

Supporting Information

Urea-Assisted Synthesis of Carbon-Doped BiNbO₄ with Oxygen Vacancies and Visible Light Photocatalytic Applications

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SI1. Micro-Raman spectra of BiNbO₄ and U-BiNbO₄.

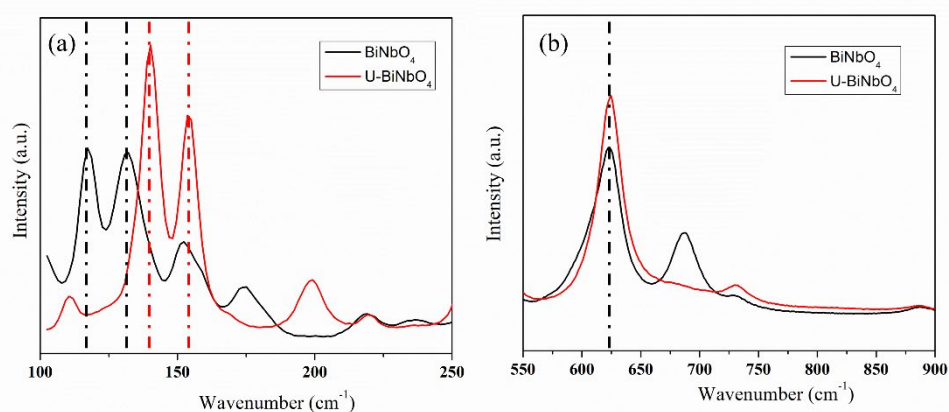


Figure S1. Raman spectra in the (a) 100 – 250 cm⁻¹ and (b) 550 – 900 cm⁻¹ regions for pristine BiNbO₄ and U-BiNbO₄.

S12. XPS spectrum of Bi4f and Nb3d for pristine BiNbO₄ (panels A, C) and U-BiNbO₄ (panels B, D), respectively.

