

Supporting Information

Enabling spectral barcoding of SERS nanotags using gold nanostar

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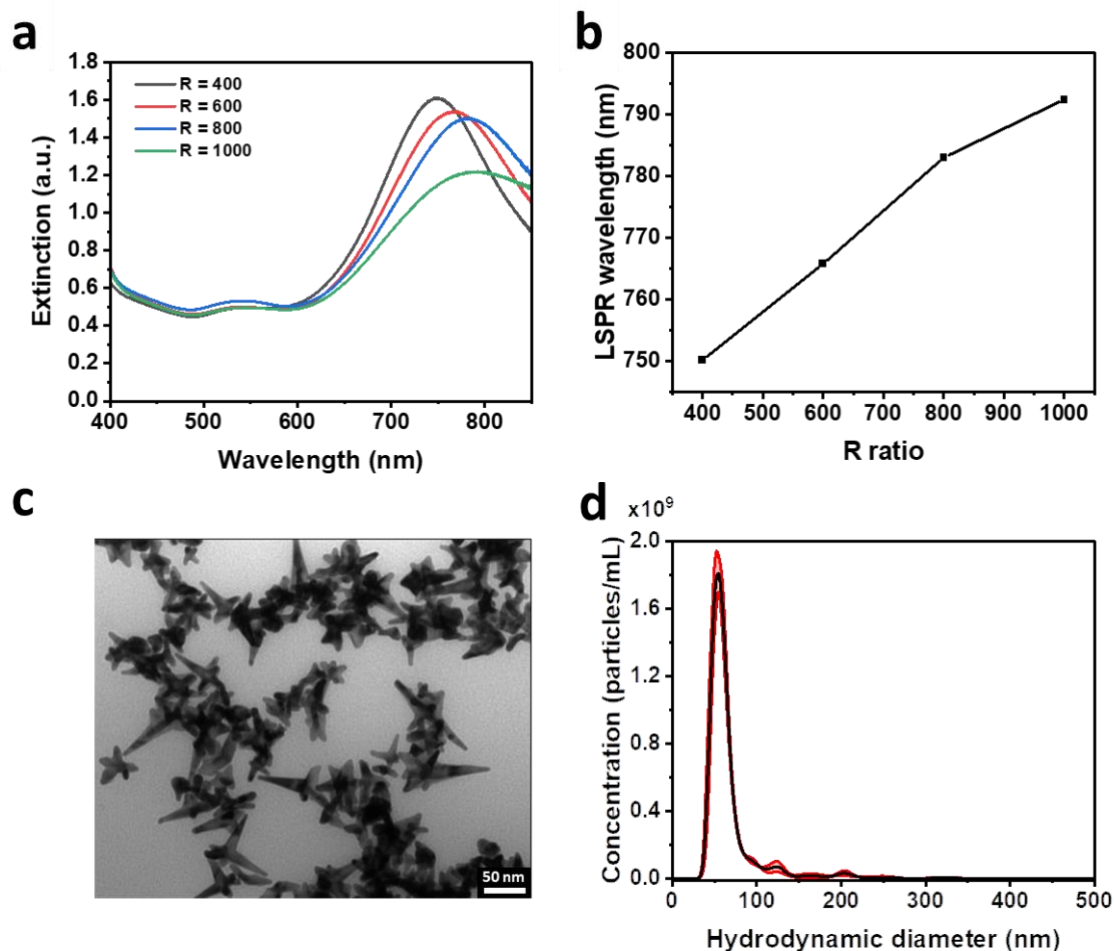


Figure S1. (a) UV-Vis absorption spectra of as-synthesized AuNSs with various molar ratio of HEPES to HAuCl₄ (R); (b) LSPR peak wavelength of the synthesized AuNS as function of R value; (c) TEM image of AuNS synthesized at the R = 800 as the LSPR wavelength of this solution was nearest to the laser excitation wavelength of 785 nm (at magnification 130,000 and applied voltage of 100 kV); (d) NTA of three replicates of AuNSs synthesized at R = 800 indicating the average hydrodynamic diameter of each AuNS at 53.5 ± 2.3 nm and the average concentration of AuNSs at $(4.99 \pm 0.06) \times 10^{10}$ particles/mL.

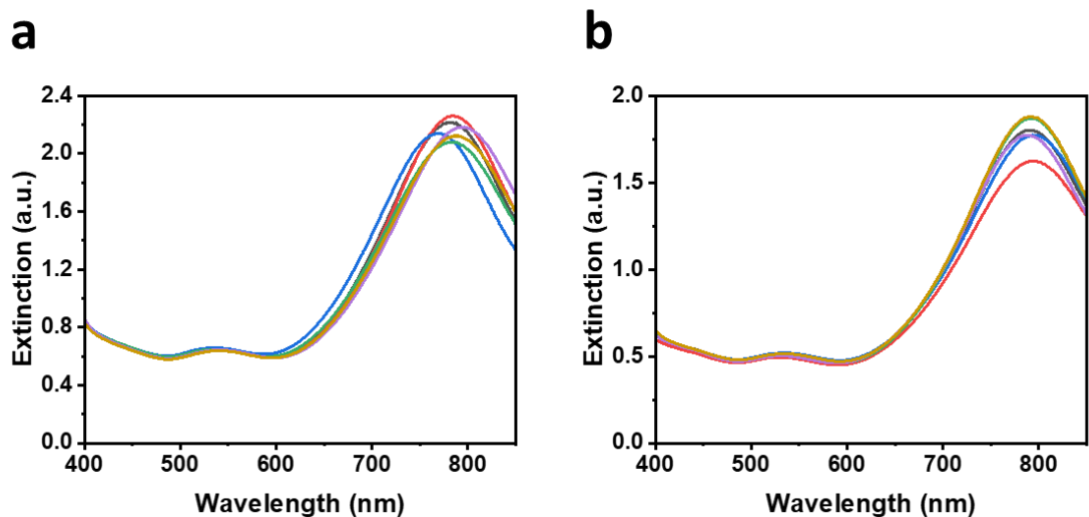


Figure S2. UV-Vis spectra of synthesized AuNSs in different volumes of batches. Reproducibility of synthesized AuNSs as changing the batch size from 10 mL to 20 mL with six vials of AuNS solution at each size. LSPR wavelengths of 10 mL batches and 20 mL batches were in range of 775 – 795 nm (a) and 788 – 794 nm (b).

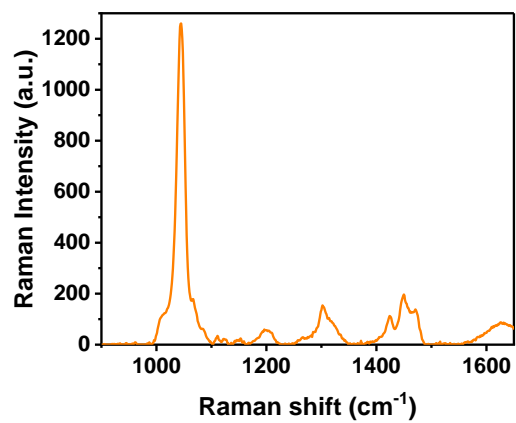
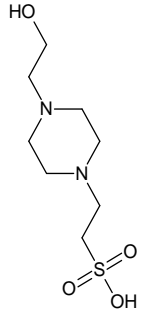


Figure S3. Raman spectrum of 0.2 M of HEPES solution using laser excitation of 785 nm.

