

Photoelectron velocity-map imaging spectroscopy of nickel carbide: Examination of the low-lying electronic states

Zhiling Liu,^{*,1} Qingyang Lin,¹ Ya Li,¹ Jing He,¹ Jingmei Jiao,¹ Lianxia Liu,¹

Yonghong Yan,¹ Hai-Shun Wu,¹ Fuqiang Zhang,¹ Jianfeng Jia,¹ and Hua Xie^{*,2}

¹School of Chemical and Material Science, Key Laboratory of Magnetic Molecules & Magnetic Information Materials, the Ministry of Education, Shanxi Normal University. No. 339, Taiyu Road, Taiyuan, Shanxi 030031, China.

Electronic mail: lzling@sxnu.edu.cn.

²State Key Laboratory of Molecular Reaction Dynamics, Dalian Institute of Chemical Physics, Chinese Academy of Sciences, 457 Zhongshan Road, Dalian, Liaoning 116023, China.

Electronic mail: xiehua@dicp.ac.cn.

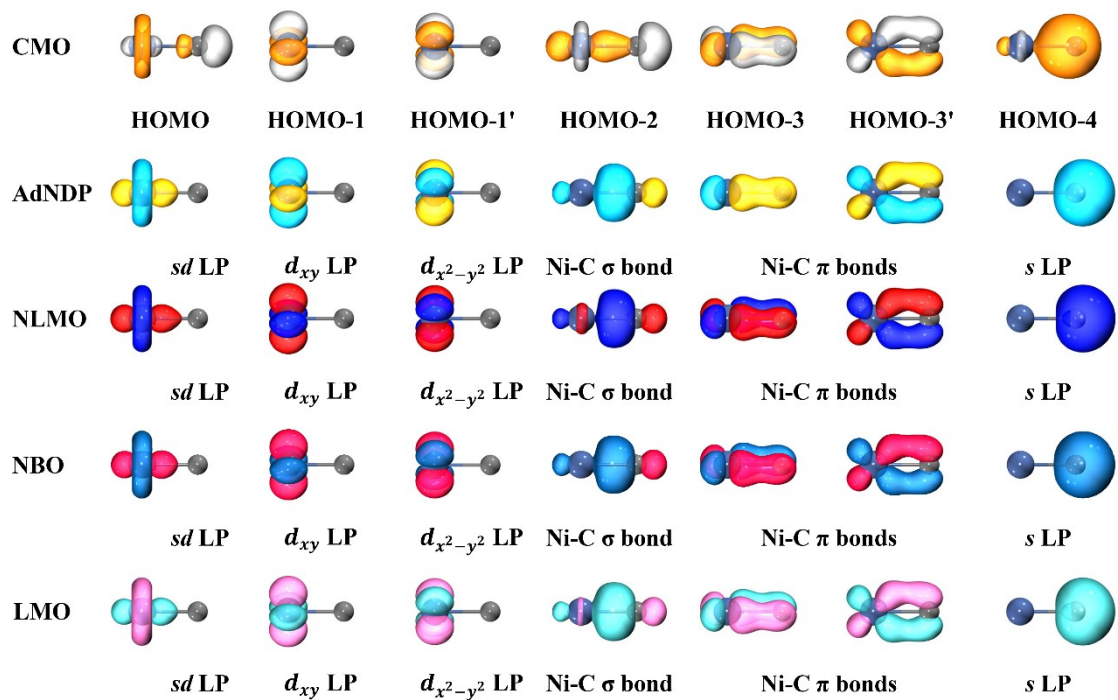


FIG. S1. A comparison of selected CMOs, AdNDP, NLMO, NBO, LMO for the NiC molecule calculated at the BP86-D3(BJ)/def2-TZVPP level. (isosurface = 0.1 au)