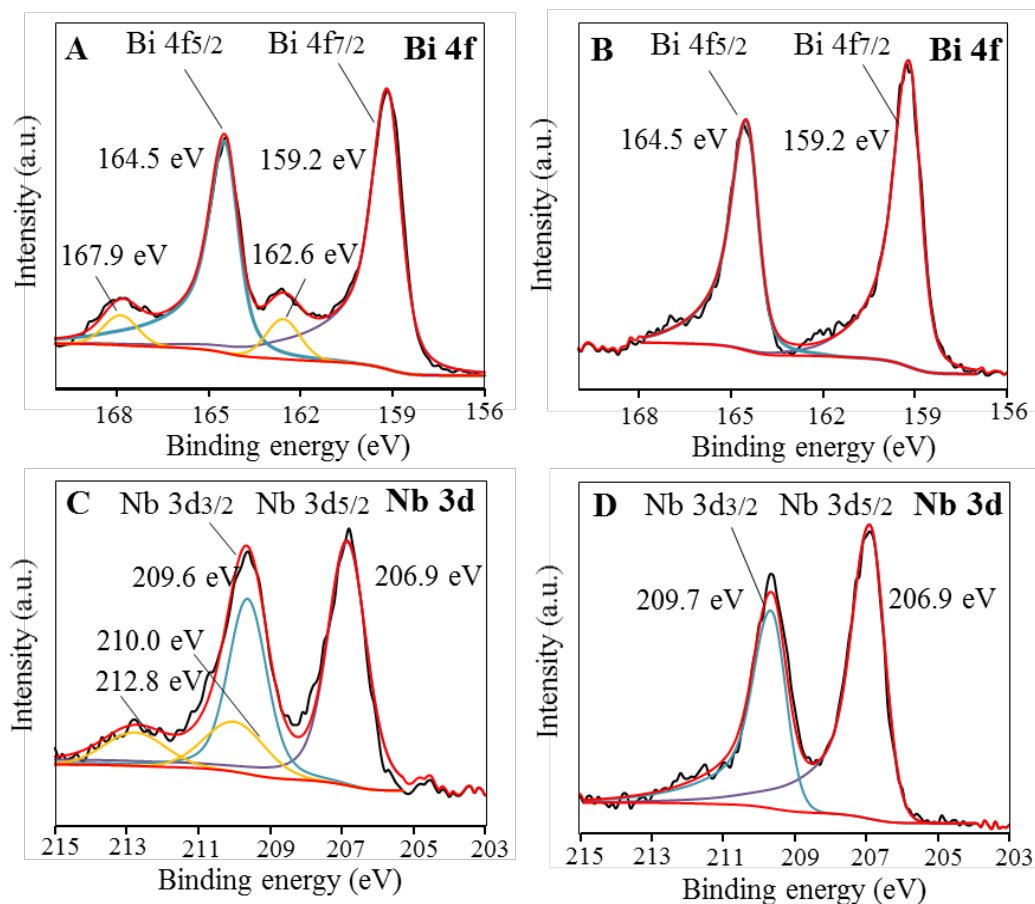


Figure S2. XPS spectra of Bi4f and Nb3d for pristine BiNbO₄ (panels A, C) and U-BiNbO₄ (panels B, D), respectively

The Bi4f spectrum of pristine BiNbO₄ suggest the presence of Bi³⁺ (with main peaks at 159.2 eV and 164.5 eV, respectively), while the Bi4f spectrum of U-BiNbO₄ also consist of Bi(+3) only (with more predominate peaks at 159.2 eV and 164.5 eV, respectively)¹. The Nb3d spectrum of pristine BiNbO₄ also propose the presence of two chemical states of Nb(+4) (206.9 eV and 209.6 eV, respectively) and Nb(+5) (210.0 eV and 212.8 eV, respectively), while the U-BiNbO₄ has only Nb(+4) on the surface². Reduced amounts of oxygen at Bi4f and Nb3d chemical states for U-BiNbO₄ catalyst is another evidence of the enhanced presence of surface oxygen vacancies. No traces of nitrogen were detected in both samples.

SI3. EDS mapping images and spectrum of BiNbO₄ (panels (A) and (C)) and U-BiNbO₄ (panels (B) and (D)), respectively.

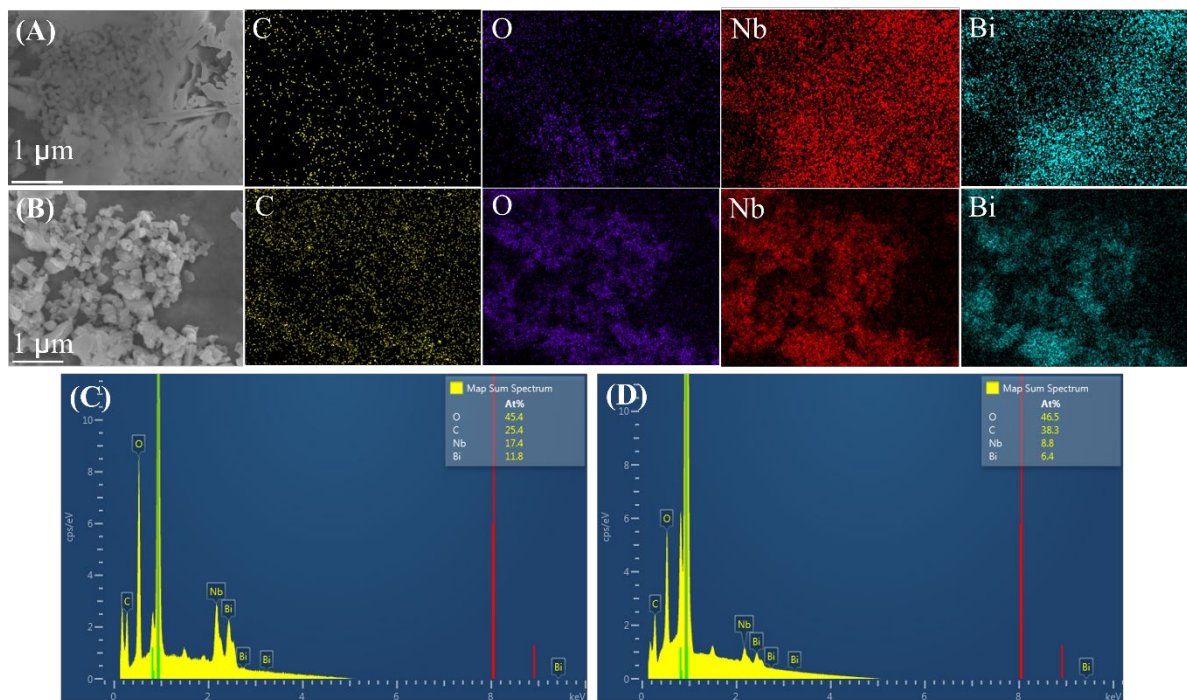


Figure S3. EDS mapping and spectral images for pristine BiNbO₄ (panels A and C) and U-BiNbO₄ (panels B and D); EDS spectra taken on Cu substrate indicated by green lines.

