

Supporting Information

Novel Fe_{2.55}Sb₂ Alloy Nanoparticles incorporated in N-doped Carbon as Bifunctional Oxygen Electrocatalyst for Rechargeable Zn-Air Battery

Xing Xie^a, Panpan Sun^{a,c*}, Weitao Liu^a, Tao Gong^a, Xiaowei Lv^{a,c}, Liang Fang^b, Yongan Wei^{a*}, Xiaohua Sun^{a,c*}

a College of Materials and Chemical Engineering, Key Laboratory of Inorganic Nonmetallic Crystalline and Energy Conversion Materials, China Three Gorges University, Yichang 443002, China.

b Guangxi Key Laboratory of Optical and Electronic Materials and Devices, Guilin University of Technology, Guilin 541004, China.

c Hubei Three Gorges Laboratory, Yichang, Hubei 443007.

*Corresponding author. E-mail:

sunpp705@nenu.edu.cn. weiya605@nenu.edu.cn. mksxh@163.com.

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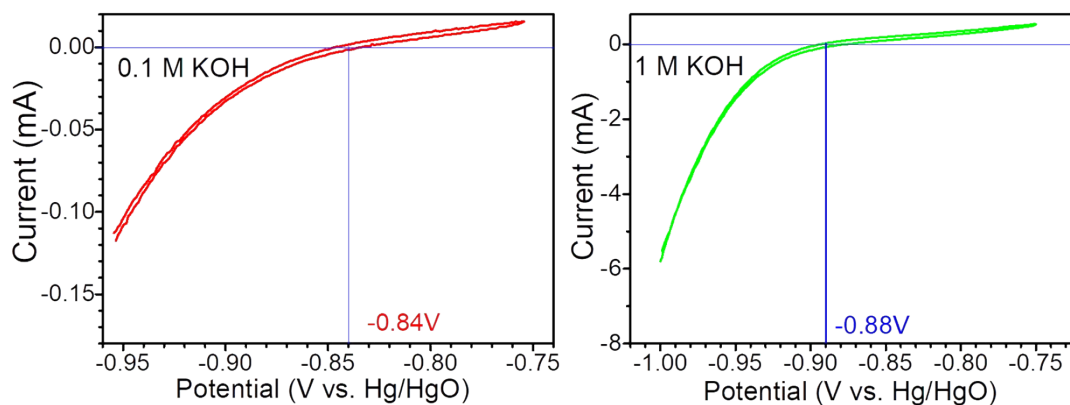


Fig. S1. Calibration of Hg/HgO reference electrode with respect to reversible hydrogen electrode (RHE).

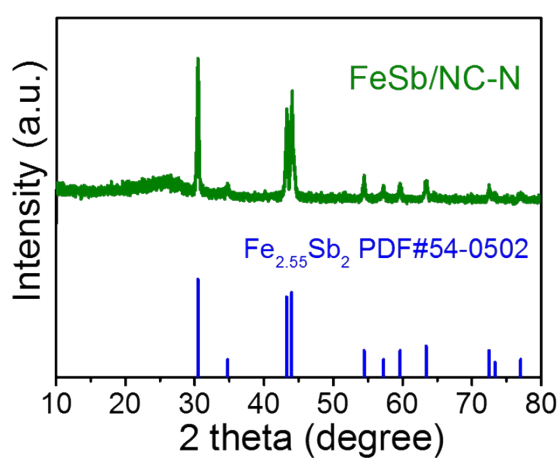


Fig. S2. XRD pattern of FeSb/NC-N catalyst.

