

Electronic Supplementary Information (ESI)

Electrocatalytic CO₂ reduction reaction on dual-metal- and nitrogen-doped graphene: Coordination environment effect of active sites

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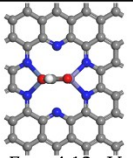
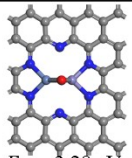
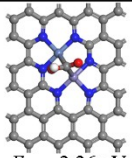
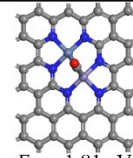
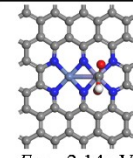
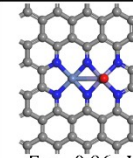
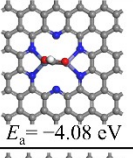
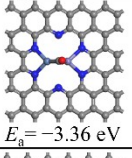
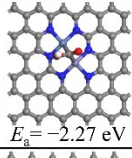
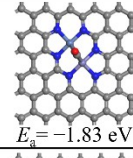
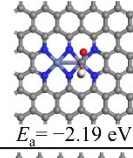
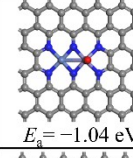
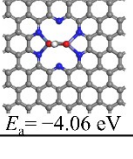
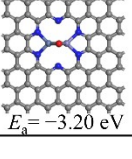
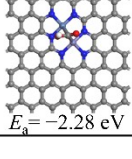
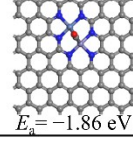
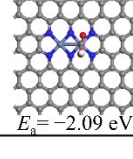
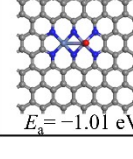
	Model 1		Model 2		Model 3	
Surpercell	*COOH	*CO	*COOH	*CO	*COOH	*CO
4 × 3	 $E_a = -4.12$ eV	 $E_a = -3.28$ eV	 $E_a = -2.26$ eV	 $E_a = -1.81$ eV	 $E_a = -2.14$ eV	 $E_a = -0.96$ eV
5 × 3	 $E_a = -4.08$ eV	 $E_a = -3.36$ eV	 $E_a = -2.27$ eV	 $E_a = -1.83$ eV	 $E_a = -2.19$ eV	 $E_a = -1.04$ eV
6 × 4	 $E_a = -4.06$ eV	 $E_a = -3.20$ eV	 $E_a = -2.28$ eV	 $E_a = -1.86$ eV	 $E_a = -2.09$ eV	 $E_a = -1.01$ eV

Fig. S1 The adsorption energies (E_a) of *COOH and *CO on 4 × 3, 5 × 3, and 6 × 4 supercells of Ni/Fe-N₆-Gra (Model 1, Model 2, Model 3) surfaces, respectively.

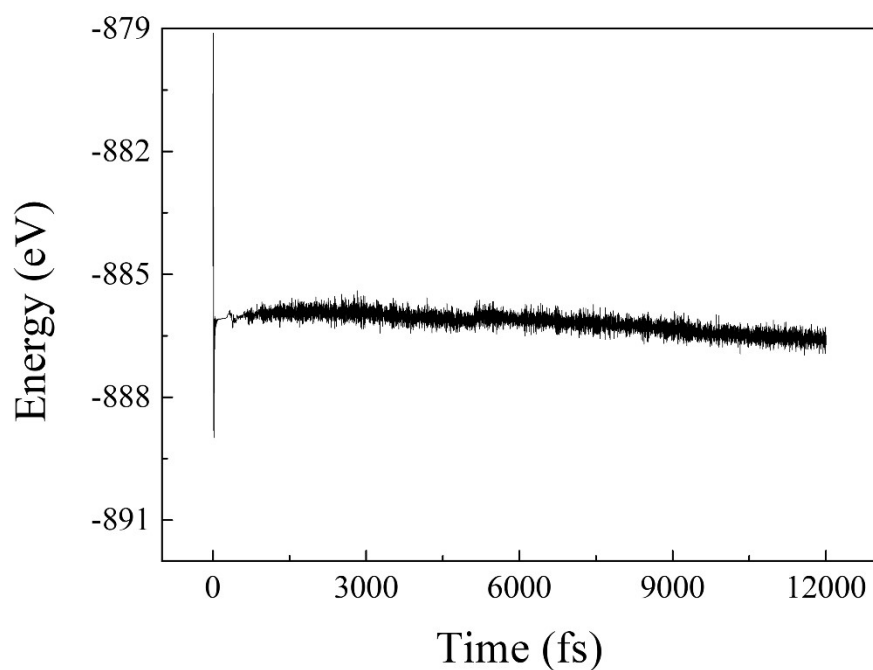


Fig. S2 The AIMD simulation result of water/NiFe-N₆-Gra-Model 2 interface.

