In-situ construction of self-supporting Ni-Fe sulfide for high-efficient oxygen evolution

Mingxin Hao,^{a,1} Huizhen Wang,^{a,1} Xiaoling Zhang,^a Yangdong Qu,^a Cuijuan Xuan,^a

Zexing Wu,^{b,*} Min Cui,^c and Jie Wang^{a,*}

^a College of Chemistry and Pharmaceutical Sciences, Qingdao Agricultural University,

Qingdao 266109, P. R. China

^b State Key Laboratory Base of Eco-chemical Engineering, College of Chemistry and

Molecular Engineering, Qingdao University of Science & Technology, 53 Zhengzhou

Road, 266042, Qingdao, P. R. China

^c Collaborative Innovation Center for Green Chemical Manufacturing and Accurate Detection, Key Laboratory of Interfacial Reaction & Sensing Analysis in Universities of Shandong, School of Chemistry and Chemical Engineering, University of Jinan, Jinan 250022, P. R. China

¹ The two authors (Y. Wang and H. Wang) contributed equally to this work.

*Corresponding authors.

E-mail address: wangjie@qau.edu.cn (J. Wang); splswzx@qust.edu.cn (Z. Wu)



Figure S1 CV curves of $(Fe,Ni)_3S_4$ -90 at potential window from 1.1 to 1.2 V under different scan rates.



Figure S2 CV curves of $(Fe,Ni)_3S_4$ -130 at potential window from 1.1 to 1.2 V under different scan rates.



Figure S3 CV curves of $(Fe,Ni)_3S_4$ -150 at potential window from 1.1 to 1.2 V under different scan rates.

Catalysts	j (mA cm ⁻²)	<i>E</i> _{10 mA cm-2} OER (V)	Electrolyte	Refs
NiCo ₂ O ₄ @NiCoFe-OH	10	235.0	1 M KOH	[1]
Co-NC@CoFeS ₂	10	275.0	1 M KOH	[2]
NiFeO _x film	10	336.0	1 M KOH	[3]
NiFe LDH	10	300.0	1 M KOH	[4]
Ni-Fe nanoparticles	10	311.0	1 M NaOH	[5]
NiFe ₂ O _x Spinels	10	356.0	1 M NaOH	[6]
FeNiS ₂	10	310.0	0.1 M KOH	[7]
Ni-Fe-O-S	10	272.0	1 M KOH	[8]
FeNi ₂ S ₄ hollow balloons	10	273.0	1 M KOH	[9]
(Ni, Fe)S ₂ @MoS ₂	10	270.0	1 M KOH	[10]
Ni-Fe-S _{3:1} -160	10	207.0	1 M KOH	[11]
NiS/Fe ₃ O ₄ /carbon nanotube	10	243.0	1 M KOH	[12]
Fe ₃ O ₄ @NiS _x /rGO	10	330.0	1 M KOH	[13]
Fe-Ni ₃ S ₂ /FeNi	10	282.0	1 M KOH	[14]
(Fe,Ni) ₃ S ₄ -110	50	240.0	1 M KOH	This work

Table S1 Comparison of key parameters for the OER performance of recent published references related to Fe-based catalysts.

References

[1] S. Li, X. Yang, S. Yang, Q. Gao, S. Zhang, X. Yu, Y. Fang, S. Yang, X. Cai, *J. Mater. Chem. A*, 2020, **8**, 5601-5611.

[2] Y. Wang, W. Jin, C. Xuan, J. Wang, J. Li, Q. Yu, B. Li, C. Wang, W. Cai, J. Wang, *J. Power Sources*, 2021, **512**, 230430.

[3] L. Trotochaud, J. K. Ranney, K. N. Williams and S. W. Boettcher, *J. Am. Chem. Soc.*, 2012, **134**, 17253-17261.

[4] F. Song and X. Hu, Nat. Commun., 2014, 5, 4477.

[5] S. L. Candelaria, N. M. Bedford, T. J. Woehl, N. S. Rentz, A. R. Showalter, S. Pylypenko, B. A. Bunker, S. Lee, B. Reinhart, Y. Ren, *ACS Catal.*, 2017, **7**, 365-379.

[6] C. N. Chervin, P. A. DeSario, J. F. Parker, E. S. Nelson, B. W. Miller, D. R. Rolison, J.W. Long, *ChemElectroChem*, 2016, **3**, 1369-1375.

[7] J. Jiang, S. Lu, H. Gao, X. Zhang and H.-Q. Yu, Nano Energy, 2016, 27, 526-534.

[8] C. Xuan, J. Wang, W. Xia, J. Zhu, Z. Peng, K. Xia, W. Xiao, H.L. Xin, D. Wang, Heteroatom (P, B, or S) incorporated NiFe-based nanocubes as efficient electrocatalysts for the oxygen evolution reaction. *J. Mater. Chem. A*, 2018, **6**, 7062-7069.

[9] H. Wang, J. Tang, Y. Li, H. Chu, Y. Ge, R. Baines, P. Dong, P. M. Ajayan, J. Shen, M. Ye, *J. Mater. Chem. A*, 2018, **6**, 19417-19424.

[10] Liu Y., Jiang S., Li S., Zhou L., Li Z., Li J., Shao M., *Appl. Catal. B-Environ.*, 2019, **247**, 107-114.
[11] C. Xuan, W. Lei, J. Wang, T. Zhao, C. Lai, Y. Zhu, Y. Sun, D. Wang, *J. Mater. Chem. A*, 2019, **7**, 12350-12357.

[12] K. Srinivas, Y. Chen, B. Wang, B. Yu, X. Wang, Y. Hu, Y. Lu, W. Li, W. Zhang D. Yang, ACS Appl. Mater. Interfaces, 2020, 12, 31552-31563.

[13] G. Zhu, X. Xie, Y. Liu, X. Li, K. Xu, X. Shen, Y. Yao, S.A. Shah, *Appl. Surf. Sci.* 2018, **442**, 256-263.
[14] C.Z. Yuan, Z.T. Sun, Y.F. Jiang, Z.K. Yang, N. Jiang, Z.W. Zhao, U.Y. Qazi, W.H. Zhang, A.W. Xu, *Small*, 2017, **13** 1604161.