

Supporting Information

Au-Pd@ZnO Alloy Nanoparticles: Promising Heterogeneous Photocatalyst toward Decarboxylative Trifluoromethylation under Visible-Light Irradiation

Zahra Bazyar^a, Mina Tavakolian^a and Mona Hosseini-Sarvari^{a*}

^a Nano Photocatalysis Lab, Department of Chemistry, Shiraz University, Shiraz 7194684795, I.R. Iran

*E-mail: hossaini@shirazu.ac.ir

Table of Contents	page
Nitrogen adsorption-desorption data of the prepared materials.....	S2
SEM image of the reused catalyst.....	S2
Experimental Data for Trifluoromethylatin.....	S2
¹ H and ¹³ C NMR spectra of the products.....	.S7

Table S1. Results of BET for Au-Pd@ZnO and ZnO

Sample	BET surface area ($\text{m}^2 \cdot \text{g}^{-1}$)	Mean pore diameter (nm)	Pore volume of pores ($\text{cm}^3 \cdot \text{g}^{-1}$)
Au-Pd@ZnO	40.3	35.0	0.43
ZnO	37.4	24.3	0.24

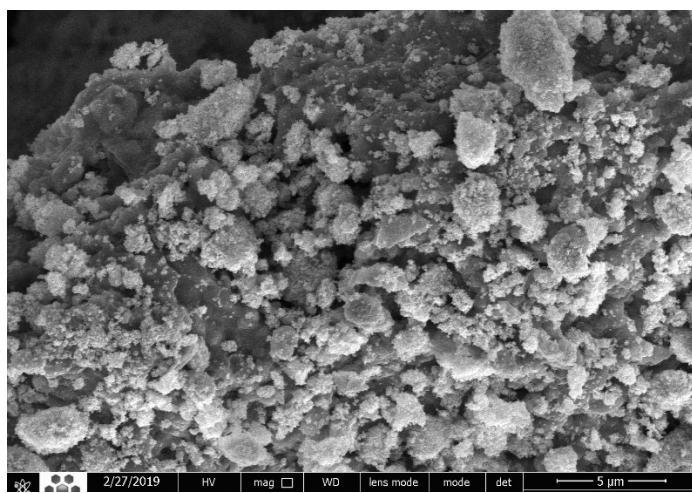
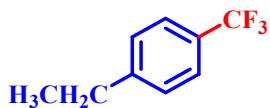


Figure S1. SEM image of the reused catalyst

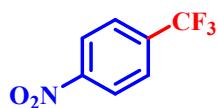
Experimental Data for Trifluoromethylation

1-Ethyl-4-(trifluoromethyl)benzene (2a) [3]



Colorless oil, TLC (Petroleumether: Ethyl acetate, 90:10 v/v): R_f = 0.72; IR (thin film) 3068, 2964, 2855, 1593, 1515, 1476, 1370, 1248, 1168, 1076 cm⁻¹; ¹H NMR (400 MHz, DMSO-d₆): δ 7.75–7.73 (d, 2H, J = 4 Hz), 7.18–7.16 (d, 2H, J = 4 Hz), 2.60–2.58 (q, 2H, J = 8 Hz), 1.19–1.15 (t, 3H, J = 8 Hz); ¹³C NMR (100 MHz, DMSO-d₆): δ 146.51, 145.67, 134.59 (q, J_{C,F} = 13 Hz), 134.24, 133.91, 133.56, 130.09 (q, J_{C,F} = 8 Hz), 126.8, 28.23, 15.47; ¹⁹F NMR (376 MHz, DMSO-d₆) δ -73.92 (s); UV/Vis (CH₃CN): λ_{max} 276 nm.

1-Nitro-4-(trifluoromethyl)benzene (2b) [3]



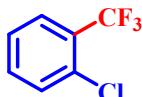
Colorless oil; TLC (Petroleumether: Ethyl acetate, 80:20 v/v): $R_f = 0.65$; IR (thin film) 3051, 1603, 1519, 1259, 1142, 1029 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3): δ 8.12–8.09 (d, 2H, $J = 12$ Hz), 6.87–6.83 (d, 2H, $J = 12$ Hz), ^{13}C NMR (100 MHz, CDCl_3): δ 138.14, 129.26 (q, $J_{\text{C}-\text{F}}^2 = 84$ Hz), 129.06 (q, $J_{\text{C}-\text{F}}^3 = 13$ Hz), 126.74 (q, $J_{\text{C}-\text{F}}^4 = 3$ Hz), 127.87 (q, $J_{\text{C}-\text{F}}^1 = 270$ Hz); ^{19}F NMR (376 MHz, CDCl_3) δ -62.99 (s); UV/Vis (CH_3CN): λ_{max} 294 nm.

1-Nitro-2-(trifluoromethyl)benzene (2c) [3]



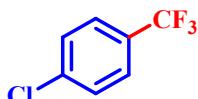
Colorless oil; TLC (Petroleumether: Ethyl acetate, 85:15 v/v): $R_f = 0.67$; IR (thin film) 3061, 1610, 1545, 1270, 1166 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3): δ 7.98–7.94 (d, 1H, $J = 10$ Hz), 7.79–7.75 (d, 1H, $J = 10$ Hz), 7.45–7.38 (t, 1H, $J = 10$ Hz), 7.23–7.16 (t, 1H, $J = 10$ Hz), ^{13}C NMR (100 MHz, CDCl_3): δ 132.88, 132.33, (d, $J_{\text{C}-\text{F}}^4 = 2$ Hz), 131.42, 129.09 (q, $J_{\text{C}-\text{F}}^2 = 52$ Hz), 127.59 (q, $J_{\text{C}-\text{F}}^3 = 5$ Hz), 126.98 (q, $J_{\text{C}-\text{F}}^1 = 271$ Hz), 126.69; ^{19}F NMR (376 MHz, CDCl_3) δ -63.26 (s); UV/Vis (CH_3CN): λ_{max} 276 nm.

1-Chloro-2-(trifluoromethyl)benzene (2d) [3]



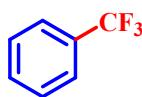
Colorless oil; TLC (Petroleumether: Ethyl acetate, 90:10 v/v): $R_f = 0.91$; IR (thin film) 3024, 1595, 1488, 1193, 1133, 1089 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3): δ 7.73–7.71 (d, 1H, $J = 8$ Hz), 7.54–7.47 (m, 2H), 7.40–7.38 (d, 1H, $J = 8$ Hz), ^{13}C NMR (100 MHz, CDCl_3): δ 144.59, 132.86, 129.18 (q, $J_{\text{C}-\text{F}}^2 = 35$ Hz), 126.53 (q, $J_{\text{C}-\text{F}}^4 = 6$ Hz), 123.73 (q, $J_{\text{C}-\text{F}}^1 = 277$ Hz), 117.48, 117.12, 113.83 (q, $J_{\text{C}-\text{F}}^3 = 29$ Hz); ^{19}F NMR (376 MHz, CDCl_3) δ -62.73 (s); UV/Vis (CH_3CN): λ_{max} 261 nm.

1-Chloro-4-(trifluoromethyl)benzene (2e) [3]



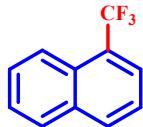
Colorless oil; TLC (Petroleumether: Ethyl acetate, 90:10 v/v): $R_f = 0.90$; IR (thin film) 3050, 1564, 1495, 1213, 1119, 1087 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3): δ 7.47–7.45 (d, 2H, $J = 8$ Hz), 7.36–7.34 (d, 2H, $J = 8$ Hz), ^{13}C NMR (100 MHz, CDCl_3): δ 138.13 (q, $J_{\text{C}-\text{F}}^4 = 1$ Hz), 129.57, (q, $J_{\text{C}-\text{F}}^2 = 33$ Hz), 129.08, 127.83 (q, $J_{\text{C}-\text{F}}^1 = 269$ Hz), 126.76 (q, $J_{\text{C}-\text{F}}^3 = 4$ Hz), ^{19}F NMR (376 MHz, CDCl_3) δ -63.26 (s); UV/Vis (CH_3CN): λ_{max} 267 nm.

(Trifluoromethyl)benzene (2f) [3]



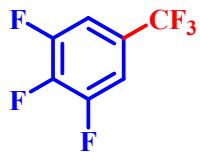
Clear colorless liquid; TLC (Petroleumether: Ethyl acetate, 90:10 v/v): $R_f = 0.25$; IR (thin film) 3039, 1550, 1229, 1138, cm^{-1} ; ^1H NMR (400 MHz, CDCl_3): δ 7.21-7.18 (t, 1H), 7.10-7.05 (t, 2H), 6.76-6.73 (d, 2H), ^{13}C NMR (100 MHz, $\text{DMSO}-d_6$): δ 138.35 (q, $J_{\text{C-F}}^2 = 11$ Hz), 134.42 (q, $J_{\text{C-F}}^1 = 203$ Hz), 133.40, 129.97, 127.38 (q, $J_{\text{C-F}}^3 = 7$ Hz), ^{19}F NMR (376 MHz, $\text{DMSO}-d_6$) δ -74.09 (s); UV/Vis (CH_3CN): λ_{max} 273 nm.

1-(Trifluoromethyl)naphthalene (2g) [3]



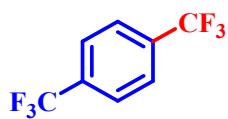
Colorless oil, TLC (Petroleumether: Ethyl acetate, 90:10 v/v): $R_f = 0.51$; IR (thin film) 3072, 1603, 1588, 1489, 1275, 1190, 1029 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3): δ 7.9 (d, 2H), 7.53 (d, 1H), 7.44 (t, 2H), 7.25 (t, 2H), ^{13}C NMR (100 MHz, CDCl_3): δ 136.12, 133.75, 133.02, 132.83, 131.73, 129.61, 128.78, 128.07, 125.58, 125.15, 124.92, 124.38; ^{19}F NMR (376 MHz, CDCl_3) δ -62.72 (s); UV/Vis (CH_3CN): λ_{max} 270 nm.

1,2,3-Trifluoro-5-(trifluoromethyl)benzene (2h) [3]



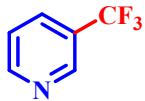
Colorless oil, TLC (Petroleumether: Ethyl acetate, 80:20 v/v): $R_f = 0.24$; IR (thin film) 3054, 1610, 1547, 1498, 1269, 1137, 1047, cm^{-1} ; ^1H NMR (400 MHz, $\text{DMSO}-d_6$): δ 7.75-7.45 (m, 2H); ^{13}C NMR (100 MHz, $\text{DMSO}-d_6$): δ 151.20 (q, $J = 9$ Hz), 148.89 (q, $J = 9$ Hz), 141.21 (q, $J = 15$ Hz), 138.71 (q, $J = 15$ Hz), 135.38, 131.42, 128.98 (q, $J = 31$ Hz), 125.95 (q, $J = 11$ Hz), 117.81 (d, $J = 17$ Hz), 116.95 (d, $J = 17$ Hz); ^{19}F NMR (376 MHz, $\text{DMSO}-d_6$) δ -164.96 (s), -141.23 (s), -67.42 (s); UV/Vis (CH_3CN): λ_{max} 277 nm.

1,4-Bis(trifluoromethyl)benzene (2i) [3]



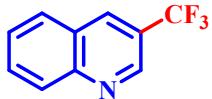
Colorless oil, TLC (Petroleumether: Ethyl acetate, 80:20 v/v): $R_f = 0.26$; IR (thin film) 3065, 1549, 1517, 1472, 1338, 1129, 1058, cm^{-1} ; ^1H NMR (400 MHz, $\text{DMSO}-d_6$): δ 8.02-7.89 (d, 4H); ^{13}C NMR (100 MHz, $\text{DMSO}-d_6$): δ 134.59 (q, $J = 180$ Hz), 130.34 (q, $J = 32$ Hz), 123.76 (q, $J = 81$ Hz); ^{19}F NMR (376 MHz, $\text{DMSO}-d_6$) δ -62.74 (d).

3-(Trifluoromethyl)pyridine (2j) [3]



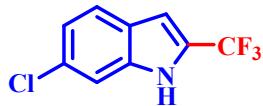
Colorless liquid, TLC (Petroleumether: Ethyl acetate, 80:20 v/v): $R_f = 0.29$; IR (thin film) 3079, 1585, 1496, 1445, 1215, 1165 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3): δ 9.01-8.99 (s, 1H), 8.53-8.50 (m, 2H), 7.80-7.72 (t, 1H), ^{19}F NMR (376 MHz, CDCl_3) δ -75.02 (s); UV/Vis (CH_3CN): λ_{\max} 268 nm.

