Electronic Supplementary Material (ESI) for New Journal of Chemistry.

Supplementary Information for

Fabrication of Fe_3O_4@poly(methyl methacrylate-co-glycidyl methacrylate) microspheres via miniemulsion polymerization using porous microspheres as templates for removal of cationic dyes

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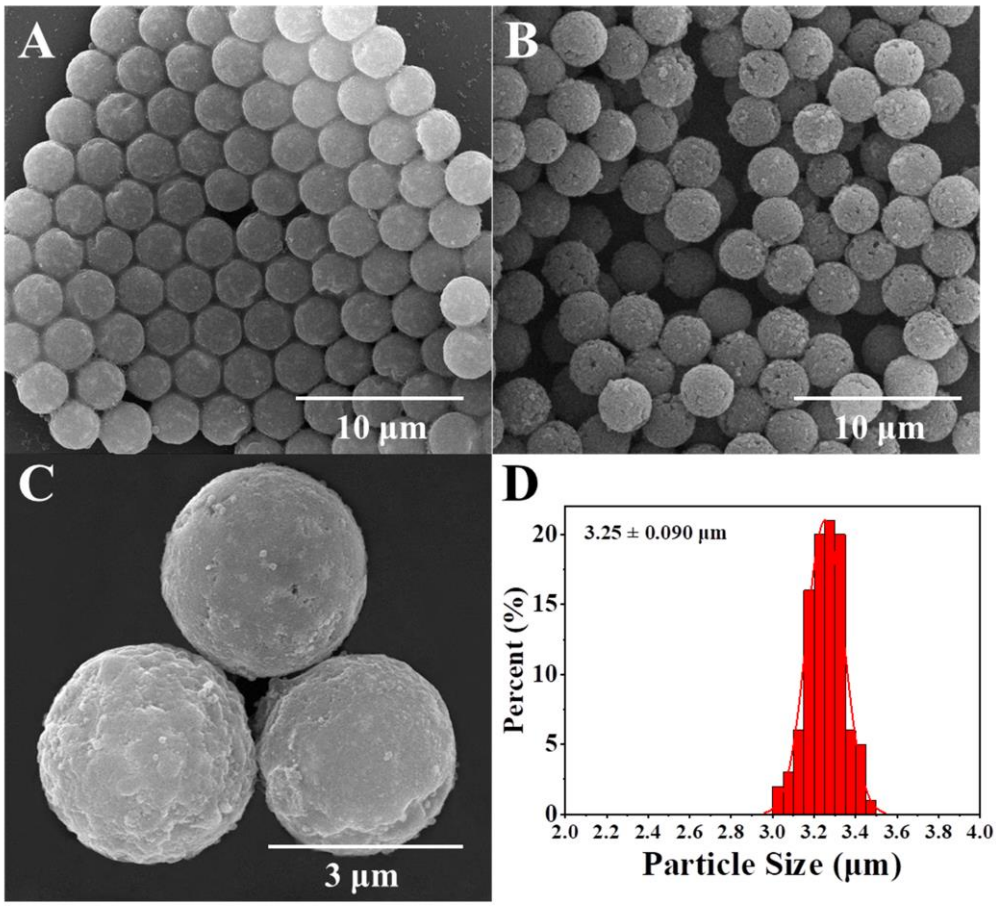


Fig. S1 SEM images of (A) magnetic Fe_3O_4@P(MMA-co-GMA) microspheres; (B) (C) magnetic Fe_3O_4@P(MMA-co-GMA) microspheres with 3/4 monomer addition volume; (D) Particle size distribution of magnetic Fe_3O_4@P(MMA-co-GMA) microspheres.

