

Supporting Information

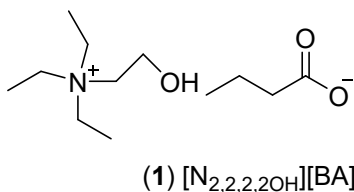
Novel hydroxyl functionalized ionic liquids as efficient catalysts for the conversion of CO₂ into cyclic carbonates under metal/halogen/cocatalyst/solvent-free conditions

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Characteristic data:

Ionic liquid 1:



C₁₂H₂₇NO₃(233). ¹H NMR (D₂O, 300MHz, RT): δ=3.91 (2H, s), 3.32 (8H, q), 2.08 (2H, t), 1.47 (2H, m), 1.21 (9H, t), 0.82 (3H, t); ¹³C NMR (75.5MHz, D₂O): δ=184.05, 57.45, 54.81, 53.39, 39.69, 19.46, 13.32, 6.74; IR (KBr): ν=1563 cm⁻¹ (C=O).

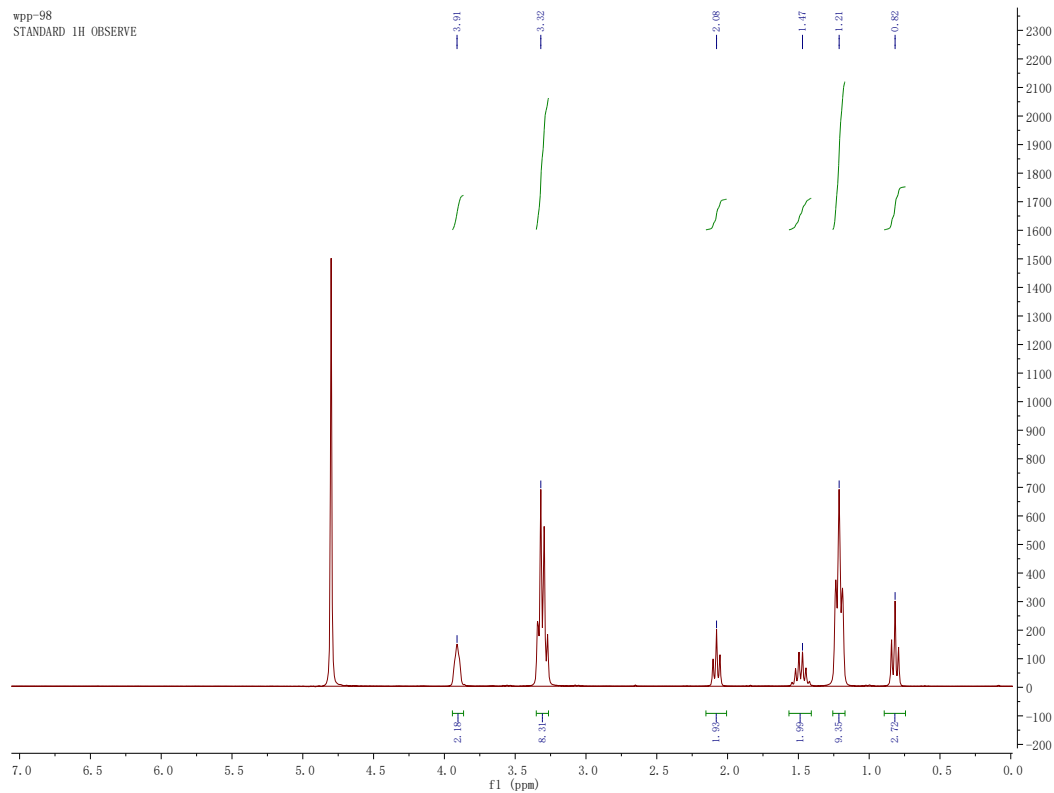


Figure S1-1. ¹H NMR spectrum of IL1

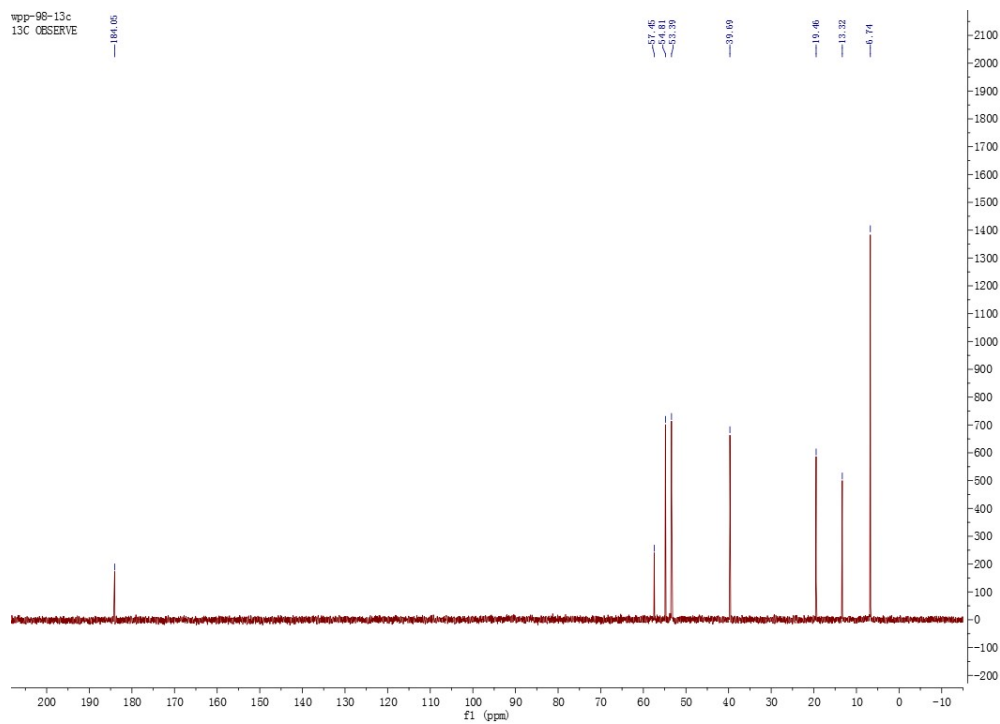


Figure S1-2. ¹³C NMR spectrum of IL1

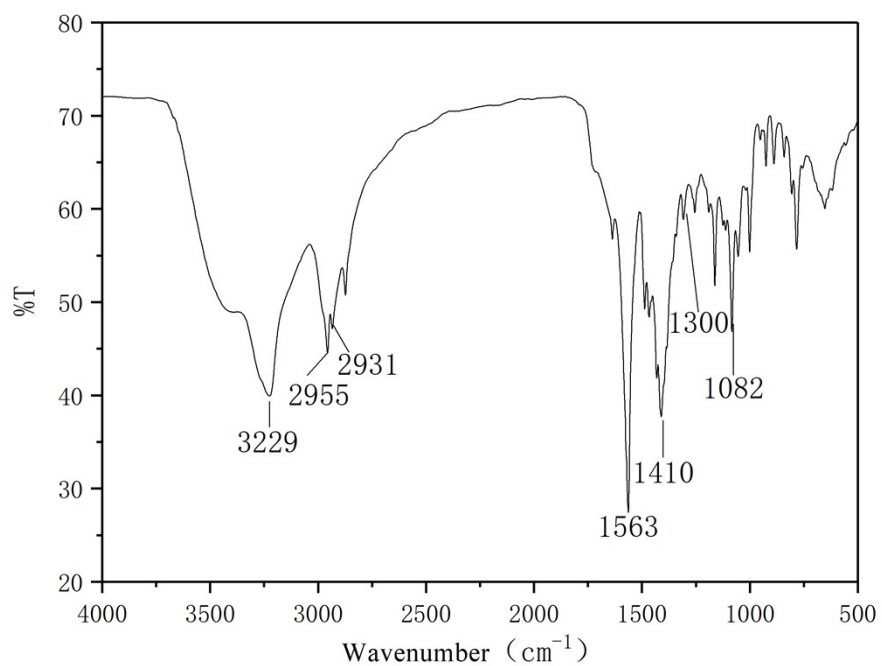
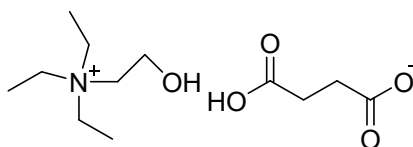


Figure S1-3. IR spectrum of IL 1

Ionic liquid 2



(2) [N_{2,2,2,2OH}][SA]

C₁₂H₂₅NO₅(263). ¹H NMR (D₂O, 300MHz, RT): δ=3.90 (2H, s), 3.33 (8H, q), 2.44 (4H, t), 1.22 (9H, t); ¹³C NMR (75.5MHz, D₂O): δ=180.08, 57.51, 54.87, 53.47, 32.03, 6.81; IR (KBr): ν=1578 cm⁻¹ (C=O).