3-(Trifluoromethyl)quinolone (2k) [3]



Clear oil, TLC (Petroleumether: Ethyl acetate, 70:30 v/v): $R_f = 0.34$; IR (thin film) 3043, 1607, 1562, 1523, 1458, 1343, 1249, 1194, 1104 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3): δ 9.10-9.08 (m, 1H), 8.25-8.21 (d, 1H), 8.08-8.07 (d, 1H), 8.04-8.01 (d, 1H), 7.63-7.55 (t, 1H), 7.53-7.50 (t, 1H); ^{13}C NMR (100 MHz, CDCl_3): δ 151.28, 136.28, 132.43, 128.72, 130.26 (q, $J_{\text{C}-\text{F}} = 162$ Hz), 128.10 (q, $J_{\text{C}-\text{F}} = 6$ Hz), 125.21, 121.95; ^{19}F NMR (376 MHz, CDCl_3) δ -59.71 (s); UV/Vis (CH_3CN): λ_{\max} 306 nm.

6-Chloro-2-(trifluoromethyl)-1H-indole (2l) [3]



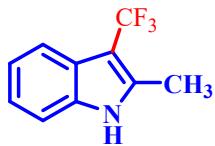
White powder; m.p: 145 °C (lit. [4] 145 °C); TLC (Petroleumether: Ethyl acetate, 85:15 v/v): $R_f = 0.13$; IR (thin film) 3411, 3063, 1676, 1542, 1465, 1344, 1289, 1236, 1126 cm^{-1} ; ^1H NMR (400 MHz, $\text{DMSO}-d_6$): δ 9.58 (s, 1H), 7.58-7.56 (d, $J = 8$ Hz, 1H), 7.50-7.48 (d, $J = 8$ Hz, 1H), 7.42-7.38 (t, 1H), 6.88-6.86 (d, 1H); ^{13}C NMR (100 MHz, $\text{DMSO}-d_6$): δ 154.69, 150.59, 145.80, 137.67 (q, $J_{\text{C}-\text{F}} = 39$ Hz), 122.11 (q, $J_{\text{C}-\text{F}} = 264$ Hz), 119.47, 116.77, 114.08, 109.19; ^{19}F NMR (376 MHz, $\text{DMSO}-d_6$) δ -61.39 (s); UV/Vis (CH_3CN): λ_{\max} 281 nm; analysis $\text{C}_9\text{H}_5\text{ClF}_3\text{N}$ (262.96): Calcd. C, 40.94; H, 1.91; N, 5.30; Found C, 40.95; H, 1.93; N, 5.28.

2-(Trifluoromethyl)-1H-indole (2m) [3]



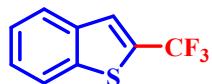
White solid, m.p: 107-108 °C (lit. [5] 107-109 °C); TLC (Petroleumether: Ethyl acetate, 70:30 v/v): $R_f = 0.28$; IR (thin film) 3409, 3053, 1618, 1509, 1428, 1383, 1337, 1265, 1141 cm^{-1} ; ^1H NMR (400 MHz, $\text{DMSO}-d_6$): δ 12.69 (s, 1H), 8.48-8.46 (s, 2H), 8.19-8.16 (d, $J = 8$ Hz, 1H), 7.60-7.58 (d, $J = 8$ Hz, 1H), 7.57-7.28 (m, 2H); ^{13}C NMR (100 MHz, $\text{DMSO}-d_6$): δ 138.11 (q, $J_{\text{C}-\text{F}} = 5$ Hz), 137.12, 126.11, 124.78, 123.87, 121.66 (q, $J_{\text{C}-\text{F}} = 298$ Hz), 121.56, 113.46, 109.31; ^{19}F NMR (376 MHz, $\text{DMSO}-d_6$) δ -71.39 (s); UV/Vis (CH_3CN): λ_{\max} 311 nm; analysis $\text{C}_9\text{H}_6\text{F}_3\text{N}$ (185.05): Calcd. C, 58.38; H, 3.27; N, 7.57; Found C, 58.37; H, 3.29; N, 7.55.

2-Methyl-3-(trifluoromethyl)-1H-indole (2n) [3]



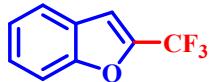
White solid, m.p: 146-147 °C (lit. [6] 147.5 °C); TLC (Petroleumether: Ethyl acetate, 85:15 v/v): $R_f = 0.17$; IR (thin film) 3405, 3058, 2934, 2867, 1644, 1599, 1532, 1458, 1357, 1278, 1158, 1035, cm^{-1} ; ^1H NMR (400 MHz, CDCl_3): δ 8.9 (s, 1H), 8.04-8.00 (d, 1H), 7.32-7.29 (d, 1H), 7.28 (t, $J = 12$ Hz, 1H), 7.26-7.18 (t, $J = 12$ Hz, 1H), 2.72 (s, 3H); ^{13}C NMR (100 MHz, CDCl_3): δ 147.34, 132.60, 123.99, 121.61, 121.27, 119.35 (q, $J^{\text{C-F}} = 284$ Hz), 119.03 (q, $J_{\text{C-F}} = 32$ Hz), 116.51, 109.18, 106.16, 13.74; ^{19}F NMR (376 MHz, CDCl_3) δ -74.33 (s); UV/Vis (CH_3CN): λ_{max} 271 nm; analysis $\text{C}_{10}\text{H}_8\text{F}_3\text{N}$ (199.06) : Calcd. C, 60.30; H, 4.05; N, 7.03; Found C, 60.31; H, 4.03; N, 7.04.

2-(Trifluoromethyl)benzo[b]thiophene (2o) [3]



Colorless oil; TLC (Petroleumether: Ethyl acetate, 80:20 v/v): $R_f = 0.39$; IR (thin film) 3079, 1521, 1458, 1284, 1171 cm^{-1} ; ^1H NMR (400 MHz, DMSO-d_6): δ 8.64 (s, 1H), 7.98-7.89 (m, 2H), 7.37-7.35 (m, 2H); ^{13}C NMR (100 MHz, DMSO-d_6): δ 142.64, 140.48, 139.36 (q, $J^2_{\text{C-F}} = 31$ Hz), 137.37 (q, $J^4_{\text{C-F}} = 5$ Hz), 132.70, 127.23 (q, $J^1_{\text{C-F}} = 304$ Hz), 124.84, 124.01 (d, $J^3_{\text{C-F}} = 6$ Hz), 122.39; ^{19}F NMR (376 MHz, DMSO-d_6) δ -73.78 (s); UV/Vis (CH_3CN): λ_{max} 330 nm.

2-(Trifluoromethyl)benzofuran (2p) [3]



Colorless oil; TLC (Petroleumether: Ethyl acetate, 80:20 v/v): $R_f = 0.39$; IR (thin film) 3058, 1575, 1449, 1308, 1280, 1127, 1126, 1071 cm^{-1} ; ^1H NMR (400 MHz, DMSO-d_6): δ 8.3-8.21 (d, H), 7.67-7.65 (d, H), 7.57-7.54 (d, H), 7.45-7.37 (m, 2H); ^{13}C NMR (100 MHz, DMSO-d_6): δ 156.42, 127.64, 125.25, 124.18 (q, $J^2_{\text{C-F}} = 138$ Hz), 122.48, 121.74, 117.41, 111.33, 111.13 (q, $J^1_{\text{C-F}} = 270$ Hz), ^{19}F NMR (376 MHz, DMSO-d_6) δ -73.74 (s).

NMR (^1H NMR, ^{13}C NMR, ^{19}F NMR purified products)

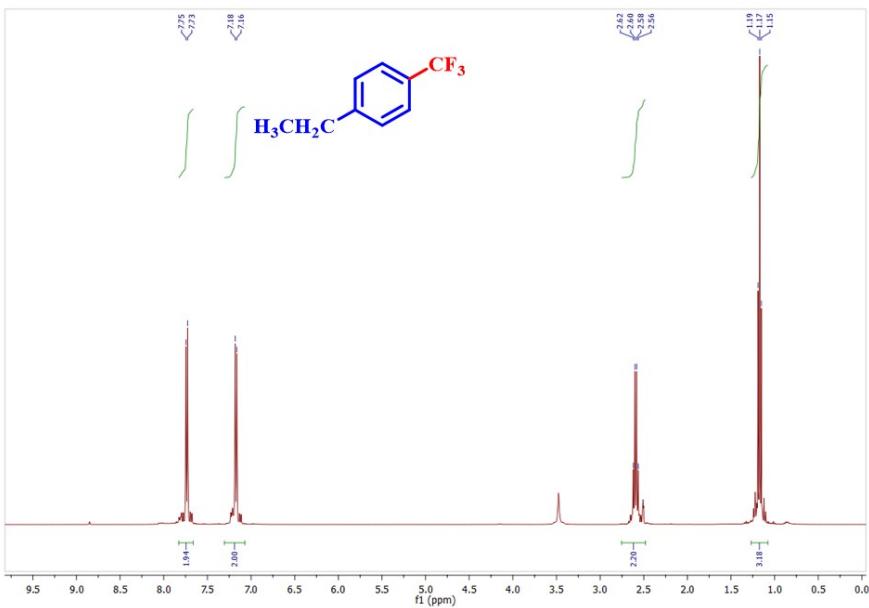


Figure S2. ^1H NMR spectrum of 1-ethyl-4-(trifluoromethyl)benzene [3].

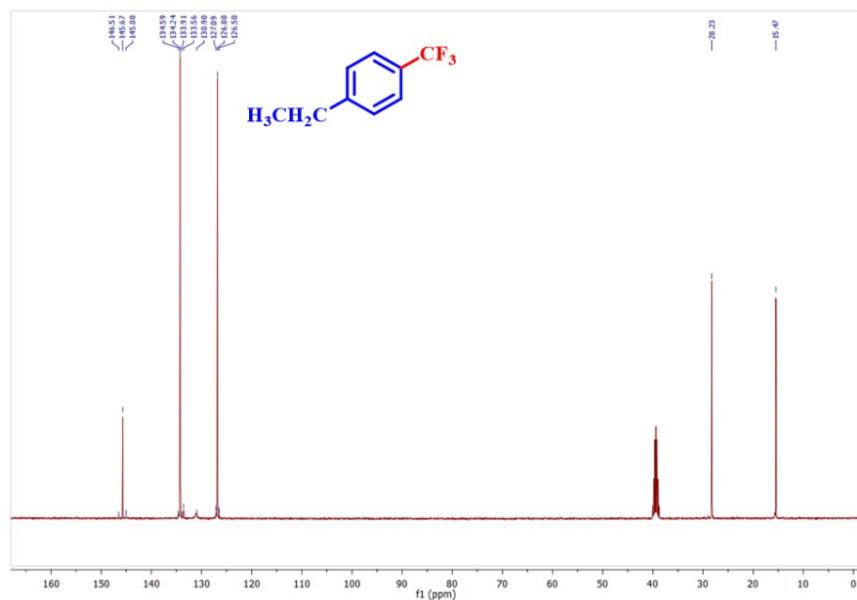


Figure S3. ^{13}C NMR spectrum of 1-ethyl-4-(trifluoromethyl)benzene [3].

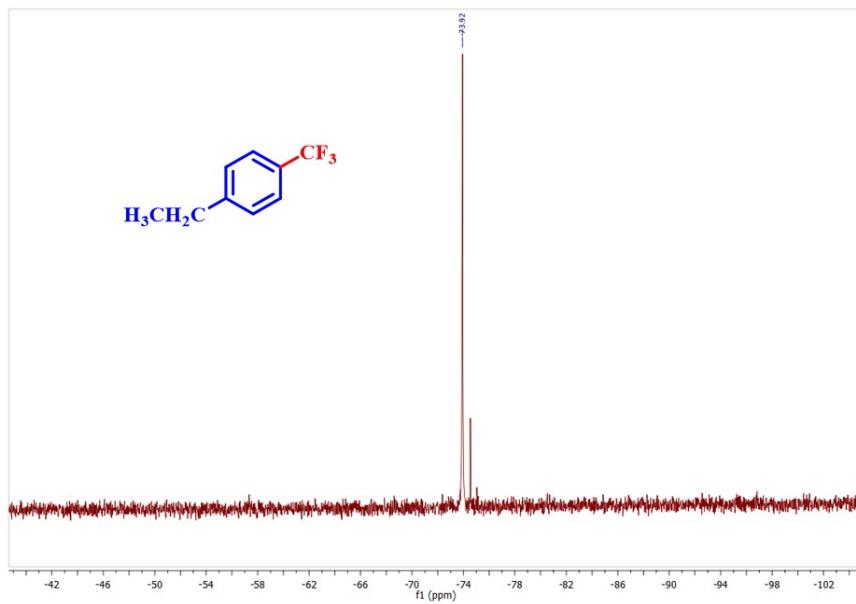


Figure S4. ^{19}F NMR spectrum of 1-ethyl-4-(trifluoromethyl)benzene [3].