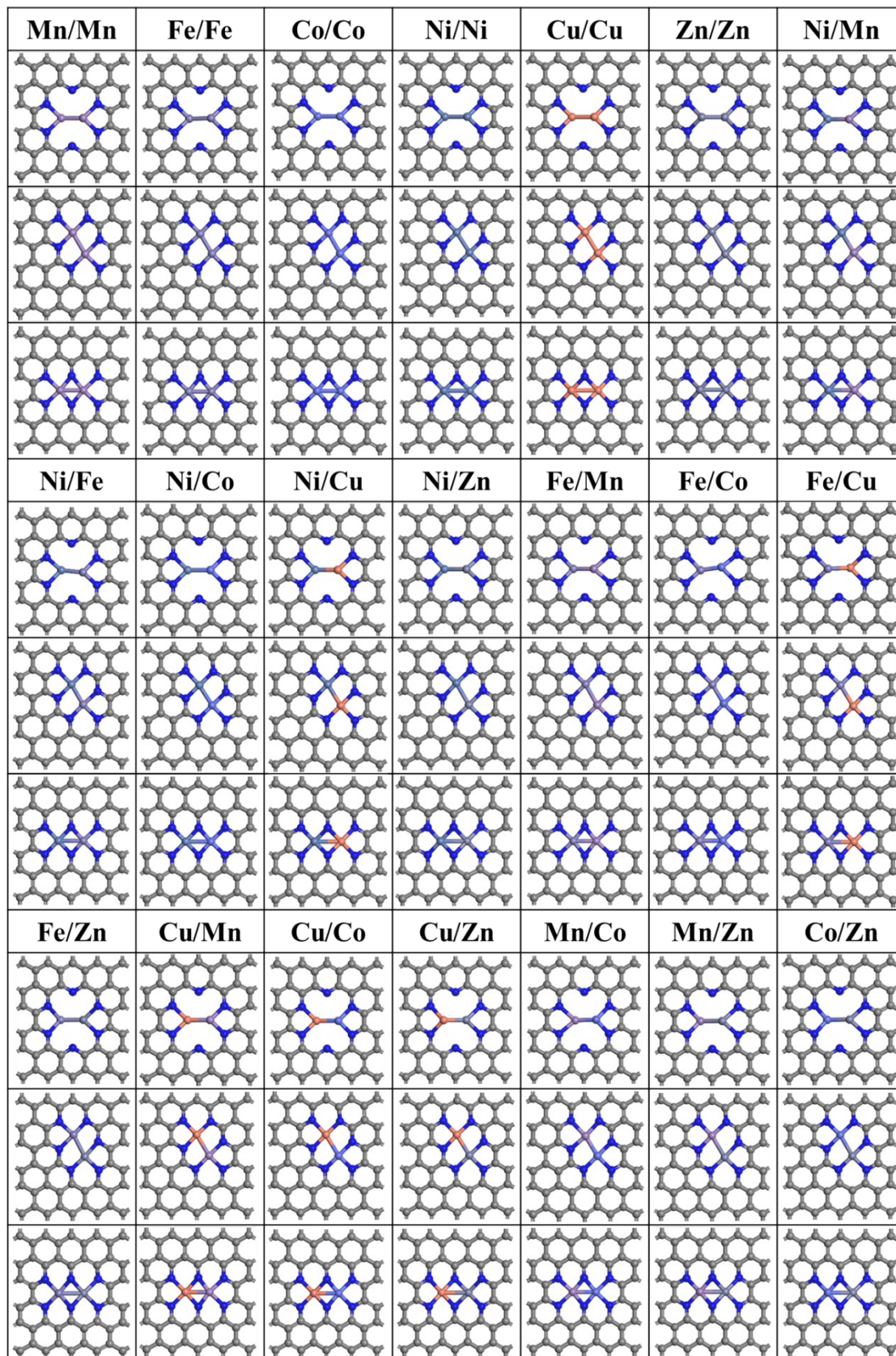


Fig. S3 Optimized geometries of M_1/M_2 -N₆-Gra substrate on Model 1, Model 2, and Model 3 (top view).

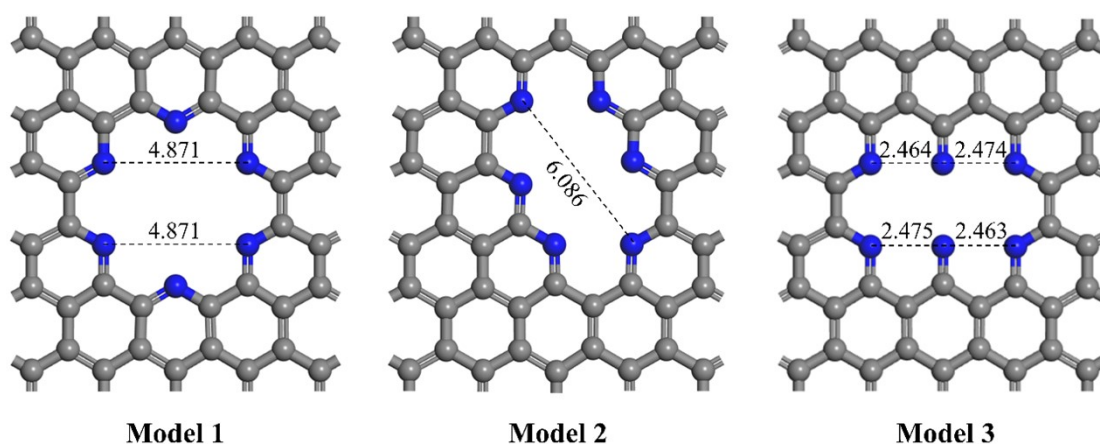


Fig. S4 Optimized geometries of N-doped graphene substrate (N_6 -Gra) on Model 1, Model 2, and Model 3. Unit is in Å.

