## A Novel CoNi<sub>7</sub>O<sub>8</sub>/MnO<sub>2</sub> Nanocomposite Supported on Ni Foam as Peroxymonosulfate Activator for Highly Efficient Singlet Oxygen Mediated Removal of Methylene Blue

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Table S1 The elemental composition in CoNi<sub>7</sub>O<sub>8</sub>/MnO<sub>2</sub>/NF obtained by SEM-EDS analysis

Element	Mass content (%)			
Ni	68.2			
Mn	7.7			
Co	7.0			
O	17.0			

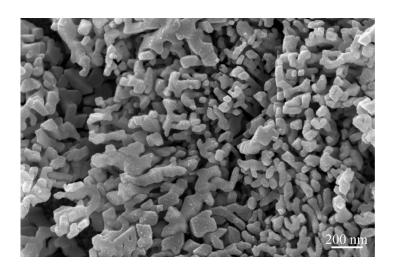


Fig. S1 SEM image of CoNi<sub>7</sub>O<sub>8</sub>/NF

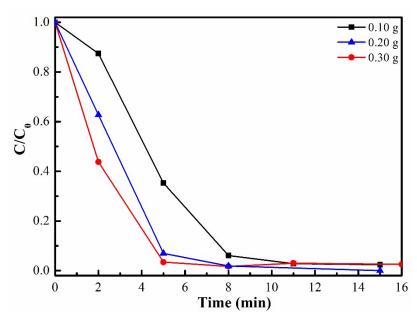


Fig. S2 Effect of  $CoCl_2 \cdot 6H_2O$  content on MB removal by  $CoNi_7O_8/MnO_2/NF$  (Reaction conditions: [MB] $_0 = 20$  mg/L, [PMS] $_0 = 0.5$  g/L, initial pH 5.9 and T = 30 °C)

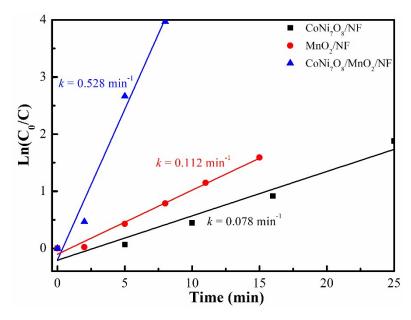


Fig. S3 The reaction kinetic curves of MB removal by different catalytic systems

Table S2 The comparison of catalytic performances of CoNi<sub>7</sub>O<sub>8</sub>/MnO<sub>2</sub>/NF with the previously reported catalysts

