

Supporting Information

Hydrothermal synthesis of Eu³⁺-doped BaMoO₄ fluorescent probe for selectively detecting Fe³⁺ ions

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Table S1. The lattice constants of the obtained BaMoO₄: Eu³⁺ phosphors.

| Sample | Lattice constant | |
|--|-------------------------|--------------|
| | <i>a</i> = <i>b</i> / Å | <i>c</i> / Å |
| BaMoO ₄ : 2 mol% Eu ³⁺ | 5.6010 | 12.830 |
| BaMoO ₄ : 1.5 mol% Eu ³⁺ | 5.5974 | 12.839 |
| BaMoO ₄ : 1 mol% Eu ³⁺ | 5.6060 | 12.837 |
| BaMoO ₄ : 0.8 mol% Eu ³⁺ | 5.6094 | 12.838 |
| BaMoO ₄ : 0.5 mol% Eu ³⁺ | 5.5959 | 12.828 |
| JCPDS: 29-0193 | 5.5802 | 12.821 |

Table S2. Comparison of different methods.

| Method | Production process | Instruments and medicines | Selectivity and Sensitivity | Response |
|--|--------------------|---------------------------|-----------------------------|----------|
| Organic Fluorescent Probe ¹ | Sophisticated | Expensive and Complex | High | Slower |
| Metal-Organic frameworks ² | Complicated | Simple | High | Fast |
| Atomic Absorption Spectrometry(AAS) ³ | Cumbersome | Expensive and complex | High | Normal |
| BaMoO ₄ : Eu ³⁺ phosphor ^{our present work} | Simple | Cheap and simple | High | Fast |

1 Y. Guo, L. Wang, J. Zhuo, B. Xu, X. Li, J. Zhang, Z. Zhang, H. Chi, Y. Dong and G. Lu, A pyrene-based dual chemosensor for colorimetric detection of Cu²⁺ and fluorescent detection of Fe³⁺. *Tetrahedron Letters*, **2017**, 58 (42), 3951-3956.

2 H. Guo, N. Wu, R. Xue, H. Liu, L. Li, M. Y. Wang, W. Q. Yao, Q. Li and W. Yang, Multifunctional Ln-MOF luminescent probe displaying superior capabilities for highly selective sensing of Fe³⁺ and Al³⁺ ions and nitrotoluene. *Colloids and Surfaces A: Physicochemical and Engineering Aspects*, **2020**, 585.

3 M.Yaman and G.Kaya, Speciation of iron (II) and (III) by using solvent extraction and flame atomic absorption spectrometry. *Analytica Chimica Acta*, **2005**, 540 (1), 77-81.

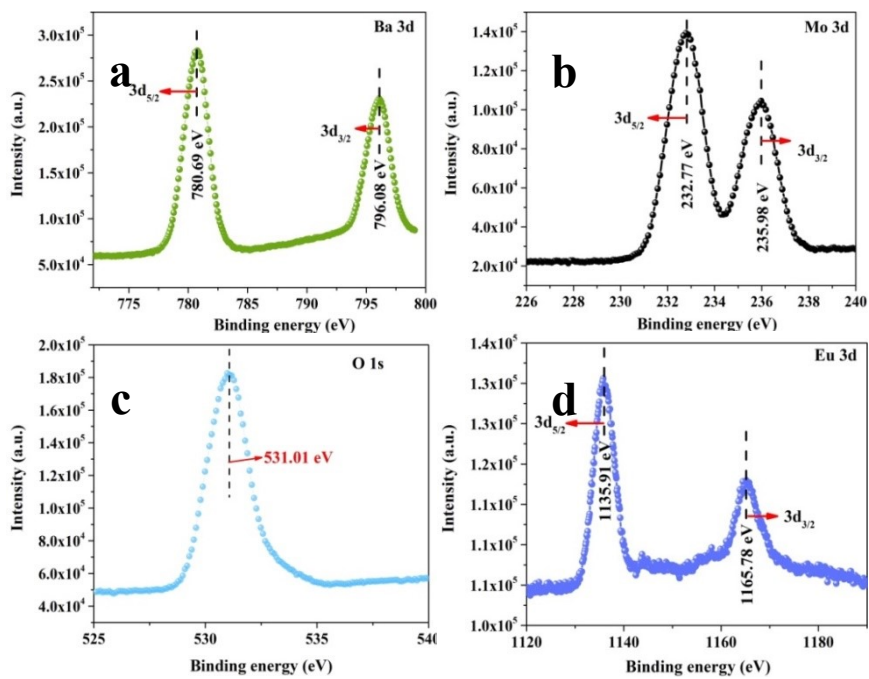


Figure S1. XPS of Ba 3d (a), Mo 3d (b), O 1s (c) and Eu 3d (d), respectively.