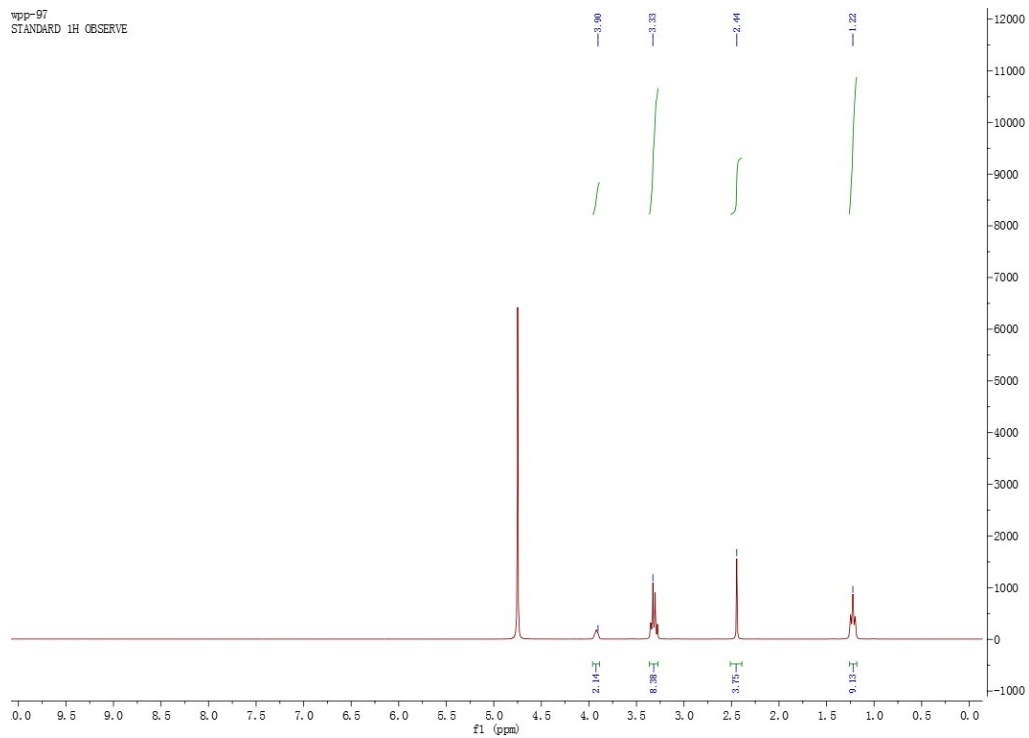


Figure S2-1. ^1H NMR spectrum of IL2

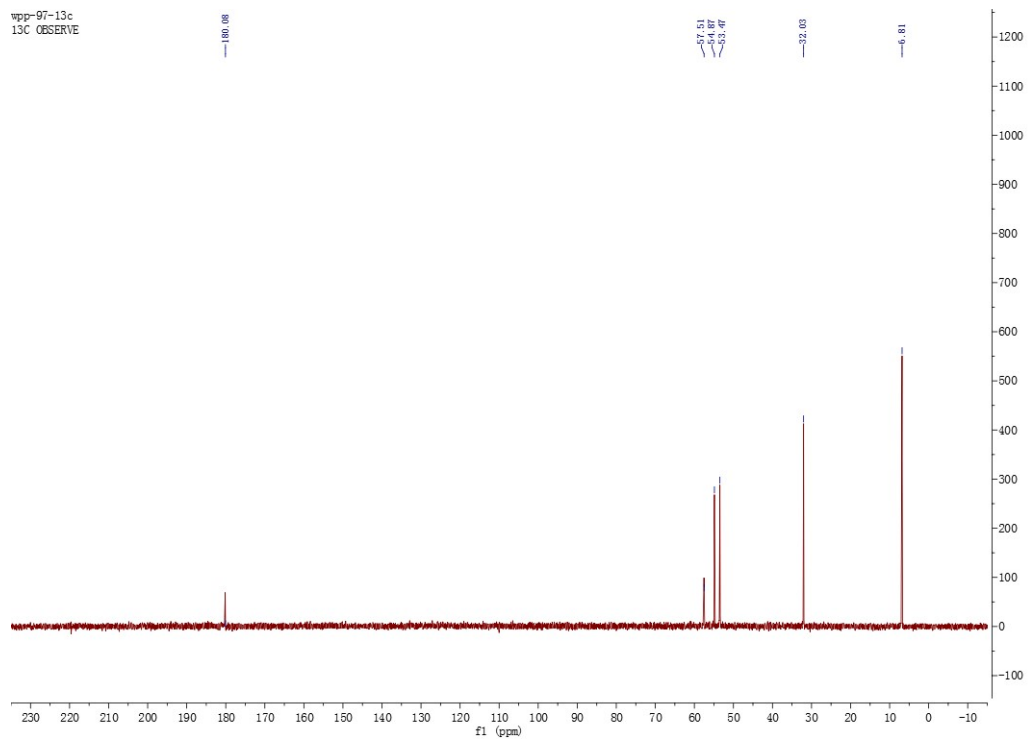


Figure S2-2. ^{13}C NMR spectrum of IL2

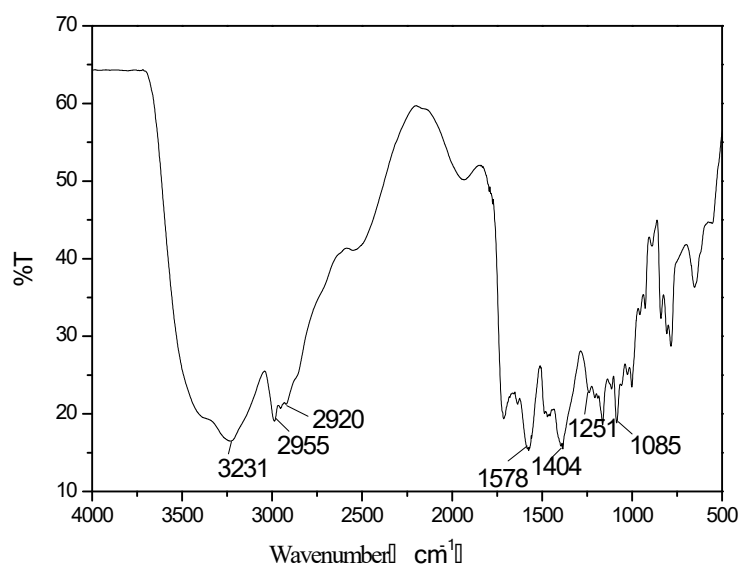
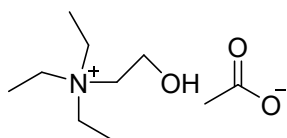


Figure S2-3. IR spectrum of IL 2

Ionic liquid 3



(3) [N_{2,2,2,2OH}][OAc]

C₁₀H₂₃NO₃(205). ¹H NMR (D₂O, 300MHz, RT): δ=3.90 (2H, s), 3.30 (8H, q), 1.83 (3H, s), 1.20 (9H, t); ¹³C NMR (75.5MHz, D₂O): δ=181.32, 62.62, 57.54, 54.87, 53.47, 23.38, 6.81; IR (KBr): ν=1581 cm⁻¹ (C=O).

