## **Supplementary Information (ESI)**

## Antioxidant mechanisms and products of four 4',5,7-

## trihydroxyflavonoids with different structural types

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Kaempferol-3-O<sup>-</sup>anion Naringenin-4'-O<sup>-</sup>anion Fig. S1 The electrostatic potential of the four flavonoid anions.

flavonoids					
Molecule	Site of the	Charge of the	Charge of the	Charge of the	bond moment
	hydroxyl	O atom	H atom	hydroxyl	of the O-H
AP	$C_5$	-0.414	0.280	-0.134	0.83087
	$C_7$	-0.408	0.317	-0.091	0.66642
	$C_{4'}$	-0.413	0.315	-0.098	0.66393
GE	$C_5$	-0.410	0.282	-0.128	0.82867
	$C_7$	-0.407	0.318	-0.089	0.66790
	$C_{4'}$	-0.429	0.310	-0.119	0.65531
KA	$C_3$	-0.413	0.293	-0.120	0.72244
	$C_5$	-0.408	0.284	-0.124	0.79888
	$C_7$	-0.407	0.318	-0.089	0.66698
	$C_{4'}$	-0.417	0.313	-0.104	0.66174
NA	$C_5$	-0.405	0.282	-0.123	0.81699
	$C_7$	-0.403	0.318	-0.085	0.66712
	$C_{4'}$	-0.429	0.310	-0.119	0.65572

Table S1 The atomic charges and bond dipole moments (Debye) of hydroxyls of the four



Fig. S2 The intrinsic reaction coordinates of apigenin to scavenging 'OH (R: reactants; Int: intermediates; TS: transition state; P: products).



Fig. S3 The intrinsic reaction coordinates of genistein to scavenging 'OH (R: reactants; Int: intermediates; TS: transition state; P: products).



Fig. S4 The intrinsic reaction coordinates of kaempferol to scavenging 'OH (R: reactants; Int: intermediates; TS: transition state; P: products).



Fig. S5 The intrinsic reaction coordinates of naringenin to scavenging 'OH (R: reactants; Int: intermediates; TS: transition state; P: products).



Fig. S6 The dose-response curves of the four flavonoids in DPPH-scavenging (A), ABTS<sup>+-</sup> scavenging (B), and  $O_2$ -scavenging (C) assays.