

Supporting Information

Synthesis of P doped NiS as Electrode Material for Supercapacitors with Enhanced Rate Capability and Cycling Stability

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Table. S1. The lattice constant and unit cell volume of the prepared material.

	a (Å)	b (Å)	c (Å)	Volume _{cell} (Å ³)
075-0613 (NiS)	3.42	3.42	5.3	53.7
NiS	3.4194	3.4194	5.3028	53.70
P-NiS	3.4253	3.4253	5.3268	54.12

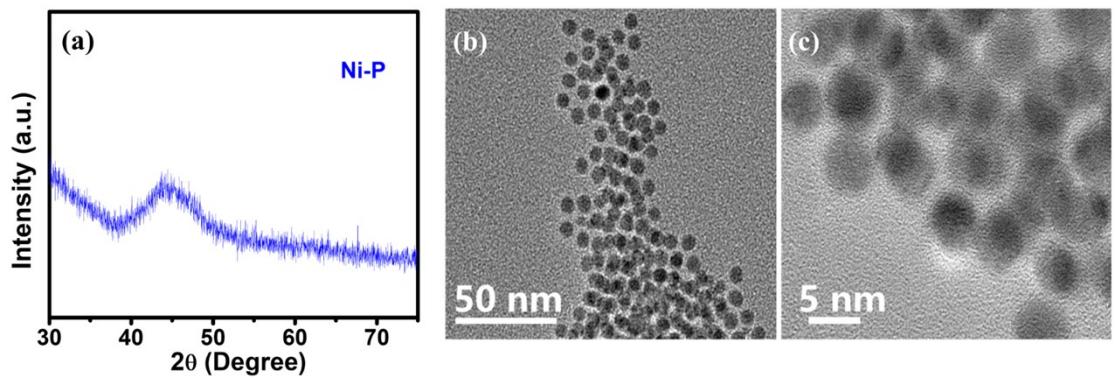


Fig. S1. (a) XRD pattern of Ni-P seeds; (b, c) TEM and HRTEM images of P-Ni seeds.

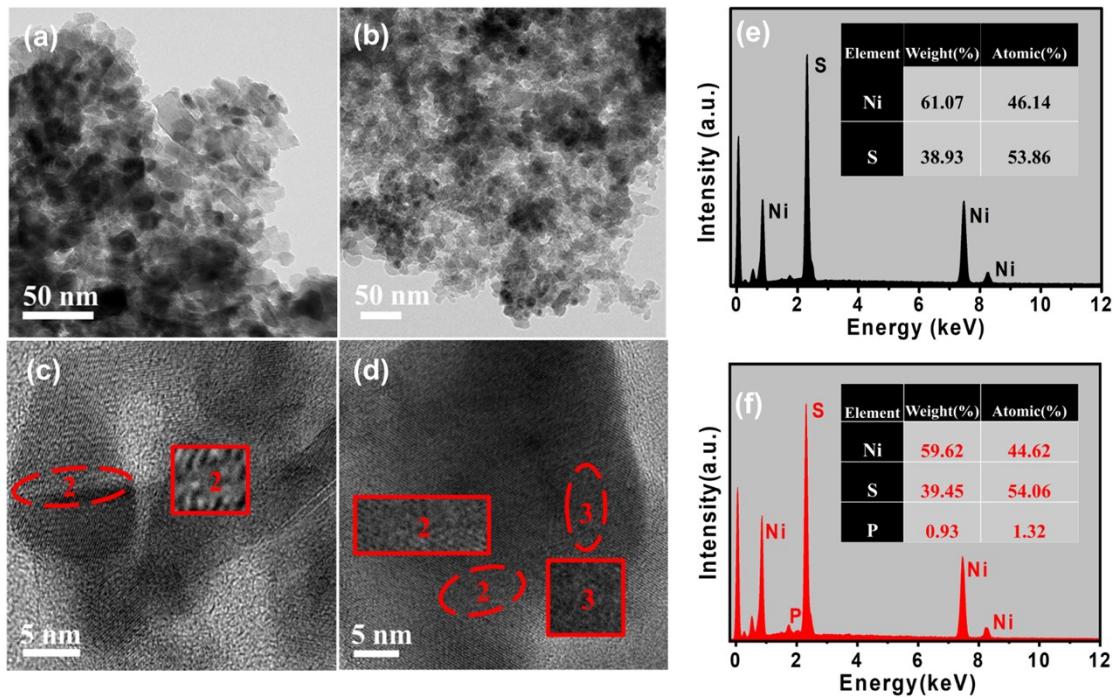


Fig. S2. (a) TEM image of NiS; (b) TEM image of P-NiS; (c, d) HRTEM images of P-NiS; (e) EDS spectrum of NiS; (f) EDS spectrum of P-NiS.