Table S1. Band assignments to the Raman fingerprint of HEPES¹⁻³

Molecular structure	Raman shift (cm ⁻¹)	Assignment
4-(2-hydroxyethyl)-1-piperazineethanesulfonic acid 	1045	SO ₃ ⁻ symmetric stretch
	1196	C-N stretch
	1302	CH ₂ twist
	1424	CH ₂ deformation
	1450	CH ₂ scissor
	1625	H-O-H bend

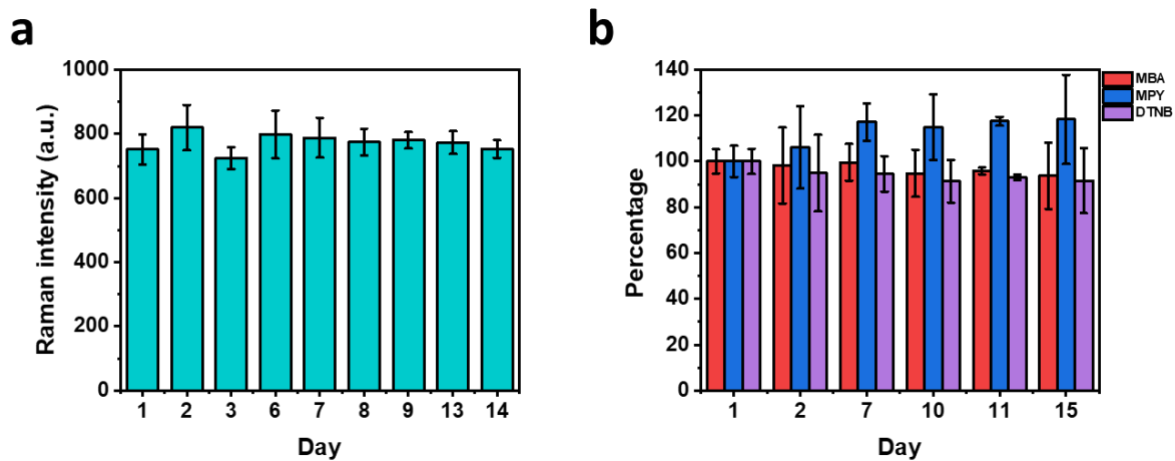


Figure S4. Stability of AuNS@TFMBA (a) and a mixture of three SERS nanotags (b) (AuNS@MBA, AuNS@MPY, and AuNS@DTNB) over two weeks. Raman intensity of each replicate was normalized by its extinction peak intensity then compared to the value on the first day to calculate the percentage. (Error bar stands for three replicates)

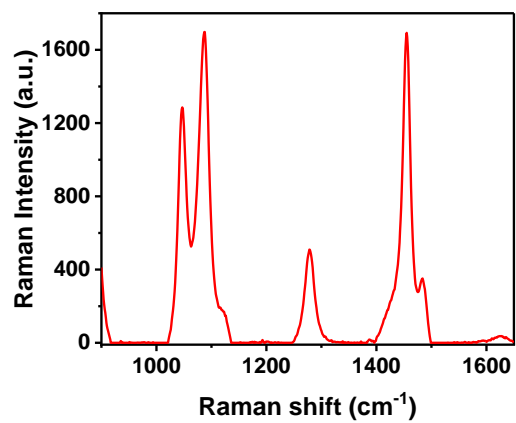


Figure S5. Raman spectrum of 1 M of TFMB in ethanol.

SI-1. Enhancement factor estimation of AuNSs

The diameter (d) and the pathlength (h) of the laser spot in the liquid: d = 40 μm, h = 10 mm.

$$V_{\text{laser spot}} = \pi \left(\frac{d}{2}\right)^2 \cdot h = \pi (20 \mu\text{m})^2 \cdot 10 \text{ mm} = 1.26 \times 10^{-5} \text{ cm}^3$$

For SERS

As can be seen in **Figure 3b**, the SERS intensity of AuNS@TFMBA increases with increasing the concentration of TFMBA, where it reaches a plateau at concentration about 0.9 mM. It means that 5 μl of TFMBA 0.9 mM could be enough for coating 1 mL of AuNS dispersion.

Thus, number of TFMBA molecules in 1 mL AuNSs solution:

$$\begin{aligned} N_{TFMBA} (AuNS@TFMBA) &= (0.9 \text{ mM} \times 5 \mu\text{L}) \times N_A \\ &= \left(0.9 \times 10^{-3} \frac{\text{mol}}{\text{L}}\right) \times (5 \times 10^{-6} \text{ L}) \times \left(6.02 \times 10^{23} \frac{\text{molecules}}{\text{mol}}\right) \\ &= 2.71 \times 10^{15} \text{ molecules} \end{aligned}$$

Number of TFMBA molecules in laser spot:

$$\begin{aligned} N_{SERS} (AuNS@TFMBA) &= V_{\text{laser spot}} \cdot \frac{N_{TFMBA} (AuNS@TFMBA)}{1 \text{ mL}} \\ &= 1.26 \times 10^{-5} \text{ cm}^3 \cdot \frac{2.71 \times 10^{15} \text{ molecules}}{1 \text{ mL}} = 3.41 \times 10^{10} \text{ molecules} \end{aligned}$$

$$I_{SERS} (AuNS@TFMBA) = 635.31$$

For Raman

Concentration of TFMBA for reference: 1 M = 1 mol/L

Number of TFMBA molecules in 1L: $N'_{TFMBA} = 6.02 \times 10^{23}$ molecules

Thus, number of TFMBA in laser spot:

$$\begin{aligned} N_{Raman} &= V_{\text{laser spot}} \cdot \frac{N'_{TFMBA}}{1 \text{ L}} = 1.26 \times 10^{-5} \text{ cm}^3 \cdot \frac{6.02 \times 10^{23} \text{ molecules}}{1 \text{ L}} \\ &= 1.26 \times 10^{-5} \text{ cm}^3 \cdot \frac{6.02 \times 10^{23} \text{ molecules}}{1000 \text{ cm}^3} = 7.59 \times 10^{15} \text{ molecules} \end{aligned}$$

$$I_{Raman} = 5863.85$$

Overall, EF of AuNS after conjugation with TFMBA can be calculated as below:

$$\begin{aligned} EF (AuNS) &= \frac{I_{SERS} (AuNS@TFMBA)}{I_{Raman}} \cdot \frac{N_{Raman}}{N_{SERS} (AuNS@TFMBA)} \\ &= \frac{635.31}{5863.85} \cdot \frac{7.59 \times 10^{15} \text{ molecules}}{3.41 \times 10^{10} \text{ molecules}} = 2.41 \times 10^4. \end{aligned}$$

