## Supplementary Information

# Sea-Urchin-like Iron-Cobalt Phosphide as Advanced Anode Material for Lithium Ion Battery 

Prakash Kumar Pathak, ${ }^{\dagger}{ }^{\dagger \mathrm{a}}$ Ved Prakash Joshi, ${ }^{\dagger, \mathrm{a}}$ Nitish Kumar, ${ }^{\text {a }}$ and Rahul R Salunkhe ${ }^{\mathrm{a}, *}$<br>${ }^{a}$ Materials Research Laboratory, Department of Physics, Indian Institute of Technology Jammu, Jammu and Kashmir, India (181121).<br>${ }^{\dagger}$ These authors contributed equally to this work.<br>*Email: rahul.salunkhe@iitjammu.ac.in



Fig. S1. FESEM images of the before and after phosphorization. (a-c) As-synthesized $\mathrm{FeCo}\left(\mathrm{Co}_{3}\right)_{2} \mathrm{OH}$ after hydrothermal reaction and (d-f) FeCoP after phosphorization at $300^{\circ} \mathrm{C}$ under the flow of $\mathrm{N}_{2}$.


Fig. S2. EDS mapping before and after phosphorization. (a) the signal from $\mathrm{FeCo}\left(\mathrm{Co}_{3}\right)_{2} \mathrm{OH}$, (b) the signal from FeCoP .


Fig. S3. EIS measurement of the device after $150^{\text {th }}$ cycle showing the deceased charge transfer resistance $\left(R_{c t}\right)$.

Table S1. FeCoP half-cell comparison with previously reported transition metal phosphides.

| S.No | Material | Voltage window (V) | Electrolyte | Discharge Capacity (mAh g ${ }^{-1}$ )@current rate | Ref. |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 1. | CoP/graphene | 0-3 | $1 \mathrm{M} \mathrm{LiPF}_{6}$ in a mixture of EC: EMC: DMC ${ }^{\text {F }}$ in the ratio of 1:1:1 | 1154@100 mA g ${ }^{-1}$ | 1 |
| 2. | CoP@ $\mathrm{GA}^{\text {* }}$ | 0-3 | $1 \mathrm{M} \mathrm{LiPF}_{6}$ in a mixture EC: $\mathrm{DEC}(1: 1 \mathrm{v} / \mathrm{v})$ | 1032.2@ $100 \mathrm{~mA} \mathrm{~g}^{-1}$ | 2 |
| 3. | CoP@ $\mathrm{GF}^{\text { }}$ | 0-3 | $1 \mathrm{M} \mathrm{LiPF}_{6}$ in a mixture of DMC: DEC: EC (1:1:1 vol\%) | $1120 @ 100 \mathrm{~mA} \mathrm{~g}{ }^{-1}$ | 3 |
| 4. | $\mathrm{CoP} / \mathrm{NC}{ }^{\text {+ }}$ | 0-3 | $1 \mathrm{M} \mathrm{LiPF}_{6}$ in a mixture of EC/DMC (1: $1, \mathrm{v} / \mathrm{v}$ ) | ~800@ 50 mA g -1 | 4 |
| 5. | $\mathrm{Co}_{2} \mathrm{P}$ | 0-3 | $1 \mathrm{M} \mathrm{LiPF}_{6}$ in a mixture of EC/DMC (1: $1, \mathrm{v} / \mathrm{v}$ ) | 780 @ 0.2 C | 5 |
| 6. | $\mathrm{CuP}_{2}$ | 0.02-2.5 | $\mathrm{LiPF}_{6}$-based electrolyte | 865 @ $100 \mathrm{~mA} \mathrm{~g}{ }^{-1}$ | 6 |
| 7. | CoP/RGO ${ }^{\text {* }}$ | 0.005-3 | $1 \mathrm{M} \mathrm{LiPF}_{6}$ in a mixture of EC/DEC (1: $1, \mathrm{v} / \mathrm{v}$ ) | 1,274@100 mA g ${ }^{-1}$ | 7 |
| 8. | CoP@ ${ }^{\text { }}$ | 1.8-2.8 | $0.2 \mathrm{M} \mathrm{Li}_{2} \mathrm{~S}_{6}+1 \mathrm{M}$ LiTFSI in 1,3-dioxolane and dimethoxyethane (1:1 in volume) | 1020 @ 0.2 C | 8 |
| 9. | $\mathrm{Co}_{\mathrm{x}} \mathrm{P}-\mathrm{NC}-800$ | 0-3 | $1 \mathrm{M} \mathrm{LiPF}_{6}$ in a mixture of EC/DEC (1: $1, \mathrm{v} / \mathrm{v}$ ) | 1224@100 mA g ${ }^{-1}$ | 9 |
| 10 | CoP@3DC | 1.7-2.8 | 1 MLLTFSI in a DOL/DOM (v/v $=1: 1)$ | 1161.79@0.2 C | 10 |
| 11 | $\mathrm{FeCoP}{ }^{\text { }}$ | 0-3 | $1 \mathrm{M} \mathrm{LiPF}_{6}$ in a mixture of EC/DMC (1: $1, \mathrm{v} / \mathrm{v}$ ) | $1653.4 @ 100 \mathrm{~mA} \mathrm{~g}$-1 | This work |

${ }^{7}$ Note: EC:EMC:DMC- Ethylene carbonate:ethylene methyl carbonate: dimethyl carbonate; RGO-Reduced graphene oxide; GA- graphene aerogel; GF-graphene framework membrane; $\boldsymbol{N C}$ - nitrogen-doped carbon; $\boldsymbol{S}$-Sulfur; 3DC-three-dimensional carbon frame embedded; CoP-Cobalt phosphide.


Table S2. Data summary for this work.

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