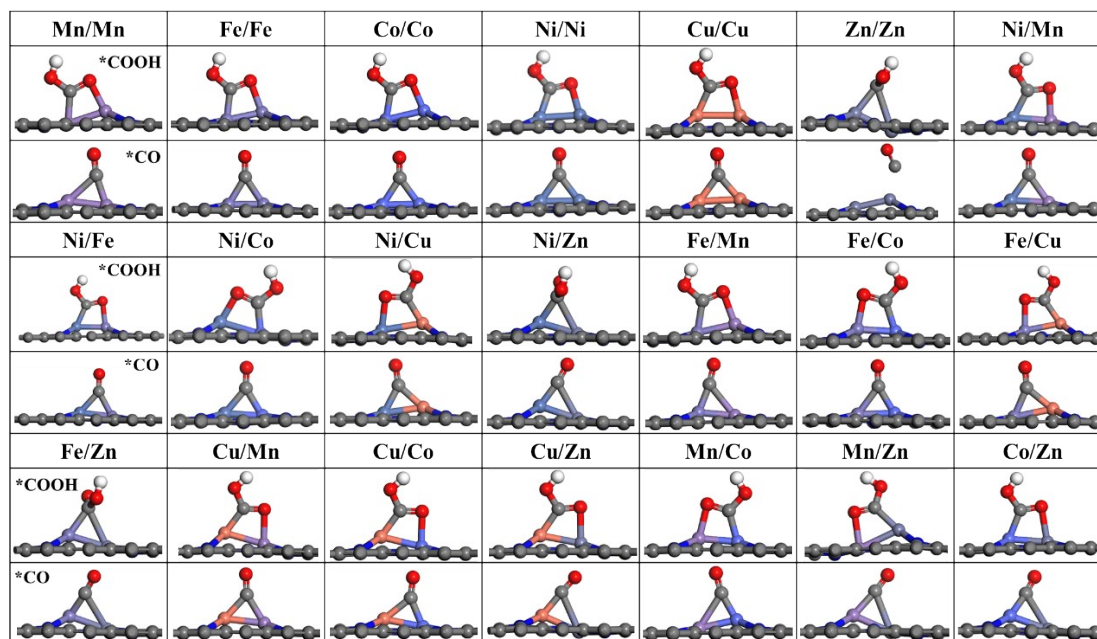


Fig. S5 Optimized geometries of *COOH and *CO adsorbed M_1/M_2 - N_6 -Gra on Model 1 (side view).

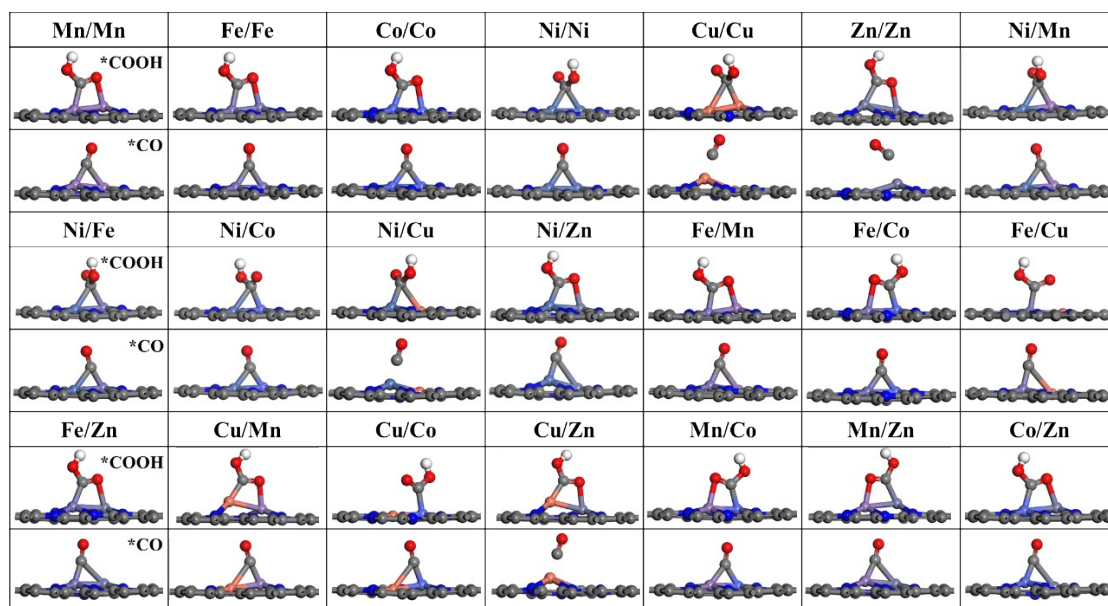


Fig. S6 Optimized geometries of *COOH and *CO adsorbed M_1/M_2 -N₆-Gra on Model 2 (side view).

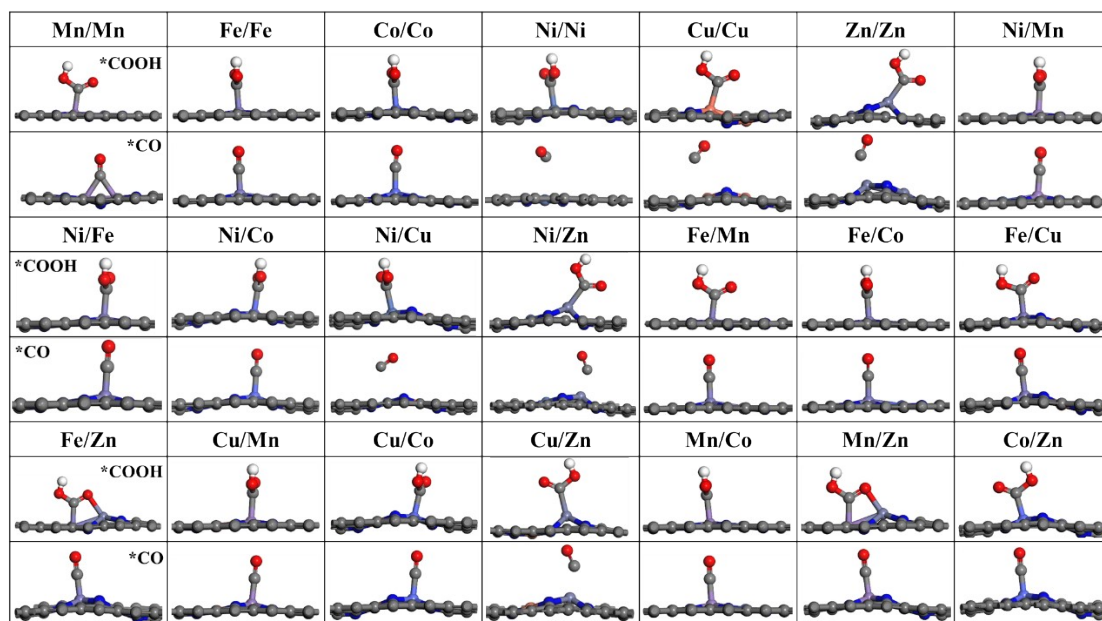


Fig. S7 Optimized geometries of *COOH and *CO adsorbed M_1/M_2 -N₆-Gra on Model 3 (side view).

