

Supporting information for
Co-doped g-C₃N₄ nanotubes decorated separators
mediate polysulfide redox for high performance
lithium sulfur batteries

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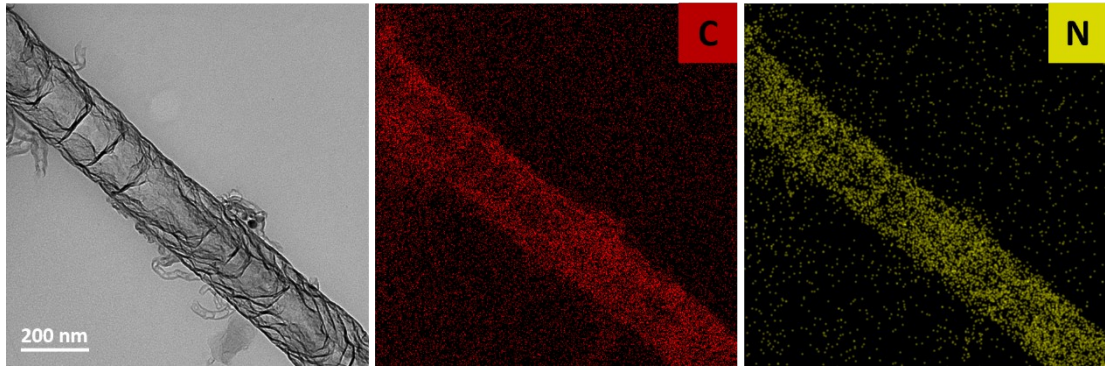


Fig. S1. STEM-EDX mapping of Co-TCN

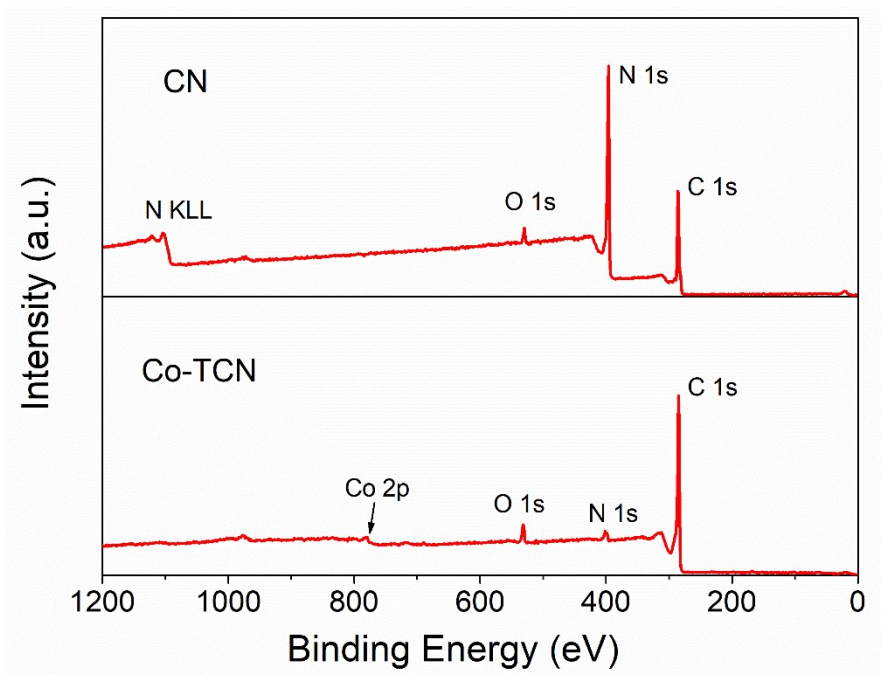


Fig. S2. XPS spectra of full spectrum for Co-TCN sample.

