

**Electronic Supplementary Material (ESI) for New Journal of
Chemistry**

**Influences of the precursor molar ratio in synthesis on the structures
and visible-light driven CO₂ reduction into solar fuel of CdS catalyst**

Cong Huang,^{a,b} Xuehua Zhang,^{*b,c} Dongyang Li,^b Mengyue Wang^{b,c} and Qiang Wu^{*a}

^a Shanghai Key Laboratory of Materials Protection and Advanced Materials in Electric
Power, Shanghai University of Electric Power, Shanghai, 200090, China

^b CAS Key Laboratory of Nanosystem and Hierarchical Fabrication, CAS Center for
Excellence in Nanoscience, National Center for Nanoscience and Technology, Beijing
100190, China.

^c University of Chinese Academy of Sciences, Beijing 100049, China

* Corresponding author. Tel.: +86 10 8254 5657; Fax: +86 10 6265 6765.

*E-mail address: zhangxh@nanoctr.cn and qiangwu@shiep.edu.cn

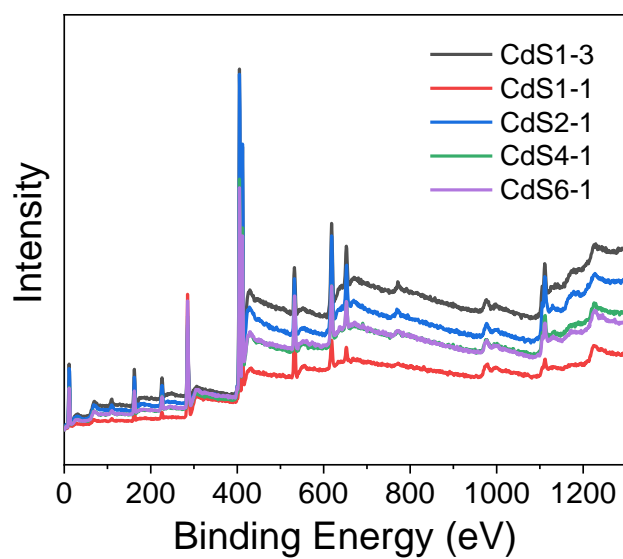


Figure S1. XPS survey spectra of the as-synthesized CdS samples.

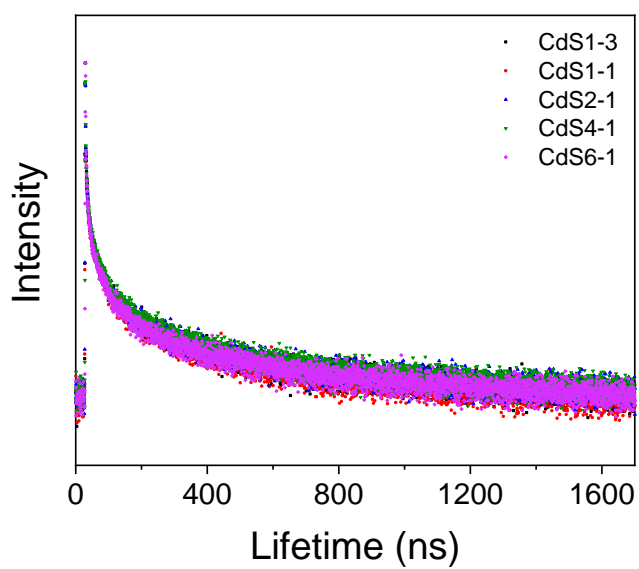


Figure S2. Time-resolved PL decay curves of the as-synthesized CdS samples.

