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Electronic Supplementary Information (ESI) for

Dialkylamino and trialkylammonium groups in close proximity: intra- and intermolecular ways of formation, structural consequences, and properties. A case of metal-free directed ortho alkylation

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Experimental Data

¹H and ¹³C NMR spectra were recorded on a Bruker DPX-250 (250 MHz) spectrometer with the solvent residual peaks as the internal standard (Scientific and Educational Laboratory of Resonance Spectroscopy, Department of Natural and High Molecular Compounds Chemistry of Southern Federal University). In some cases, a Bruker Avance 600 (600 MHz) and Varian Unity-300 (300 MHz) spectrometers were used. The HR-ESI mass-spectra were obtained on a BRUKER maXis spectrometer equipped with an electrospray ionization (ESI) source; methanol was used as the solvent (Chemical Analysis and Materials Research Centre, St. Petersburg State University). The instrument was operated in positive mode using an *m/z* range of 50–1200. The capillary voltage of the ion source was set at 4000 V. The nebulizer gas pressure was 1.0 bar, and the drying gas flow was set to 4.0 L/min. The progress of reactions and the purity of products were purified by chromatography on silica gel (230–400 mesh, grade 60) or Al₂O₃ (Brockmann activity III). The melting points were measured in sealed capillaries and are uncorrected. The solvents were purified and dried by standard methods.

1-Methylamino-8-(pyrrolidin-1-yl)naphthalene (4)



A mixture of 1-dimethylamino-8-(pyrrolidin-1-yl)naphthalene (3)^{S1} (112 mg, 0.47 mmol), KI (390 mg, 2.35 mmol) and 37% aqueous HBr (0.08 mL, 0.49 mmol) in DMF (6 mL)
 was refluxed for 1 h. The resulting mixture was poured into water (30 mL) and extracted with CH₂Cl₂ (3 × 10 mL). The solvent was evaporated, and the crude product was purified

by column chromatography (silica gel, CH₂Cl₂) to afford **4** (73 mg, 69%) as a colourless solid (its solutions possess blue fluorescence under UV light) with mp 79–81 °C.^{S2} ¹H NMR (300 MHz, CDCl₃): $\delta = 8.90$ (br s, 1H), 7.48 (dd, J = 8.1, 1.3 Hz, 1H), 7.33–7.25 (m, 2H), 7.15 (dd, J = 7.5, 1.3 Hz, 1H), 7.03 (dd, J = 8.1, 1.0 Hz, 1H), 6.41 (br d, J = 7.8 Hz, 1H), 3.40–3.35 (m, 2H), 2.94 (br s, 3H), 2.84–2.74 (m, 2H), 2.01–1.96 (m, 4H). ¹³C{¹H} NMR (150 MHz, CDCl₃): $\delta = 148.14$, 147.90, 136.71, 126.83, 125.26, 125.14, 118.99, 114.82 (2C), 102.36, 54.05, 30.28, 24.02. HRMS (ESI): m/z [M + H]⁺ calcd for C₁₅H₁₉N₂⁺: 227.1543; found: 227.1549.

Interaction of pyrrolidinonaphthalene 4 with 1,3-dibromopropane

A mixture of compound 4 (46 mg, 0.20 mmol), K_2CO_3 (21 mg, 0.15 mmol) and 1,3-dibromopropane (0.03 mL, 0.30 mmol) in acetonitrile (4 mL) was refluxed for 7 d. The resulting reaction suspension was basified with aqueous KOH and evaporated to dryness. The residue was dissolved in CH₂Cl₂ (4 mL) and filtered. Diethyl ether (9 mL) was added to the concentrated filtrate (1 mL). The precipitate formed was filtered off and washed with Et₂O (2 × 5 mL) to afford **5** (26 mg, 38%). Next, diethyl ether was removed,

and the remaining neutral compounds were separated by column chromatography (silica gel, CH₂Cl₂), after isolation of unreacted 4 (11 mg, 24%), the solvent was changed to ethyl acetate to give 6 (8 mg, 15%).