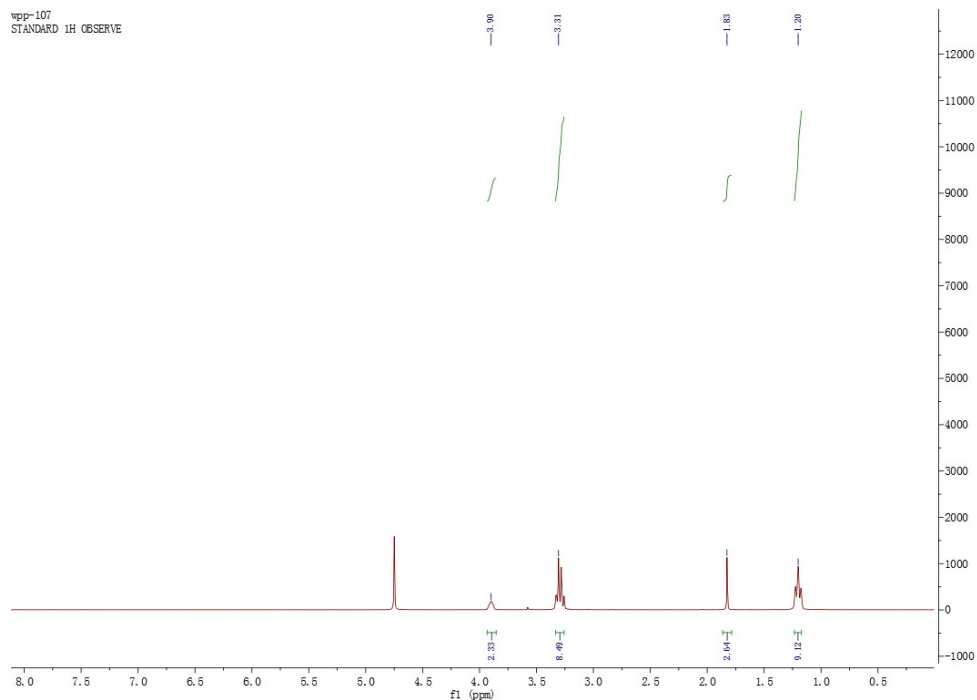


Figure S3-1. ¹H NMR spectrum of IL 3

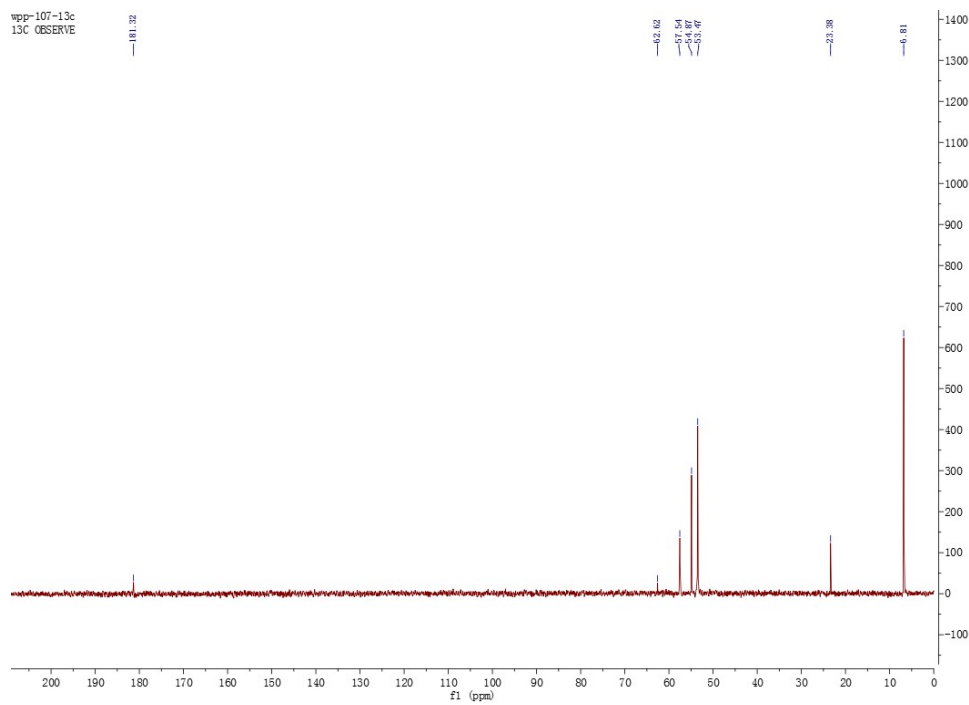


Figure S3-2. ¹³C NMR spectrum of IL 3

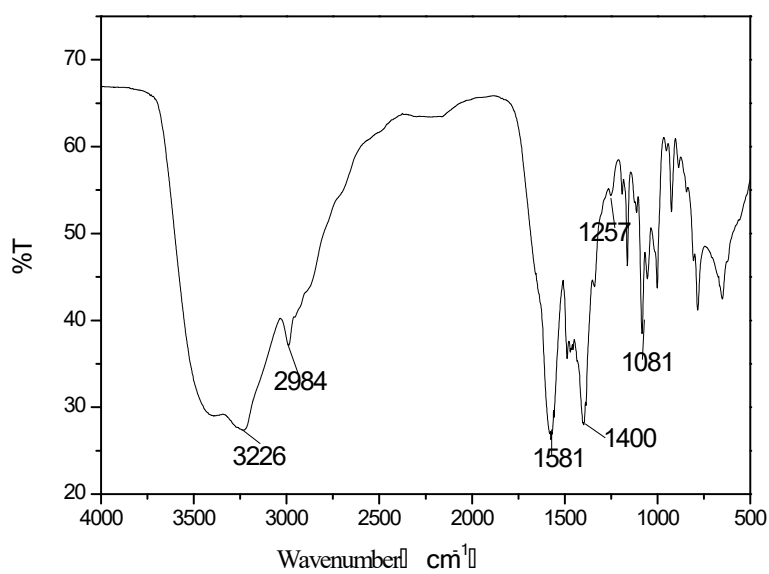
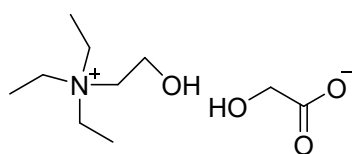


Figure S3-3. IR spectrum of IL 3

Ionic liquid 4



(4) [N_{2,2,2,2OH}][GA]

C₁₀H₂₃NO₄(221). ¹H NMR (D₂O, 300MHz, RT): δ=3.85 (4H, d), 3.31 (8H, q), 1.21 (9H, t); ¹³C NMR (75.5MHz, D₂O): δ=179.87, 61.30, 57.50, 54.87, 53.45, 6.79; IR (KBr): ν=1600cm⁻¹ (C=O).

