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## **Electronic Supplementary Information**

## Niobium and tantalum complexes derived from the acids Ph<sub>2</sub>C(X)CO<sub>2</sub>H (X = OH, NH<sub>2</sub>): Synthesis, structure and ROP capability

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Figure S1. Alternative view of four niobium cluster present in 1.



**Figure S2.** Alternative view of the four-tantalum cluster present in **2-0.5MeCN**. Atoms are drawn as 50% probability ellipsoids. Symmetry operation: i = 1-x, y,  $\frac{1}{2}$  –z.



Figure S3. IR spectrum of  $L^1H_2$ , 1 and 2.0.5MeCN in nujol.



Figure S4. ESI-MS spectrum of 1.



Figure S5. ESI-MS spectrum of 2.0.5MeCN.



Figure S6. Alternative view of the asymmetric unit of 3.2MeCN



Figure S7. Asymmetric unit of 4.2.25MeCN with atoms drawn as 50% probability ellipsoids.



Figure S8. IR spectrum of  $L^2H_3$ , 3·2MeCN and 4·2.25MeCN in nujol.



Figure S9. ESI-MS spectrum of 3.2MeCN.



Figure S10. ESI-MS spectrum of 4.2.25MeCN.



Figure S11. Plot of relationship between conversion and time (min) for the polymerization of  $\epsilon$ -CL (Table 1, entry 1-2).



Figure S12. <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>) of PCL catalyzed by 1 (Table 1, entry 1).



**Figure S13.** MALDI-TOF spectrum of the PCL catalysed by **1** in the absence of BnOH (Table 1, entry 5); n is the degree of polymerization.



**Figure S14**. Plot of relationship between conversion and time (min) for the polymerization of *r*-LA (Table 2, entry 5-8).



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