## **Supplementary Information**

## Sea-Urchin-like Iron-Cobalt Phosphide as Advanced Anode Material for Lithium Ion Battery

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**Fig. S1.** FESEM images of the before and after phosphorization. (a-c) As-synthesized  $FeCo(Co_3)_2OH$  after hydrothermal reaction and (d-f) FeCoP after phosphorization at 300 °C under the flow of N<sub>2</sub>.



**Fig. S2.** EDS mapping before and after phosphorization. (a) the signal from  $FeCo(Co_3)_2OH$ , (b) the signal from FeCoP.



Fig. S3. EIS measurement of the device after  $150^{\text{th}}$  cycle showing the deceased charge transfer resistance ( $R_{ct}$ ).

S.No	Material	Voltage window (V)	Electrolyte	Discharge Capacity (mAh g <sup>-1</sup> ) @ current rate	Ref.
1.	CoP/graphene	0-3	1 M LiPF <sub>6</sub> in a mixture of EC: EMC: DMC <sup>‡</sup> in the ratio of 1: 1: 1	1154 @ 100 mA g <sup>-1</sup>	1
2.	CoP@GA <sup>‡</sup>	0-3	1 M LiPF <sub>6</sub> in a mixture EC: DEC (1: 1 $v/v$ )	1032.2 @ 100 mA g <sup>-1</sup>	2
3.	CoP@GF <sup>‡</sup>	0-3	1 M LiPF <sub>6</sub> in a mixture of DMC: DEC: EC $(1:1:1 \text{ vol}\%)$	1120 @ 100 mA g <sup>-1</sup>	3
4.	CoP/NC <sup>‡</sup>	0-3	1 M LiPF <sub>6</sub> in a mixture of EC/DMC (1: 1, $v/v$ )	$\sim 800 @ 50 \text{ mA g}^{-1}$	4
5.	Co <sub>2</sub> P	0-3	1 M LiPF <sub>6</sub> in a mixture of EC/DMC (1: 1, $v/v$ )	780 @ 0.2 C	5
6.	CuP <sub>2</sub>	0.02–2.5	LiPF <sub>6</sub> -based electrolyte	865 @ 100 mA g <sup>-1</sup>	6
7.	CoP/RGO <sup>‡</sup>	0.005-3	1 M LiPF <sub>6</sub> in a mixture of EC/DEC (1: 1, $v/v$ )	1,274 @ 100 mA g <sup>-1</sup>	7
8.	CoP@S <sup>‡</sup>	1.8-2.8	0.2 M $Li_2S_6$ +1M LiTFSI in 1,3-dioxolane and dimethoxyethane (1:1 in volume)	1020 @ 0.2 C	8
9.	Co <sub>x</sub> P-NC-800	0-3	1 M LiPF <sub>6</sub> in a mixture of EC/DEC (1: 1, $v/v$ )	1224 @100 mA $g^{-1}$	9
10	CoP@3DC <sup>‡</sup>	1.7–2.8	1 M LiTFSI in a DOL/DOM ( $v/v = 1:1$ )	1161.79 @ 0.2 C	10
11	FeCoP <sup>‡</sup>	0-3	1 M LiPF <sub>6</sub> in a mixture of EC/DMC (1: 1, $v/v$ )	1653.4 @ 100 mA g <sup>-1</sup>	This work

Table S1. FeCoP half-cell comparison with previously reported transition metal phosphides.

<sup>‡</sup>Note: *EC:EMC:DMC- Ethylene carbonate:ethylene methyl carbonate: dimethyl carbonate; RGO- Reduced graphene oxide; GA- graphene aerogel; GF- graphene framework membrane;NC- nitrogen-doped carbon;S-Sulfur; 3DC- three-dimensional carbon frame embedded; CoP-Cobalt phosphide.* 

XPS Analy	ysis			
Sample	Element		Binding energy (eV)	Atomic percentage (%)
FeCoP				· · · ·
	Р		133.75	11.4
	С		284.8	25.46
	0		531.1	44.68
	Fe		710.91	9.16
	Со		781.1	9.3
BET surfa	ice area measureme	ents		
Material	S <sub>BET</sub>	Smicro	V <sub>pore</sub>	V <sub>micro</sub>
	$(m^2 g^{-1})$	$(m^2 g^{-1})$	$(cm^3 g^{-1})$	$(cm^3 g^{-1})$
FeCoP	29.3	0	0.034	0
Half cell p	oerformance			
			Current density	Specific discharge capacity
			$(mA g^{-1})$	$(mAh g^{-1})$
			100	1653.4
			400	1091.2
			500	1030.4
			600	884.2
			700	881.0
			1000	820.8

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 Table S2. Data summary for this work.

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