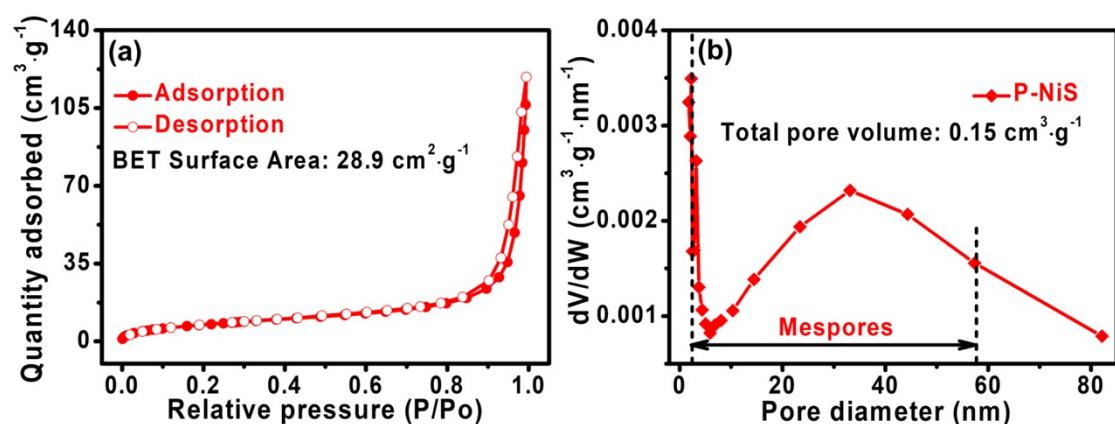


Fig. S3. (a) Nitrogen adsorption-desorption isotherms; (b) pore size distribution curves of P-NiS.

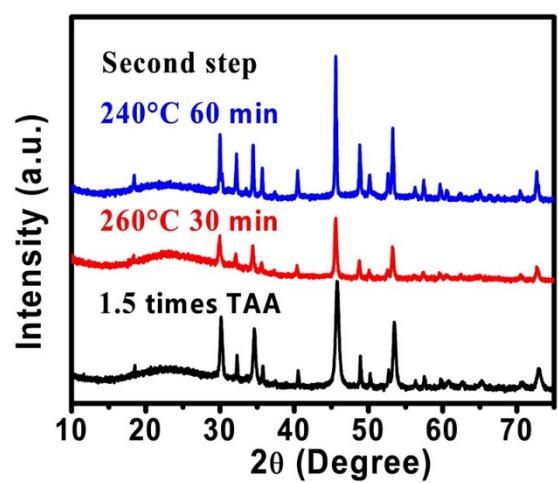


Fig. S4. XRD patterns of products obtained under different reaction conditions.

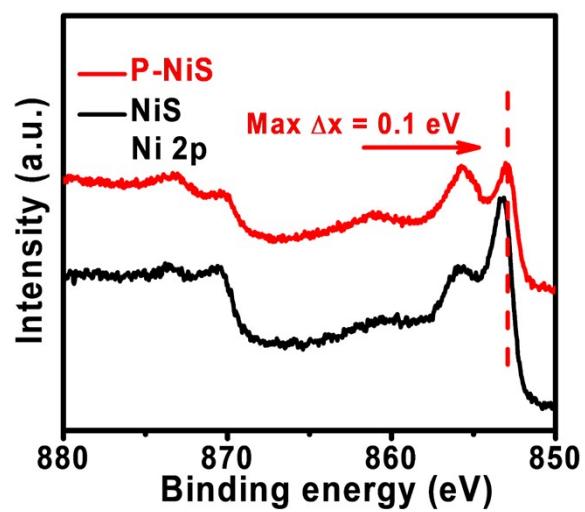


Fig. S5. High-resolution XPS of Ni 2p.

