

Electronic supplementary information (ESI)

**Colorimetric determination of biothiols with AuNPs@MoS₂ NSs
as peroxidase mimetic enzyme**

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Experimental Section

Kinetic assay

The steady-state kinetic parameters were examined to analyze the catalytic performance of AuNPs@MoS₂ NSs. The experiment was performed by varying the concentration of H₂O₂ from 0.02 to 0.12 mmol·L⁻¹ while the TMB concentration was fixed at 1.0 mmol·L⁻¹. The experiment was also conducted by varying the working concentration of TMB from 0.2 to 1.2 mmol·L⁻¹ while the working concentration of H₂O₂ was fixed at 0.1 mmol·L⁻¹. The absorbance of 652 nm of the reaction was recorded for 10 min (at an interval of 1 min). Then obtained the kinetic parameters by Lineweaver–Burk double-reciprocal formula from the Michaelis–Menten equation:

$$\frac{1}{V} = \frac{K_m}{V_{max}[S]} + \frac{1}{V_{max}}$$

in which V is the initial velocity, K_m is the Michaelis constant, and V_{max} is the maximal reaction velocity, and $[S]$ is the concentration of substrate.

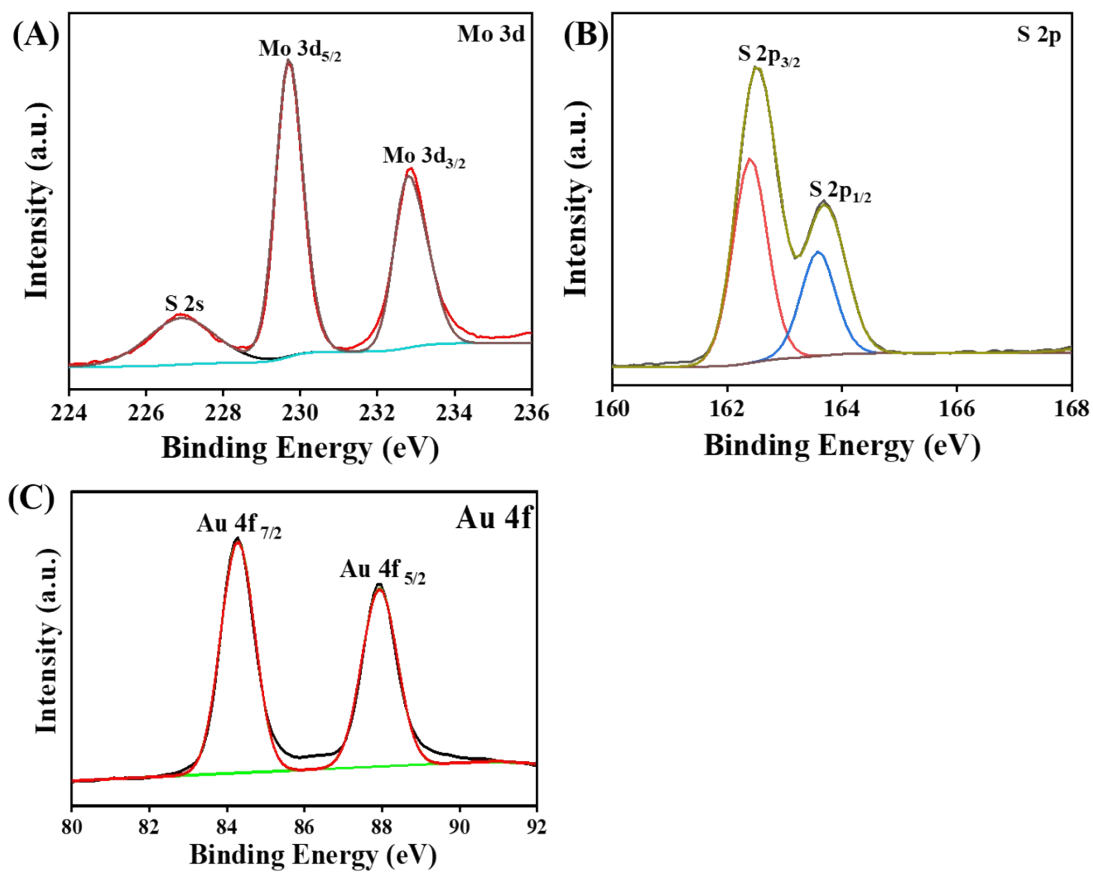


Fig. S1 XPS high-resolution spectra of AuNPs@MoS₂ NSs: (A) Mo 3d, (B) S 2p and (C) Au

