

## *Electronic Supplementary Information*

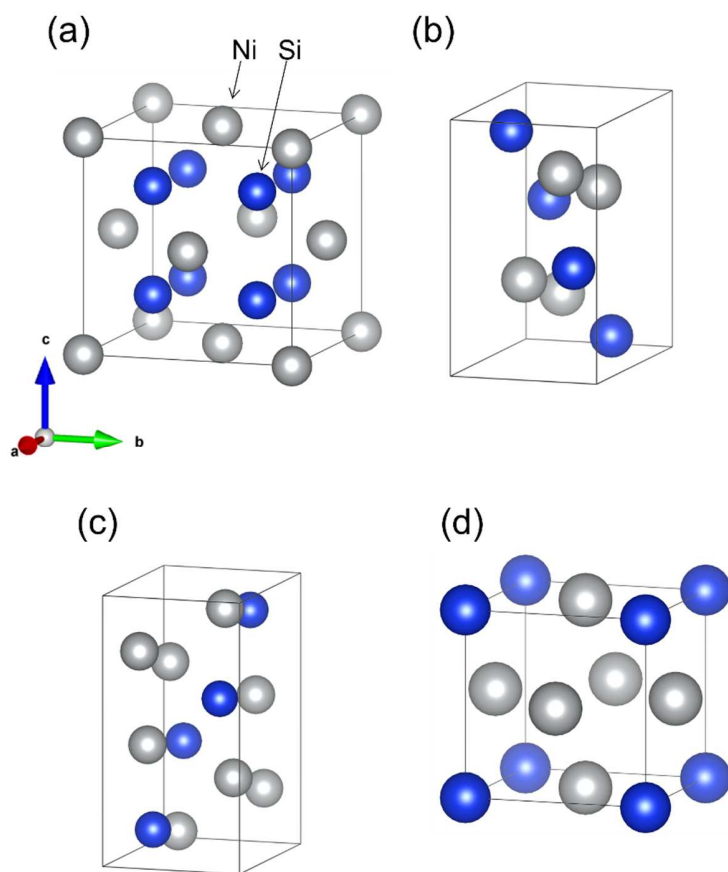
# Impact of the Crystal Phase of Binary Silicide on its Lithiation and Delithiation Properties

*Yasuhiro Domi,<sup>a,b</sup> Hiroyuki Usui,<sup>a,b</sup> Takumi Ando,<sup>b,c</sup> Hiroki Sakaguchi<sup>a,b,\*</sup>*

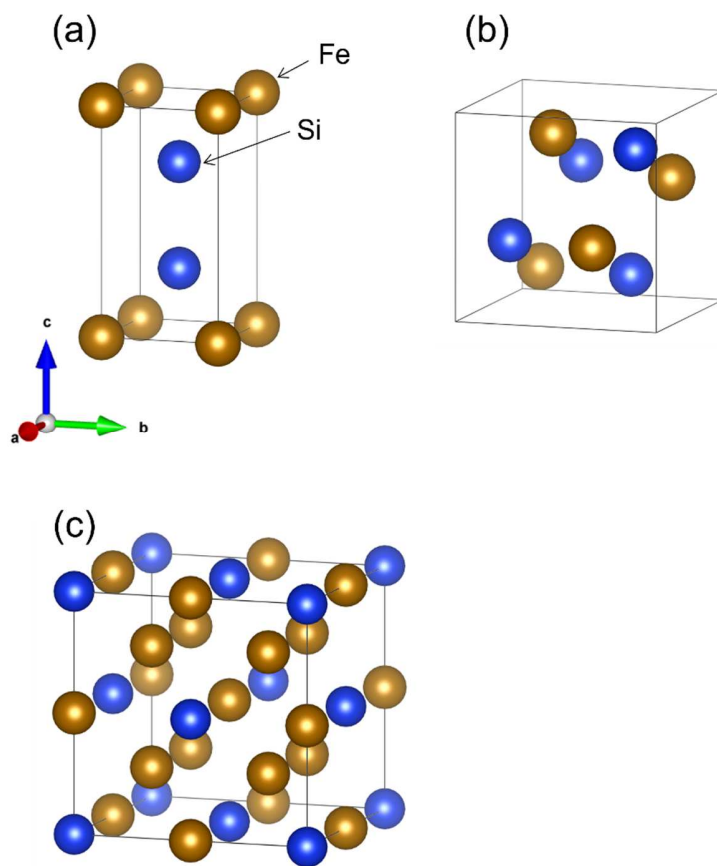
<sup>a</sup>Department of Chemistry and Biotechnology, Graduate School of Engineering, Tottori University, 4-101 Minami, Koyama-cho, Tottori 680-8552, Japan.

<sup>b</sup>Center for Research on Green Sustainable Chemistry, Tottori University, 4-101 Minami, Koyama-cho, Tottori 680-8552, Japan.