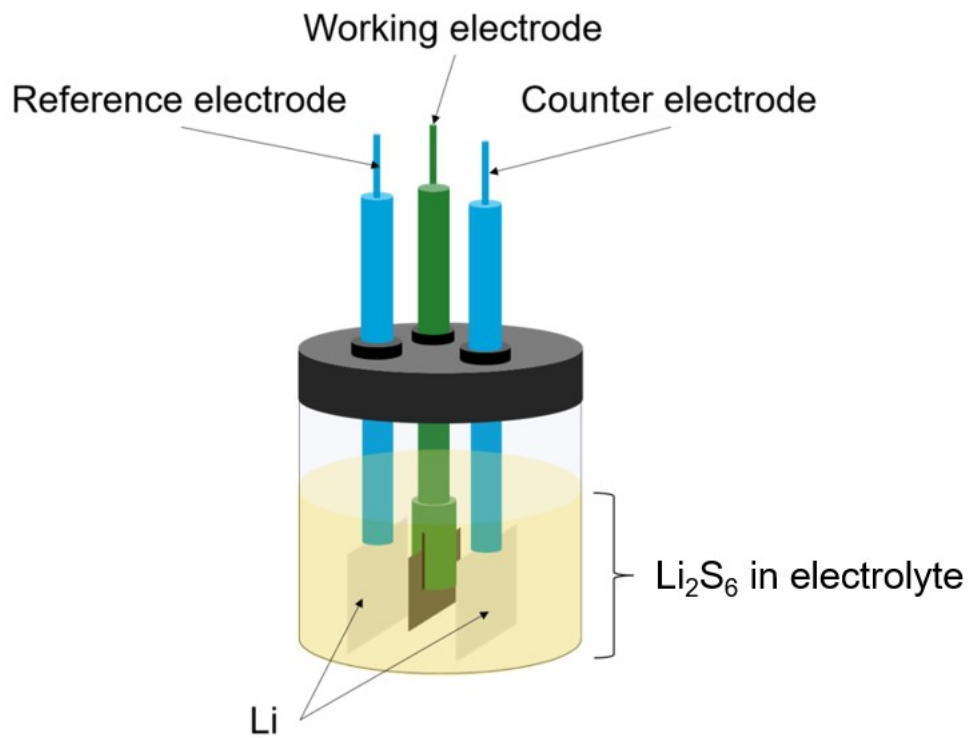


Fig. S3. Diagram of elucidated by exchange current measurements using linear scanning voltammetry (LSV) with a Li_2S_6 solution in a three-electrode cell.

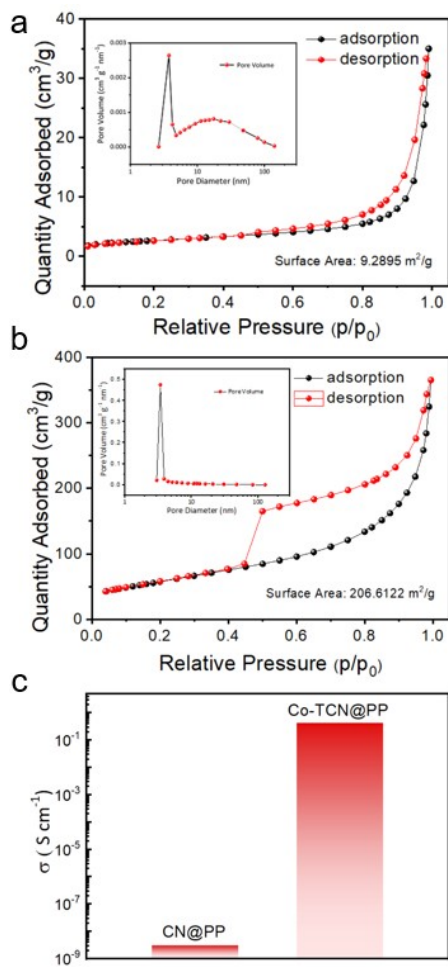


Fig. S4. N_2 adsorption/desorption isotherms of g) CN and h) Co-TCN. i) Conductivity of CN and Co-TCN.