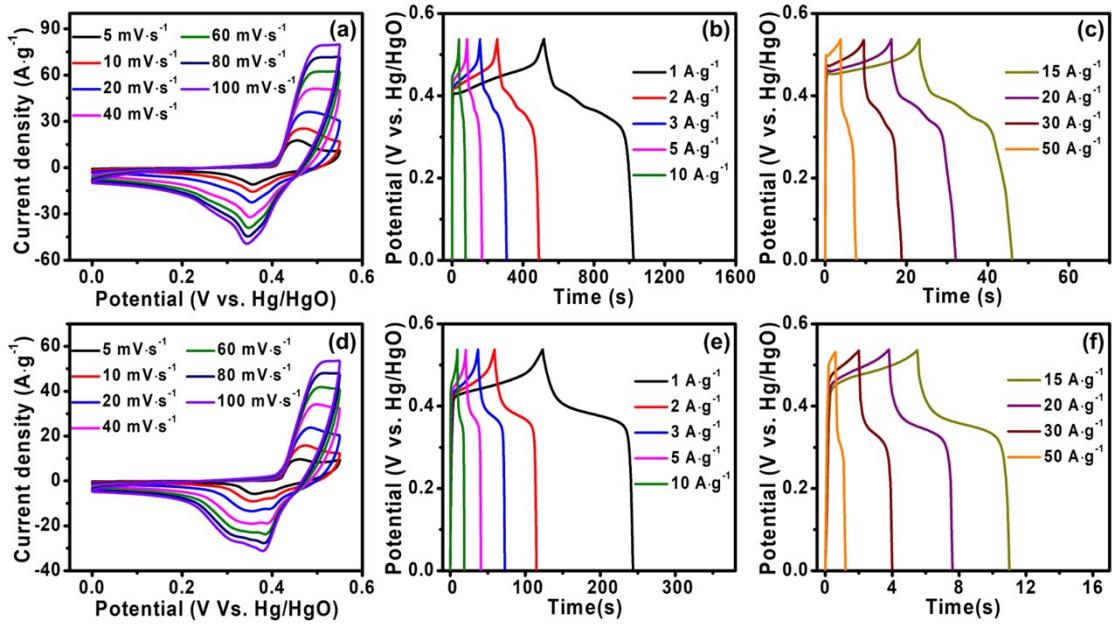


Fig. S6. (a) CV curves at different scan rates and (b, c) GCD curves at different densities of NiS; (d) CV curves at different scan rates and (e, f) GCD curves at different densities of Ni-P.

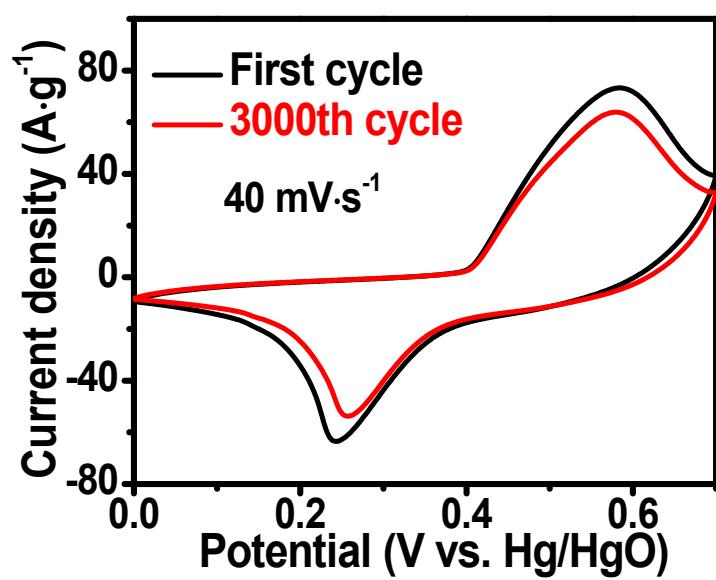


Fig. S7. CV comparison before and after stability test of P-NiS.