SI4. CHNS (Carbon, Hydrogen, Nitrogen and Sulphur) Elemental Analysis

CHNS elemental analysis was carried out using FLASH EA 1112 Series, CHNS-O Analyzer. Furnace temp is at 950 0C, testing time is 720 sec per sample. The weight percentage of C for BiNbO₄ and U-BiNbO₄ is 0.0385 % and 0.1077 %, respectively.

Table S1. Percentage of C and H.

Sample	C (wt%)	H(wt%)
BiNbO ₄	0.0385	0.0763
U-BiNbO ₄	0.1077	0.0865

SI5. Overall XPS survey spectra of U-BiNbO₄

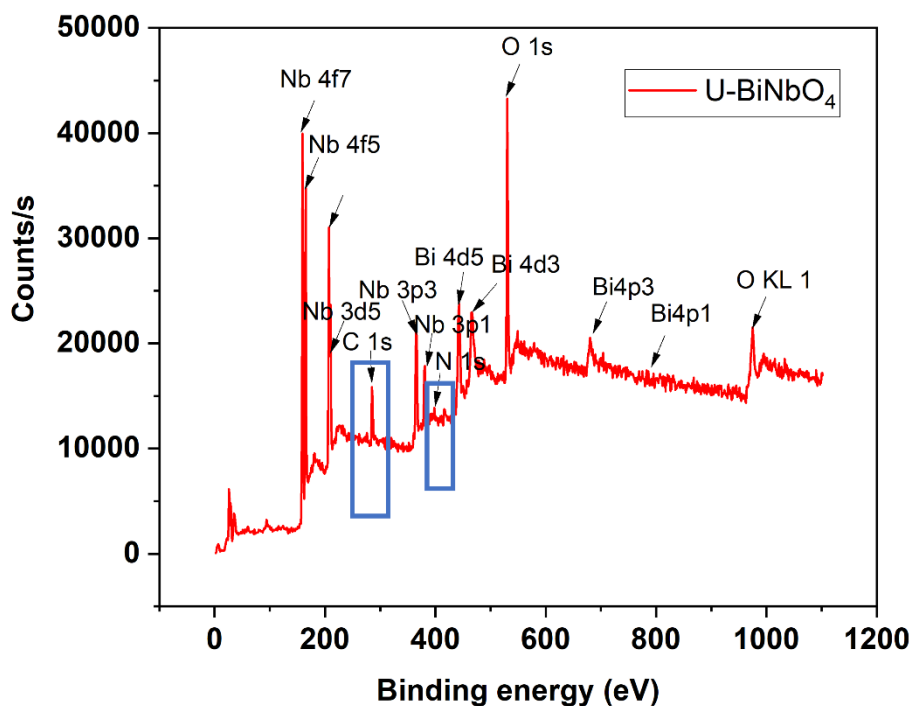


Figure S4: XPS survey spectra of U-BiNbO₄

SI6. Characteristic absorption bands of cationic A) MB and B) BG dyes for U-BiNbO₄ catalyst