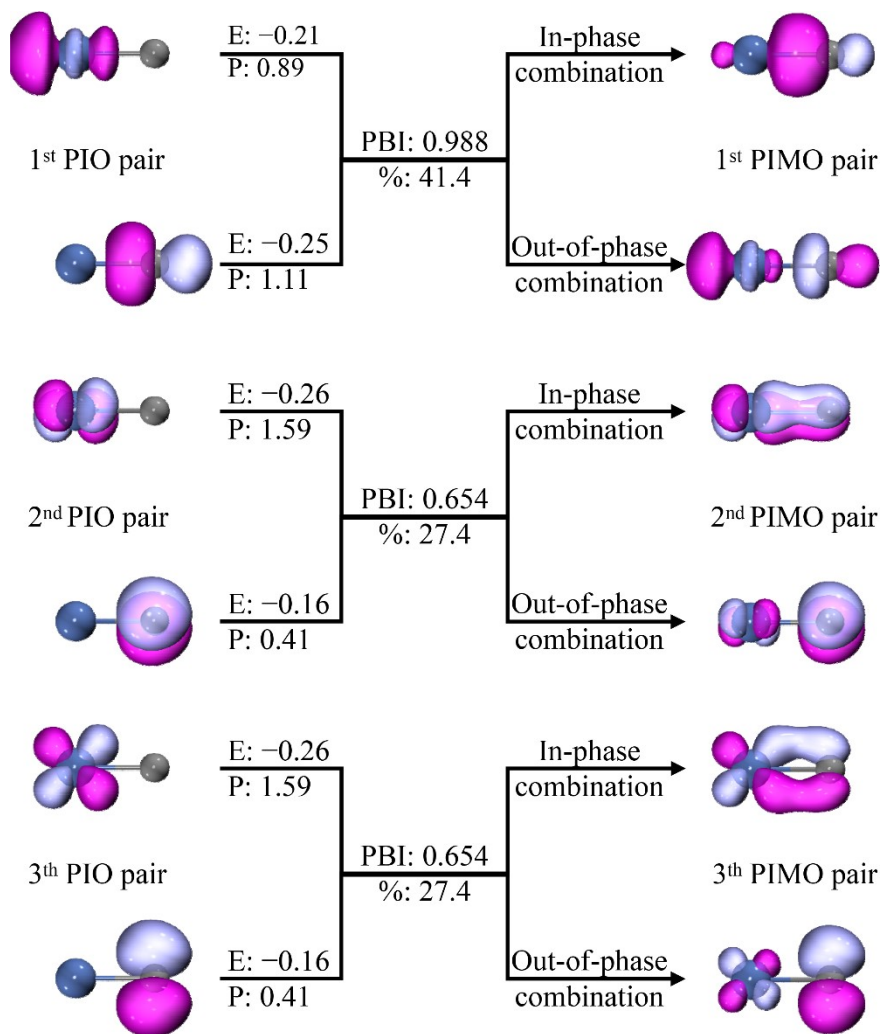


FIG. S2. Results of PIO analysis on NiC with Ni atom and C atom as two fragments at the BP86-D3(BJ)/def2-TZVPP level. (isosurface = 0.1 au)

TABLE S1. EDA-NOCV results for NiC at the BP86-D3(BJ)/TZ2P level using charged nickel ion and carbon ion in different electronic reference configuration as interacting moieties. Energy values are given in kcal·mol⁻¹.

Fragments	C ⁺ [<i>D</i> , (2 <i>s</i> _σ) ² (2 <i>p</i> _π) ⁰ (2 <i>p</i> _σ) ¹]	C ⁻ [<i>Q</i> , (2 <i>s</i> _σ) ² (2 <i>p</i> _π) ² (2 <i>p</i> _σ) ¹]	C ⁻ [<i>D</i> , (2 <i>s</i> _σ) ² (2 <i>p</i> _π) ³ (2 <i>p</i> _σ) ⁰]	C ⁻ [<i>D</i> , (2 <i>s</i> _σ) ² (2 <i>p</i> _π) ¹ (2 <i>p</i> _σ) ²]
	Ni ⁻ [<i>D</i> , (3 <i>d</i> _π) ⁴ (3 <i>d</i> _σ) ² (3 <i>d</i> _δ) ⁴ (4 <i>s</i> _σ) ¹]	Ni ⁺ [<i>Q</i> , (3 <i>d</i> _π) ² (3 <i>d</i> _σ) ² (3 <i>d</i> _δ) ⁴ (4 <i>s</i> _σ) ¹]	Ni ⁺ [<i>D</i> , (3 <i>d</i> _π) ¹ (3 <i>d</i> _σ) ² (3 <i>d</i> _δ) ⁴ (4 <i>s</i> _σ) ²]	Ni ⁺ [<i>D</i> , (3 <i>d</i> _π) ³ (3 <i>d</i> _σ) ² (3 <i>d</i> _δ) ⁴ (4 <i>s</i> _σ) ⁰]
Symmetry	<i>C</i> _{∞v}	<i>C</i> _{∞v}	<i>C</i> _{2v}	<i>C</i> _{2v}
Δ <i>E</i> _{int}	-432.12	-303.90	-559.88	-414.52
Δ <i>E</i> _{pauli}	271.92	331.01	236.59	324.86
Δ <i>E</i> _{disp} ^a	-1.19 (0.17%)	-1.19 (0.19%)	-1.19 (0.15%)	-1.19 (0.16%)
Δ <i>E</i> _{elstat} ^a	-311.89 (43.30%)	-391.47 (61.66%)	-230.74 (28.97%)	-393.66 (53.24%)
Δ <i>E</i> _{orb} ^a	-390.95 (55.53%)	-242.25 (38.16%)	-564.54 (70.88%)	-344.54 (46.60%)
Δ <i>E</i> _{orb(σ)} ^b	-129.30 (33.07%)	-64.61 (26.67%)	-344.50 (61.02%)	-69.18 (20.08%)
Δ <i>E</i> _{orb(π)} ^b	-120.75 (30.89%)	-80.12 (33.07%)	-100.15 (17.74%)	-234.46 (68.05%)
Δ <i>E</i> _{orb(π')} ^b	-120.75 (30.89%)	-80.12 (33.07%)	-119.65 (21.19%)	-23.92 (6.94%)

^aThe italic capital letters *D* and *Q* in square brackets stand for the electronic doublet and quartet states, respectively.

^bThe value in parentheses gives the percentage contribution to the total attractive interactions Δ*E*_{disp} + Δ*E*_{elstat} + Δ*E*_{orb}.

^cThe value in parentheses gives the percentage contribution to the total orbital interactions Δ*E*_{orb}.