

## Electronic Supplementary Information (ESI)

### CF<sub>3</sub> H-bonding Locked Aromatic Stacking of Picric Acid with Mechanofluorochromic fluorophores: Highly Selective Reusable Sensor and Rewritable Fluorescence Platform

Parthasarathy Gayathri,<sup>a</sup> Sasikala Ravi,<sup>a</sup> Periyappan Nantheeswaran,<sup>b</sup> Mariappan Mariappan,<sup>b</sup> Subramanian Karthikeyan,<sup>c</sup> Mehboobali Pannipara,<sup>d,e</sup> Abdullah G. Al-Sehemi,<sup>d,e</sup> Dohyun Moon\*<sup>f</sup> Savarimuthu Philip Anthony\*<sup>a</sup>

<sup>a</sup>)School of Chemical & Biotechnology, SASTRA Deemed University, Thanjavur-613401, Tamil Nadu, India. Fax: +914362264120; Tel: +914362264101; E-mail: philip@biotech.sastra.edu

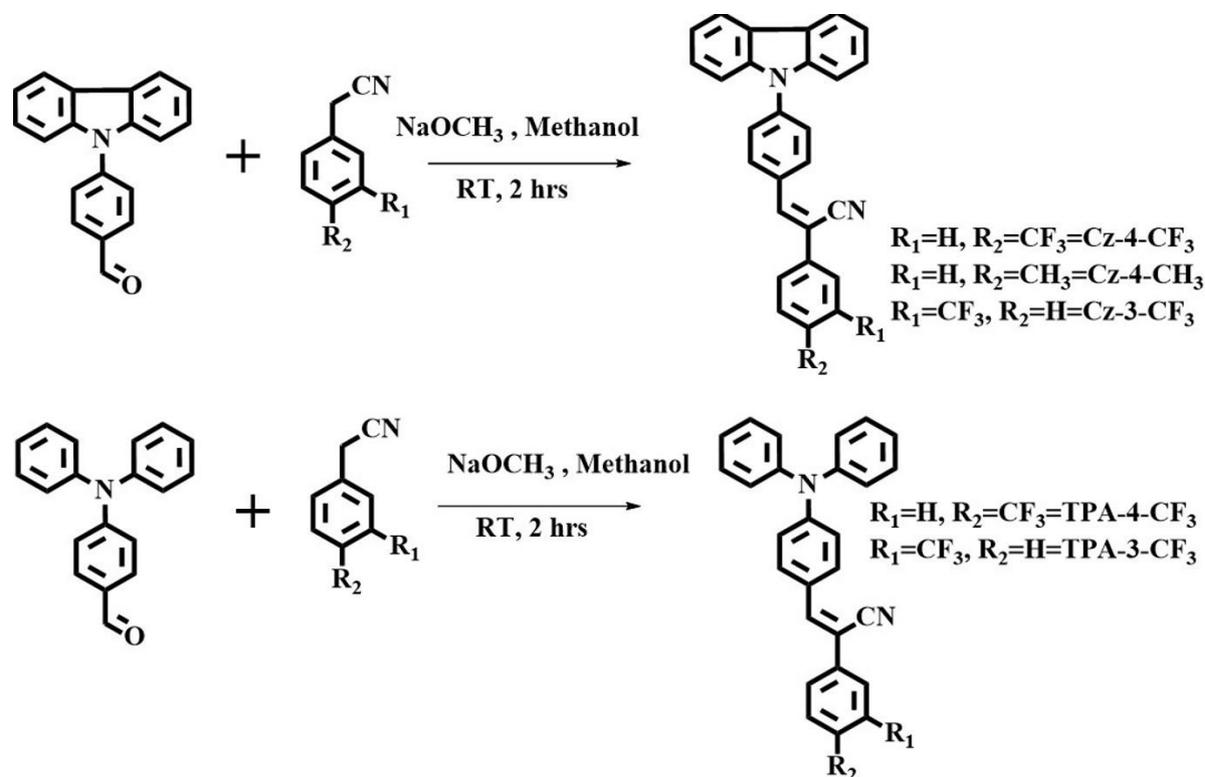
<sup>b</sup>)Department of Chemistry, SRM IST, Kattankulathur, Chennai-603203, Tamil Nadu, India.

<sup>c</sup>)PG and Research Department of Chemistry, KhadirMohideen College, Adirampattinam, Tamil Nadu, India.

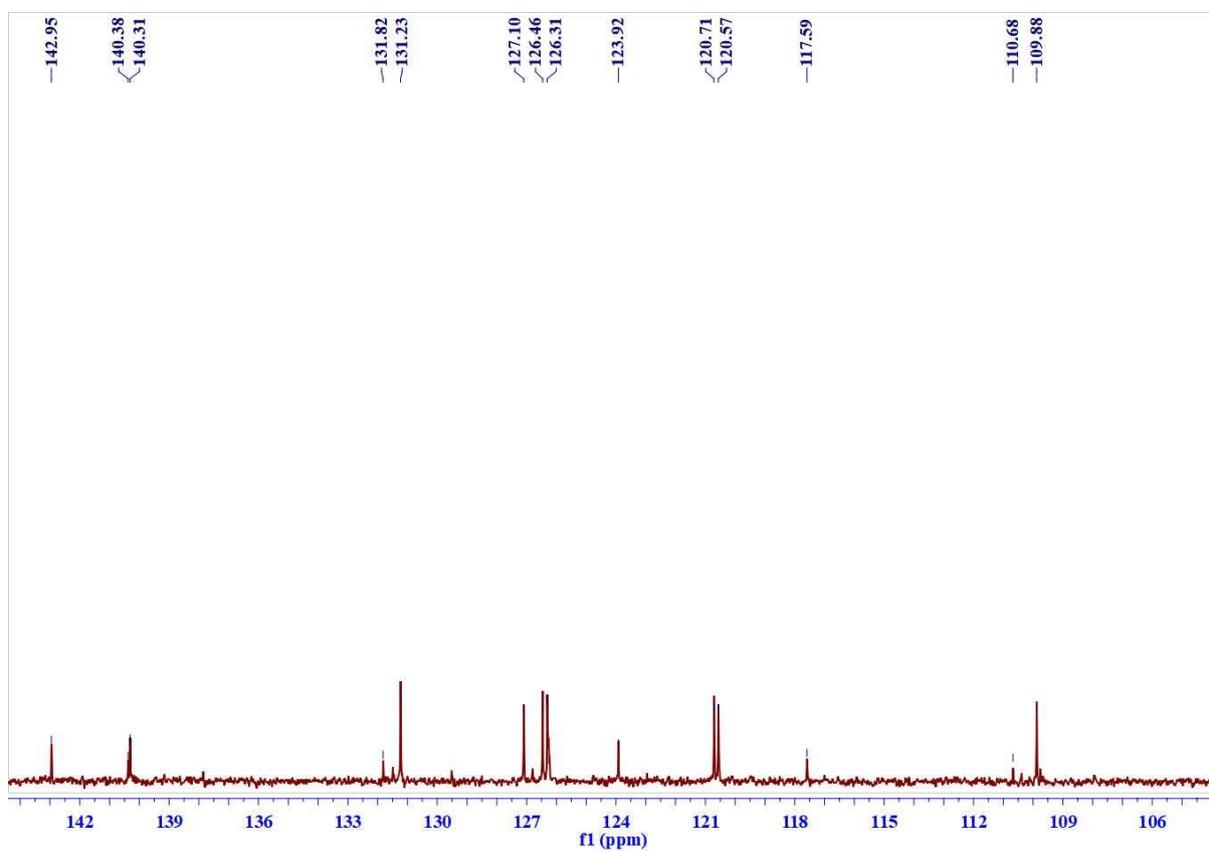
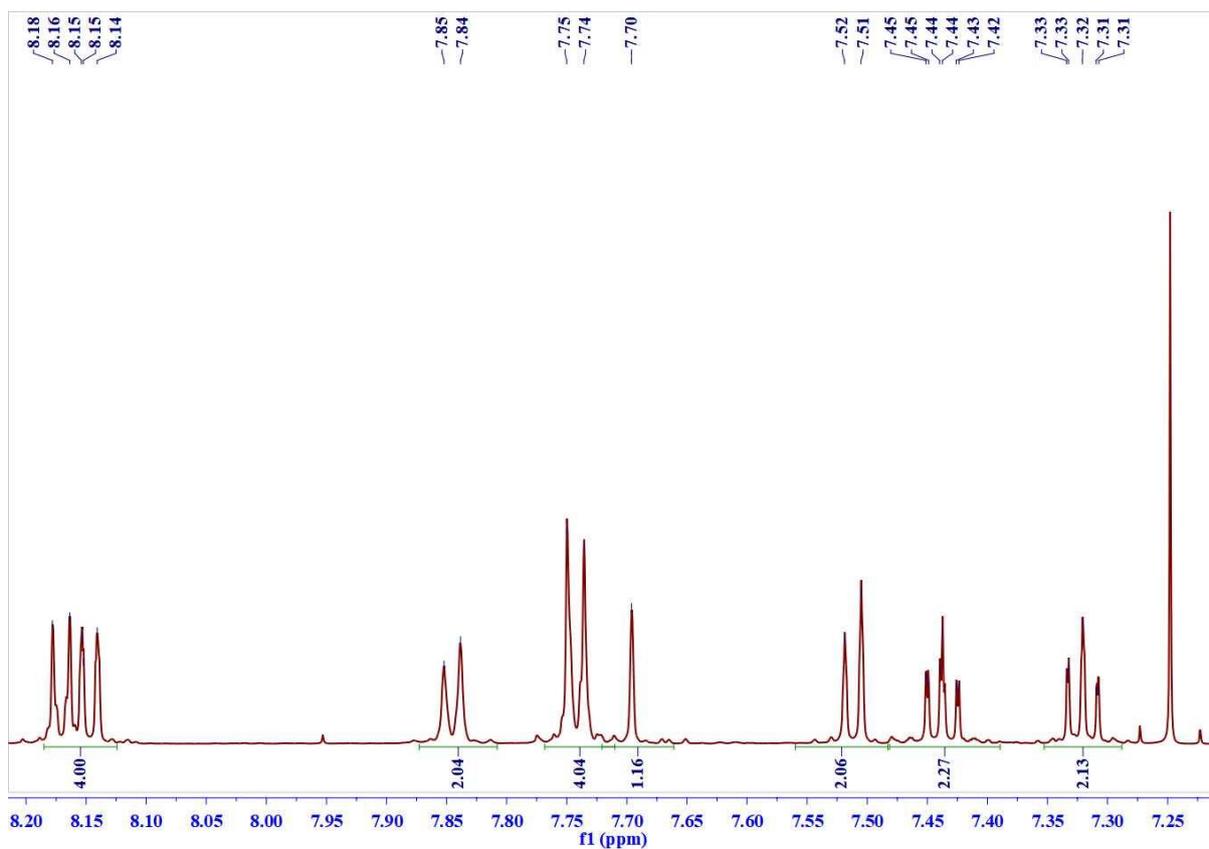
<sup>d</sup>)Research center for Advanced Materials Science, King Khalid University, Abha 61413, Saudi Arabia.

<sup>e</sup>)Department of chemistry, King Khalid University, Abha 61413, Saudi Arabia.