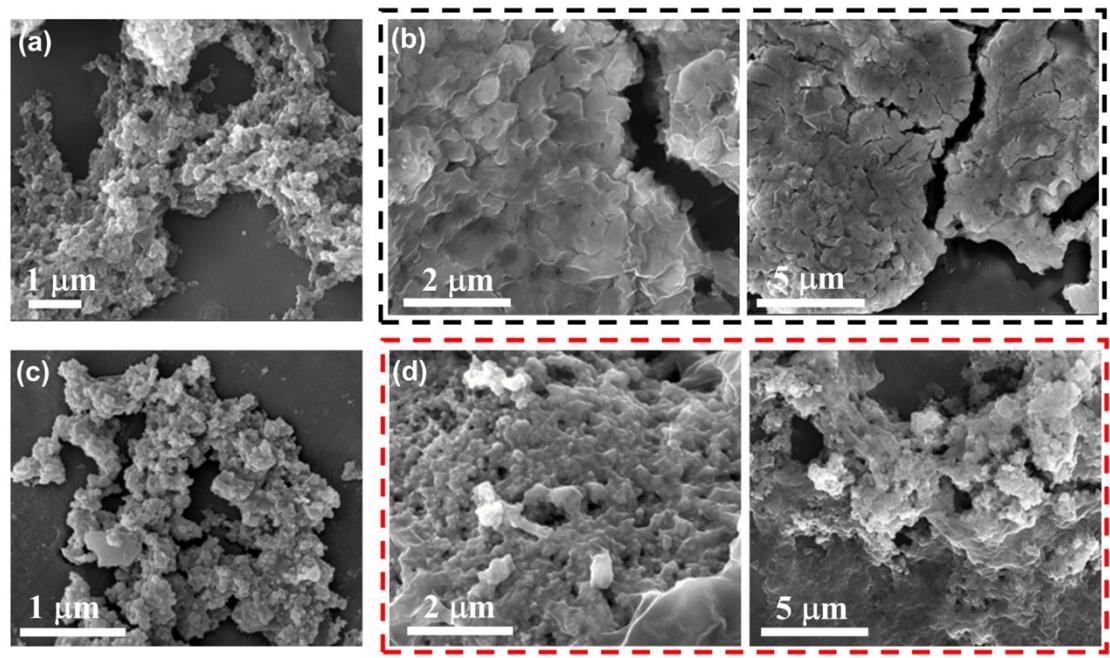


Fig. S8. (a, b) SEM images of the NiS electrode before and after cycle stability test; (c, d) SEM images of the P-NiS electrode before and after cycle stability test.

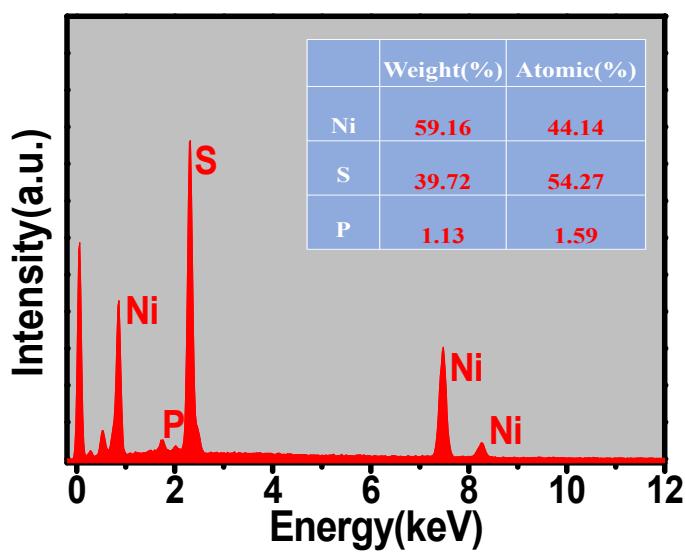


Fig. S9. EDS image of the P-NiS electrode after cycle stability test.

Table. S2. Comparison of electrochemical performance for NiS based electrode.

