0.0017(5)	-0.0035(5)	0.0006(4)
O3	0.0181(7)	0.0300(9)	0.0337(8)	-0.0006(5)	0.0000(6)	-0.0100(6)
04	0.0228(7)	0.0336(9)	0.0322(8)	-0.0046(6)	0.0044(6)	-0.0073(6)
N1	0.0280(7)	0.0138(7)	0.0139(7)	0.0001(6)	0.0006(6)	0.0011(5)
N2	0.0281(7)	0.0142(7)	0.0126(7)	0.0005(6)	-0.0020(6)	0.0001(5)
Lil	0.036(3)	0.038(3)	0.028(3)	0.003(2)	0.001(2)	0.003(2)

LIZ	0.044(4)	0.062(6)	0.050(5)	-0.020(2)	-0.018(2)	0.018(2)
Li3	0.027(2)	0.016(2)	0.043(2)	0.000(2)	-0.018(2)	-0.001(2)
Li5	0.058(2)	0.029(2)	0.016(2)	0.027(2)	-0.001(2)	0.005(2)
Li6	0.048(2)	0.018(2)	0.026(2)	-0.014(2)	-0.014(2)	0.005(2)

Table S4: Interatomic distances of orthorhombic $\beta\text{-Li}_3\text{SiNO}_2$ (standard deviations in parentheses).

Si1	– N1	1.716(2)		
	– N1	1.720(2)	Ø Si1–N	1.718
	- O1	1.680(2)		
	- 03	1.656(2)	Ø Si1-O	1.668
Si2	– N2	1.726(2)		
	– N2	1.718(2)	Ø Si2–N	1.722
	- O2	1.678(2)		
	- 04	1.652(2)	Ø Si2–O	1.665
111	- 01	1.941(3)		
	- 02	1.922(3)		
	- 03	1.912(3)		
	- 04	1.929(3)	Ø Li1–O	1.926
1i2	- 03	1 970(4)		
	- 03	2.056(4)		
	- 04	1.943(4)		
	- 04	2.178(4)	Ø Li2–O	2.037
		0.010///		
LI3	- N I	2.210(4)		
	- N I	2.493(4)		
	- INZ	2.210(4)		2 3 5 0
	= N2 = 01	2.517(4) 2.053(4)	Ø LIJ-N	2.557
	- 02	2.053(4)	Ø Li3-0	2 0.56
	02	2.000(4)		2,000
Li4	– N1	2.245(5)	~	
	– N2	2.172(5)	Ø LÍ4–N	2.209
	-01	1.993(5)		
	- 02	1.995(5)		
	- 03	3.03(6)		0.47
	- 04	2.03(0)	Ø L14-0	2.4/
Li4a	– N1	2.65(2)		
	– N2	2.63(2)	Ø Li4a–N	2.64
	- 01	2.02(2)		
	- O2	2.07(2)		
	- O3	2.55(2)		
	- 04	2.35(2)	Ø Li4a–O	2.25
Li5	– N1	1.960(4)		
	- 01	1.910(4)		



Figure S3: Comparison of excerpts of layer A (left) and C (right) of β -Li₃SiNO₂ with the differences relevant to the changed connection scheme marked in red. For clarity, the anions are not shown here.



Figure S4: Bimodal Gauß-fits of the emission spectra obtained from powder samples of α - and β -Li₃SiNO₂:Eu²⁺.

Table	S5:	Paramet	ters	of the	two	Gauß-cur	ves	of the	e decomposition	of	the
emissi	on s	pectra fr	om p	bowde	r san	nples of α -	and	β-Li₃Si	NO2:EU ²⁺ .		

	α-Li ₃ S	SiNO ₂	β-Li ₃ SiNO ₂		
	Peak 1	Peak 2	Peak 1	Peak 2	
rel. Intensity	0.72	0.28	0.50	0.50	
E _{peak} / eV	2.04	1.92	2.12	1.99	
ṽ _{peak} / cm ^{−1}	16488	15456	17072	16071	
FWHM / eV	0.29	0.36	0.26	0.35	
FWHM / cm ⁻¹	2323	2886	2130	2850	



Figure S5: The local environment of the Li2-position in β -Li₃SiNO₂:Eu²⁺is shown with its bond lengths in a 4+2 coordination.