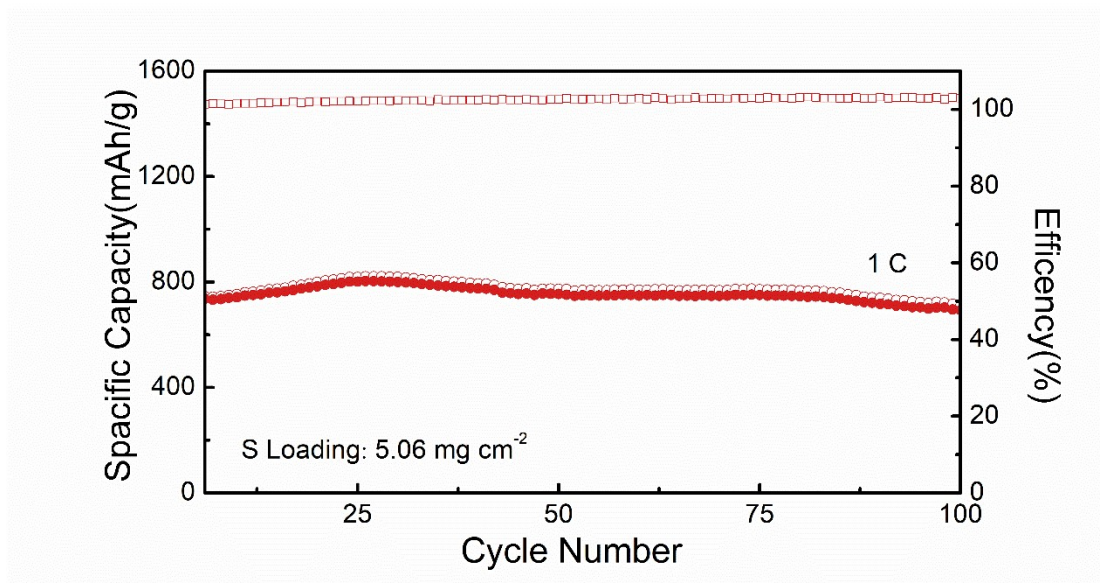


Fig. S5. Cycling performances of Co-TCN with high loading at current density of 1 C.

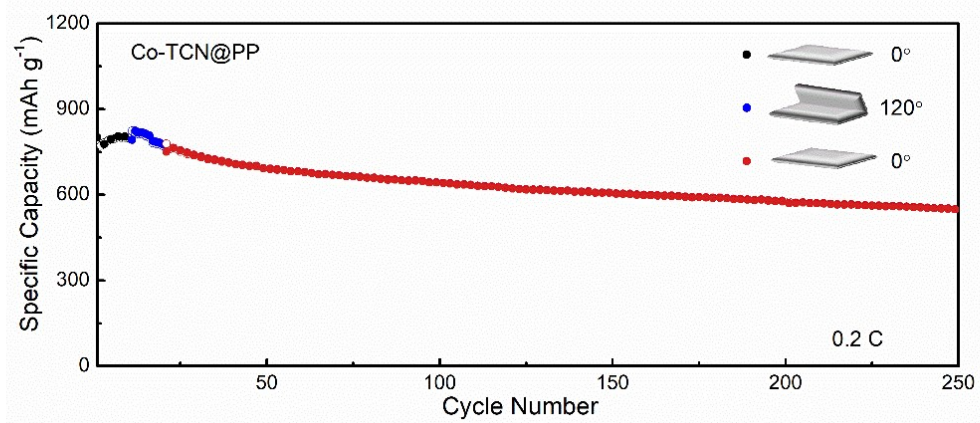


Fig. S6. Cycling performances of Co-TCN assembled Li-S pouch cell at 0.2 C.