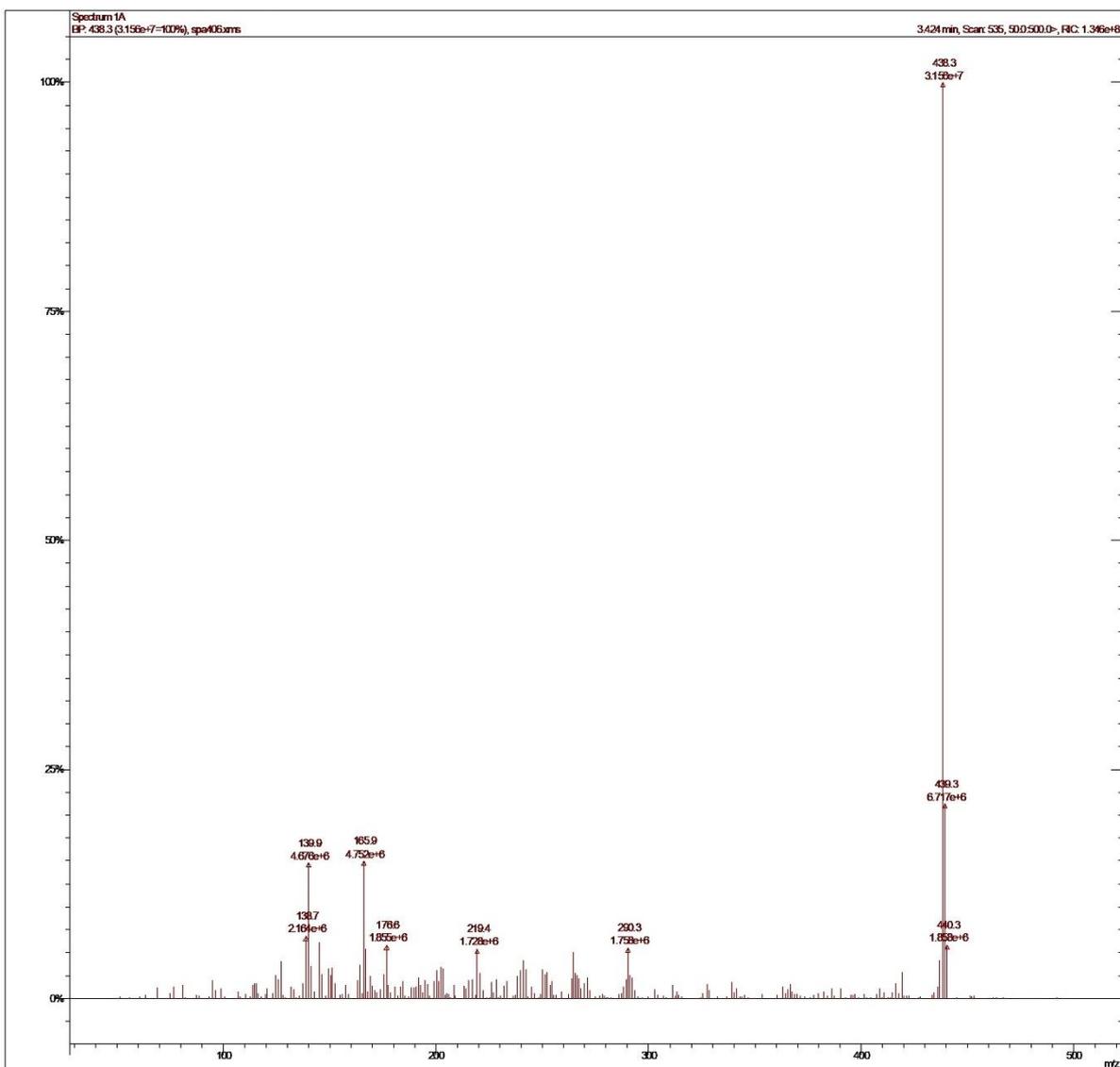
<sup>f</sup>)Beamline Department, Pohang Accelerator Laboratory, 80 Jigokro-127beongil, Nam-gu, Pohang, Gyeongbuk, Korea, Email: dmoon@postech.ac.kr



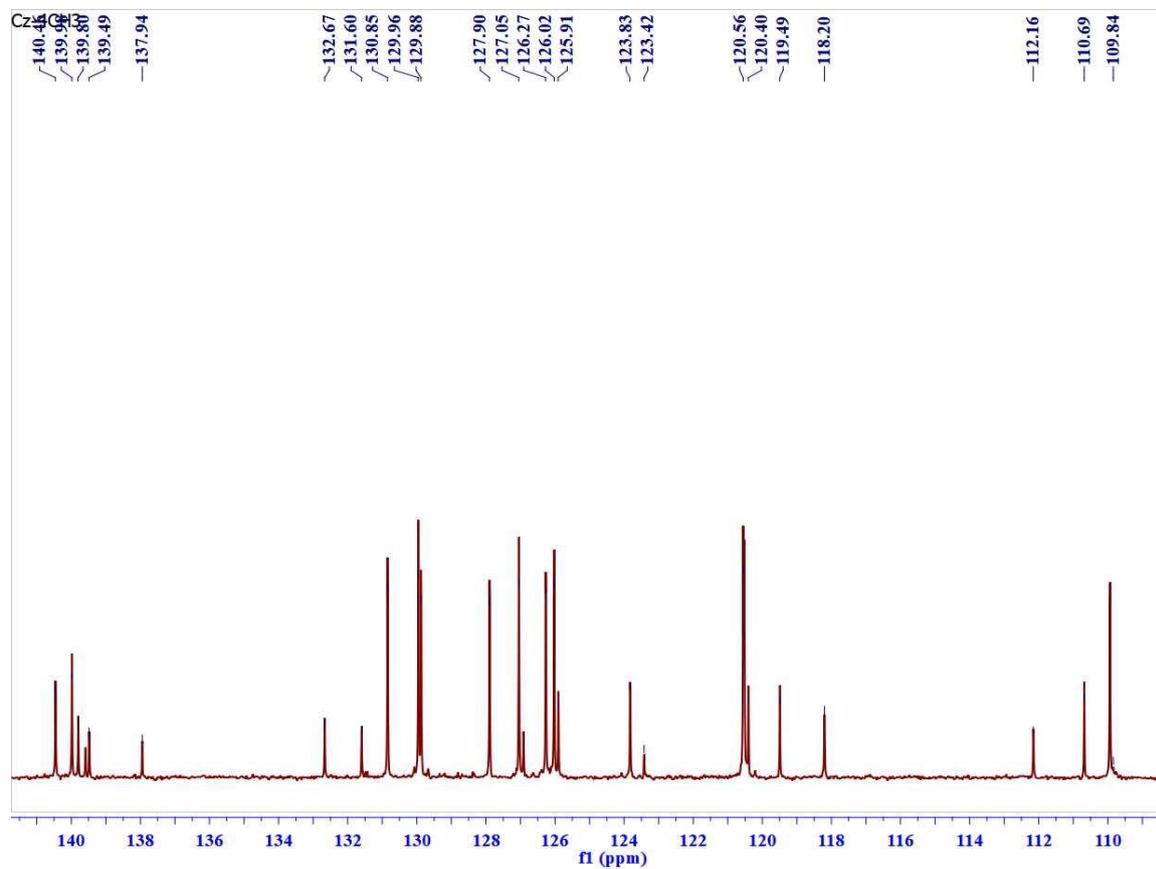
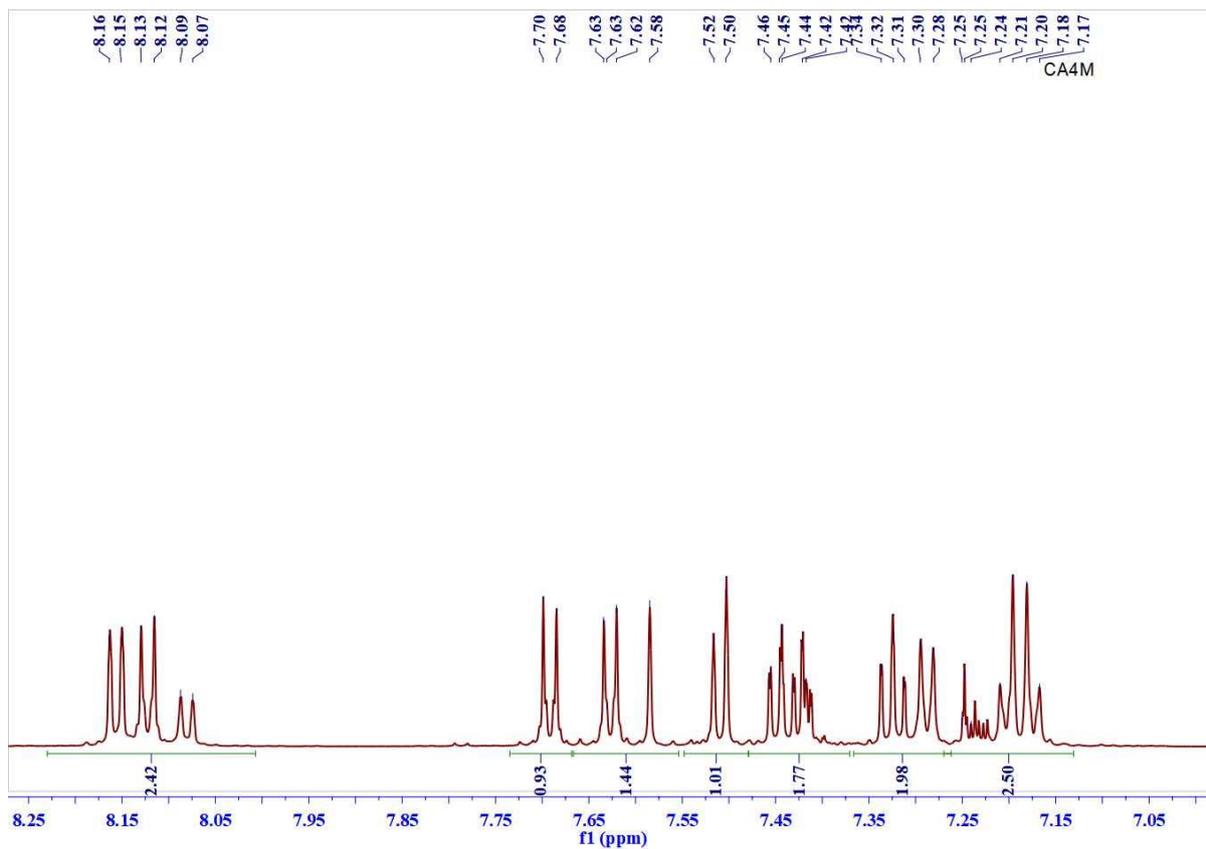
Scheme S1. Synthesis of Cz-4-CF<sub>3</sub>, Cz-4-CH<sub>3</sub>, Cz-3-CF<sub>3</sub>, TPA-4-CF<sub>3</sub> and TPA-3-CF<sub>3</sub>.



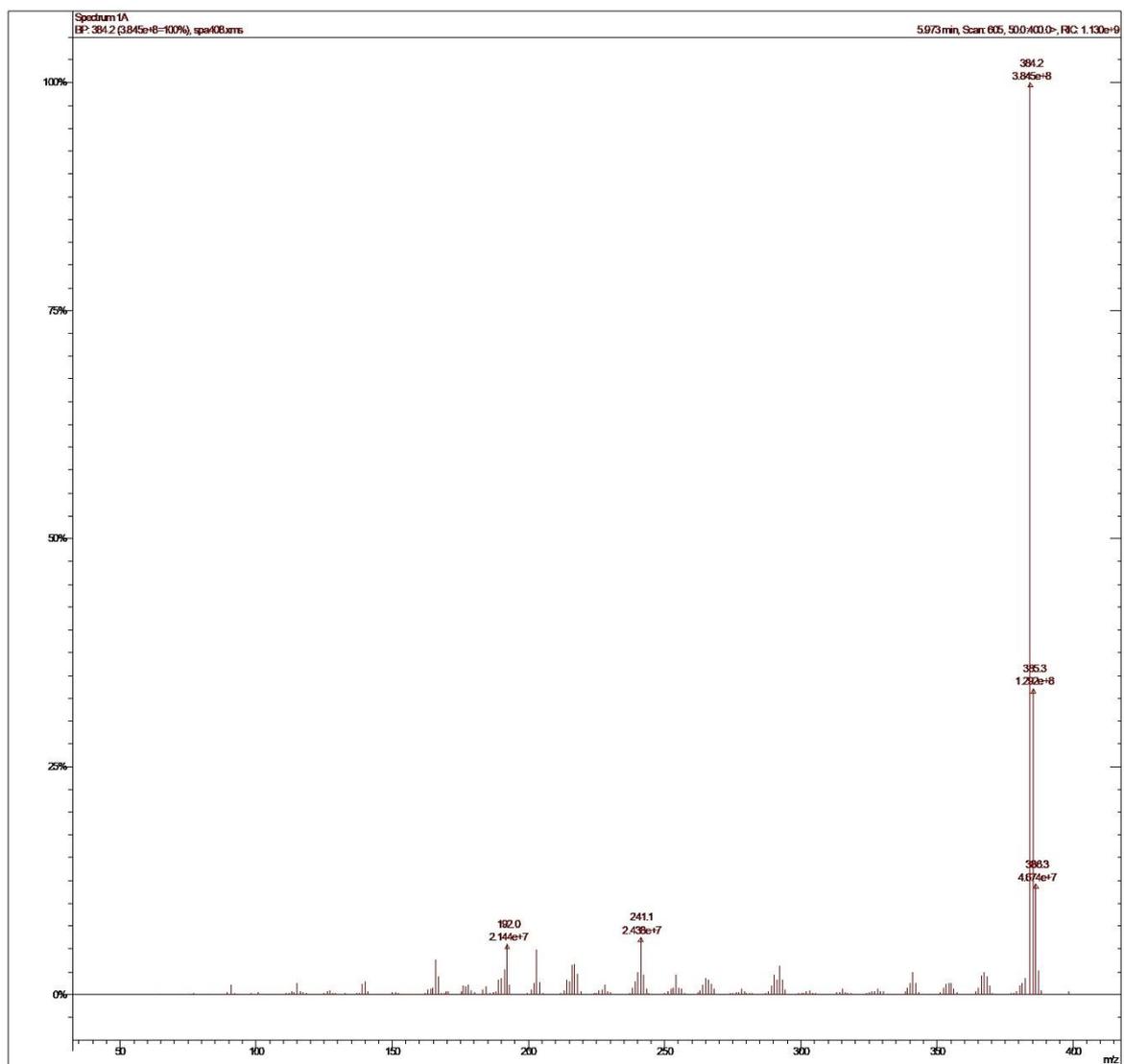
$^1\text{H}$  and  $^{13}\text{C}$  NMR of **Cz-4-CF<sub>3</sub>** (solvent =  $\text{CDCl}_3$ ).