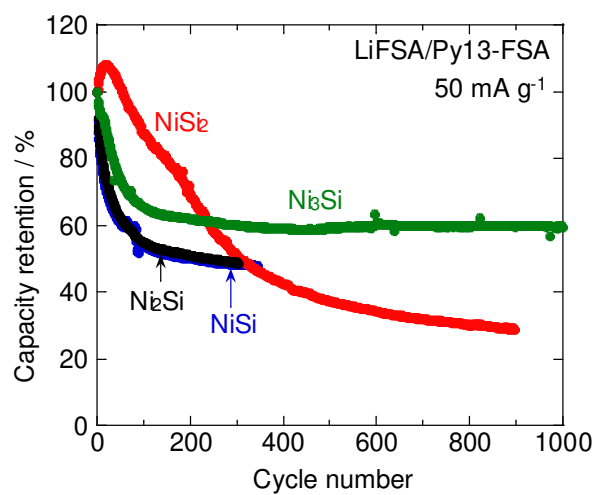
<sup>c</sup>Department of Engineering, Graduate School of Sustainability Science, Tottori University, 4-101 Minami, Koyama-cho, Tottori 680-8552, Japan.



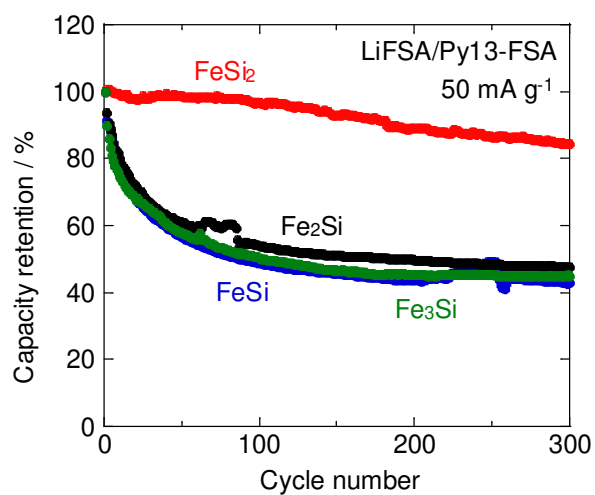
**Fig. S1.** Crystal structure of (a)  $\text{NiSi}_2$ , (b)  $\text{NiSi}$ , (c)  $\text{Ni}_2\text{Si}$  and (d)  $\text{Ni}_3\text{Si}$ .



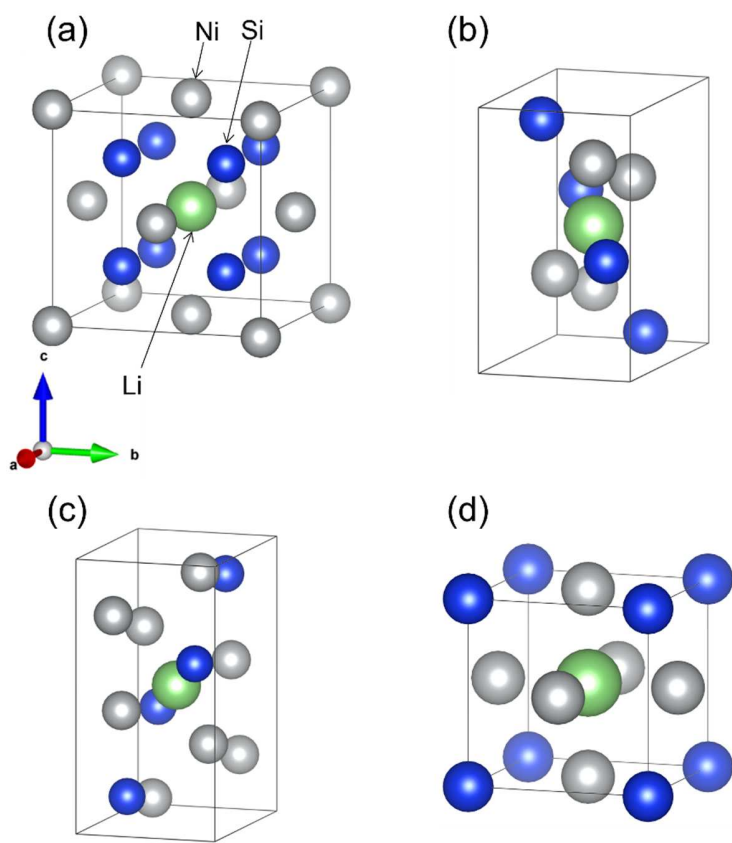
**Fig. S2.** Crystal structure of (a)  $\text{FeSi}_2$ , (b)  $\text{FeSi}$  and (c)  $\text{Fe}_3\text{Si}$ . The structure of  $\text{Fe}_2\text{Si}$  cannot be determined because  $Z$  number is unknown.