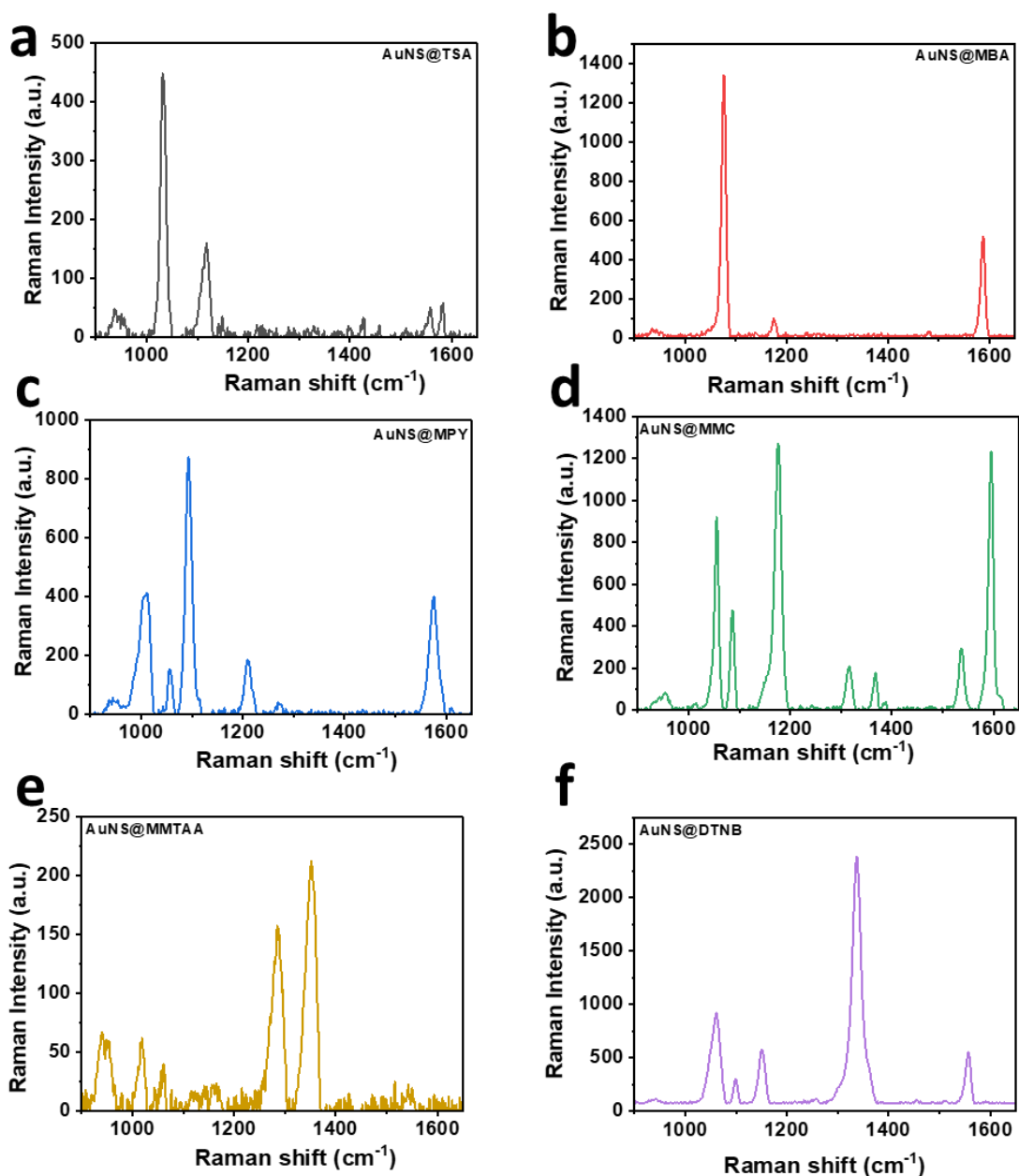


Figure S6. SERS spectra of AuNS coated with seven Raman reporter molecules. The most significant peak of each SERS nanotag was recorded, including (a) 1032 cm^{-1} of AuNS@TSA (b) 1077 cm^{-1} of AuNS@MBA (c) 1093 cm^{-1} AuNS@MPY (d) 1176 cm^{-1} of AuNS@MMC (e) 1351 cm^{-1} of AuNS@MMTAA (f) 1337 cm^{-1} of AuNS@DTNB (g) 1378 cm^{-1} of AuNS@TFMBA

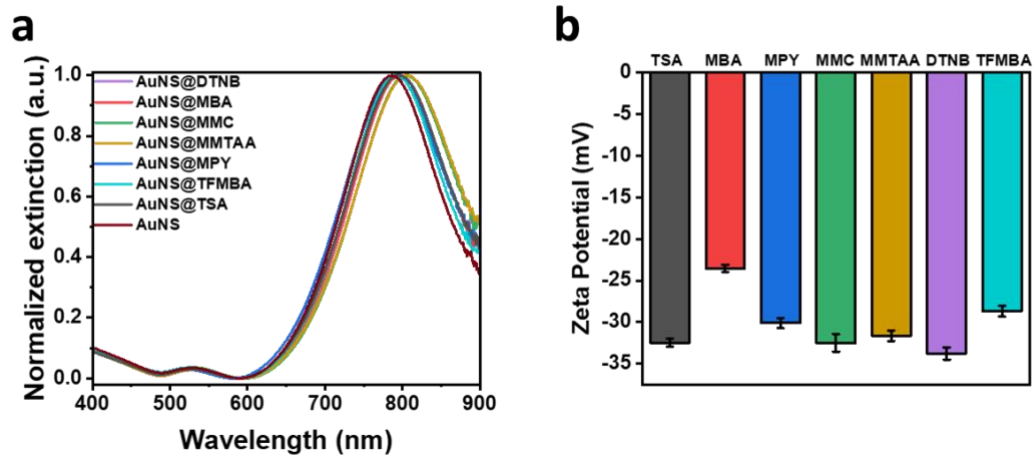


Figure S7. UV-Vis spectra (a) and zeta potential (b) of seven different SERS nanotags. UV-Vis spectra of AuNSs presented red shifts after conjugation with Raman reporter molecules (a) and all SERS nanotags had negative charges (b)