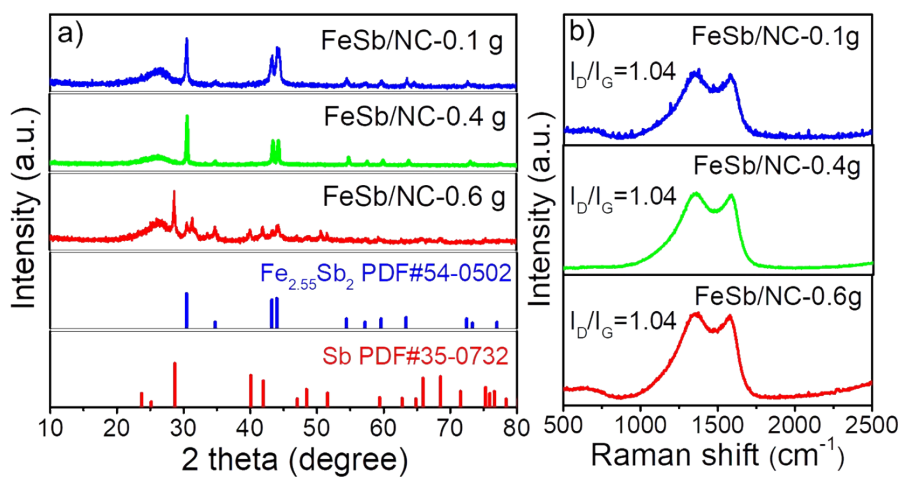


Fig. S3. a) XRD patterns and b) Raman spectra of FeSb/NC with different feeding amount of SbCl_3 .

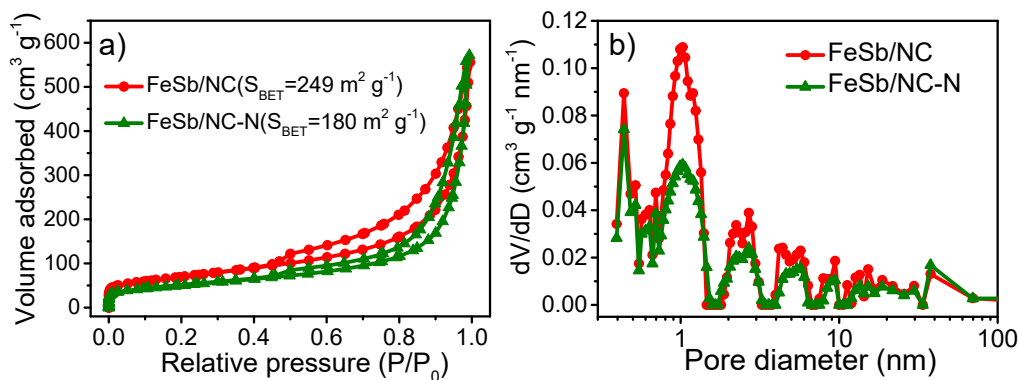


Fig. S4. a) N₂ adsorption-desorption isotherms and b) the corresponding pore size distribution curves of FeSb/NC and FeSb/NC-N catalysts.

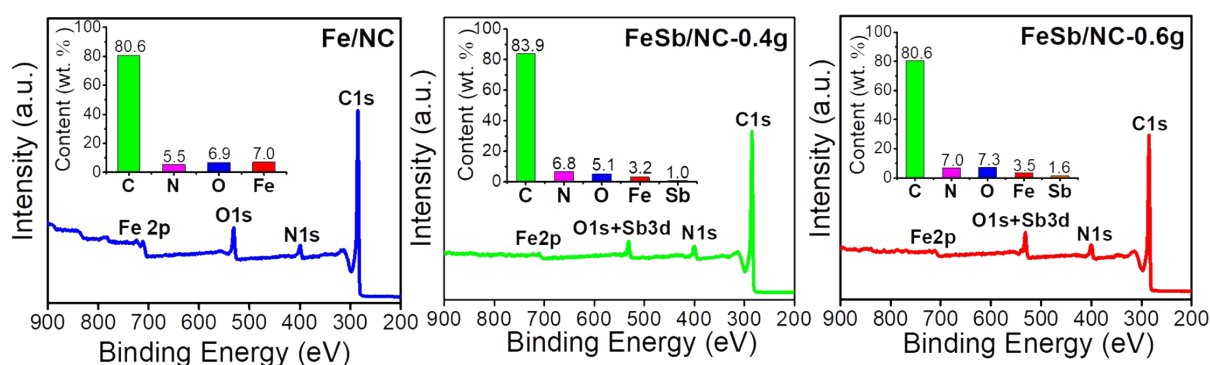


Fig. S5. The XPS survey spectra and the corresponding elemental contents of Fe/NC, FeSb/NC-0.4g and FeSb/NC-0.6g.