5-Methyl-2,3,4,5-tetrahydrospiro[naphtho[1,8-bc][1,5]diazocine-1,1'-pyrrolidin]-1-ium bromide (5)



Beige solid; mp 168–170 °C (rearranges to 6 on heating; see below). Crystals suitable for XRD analysis were obtained from a 1:1 mixture of CH₂Cl₂/Et₂O. ¹H NMR (250 MHz, $CDCl_3$): $\delta = 8.10$ (br d, J = 7.8 Hz, 1H), 7.90 (br d, J = 8.1 Hz, 1H), 7.67 (dd, J = 8.0, 1.3Hz, 1H), 7.58–7.49 (m, 2H), 7.42 (dd, J = 7.5, 1.3 Hz, 1H), 5.12–5.00 (m, 2H), 4.74–4.65 (m, 2H), 4.46–4.28 (m, 2H), 3.27–3.15 (m, 1H), 3.05 (s, 3H), 2.79–2.71 (m, 1H), 2.47–

2.35 (m, 2H), 2.12–1.99 (m, 2H), 1.90–1.80 (m, 1H), 1.68–1.61 (m, 1H). ¹³C{¹H} NMR (62.9 MHz, $CDCl_3$): $\delta = 146.1, 137.24, 137.15, 132.6, 127.7, 126.1, 125.1, 125.0, 122.3, 121.0, 72.3, 71.0, 68.7, 57.0, 122.3, 121.0, 72.3, 71.0, 68.7, 57.0, 122.3, 121.0, 72.3, 71.0, 68.7, 57.0, 122.3, 121.0, 72.3, 71.0, 68.7, 57.0, 122.3, 121.0, 72.3, 71.0, 68.7, 57.0, 122.3, 122$ 43.6, 26.4, 22.5, 21.1. HRMS (ESI): *m/z* [M]⁺ calcd for C₁₈H₂₃N₂⁺: 267.1861; found: 267.1856.

The conversion of salt 5 (10 mg, 0.03 mmol) to 6 was conducted in a pre-heated silicone oil bath at 170 °C for 10 min. The reaction mixture was then allowed to cool to room temperature, dissolved in ethyl acetate (2 mL), poured into water, and basified with aqueous KOH to pH 10-11. The aqueous layer was additionally extracted with ethyl acetate $(2 \times 2 \text{ mL})$ to give crude tetrahydrobenzo[h]quinoline 6 (7 mg, 88%).

1-Methyl-10-(pyrrolidin-1-yl)-1,2,3,4-tetrahydrobenzo[*h*]quinoline (6)



Dark beige oil. ¹H NMR (250 MHz, CDCl₃): $\delta = 7.22-7.15$ (m, 3H), 7.03 (d, J = 8.1 Hz, 1H), 6.80 (dd, *J* = 6.0, 2.7 Hz, 1H), 3.73–3.02 (br m, 5H), 2.91–2.86 (m, 3H), 2.75 (s, 3H), 2.06–1.81 (m, 6H). ¹³C{¹H} NMR (62.9 MHz, CDCl₃): $\delta = 147.0, 144.9, 136.4, 127.4,$ 124.7, 123.2, 120.1, 119.5, 119.0, 109.9, 52.3, 51.3, 45.3, 29.1, 24.8, 20.0. HRMS (ESI): m/z [M + H]⁺ calcd for C₁₈H₂₃N₂⁺: 267.1856; found: 267.1860.

Interaction of pyrrolidinonaphthalene 4 with o-xylylene dibromide

A mixture of compound 4 (30 mg, 0.14 mmol), K₂CO₃ (19 mg, 0.14 mmol) and o-xylylene dibromide (45 mg, 0.17 mmol) in 1,4-dioxane (4 mL) was stirred and refluxed for 15 hrs. The resulting mixture was poured into water (15 mL), basified with aqueous KOH to pH 10-11 and extracted with ethyl acetate (3 \times 6 mL). The solvent was evaporated, and the crude product was purified by column chromatography (silica gel, CH₂Cl₂, after isolation of mobile precursors, the solvent was changed to ethyl acetate). TLC plate showed only one spot, but NMR and ESI-MS spectra indicated the presence of two compounds, salt 9 and bromide 10 (See Figs. S9–S10).

