## Supporting Information

# Cocrystal Engineering Strategy for Sustained Release and Leaching Reduction of Herbicides: A Case Study of Metamitron

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Agrichemical name	Coformer name	Result	
Metamitron	Cinnamic acid	$\sqrt{\sqrt{1}}$	
Metamitron	Ferulic acid	×	
Metamitron	Syringic acid	×	
Metamitron	Vanillic acid	×	
Metamitron	P-coumaric acid	×	
Metamitron	Hesperetin	×	
Metamitron	Gallic acid	×	
Metamitron	Salicylic acid	×	
Metamitron	Ellagic acid	×	
Metamitron	Quercetin	×	
Metamitron	Hesperetin	×	
Metamitron	Kaempferol	×	
Metamitron	Baicalein	$\sqrt{\sqrt{1}}$	
Metamitron	Resveratrol	$\checkmark$	
Metamitron	Naringenin	$\sqrt{\sqrt{2}}$	
Metamitron	Catechins	×	
Metamitron	Glycolic acid	×	

 Table S1. Overview of Screening Experiments

 $\sqrt{\text{represents a new cocrystal/salt determined by PXRD}}; \sqrt{\sqrt{\text{represents a new cocrystal/salt}}}$ determined by PXRD and its structure is determined by SCXRD; × represents no new cocrystal/salt found by PXRD

Parameter	Details
System	Agilent 1260 LC
Column	C18 column (6 $\mu$ m, 4.6 mm × 150 mm)
Mobile phase	Methanol-water (70:30)
Flow rate	1 mL/min
Inject volume	20 µL
Column temperature	25 °C
Sample temperature	25 °C
$\lambda_{max}$	254 nm
Retention time	4.6 min
Equation	y = 22311.5587x + 21.10418
Regression coefficient (R <sup>2</sup> )	0.99999
Calibration range	0.01-0.4  mg/mL

 Table S2. HPLC method parameters for MET

Table S3. Physical and chemical parameters of the soil column

Properties	Indicators
Soli particle size (mm)	≤2
Weight (g)	250
Volumetric weight (g/cm <sup>3</sup> )	$1.29\pm0.02$
Loading height (cm)	$16.0\pm0.3$
Pore volume (mL)	$116.2 \pm 3.1$
Flow velocity (mL/min)	1.0

Parameter	Details
System	Waters ACQUITY
Column	C18 column (6 $\mu$ m, 4.6 mm × 150 mm)
Mobile phase	Methanol-water (70:30)
Flow rate	1 mL/min
Inject volume	20 µL
Column temperature	25 °C
Sample temperature	25 °C
$\lambda_{\max}$	254 nm
Retention time	4.2 min
Equation	y = 111.1247x + 0.35979
Regression coefficient (R <sup>2</sup> )	0.99998
Calibration range	$0.1-1.0~\mu g/mL$

**Table S4.** UPLC method parameters for MET

 Table S5. Hydrogen bond geometrical parameters of cocrystals

Compound	D–H···A	d(H…D)/Å	d(H···A)/Å	d(D····A)/Å	$\theta$ (D–H···A)/°	symmetry code
MET-CA	N(4)-H(4A)…N(1)	0.865	2.481	3.121	131.34	x+1/2, -y+3/2, z+1/2
	N(4)-H(4A)····O(3)	0.865	2.397	3.012	128.51	x+1/2, -y+3/2, z+1/2
	N(4)-H(4B)…O(1)	0.924	2.209	2.698	112.26	x-1,y,z
	N(4)-H(4B)…O(6)	0.924	2.453	3.189	136.67	x+2, y, z+1
	O(2)-H(2)···N(2)	0.840	1.828	2.656	168.41	

	N(8)-H(8A)····O(3)	0.860	2.291	3.037	145.12	
	N(8)-H(8A)····O(4)	0.860	2.264	2.688	110.39	
	O(5)-H(5)…N(6)	0.840	1.857	2.683	167.21	
MET-NA	N(1)-H(1A) ···O(6)	0.865	2.213	2.902	136.43	x+1, -y+1, z+1/2
	N(1)-H(1B)····O(4)	0.865	2.126	2.869	143.70	x, y+1, z
	O(2)-H(2)····N(3)	0.840	1.811	2.632	165.37	
	O(3)-H(3)····O(4)	0.840	1.846	2.586	146.24	
	O(3)-H(3)····O(6)	0.840	2.540	3.042	119.48	x+1, -y, z+1/2
	O(6)-H(6)····O(3)	0.840	1.922	2.734	162.21	x-1, y, z-1
MET-BA	N(4)-H(4A)····O(1)	0.945	2.219	2.709	111.31	
	N(4)-H(4A)····O(3)	0.945	2.376	3.223	149.00	-x+3/2, y+1/2, -z+3/2
	N(4)-H(4B)····O(2)	0.961	2.002	2.954	170.83	-x+1/2, y+1/2, -z+3/2
	O(2)-H(2B)····O(6)	0.894	1.715	2.560	156.64	
	O(3)-H(3B)····O(1)	0.891	1.976	2.782	149.56	-x+3/2, y-1/2, -z+3/2
	O(3)-H(3B)····O(4)	0.891	2.189	2.678	113.94	
	O(4)-H(4C)…N(2)	0.952	2.623	3.519	157.00	
	O(4)-H(4C)… N(1)	0.952	1.725	2.673	173.54	