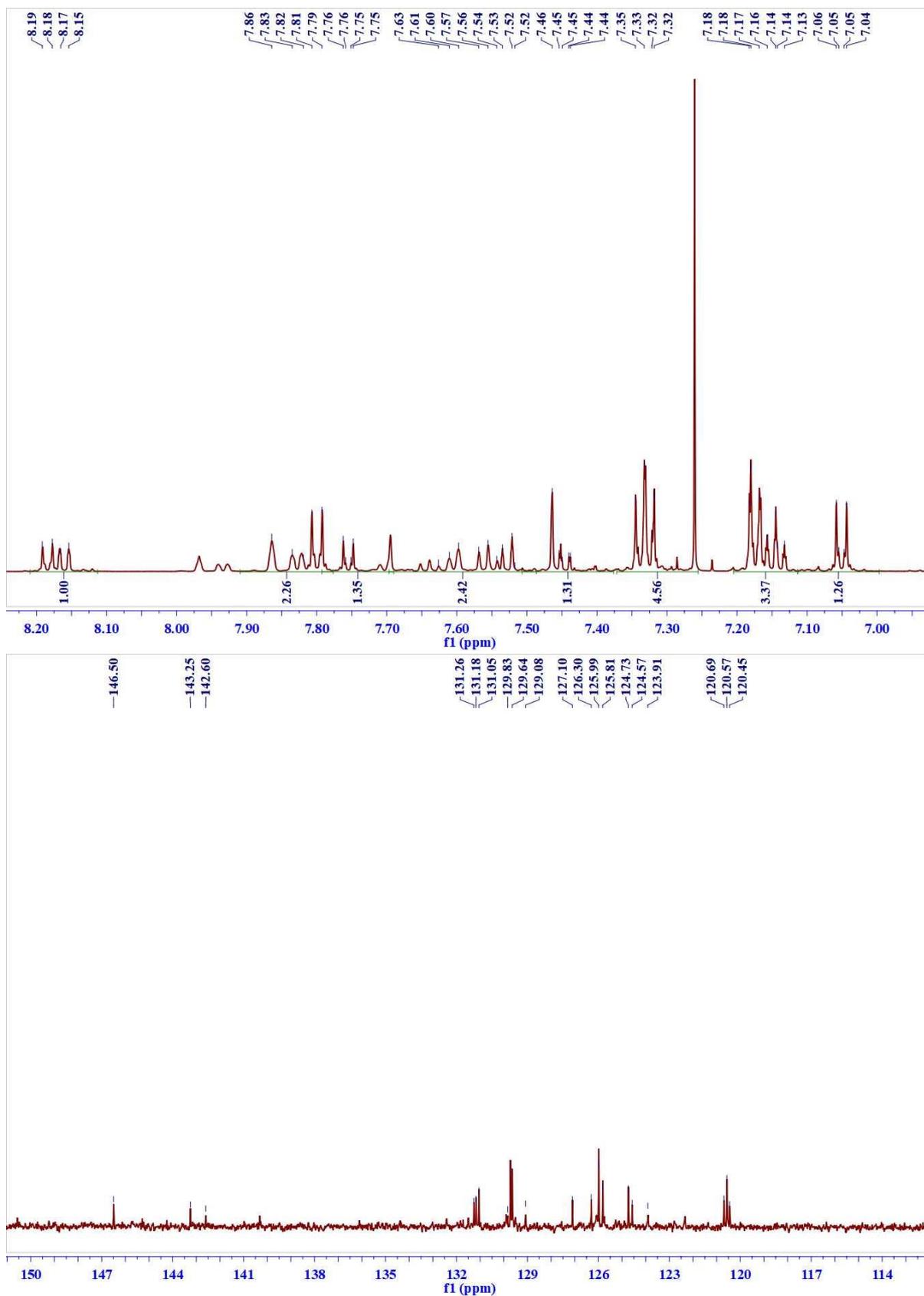
**Cz-4-CF<sub>3</sub>**: m/z calcd for C<sub>28</sub>H<sub>17</sub>F<sub>3</sub>N<sub>2</sub> (M + H): 438.13, found: 438.3.



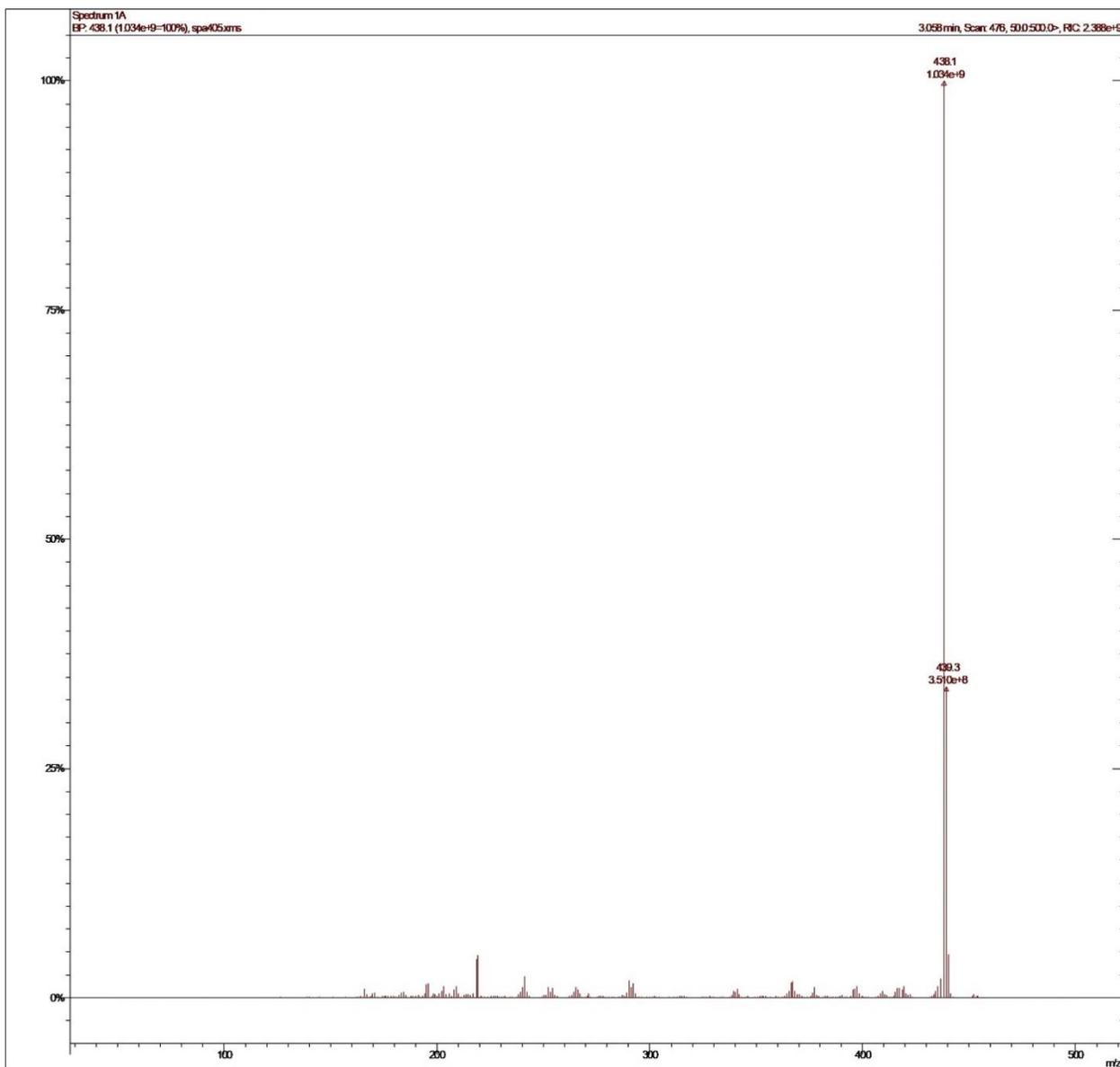
$^1\text{H}$  and  $^{13}\text{C}$  NMR of **Cz-4-CH<sub>3</sub>** (solvent =  $\text{CDCl}_3$ ).