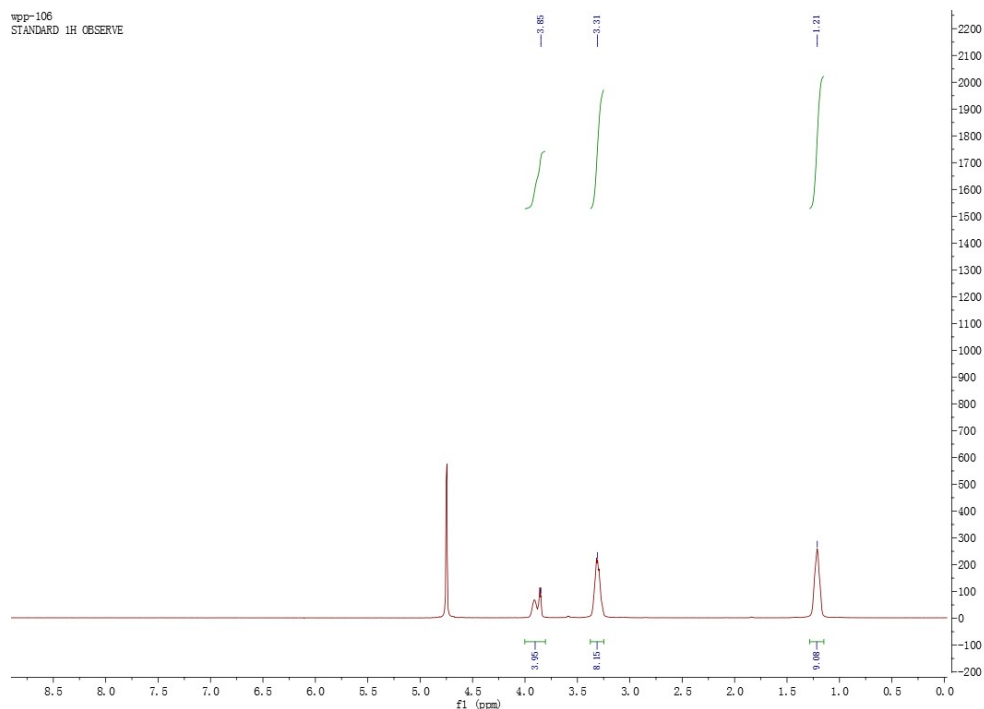


Figure S4-1. ^1H NMR spectrum of IL 4

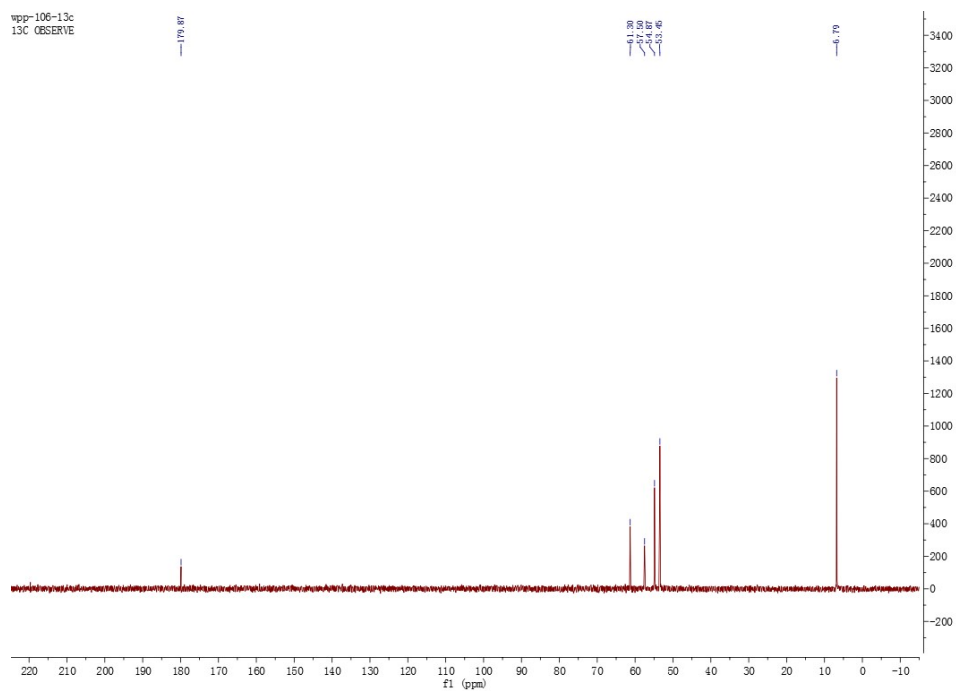


Figure S4-2. ^{13}C NMR spectrum of IL 4

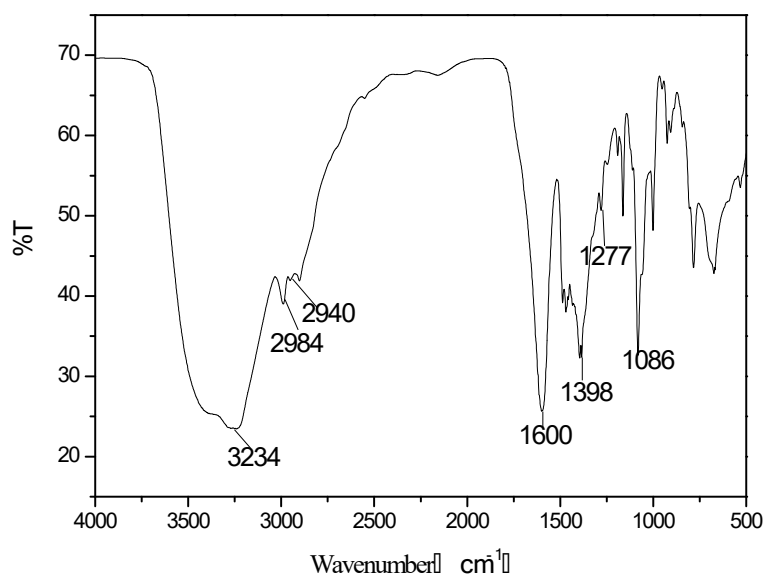
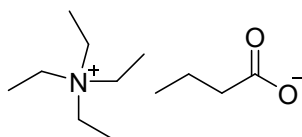


Figure S4-3. IR spectrum of IL 4

Ionic liquid 5



(5) $[N_{2,2,2,2}][BA]$

$C_{12}H_{27}NO_2$ (217). 1H NMR (D_2O , 300 MHz, RT): δ = 3.17 (8H, q), 2.08 (2H, m), 1.48 (2H, m), 1.18 (12H, m), 0.82 (3H, m); ^{13}C NMR (75 MHz, D_2O) δ = 183.59, 52.07, 39.69, 19.58, 13.55, 6.82; IR (KBr): ν = 1564 cm^{-1} (C=O).