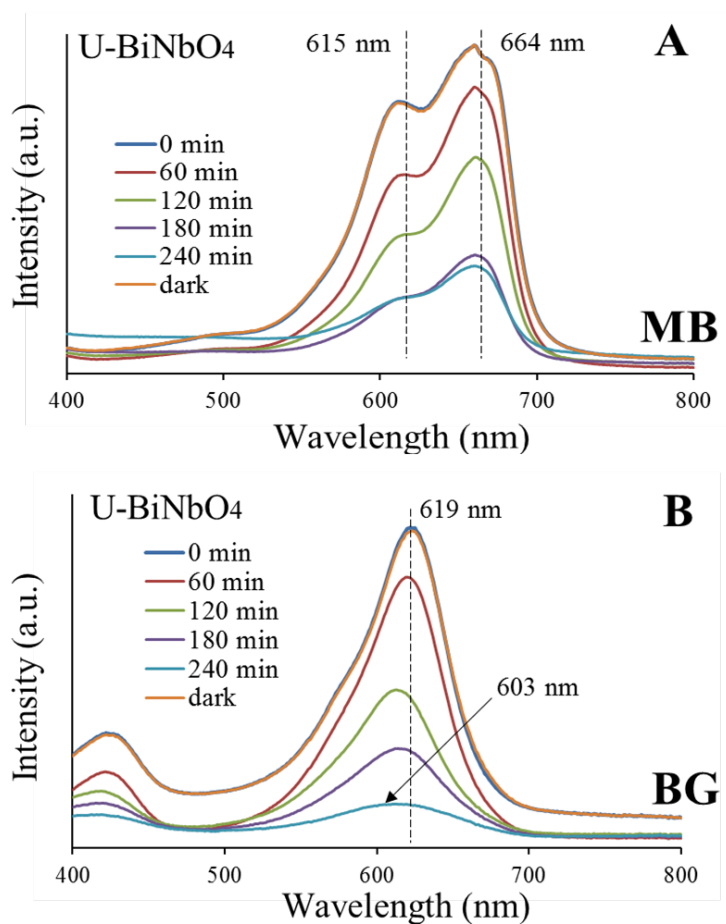


Figure S5: Characteristic absorption bands of cationic A) MB and B) BG dyes for U-BiNbO₄ catalyst.

SI7. Normalized photocatalytic degradation using 10, 15, 30, and 45 mg of U-BiNbO₄ for A) MB and B) BG dyes.

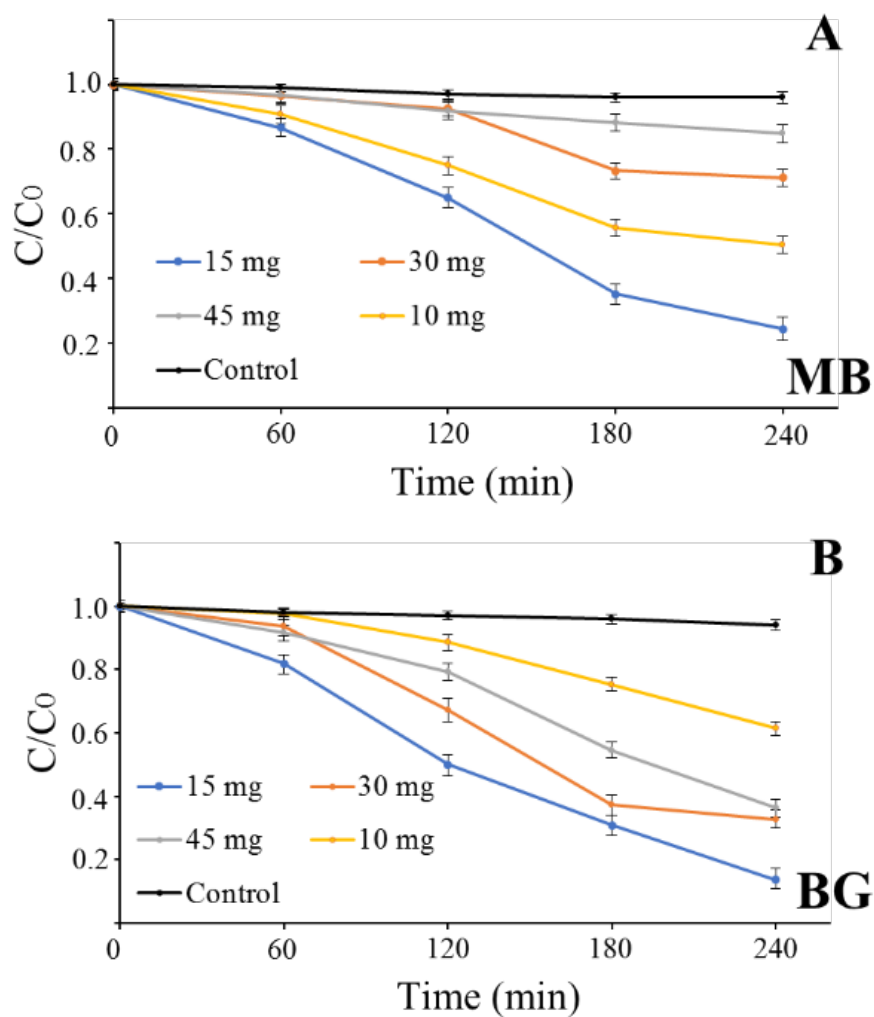


Figure S6: Photocatalytic degradation using 10, 15, 30, and 45 mg of U-BiNbO₄ for A) MB and B) BG dyes.