N-(2-(Bromomethyl)benzyl)-*N*-methyl-8-(pyrrolidin-1-yl)naphthalen-1-amine hydrogen tetrafluoroborate (11)



The isolated mixture of **9** and **10** (20 mg) was treated with 40% aqueous HBF₄ (0.01 mL, 0.06 mmol, 1.2 equiv) in a minimum volume of acetonitrile (0.4 mL). The solvent was removed, and the oily residue was washed with Et₂O (2 × 5 mL) to afford salt **11** (20 mg, 83%) as colourless oil. ¹H NMR (250 MHz, CD₃CN): δ = 19.96 (br s, 1H),

¹¹ 7.80–7.76 (m, 4H), 7.61–7.54 (m, 2H), 7.01–6.89 (m, 4H), 5.10–5.00 (m, 2H), 4.30– 4.17 (m, 2H), 3.77–3.68 (m, 2H), 3.46–3.40 (m, 2H), 3.33 (d, J = 2.7 Hz, 3H), 1.91–1.83 (m, 2H), 1.79– 1.51 (m, 2H). ¹³C{¹H} NMR (62.9 MHz, CD₃CN): $\delta = 141.7$, 139.3, 134.2, 133.4, 133.2, 131.9 (2C), 129.23, 129.20, 128.7, 128.6, 126.9, 126.8, 124.3, 121.7, 121.5, 63.5, 62.5, 53.9, 41.4, 33.2, 29.8, 25.5.

1-(8-(Dimethylamino)naphthalen-1-yl)-1-methylpyrrolidin-1-ium iodide (14)

Iodomethane was added (0.24 mL, 3.80 mmol) to a solution of 1-dimethylamino-8-(pyrrolidin-1-yl)naphthalene (3)^{S1} (91 mg, 0.38 mmol) in acetonitrile (4 mL). The reaction Me₂N Me mixture was kept tightly closed for 21 d at room temperature. The resulting solution was treated with aqueous 10% KOH (0.2 mL) and evaporated to dryness. The residue was 14 dissolved in CH₂Cl₂ (4 mL) and filtered. The solvent was removed, and the crude product was washed with diethyl ether (2 \times 5 mL). Further purification was carried out by dissolving iodide 14 in CH₂Cl₂ (1 mL) and addition of diethyl ether (8 mL). The precipitate formed was washed with Et₂O (2×8 mL) to afford 14 (19 mg, 13%) as a beige solid with mp 134–136 °C (quantitatively decomp. to 3 on melting). Crystals suitable for XRD analysis were obtained from a 1:1 mixture of CH₂Cl₂/Et₂O. ¹H NMR (250 MHz, CDCl₃): $\delta = 8.55$ (br d, J = 7.8 Hz, 1H), 7.87 (br d, J = 8.0 Hz, 1H), 7.72 (dd, J = 7.2, 2.2 Hz, 1H), 7.62–7.50 (m, 3H), 5.36–5.26 (m, 2H), 4.27 (s, 3H), 3.85–3.73 (m, 2H), 2.66 (s, 6H), 2.51–2.39 (m, 2H), 2.16–2.05 (m, 2H). ¹³C{¹H} NMR (62.9 MHz, CDCl₃): $\delta = 149.9$, 144.1, 137.5, 131.6, 127.3, 127.1, 125.6, 124.8, 122.8, 121.5, 70.1, 54.0, 46.8, 21.8. HRMS (ESI): *m/z* [M]⁺ calcd for C₁₇H₂₃N₂⁺: 255.1856; found: 255.1856.

1,1'-(2-Methylnaphthalene-1,8-diyl)dipyrrolidine (17)



Iodomethane (0.19 mL, 3.00 mmol) was added to dipyrrolidinonaphthalene 15^{S3} (40 mg, 0.15 mmol) and K₂CO₃ (21 mg, 0.15 mmol) in acetonitrile (5 mL). The reaction mixture was stirred in inert atmosphere for 10 d at 50 °C. The resulting mixture was poured into water (20 mL), basified with aqueous KOH to pH 10–11 and extracted with CH₂Cl₂ (3 ×

7 mL). The solvent was evaporated, and the crude product was purified by column chromatography (Al₂O₃, CH₂Cl₂), after isolation of dipyrrolidinonaphthalene **15** (6 mg, 15%), the solvent was changed to ethyl acetate to give **17** (35 mg, 83%) as a beige solid with mp 40–42 °C. ¹H NMR (250 MHz, CDCl₃): δ

= 7.43–7.36 (m, 2H), 7.26–7.21 (m, 2H), 7.04 (br d, J = 7.3 Hz, 1H), 3.44–3.39 (m, 4H), 3.13–2.90 (br m, 4H), 2.41 (s, 3H), 1.98–1.90 (m, 8H). ¹³C{¹H} NMR (62.9 MHz, CDCl₃): δ = 148.2, 142.6, 135.9, 131.7, 130.3, 125.7, 124.6, 123.4, 122.4, 112.9, 53.1, 51.2, 26.5, 24.3, 20.7. HRMS (ESI): m/z [M + H]⁺ calcd for C₁₉H₂₅N₂⁺: 281.2012; found: 281.2012.