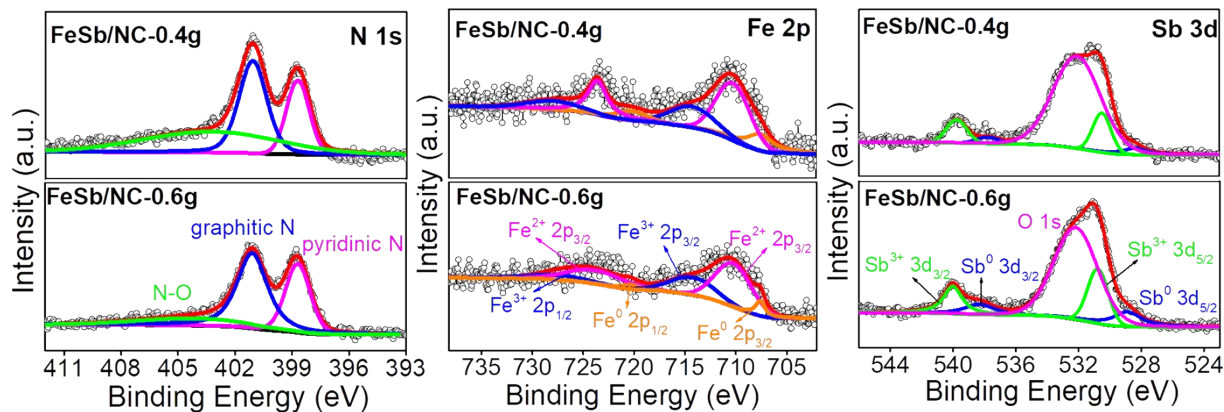


Fig. S6. High-resolution N 1s, Fe 2p and Sb 3d spectra of FeSb/NC-0.4g and FeSb/NC-0.6g.

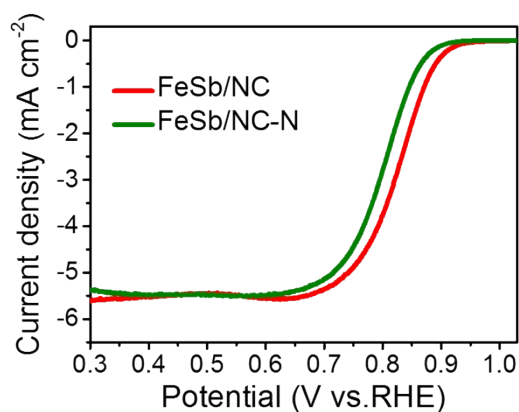


Fig. S7. LSV curves for ORR of FeSb/NC and FeSb/NC-N catalysts.