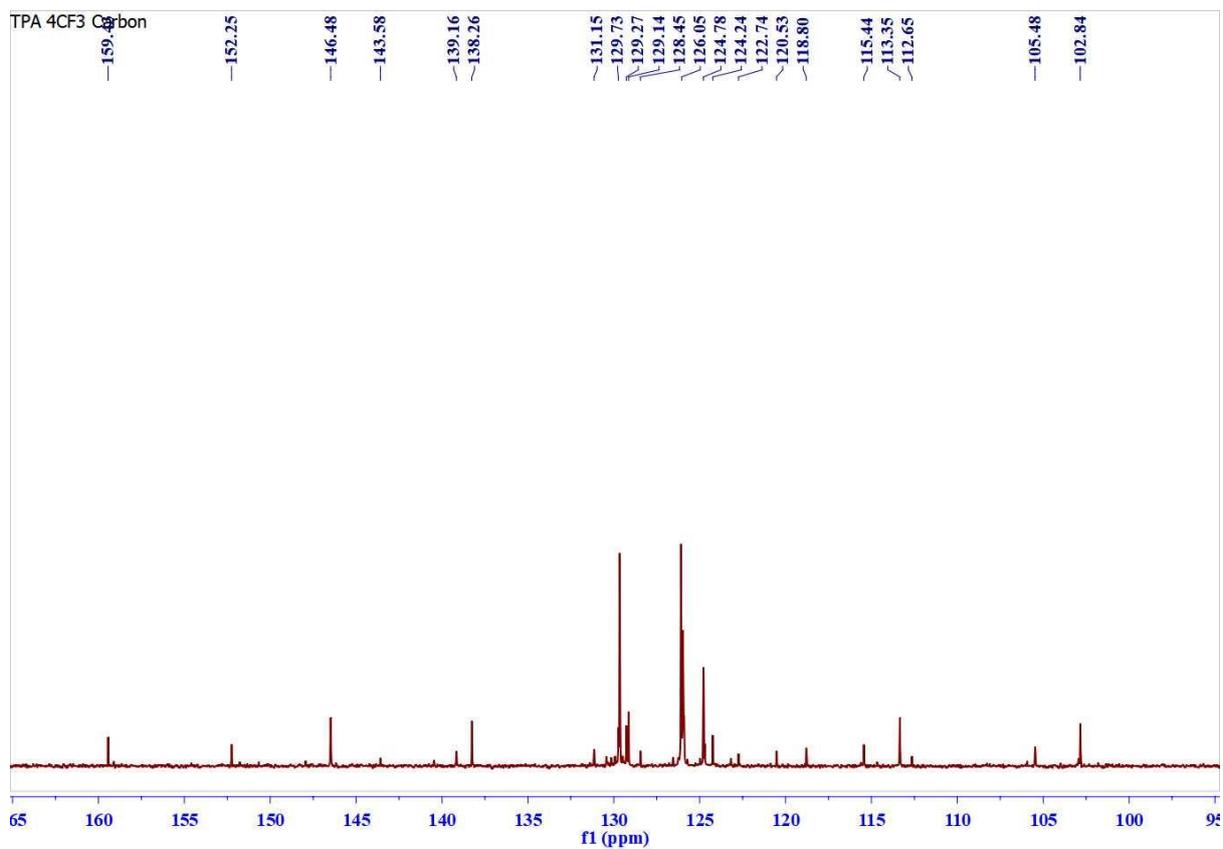
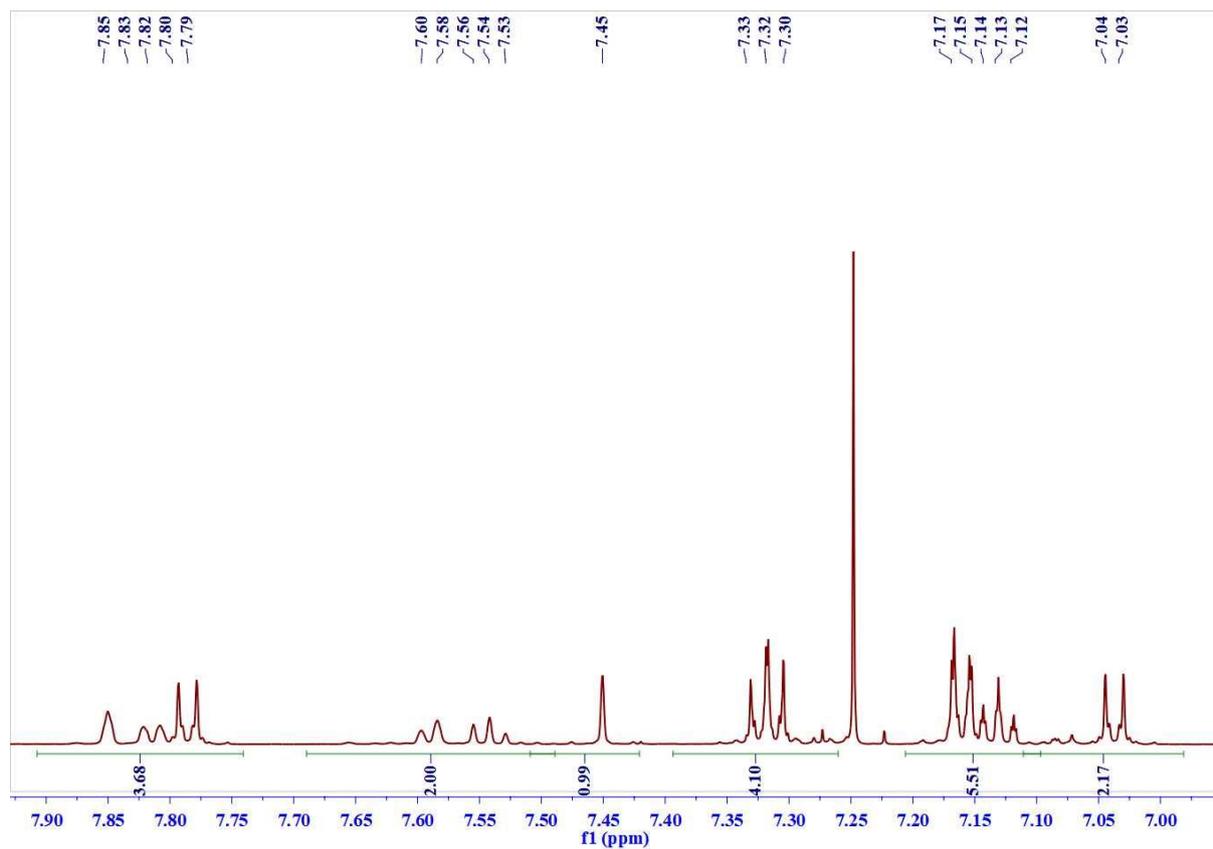
**Cz-4-CH<sub>3</sub>**: m/z calcd for C<sub>28</sub>H<sub>20</sub>N<sub>2</sub> (M + H): 384.16, found: 384.2.



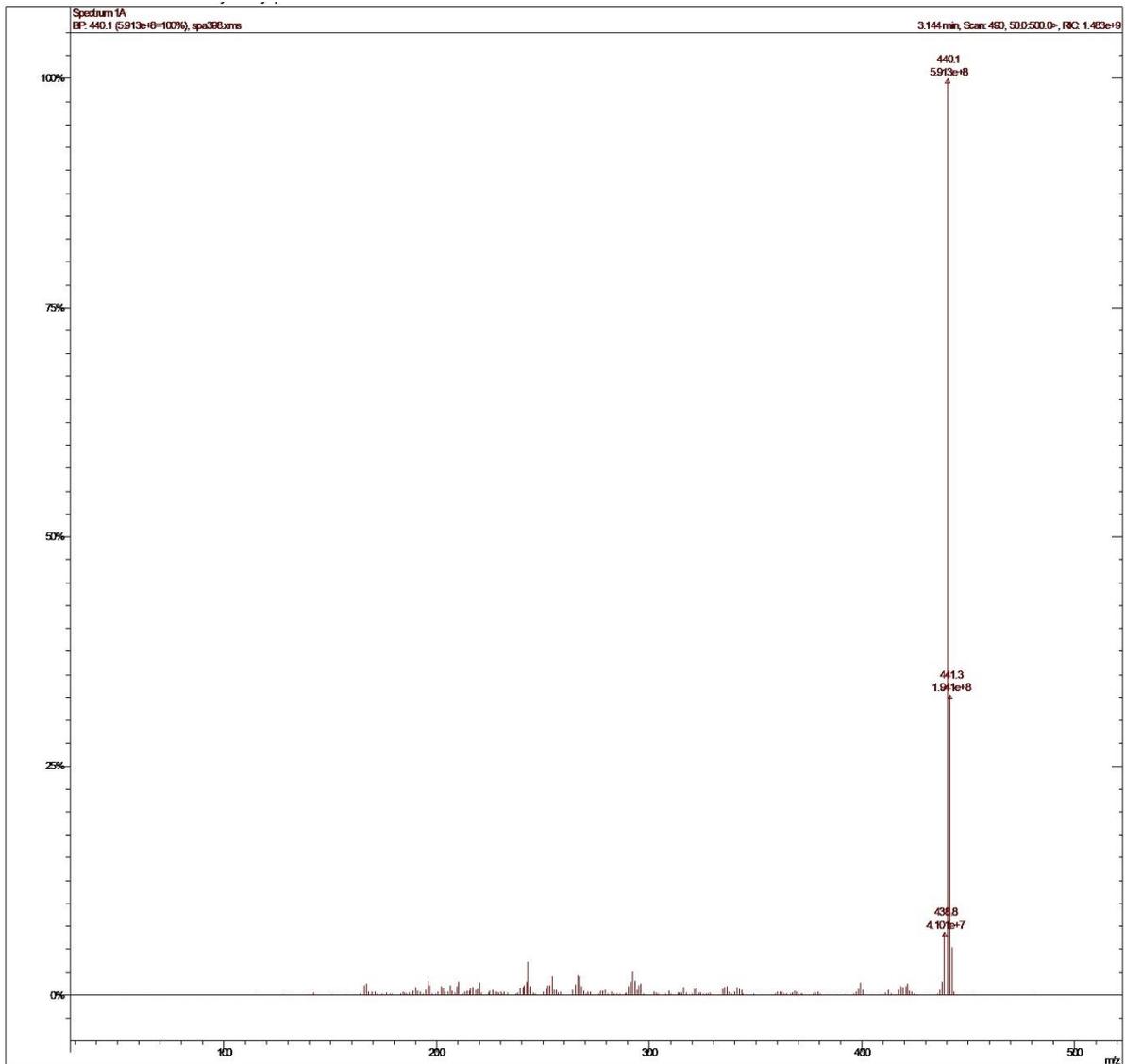
$^1\text{H}$  and  $^{13}\text{C}$  NMR of **Cz-3-CF<sub>3</sub>** (solvent = CDCl<sub>3</sub>).



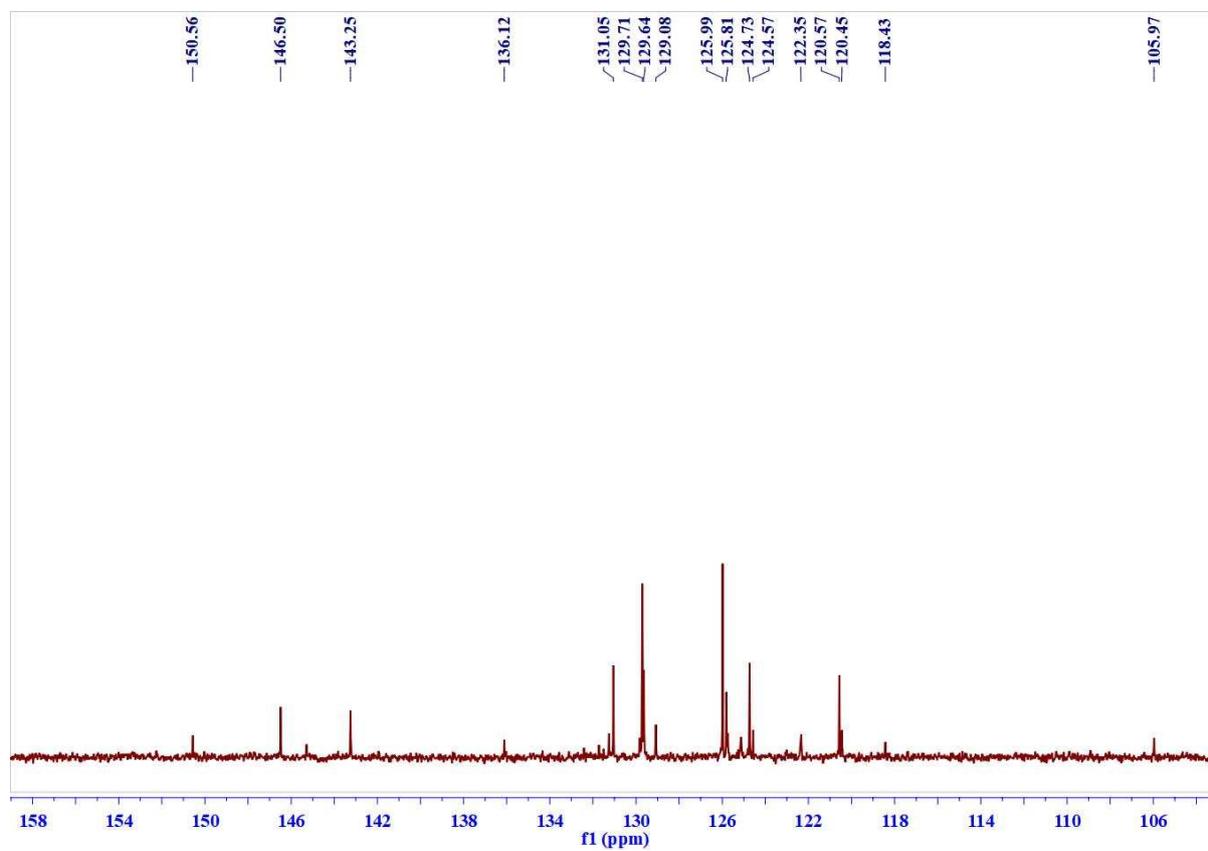
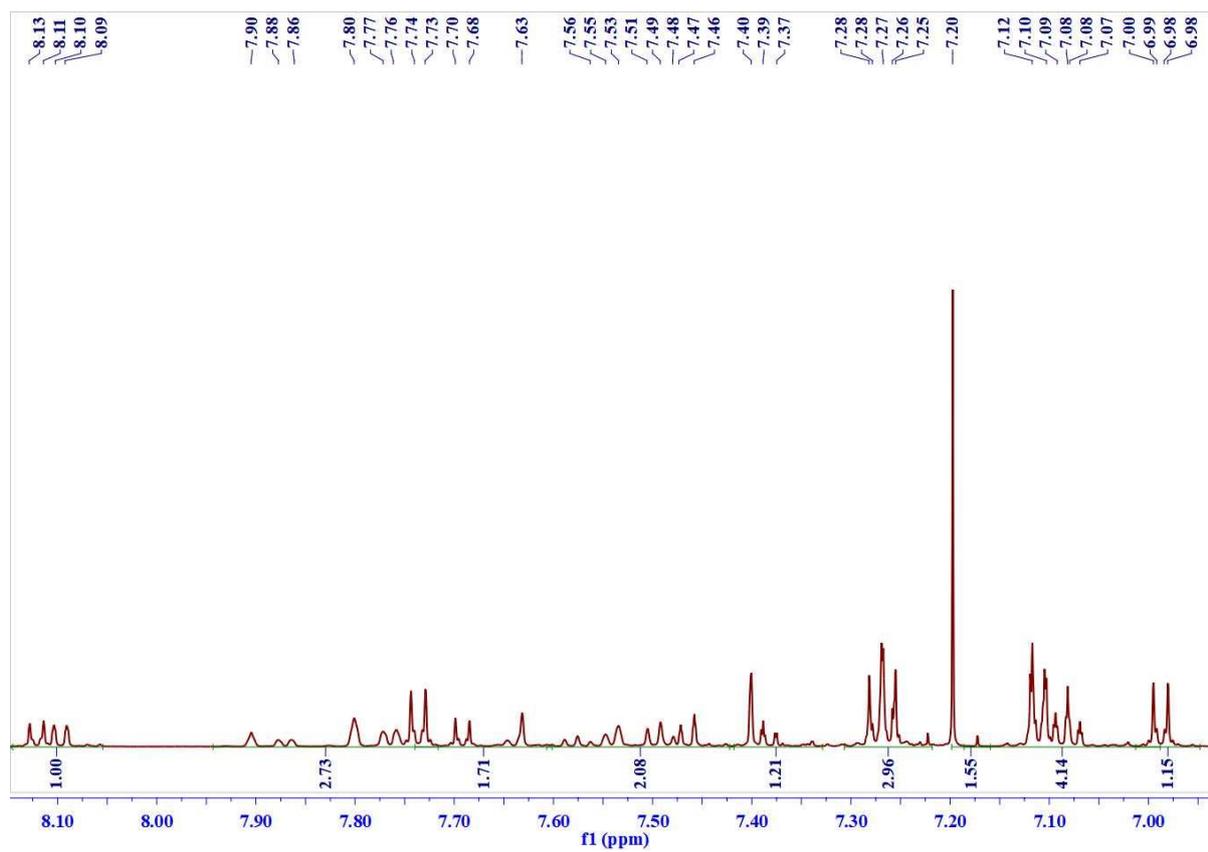
**Cz-3-CF<sub>3</sub>**: m/z calcd for C<sub>28</sub>H<sub>17</sub>F<sub>3</sub>N<sub>2</sub> (M + H): 438.13, found: 438.1.