1,1'-(2-Methylnaphthalene-1,8-diyl)dipyrrolidine hydrogen tetrafluoroborate (18b)



Compound 17 (30 mg, 0.11 mmol) was treated with 40% aqueous HBF₄ (0.02 mL, 0.12 mmol, 1.1 equiv) in a minimum volume of acetonitrile (0.4 mL) followed by 3-fold dilution with Et₂O. The residue formed was washed with Et₂O and vacuum-dried to give 18b (33 mg, 80%) as a beige solid with mp 214–216 °C (from EtOH). Crystals

suitable for XRD analysis were obtained from CH₂Cl₂. ¹H NMR (250 MHz, CD₃CN): $\delta = 19.80$ (br s, 1H), 8.01–7.94 (m, 2H), 7.84 (dd, J = 7.7, 1.1 Hz, 1H), 7.65 (t, J = 7.9 Hz, 1H), 7.52 (d, J = 8.6 Hz, 1H) 3.87–3.77 (m, 2H), 3.69–3.55 (m, 4H), 3.50–3.37 (m, 2H), 2.59 (s, 3H), 2.45–2.33 (m, 4H), 2.24–2.12 (m, 4H). ¹³C{¹H} NMR (62.9 MHz, CD₃CN): $\delta = 141.0$, 136.6, 134.4, 133.9, 132.0, 129.3, 129.2, 126.4, 122.8, 122.6, 58.9, 54.0, 25.7, 24.2, 18.8.

Interaction of dipyrrolidinonaphthalene 15 with iodoethane

Iodoethane (1 mL, 12.40 mmol) was added to dipyrrolidinonaphthalene **15** ^{S3} (115 mg, 0.43 mmol) and K₂CO₃ (59 mg, 0.43 mmol) in acetonitrile (7 mL). The reaction mixture was refluxed for 4 d. The resulting mixture was poured into water (30 mL), basified with aqueous KOH to pH 10–11 and extracted with CH₂Cl₂ (3 × 10 mL). The solvent was evaporated, and the crude products were purified by column chromatography (Al₂O₃, CH₂Cl₂), after isolation of starting compound **15** (38 mg, 33%) and *para*-ethylated product **20** (18 mg, 14%), the solvent was changed to ethyl acetate to give *ortho*-isomer **19** (48 mg, 38%).

1,1'-(2-Ethylnaphthalene-1,8-diyl)dipyrrolidine (19)



Beige solid; mp 33–35 °C. ¹H NMR (250 MHz, CDCl₃): δ = 7.45 (d, *J* = 8.4 Hz, 1H), 7.38 (d, *J* = 7.7 Hz, 1H), 7.28–7.22 (m, 2H), 7.07 (d, *J* = 7.5 Hz, 1H), 3.43–3.36 (m, 4H), 3.21–2.95 (m, 4H), 2.78 (q, *J* = 7.5 Hz, 2H), 2.00–1.89 (m, 8H), 1.24 (t, *J* = 7.5 Hz, 3H).

¹³C{¹H} NMR (62.9 MHz, CDCl₃): δ = 148.5, 142.1, 138.0, 135.9, 128.8, 125.5, 124.7,

123.7, 122.4, 113.0, 53.1, 51.8, 26.4 (3C), 24.4, 16.0. HRMS (ESI): m/z [M + H]⁺ calcd for C₂₀H₂₇N₂⁺: 295.2169; found: 295.2173.