Table S2. 21 SERS spectra of two types of SERS nanotags

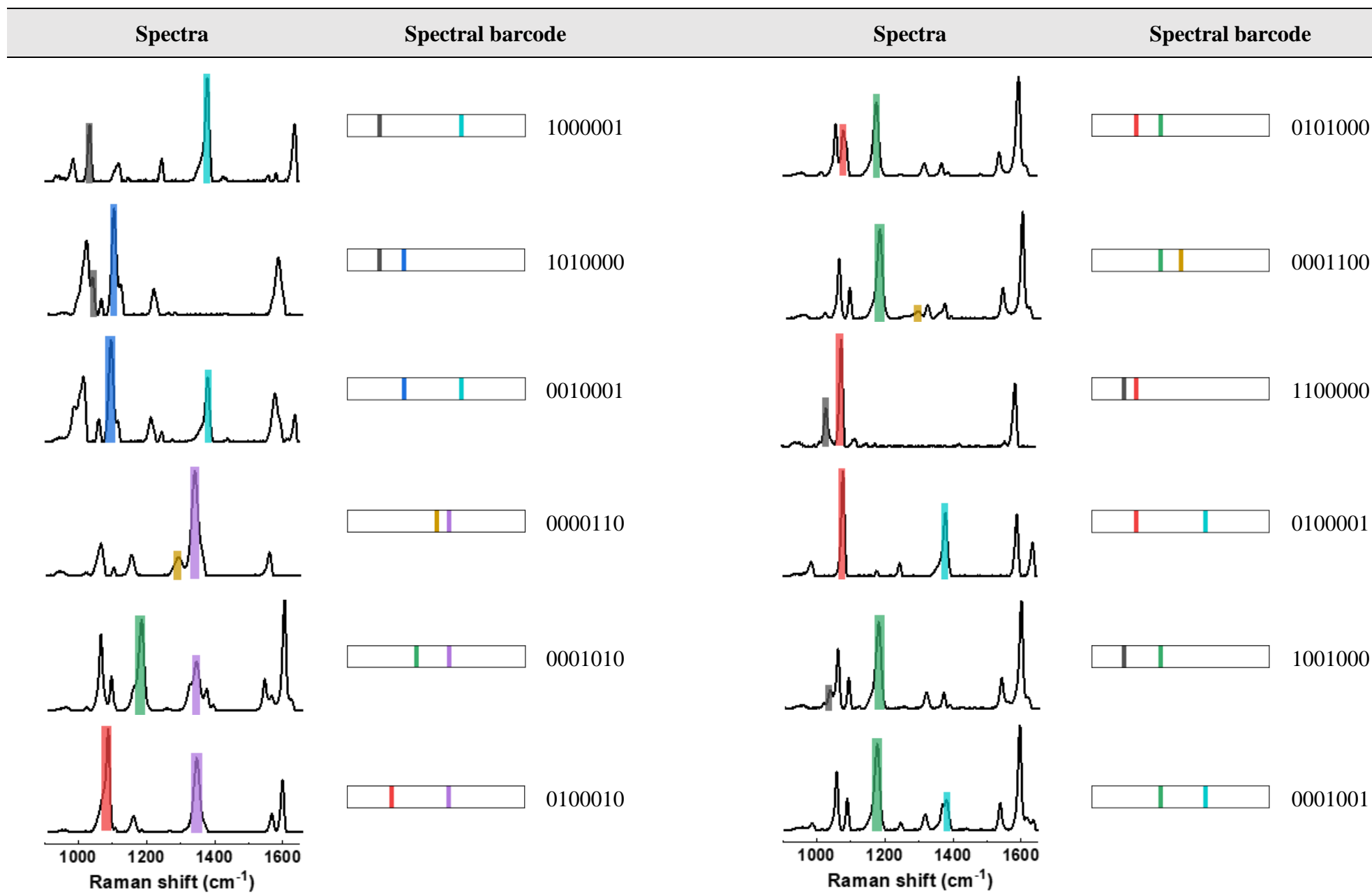


Table S2. 21 SERS spectra of two types of SERS nanotags (continued)