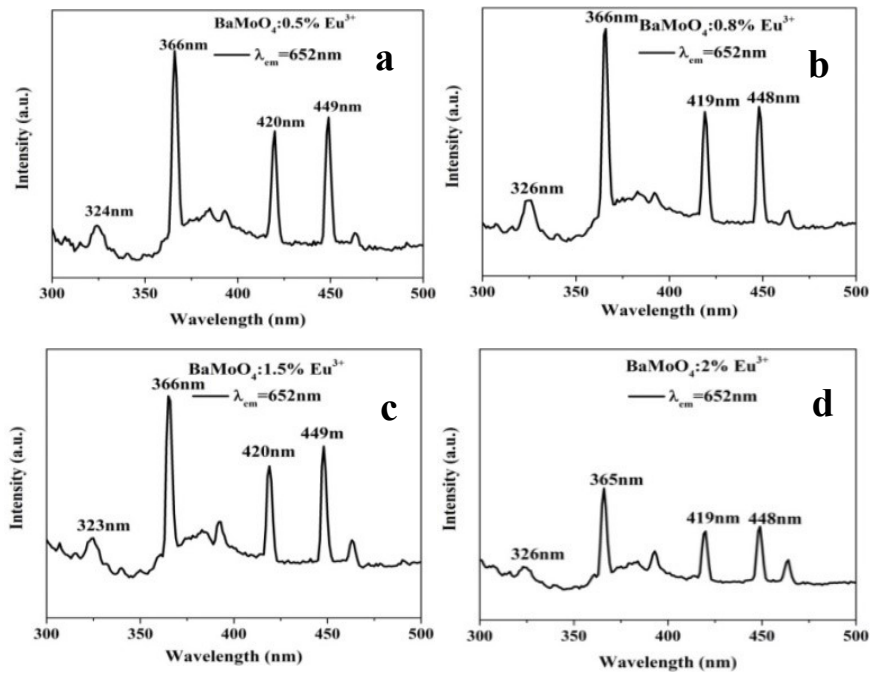


Figure S2. The excitation spectra of various concentrations of Eu^{3+} .

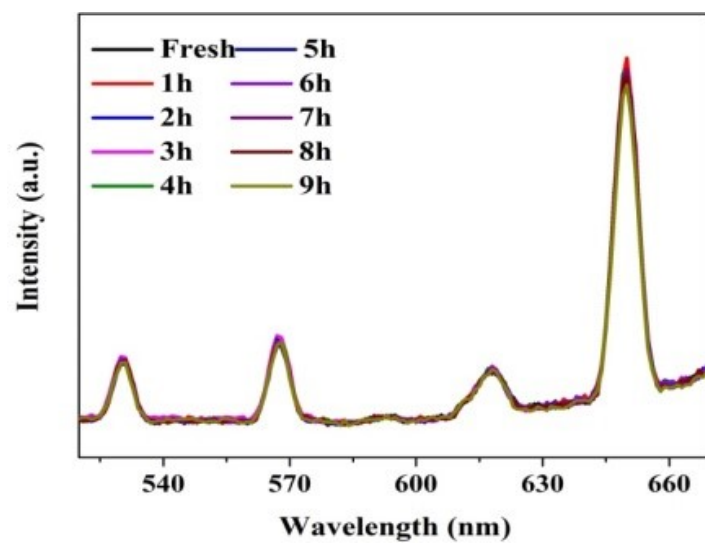


Figure S3. Hour-to-hour photostability of BaMoO₄:Eu³⁺ phosphors after immersing in aqueous solution for several hours.

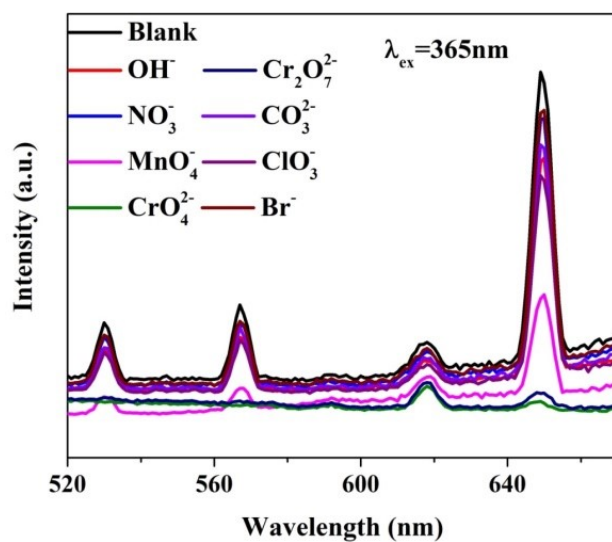


Figure S4. Emission spectra of BaMoO₄: Eu³⁺ suspensions after mixing with different anions.