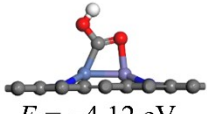
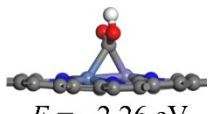
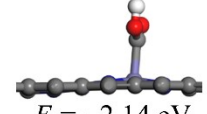
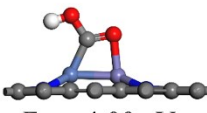
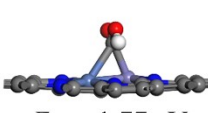
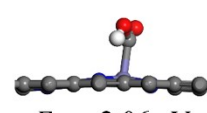
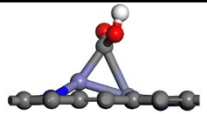
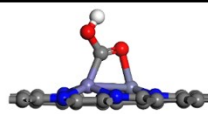
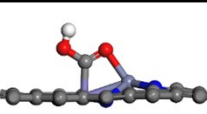
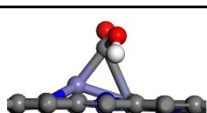
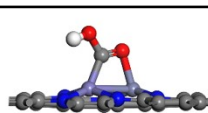
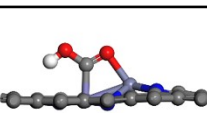
	*COOH	Model 1	Model 2	Model 3
Ni/Fe-N ₆ -Gra	H-up	 $E_a = -4.12$ eV	 $E_a = -2.26$ eV	 $E_a = -2.14$ eV
	H-down	 $E_a = -4.00$ eV	 $E_a = -1.77$ eV	 $E_a = -2.06$ eV
Fe/Zn-N ₆ -Gra	H-up	 $E_a = -2.97$ eV	 $E_a = -3.08$ eV	 $E_a = -2.35$ eV
	H-down	 $E_a = -2.88$ eV	 $E_a = -2.98$ eV	 $E_a = -2.26$ eV

Fig. S8 Optimized geometries and adsorption energies (E_a) of adsorbed cis-COOH (H-up) and trans-COOH (H-down) on Ni/Fe-N₆-Gra and Fe/Zn-N₆-Gra.

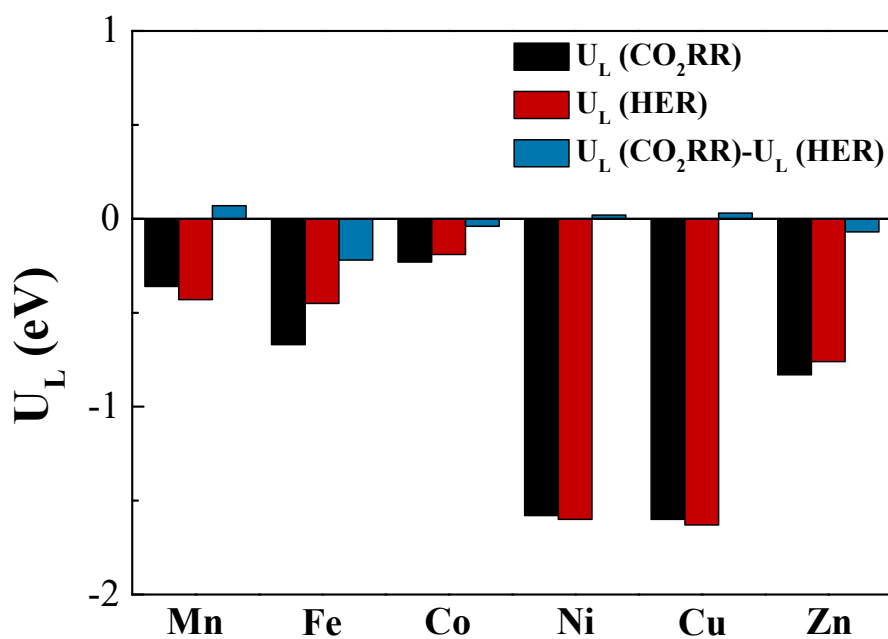


Fig. S9 Calculated limiting potentials for HER (U_L (HER)), CO₂RR (U_L (CO₂RR)), and (U_L (CO₂RR) - U_L (HER)) on M-N₄-Gra (M = Mn, Fe, Co, Ni, Cu, and Zn).