Materials	Electrolyte	Max. SC	Rate Performance	Cycling results	Refs.
NiS	2 MKOH	927 F·g ⁻¹ /4.08 A·g ⁻¹	62.8%; 583 F g ⁻¹ /10.2 A g ⁻¹	74.1%; 3000cycles/4.2 A g ⁻¹	[1]
NiS	2 M KOH	857.76 F·g ⁻¹ /2 A·g ⁻¹	59.8%; 512.96 F·g ⁻¹ /5 A g ⁻¹	40.7%; 1000cycles/2 A g ⁻¹	[2]
NiS/CRs	2 M KOH	514 F·g ⁻¹ /1 A·g ⁻¹	56.4%; 290 F g ⁻¹ /5 A g ⁻¹	49.7%; 2000cycles/5 A g ⁻¹	[3]
NiS/rGO	2 M KOH	905.3 F·g ⁻¹ /0.5 A·g ⁻¹	63.95%; 579 F g ⁻¹ /5 A g ⁻¹	90.9%; 2000cycles/4 A g ⁻¹	[4]
NiS/GNS	6 M KOH	775 F·g ⁻¹ /0.5 A·g ⁻¹	64.9%; 503 F g ⁻¹ /5 A g ⁻¹	88.1%; 1000cycles/2 A g ⁻¹	[5]
NiS	3 M KOH	717.3 F·g ⁻¹ /0.6 A·g ⁻¹	17.5%; 125.4 F g ⁻¹ /6 A g ⁻¹	98.5%; 1000cycles/1.2 A g ⁻¹	[6]
NiS	6 M KOH	845 F·g ⁻¹ /1 A·g ⁻¹	44%; 375 F g ⁻¹ /10 A g ⁻¹	81.6%; 1000cycles/1 A g ⁻¹	[7]
NiS	3 M KOH	1122.7 F·g ⁻¹ /1 A·g ⁻¹	28%; 323.5 F g ⁻¹ /30 A g ⁻¹	97.8%; 1000cycles/10 A g ⁻¹	[8]
NiS/NHCS	2 M KOH	1150 F·g ⁻¹ /1 A·g ⁻¹	52.2%; 600.3F g ⁻¹ /20 A g ⁻¹	76%; 4000cycles/5 A g ⁻¹	[9]
NiS	3 M KOH	1315.4 F·g ⁻¹ /1 A·g ⁻¹	24.2%; 317.8 F g ⁻¹ /30 A g ⁻¹	89.2%; 5000cycles/10 A g ⁻¹	[10]
NiS	2 M KOH	515.98 C·g ⁻¹ /1 A·g ⁻¹	37.6%; 194.01 C·g ⁻¹ /50 A·g ⁻¹	76.7%; 3000cycles/30 A·g ⁻¹	This
P-NiS		727.79 C·g ⁻¹ /1 A·g ⁻¹	50.6%; 368.26 C·g ⁻¹ /50A·g ⁻¹	93.9%; 3000cycles/30 A·g ⁻¹	Work

References

- [1] B.T. Zhu, Z. Wang, S. Ding, J.S. Chen and X.W. Lou, *RSC Adv.*, 2011, **1**, 397.
- [2] J. Yang, X. Duan, Q. Qin and W. Zheng, *J. Mater. Chem. A*, 2013, **1**, 7880.
- [3] C. Sun, M. Ma, J. Yang, Y. Zhang, P. Chen, W. Huang and X. Dong, *Sci. Rep.*, 2014, **4**, 7054.
- [4] J. Yang, X. Duan, W. Guo, D. Li, H. Zhang and W. Zheng, *Nano Energy*, 2014, **5**, 74.
- [5] Y. Li, K. Ye, K. Cheng, J. Yin, D. Cao and G. Wang, *J. Power Sources*, 2015, **274**, 943.
- [6] C. Wei, C. Cheng, Y. Cheng, Y. Wang, Y. Xu, W. Du and H. Pang, *Dalton Trans.*, 2015, **44**, 17278.
- [7] L. Peng, X. Ji, H. Wan, Y. Ruan, K. Xu, C. Chen, L. Miao and J. Jiang, *Electrochim. Acta*, 2015, **182**, 361.
- [8] B. Guan, Y. Li, B. Yin, K. Liu, D. Wang, H. Zhang and C. Cheng, *Chem. Eng. J.*, 2017, **308**, 1165.
- [9] T. Liu, C. Jiang, B. Cheng, W. You and J. Yu, *J. Mater. Chem. A*, 2017, **5**, 21257.
- [10] J. Zhao, B. Guan, B. Hu, Z. Xu, D. Wang and H. Zhang, *Electrochim. Acta*, 2017, **230**, 428.