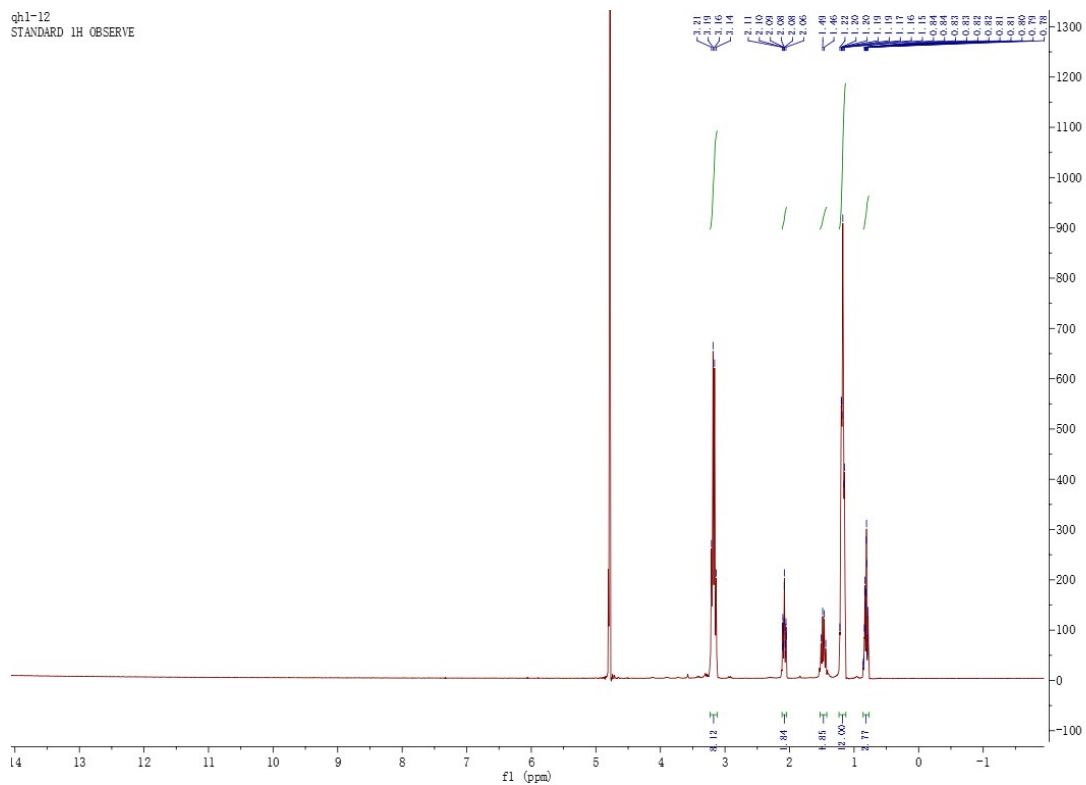


Figure S5-1. ^1H NMR spectrum of IL 5

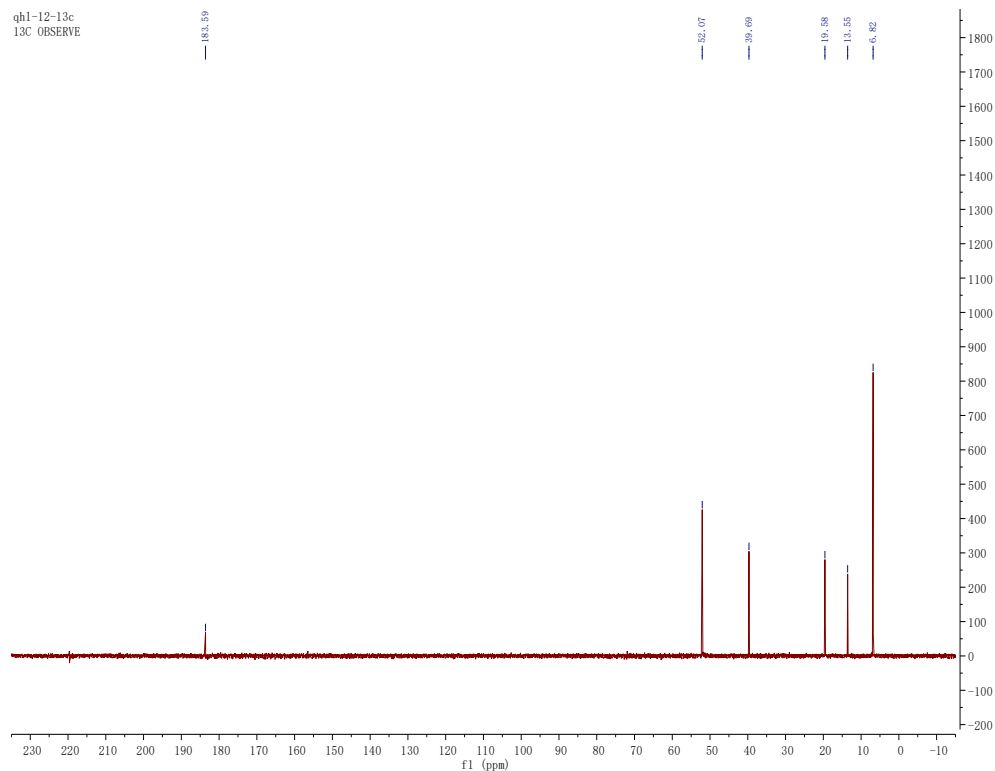


Figure S5-2. ^{13}C NMR spectrum of IL 5

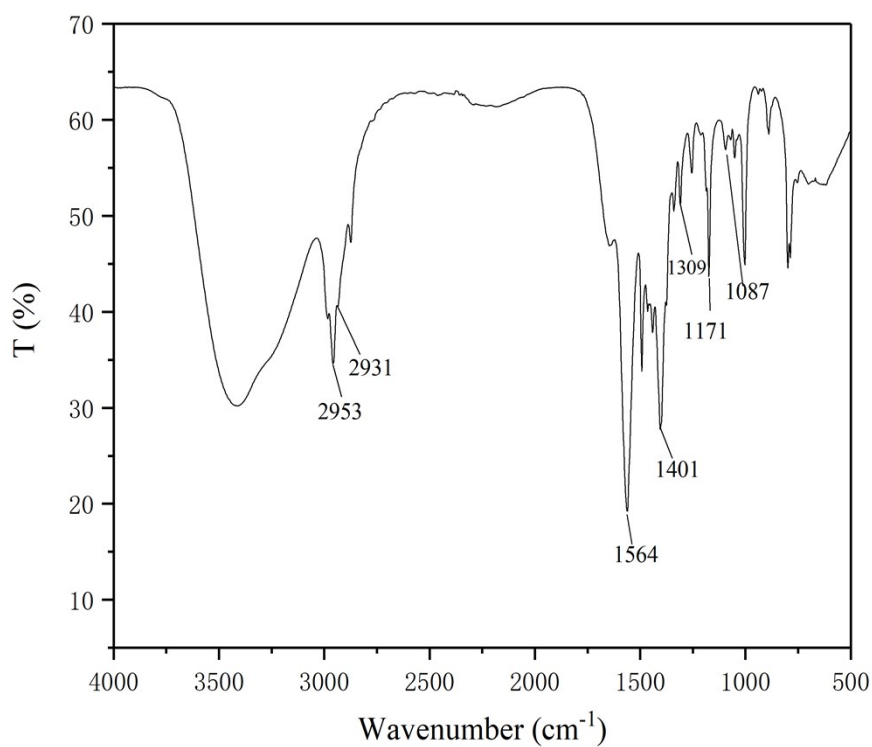


Figure S5-3. IR spectrum of IL 5

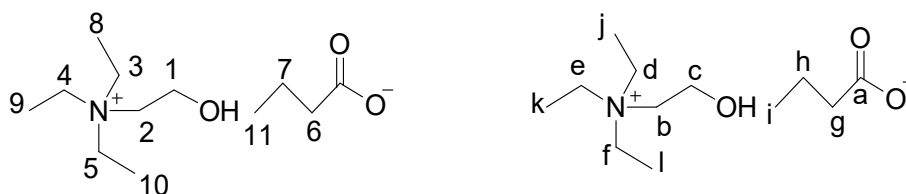


Figure S6. Distribution of ^1H NMR and ^{13}C NMR spectra of ionic liquid 1.

Table S1

^1H NMR (D_2O) of IL 1 (δ , ppm).

H(1)	H(2,3,4,5)	H(6)	H(7)	H(8,9,10)	H(11)
3.91	3.32	2.08	1.47	1.21	0.82
(2H,s)	(8H,q)	(2H,t)	(2H,m)	(9H,t)	(3H,t)

^{13}C NMR (D_2O) of IL 1 (δ , ppm).

C^{a}	C^{b}	C^{c}	C^{d}	C^{e}	C^{f}	$\text{C}^{\text{g}}, \text{C}^{\text{h}}, \text{C}^{\text{i}}$	$\text{C}^{\text{j}}, \text{C}^{\text{k}}, \text{C}^{\text{l}}$
184.05	57.45	54.81	53.39	39.69	19.46	13.32	6.74

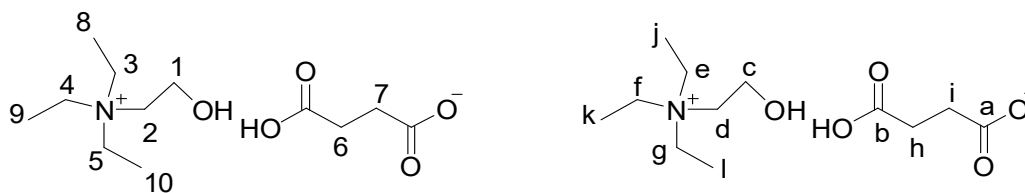


Figure S7. Distribution of ^1H NMR and ^{13}C NMR spectra of ionic liquid 2.

Table S2

^1H NMR (D_2O) of IL 2 (δ , ppm).

H(1)	H(2,3,4,5)	H(6,7)	H(8,9,10)
3.90	3.33	2.44	1.22
(2H,s)	(8H,q)	(4H,t)	(9H,t)

^{13}C NMR (D_2O) of IL 2 (δ , ppm).

$\text{C}^{\text{a}}, \text{C}^{\text{b}}$	C^{c}	C^{d}	$\text{C}^{\text{e}}, \text{C}^{\text{f}}, \text{C}^{\text{g}}$	$\text{C}^{\text{h}}, \text{C}^{\text{i}}$	$\text{C}^{\text{j}}, \text{C}^{\text{k}}, \text{C}^{\text{l}}$
180.08	57.51	54.87	53.47	32.03	6.81

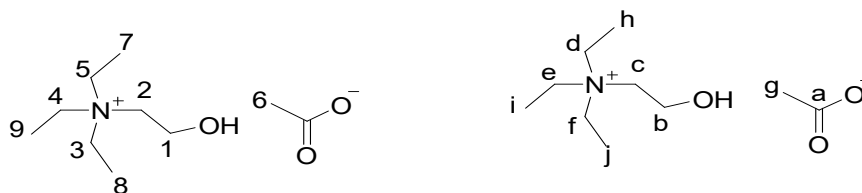


Figure S8. Distribution of ^1H NMR and ^{13}C NMR spectra of ionic liquid 3.

Table S3

^1H NMR (D_2O) of IL 3 (δ , ppm).

H(1)	H(2,3,4,5)	H(6)	H(7,8,9)
3.90	3.30	1.83	1.20
(2H,s)	(8H,q)	(3H,s)	(9H,t)

^{13}C NMR (D_2O) of IL 3 (δ , ppm).

C^{a}	C^{b}	C^{c}	C^{d}	$\text{C}^{\text{e}}, \text{C}^{\text{f}}$	C^{g}	$\text{C}^{\text{h}}, \text{C}^{\text{i}}, \text{C}^{\text{j}}$
181.32	62.62	57.54	54.87	53.47	23.38	6.81

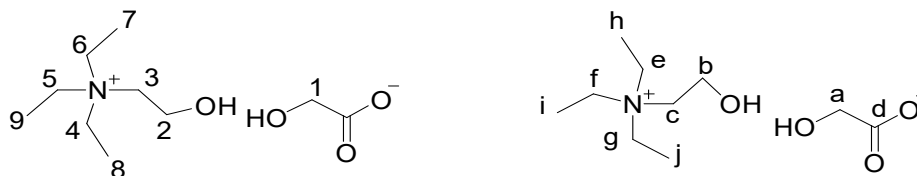


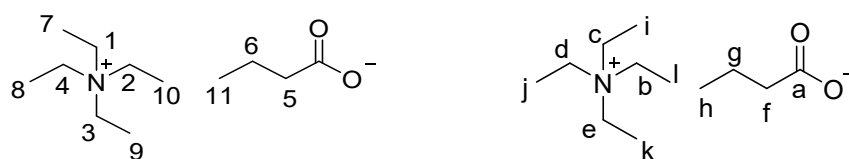
Figure S9. Distribution of ^1H NMR and ^{13}C NMR spectra of ionic liquid 4.

Table S4¹H NMR (D₂O) of IL **4** (δ, ppm).

H(1,2)	H(3,4,5,6)	H(7,8,9)
3.85	3.31	1.21
(4H,d)	(8H,q)	(9H,t)

¹³C NMR (D₂O) of IL **4** (δ, ppm).

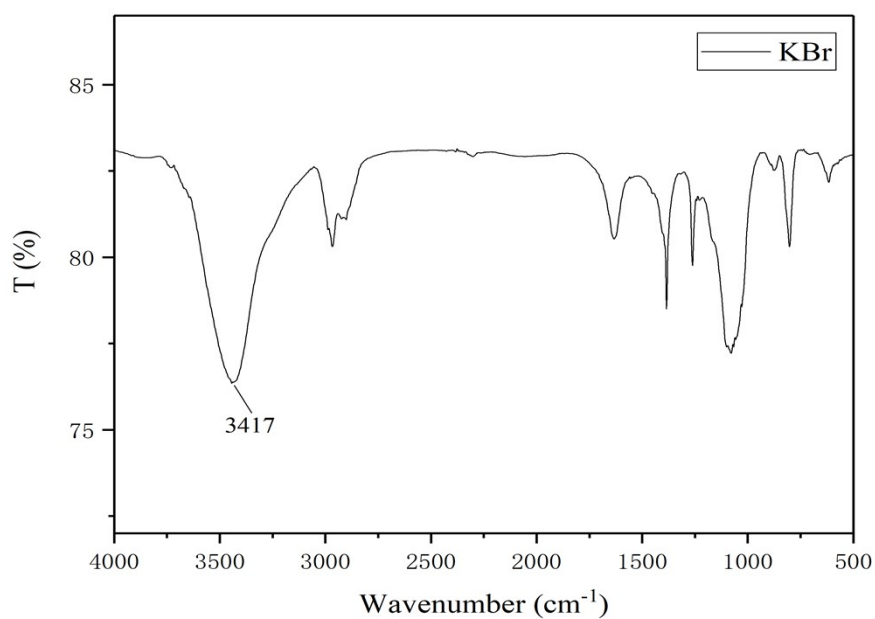
C ^a	C ^b	C ^c	C ^d	C ^e , C ^f , C ^g	C ^h , C ⁱ , C ^j
179.87	61.30	57.50	54.87	53.45	6.79

**Figure S10.** Distribution of ¹H NMR and ¹³C NMR spectra of ionic liquid **5**.**Table S5**¹H NMR (D₂O) of IL **5** (δ, ppm).

H (1,2,3,4)	H(5)	H(6)	H(7,8,9,10)	H(11)
3.17	2.08	1.48	1.18	0.82
(8H,q)	(2H,m)	(2H,m)	(12H,m)	(3H,m)

¹³C NMR (D₂O) of IL **5** (δ, ppm).