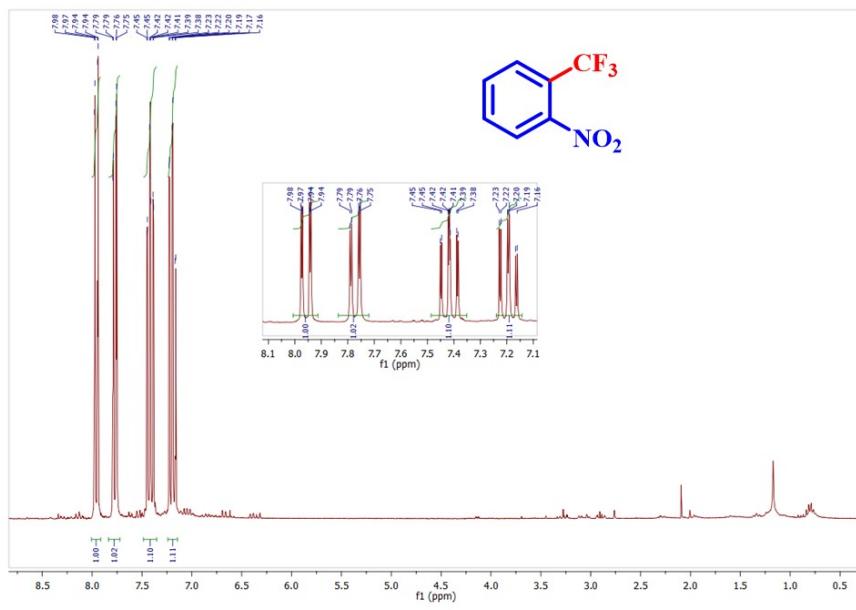


Figure S5. ^1H NMR spectrum of 1-nitro-2-(trifluoromethyl)benzene [3].

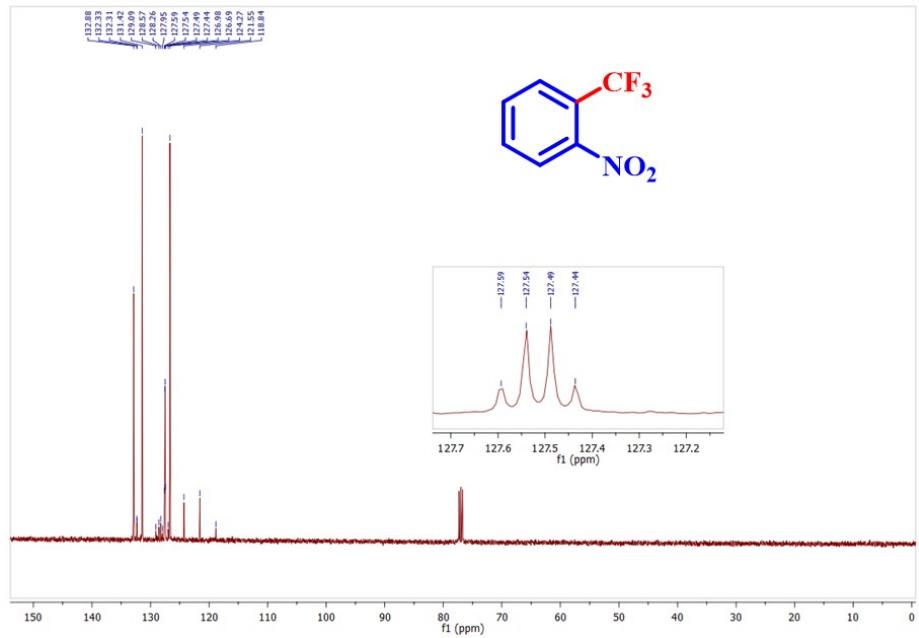


Figure S6.¹³C NMR spectrum of 1-nitro-2-(trifluoromethyl)benzene [3].

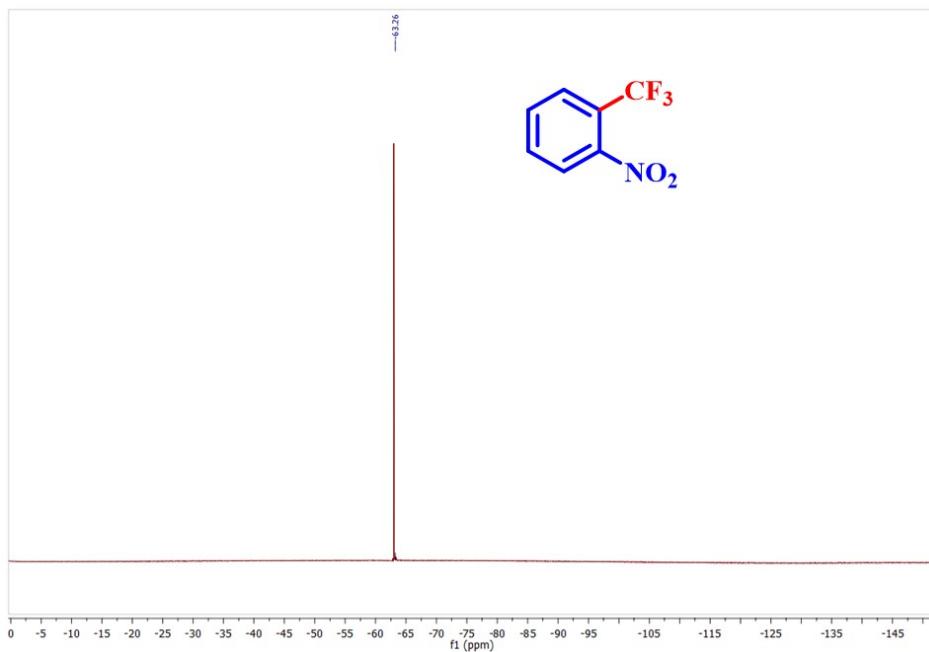


Figure S7. ^{19}F NMR spectrum of 1-nitro-2-(trifluoromethyl)benzene [3].

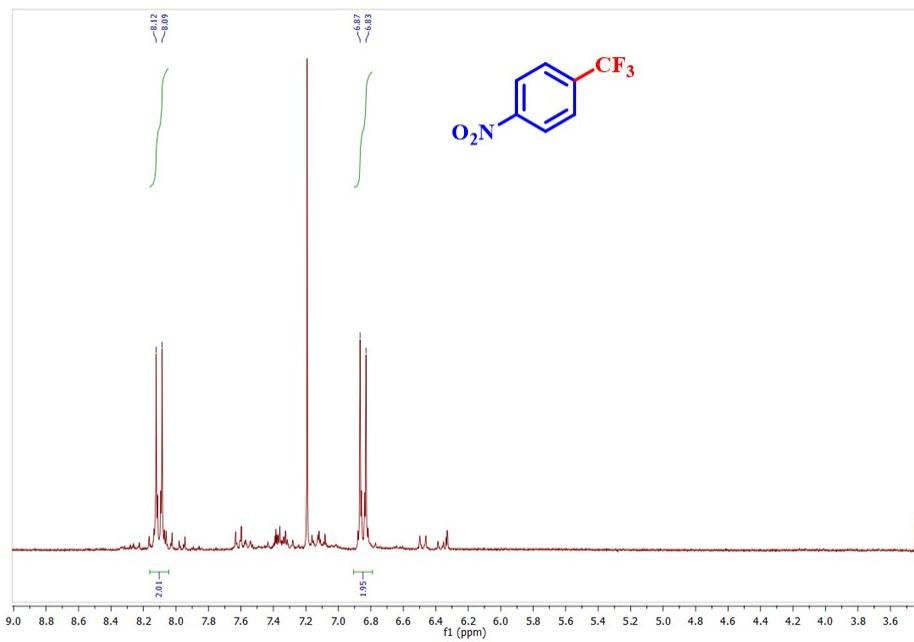


Figure S8. ¹H NMR spectrum of 1-nitro-4-(trifluoromethyl)benzene [3].

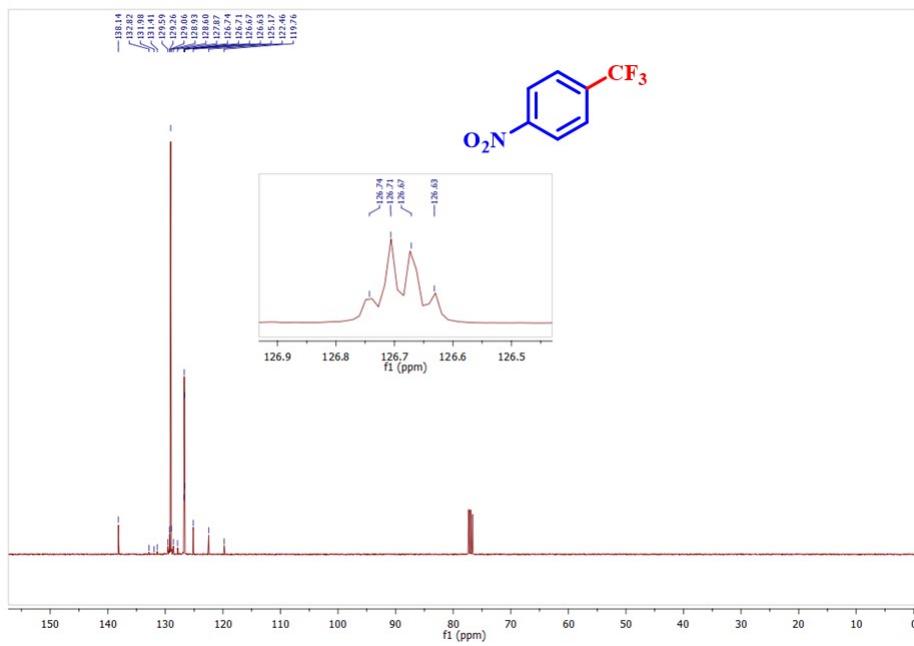


Figure S9. ¹³C NMR spectrum of 1-nitro-4-(trifluoromethyl)benzene [3].

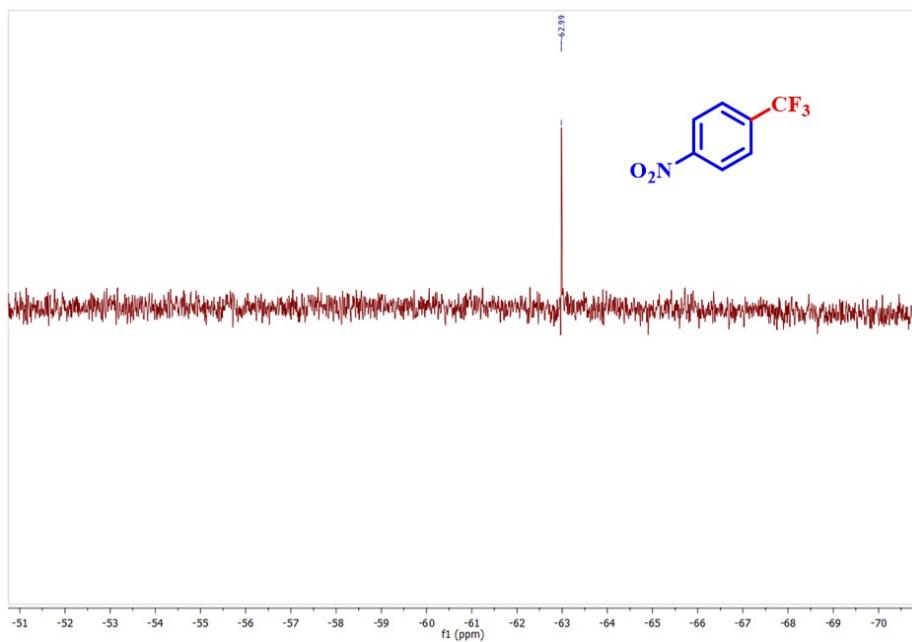


Figure S10. ¹⁹F NMR spectrum of 1-nitro-4-(trifluoromethyl)benzene [3].

