Supporting Information

for

Biaxial Pseudorotaxane Secondary Assembly for

Phosphorescent Cellular Imaging

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Instruments.

NMR spectra were recorded on Bruker 400 MHz instrument, and chemical shifts were recorded in parts per million (ppm). High resolution mass (HRMS) spectra were performed on Varian 7.0T FTMS with ESI or MALDI source. TEM images were acquired by a high-resolution transmission electron microscope (Philips Tecnai G2 20S-TWIN microscope) operating at an accelerating voltage of 200 keV. The samples were prepared by placing a drop of solution onto a carbon-coated copper grid and air-dried. The morphological information was directly obtained from the fresh TEM samples without staining. Transmission spectra were recorded on a Shimadzu UV-3600 spectrophotometer in a quartz cell (light path 10 mm) at 25 °C with a PTC-348WI temperature controller. Dynamic light scattering (DLS) was recorded on BI-200SM (Brookhaven Company) at 25 °C. Confocal florescence imaging was recorded with Olympus FV1000.

Synthesis of BPTN



Figure S1 Synthetic route of BPTN.



Figure S2 ¹H NMR spectra (400 MHz, D₂O, 298K) of BPTN.



Figure S3 ¹³C NMR spectra (100 MHz, D₂O, 298K) of BPTN.



Figure S4 ESI-HRMS spectrum of BPTN.



Figure S5 ¹H NMR spectroscopy of BPTN with (bottom) 0, (middle) 0. 5 and (upper) 1 eq. CB[7] in D_2O at 298 K.



Figure S6 ¹H NMR spectra (400 MHz, D_2O , 298K) of BPTN \subset CB[8]. The binding ratio of BPTN and CB[8] can be calculated as 2:1 via the integral area.



Figure S7 Job's plot of BPTN and CB[7] in PBS solution ([BPTN]+[CB[7]] = 50 μ M) at 298 K.



Figure S8 UV-vis absorbtion spectra of BPTN ([BPTN] = 20 μ M) upon the addition of CB[7] in PBS solution.



Figure S9 UV-vis absorbtion spectra of BPTN ([BPTN] = $20 \mu M$) upon the addition of CB[8] in PBS solution.



Figure S10 Excitation spectrum of BPPY \subset CB[8] in PBS solution ([BPTN] = 2[CB[8]] = 10 μ M).



Figure S11 Fluorescence spectra of BPTN (10 $\mu M)$ and CB[8] at concentrations of 0, 2.5 and 5 $\mu M.$



Figure S12 Photoluminescence spectra of BPTNCCB[8] at different temperature.



Figure S13 The quantum yield of BPTN \subset CB[8] ([BPTN] = 2[CB[8]] = 10 μ M).



Figure S14 Photoluminescence spectra of BPTN (10 $\mu M)$ and CB[7] at concentrations of 0, 5 and 10 $\mu M.$



Figure S15 (a) Fluorescence spectra and (b) time-lapse photoluminescence spectra of BPTN (10 μ M) and CB[7] at concentrations of 0, 5 and 10 μ M.



Figure S16 The Tyndall effect of solution of BPTN \subset CB[8] and BPTN \subset CB[8]@SSP[4] ([BPTN] = 2[CB[8]] = 10 μ M, [SSP[4]] = 20 μ M).



Figure S17 Stern-Volmer curves of BPTNCCB[8]@SSP[4] system.



Figure S18 DLS data of BPTN \subset CB[8]@SSP[4] with GSH and BPTN \subset CB[8]@SSP[4] in pH = 6.5 solution ([BPTN] = 2[CB[8]] = 10 μ M, SSP[4] = 20 μ M, GSH = 2 mM).



Figure S19 The Tyndall effect of solution of BPTN \subset CB[8]@SSP[4] with (a) GSH or (b) under pH = 6.5 ([BPTN] = 2[CB[8]] = 10 μ M, [SSP[4]] = 20 μ M).



Figure S20 Time-lapse photoluminescence spectra of BPTN \subset CB[8]@SSP[4] with GSH or under pH = 6.5 under argon atmosphere.



Figure S21 The quantum yield of BPTN \subset CB[8]@SSP[4] with (a) GSH or (b) adjusting pH to 6.5 ([BPTN] = 2[CB[8]] = 10 μ M, [SSP[4]] = 20 μ M, [GSH] = 2 mM).



Figure S22 Time resolved photoluminescence decay curves of BPTN \subset CB[8]@SSP[4] at pH 6.5 in PBS solution ([BPTN] = 2[CB[8]] = 10 μ M, [SSP[4]] = 20 μ M).

	Lifetime (ms)	Quantum Yield (%)	Normalized Intensity of Emission at 505 nm
BPTN⊂CB[8]	0.60	2.10	1.00
BPTN⊂CB[8]@SSP[4] with GSH	4.03	1.35	0.28
BPTN⊂CB[8]@SSP[4] at pH 6.5	1.33	0.81	0.46

Table S1 Phosphorescent properties of BPTN_CB[8] and BPTN_CB[8]@SSP[4] with GSH or adjusting pH to 6.5.



Figure S23 Cell viability of (a) A549, (b) Hela, (c)MCF-7 and (d) 293T cells incubated with BPTN, BPTN_CB[8] and BPTN_CB[8]@SSP[4] at different concentration.



Figure S24 Confocal fluorescence images of living A549 cells incubated with (a) BPTN and (b) BPTN@SSP[4] (scale bar = $20 \ \mu m$).