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Supporting Information

Visible-Light-Induced Tandem Difluoroalkylated Spirocyclization of

N-arylpropiolamides: Access to C3-Difluoroacetylated

Spiro[4,5]trienones

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1 General information

All chemicals were commercially available and used as received without further. Column chromatography was performed using 300-400 mesh silica. Nuclear magnetic resonance spectra were recorded on Bruker Avance 400 MHz spectrometer. ¹H NMR spectra are recorded in parts per million from tetramethylsilane. Data were reported as follows: chemical shift, multiplicity (s = singlet, d = doublet, t = triplet, m = multiplet and br = broad), coupling constant in Hz and integration. ¹³C NMR spectra were recorded in parts per million from tetramethylsilane. ¹⁹F NMR spectra were recorded in parts per million from tetramethylsilane. ¹⁹F NMR spectra were recorded in parts per million from tetramethylsilane. ¹⁹F NMR spectra were recorded in parts per million from tetramethylsilane. ¹⁹F NMR spectra were recorded in parts per million from tetramethylsilane. ¹⁹F NMR spectra were recorded in parts per million with fluorobenzene as external standard. High resolution mass spectra (HR MS) were obtained on Thermo Scientific LTQ Orbitrap XL instrument using the ESI technique. IR spectra were recorded on WQF-510 Fourier transform infrared spectrophotometer. Melting points were measured on an XT4A microscopic apparatus uncorrected.

2 Screening the reaction conditions



Entry	the molar ratio of 1a and 2a	Yields (%) ^b
1	1:1	22
2	1:1.5	37
3	1:2	44
4	1:2.5	50
5	1:3	56
6	1:3.5	55

 Table S1 Screening the molar ratio of 1a and 2a^a

^a Reaction conditions: N-(4-methoxyphenyl)-N-methyl-3-phenylpropiolamide 1a (0.2 mmol, 53.0 mg), ethyl 2-

bromo-2,2-difluoroacetate **2a**, *fac*-[Ir(ppy)] (0.002 mmol, 1.31 mg), K₃PO₄ (0.3 mmol, 63.6 mg) and DMF (2.0 mL), 5 W blue LEDs at room temperature under N₂ atmosphere for 4.5 h. ^{*b*} Isolated yield.

3 Luminescence quenching study

Emission intensities were recorded using a F-7000 FL Spectrophotometer. First, the emission intensity of fluorescein solutions was observed at 369 nm. The solutions were irradiated at 510 nm (Maximum absorption wavelength of fluorescein) and fluorescence was measured from 450 nm to 650 nm. In a typical experiment, the emission spectrum of a 1×10^{-4} M solution of fluorescein with different concentration of **2a** in degassed anhydrous CH₂Cl₂ and the linear relationship between I₀/I and the increasing concentration of **2a** from 0 M to 5.0×10^{-2} M.