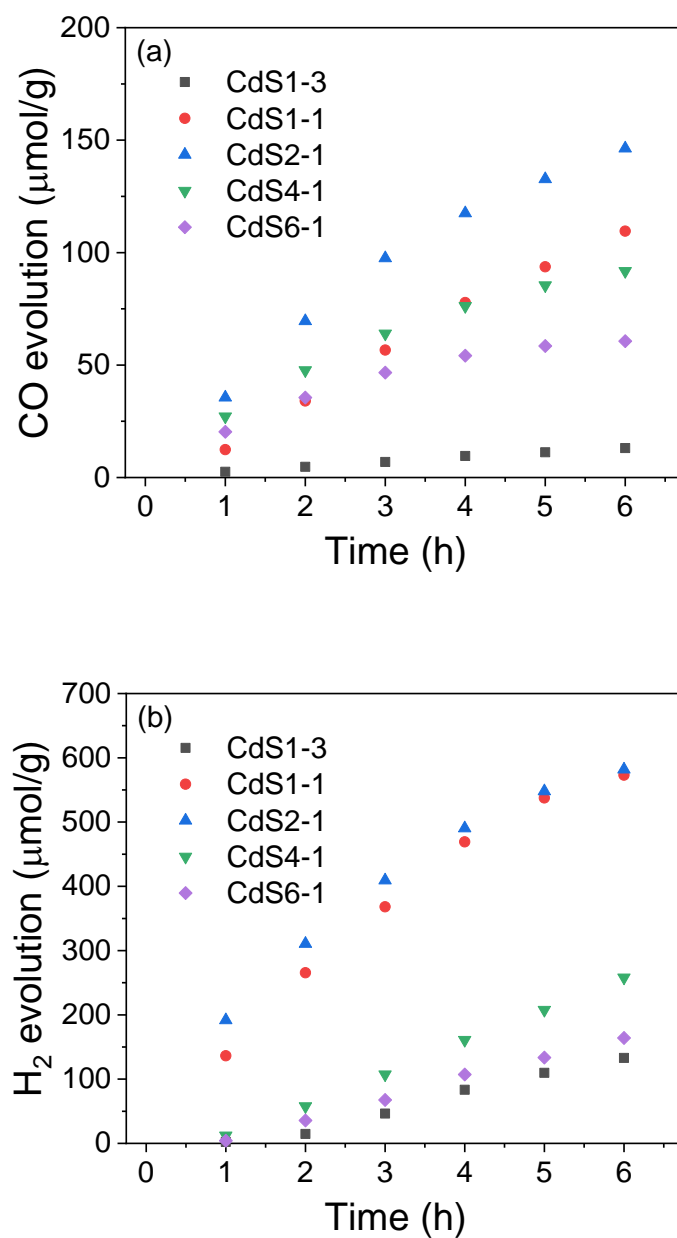


Figure S3. Time-dependent gas evolution yield of (a) CO and (b) H₂ for photocatalytic reduction of CO₂ over the as-synthesized CdS catalysts in 6 h in CH₃CN:H₂O:TEOA (3:1:1, 100 mL) mixed solution under visible-light irradiation ($420 \text{ nm} \leq \lambda \leq 780 \text{ nm}$).

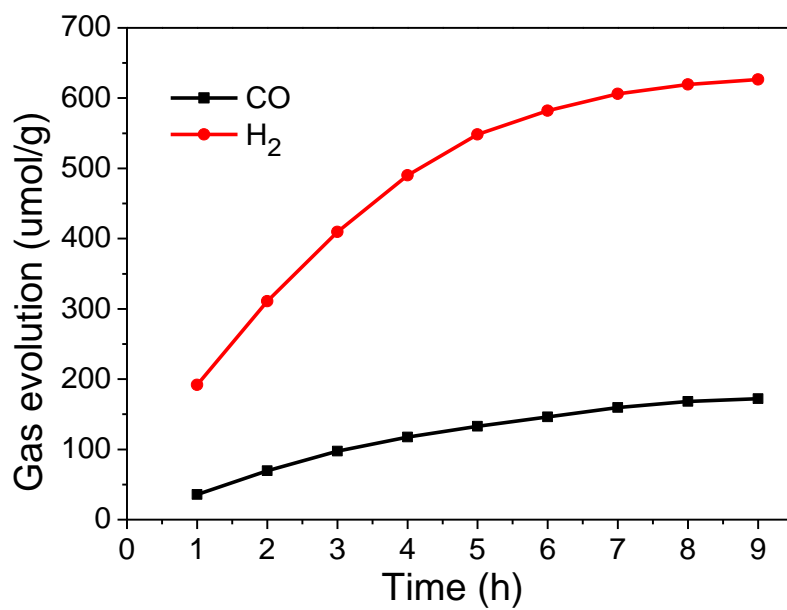


Figure S4. Time dependence of CO and H₂ evolution amount using CdS2-1 catalyst in CH₃CN:H₂O:TEOA (3:1:1, 100 mL) mixed solution under visible-light illumination for 9 h.