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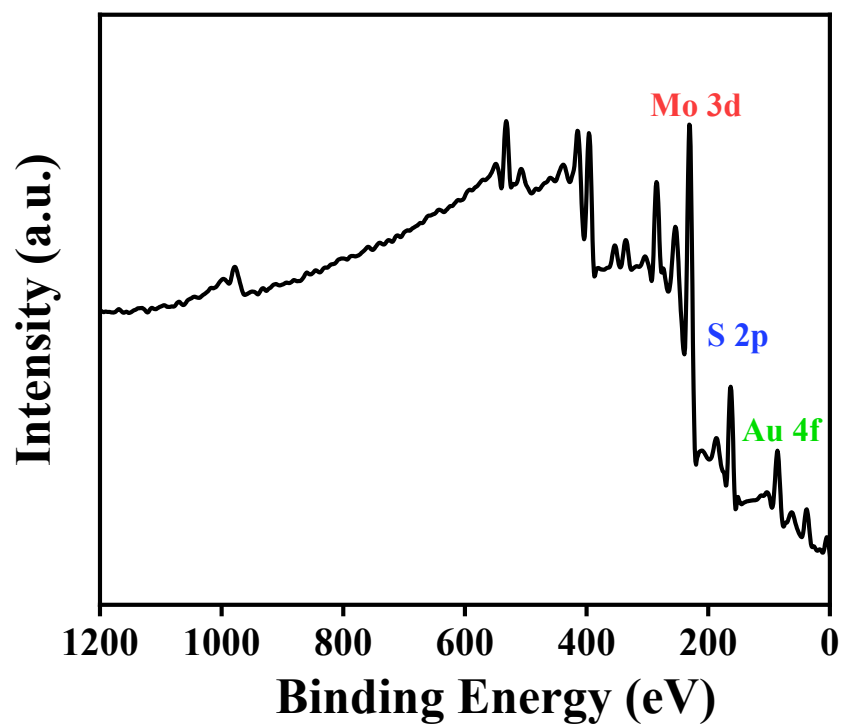


Fig.S2 XPS spectra of AuNPs@MoS₂ NSs after the catalytic reaction

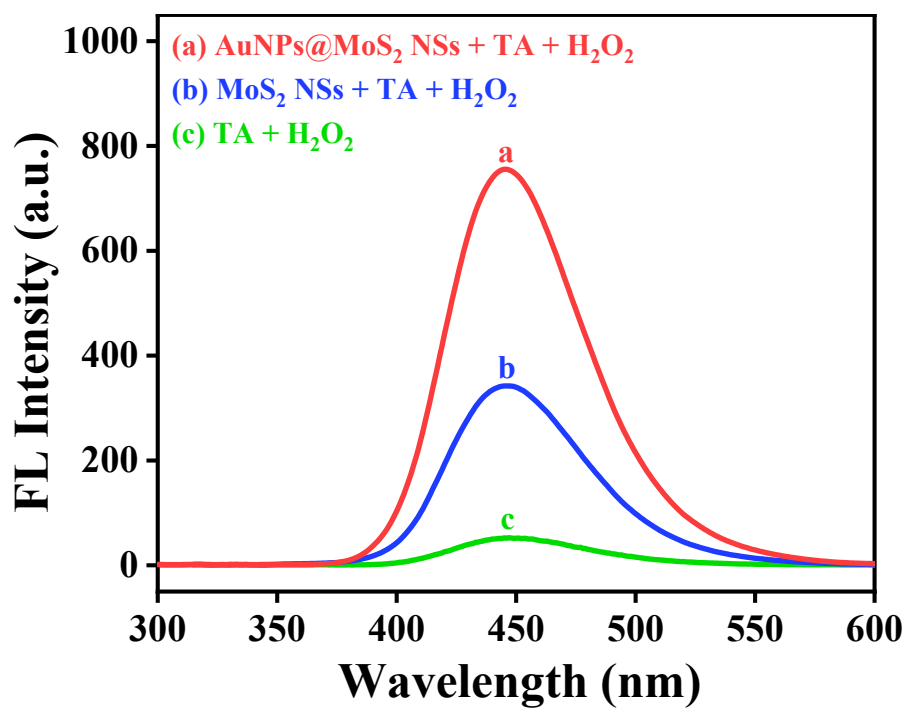


Fig.S3 Fluorescence spectra of different reaction systems. Experimental conditions: $1.0 \mu\text{g}\cdot\text{mL}^{-1}$ AuNPs@MoS₂ NSs or MoS₂ NSs; $0.5 \text{ mmol}\cdot\text{L}^{-1}$ TA; $1.0 \text{ mmol}\cdot\text{L}^{-1}$ H₂O₂; pH 4.0 HAC-NaAc buffer; incubated at 30 °C for 12 h (in dark).