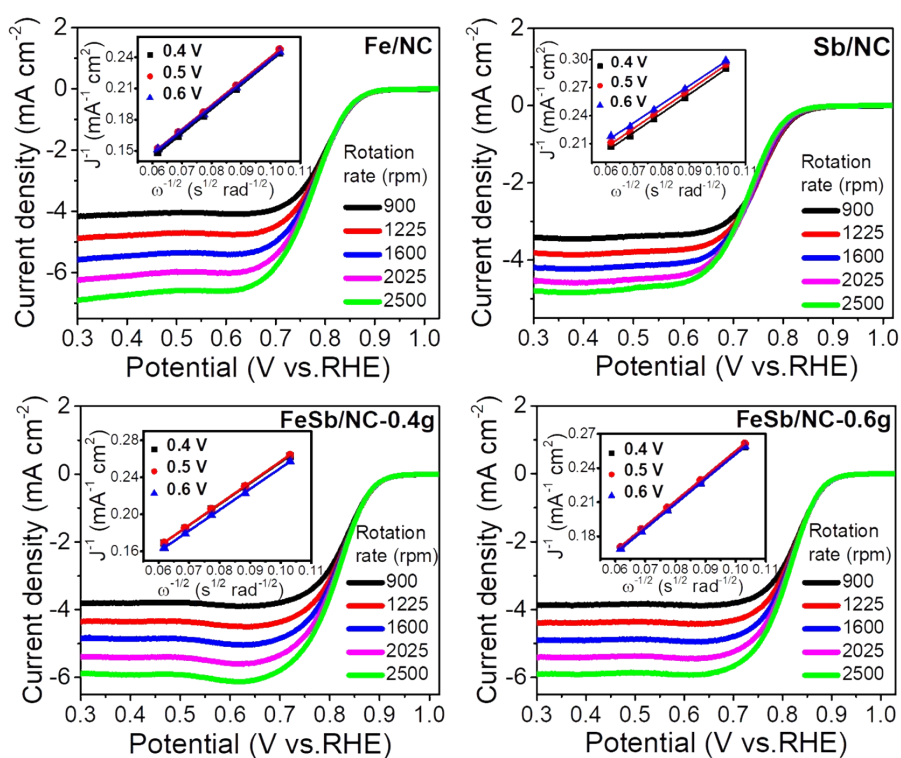


Fig. S8. LSV curves of Fe/NC, Sb/NC, FeSb/NC-0.4g and FeSb/NC-0.6g catalysts measured at different rotation rates and the corresponding K-L plots.

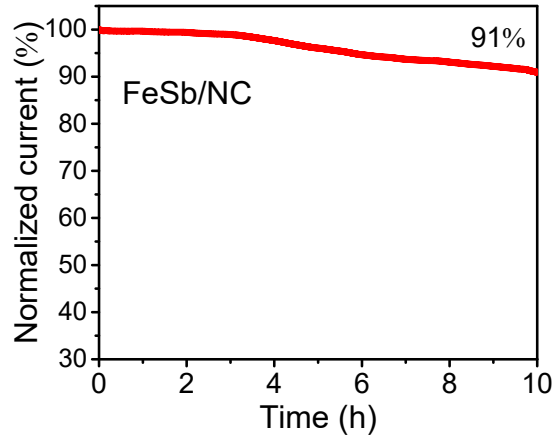


Fig. S9. ORR durability test of FeSb/NC performed on carbon paper.

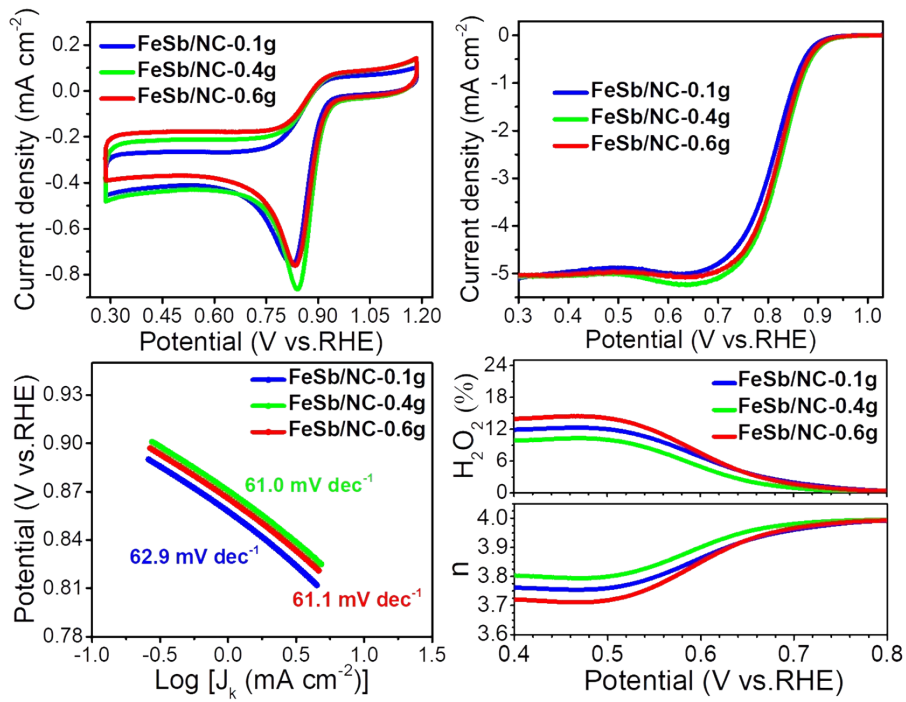


Fig. S10. ORR performance of FeSb/NC with different feeding amount of SbCl_3 .