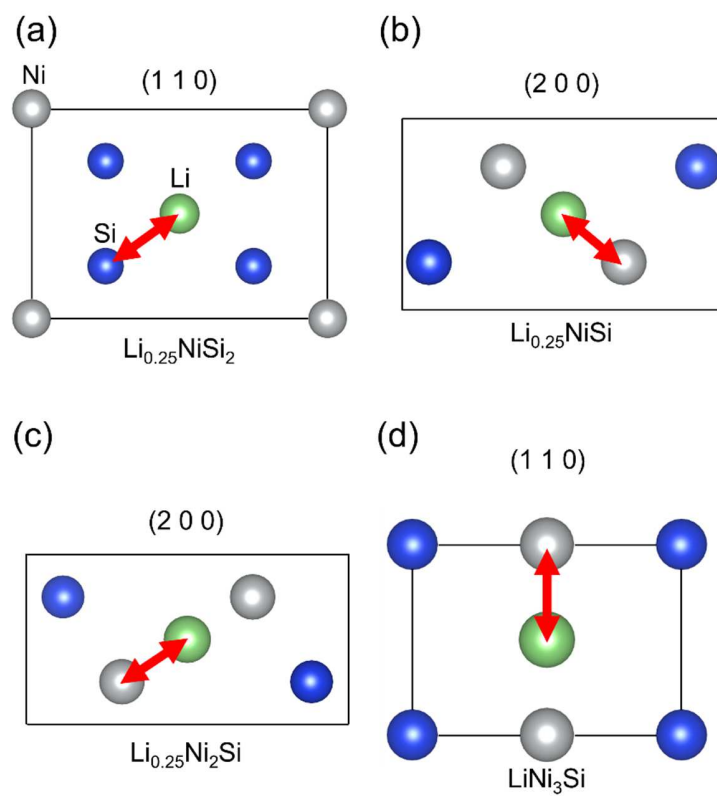
**Fig. S3.** Capacity retention of NiSi<sub>x</sub> ( $x = 2, 1, 1/2$  and  $1/3$ ) electrodes under a current density of 50 mA g<sup>-1</sup> in 1 M LiFSA/Py13-FSA.



**Fig. S4.** Capacity retention of FeSi<sub>x</sub> ( $x = 2, 1, 1/2$  and  $1/3$ ) electrodes under a current density of 50 mA g<sup>-1</sup> in 1 M LiFSA/Py13-FSA.



**Fig. S5.** Crystal structure of each lithiated  $\text{NiSi}_x$  ( $x = 2, 1, 1/2$  or  $1/3$ ). While  $\text{Li}_{0.25}\text{NiSi}_2$ ,  $\text{Li}_{0.25}\text{NiSi}$  and  $\text{Li}_{0.25}\text{Ni}_2\text{Si}$  were optimized crystal structure,  $\text{LiNiSi}_3$  did not converge. Hence, the charge density was calculated with Li at the center of the lattice as in the former three crystals.



**Fig. S6.** Distance of between Li and nearest-neighbour atom in each Ni-Si lattice.

**Table S1.** Molar ratio, rotational speed, and treatment time for preparation of each silicide powder.

Silicide	Molar ratio	Rotational speed / rpm	Treatment time / h
NiSi <sub>2</sub>	Ni : Si = 1 : 2	380	20
NiSi	Ni : Si = 1 : 1	380	10
Ni <sub>2</sub> Si	Ni : Si = 2 : 1	380	10
Ni <sub>3</sub> Si	Ni : Si = 3 : 1	380	40
FeSi <sub>2</sub>	Fe : Si = 1 : 3.3	380	100
FeSi	Fe : Si = 1 : 1	380	10
Fe <sub>2</sub> Si	Fe : Si = 2 : 1	380	10
Fe <sub>3</sub> Si	Fe : Si = 3 : 1	380	10



**Table S2.** Crystallite sizes of (upper) NiSi<sub>x</sub> and (lower) FeSi<sub>x</sub> ( $x = 2, 1, 1/2$  or  $1/3$ ).

	NiSi <sub>2</sub>	NiSi	Ni <sub>2</sub> Si	Ni <sub>3</sub> Si
Crystallite size / nm	9.2	11.8	13.9	9.0
	FeSi <sub>2</sub>	FeSi	Fe <sub>2</sub> Si	Fe <sub>3</sub> Si
Crystallite size / nm	13.7	22.2	6.3	11.7

**Table S3.** Molar mass, density and crystal structure for each silicide.

Silicide	Molar mass / g mol <sup>-1</sup>	Density / g cm <sup>-3</sup>	Crystal system
NiSi <sub>2</sub>	114.87	4.83	Cubic
NiSi	86.78	5.96	Orthorhombic
Ni <sub>2</sub> Si	145.47	7.37	Orthorhombic
Ni <sub>3</sub> Si	204.17	7.84	Cubic
FeSi <sub>2</sub>	112.02	5.04	Tetragonal
FeSi	83.93	6.17	Cubic
Fe <sub>2</sub> Si	139.78	–	Cubic
Fe <sub>3</sub> Si	195.62	7.1 <sup>40</sup>	Cubic