S18. Normalized photocatalytic degradation of MB in different pH using U-BiNbO₄

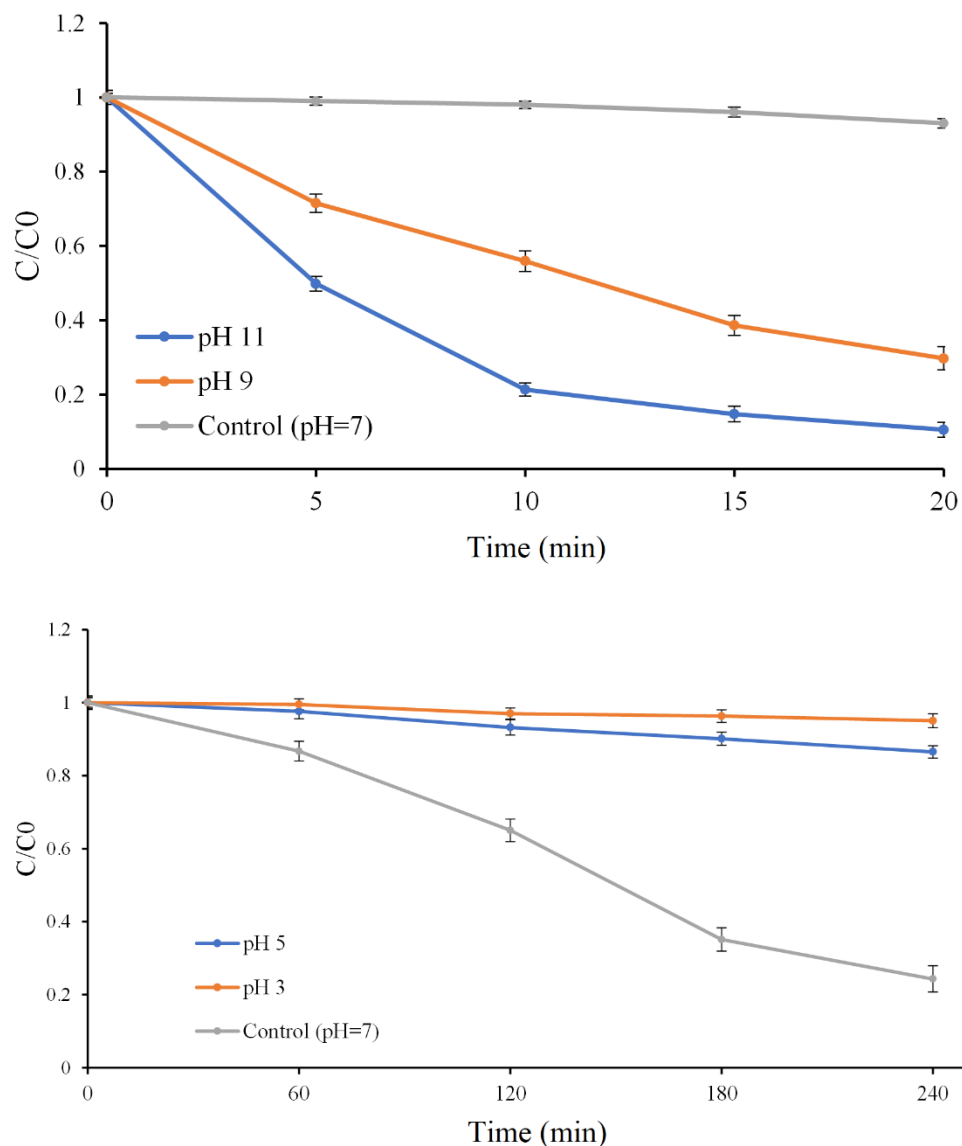


Figure S7: Photocatalytic degradation using pH 11.5,9.5,5.5 and 3.5 solutions of U-BiNbO₄ for MB.

References

1. M. Aslam, M. T. Soomro, I. M. I. Ismail, H. A. Qari, M. A. Gondal and A. Hameed, *RSC Advances*, 2015, **5**, 102663-102673.
2. K. Senevirathne, R. Hui, S. Campbell, S. Ye and J. Zhang, *Electrochimica Acta*, 2012, **59**, 538-547.