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$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	Catalysts			ьU		k	Pof
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	Catalysts	•		рп	•	(min <sup>-1</sup> )	Kel.
CoNi <sub>7</sub> O <sub>8</sub> /MnO <sub>2</sub> /NF       0.5       20       5.9       (15 min)       0.528 work         Cu@Co-MOFs       0.614       64       5.0 $\frac{100}{(30 \text{ min})}$ 0.109       [1]         FexOy/N-GN/CNTs       1.228       100       7.62 $\frac{100}{(20 \text{ min})}$ 0.33444       [2]         Mn <sub>3</sub> O <sub>4</sub> 1.88       62       4.0 $\frac{86.71}{(20 \text{ min})}$ 0.005       [3]         Ag-La <sub>0.8</sub> Ca <sub>0.2</sub> Fe <sub>0.9</sub> 4O <sub>3-δ</sub> (Ag-LCF)       0.368       10       /       90 (75 min)       /       [4]         CoFe <sub>2</sub> O <sub>4</sub> (Ag-CO)       0.491       200       7.0       93.3%       /       [6]		(g/L)	(mg/L)		(%)		
Cu@Co-MOFs $0.614$ $64$ $5.0$ $(15 \text{ min})$ ( $30 \text{ min})$ work         FexOy/N-GN/CNTs $1.228$ $100$ $7.62$ $100$ ( $20 \text{ min}$ ) $0.33444$ $[2]$ Mn <sub>3</sub> O <sub>4</sub> $1.88$ $62$ $4.0$ $86.71$ ( $20 \text{ min}$ ) $0.005$ $[3]$ Ag-La <sub>0.8</sub> Ca <sub>0.2</sub> Fe <sub>0.9</sub> 4O <sub>3-δ</sub> (Ag-LCF) $0.368$ $10$ $/$ $90$ ( $75 \text{ min}$ ) $/$ $[4]$ CoFe <sub>2</sub> O <sub>4</sub> @GO $0.491$ $200$ $7.0$ $93.3\%$ ( $93.3\%$ (	CoNi <sub>7</sub> O <sub>8</sub> /MnO <sub>2</sub> /NF	0.5	20	5.9	100	0.528	This
Cu@Co-MOFs $0.614$ $64$ $5.0$ $(30 \text{ min})$ $0.109$ [1]         FexOy/N-GN/CNTs $1.228$ $100$ $7.62$ $100$ $0.33444$ [2]         Mn <sub>3</sub> O <sub>4</sub> $1.88$ $62$ $4.0$ $86.71$ ( $20 \text{ min}$ ) $0.005$ [3]         Ag-La <sub>0.8</sub> Ca <sub>0.2</sub> Fe <sub>0.9</sub> 4O <sub>3-\delta</sub> (Ag-LCF) $0.368$ $10$ $/$ $90$ ( $75 \text{ min}$ ) $/$ [4]         CoFe <sub>2</sub> O <sub>4</sub> $0.3$ $20$ $6.3$ $97.9\%$ ( $75 \text{ min}$ ) $0.114$ [5]         CuFe <sub>2</sub> O <sub>4</sub> @GO $0.491$ $200$ $7.0$ $93.3\%$ $/$ [6]					(15 min)		work
FexOy/N-GN/CNTs 1.228 100 7.62 $\frac{(30 \text{ min})}{100}$ 0.33444 [2]  Mn <sub>3</sub> O <sub>4</sub> 1.88 62 4.0 $\frac{86.71}{(20 \text{ min})}$ 0.005 [3]  Ag-La <sub>0.8</sub> Ca <sub>0.2</sub> Fe <sub>0.9</sub> 4O <sub>3-δ</sub> 0.368 10 / $\frac{90}{(75 \text{ min})}$ 7.62 $\frac{90}{(75 \text{ min})}$ 6.114 [5]  CuFe <sub>2</sub> O <sub>4</sub> @GO 0.491 200 7.0 $\frac{93.3\%}{(75 \text{ min})}$ 6.114 [5]	Cu@Co-MOFs	0.614	64	5.0	100	0.109	[1]
FexOy/N-GN/CNTs 1.228 100 7.62 (20 min) 0.33444 [2]  Mn <sub>3</sub> O <sub>4</sub> 1.88 62 4.0 86.71 (20 min) 0.005 [3]  Ag-La <sub>0.8</sub> Ca <sub>0.2</sub> Fe <sub>0.9</sub> 4O <sub>3-<math>\delta</math></sub> 0.368 10 / 90 (75 min)					(30 min)		
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	FexOy/N-GN/CNTs	1.228	100	7.62	100	0.33444	[2]
$\begin{array}{cccccccccccccccccccccccccccccccccccc$					(20 min)		
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$Mn_3O_4$	1.88	62	4.0	86.71	0.005	[3]
Ag-La <sub>0.8</sub> Ca <sub>0.2</sub> Fe <sub>0.9</sub> 4O <sub>3-δ</sub> 0.368 10 / $\frac{90}{(75 \text{ min})}$ [4] $CoFe2O4 0.3 20 6.3 \frac{97.9\%}{(75 \text{ min})} 0.114 [5]$ $CuFe2O4@GO 0.491 200 7.0 \frac{93.3\%}{(75 \text{ min})}$					(20 min)		
(Ag-LCF) $0.368$ $10$ $7$ $7$ $14$ $15$ $10$ $10$ $10$ $10$ $10$ $10$ $10$ $10$	Ag-Lao «Ca0.2Fe0.94O <sub>2.8</sub>				, ,		
CoFe <sub>2</sub> O <sub>4</sub> 0.3 20 6.3 $\frac{97.9\%}{(75 \text{ min})}$ 0.114 [5]		0.368	10	/		/	[4]
$CoFe_2O_4$ 0.3 20 6.3 (75 min) 0.114 [5] $CuFe_2O_4@GO$ 0.491 200 7.0 93.3%	(rig Let)				` '		
(75 min) 93.3% CuFe <sub>2</sub> O <sub>4</sub> @GO 0.491 200 7.0	$CoFe_2O_4$	0.3	20	6.3		0.114	[5]
$CuFe_2O_4@GO = 0.491 = 200 = 7.0$	- 1				(75 min)		
Curc <sub>2</sub> O <sub>4</sub> (u)OO 0.491 200 7.0 (20iv) / [0]	CuFe <sub>2</sub> O <sub>4</sub> @GO	0.491	200	7.0	93.3%	/	[6]
(30 min)					(30 min)		

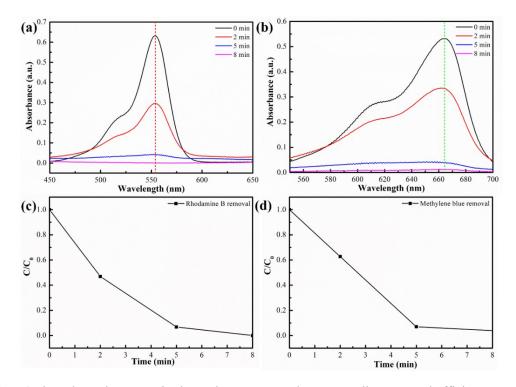


Fig. S4 Time-dependent UV-vis absorption spectra and corresponding removal efficiency curves of RhB (a, c) and MB (b, d)