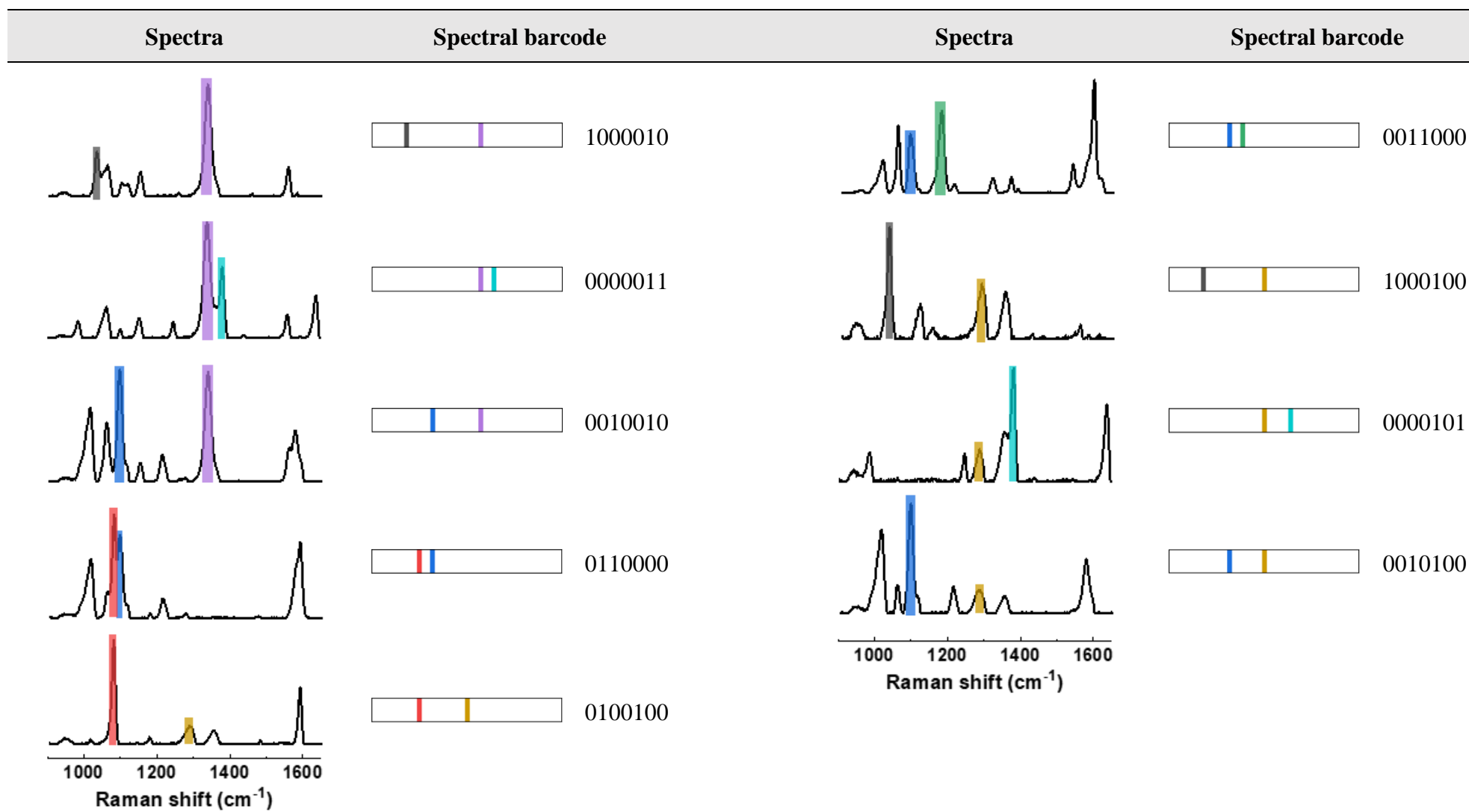


Table S3. 35 SERS spectra of three types of SERS nanotags

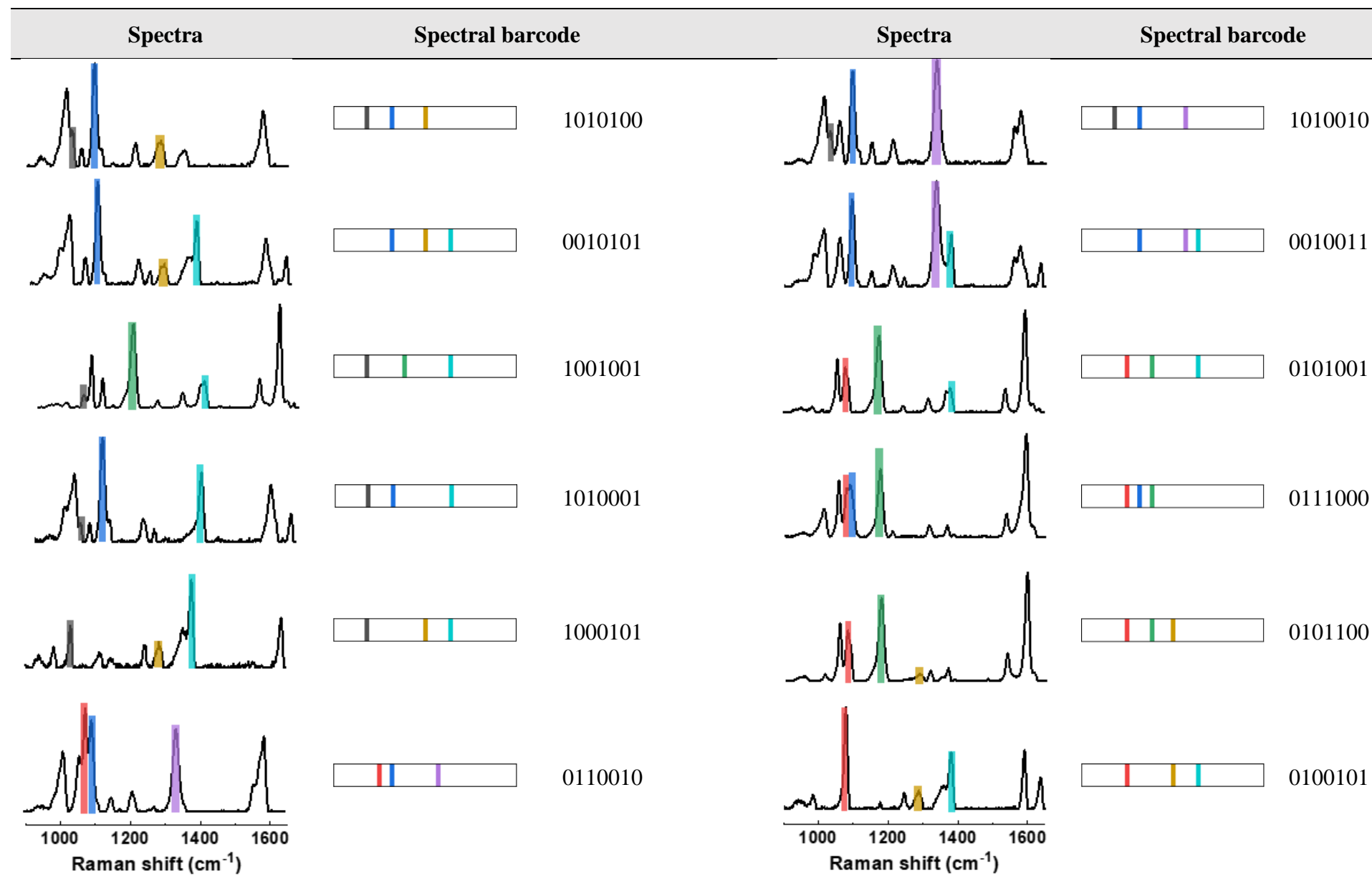


Table S3. 35 SERS spectra of three types of SERS nanotags (continued)

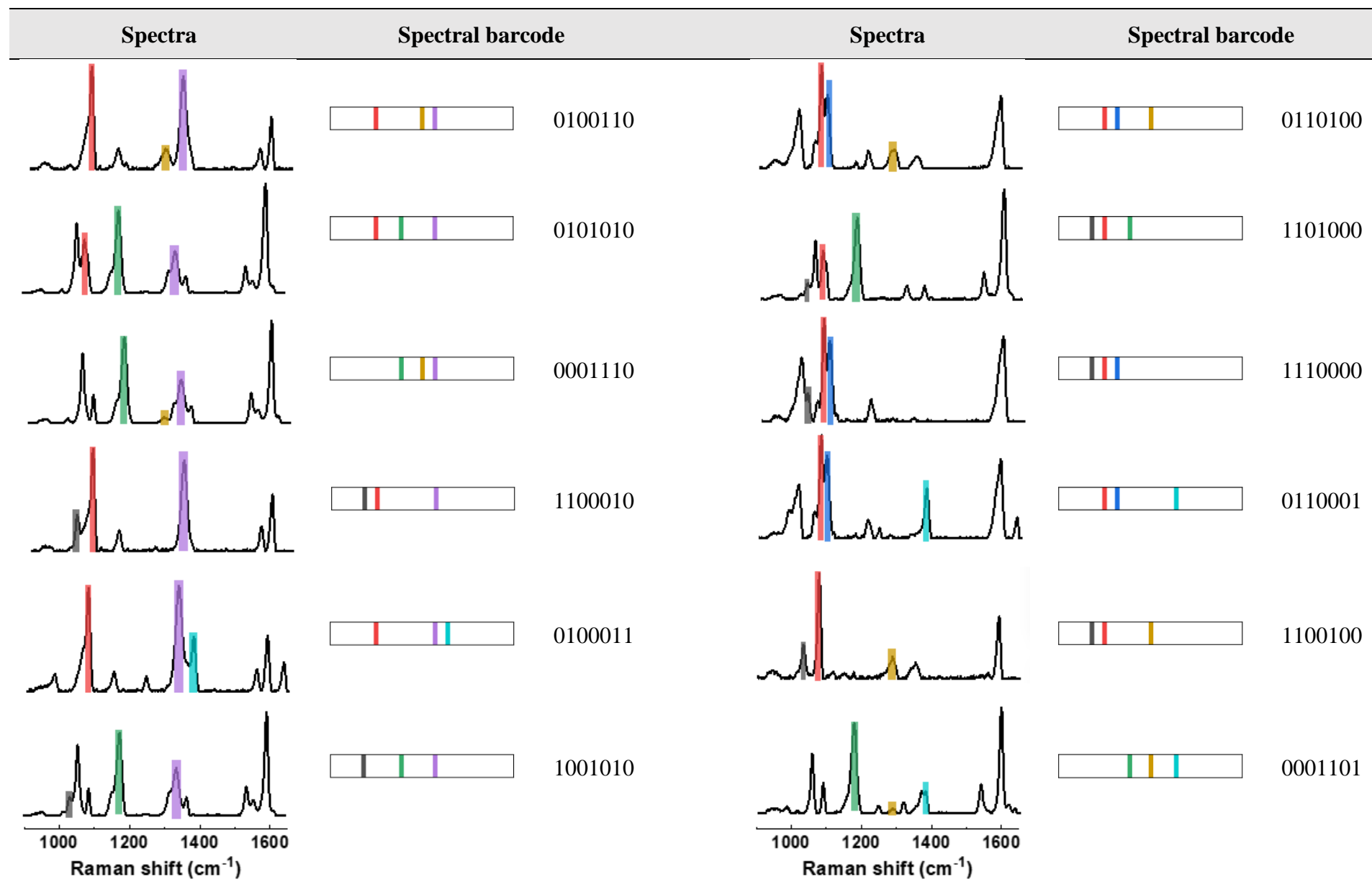
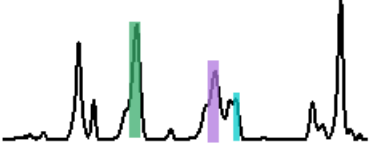