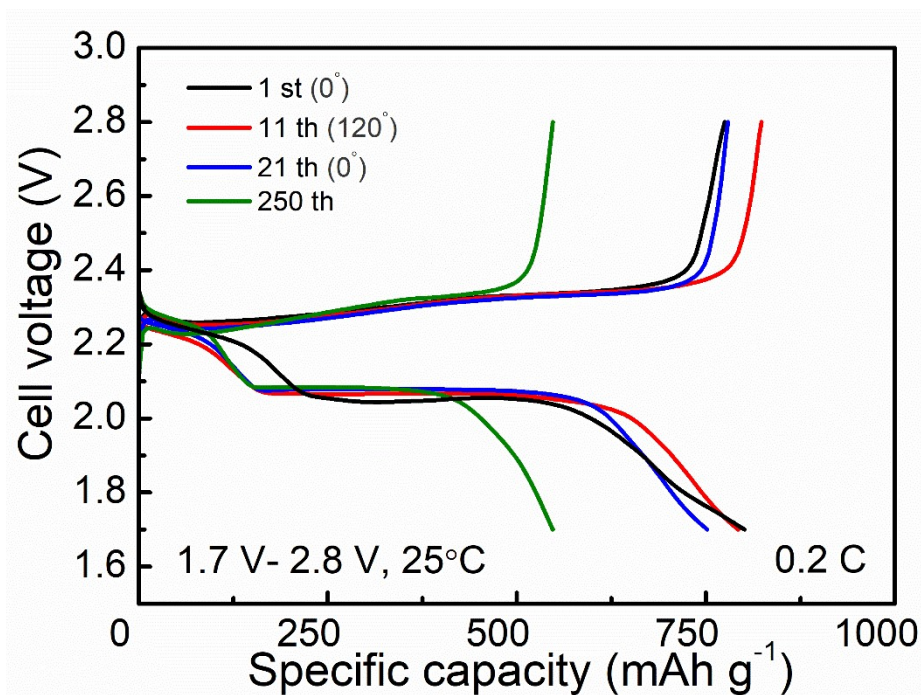


Fig. S7. Discharge-charge profiles of Co-TCN assembled Li-S pouch cell at 0.2 C.

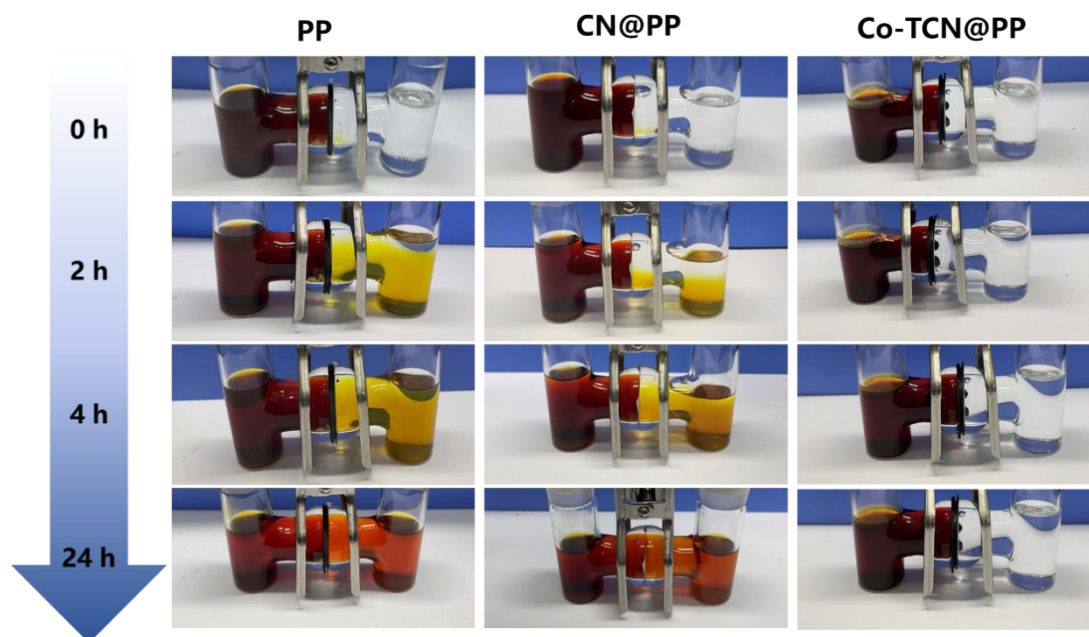


Fig.S8. Photographs of the glass cells with Li_2S_6 in DOL/DME solution and pure DOL/DME solvent in the left and right chambers, respectively, separated by pp, CN@PP or Co-TCN@PP.

Table S1. The setup parameter of coin cells and pouch cells.

Coin cell test information	
Active material	Cathode: Sulful (1675 mAh/g)
Additive	Ketjenblack
Binder	PVDF
Electrolyte	1M LiTFSi and 1wt% LiNO ₃ in DOL and DME (v:v,1:1)
Current collector	Al
Separator	Co-TCN@PP
Pouch cell test information	
	Anode: Li
	Cathode: S/KB
Comments	We tested both pouch cells and coin cells. The S loading was controlled to 1 mg/cm ² .The highest can reach to 5.6 mg/cm ² . The area of the electrode was 1.3 cm ² for coin cell and 4 cm ² for pouch cell.
Measurements and Calculations	
1C = 1675 mA/g Capacity = current density*time (h)	
Initial capacity (mAh/g)	863
Capacity achieved (mAh/g)	621
@rate	2C
Number of cycles tested	400
Capacity retention	71%
Testing temperature	30°C