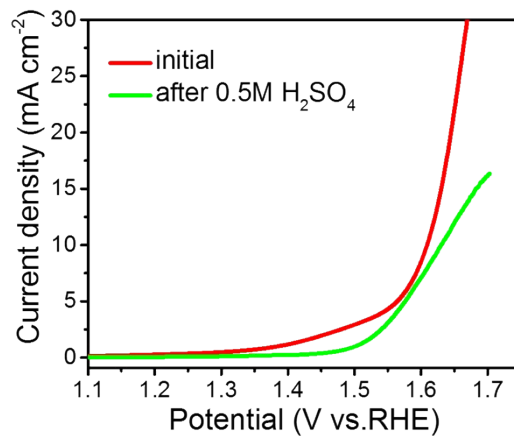


Fig. S11. OER LSV curves of FeSb/NC before and after acid leaching.

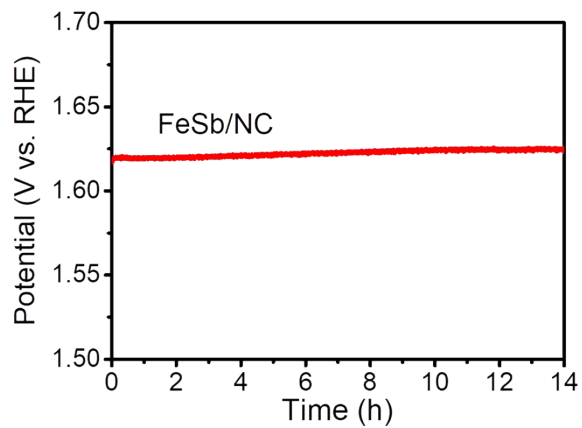


Fig. S12. Chronopotentiometric curve at a constant current density of 10 mA cm^{-2} for on FeSb/NC.

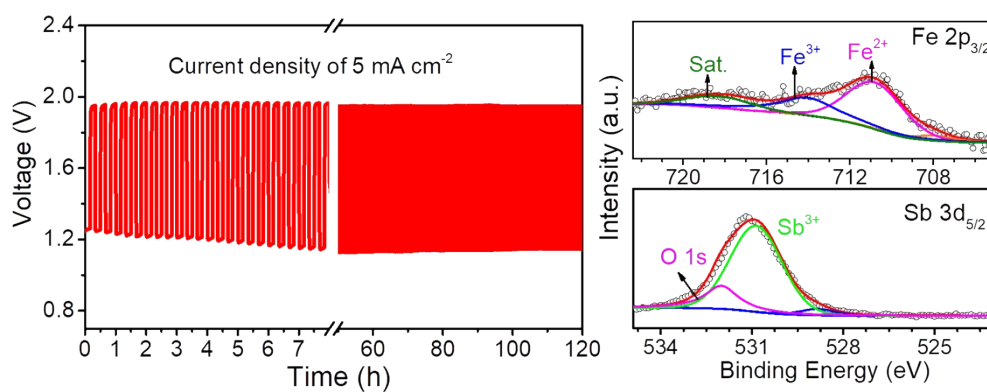


Fig. S13. High-resolution Fe 2p_{3/2} and Sb 3d_{5/2} spectra of FeSb/NC after 120 h of charge-discharge cycling test at 5 mA cm^{-2} for 120 h.

Table S1 ORR parameters for for FeSb/NC-0.1g, FeSb/NC-0.4g, FeSb/NC-0.6g.

