

## *Supplementary materials*

### **Inorganic hydrogen-bonded SnO(OH)<sub>2</sub> as molecular springs boosted the piezocatalytic degradation of contaminants**

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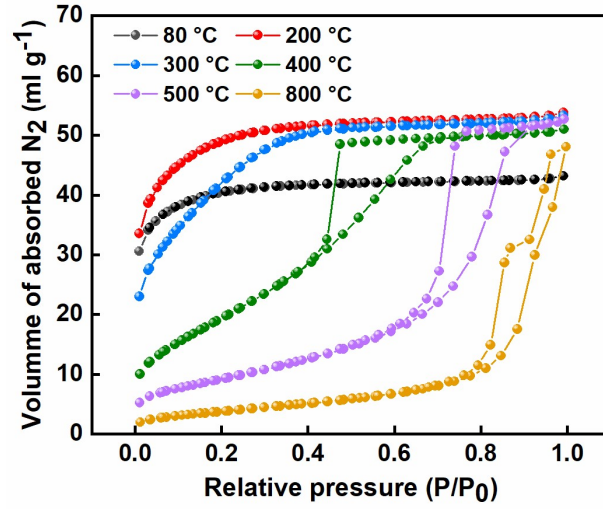
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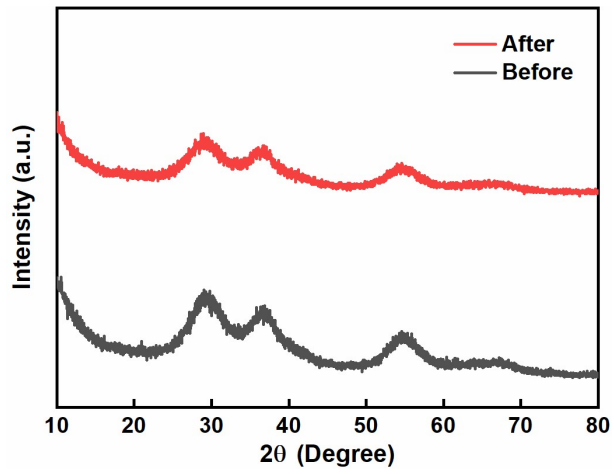
c Guangdong Provincial Key Laboratory of Environmental Pollution Control and Remediation Technology, Sun Yat-Sen University, No. 135, Xingang Xi Road, Guangzhou 510275, P. R. China

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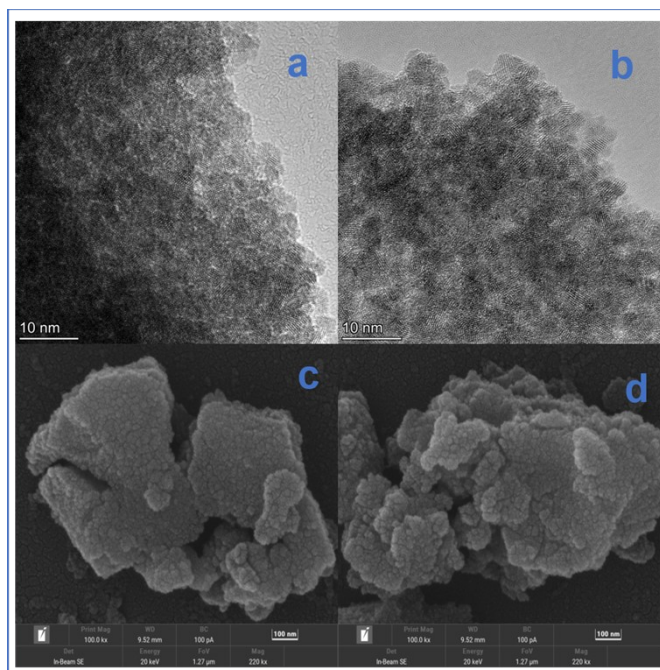
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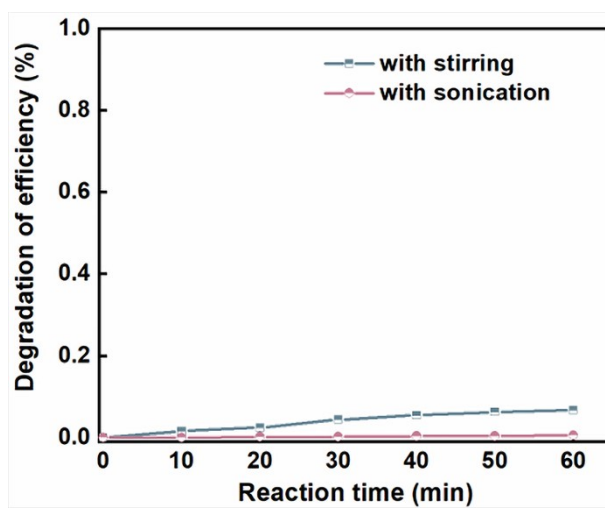
**Fig. S1.** N<sub>2</sub> adsorption-desorption isotherms of SnO<sub>x</sub>H<sub>y</sub>.



**Fig. S2.** XRD pattern of SnO(OH)<sub>2</sub> before and after cycles.



**Fig. S3.** The morphology and structure of the SnO(OH)<sub>2</sub> before and after cycles. (a: TEM before cycle; b: TEM after cycles; c: SEM before cycle; d: SEM after cycles)



**Fig. S4.** The degradation of TCZ after addition H<sub>2</sub>O<sub>2</sub> with stirring and with sonication.

**Table S1.** Surface characteristics of SnO<sub>x</sub>H<sub>y</sub>.

Samples	S <sub>BET</sub> (m <sup>2</sup> g <sup>-1</sup> )	S <sub>ext</sub> (m <sup>2</sup> g <sup>-1</sup> )	V <sub>total</sub> (cm <sup>3</sup> g <sup>-1</sup> )	V <sub>meso</sub> (cm <sup>3</sup> g <sup>-1</sup> )
80 °C	149.12	108.14	0.0668	0.0206
200 °C	173.86	76.23	0.0832	0.0373
300 °C	153.16	160.81	0.0824	0.1032
400 °C	75.25	94.07	0.0789	0.0777
500 °C	58.63	60.13	0.0815	0.0820
800 °C	33.73	35.65	0.0743	0.0725