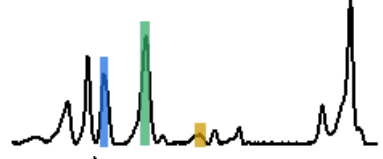

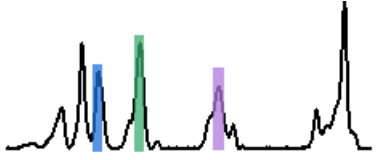

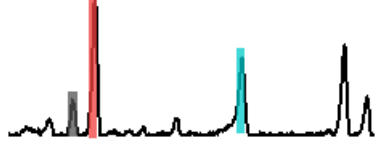

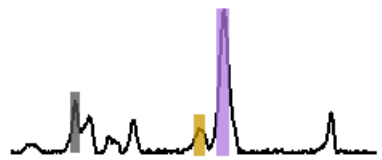
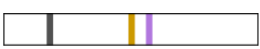
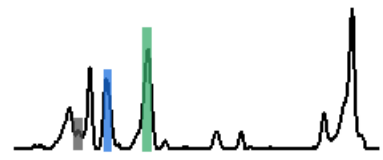

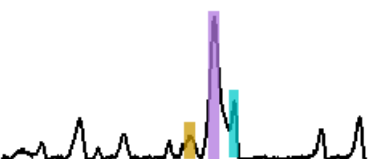

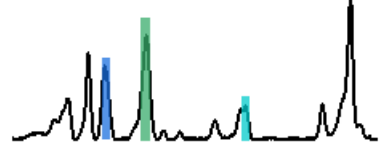



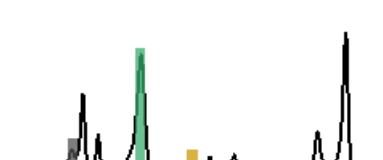

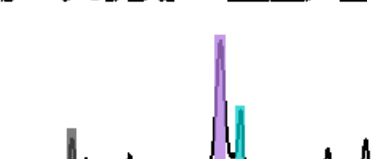

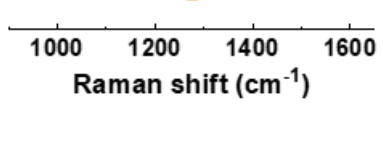



Table S3. 35 SERS spectra of three types of SERS nanotags (continued)

Spectra	Spectral barcode		Spectra	Spectral barcode
		0001011		
		0011010		
		1000110		
		0000111		
		0010110		
		1000011		

1000 1200 1400 1600
Raman shift (cm⁻¹)

Table S4. 35 SERS spectra of four types of SERS nanotags

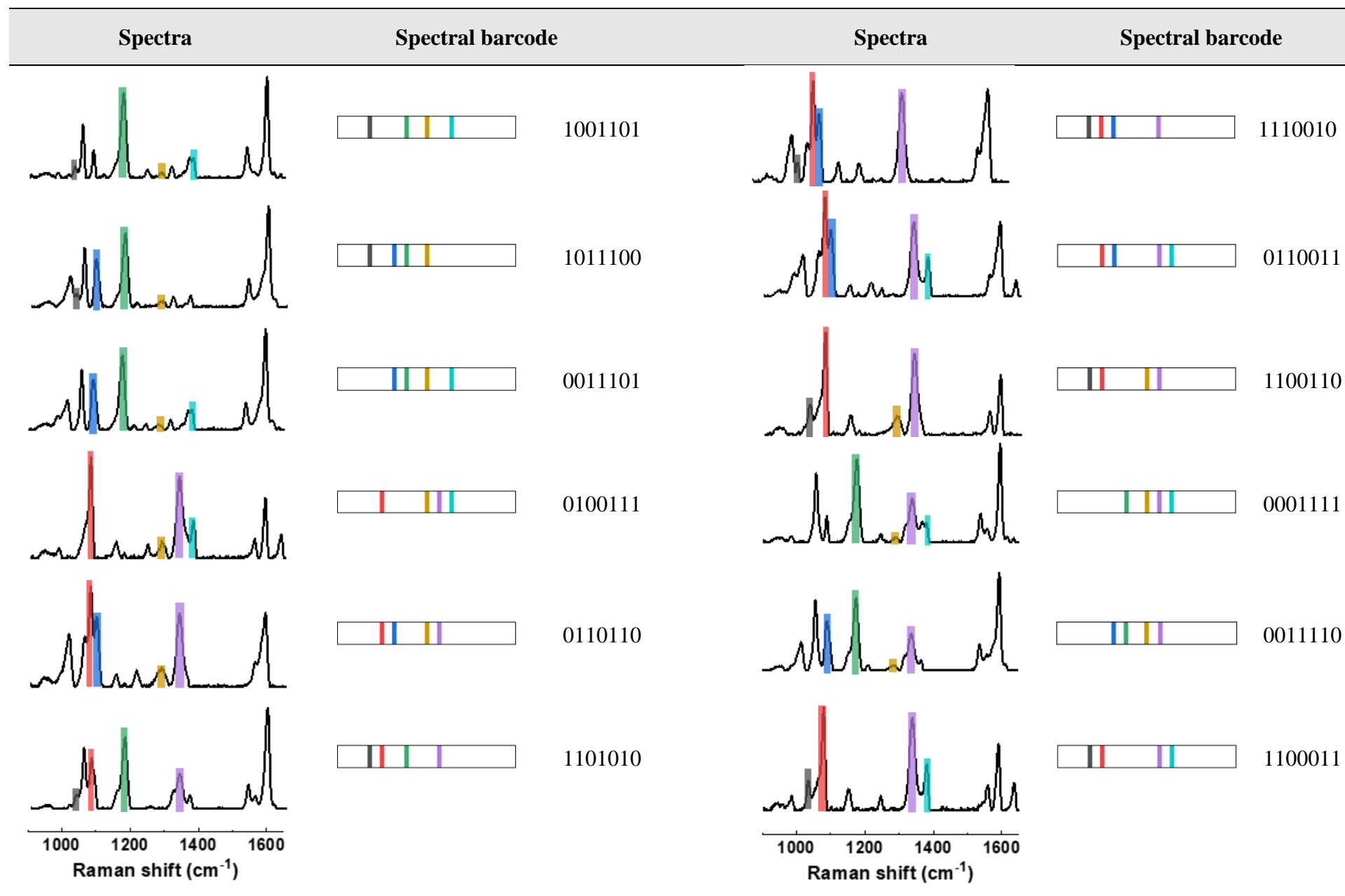


Table S4. 35 SERS spectra of four types of SERS nanotags (continued)