**Table S6.** Summary tables of significance analysis of leaching behavior

MET - CA:

Analysis of variance: one-way analysis of variance

SUMMARY

Groups	Counts	Sum	Average	Variance
Column 1	12	1053.86658	87.822215	437.199827

		9	7	4		
Column 2	12	850.035210 2	70.836267 5	939.498210 9		
ANOVA						
Source of Variation	SS	df	MS	F	P-value	F crit
Between Groups	1731.13461 9	1	1731.1346 2	2.51490823 9	0.12704 4	4.3009 5
With Groups	15143.6784 2	22	688.34901 9			
Total	16874.8130 4	23				

## MET - NA:

Analysis of variance: one-way analysis of variance

#### SUMMARY

Groups	Counts	Sum	Average	Variance
Column 1	12	1053.867	87.82222	437.1998
Column 2	12	97.84546	8.153788	43.03884

ANOVA

Source of Variation	SS	df	MS	F	P-value	F crit
Between Groups	38082.35	1	38082.35	158.5976	1.56E-11	4.30095
With Groups	5282.625	22	240.1193			
Total	43364.98	23				

## MET - BA:

Analysis of variance: one-way analysis of variance

SUMMARY

Groups	Counts	Sum	Average	Variance	-	
Column 1	12	1053.867	87.82222	437.1998	-	
Column 2	12	165.6016	13.80014	85.86715		
ANOVA						
C CV : .:	66	10	1.00		D 1	
Source of Variation	55	df	MS	F	P-value	F crit
Source of Variation Between Groups	32875.61	<i>df</i> 1	MS 32875.61	F 125.7033	<i>P-value</i> 1.45E-10	<i>F crit</i> 4.30095
Source of Variation Between Groups With Groups	32875.61       5753.737	<i>df</i> 1 22	MS 32875.61 261.5335	F 125.7033	<i>P-value</i> 1.45E-10	<i>F crit</i> 4.30095

**Table S7.** Analysis of significant differences between different cocrystal treatment groups

### Post Hoc Tests

## Homogeneous Subsets

Duncan

		Subset for $alpha = 0.05$					
treatment	N						
		1	2	3			
MET	74	3.4789					
MET-CA	72	4.8593					
MET-BA	74		66.9409				
MET-NA	71			82.1775			
Sig.		0.823	1.000	1.000			

Compound	pH	CEC [mmol/kg]
MET	6.95±0.16	171±4
MET-CA	6.89±0.12	182±2
MET-NA	7.06±0.01	174±11
MET-BA	7.04±0.12	181±7

**Table S8.** Relevant properties such as soil pH and CEC after all kinds of herbicides have been applied.

#### **Supporting Figures:**



Figure S1. Standard curve of MET content (concentration-peak area) determined by HPLC.



Figure S2. Schematic diagram of the soil column experiment.



Figure S3. Standard curve of MET content (concentration-peak area) determined by UPLC.



Figure S4. Supramolecular synthons with similar structures reported in the literature.<sup>1-9</sup>



**Figure S5.** PXRD pattern of the MET-CA neat grinding product (yellow) compared with MET (blue), CA (orange) and the calculated pattern obtained from the single crystal data (purple).



**Figure S6.** PXRD patterns of the MET-NA neat grinding product (yellow) compared with MET (blue), NA (orange) and the calculated pattern obtained from the single crystal data (purple).



**Figure S7.** PXRD pattern of the MET-NA neat grinding product (yellow) compared with MET (blue), NA (orange) and the calculated pattern obtained from the single crystal data (purple).



**Figure S8.** Confocal Raman micro-spectra of the MET-CA neat grinding product (yellow) compared MET (blue) and CA (orange).



**Figure S9.** Confocal Raman micro-spectra of the MET-NA neat grinding product (yellow) compared MET (blue) and NA.



**Figure S10.** Confocal Raman micro-spectra of the MET-BA neat grinding product (yellow) compared MET (blue) and BA (orange).



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**Figure S13.** Comparison of DSC plots of MET (blue) and CA (orange) with the MET-BA cocrystal (yellow).



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Figure S22. Dissolution profiles of MET, and three cocrystals in water at 45 °C.



Figure S23. Morphological pictures of Kentucky bluegrass after 14 days of different treatments (data represent germination rate: Control, 100%; MET, 98.67%; MET-CA, 96.00%; MET-BA, 98.67%; MET-NA, 94.67%).

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