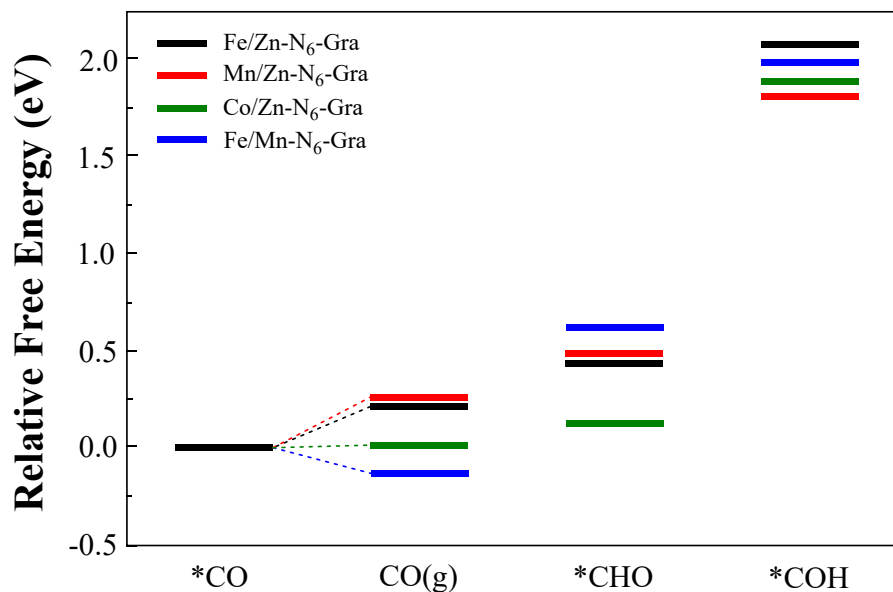


Fig. S10 Relative free energy diagrams of the hydrogenation of CO to *CHO and *COH.

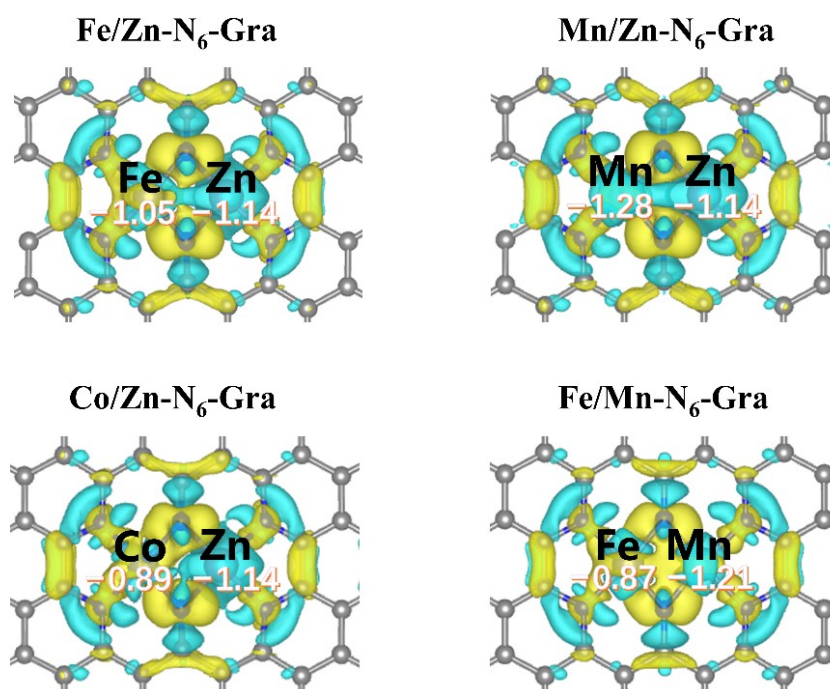


Fig. S11 The charge density difference for Fe/Zn-N₆-Gra, Mn/Zn-N₆-Gra, Co/Zn-N₆-Gra, and Fe/Mn-N₆-Gra. Yellow: charge accumulation; Cyan: charge depletion. The isosurface value is set to 0.003 e/Bohr³.

Table S1. The formation energies (E_f) on M_1/M_2-N_6 -Gra.

M_1/M_2-N_6 -Gra	E_f (eV)		
	Model 1	Model 2	Model 3
Ni/Mn	1.12	-2.20	-6.91
Ni/Fe	3.01	-0.91	-5.75
Ni/Co	2.09	-1.27	-5.94
Ni/Ni	1.70	-1.93	-6.03
Ni/Cu	1.60	-1.39	-4.58
Ni/Zn	0.56	-1.86	-5.22
Fe/Mn	1.98	-0.70	-7.27
Fe/Fe	2.83	-0.02	-6.19
Fe/Co	3.26	-0.68	-6.27
Fe/Cu	2.16	-0.43	-4.00
Fe/Zn	0.97	-0.58	-4.79
Cu/Mn	2.33	-1.47	-5.15
Cu/Co	2.11	-0.98	-4.27
Cu/Cu	1.65	-4.90	-3.40
Cu/Zn	0.42	-1.90	-4.07
Mn/Mn	1.11	-1.42	-8.12
Mn/Co	1.39	-1.41	-7.46
Mn/Zn	1.10	-1.70	-5.96
Co/Co	2.61	-0.87	-6.24
Co/Zn	0.88	-1.21	-4.89
Zn/Zn	-1.47	-2.22	-4.58

Table S2. Bond lengths between bimetals on M₁/M₂-N₆-Gra.

M ₁ /M ₂ -N ₆ -Gra	$d(\text{M}_1\text{—M}_2)$ (Å)		
	Model 1	Model 2	Model 3
Ni/Mn	2.252	2.264	2.462
Ni/Fe	2.216	2.274	2.438
Ni/Co	2.247	2.307	2.423
Ni/Ni	2.342	2.341	2.580
Ni/Cu	2.298	2.381	2.547
Ni/Zn	2.294	2.425	2.546
Fe/Mn	2.134	2.337	2.236
Fe/Fe	2.281	2.208	2.221
Fe/Co	2.108	2.166	2.250
Fe/Cu	2.289	2.397	2.445
Fe/Zn	2.303	2.430	2.451
Cu/Mn	2.230	2.398	2.325
Cu/Co	2.290	2.395	2.428
Cu/Cu	2.329	2.382	2.576
Cu/Zn	2.241	2.386	2.632
Mn/Mn	2.217	2.372	2.266
Mn/Co	2.184	2.224	2.258
Mn/Zn	2.315	2.491	2.514
Co/Co	2.236	2.229	2.258
Co/Zn	2.310	2.408	2.470
Zn/Zn	2.221	2.546	2.553

Table S3. Free energy change values of elementary steps for CO₂RR at $U = 0$ V on M₁/M₂-N₆-Gra. Unit is in eV.