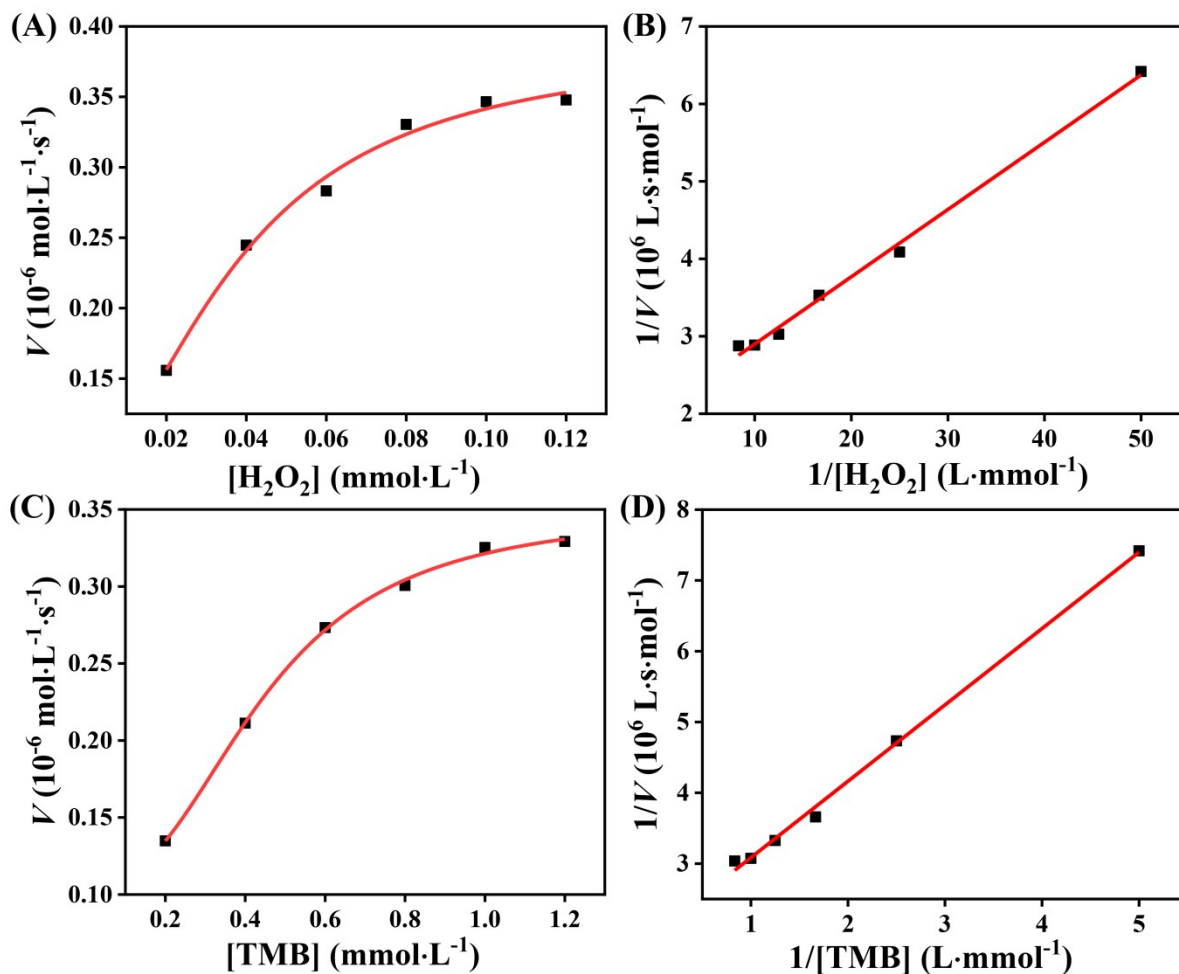


Fig.S4 Steady-state kinetic assay of AuNPs@MoS₂ NSs by using (A, C) Michaelis-Menten curves and (B, D) Lineweaver-Burk plots. (A, B) The concentration of TMB was 1.0 mmol·L⁻¹ and H₂O₂ concentration was varied. (C, D) The concentration of H₂O₂ was 0.1 mmol·L⁻¹ and TMB concentration was varied.

