

Electronic Supplementary Information

Microstructured layered-kagome $\text{BaCo}_3(\text{VO}_4)_2(\text{OH})_2$ with variable crystallite size: alternative synthetic route and comparison with nanostructured samples

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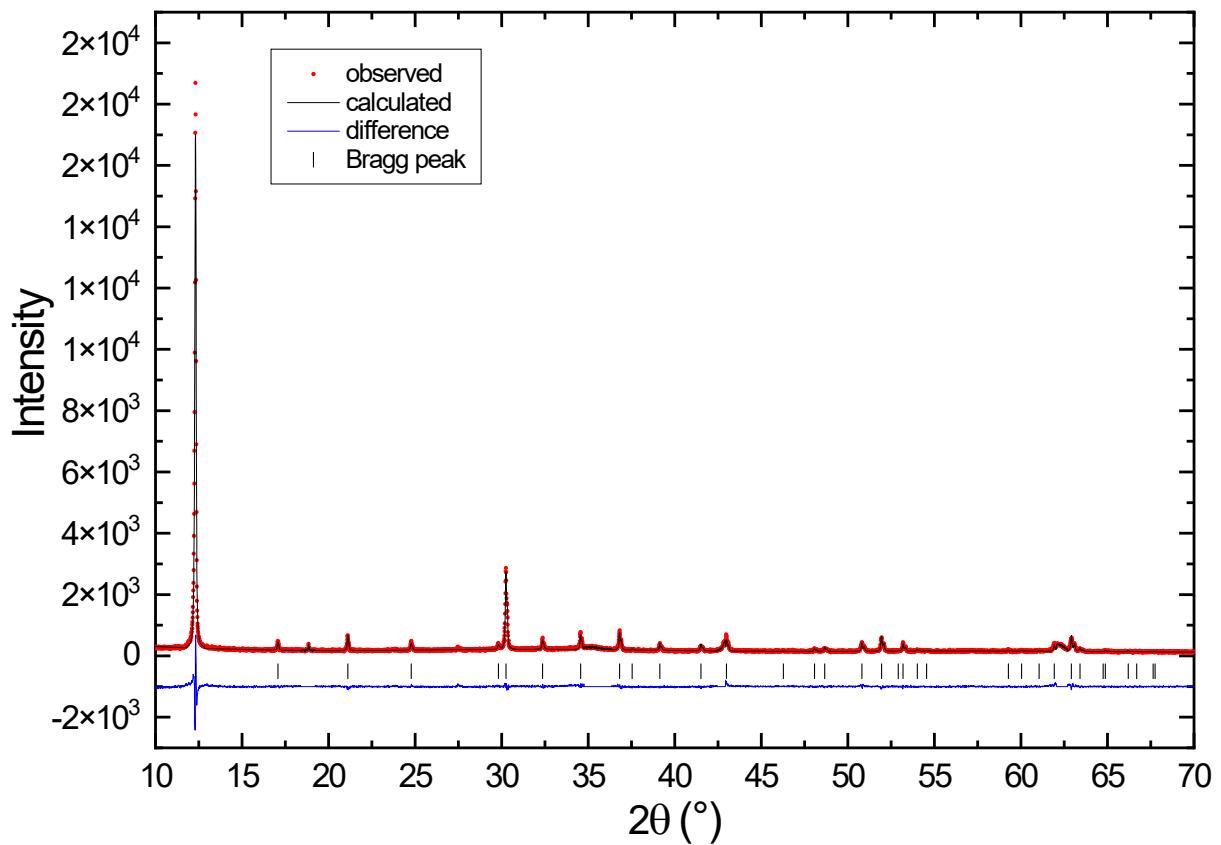


Figure S1. Rietveld refinement plot for $\text{Co}_3\text{V}_2\text{O}_7(\text{OH})_2 \cdot 2\text{H}_2\text{O}$ showing observed, calculated and difference patterns. The high-level noisy pattern is due to cobalt fluorescence.