M ₁ /M ₂ - N ₆ -Gra	Model 1			Model 2			Model 3		
	ΔG_1	ΔG_2	ΔG_3	ΔG_1	ΔG_2	ΔG_3	ΔG_1	ΔG_2	ΔG_3
Ni/Mn	-1.13	-0.30	1.56	0.30	-0.56	0.39	0.31	-0.36	0.18
Ni/Fe	-1.61	-0.61	2.35	0.23	-0.98	0.87	0.39	-0.29	0.03
Ni/Co	-0.47	-1.13	1.73	-0.08	-0.81	1.02	0.54	-0.08	-0.33
Ni/Ni	-0.19	-1.33	1.65	0.62	-0.58	0.09	1.47	-0.66	-0.68
Ni/Cu	-0.23	-1.10	1.46	0.73	-0.58	-0.03	1.49	-0.63	-0.73
Ni/Zn	-0.73	-0.87	1.73	0.09	-0.55	0.59	1.06	-0.34	-0.59
Fe/Mn	-1.20	-0.29	1.62	-0.65	-0.68	1.46	0.26	0.01	-0.14
Fe/Fe	-0.82	-0.55	1.50	-0.32	-0.77	1.22	0.48	-0.06	-0.29
Fe/Co	-1.54	-0.65	2.32	0.10	-0.91	0.94	0.42	0.01	-0.30
Fe/Cu	-0.62	-0.72	1.47	0.01	-0.68	0.80	0.22	-0.34	0.25
Fe/Zn	-0.47	-0.75	1.35	-0.56	-0.65	1.34	0.19	-0.30	0.24
Cu/Mn	-2.32	-0.05	2.50	0.17	-0.55	0.51	0.25	-0.39	0.27
Cu/Co	-0.57	-0.97	1.67	0.13	-0.64	0.64	0.42	0.04	-0.33
Cu/Cu	-0.33	-0.97	1.43	1.09	-0.79	-0.17	1.83	-0.98	-0.72
Cu/Zn	-0.59	0.07	0.65	0.40	-0.19	-0.08	0.86	-0.10	-0.63
Mn/Mn	-0.81	-0.38	1.32	-0.57	-0.75	1.45	0.08	0.32	-0.27
Mn/Co	-1.15	-0.46	1.74	-0.53	-0.56	1.22	0.34	-0.42	0.21
Mn/Zn	-1.26	-0.60	1.99	-0.63	-0.32	1.08	0.15	-0.28	0.26
Co/Co	-0.85	-0.84	1.82	0.09	-1.31	1.35	0.89	0.06	-0.82
Co/Zn	-0.55	-0.99	1.67	-0.45	-0.62	1.20	0.20	-0.09	0.02
Zn/Zn	0.81	-0.18	-0.50	-0.50	1.08	-0.45	1.01	-0.37	-0.51

Note: The red values represent the maximum free energy change values on the reaction pathway.

Table S4. Adsorption energies of *CO and *COOH on M₁/M₂-N₆-Gra. Unit is in eV.

M ₁ /M ₂ -N ₆ - Gra	Model 1		Model2		Model 3	
	E_a (*COOH)	E_a (*CO)	E_a (*COOH)	E_a (*CO)	E_a (*COOH)	E_a (*CO)
Ni/Mn	-3.65	-2.49	-2.17	-1.34	-2.18	-1.11
Ni/Fe	-4.12	-3.28	-2.26	-1.81	-2.14	-0.96
Ni/Co	-2.98	-2.66	-2.58	-1.96	-1.96	-0.58
Ni/Ni	-2.72	-2.61	-1.90	-1.00	-1.01	-0.09
Ni/Cu	-2.74	-2.40	-1.77	-0.89	-1.01	-0.07
Ni/Zn	-3.20	-2.69	-2.42	-1.49	-1.40	-0.27
Fe/Mn	-3.69	-2.53	-3.18	-2.40	-2.25	-0.80
Fe/Fe	-3.33	-2.41	-2.83	-2.17	-2.04	-0.64
Fe/Co	-4.06	-3.25	-2.45	-1.89	-2.09	-0.64
Fe/Cu	-3.13	-2.39	-2.47	-1.76	-2.28	-1.18
Fe/Zn	-2.97	-2.27	-3.08	-2.28	-2.35	-1.17
Cu/Mn	-4.86	-3.46	-2.34	-1.42	-2.25	-1.20
Cu/Co	-3.07	-2.59	-2.38	-1.56	-2.09	-0.76
Cu/Cu	-2.82	-2.36	-1.37	-0.75	-0.65	-0.08
Cu/Zn	-3.12	-1.60	-2.13	-0.85	-1.63	-0.21
Mn/Mn	-3.32	-2.23	-3.09	-2.34	-2.41	-0.65
Mn/Co	-3.65	-2.66	-3.09	-2.16	-2.16	-1.15
Mn/Zn	-3.75	-2.93	-3.13	-2.00	-2.39	-1.86
Co/Co	-3.35	-2.75	-2.45	-2.30	-1.61	-0.11
Co/Zn	-3.04	-2.61	-2.97	-2.14	-2.29	-0.96
Zn/Zn	-1.68	-0.38	-3.03	-0.40	-1.48	-0.34

Table S5. Adsorption free energies of *CO and *COOH on M₁/M₂-N₆-Gra. Unit is in eV.