Samples	Peak potential (V vs. RHE)	E_{onset} (V vs. RHE)	$E_{1/2}$ (V vs. RHE)	J_L (mA cm ⁻²)	Tafel slope (mV dec ⁻¹)
FeSb/NC-0.1g	0.82	0.89	0.81	5.00	62.9
FeSb/NC-0.4g	0.84	0.90	0.82	5.24	61.0
FeSb/NC-0.6g	0.83	0.90	0.82	5.07	61.1

Table S2 Comparison of Zn-air battery performance of the FeSb/NC catalyst with recently reported electrocatalysts.

Catalysts	OCV (V)	Peak power density (mW cm ⁻²)	Specific capacity (mA h g ⁻¹ Zn)	Cycling durability @J(mA cm ⁻²)	Ref.
FeSb/NC	1.46	175	751	887 h @5 616 h @10	This work
N-CoS ₂ YSSs	1.41	81	744	165 h @10	[1]
FeNi ₃ @NC	1.39	139	756	30 h @5	[2]
Co@hNCTs-800	1.45	149	746	500 h @5	[3]
Zn-Ni ₃ FeN/NG	1.60	158	650	180 h @10	[4]
Fe-enriched-FeNi ₃ /NC	1.43	89	734	14 h @10	[5]
CoSe ₂ @NC	1.48	137	751	83 h @10	[6]
CoFe@NC/NC HNSs-700	1.49	184	774	50 h @5	[7]
Mn-RuO ₂	1.55	181	812	2500 h @10	[8]
Co ₆ Mo ₆ C ₂ -Co@NC	1.38	45	760	300 h @10	[9]
CoFe/S-N-C	1.48	130	814	100 h @10	[10]
NiCo/NHCS-TUC-3	1.59	256	757	70 h @5	[11]
CS350-10Ar	1.36	135		120 h @5	[12]
Co/NGC-3		134	716	120 h @5	[13]
FeCo/Co ₂ P@NPCF	1.44	154		107 h @10	[14]
MnSAC		258	675	100 h @5	[15]
Co ₉ S ₈ @Co/Mn-S,N-PC	1.49	80		210 h @10	[16]
CoFe-NCNFs	1.52	116		110 h @5	[17]
Fe-NC SAC		180	786	160 cycles @20	[18]
FeCo/Se-CNT	1.54	173		70 h @5	[19]
FeCu _{0.3} -N/C	1.50	111	682	75 h @5	[20]
NiFe ₃ @NGHS-NCNT _s	1.51	127		166 h @10	[21]
f-FeCo-CNT	1.48	196	754	180 h @20	[22]
Catalysts	OCV (V)	Peak power density (mW cm ⁻²)	Specific capacity (mA h g ⁻¹ Zn)	Cycling durability @J(mA cm ⁻²)	Ref.

CoSb/NC-0.8	1.42	180		160 h @10	[23]
CoSb₃/NCL-30		211		60 h @10	[24]
S-LaCoO₃	1.47	92	747	100 h @2	[32]
LSCP-3	1.50	52	740	60 h @10	[33]
Pb₂Ru₂O_{6.5}		195		33 h @10	[34]
YRO	1.17	145		34 h @10	[35]

Table S3 Comparison of BI and SBI values of FeSb/NC catalyst with recently reported electrocatalysts.

Catalysts	ORR (V) @-3 mA/cm²_{geo}	OER (V) @10 mA/cm²_{geo}	BI (V)	SBI (uA/cm²_{BET})	Ref.
FeSb/NC	0.82	1.72	0.90	4.9	This work
Co₉S₈/(N,S-doped carbon)	0.84	1.64	0.80	1.3	[25]
N-doped CNT framework	0.83	1.60	0.77	6.8	[26]
N,P-doped C foam	0.80	1.90	1.10	0.8	[27]
Ba_{0.5}Sr_{0.5}Co_{0.8}F_{e_{0.2}O_{3-δ}}	0.74	1.58	0.84	2.7	[28]
α-MnO₂	0.76	1.72	0.96	14.7	[29]
Co_xMn_{3-x}O₄/C	0.82	1.80	0.98	92.6	[30]
Pb₂Ru₂O_{7-x}	0.74	1.43	0.69	274.0	[31]

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