C ^a	C ^b , C ^c , C ^d , C ^e	C ^f	C ^g	C ^h	C ⁱ , C ^j , C ^k , C ^l
183.59	52.07	39.69	19.58	13.55	6.82

**Figure S11.** IR spectrum of IL KBr

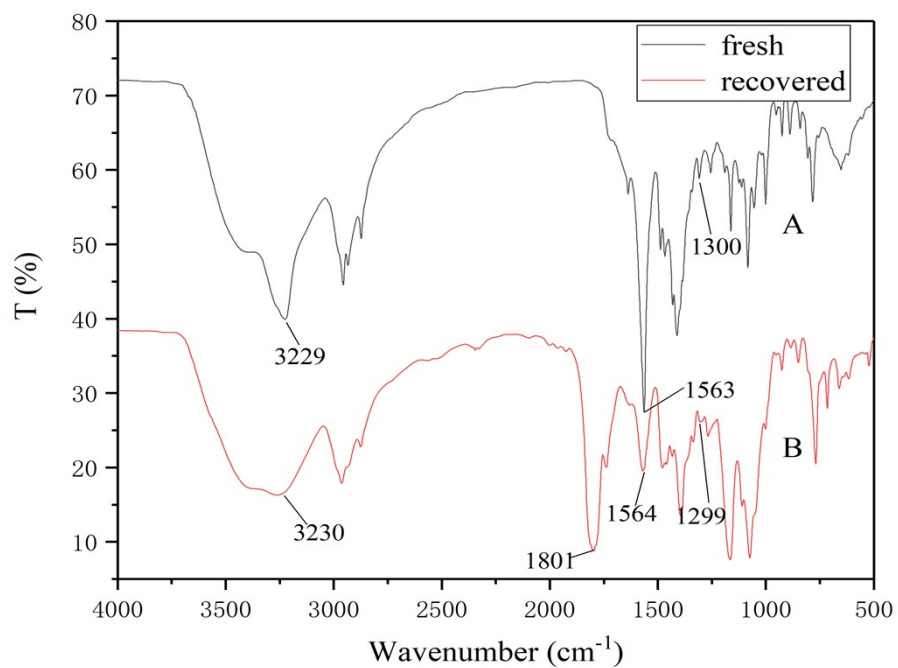


Figure S12. FT-IR spectra of catalyst: A: fresh, B: recovered.

