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Supporting Information

Enhanced Photocatalytic Performance of ZnO/AgCl Composites Prepared by High-Energy Mechanical Ball Milling

Xuemei Tian^a, Hanliu Wu^a, Xiaoyan Hu^a, Zhonghua Wang^a*, Chunguang Ren^b**, Zhengjun Cheng^a, Lin Dou^c, Ying-Wu Lin^d

^a Chemical Synthesis and Pollution Control Key Laboratory of Sichuan Province, College of Chemistry and Chemical Engineering, China West Normal University, Nanchong 637002, Sichuan, China

^b Yantai Institute of Materia Medica, Yantai 264000, Shandong, China.

^c Key Laboratory of Green Chemistry of Sichuan Institutes of Higher Education, College of Chemistry and Environmental Engineering, Sichuan University of Science and Engineering, Zigong 643000, Sichuan, China

^d School of Chemistry and Chemical Engineering, University of South China, Hengyang 421001, Hunan, China

*Corresponding author, Email: zhwangs@163.com (Z. Wang),

Tel: (+86) 817-2568081, Fax: (+86) 817-2445233.

**Corresponding author, Email: cgren@yimm.ac.cn (C. Ren)



Fig. S1 SEM image of mapping area (A), EDS elements mapping of Ag (B), Cl (C), O (D) and Zn (E) of the BM-ZA-60% sample.



Fig. S2 Calculated band gap of BM-ZA-10% (A), BM-ZA-30% (B), BM- ZA-50% (C) and BM-ZA-70% (D) with $(\alpha hv)^2$ graph of relationship with energy (*hv*).



Fig. S3 ZnO, AgCl and BM-ZA-*x*% photocatalytic degradation of RhB under visible light: (A) Kinetic diagram, (B) Degradation efficiency graph, (C) The linear kinetic fitting graph, (D) Comparison of the pseudo first-order rate constants.



Fig. S4 Photo-dehydrogenation of 1,4-DHP under visible light: (A) Kinetic diagram of ZnO, AgCl and BM-ZA-60%; (B) The linear kinetic fitting graph of ZnO, AgCl and BM-ZA-*x*%.



Fig. S5 The UV-Vis spectra changes of RhB solution under visible light irradiation ($\lambda \ge 400$ nm) in the presence of different photocatalysts: (A) SM-ZA-60%, (B) DP-ZA-60%; The UV-Vis spectra changes of 1,4-DHP solution under visible light irradiation ($\lambda \ge 400$ nm) in the presence of different photocatalysts: (C) SM-ZA-60%, (D) DP-ZA-60%.



Fig. S6 (A) Comparison of the pseudo first-order rate constants of photodegradation of RhB with ZnO, AgCl, BM-ZA-60%, DP-ZA-60% and SM-ZA-60%; (B) Comparison of the pseudo first-order rate constants of photooxidation of 1,4-DHP with ZnO, AgCl, BM-ZA-60%, DP-ZA-60% and SM-ZA-60%.



Fig. S7 The fluorescence spectra changes of TA alkaline solution in the presence of $TiO_2(A)$ and BM-ZA-60% (B).