M ₁ /M ₂ -N ₆ -Gra	Model 1		Model2		Model 3	
	ΔG_{*COOH}	ΔG_{*CO}	ΔG_{*COOH}	ΔG_{*CO}	ΔG_{*COOH}	ΔG_{*CO}
Ni/Mn	-3.12	-2.35	-1.69	-1.18	-1.68	-0.97
Ni/Fe	-3.60	-3.14	-1.78	-1.64	-1.59	-0.81
Ni/Co	-2.46	-2.52	-2.07	-1.81	-1.46	-0.45
Ni/Ni	-2.18	-2.44	-1.38	-0.88	-0.52	-0.11
Ni/Cu	-2.23	-2.25	-1.27	-0.77	-0.50	-0.06
Ni/Zn	-2.73	-2.52	-1.90	-1.38	-0.70	-0.18
Fe/Mn	-3.19	-2.41	-2.64	-2.25	-1.74	-0.65
Fe/Fe	-2.81	-2.29	-2.31	-2.01	-1.52	-0.50
Fe/Co	-3.54	-3.11	-1.90	-1.73	-1.57	-0.49
Fe/Cu	-2.61	-2.26	-1.99	-1.59	-1.78	-1.04
Fe/Zn	-2.46	-2.14	-2.55	-2.13	-1.81	-1.03
Cu/Mn	-4.31	-3.29	-1.83	-1.30	-1.74	-1.06
Cu/Co	-2.56	-2.46	-1.86	-1.43	-1.58	-0.46
Cu/Cu	-2.32	-2.22	-0.91	-0.62	-0.17	-0.07
Cu/Zn	-2.58	-1.44	-1.60	-0.71	-1.13	-0.17
Mn/Mn	-2.81	-2.11	-2.56	-2.24	-1.92	-0.52
Mn/Co	-3.14	-2.53	-2.53	-2.01	-1.66	-1.00
Mn/Zn	-3.25	-2.78	-2.62	-1.87	-1.84	-1.05
Co/Co	-2.84	-2.61	-1.91	-2.14	-1.11	0.03
Co/Zn	-2.55	-2.46	-2.44	-1.99	-1.79	-0.81
Zn/Zn	-1.18	-0.28	-2.50	-0.34	-0.99	-0.28

Table S6. Bond lengths ($d(M_1-C)$ and $d(M_2-C)$) of *CO adsorbed M_1/M_2-N_6 -Gra. Unit is in Å.

M_1/M_2-N_6 -Gra	Model 1		Model 2		Model 3	
	$d(M_1-C)$	$d(M_2-C)$	$d(M_1-C)$	$d(M_2-C)$	$d(M_1-C)$	$d(M_2-C)$
Ni/Mn	1.832	1.986	2.043	1.816	—	1.745
Ni/Fe	1.905	1.871	2.119	1.747	—	1.734
Ni/Co	1.901	1.835	1.992	1.766	—	1.739
Ni/Ni	1.852	1.852	1.929	1.924	—	—
Ni/Cu	1.810	1.911	1.771	2.829	—	—
Ni/Zn	1.701	2.378	1.736	2.745	—	—
Fe/Mn	1.752	2.313	1.728	2.318	1.725	—
Fe/Fe	1.891	1.912	1.879	1.879	1.726	—
Fe/Co	2.017	1.807	1.859	1.868	1.726	—
Fe/Cu	1.840	1.942	1.719	2.728	1.722	—
Fe/Zn	1.750	2.508	1.740	2.403	1.721	—
Cu/Mn	1.907	1.935	2.404	1.804	—	1.747
Cu/Co	1.962	1.787	2.512	1.748	—	1.740
Cu/Cu	1.855	1.862	—	—	—	—
Cu/Zn	1.777	2.528	—	—	—	—
Mn/Mn	2.173	1.801	1.785	2.158	1.973	1.973
Mn/Co	2.350	1.736	1.938	1.845	1.749	—
Mn/Zn	1.865	2.489	1.795	2.294	1.765	—
Co/Co	1.858	1.886	1.846	1.848	1.818	—
Co/Zn	1.719	2.472	1.725	2.442	1.731	—
Zn/Zn	—	—	—	—	—	—

Table S7. Bond lengths of *COOH adsorbed M₁/M₂-N₆-Gra on Model 1, Model 2, and Model 3. Unit is in Å.

Model 1	$d(M_1-C)$	$d(M_1-O)$	$d(M_2-C)$	$d(M_2-O)$
Ni/Mn	1.829	—	—	1.920
Ni/Fe	1.854	—	—	2.007
Ni/Co	—	1.927	1.797	—
Ni/Ni	1.821	—	—	1.955
Ni/Cu	—	2.022	1.871	—
Ni/Zn	1.888	—	2.434	—
Fe/Mn	1.870	—	—	1.990
Fe/Fe	1.832	—	—	1.983
Fe/Co	—	1.982	1.853	—
Fe/Cu	—	2.045	1.880	—
Fe/Zn	1.951	—	2.484	—
Cu/Mn	1.871	—	—	2.101
Cu/Co	1.879	—	—	2.042
Cu/Cu	1.870	—	—	2.001
Cu/Zn	1.878	—	—	2.105
Mn/Mn	1.870	—	1.986	—
Mn/Co	—	2.004	1.836	—
Mn/Zn	—	2.198	1.968	—
Co/Co	1.853	—	—	1.957
Co/Zn	1.872	—	—	2.151
Zn/Zn	2.025	—	2.733	—

Model 2	$d(M_1-C)$	$d(M_1-O)$	$d(M_2-C)$	$d(M_2-O)$
Ni/Mn	2.261	—	2.006	—
Ni/Fe	2.189	—	1.952	—
Ni/Co	2.365	—	1.909	—
Ni/Ni	2.046	—	2.070	—
Ni/Cu	1.941	—	2.515	—
Ni/Zn	1.877	—	—	2.139
Fe/Mn	1.840	—	—	2.078
Fe/Fe	1.854	—	—	2.014
Fe/Co	—	2.260	1.857	—
Fe/Cu	1.864	—	—	—
Fe/Zn	1.916	—	—	2.070
Cu/Mn	1.869	—	—	2.052
Cu/Co	—	—	1.862	—