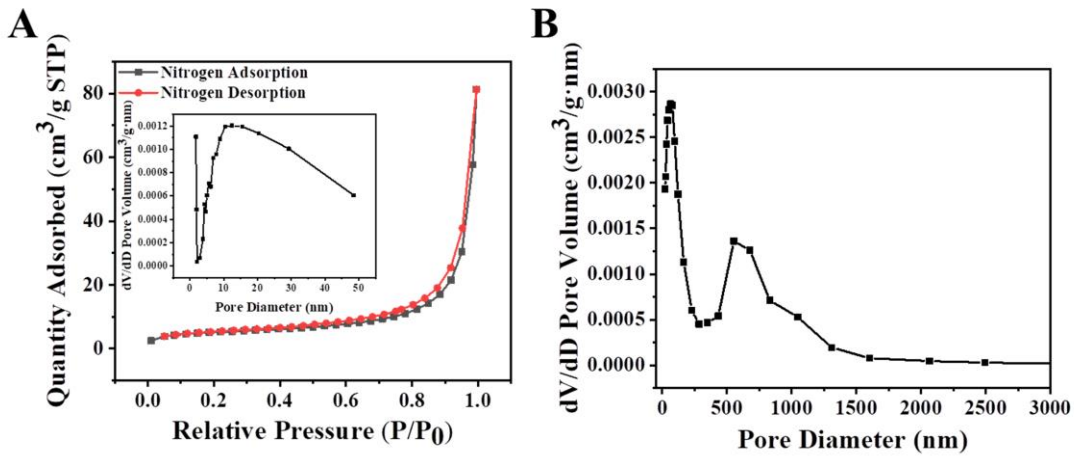
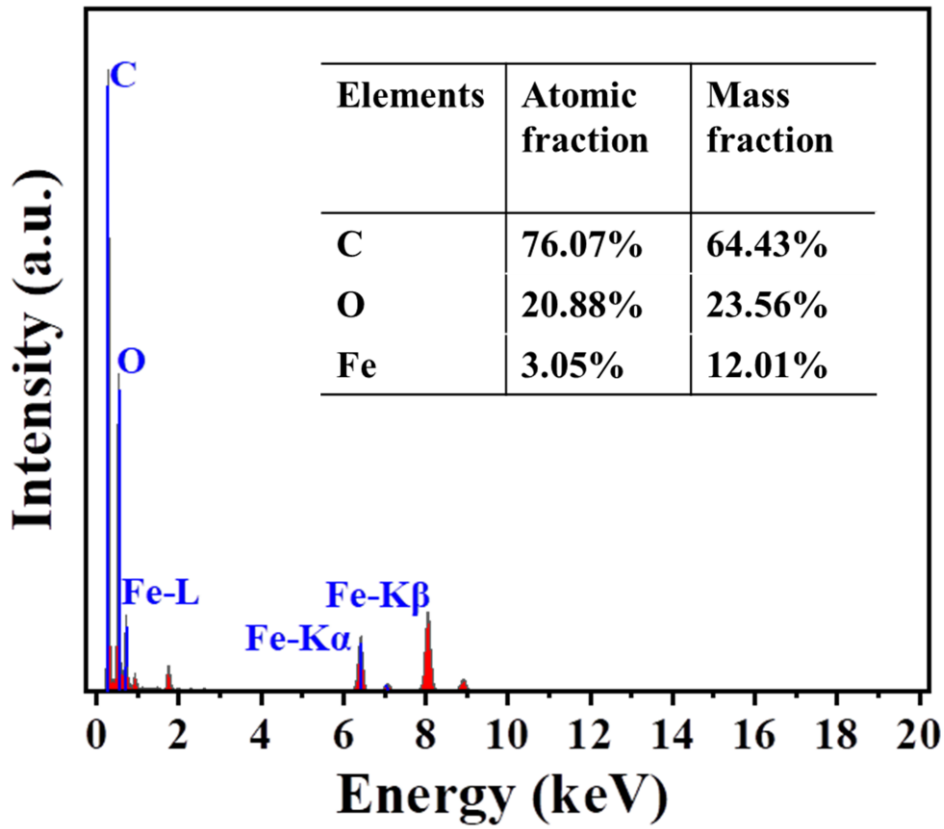


Fig. S2 (A) Adsorption-desorption isotherms of N_2 and the mesopore pore size distribution (inset) of porous P(MMA-co-GMA) microspheres; (B) Macropore pore size distribution of porous P(MMA-co-GMA) microspheres by MIP.



Elements	Atomic fraction	Mass fraction
C	76.07%	64.43%
O	20.88%	23.56%
Fe	3.05%	12.01%

Fig. S3 EDX analysises of magnetic Fe_3O_4@P(MMA-co-GMA) microspheres.

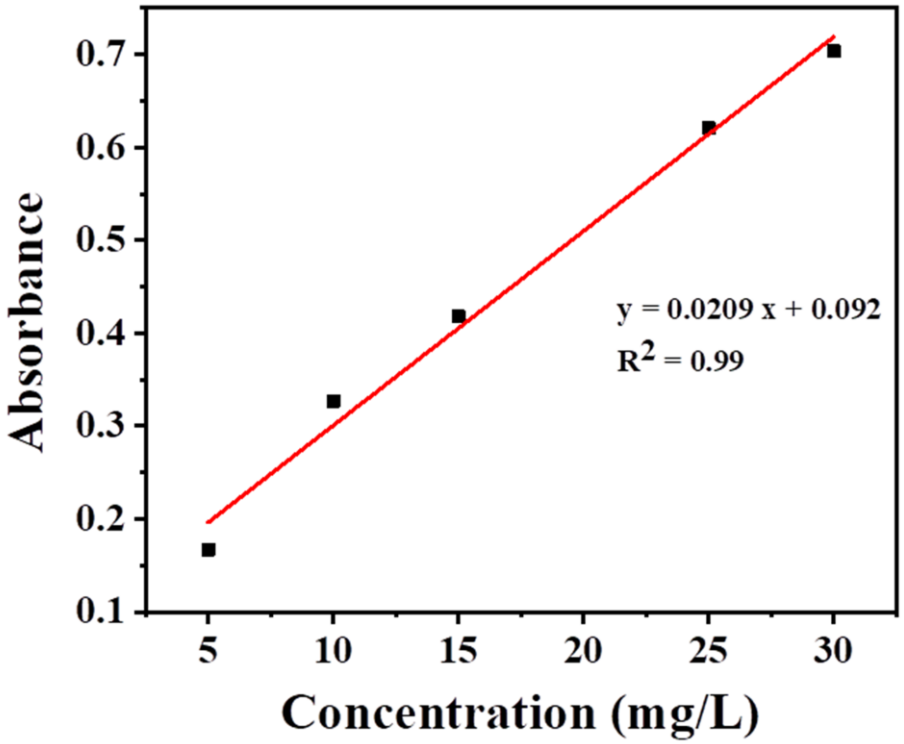


Fig. S4 Standard curve of UV absorption of nile blue dyes.

Table S1 Recycle performance of magnetic Fe_3O_4@P(MMA-co-GMA) microspheres in adsorption-desorption cycles.

Recycle time	Dye adsorption capacity (mg/g)	Dye adsorption efficiency (%)	Dye desorption capacity (mg/g)	Dye desorption efficiency (%)
1	99.5	99.5	91.0	91.5
2	97.4	97.4	87.3	89.6
3	97.1	97.1	85.2	87.7
4	95.2	95.2	82.1	86.2
5	94.3	94.3	79.7	84.6
6	95.0	95.0	79.3	83.5
7	93.3	93.3	75.3	80.7