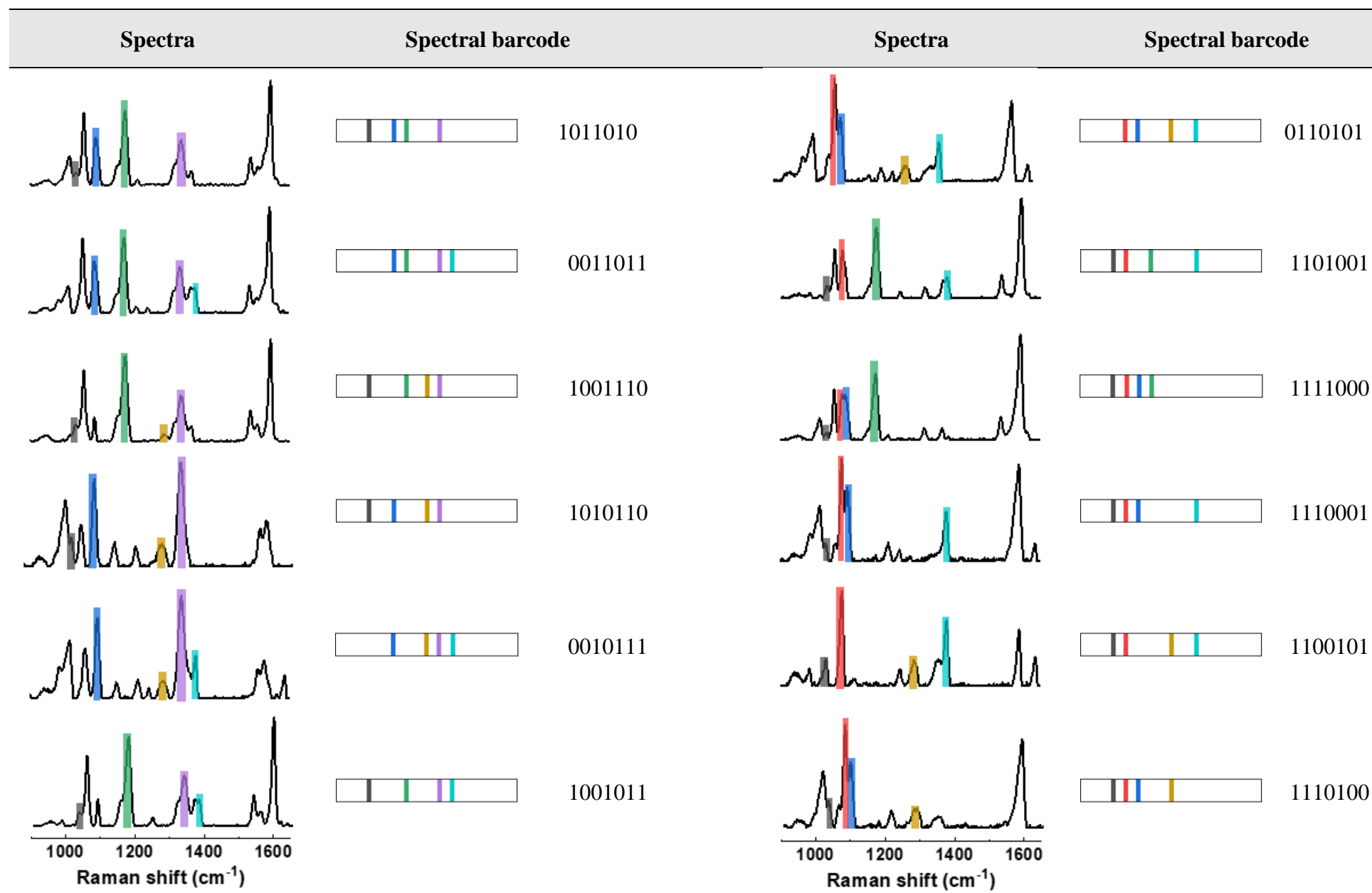


Table S4. 35 SERS spectra of four types of SERS nanotags (continued)