1,1'-(2-Ethylnaphthalene-1,8-diyl)dipyrrolidine hydrogen tetrafluoroborate (19·HBF₄)



Compound **19** (50 mg, 0.17 mmol) was treated with 40% aqueous HBF₄ (0.032 mL, 0.19 mmol, 1.1 equiv) in a minimum volume of acetonitrile (0.8 mL) followed by 3-fold dilution with Et₂O. The residue formed was washed with Et₂O and vacuum-dried to give **19**·**HBF**₄ (47 mg, 72%) as a beige solid with mp 222–224 °C (from EtOH). ¹H

NMR (250 MHz, CD₃CN): δ = 19.81 (br s, 1H), 8.02 (d, *J* = 8.6 Hz, 1H), 8.00 (dd, *J* = 8.1, 1.2 Hz, 1H), 7.85 (dd, *J* = 7.7, 1.2 Hz, 1H), 7.70–7.64 (m, 2H), 3.88–3.67 (m, 4H), 3.55–3.38 (m, 4H), 2.86 (q, *J* = 7.5 Hz, 2H), 2.47–2.37 (m, 4H), 2.27–2.16 (m, 4H), 1.39 (t, *J* = 7.5 Hz, 3H). ¹³C{¹H} NMR (62.9 MHz, CD₃CN): δ = 141.7, 141.1, 136.7, 134.3, 130.7, 130.1, 129.9, 127.0, 123.3, 123.2, 59.5, 55.9, 26.1, 24.8, 24.7, 15.0.

1,1'-(4-Ethylnaphthalene-1,8-diyl)dipyrrolidine (20)



Dark beige oil. ¹H NMR (250 MHz, CDCl₃): δ = 7.40 (dd, *J* = 8.3, 1.2 Hz, 1H), 7.32–7.25 (m, 1H), 7.12 (d, *J* = 7.7 Hz, 1H), 6.76 (dd, *J* = 7.5, 1.1 Hz, 1H), 6.69 (d, *J* = 7.8 Hz, 1H), 3.69–2.70 (m, 10H), 1.93–1.82 (m, 8H), 1.34 (t, *J* = 7.5 Hz, 3H). ¹³C{¹H} NMR (62.9 MHz, CDCl₃): δ = 148.3, 146.4, 135.4, 130.2, 125.3, 124.5, 119.2, 114.5, 108.4, 108.2,

51.2, 51.1, 25.8, 25.0, 24.9, 14.8. HRMS (ESI): m/z [M + H]⁺ calcd for C₂₀H₂₇N₂⁺: 295.2169; found: 295.2170.



Fig. S1 ¹H NMR spectrum of 4 (300 MHz, CDCl₃).



Fig. S2 ${}^{13}C{}^{1H}$ APT-NMR spectrum of 4 (150 MHz, CDCl₃).

2029.25 1921.47 1922.58 1922.58 1922.58 1921.58 1921.58 1921.58 1933.88 1988.38 1888.38 1888.38 1888.38 1888.38 1888.38 1889.36 1888.38 1889.36 1889.37 1115.59 1115.59 1115.59 1115.59 1115.59 1125.75 1125.7



Fig. S3 ¹H NMR spectrum of 5 (250 MHz, CDCl₃).



Fig. S4 ¹³C{¹H} APT-NMR spectrum of 5 (62.9 MHz, CDCl₃).







Fig. S7 ¹³C{¹H} APT-NMR spectrum of **6** (62.9 MHz, CDCl₃).



Fig. S8 HRMS (ESI) of 6.



Fig. S9 ¹H NMR spectrum of an inseparable mixture of 9 and 10 (600 MHz, CDCl₃).



Fig. S10 HRMS (ESI) of a 3:7 mixture of 9 and 10.

-4989.56







Fig. S12 $^{13}C\{^{1}H\}$ APT-NMR spectrum of 11 (62.9 MHz, CD₃CN).







Fig. S14 ¹³C{¹H} APT-NMR spectrum of 14 (62.9 MHz, CDCl₃).



Fig. S15 HRMS (ESI) of 14.



Fig. S16 ¹H NMR-monitoring of the **3** + MeI (1:10) and **3** + Me₂SO₄ (1:2) mixtures (250 MHz, 22 °C, only aliphatic regions are shown).

1856.67 1848.35 1840.11 1816.13 1816.13 1816.13 1809.070 11609.070 11763.04 1755.75

859.89 847.12 847.12 763.42 723.80 603.52 603.52 499.75 474.17















Fig. S20 ¹H NMR spectrum of 18b (250 MHz, CD₃CN).



Fig. S21 ${}^{13}C{}^{1}H$ APT-NMR spectrum of 18b (62.9 MHz, CD₃CN).