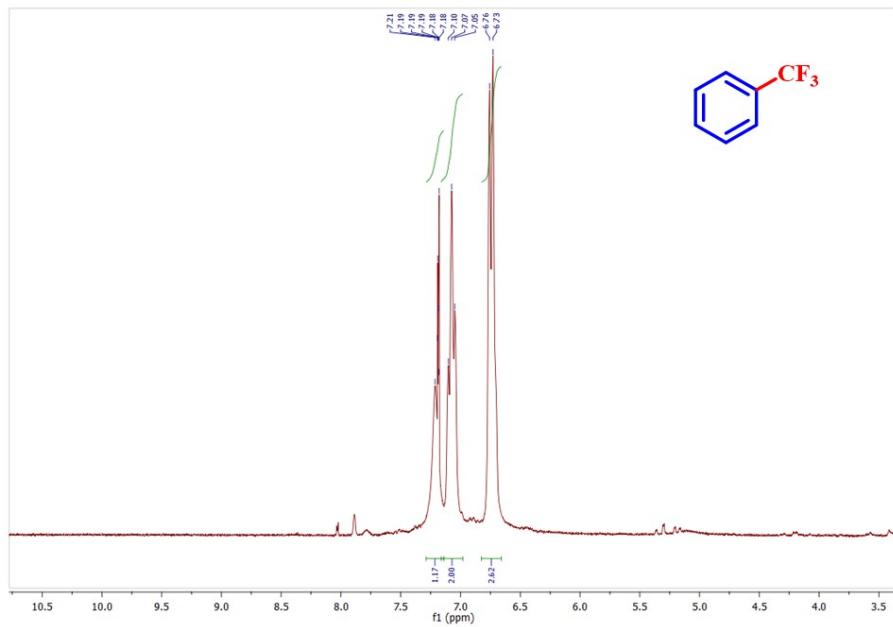


Figure S11. ¹H NMR spectrum of (trifluoromethyl)benzene [3].

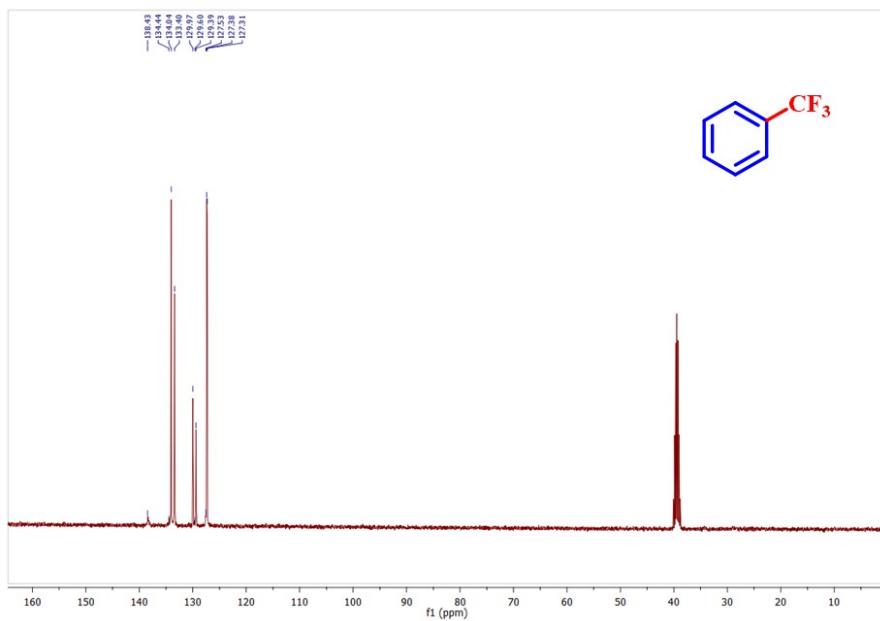


Figure S12. ^{13}C NMR spectrum of (trifluoromethyl)benzene [3].

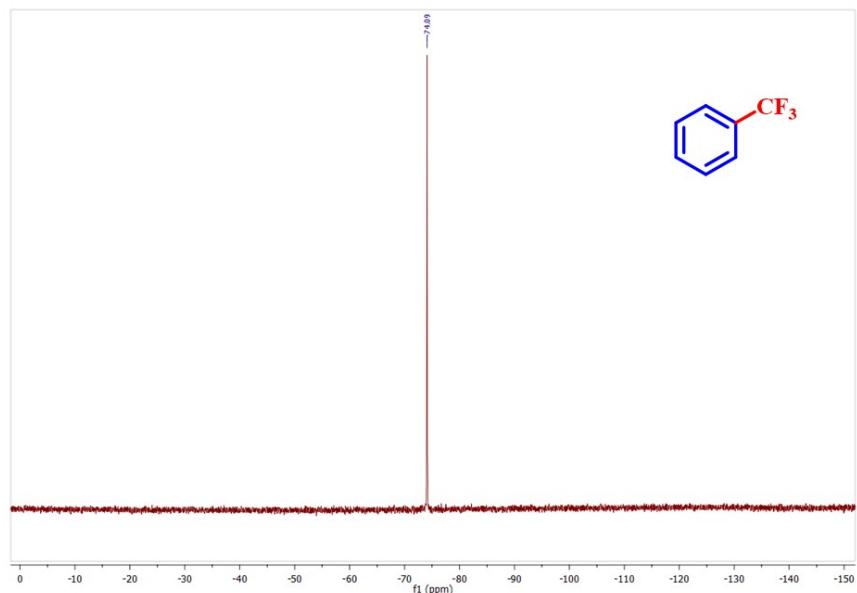


Figure S13. ^1F NMR spectrum of (trifluoromethyl)benzene [3].

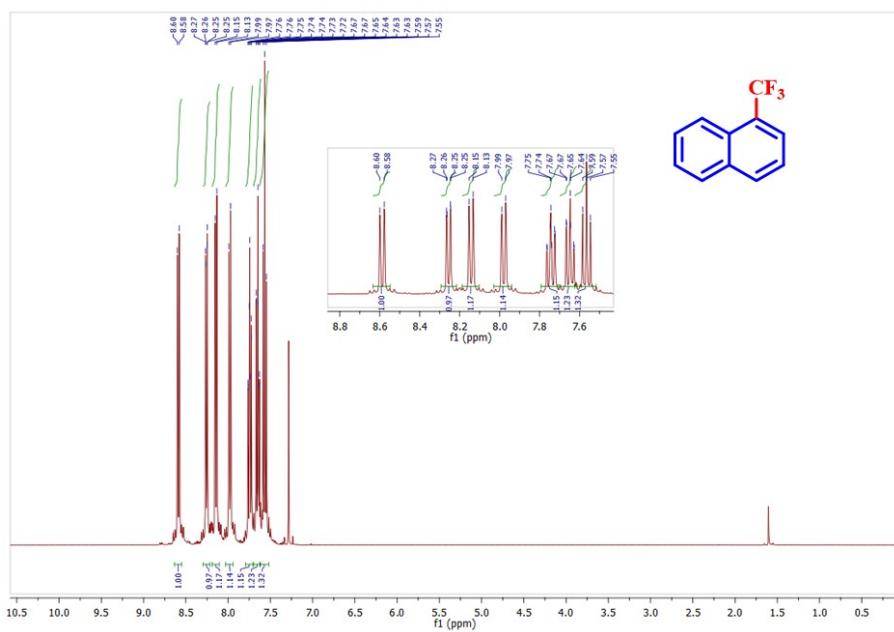


Figure S14. ¹H NMR spectrum of 1-(trifluoromethyl)naphthalene [3].

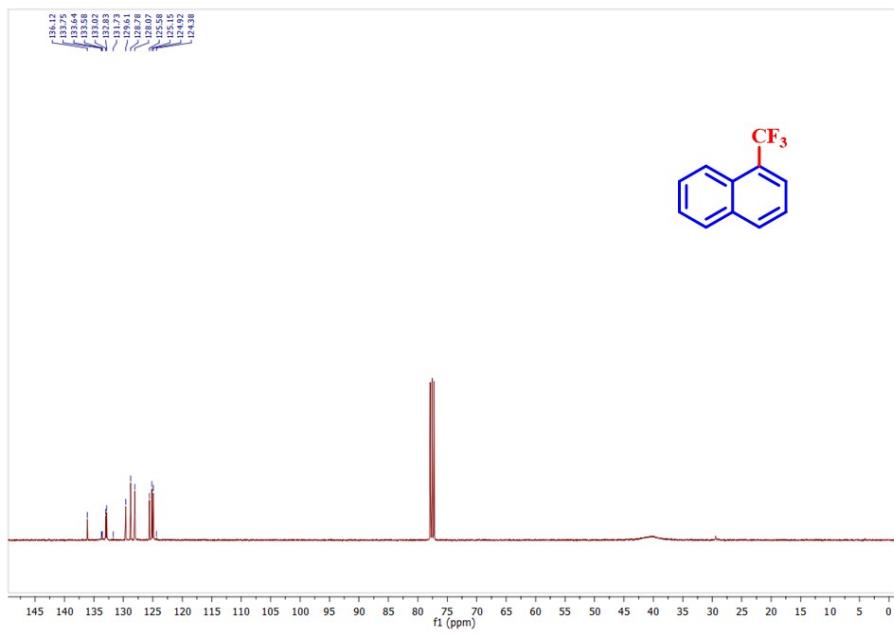


Figure S15. ¹³C NMR spectrum of 1-(trifluoromethyl)naphthalene [3].

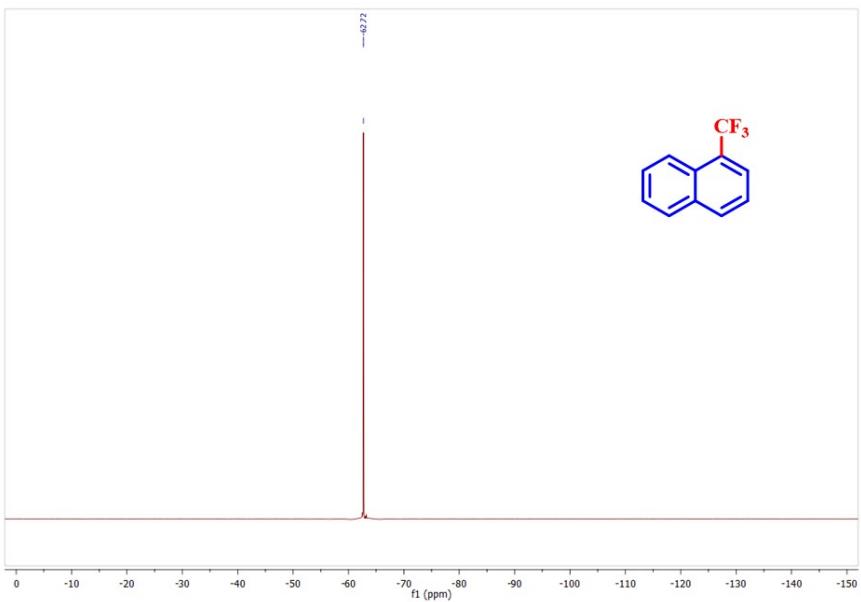


Figure S16. ^{19}F NMR spectrum of 1-(trifluoromethyl)naphthalene [3].

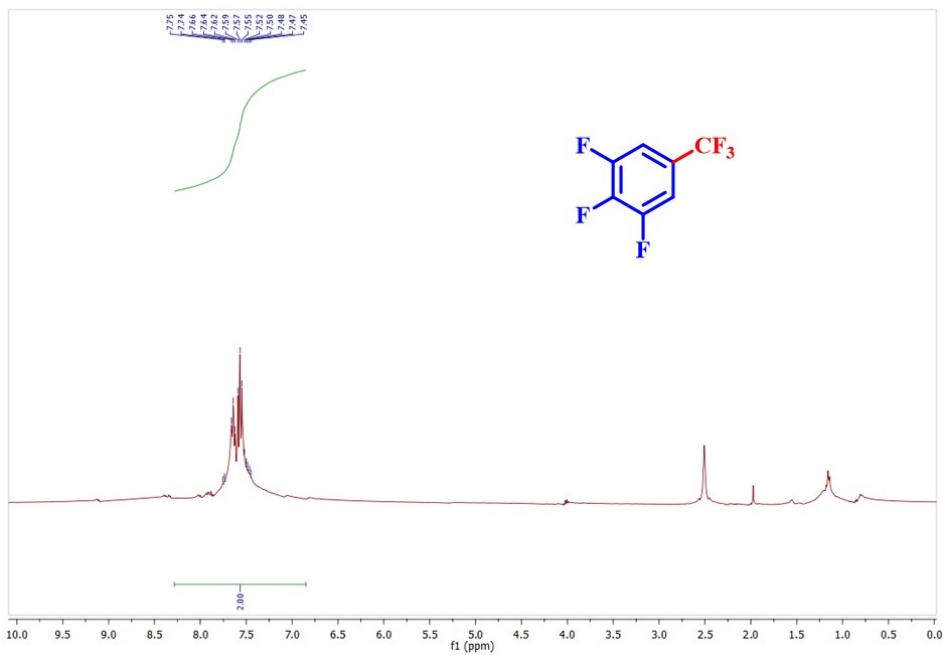


Figure S17. ^1H NMR spectrum of 1,2,3-Trifluoro-5-(trifluoromethyl)benzene [3].