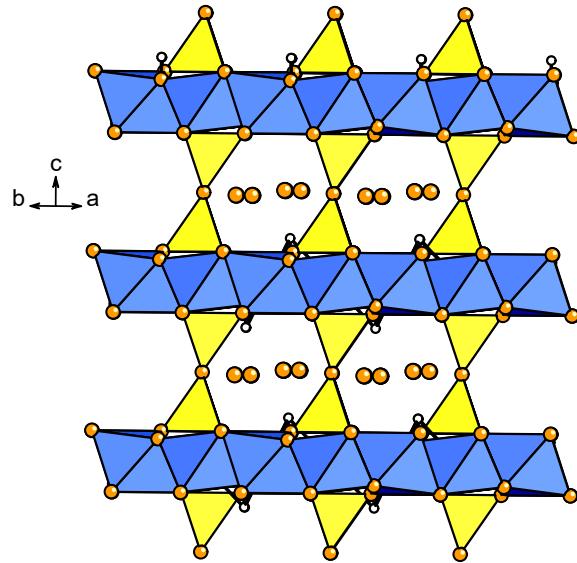


Figure S2. Crystal structure of $\text{Co}_3\text{V}_2\text{O}_7(\text{OH})_2 \cdot 2\text{H}_2\text{O}$. Blue octahedra: CoO_6 , yellow tetrahedra: VO_4 , orange sphere: oxygen, white sphere: hydrogen.

Table S1. Structural parameters and crystallite size of $\text{Co}_3\text{V}_2\text{O}_7(\text{OH})_2 \cdot 2\text{H}_2\text{O}$.

SG $P\bar{3}m$: $a=5.9886(3)$ Å, $c=7.1804(4)$ Å, Cryst. size=217(5) nm $\perp [001]$					
Atom	x	y	z	SOF	U _{iso} Equiv (Å ²)
Co ₁	0.5	0	0	1	0.019(7)
V ₁	0	0	0.2494(16)	1	0.034(7)
O ₁	0.6667	0.3333	0.874(5)	1	0.000(7)
O ₂	0.155(3)	0.311 (5)	0.829 (2)	1	0.000
O ₃	0	0	0.5	1	0.000
O ₄	0.7548(13)	0.374(6)	0.482(12)	1	0.000
H ₁	0.6667	0.3333	0.75*	1	0.02*
GOF= 1.41, Rwp= 9.49, Rp= 7.43					

Aniso. ADP (Å ²)	U ¹¹	U ²²	U ³³	U ¹²	U ¹³	U ²³
Co ₁	0.013(9)	0.002(10)	0.039(6)	0.001(5)	0.014(11)	0.03(2)
V ₁	0.032(10)	0.032	0.037(9)	0.0216(5)	0	0

*not refined

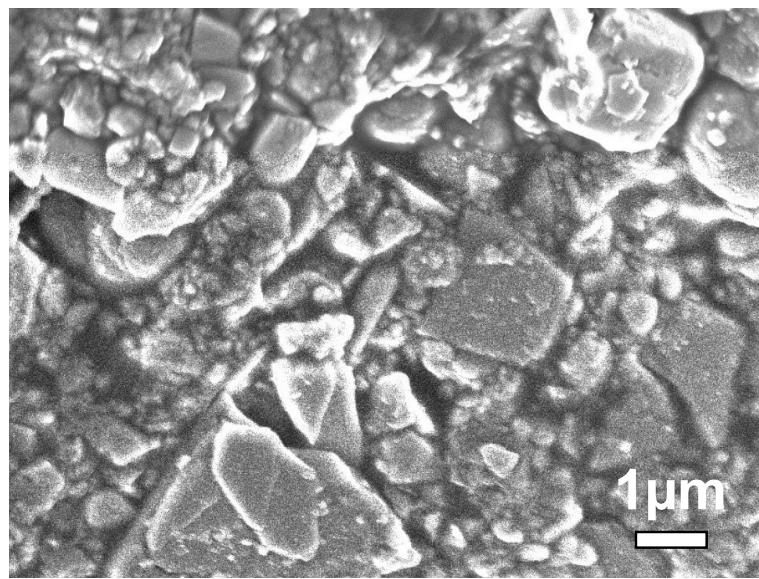


Figure S3. SEM image of $\text{Co}_3\text{V}_2\text{O}_7(\text{OH})_2 \cdot 2\text{H}_2\text{O}$.

