

Performances of Sodium-ion Supercapattery using LaMnO_3 and rGO in non-Aqueous Electrolyte

*Vaithiyathan Sankar Devi, Kalidoss Kannadasan, Pamangadan C Sharafudeen and Perumal Elumalai**

Electrochemical Energy Storage Lab, Department of Green Energy Technology,
Madanjeet School of Green Energy Technologies, Pondicherry University,
Puducherry – 605014, India

Supplementary data

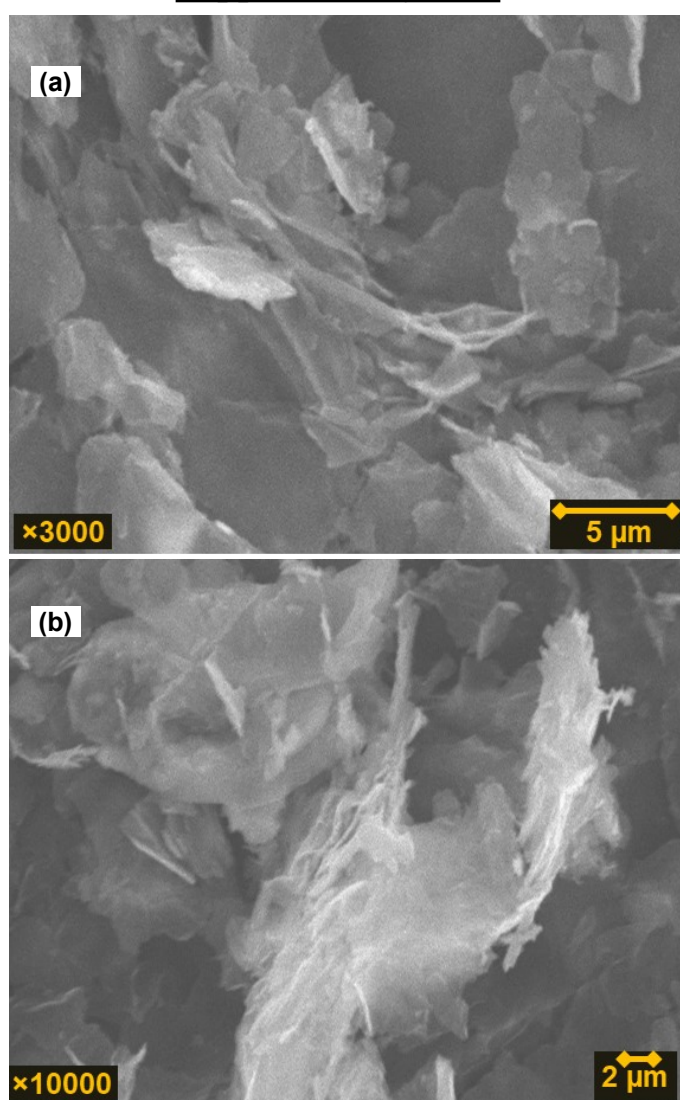


Fig. S1 (a, b) SEM images recorded for the rGO sample at different magnifications. The characteristic folded rGO flakes are visible in the SEM images indicating better surface area which has reflected in high specific capacitance of the rGO electrode.

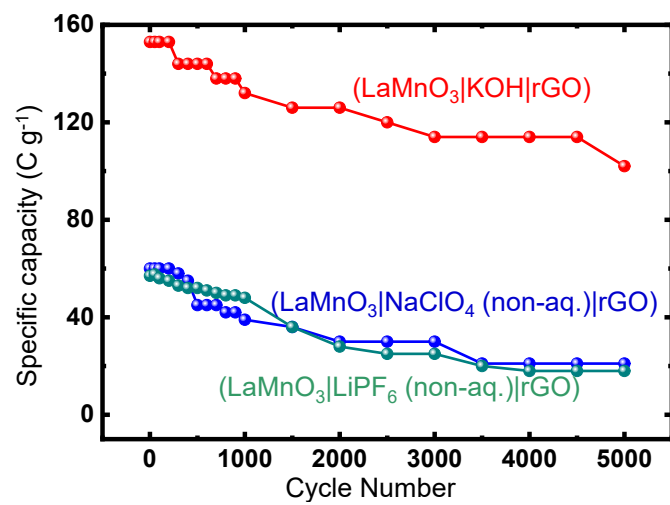


Fig. S2 Specific capacity vs. cycle number for all the three devices including the aqueous (KOH) and non-aqueous (NaClO₄ and LiPF₆) systems.