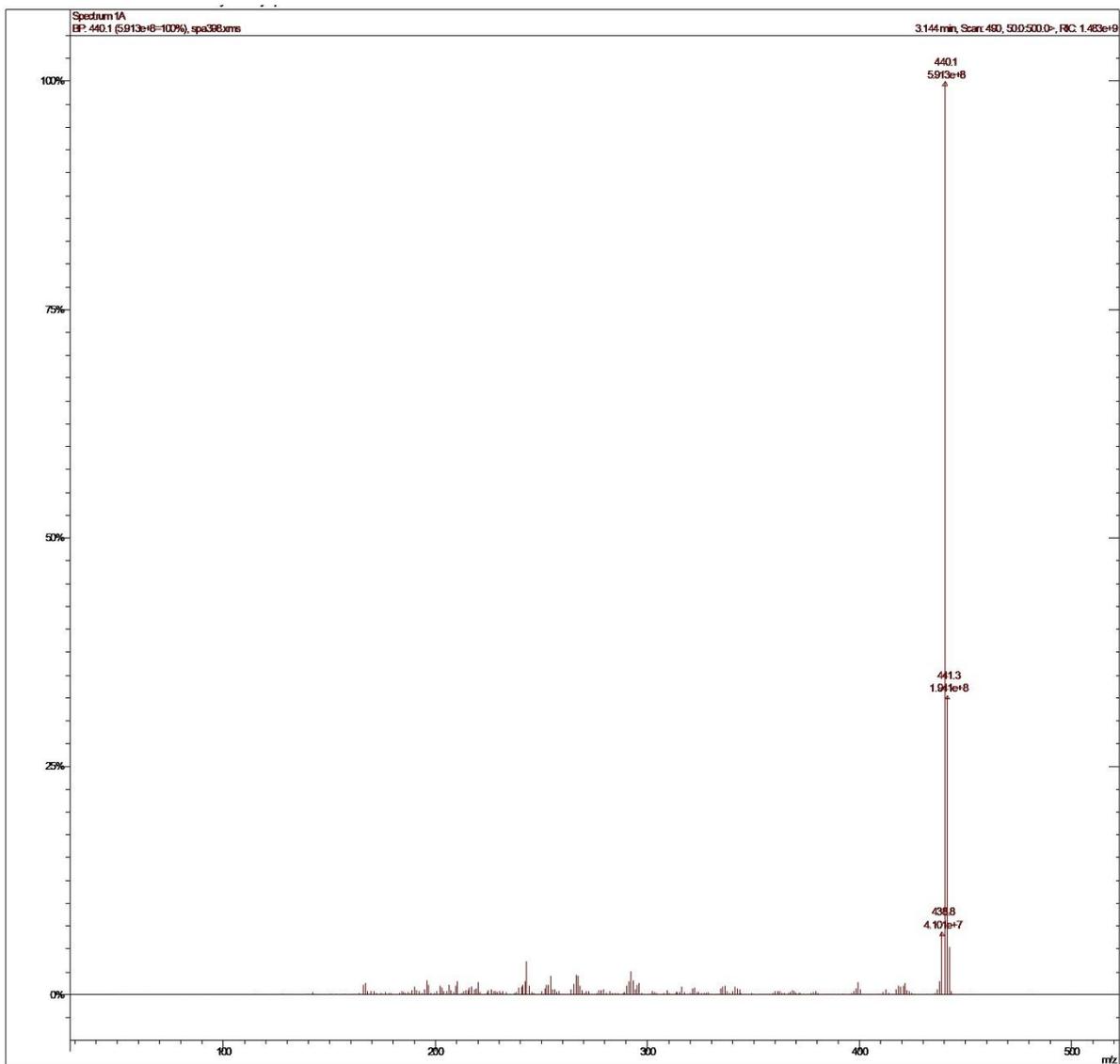
<sup>1</sup>H and <sup>13</sup>C NMR of TPA-4-CF<sub>3</sub> (solvent = CDCl<sub>3</sub>).



**TPA-4-CF<sub>3</sub>**: m/z calcd for C<sub>28</sub>H<sub>19</sub>F<sub>3</sub>N<sub>2</sub> (M + H): 440.15, found: 440.1.



<sup>1</sup>H and <sup>13</sup>C NMR of TPA-3-CF<sub>3</sub> (solvent = CDCl<sub>3</sub>).



**TPA-3-CF<sub>3</sub>**: m/z calcd for C<sub>28</sub>H<sub>19</sub>F<sub>3</sub>N<sub>2</sub> (M + H): 440.15, found: 440.1.

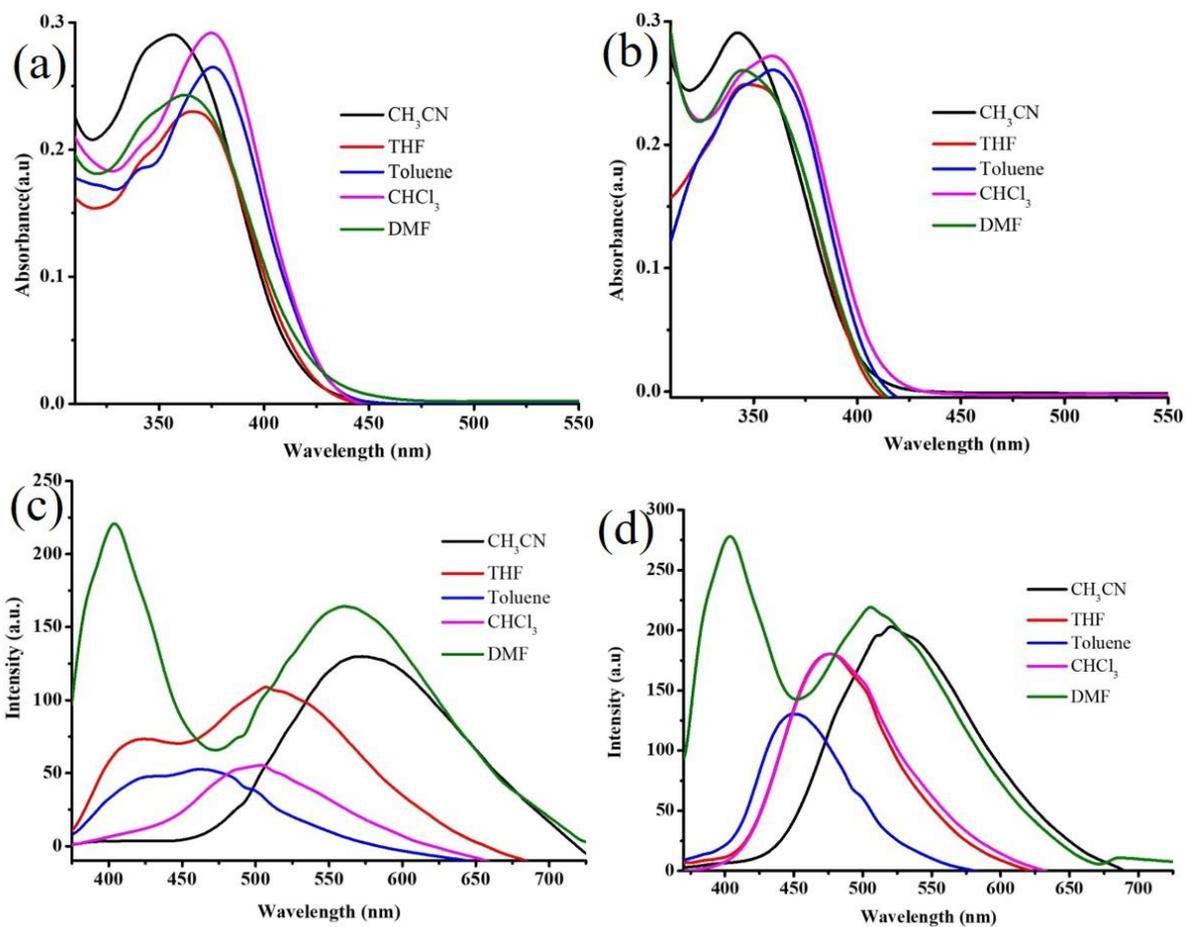


Fig. S1. Absorption (a,b) and fluorescence (c, d) spectra of (a, c) Cz-4-CF<sub>3</sub> and (b, d) Cz-4-CH<sub>3</sub> in different solvents.

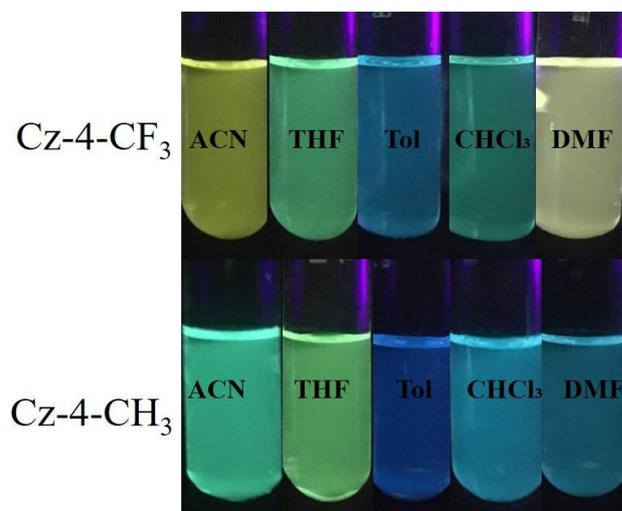


Fig. S2. Digital fluorescence images of Cz-4-CF<sub>3</sub> and Cz-4-CH<sub>3</sub> in different solvents.