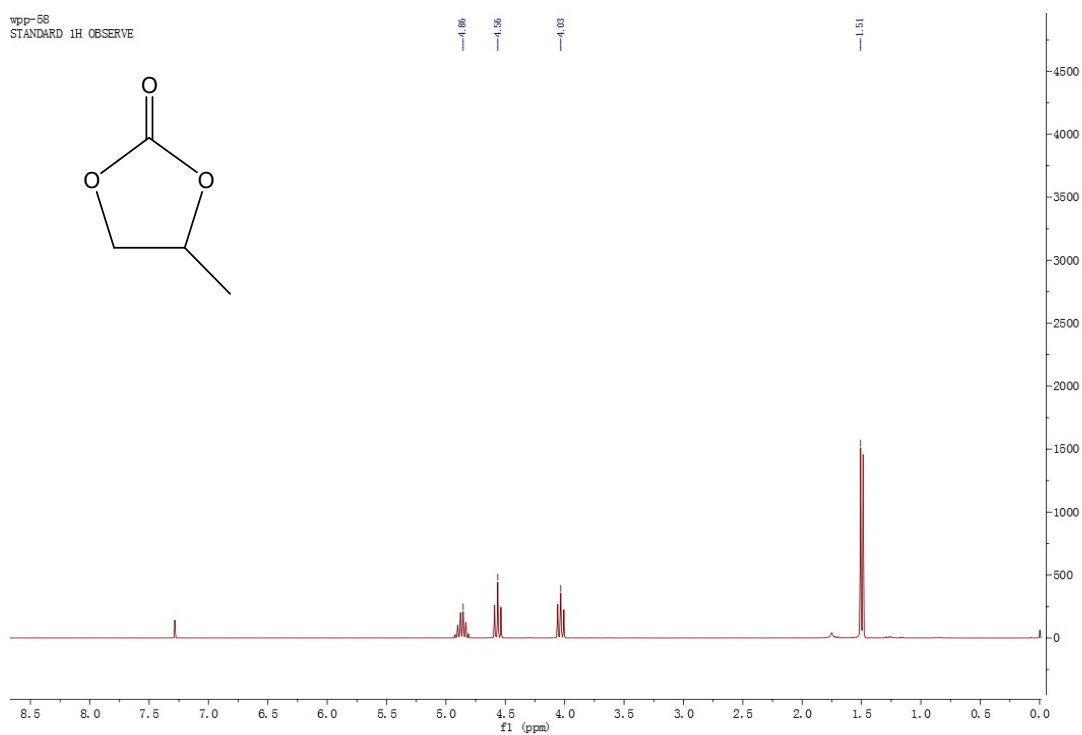


Figure S13. ^1H NMR spectrum of 4-methyl-1,3-dioxolan-2-one

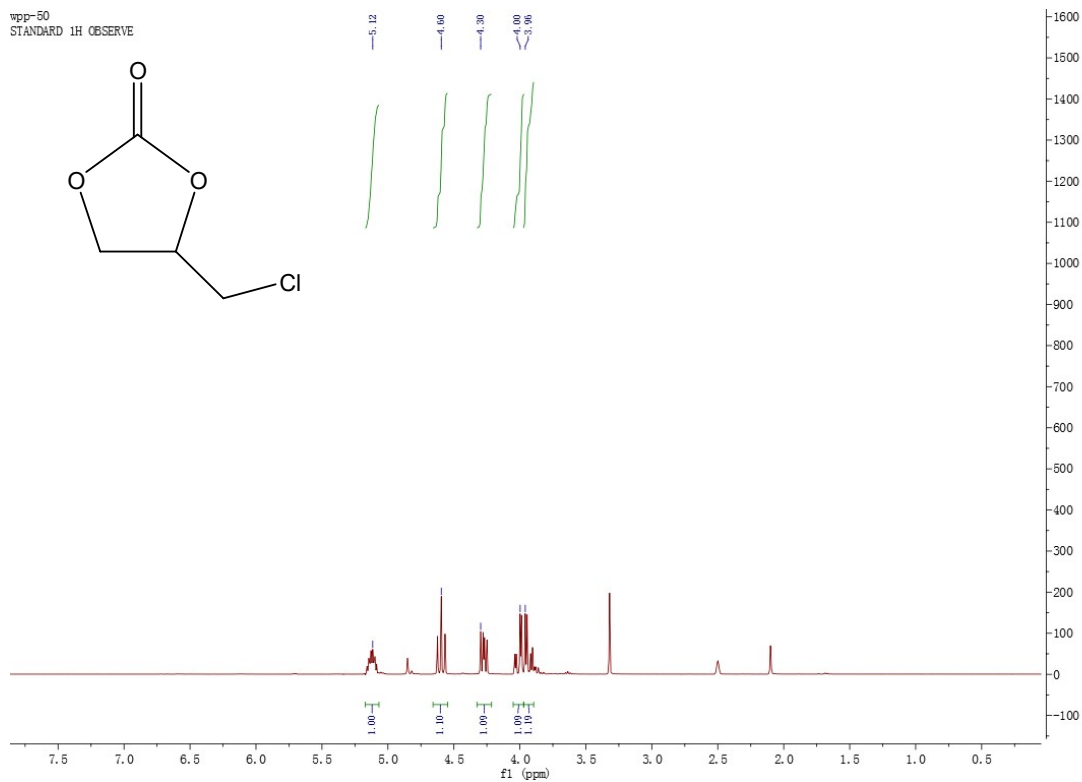


Figure S14. ^1H NMR spectrum of 4-(chloromethyl)-1,3-dioxolan-2-one

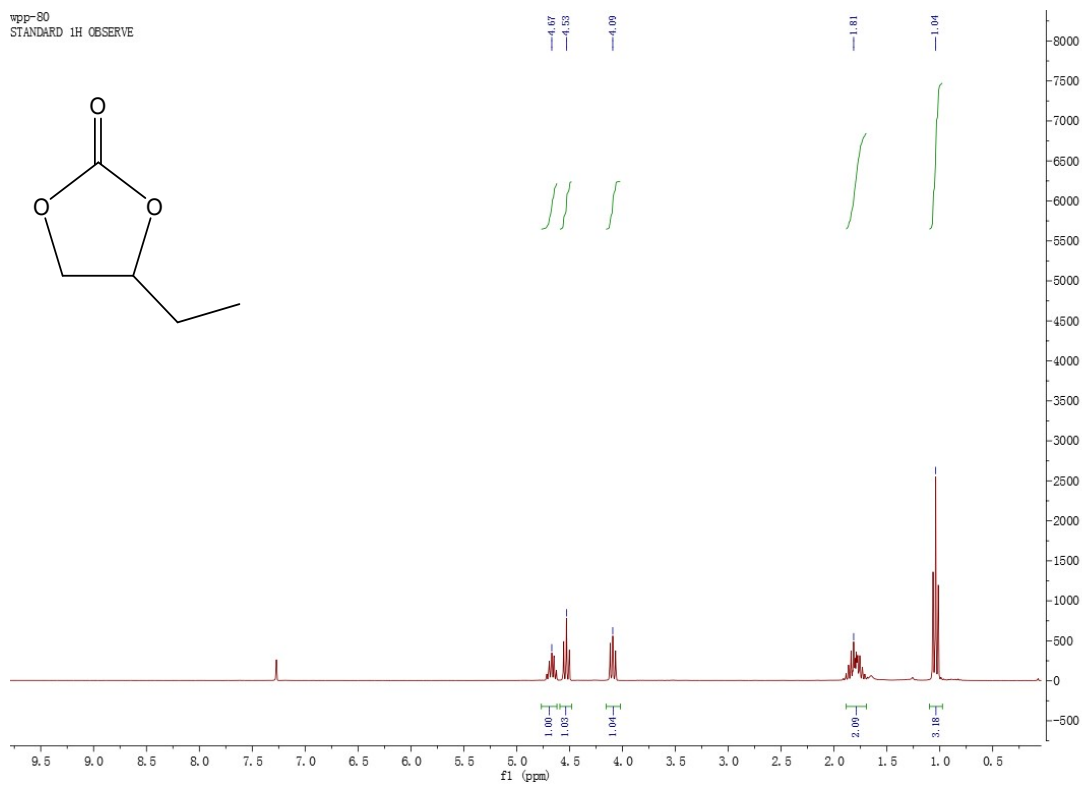


Figure S15. ^1H NMR spectrum of 4-ethyl-1,3-dioxolan-2-one

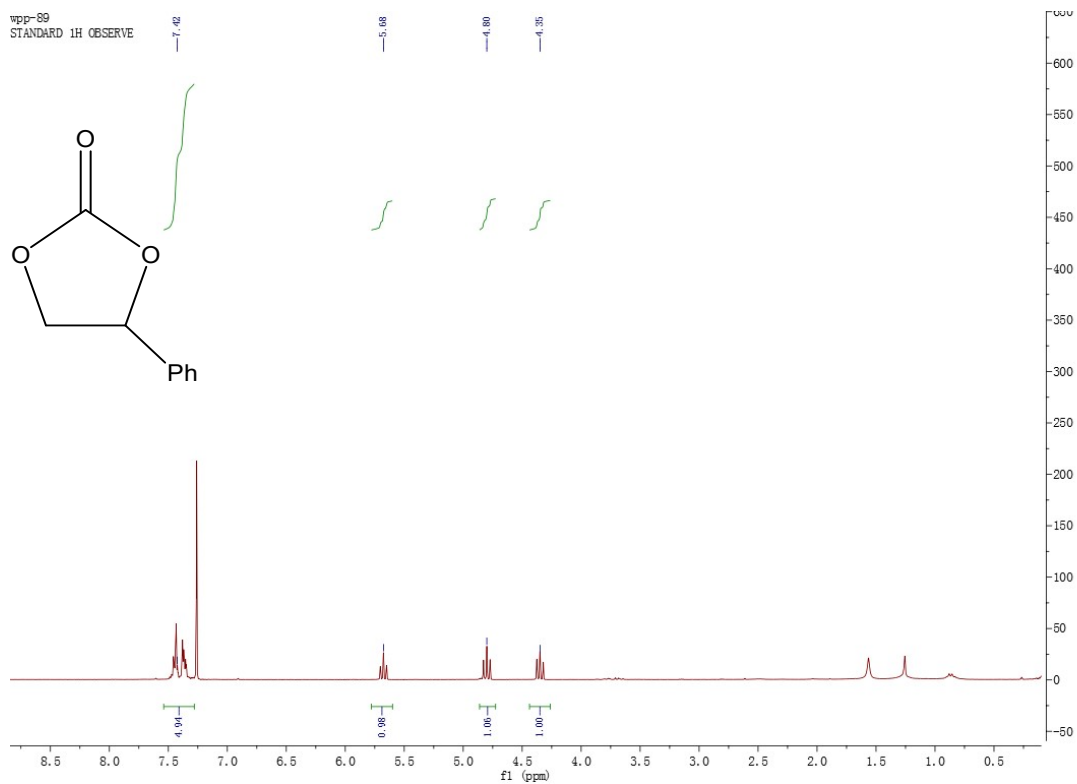


Figure S16. ^1H NMR spectrum of 4-phenyl-1,3-dioxolan-2-one

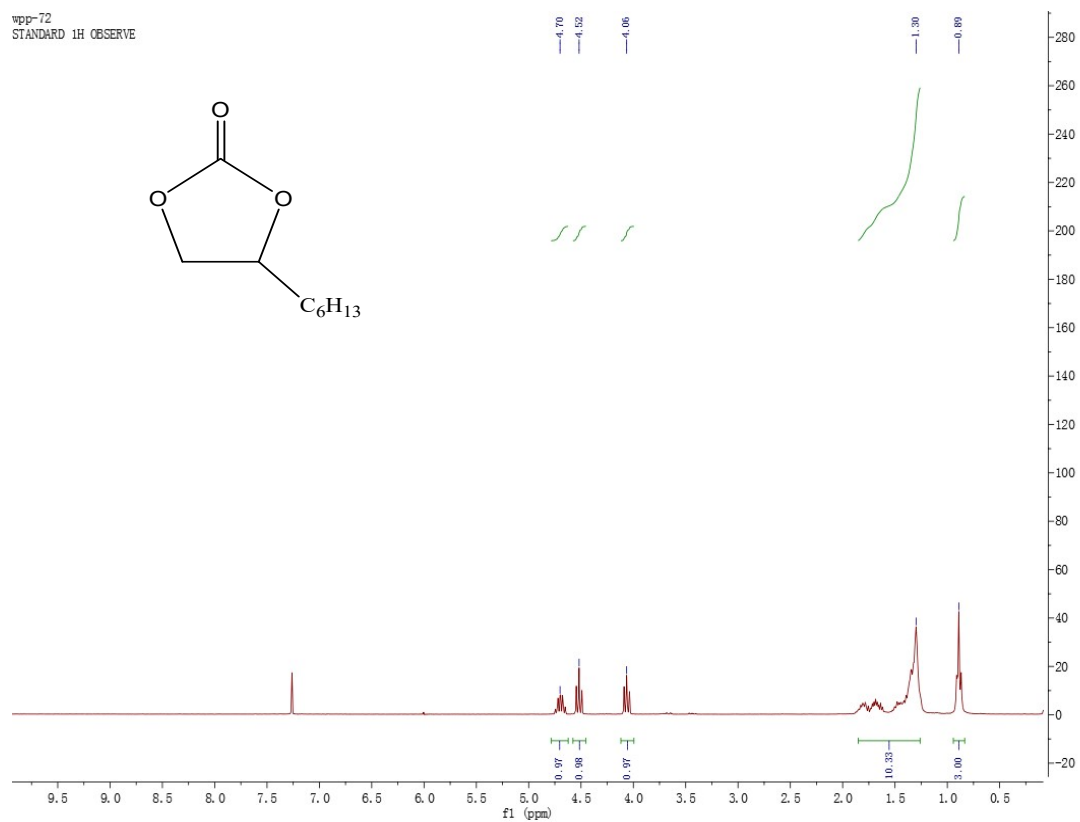


Figure. S17. ^1H NMR spectrum of 4-Hexyl-1,3-dioxolan-2-one

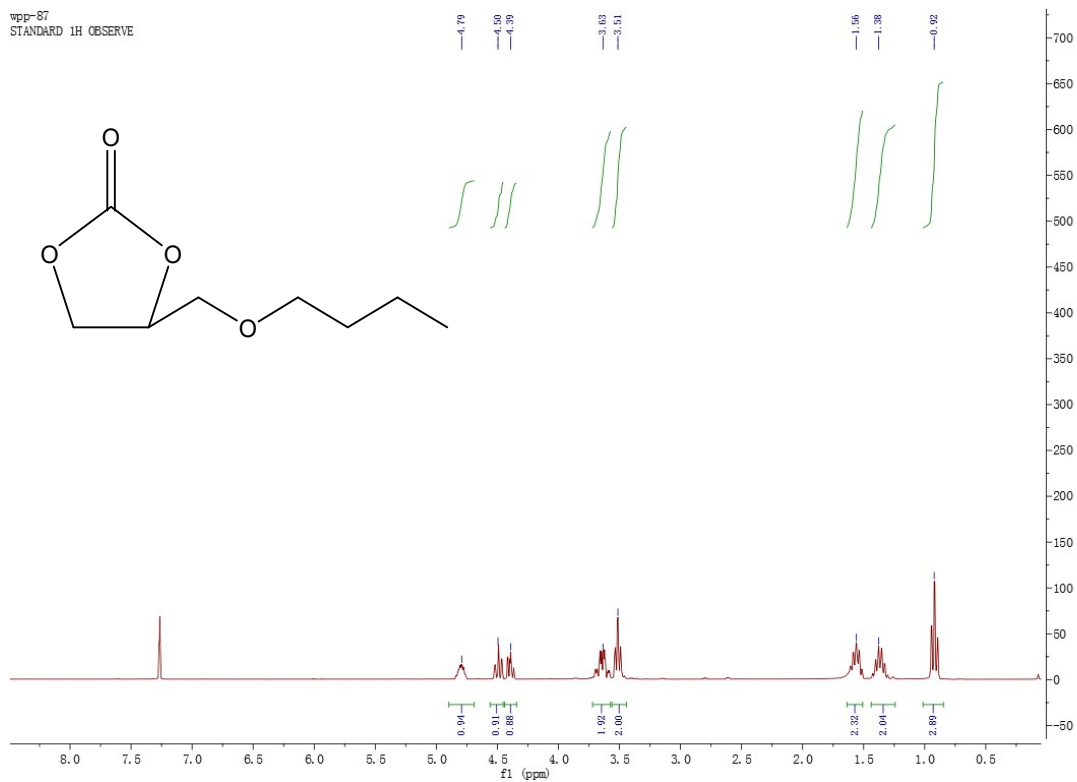


Figure S18. ^1H NMR spectrum of 4-butoxymethyl-1,3-dioxolan-2-one

Ionic liquid 1:

^1H NMR (400 MHz, DMSO) δ 3.78 (s, 2H), 3.34 (m, 8H), 1.80 (d, $J = 4.9$ Hz, 2H), 1.40 (m, 2H), 1.17 (m, 9H), 0.80 (t, 3H).