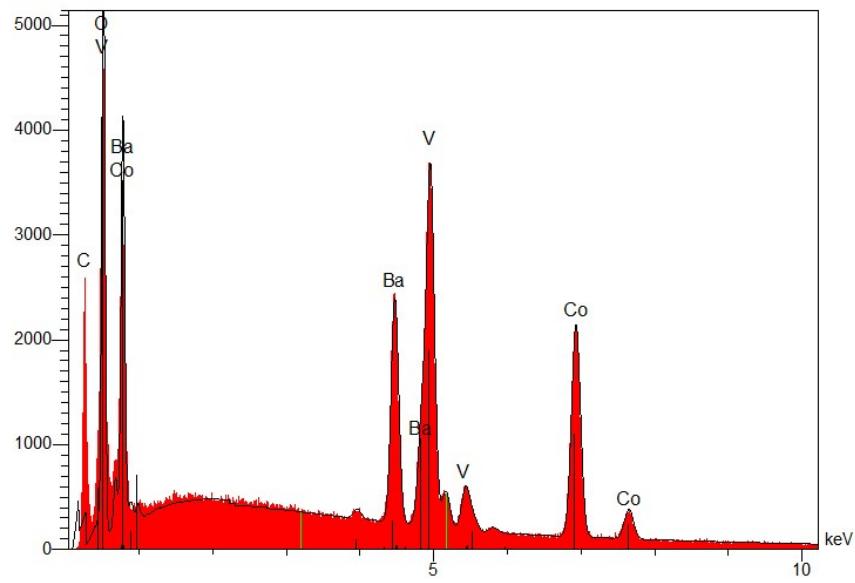


Figure S4. EDS spectrum acquired on micrometric particles of **BaCo₃-A** (a similar spectrum has been obtained for **BaCo₃-B**).

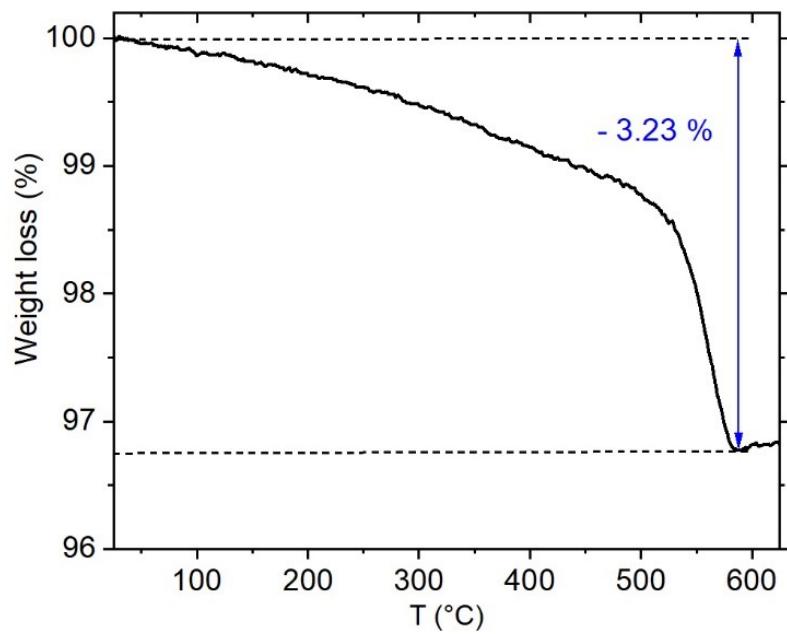


Figure S5. Thermogravimetric curve of microstructured **BaCo₃-B** recorded in air.