Cu/Cu	2.009	—	2.575	—
Cu/Zn	1.881	—	—	2.044
Mn/Mn	1.924	—	—	2.106
Mn/Co	—	2.181	1.845	—
Mn/Zn	—	2.119	2.020	—
Co/Co	1.855	—	—	2.152
Co/Zn	1.854	—	—	2.084
Zn/Zn	2.003	—	—	2.098

Model 3	$d(M_1—C)$	$d(M_1—O)$	$d(M_2—C)$	$d(M_2—O)$
Ni/Mn	—	—	1.947	—
Ni/Fe	—	—	1.923	—
Ni/Co	—	—	1.897	—
Ni/Ni	1.963	—	—	—
Ni/Cu	1.952	—	—	—
Ni/Zn	—	—	2.019	—
Fe/Mn	1.882	—	—	—
Fe/Fe	1.920	—	—	—
Fe/Co	1.914	—	—	—
Fe/Cu	1.918	—	—	—
Fe/Zn	1.880	—	—	2.091
Cu/Mn	—	—	—	1.951
Cu/Co	—	—	—	1.894
Cu/Cu	2.028	—	—	—
Cu/Zn	—	—	—	2.071
Mn/Mn	1.950	—	—	—
Mn/Co	1.960	—	—	—
Mn/Zn	1.917	—	—	2.065
Co/Co	1.912	—	—	—
Co/Zn	1.886	—	—	—
Zn/Zn	2.020	—	—	—

Table S8. The limiting potentials of HER at $U = 0$ V on Model 2, and Model 3. Unit is in V.

M_1/M_2 -N ₆ -Gra	Model 2	Model 3
Ni/Mn	-0.14	-0.31
Ni/Fe	-0.08	-0.33
Ni/Co	-0.35	-0.50
Ni/Ni	-0.10	-1.37
Ni/Cu	-0.41	-1.37
Ni/Zn	-0.16	-1.15
Fe/Mn	-0.61	-0.29
Fe/Fe	-0.40	-0.41
Fe/Co	-0.10	-0.37
Fe/Cu	-0.17	-0.20
Fe/Zn	-0.67	-0.32
Cu/Mn	-0.07	-0.22
Cu/Co	-0.16	-0.37
Cu/Cu	-0.77	-1.94
Cu/Zn	-0.25	-0.76
Mn/Mn	-0.71	-0.05
Mn/Co	-0.54	-0.31
Mn/Zn	-0.59	-0.33
Co/Co	-0.21	-0.82
Co/Zn	-0.50	-0.25
Zn/Zn	-0.61	-1.00

Table S9. The d -band center (ε_d) of M_1/M_2 -N₆-Gra-Model 3. Unit is in eV.

M_1/M_2 -N ₆ -Gra	$\varepsilon_d (M_1)$	$\varepsilon_d (M_2)$
Mn/Mn	-0.74	-0.74
Fe/Fe	-1.53	-1.52
Co/Co	-1.96	-1.97
Ni/Ni	-1.93	-1.94
Cu/Cu	-3.37	-3.37
Zn/Zn	-5.86	-5.87
Ni/Mn	-2.00	-0.86
Ni/Fe	-2.22	-1.49
Ni/Co	-2.38	-1.67
Ni/Cu	-1.86	-3.69
Ni/Zn	-1.33	-5.88
Fe/Mn	-1.42	-0.95
Fe/Co	-1.33	-1.80
Fe/Cu	-1.53	-3.53
Fe/Zn	-1.55	-6.34
Cu/Mn	-3.68	-1.07
Cu/Co	-3.44	-1.38
Cu/Zn	-3.78	-6.28
Mn/Co	-1.27	-1.75
Mn/Zn	-0.76	-6.18
Co/Zn	-1.18	-6.07

Table S10. The adsorption energies of intermediates, free energy change values of elementary steps and limiting potentials of Fe/Zn-N₆-Gra-Model 3 and Co/Zn-N₆-Gra-Model 3 at different U values and Cu/Fe-N₆-Gra-Model 2 at U = 4 eV. Unit is in eV.

	U value	$E_a(*\text{COOH})$	$E_a(*\text{CO})$	ΔG_1	ΔG_2	ΔG_3	U_L (V)
Fe/Zn- Model 3	U = 3	-1.70	-0.45	0.85	-0.28	-0.44	-0.85
	U = 4	-1.54	-0.32	1.00	-0.25	-0.61	-1.00
	U = 5	-1.09	-0.27	1.45	-0.73	-0.59	-1.45
	U = 6	-0.91	-0.36	1.61	-0.97	-0.50	-1.61
Co/Zn- Model 3	U = 3	-2.29	-0.96	0.44	-0.14	-0.17	-0.44
	U = 4	-2.05	-0.72	0.61	-0.23	-0.25	-0.61
	U = 5	-1.90	-0.64	0.81	-0.35	-0.34	-0.81
	U = 6	-1.69	-0.54	1.01	-0.46	-0.43	-1.01
Cu/Fe- Model 2	U = 4	-1.40	-0.80	1.08	-0.84	-0.11	-1.08

Table S11. Magnetic moments of metal atoms at different U values on Fe/Zn and Co/Zn-N₆-Gra-Model 3 Unit is in μ_B .

U value	Fe/Zn		Co/Zn	
	mag(M ₁)	mag(M ₂)	mag(M ₁)	mag(M ₂)
U = 3	2.625	0.011	-1.123	0.077
U = 4	2.724	0.015	-1.060	0.002
U = 5	2.992	0.001	-1.089	0.001
U = 6	3.198	0.002	-1.733	0.001