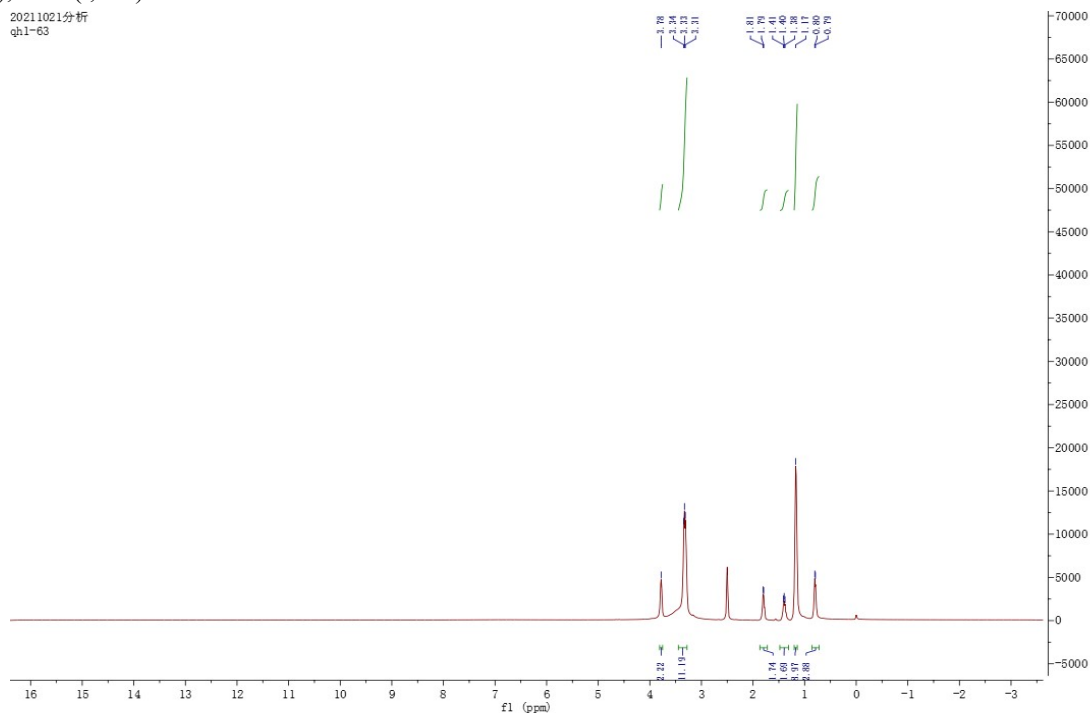


Figure S19-1. ^1H NMR spectrum of IL1 in DMSO

^1H NMR (400 MHz, DMSO) δ 3.79 (s, 3H), 3.34 (m, 8H), 1.80 (t, 2H), 1.40 (m, 2H), 1.17 (d, 9H), 0.80 (t, 3H). epichlorohydrin: 3.90 (m, 1H), 3.54 (m, 1H), 3.24 (m, 1H), 2.85 (m, 1H), 2.78 (m, 1H).

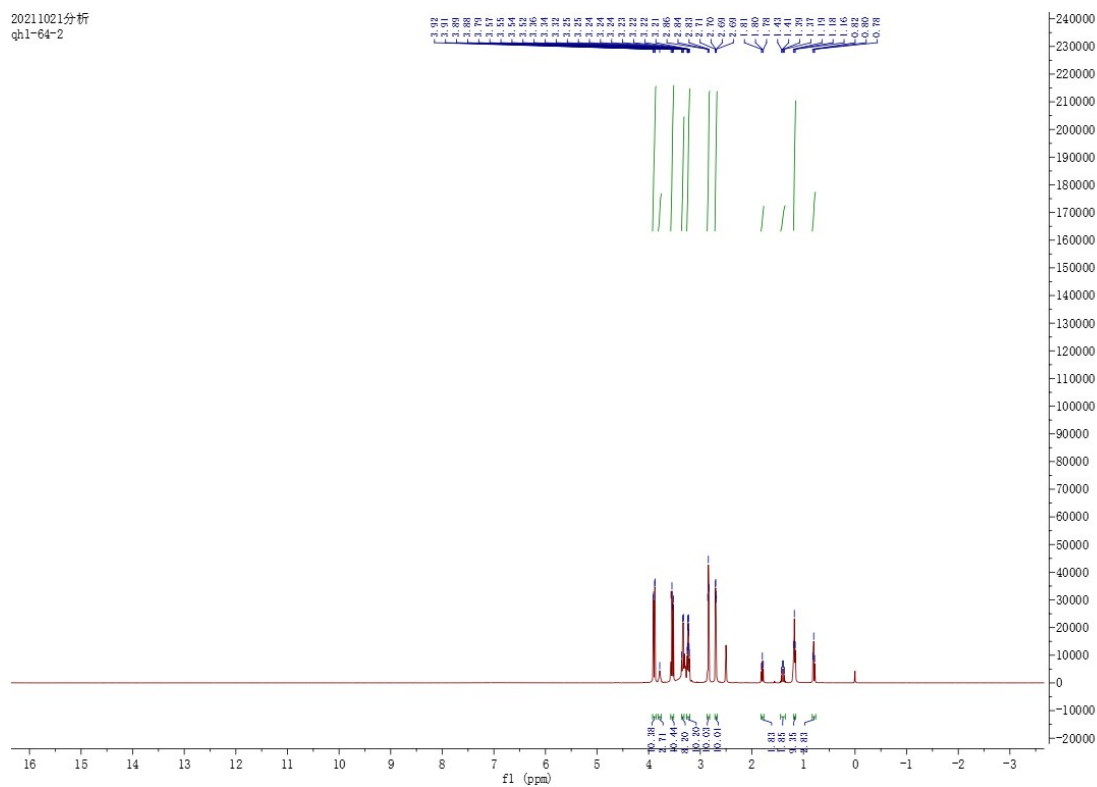


Figure S19-2 ^1H NMR spectrum for IL **1** + epichlorohydrin at 25 °C and $t = 0.5$ h in DMSO

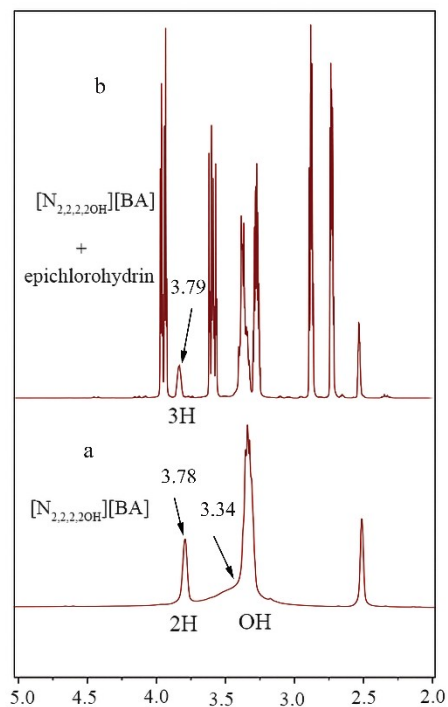


Figure S19-3. (a) ^1H -NMR spectrum for $[\text{N}_{2,2,2,2\text{OH}}][\text{BA}]$ in DMSO. (b) ^1H -NMR spectrum for

$[\text{N}_{2,2,2,2\text{OH}}][\text{BA}]$ + epichlorohydrin at 25 °C and $t = 0.5$ h in DMSO.