Figure 1 Luminescence quenching study

4 Copies of spectra of products



Fig. 2 ¹H NMR spectrum of compound 3a



Fig. 3 ¹³C NMR spectrum of compound 3a



Fig. 4 ¹⁹F NMR spectrum of compound 3a



Fig. 5 ¹H NMR spectrum of compound 3b



Fig. 6 ¹³C NMR spectrum of compound 3b



Fig. 7¹⁹F NMR spectrum of compound 3b



Fig. 8 ¹H NMR spectrum of compound 3c



Fig. 9 $^{\rm 13}{\rm C}$ NMR spectrum of compound 3c



Fig. 10¹⁹F NMR spectrum of compound 3c



Fig. 11 ¹H NMR spectrum of compound 3d



Fig. 12 ¹³C NMR spectrum of compound 3d



Fig. 13 ¹⁹F NMR spectrum of compound 3d



Fig. 14 ¹H NMR spectrum of compound 3e



Fig. 15 ¹³C NMR spectrum of compound 3e



Fig. 16 ¹⁹F NMR spectrum of compound 3e



Fig. 17 ¹H NMR spectrum of compound 3f



Fig. 18 ¹³C NMR spectrum of compound 3f



Fig. 19¹⁹F NMR spectrum of compound 3f



Fig. 21 ¹³C NMR spectrum of compound 3g



Fig. 22 ¹⁹F NMR spectrum of compound 3g



Fig. 23 ¹H NMR spectrum of compound 3h



Fig. 24 ¹³C NMR spectrum of compound 3h



Fig. 25 ¹⁹F NMR spectrum of compound 3h



Fig. 26 ¹H NMR spectrum of compound 3i



Fig. 27 ¹³C NMR spectrum of compound 3i



Fig. 28 ¹⁹F NMR spectrum of compound 3i



Fig. 29 ¹H NMR spectrum of compound 3j



Fig. 30 ¹³C NMR spectrum of compound 3j



Fig. 31 ¹⁹F NMR spectrum of compound 3j



Fig. 32 ¹H NMR spectrum of compound 3k



Fig. 33 ¹³C NMR spectrum of compound 3k



Fig. 34 ¹⁹F NMR spectrum of compound 3k



Fig. 35 ¹H NMR spectrum of compound 3I



Fig. 36 ¹³C NMR spectrum of compound 3I



Fig. 37 ¹⁹F NMR spectrum of compound 3I



Fig. 38 ¹H NMR spectrum of compound 3m



Fig. 39 ¹³C NMR spectrum of compound 3m



Fig. 40 ¹⁹F NMR spectrum of compound 3m



Fig. 41 ¹H NMR spectrum of compound 3n



Fig. 42 ¹³C NMR spectrum of compound 3n



Fig. 43 ¹⁹F NMR spectrum of compound 3n



Fig. 44 ¹H NMR spectrum of compound 30



Fig. 45 ¹³C NMR spectrum of compound 30



Fig. 46 ¹⁹F NMR spectrum of compound 30



Fig. 47 ¹H NMR spectrum of compound 3p



Fig. 48 ¹³C NMR spectrum of compound 3p



Fig. 49 ¹⁹F NMR spectrum of compound 3p





Fig. 51 ¹³C NMR spectrum of compound 3q



Fig. 52 ¹⁹F NMR spectrum of compound 3q



Fig. 53 ¹H NMR spectrum of compound 3r



Fig. 54 ¹³C NMR spectrum of compound 3r



Fig. 55 ¹⁹F NMR spectrum of compound 3r



Fig. 56 ¹H NMR spectrum of compound 3s



Fig. 57 ¹³C NMR spectrum of compound 3s



Fig. 59 ¹H NMR spectrum of compound 3t



Fig. 60 ¹³C NMR spectrum of compound 3t



Fig. 61 ¹⁹F NMR spectrum of compound 3t



Fig. 63 ¹³C NMR spectrum of compound 3u



Fig. 64 ¹⁹F NMR spectrum of compound 3u



Fig. 65 ¹H NMR spectrum of compound 3v



Fig. 66 ¹³C NMR spectrum of compound 3v



Fig. 67 ¹⁹F NMR spectrum of compound 3v



Fig. 68 ¹H NMR spectrum of compound 3w



Fig. 69 ¹³C NMR spectrum of compound 3w



Fig. 70 $^{\rm 19}{\rm F}$ NMR spectrum of compound 3w



Fig. 71 ¹H NMR spectrum of compound 3x



Fig. 72 ¹³C NMR spectrum of compound 3x



Fig. 73 ¹⁹F NMR spectrum of compound 3x



Fig. 74 ¹H NMR spectrum of compound 3y



Fig. 75 ¹³C NMR spectrum of compound 3y



Fig. 76 ¹⁹F NMR spectrum of compound 3y



Fig. 77 ¹H NMR spectrum of compound 3z



Fig. 78 ¹³C NMR spectrum of compound 3z



Fig. 79¹⁹F NMR spectrum of compound 3z



Fig. 80 ¹H NMR spectrum of compound 3ab



Fig. 81 ¹³C NMR spectrum of compound 3ab



Fig. 82 ¹⁹F NMR spectrum of compound 3ab



Fig. 83 ¹H NMR spectrum of compound 3ac



Fig. 84 ¹³C NMR spectrum of compound 3ac



Fig. 85¹⁹F NMR spectrum of compound 3ac



Fig. 87 ¹³C NMR spectrum of compound 3ad



Fig. 88 ¹H NMR spectrum of compound 5a



Fig. 89¹³C NMR spectrum of compound 5a



Fig. 90 ¹⁹F NMR spectrum of compound 5a



Fig. 91 ¹H NMR spectrum of compound 5b



Fig. 92 ¹³C NMR spectrum of compound 5b



Fig. 93 ¹⁹F NMR spectrum of compound 5b



Fig. 94 ¹H NMR spectrum of compound 5c



Fig. 95 ¹³C NMR spectrum of compound 5c



Fig. 96 ¹⁹F NMR spectrum of compound 5c



Fig. 97 ¹H NMR spectrum of compound 5d



Fig. 98 ¹³C NMR spectrum of compound 5d



Fig. 99 ¹⁹F NMR spectrum of compound 5d



Fig. 100 ¹H NMR spectrum of compound 5e



Fig. 101 ¹³C NMR spectrum of compound 5e



Fig. 102 ¹⁹F NMR spectrum of compound 5e



Fig. 103 ¹H NMR spectrum of compound 7



Fig. 104 ¹³C NMR spectrum of compound 7



-70 -75 -80 -85 -90 -95 -100 -105 -110 -115 -120 -125 -130 -135 -140 -145 -150 -155 ppm

Fig. 105 ¹⁹F NMR spectrum of compound 7



Fig. 106 ¹H NMR spectrum of compound 9



Fig. 107 ¹³C NMR spectrum of compound 9











Fig. 110 ¹³C NMR spectrum of compound 10



Fig. 111 ¹⁹F NMR spectrum of compound 10

5¹⁹F NMR spectrum of TEMPO-CF₂CO₂Et adduct



Fig. 112 ¹⁹F NMR spectrum of TEMPO-CF₂CO₂Et adduct

6 HR MS of BHT-CF₂CO₂Et adduct