Table S1. Quantum yield of Cz-4-CF<sub>3</sub> and Cz-4-CH<sub>3</sub> in solvents with respect to quinine sulphate.

Solvent	Quantum yield ( $\Phi_F$ )	
	Cz-4-CF <sub>3</sub>	Cz-4-CH <sub>3</sub>
CH <sub>3</sub> CN	0.023	0.040
CHCl <sub>3</sub>	0.010	0.045
DMF	0.040	0.064
THF	0.018	0.039
Toluene	0.012	0.029

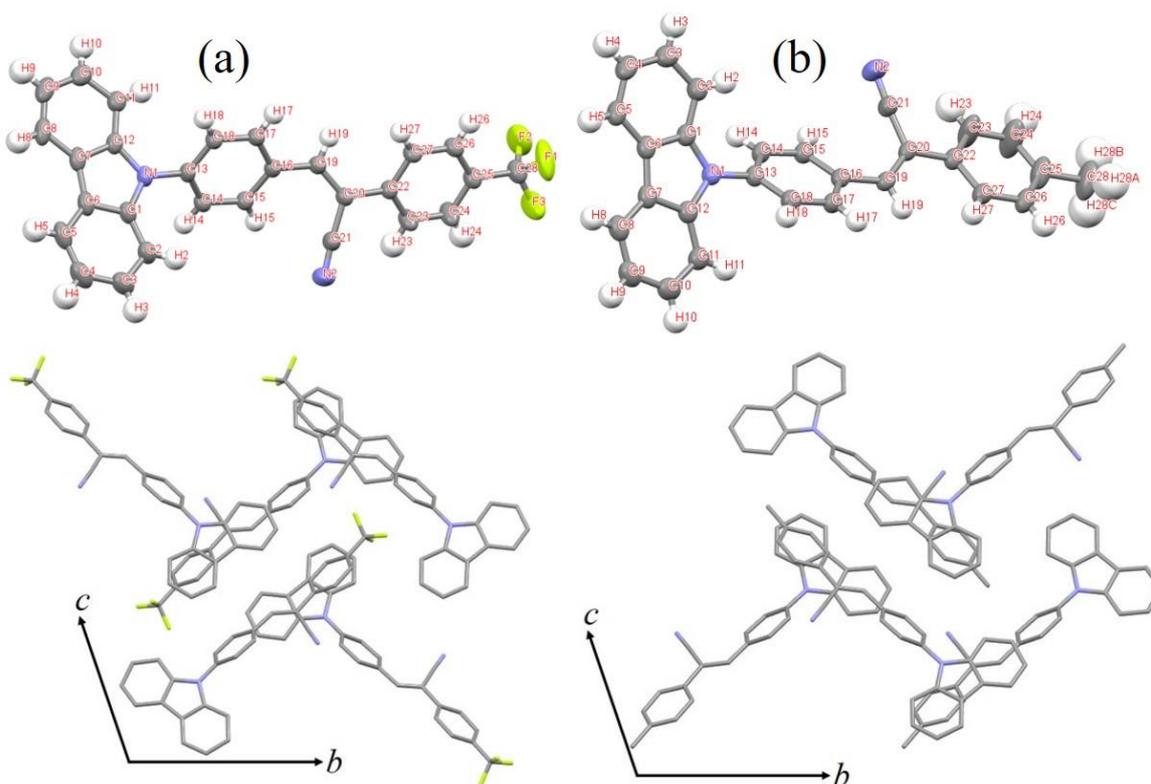


Fig. S3. Thermal ellipsoid and molecular packing in the crystal lattice of (a) **Cz-4-CF<sub>3</sub>** and (b) **Cz-4-CH<sub>3</sub>**. Hydrogen atoms are omitted for clarity. C (grey), N (blue) and F (yellow).

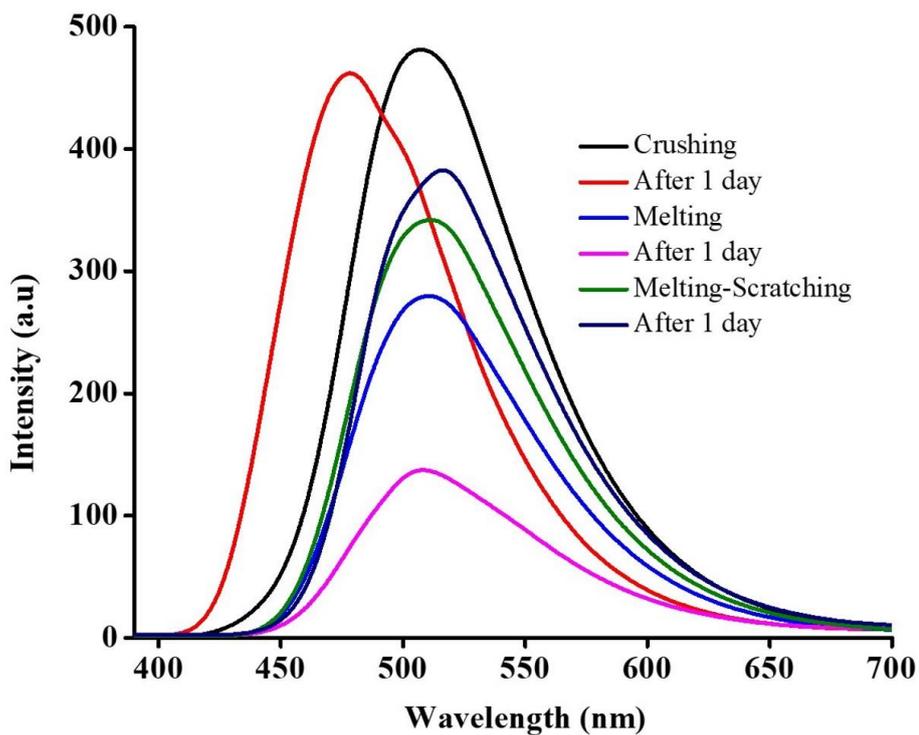


Fig. S4. Self-reversible fluorescence spectra of Cz-4-CF<sub>3</sub> after different external treatment.

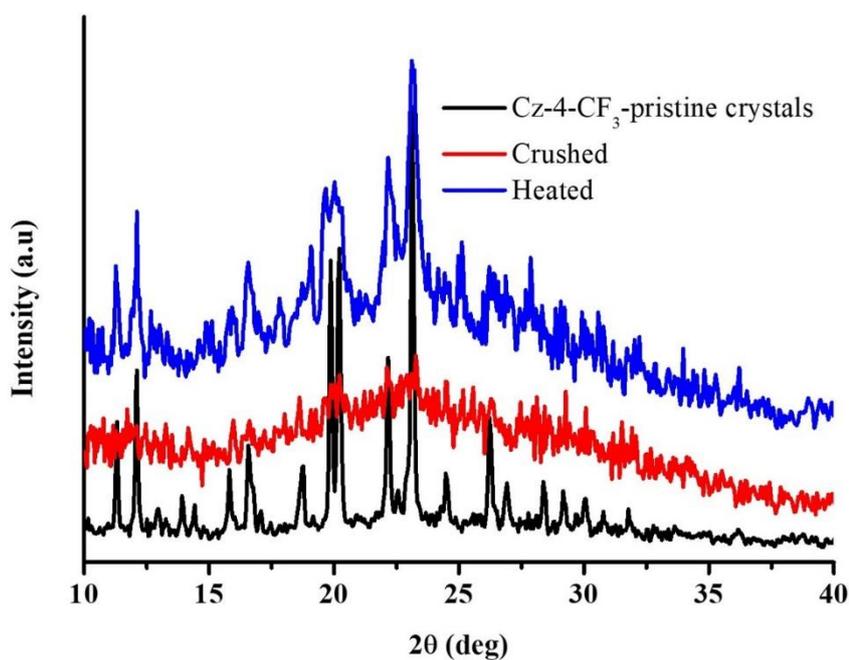


Fig. S5. PXRD pattern of Cz-4-CF<sub>3</sub>. Crushing sample measurement was performed immediately after crushing since it would self-reverse to initial state with time.

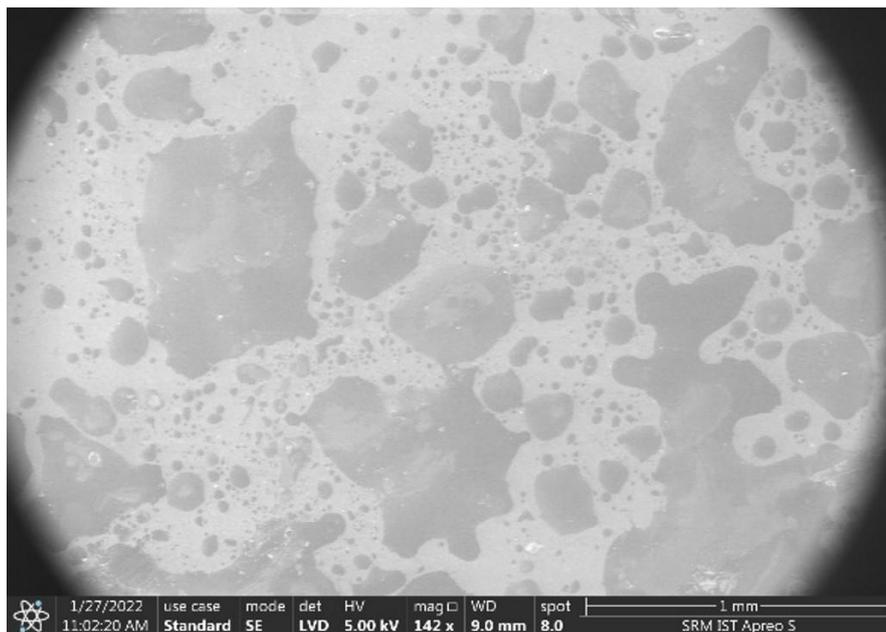


Fig. S6. SEM image of Cz-4-CF<sub>3</sub> at melt state.

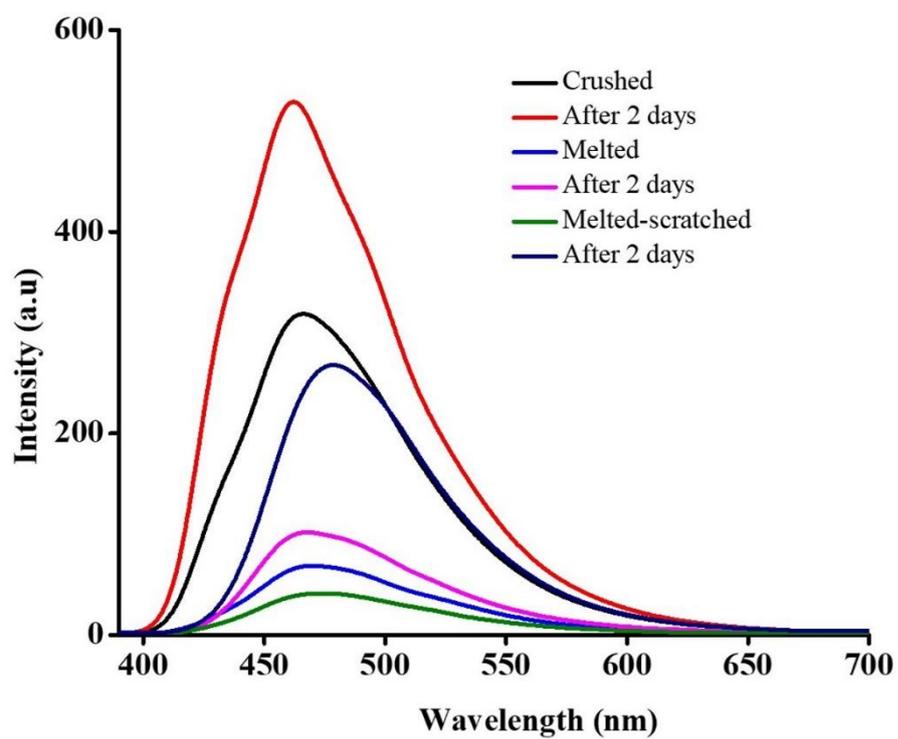


Fig. S7. Self-reversible fluorescence spectra of Cz-4-CH<sub>3</sub> after different external treatment.