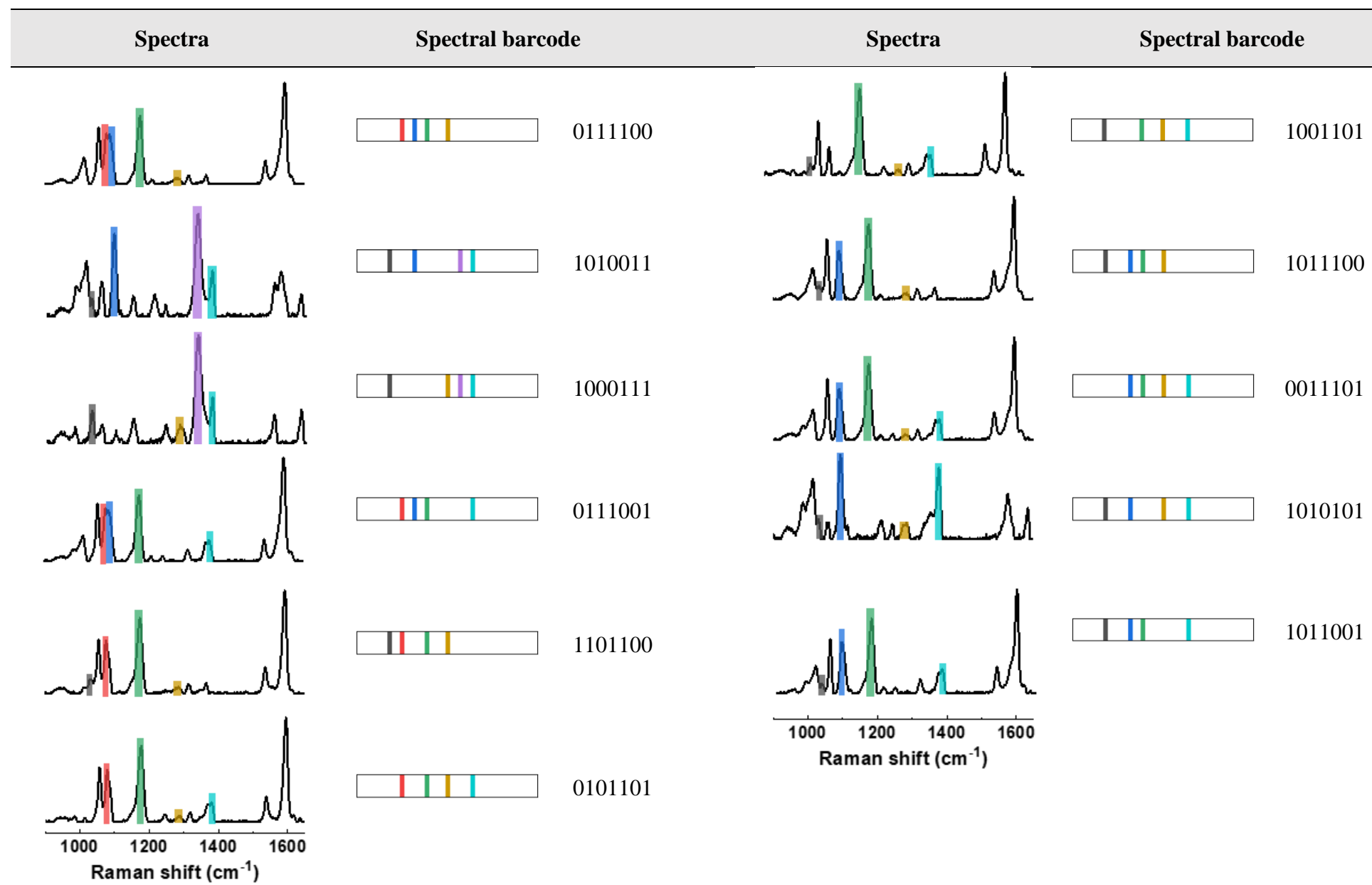


Table S5. 21 SERS spectra of five types of SERS nanotags

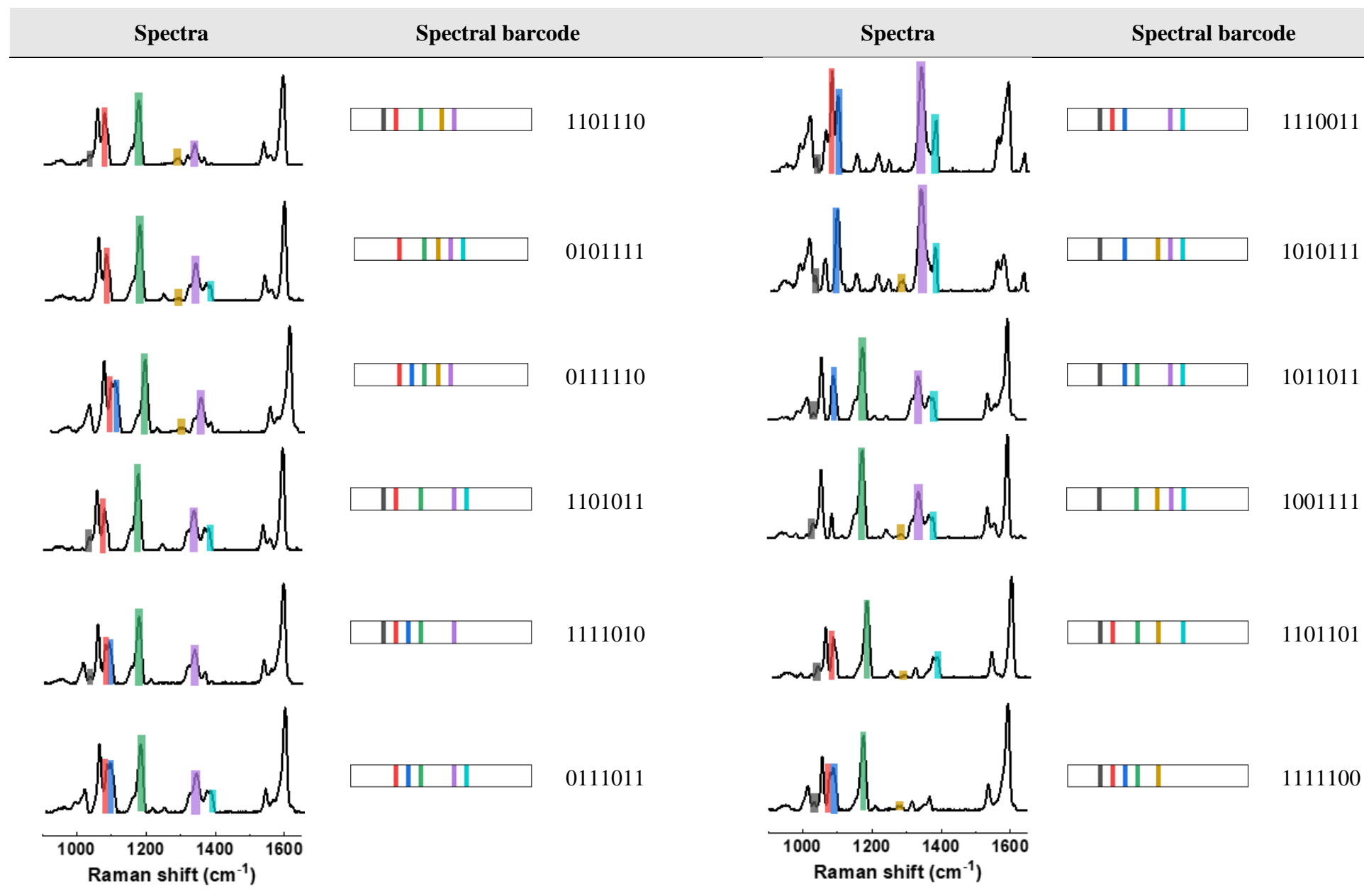
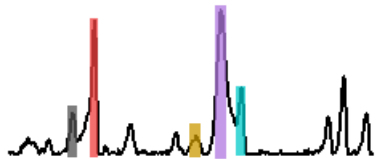
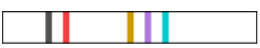
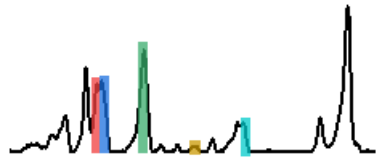

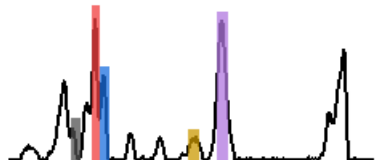

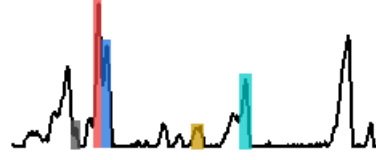

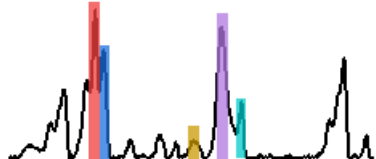

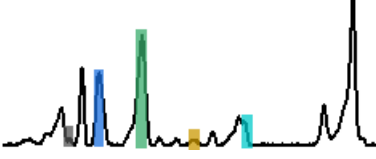

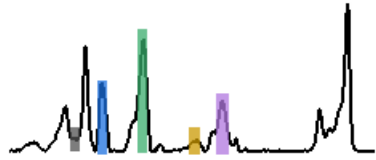

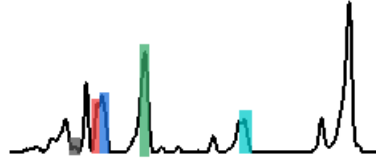
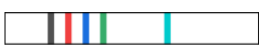
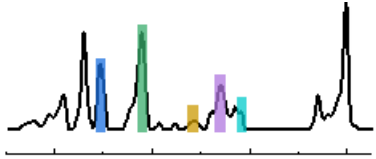



Table S5. 21 SERS spectra of five types of SERS nanotags (continued)

Spectra	Spectral barcode	Spectra	Spectral barcode
	 1100111		 0111101
	 1110110		 1110101
	 0110111		 1011101
	 1011110		 1111001
	 0011111		

1000 1200 1400 1600
Raman shift (cm⁻¹)

References

1. G. Socrates, *Infrared and Raman characteristic group frequencies: tables and charts*, John Wiley & Sons Ltd, Third edn., 2001.
2. W. Xi and A. J. Haes, *J Am Chem Soc*, 2019, **141**, 4034-4042.
3. A. A. Kananenka and J. L. Skinner, *J Chem Phys*, 2018, **148**, 244107.