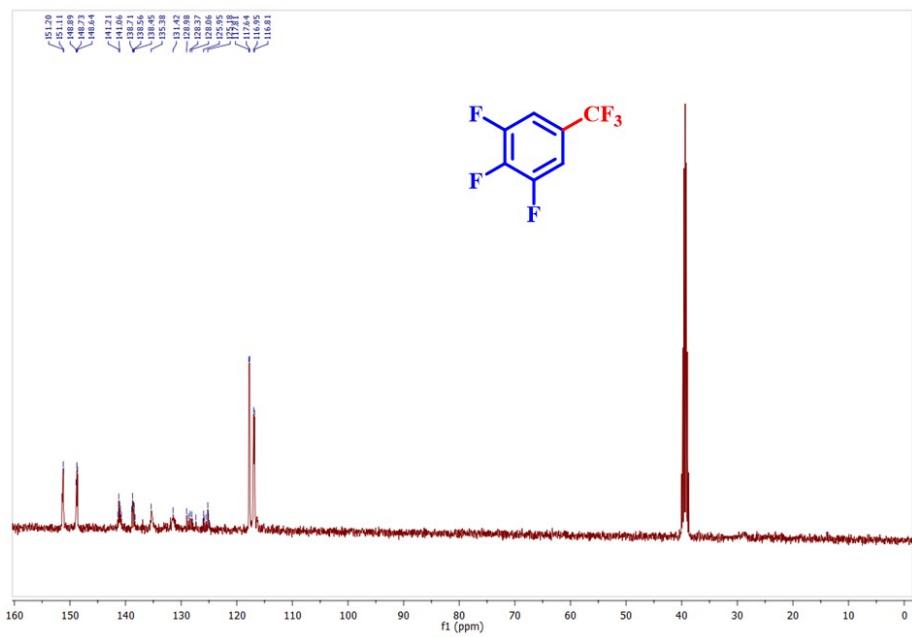


Figure S18. ¹³C NMR spectrum of 1,2,3-Trifluoro-5-(trifluoromethyl)benzene [3].

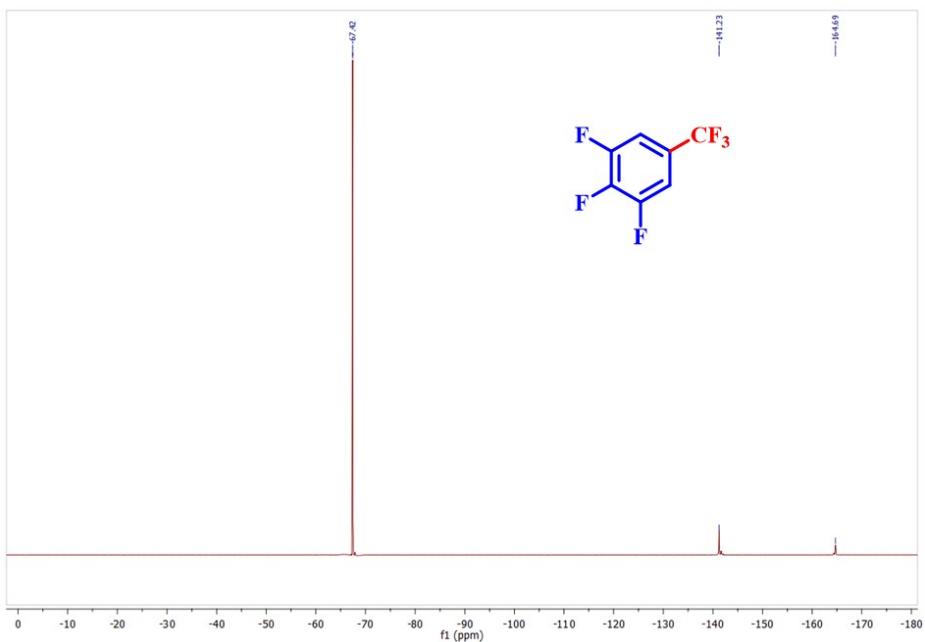


Figure S19. ¹⁹F NMR spectrum of 1,2,3-Trifluoro-5-(trifluoromethyl)benzene [3].

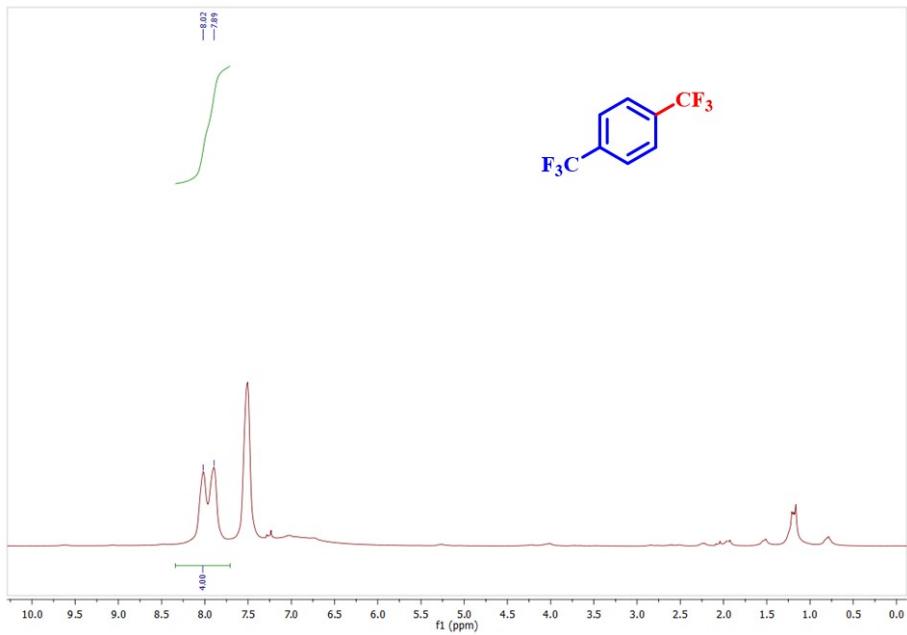


Figure S20. ¹H NMR spectrum of 1,4-bis(trifluoromethyl)benzene [3].

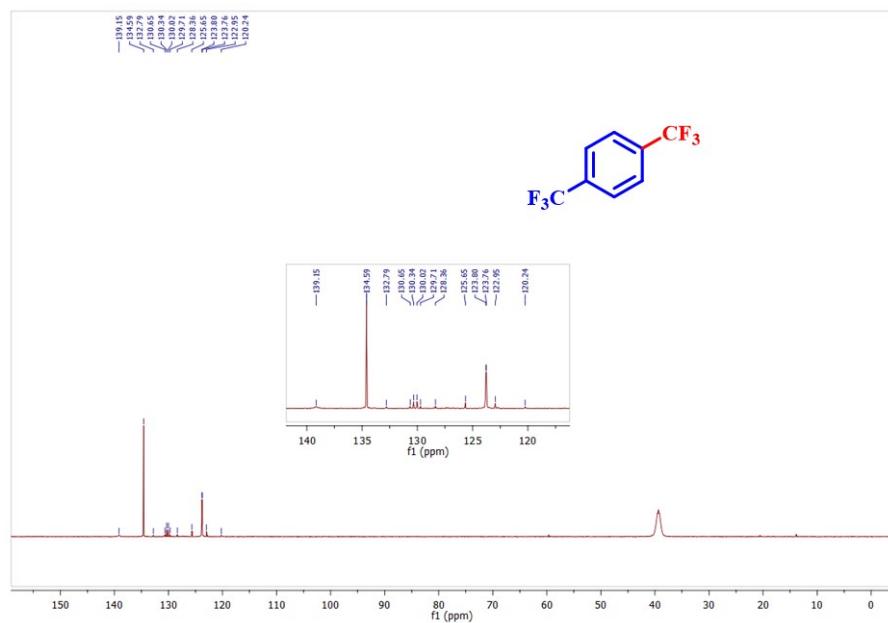


Figure S21. ¹³C NMR spectrum of 1,4-bis(trifluoromethyl)benzene [3].

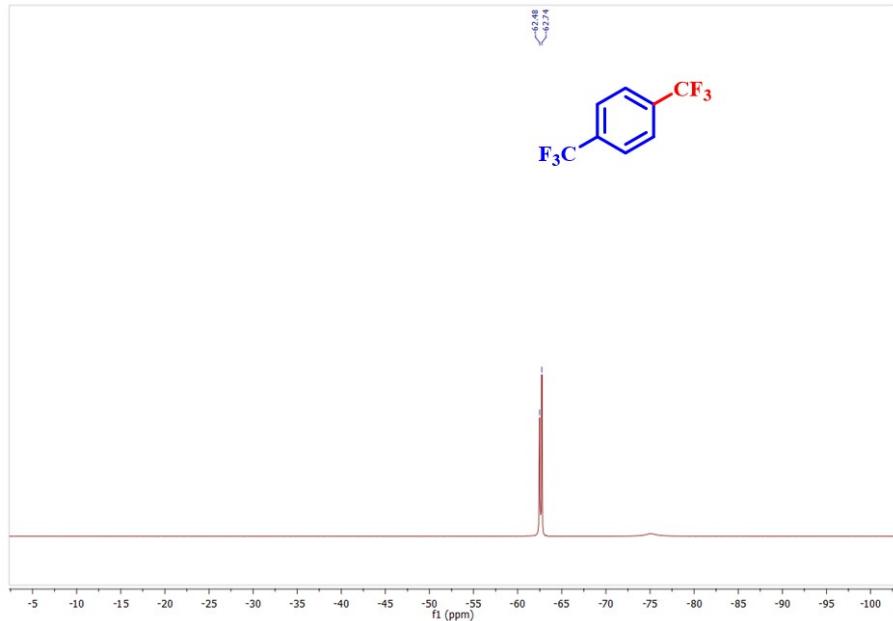


Figure S22.¹⁹F NMR spectrum of 1,4-bis(trifluoromethyl)benzene [3].

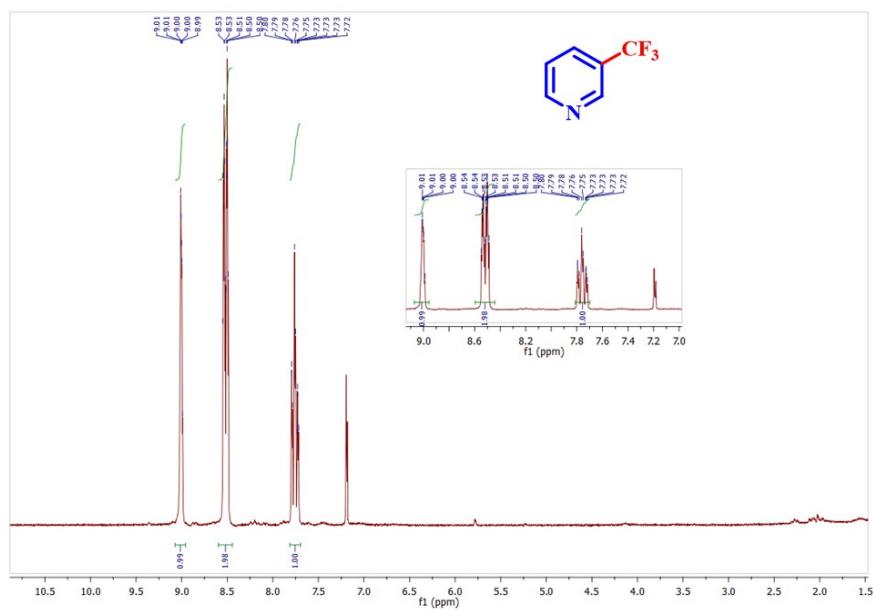


Figure S23.¹H NMR spectrum of 3-(trifluoromethyl)pyridine [3].