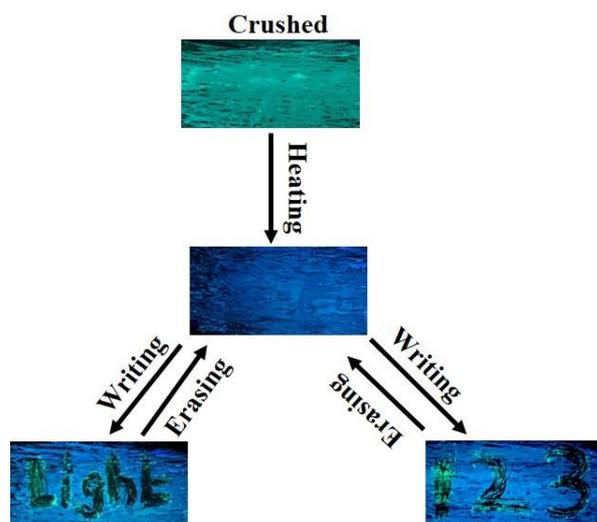


Fig. S8. Demonstrating rewritable fluorescent platforms using **Cz-4-CF<sub>3</sub>** by crushing and heating.  $\lambda_{\text{exc}} = 365 \text{ nm}$ .

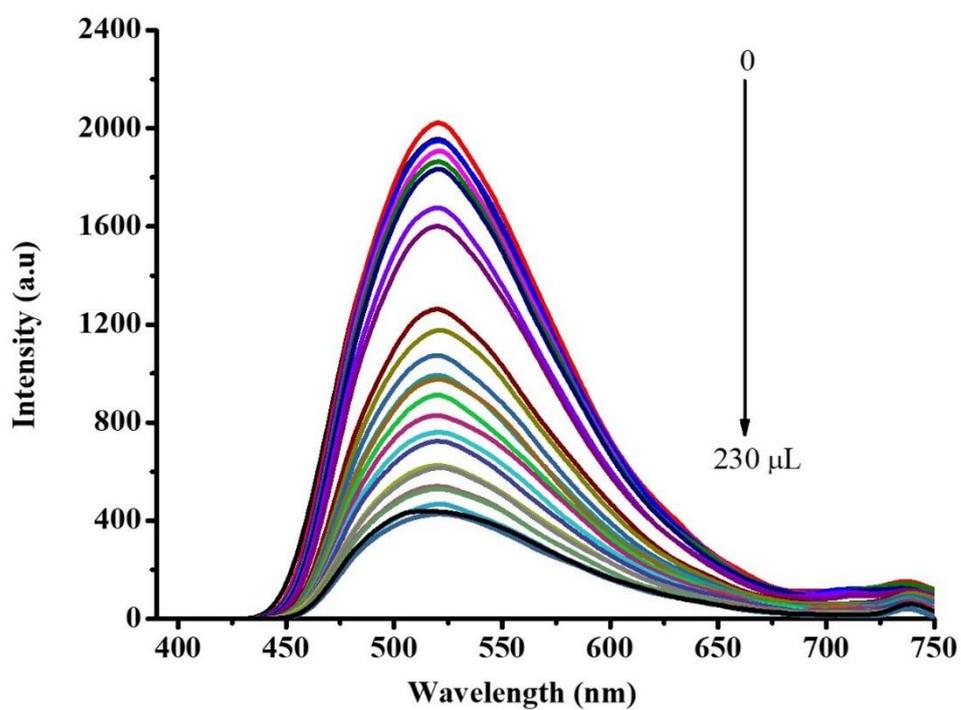


Fig. S9. PA ( $10^{-4} \text{ M}$ ) concentration dependent fluorescence change of Cz-4-CF<sub>3</sub> ( $10^{-4} \text{ M}$ ).

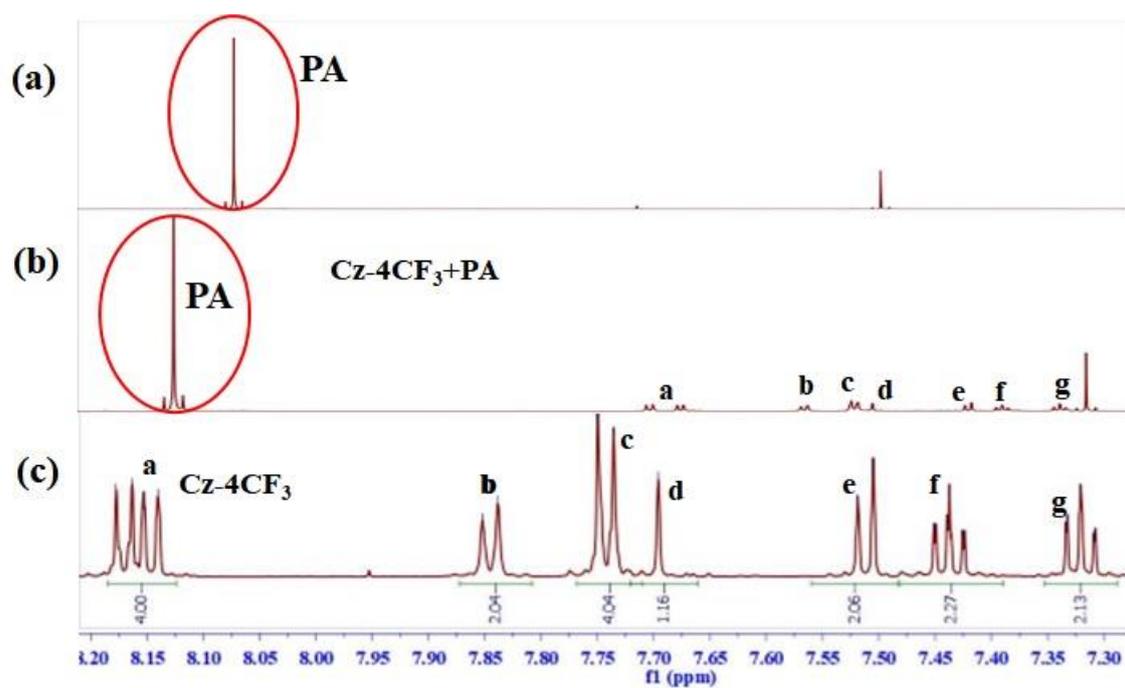


Fig. S10. (a)  $^1\text{H}$  NMR spectra of **PA** (ii) upon the addition of 1 equiv of **PA** into **Cz-4-CF<sub>3</sub>**(1:1 ratio); (iii)  $^1\text{H}$  NMR spectra of **Cz-4-CF<sub>3</sub>** in  $\text{CDCl}_3$ .

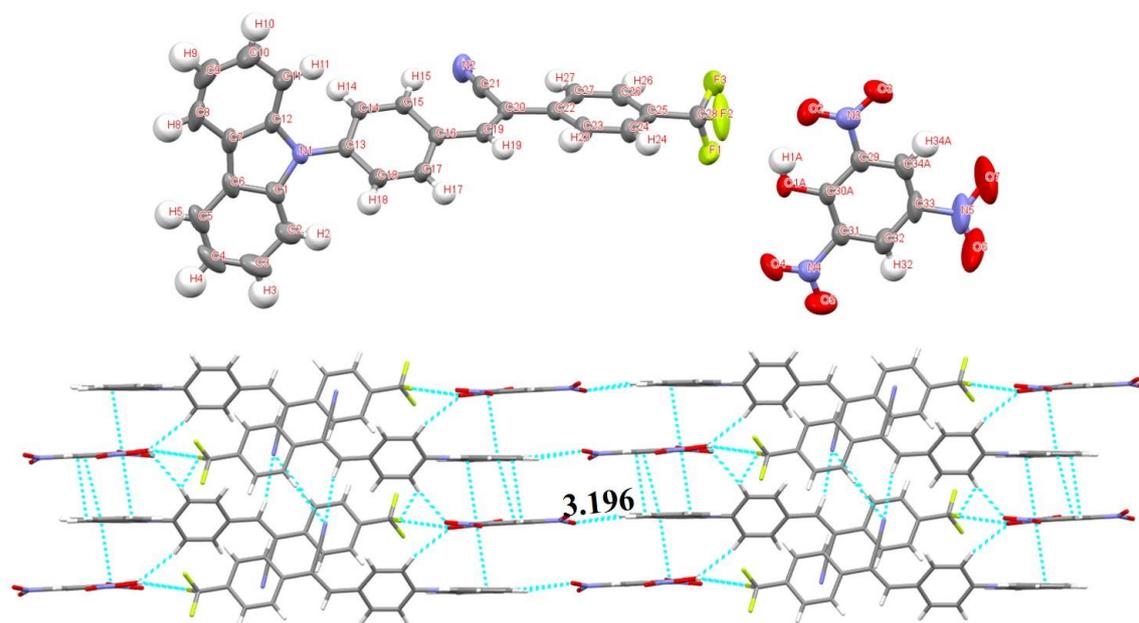


Fig. S11. Thermal ellipsoid and aromatic  $\pi$ -stacking between carbazole and **PA** aromatic unit and intermolecular interactions in the crystal lattice of **Cz-4-CF<sub>3</sub>-PA**. C (grey), H (white), N (blue), O (red) and F (yellow). Dotted lines indicate the hydrogen bonding and  $\pi \dots \pi$  interactions in  $\text{\AA}$  and distances along with e.s.d values are 3.196 (6).

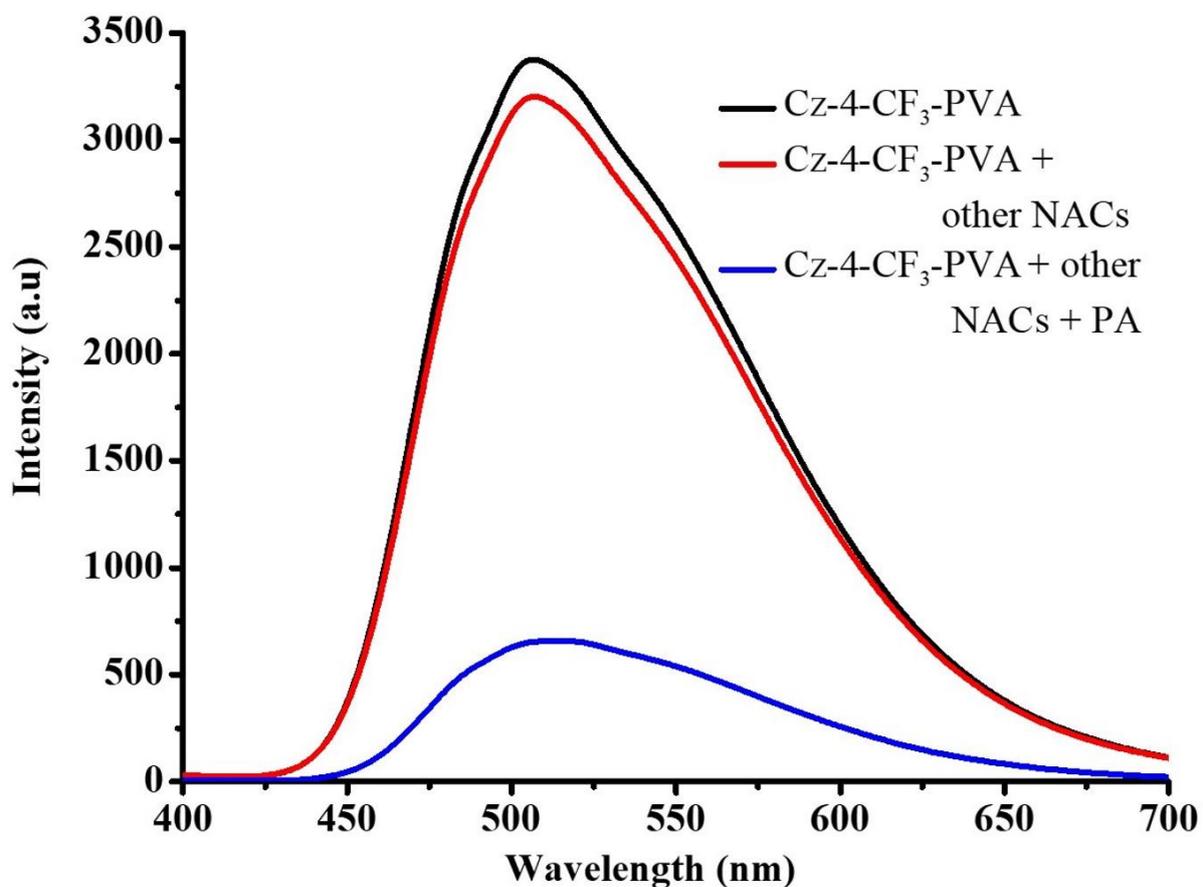


Fig. S12. Fluorescence sensing of Cz-4-CF<sub>3</sub>-PVA thin film towards PA ( $10^{-4}$  M) in presence of other nitroaromatics (NACs,  $10^{-4}$  M).