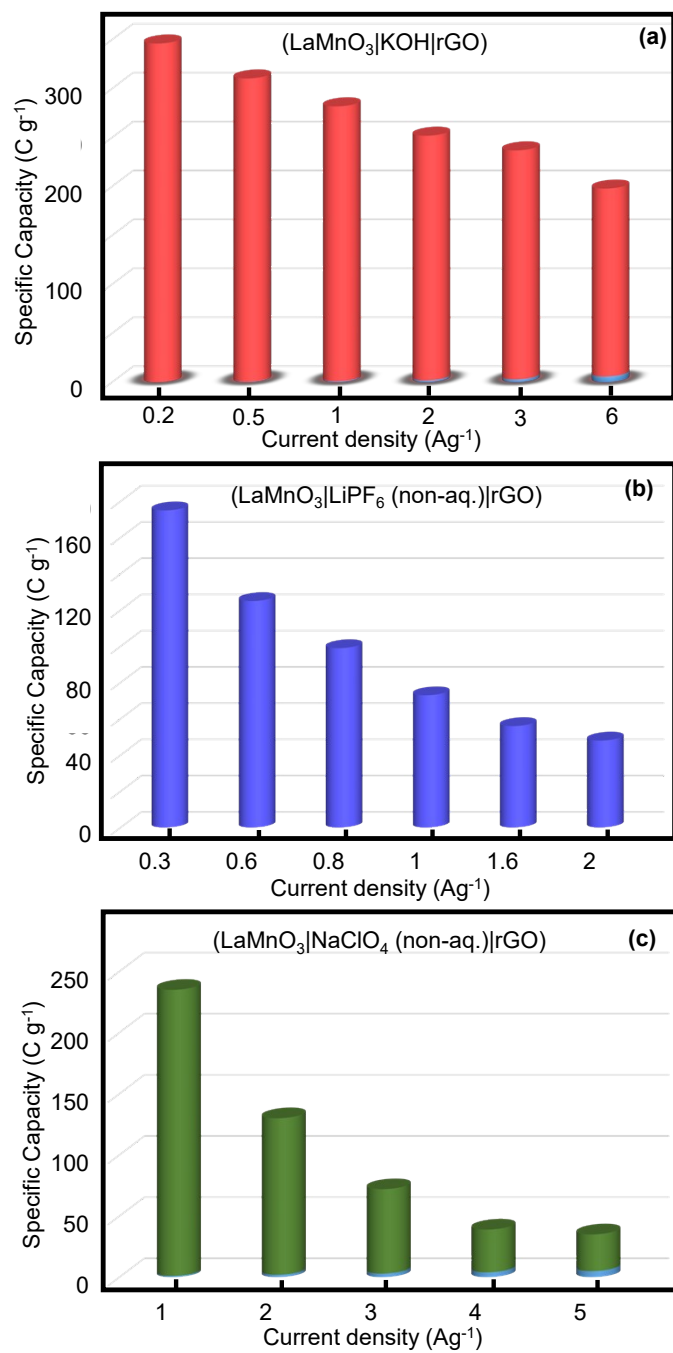


Fig. S3. Specific capacity vs. current density recorded for (a) aqueous supercapattery, (b) Lithium-ion supercapattery and (c) Sodium-ion supercapattery.

Table. S1. Mass loadings of different materials and devices fabricated.

System	Aqueous three electrode study (mg/ cm²)	Aqueous two electrode device (mg)	Non-aqueous device – LIS (mg)	Non-aqueous device – SIS (mg)
Positive Electrode (LaMnO₃)	5.29	3.12	5.94	4.25
Negative Electrode (rGO)	1.64	3.53	6.87	3.81