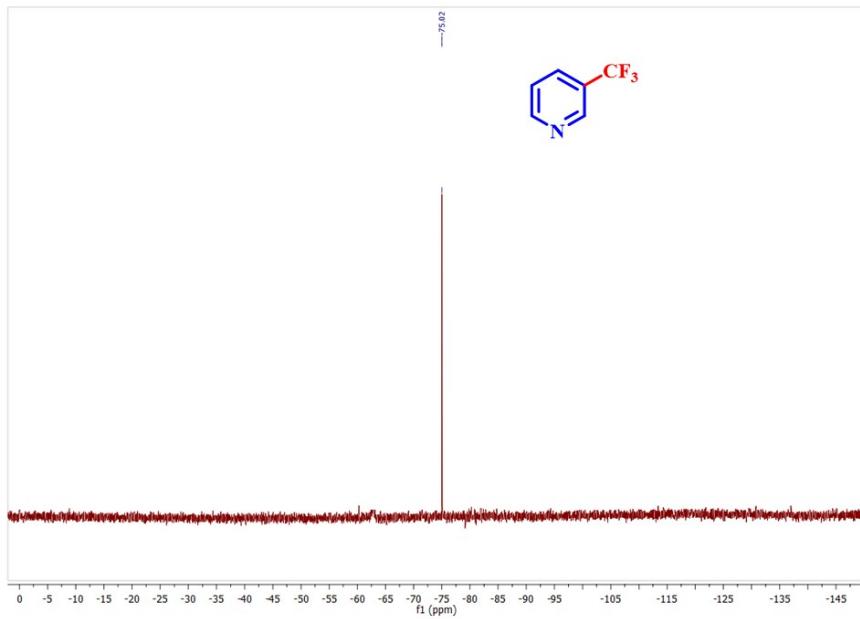


Figure S24.¹⁹F NMR spectrum of 3-(trifluoromethyl)pyridine [3].

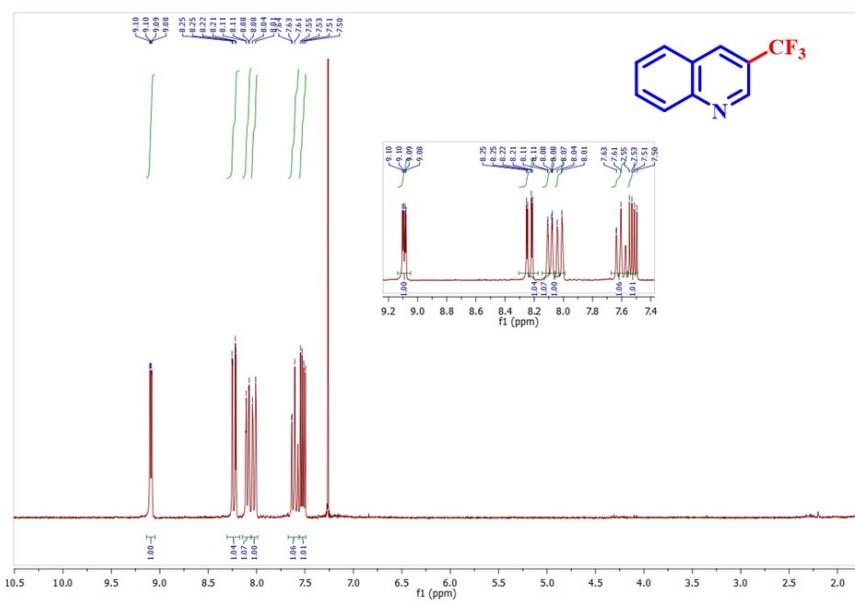


Figure S25.¹H NMR spectrum of 3-(trifluoromethyl)quinolone [3].

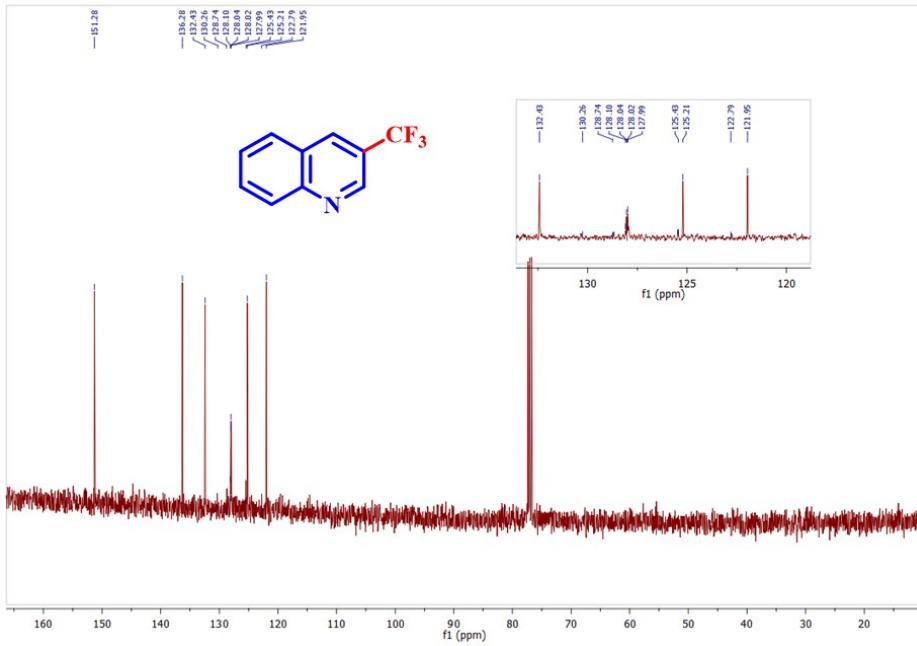


Figure S26.¹³C NMR spectrum of 3-(trifluoromethyl)quinolone [3].

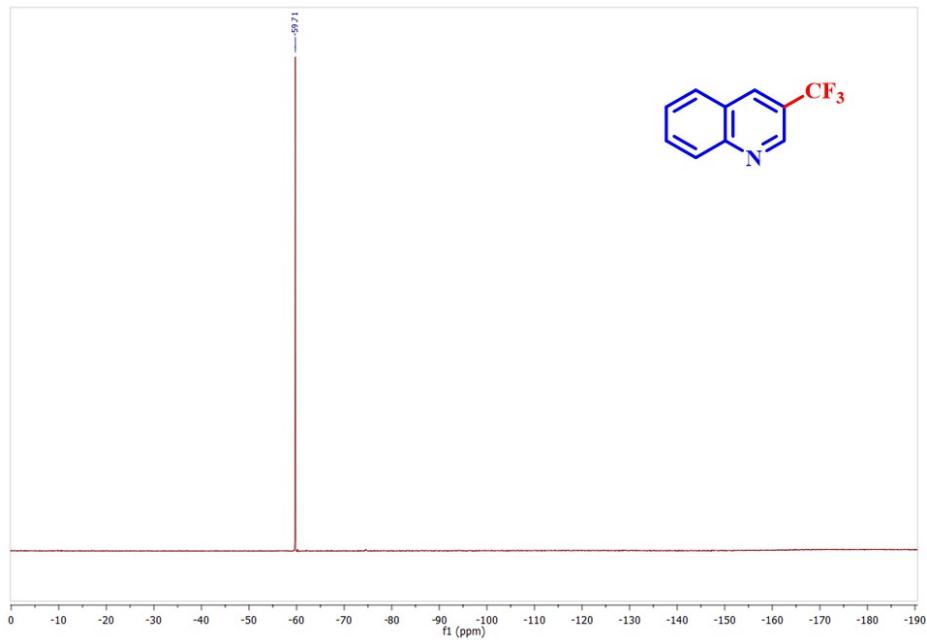


Figure S27. ^{19}F NMR spectrum of 3-(trifluoromethyl)quinolone [3].

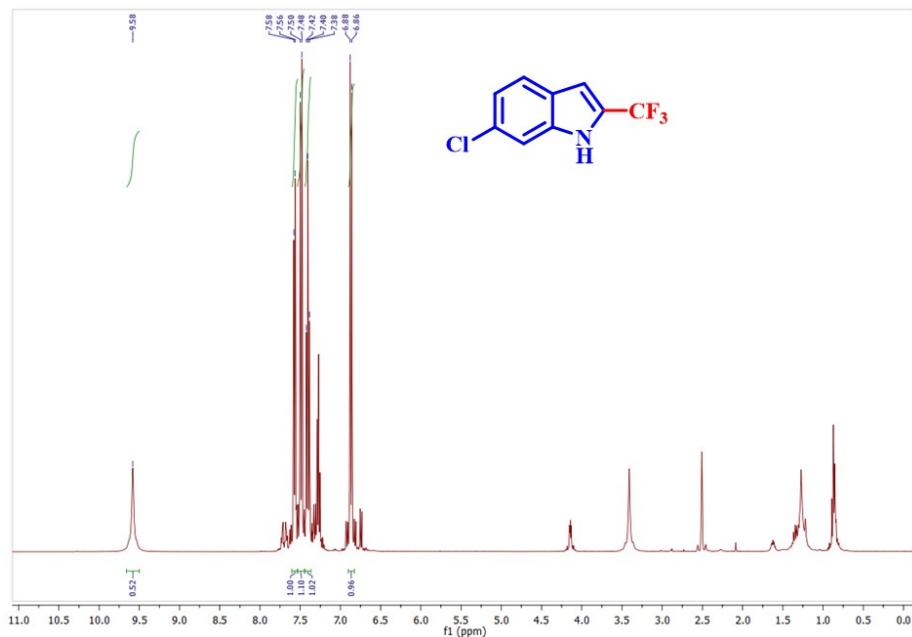


Figure S28. ^1H NMR spectrum of 6-Chloro-2-(trifluoromethyl)-1H-indole [3].

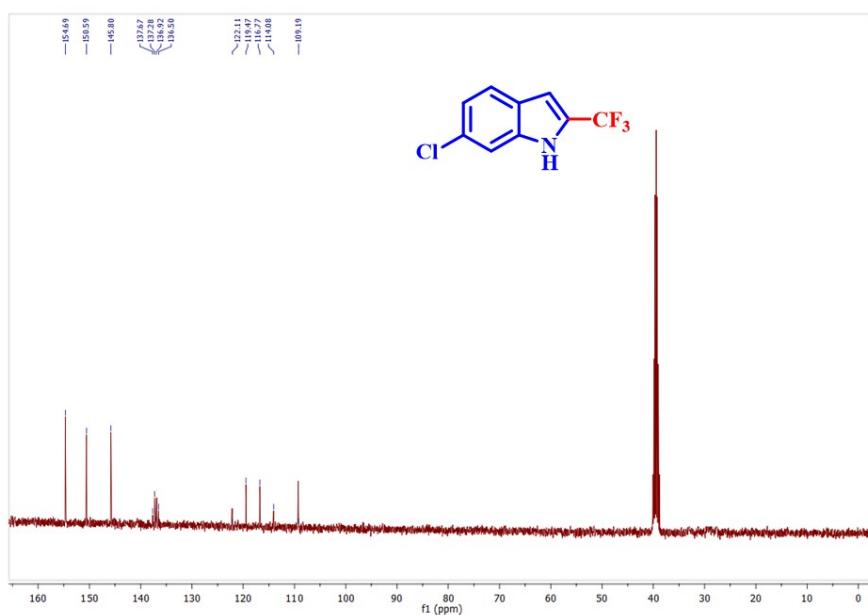


Figure S29.¹³C NMR spectrum of 6-Chloro-2-(trifluoromethyl)-1H-indole [3].