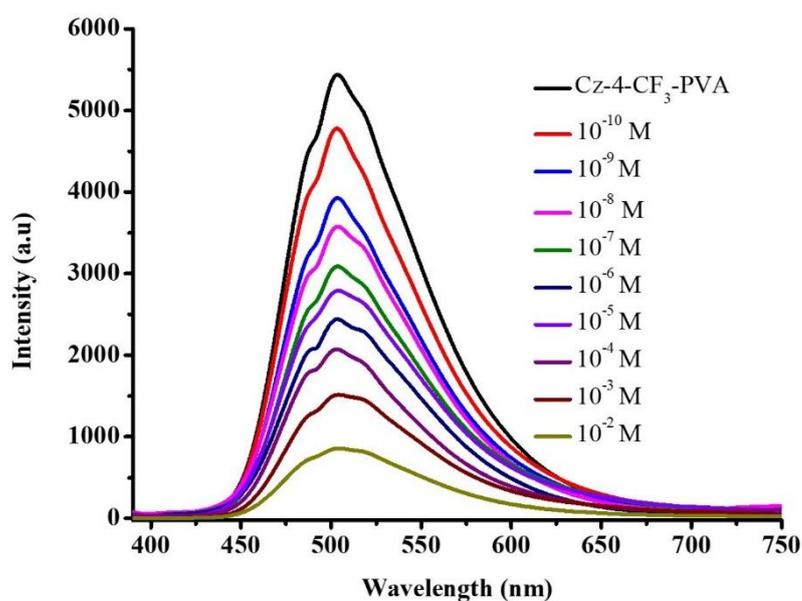


Fig. S13. Fluorescence sensing of Cz-4-CF<sub>3</sub>-PVA thin film towards PA in different concentration. The film was dipped for 30s and recovered back upon dipping in pure water for 1-2 min.

Table S2. Cz-4-CF<sub>3</sub>-PVA thin film sensing of PA in real water samples with known concentration.

Sample	Spiked (M)	Detected (M)	Recovery (%)
Sea water	10 <sup>-3</sup>	1.05x10 <sup>-3</sup>	105%
	10 <sup>-6</sup>	1.04x10 <sup>-6</sup>	104%
River water	10 <sup>-3</sup>	0.98x10 <sup>-3</sup>	98%
	10 <sup>-6</sup>	1.01x10 <sup>-6</sup>	101%
Pond water	10 <sup>-4</sup>	0.99x10 <sup>-4</sup>	99%
	10 <sup>-7</sup>	1.02x10 <sup>-7</sup>	102%
Lake water	10 <sup>-4</sup>	0.96x10 <sup>-4</sup>	96%
	10 <sup>-7</sup>	1.03x10 <sup>-7</sup>	103%

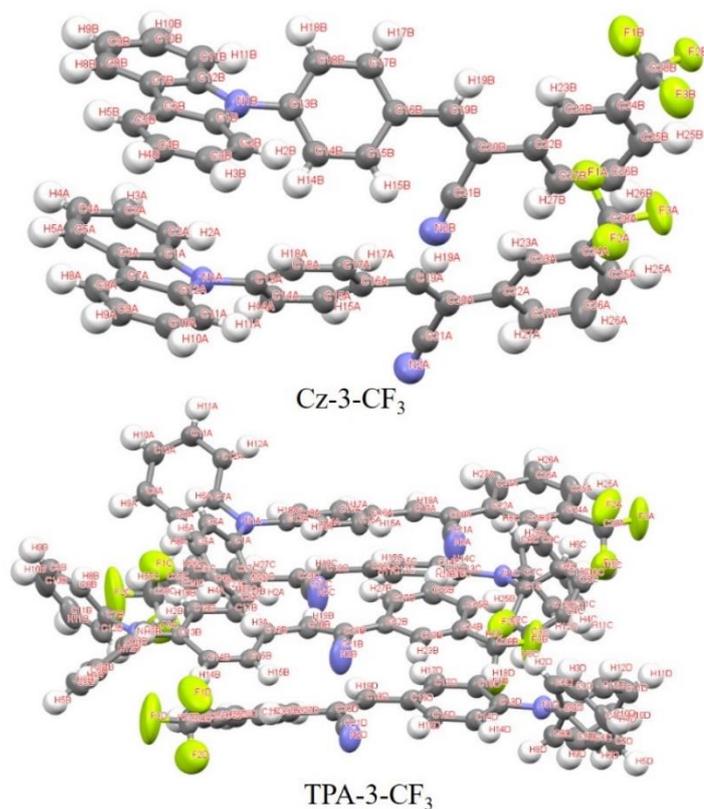


Fig. S14. Thermal ellipsoids of Cz-3-CF<sub>3</sub> and TPA-3-CF<sub>3</sub>.

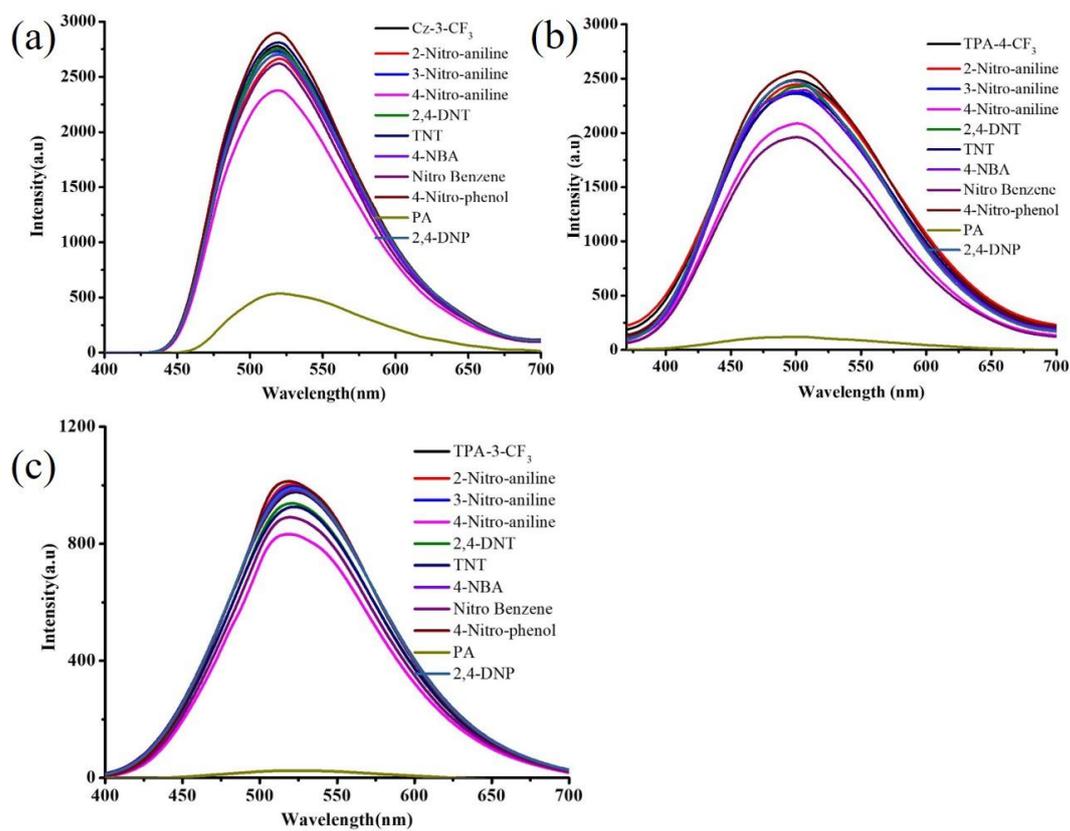


Fig. S15. Fluorescence sensing of (a) Cz-3-CF<sub>3</sub>, (b) TPA-4-CF<sub>3</sub> and (c) TPA-3-CF<sub>3</sub> towards PA (10<sup>-4</sup> M).

Table S3. Comparative study with various types of PA probes reported recently.

Chemosensors	Other NACs interferences	LOD	Mode of detection	Reusability	Reference
Cz-4CF <sub>3</sub>	Nil	51.4nm	Solution, Solid and polymer film	Reusable in filter paper, thin film by washing with water	Present work
Organogelator	4-NP, 2,4-DNP	700ppt	Gel/solution state	-----	[1]
Conjugated polymer nanoparticles	Nil	Not reported	Solution, solid and vapor phase	-----	[2]
Curcumin-BF <sub>2</sub> complexes	Nil	4.21nM 11.61 nM	Test Strips, solution	-----	[3]
Conjugated polyelectrolyte (PMI)	4-NP, 2,4-DNP	$56.11 \times 10^{-11}M$	Test Strips, solution	-----	[4]
N,N,N-trimethyl-2-(pyren-1-yloxy)ethanaminium bromide (PyOEA)	Nil	23.2 nM	Test Strips, solution	-----	[5]
Triphenylamine based fluorophore	DNP	~5ppb	Solid and solution state	-----	[6]
Cyanostilbene derivatives	Not reported	$2.85 \times 10^{-7}$ $1.96 \times 10^{-6} M$	Solid and solution state	-----	[7]
Trifluoromethyle decorated MOFs	DNP, ONP, PNP	$3 \times 10^{-5}M$ $9 \times 10^{-6}M$	Solid and solution state	Reusable in solution by centrifuged and washed with methanol	[8]
Conjugated polymer (PFAM)	2,4-DNP, 4-NP	57.8nm	Solid, strips, solution	Reusable in filter paper, by washing with water	[9]
Pyrene-derived pH-responsive polymeric probe	4-NP, 2,4-DNP	56 $\mu M$	Solution and polymer film	Reusable in film, by washing with water	[10]
1,8-naphthalimide-conjugated sulfonamide probe	2,4-DNP 4-NP (at high conc)	25.6 $\mu M$	Strips and solution state	-----	[11]
Naphthalene based Schiff base	DNP, NP, DNT, DNBA, DNB	0.11 $\mu M$	Strips and solution state	-----	[12]
Ln(III) based probes	Nil	0.5 $\mu M$	Strips, sol-gel and solution state	-----	[13]
$\pi$ -Conjugated polymers	DNT, DNP, NB, NT	$47.39 \times 10^{-8} M$	Filter paper, solution	Reusable in filter paper by washing with water	[14]
Arylene-vinylene Terpyridine Conjugates	NB, NT, DNT, NBA, HNB, DNT	$1.31-2.94 \times 10^{-7} M$	Test Strips, solution	Reusable in filter paper by washing with water	[15]
Dansyl tagged copolymer	DNT, TNT	3.7 $\mu M$	Solution state, thin film and filter paper	-----	[16]
Quinoxaline-based luminogen	DNB, TNT	28.7nM	Test Strips, solution	-----	[17]
Pyrene appended imidazolium probe	4-NP, 2,4-DNP	10nM	solution	-----	[18]
Quinoline-benzimidazole conjugate	2,4-DNP, NA	4.86ppb	Test Strips, solution	-----	[19]
Fluorene based chemosensor	2,4-DNP	22ppt, 0.23ppt	Solution	-----	[20]
Cyanine based chemosensor	Nil	8.24nm, 8.44nm <sub>21</sub>	Test Strips, solution	-----	[21]

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