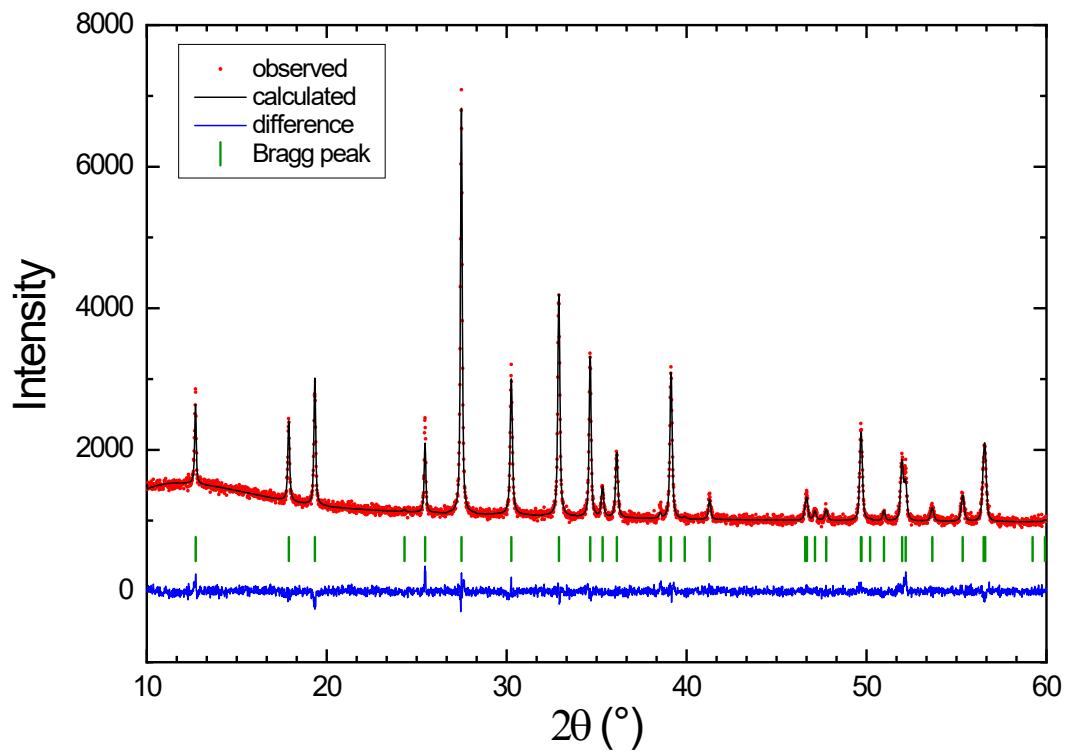


Figure S6. Rietveld refinement plot for microstructured **BaCo₃-B** showing observed, calculated and difference patterns. The high-level noisy pattern is due to cobalt fluorescence.

Table S2. Structural parameters and crystallite size of **BaCo₃-B**.

SG $R\bar{3}m$: $a=5.9238(2)$ Å, $c=21.0609(9)$ Å, Cryst. size=92(3) nm \perp [001], 47(5) nm // [001]					
Atom	x	y	z	SOF	U_{iso} Equiv (Å ²)
Ba ₁	0.6667	0.3333	0.8333	1	0.033(2)
Co ₁	0.8333	0.6667	0.6667	1	0.013 (2)
V ₁	0.3333	0.6667	0.7527(3)	1	0.015(3)
O ₁	0.3333	0.6667	0.832(8)	1	0.005 (3)
O ₂	0.4930(9)	0.5070(10)	0.7275(4)	1	0.005
O ₃	0	0	0.7030(10)	1	0.005
H ₁	0	0	0.7494*	1	0.008*

GOF= 1.11, Rwp= 3.32, Rp= 2.62

*not refined

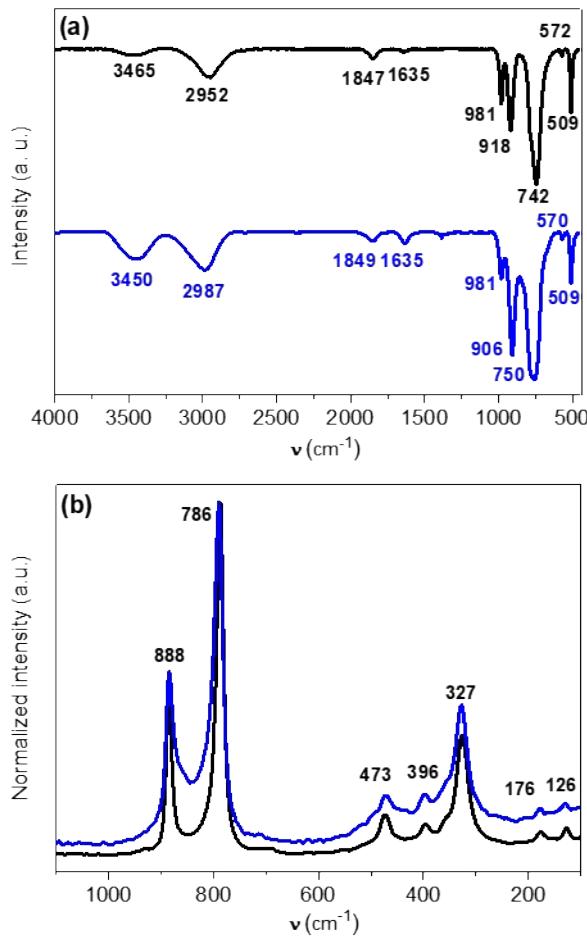


Figure S7. Comparison of (a) FT-IR and (b) Raman spectra of microstructured **BaCo₃-B** (black line) and **BaCo₃(VO₄)₂(OH)₂** nanoparticles^{S1} (NPs) (blue line).