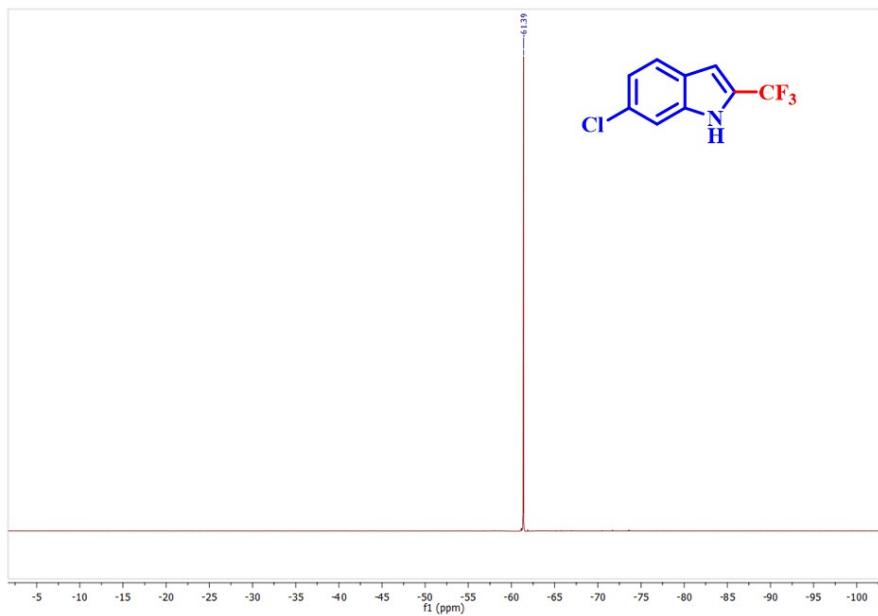


Figure S30. ^{19}F NMR spectrum of 6-Chloro-2-(trifluoromethyl)-1H-indole [3].

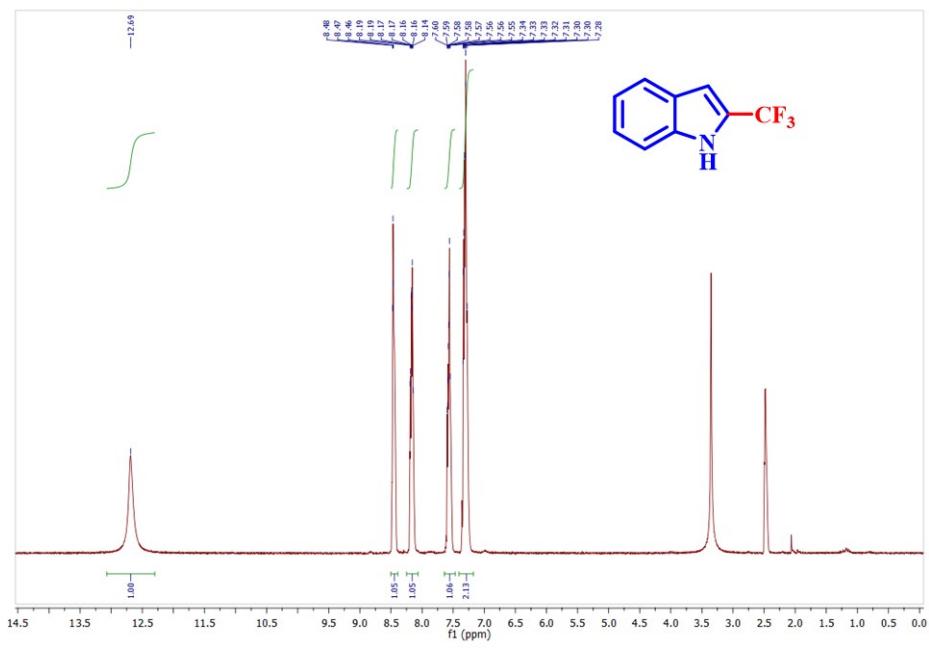


Figure S31. ^1H NMR spectrum of 2-(trifluoromethyl)-1H-indole [3].

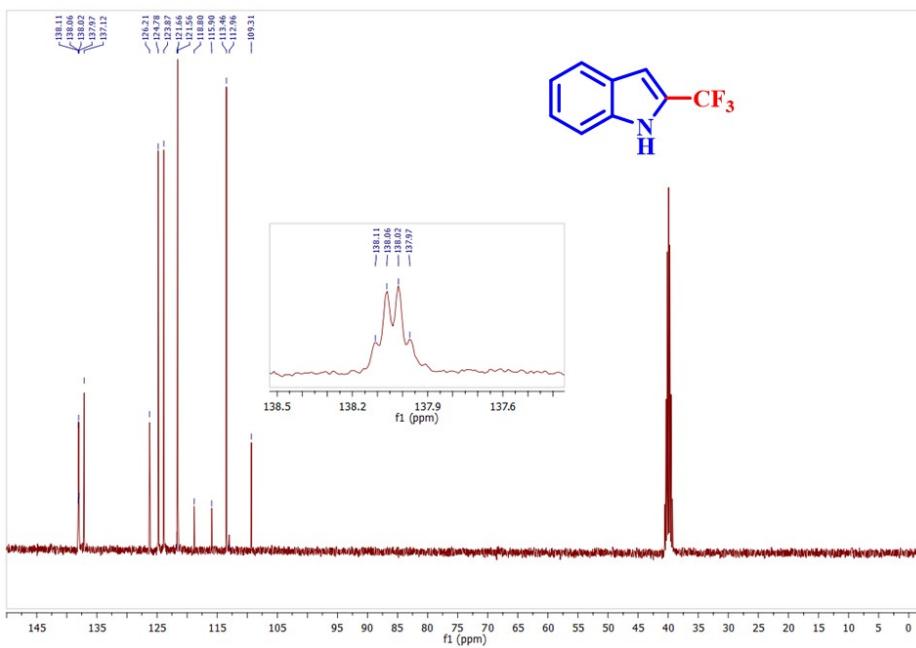


Figure S32. ^{13}C NMR spectrum of 2-(trifluoromethyl)-1H-indole [3].

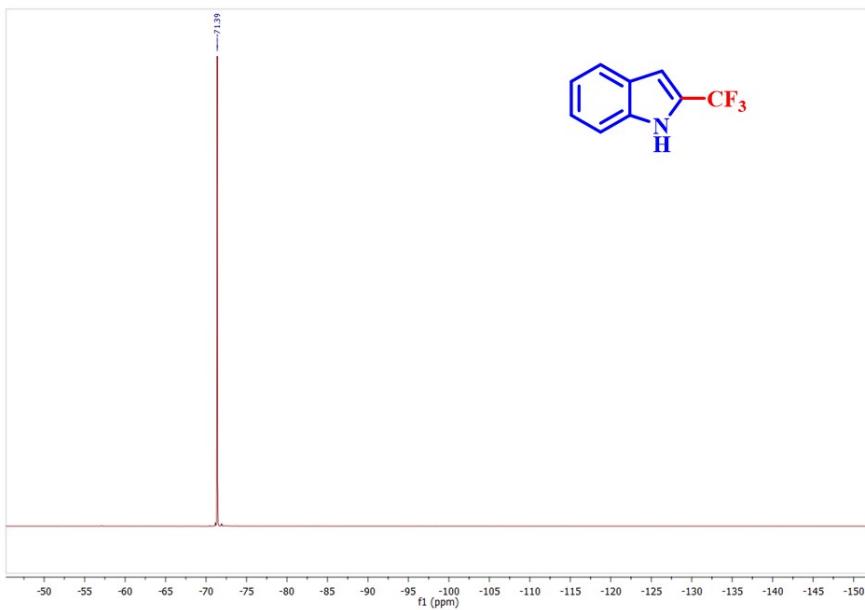


Figure S33. ^{19}F NMR spectrum of 2-(trifluoromethyl)-1H-indole [3].

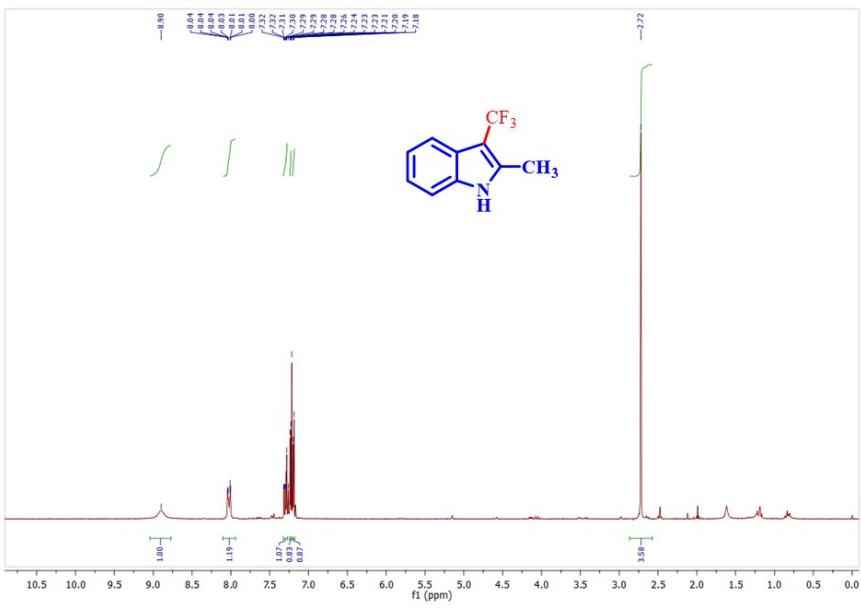


Figure S34. ¹H NMR spectrum of 2-Methyl-3-(trifluoromethyl)-1H-indole [3].

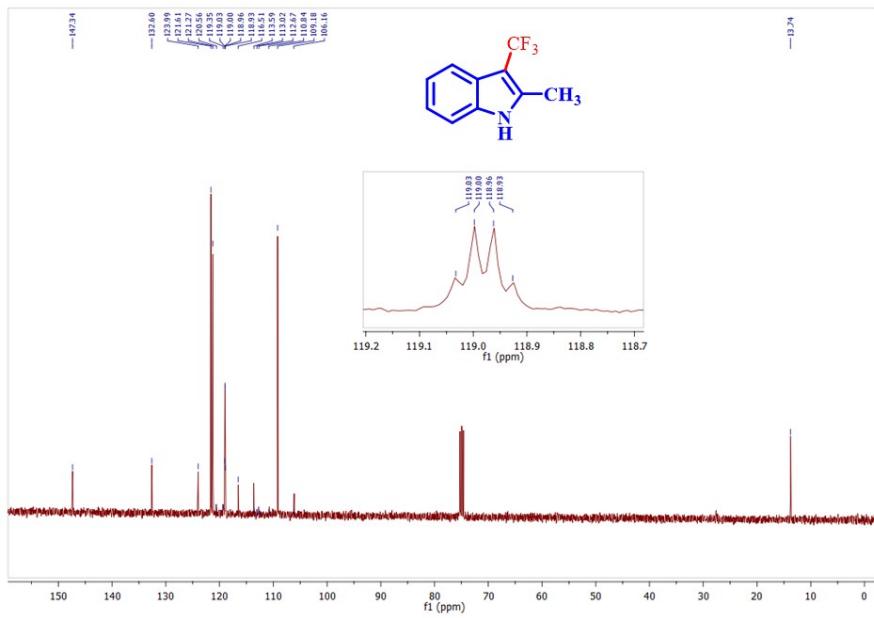


Figure S35. ¹³C NMR spectrum of 2-Methyl-3-(trifluoromethyl)-1H-indole [3].

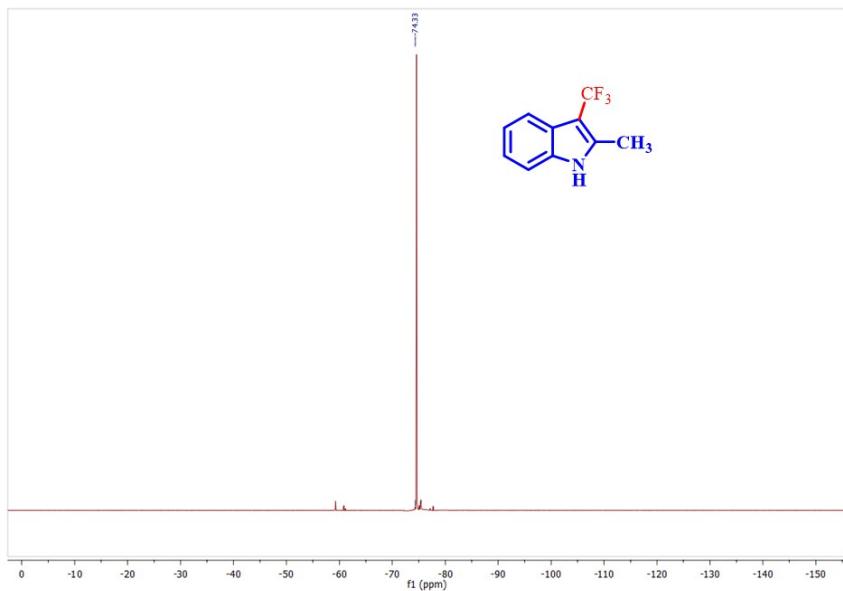


Figure S36. ^{19}F NMR spectrum of 2-Methyl-3-(trifluoromethyl)-1H-indole [3].