(Reaction conditions: [MB] $_0$  and [RhB] $_0$  = 20 mg/L, [PMS] $_0$  = 0.5 g/L, initial pH 5.9 and T = 30



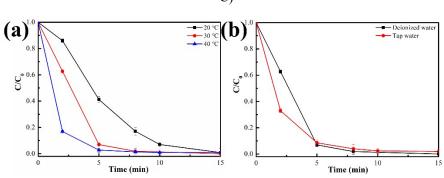


Fig. S5 Influences of temperature and water type on MB removal

(Reaction condition: [MB] $_0$  = 20 mg/L, [PMS] $_0$  = 0.5 g/L, initial pH 5.9 and T = 30 °C)

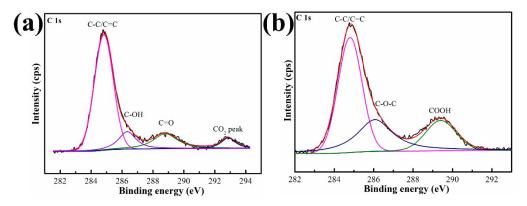


Fig. S6 High-resolution C 1s XPS spectra of fresh (a) and used (b) CoNi<sub>7</sub>O<sub>8</sub>/MnO<sub>2</sub>/NF

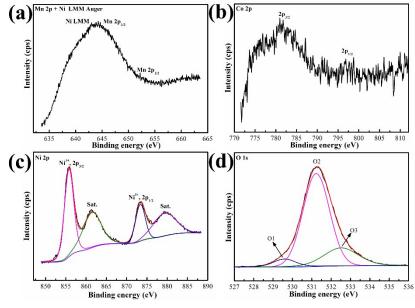


Fig. S7 High-resolution Mn 2p (a), Co 2p (b), Ni 2p (c) and O 1s (d) XPS spectra of used  $CoNi_7O_8/MnO_2/NF$ 

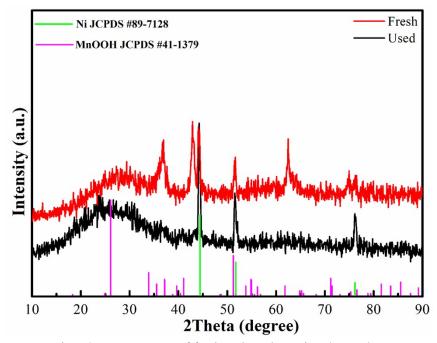


Fig. S8 XRD patterns of fresh and used  $CoNi_7O_8/MnO_2/NF$ 

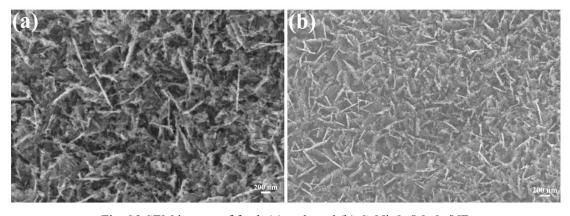


Fig. S9 SEM images of fresh (a) and used (b)  $CoNi_7O_8/MnO_2/NF$ 

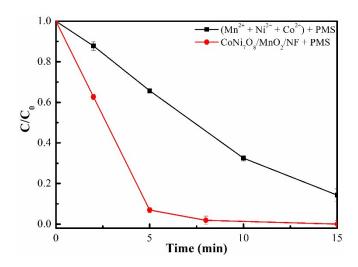


Fig. S10 The impact of homogeneous catalysis on MB removal (Reaction conditions:  $[MB]^0 = 20 \text{ mg/L}$ ,  $[PMS]^0 = 0.5 \text{ g/L}$ , initial pH 5.9 and T = 30 °C)

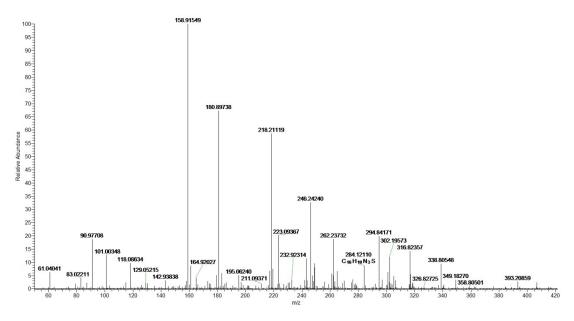


Fig. S11 LC-MS spectrum of MB degradation solution after reaction of 15min

## References

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