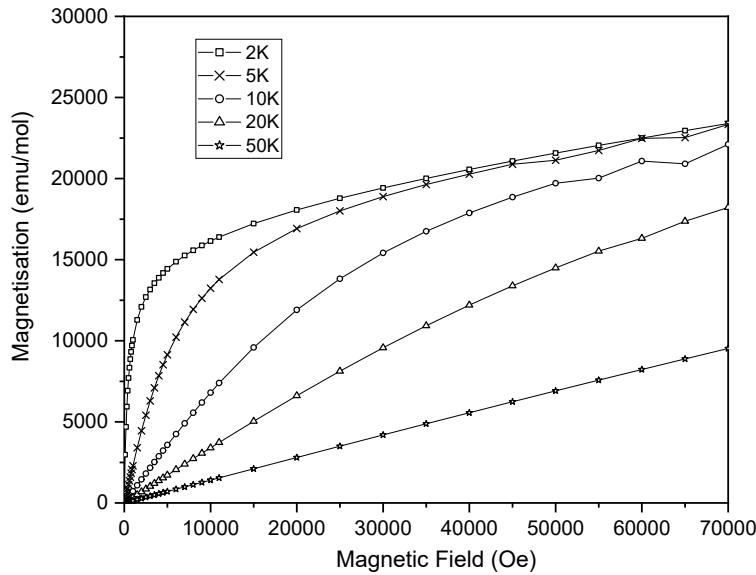


Figure S8. Magnetic field dependence of magnetisation measured at several temperatures for $\text{BaCo}_3\text{-B}$.

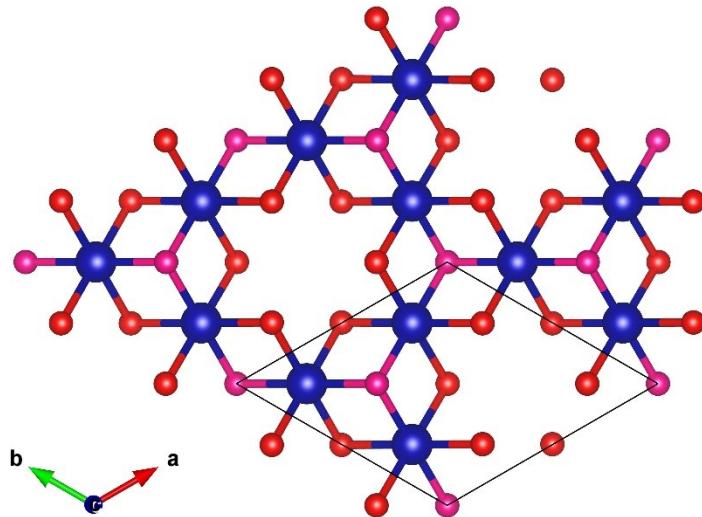


Figure S9. Fragment of a kagome layer in the $\text{BaCo}_3(\text{VO}_4)_2(\text{OH})_2$ crystal structure (projection along [001] direction). Blue, red, and pink spheres represent Co, O₂, and O₃ atoms (see Tables 1 and S2), respectively. Barium, vanadium, and hydrogen atoms have been removed for clarity. Each kagome layer is formed from compressed $\text{Co}(\text{O}_2)_4(\text{O}_3)_2$ octahedra with $d_{\text{Co-O}2} = 2.176(4)$ Å and $d_{\text{Co-O}3} = 1.9407(6)$ Å.

References

- [S1] R. Dessapt, L. Lajaunie, J. J. Calvino, P. Deniard, I. Trenque and C. Payen, *J. Mater. Chem. C*, 2022, **10**, 3287-3291.