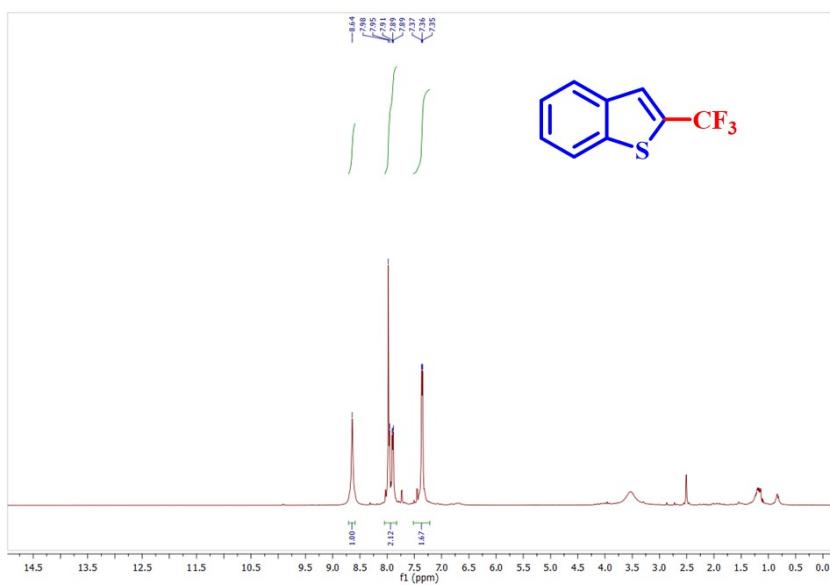


Figure S37. ^1H NMR spectrum of 2-(Trifluoromethyl)benzo[b]thiophene [3].

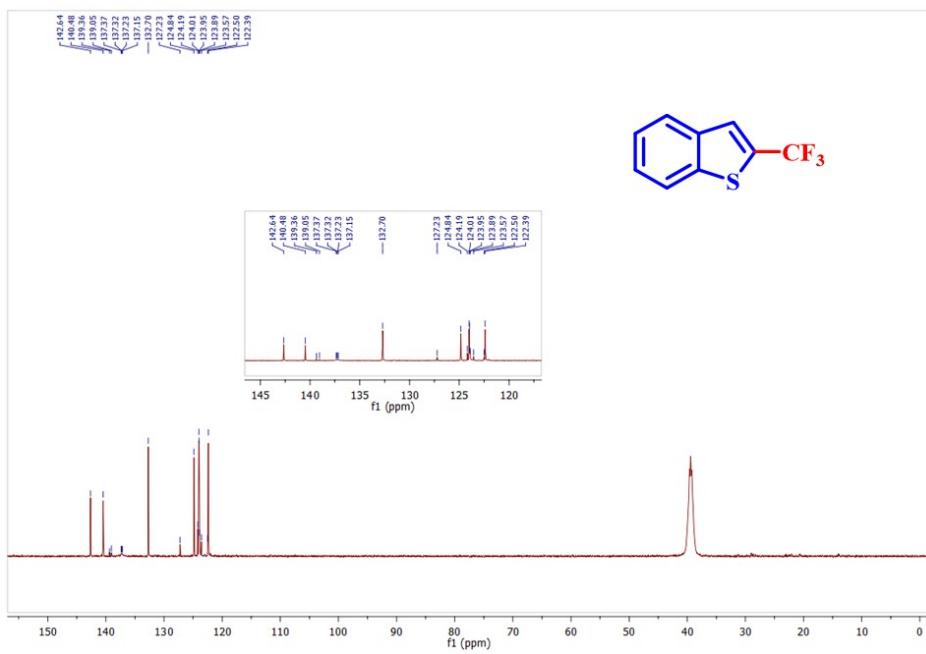


Figure S38. ^{13}C NMR spectrum of 2-(Trifluoromethyl)benzo[b]thiophene [3].

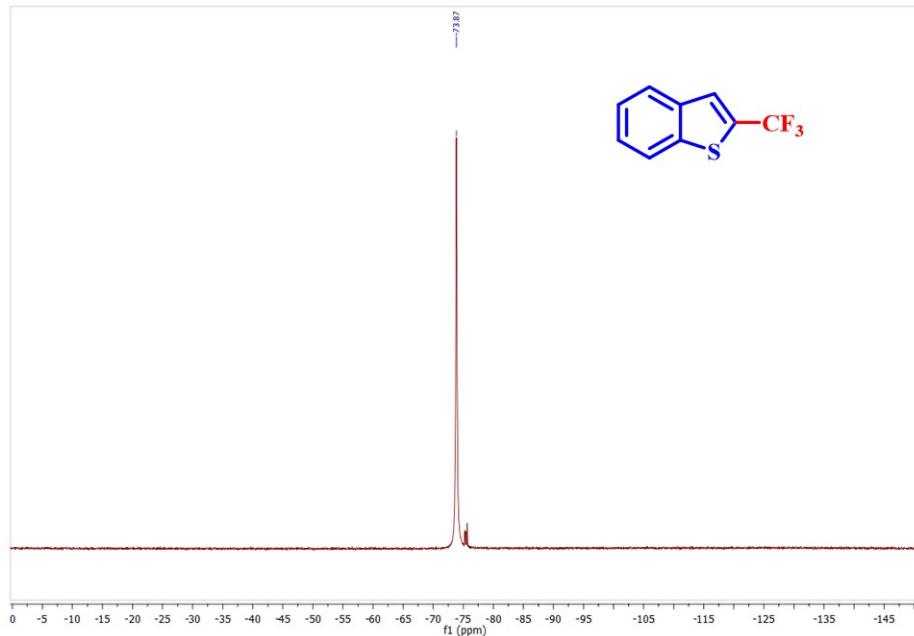


Figure S39. ^{19}F NMR spectrum of 2-(Trifluoromethyl)benzo[b]thiophene [3].

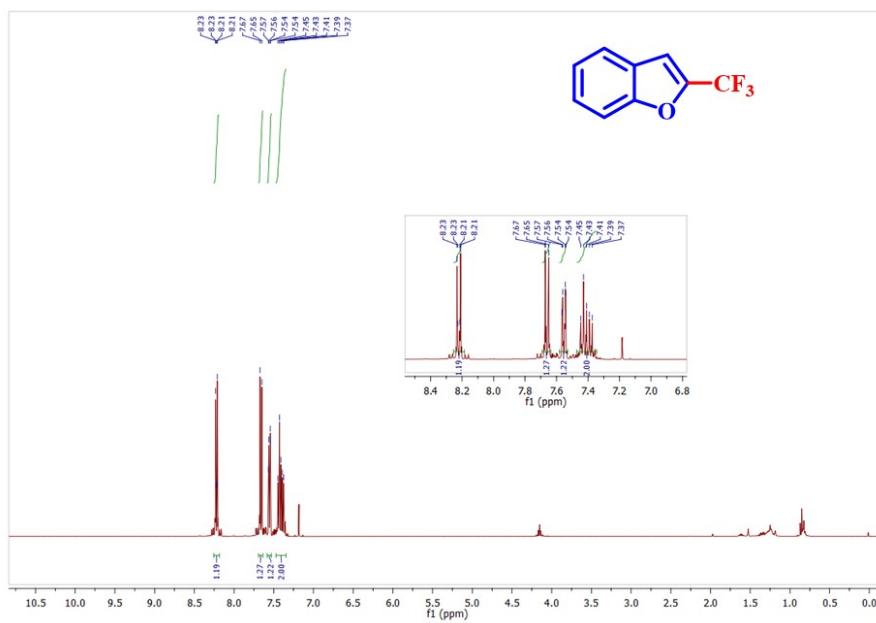


Figure S40. ¹H NMR spectrum of 2-(Trifluoromethyl)benzofuran [3].

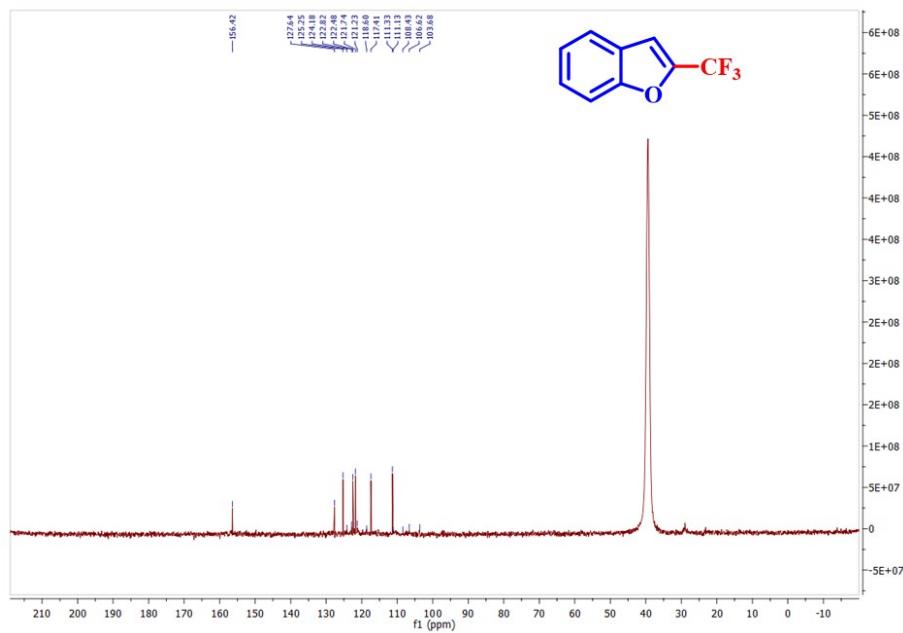


Figure S41. ¹³C NMR spectrum of 2-(Trifluoromethyl)benzofuran [3].

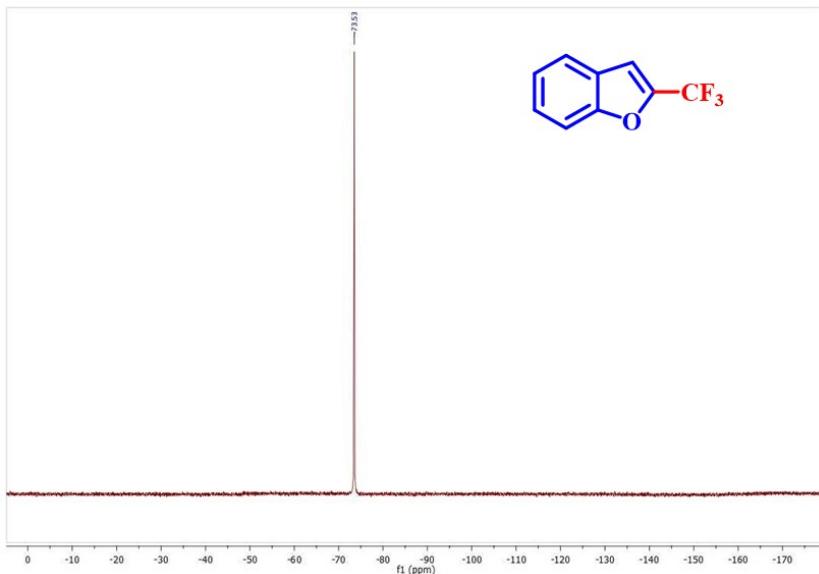


Figure S42. ¹⁹F NMR spectrum of 2-(Trifluoromethyl)benzofuran [3].

- [1] Q. Xiao, S. Sarina, E. Jaatinen, J. Jia, D. P. Arnold, H. Liu, H. Zhu, *Green Chem.* 2014, **16**, 4272.
- [2] A. Maji, A. Hazra, *Org. Lett.* 2014, **16**, 4524.
- [3] Z. Bazyar, M. Hosseini-Sarvari, *Org. Process Res. Dev.* 2019, **23**, 2345.
- [4] Z. Wang, G. Fenglian, W. Wan, H. Jiang, J. Hao, *J Fluor Chem.* 2007, **128**, 1143.
- [5] V. M. Muzalevskiy, A. V. Shastin, E. S. Balenkova, G. Haufe, V. J. Nenajdenko, *Russ. Chem. Bull.* 2008, **57**, 2175.
- [6] I. Abdiaj, C. Bottecchia, J. Alcazar, T. Noel, *Synthesis* 2017, **49**, 4978.