1865.77 1855.42 1857.42 1849.57 1849.57 1849.65 1841.90 1841.90 1814.85 1814.85 1814.85 1814.85 1814.65 1814.65 1814.65 1711.24

857.64 845.93 845.93 801.67 801.67 707.07 7737.07 697.54 697.54 692.21 481.91 472.87 472.87 493.21 472.87 313.05 313.07 313.07 313.07 313.07 313.07 313.07 313.07 472.87 313.07 472.87 313.07 472.87 472.97 473.97 4



Fig. S22 ¹H NMR spectrum of 19 (250 MHz, CDCl₃).



Fig. S23 ${}^{13}C{}^{1}H$ APT-NMR spectrum of 19 (62.9 MHz, CDCl₃).







Fig. S26 ${}^{13}C{}^{1}H$ APT-NMR spectrum of 19·HBF₄ (62.9 MHz, CD₃CN).

19854.72 1985.54 1986.526 1986.526 1987.526 1982.87 1820.55 1820.538 1820.538 1820.538 1820.538 1820.538 1820.538 1820.538 1820.538 1820.538 1668.4816 1668.48 1668.48 1668.48 1668.48 1668.4816 1668.48 166



Fig. S28 ¹³C{¹H} APT-NMR spectrum of 20 (62.9 MHz, CDCl₃).



Fig. S29 HRMS (ESI) of 20.

XRD Data

	A A Br		F1 F2 F3 F4
			NI HI N2
		A A	A A A
Identification code	Structure-5	Structure-14	Structure-18b
Empirical formula	$C_{18}H_{23}BrN_2$	$C_{17}H_{23}IN_2$	$C_{19}H_{25}BF_4N_2$
Formula weight	347.29	382.27	368.22
Temperature/K	99.97(15)	100.00(10)	100.00(10)
Crystal system	monoclinic	monoclinic	monoclinic
Space group	Сс	$P2_{1}/c$	$P2_{1}/n$
a/Å	7.79650(10)	11.96030(10)	9.7812(1)
<i>b</i> /Å	18.2667(2)	9.88890(10)	10.3437(1)
c/Å	11.37140(10)	27.9480(2)	17.7597(2)
$\alpha/^{\circ}$	90	90	90
β/°	91.6140(10)	93.0630(10)	95.106(1)
γ/°	90	90	90
Volume/Å ³	1618.83(3)	3300.81(5)	1789.69(3)
Z	4	8	4
$\rho_{\rm calc} {\rm g/cm^3}$	1.425	1.538	1.367
μ/mm^{-1}	3.405	15.169	0.917
F(000)	720.0	1536.0	776.0
Crystal size/mm ³	0.14 imes 0.1 imes 0.08	0.12 imes 0.1 imes 0.08	$0.24 \times 0.18 \times 0.15$
Radiation	$CuK_{a} (\lambda = 1.54184)$	CuK_{α} ($\lambda = 1.54184$)	CuK_{α} ($\lambda = 1.54184$)
2Θ range for data collection/°	9.684 to 139.948	6.334 to 139.986	9.906 to 152.564
Index <i>h k l</i> max	9, 22, 13	14, 12, 30	12, 13, 22
Reflections collected	22273	24250	18984
	2966	6246	3735
Independent reflections	$[R_{\rm int} = 0.0280,$	$[R_{\rm int} = 0.0339,$	$[R_{\rm int} = 0.0263,$
	$R_{\rm sigma} = 0.0146$]	$R_{\rm sigma} = 0.0265$]	$R_{\rm sigma} = 0.0166$]
Data/restraints/parameters	2966/2/191	6246/0/367	3735/0/259
Goodness-of-fit on F^2	1.111	1.068	1.046
Final <i>R</i> indexes [$I >= 2\sigma(I)$]	$R_1 = 0.0139, wR_2 = 0.0359$	$\begin{vmatrix} R_1 = 0.0343, wR_2 = 0.0978\\ R_1 = 0.0362, wR_2 = 0.0993 \end{vmatrix}$	$R_1 = 0.0384, wR_2 = 0.1024$ $R_1 = 0.0407, wR_2 = 0.1044$
Final <i>R</i> indexes [all data]	$R_1 = 0.0139, wR_2 = 0.0360$	1.23/-1.79	0.35/-0.22
Largest diff. peak/hole / e Å ⁻³	0.13/-0.18		

 Table S1. Crystal data and structure refinement for the studied compounds.

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