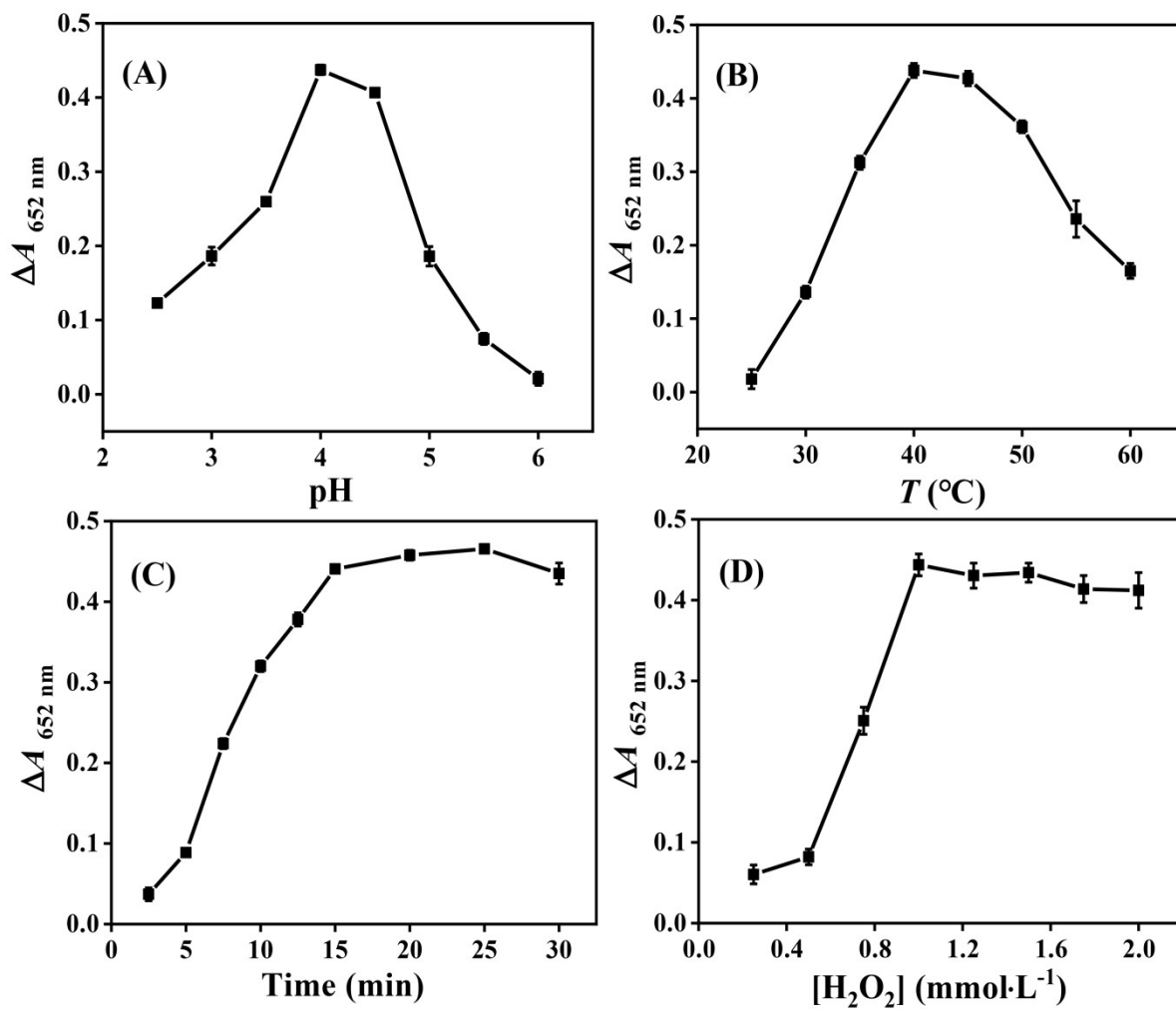


Fig.S5 The effects of (A) pH value, (B) temperature, (C) time and (D) the concentration of H₂O₂ on the catalytic activity AuNPs@MoS₂ NSs. Experimental conditions: 1.0 μg·mL⁻¹ AuNPs@MoS₂ NSs; 1.0 mmol·L⁻¹ TMB; 50 μmol·L⁻¹ Hcy.

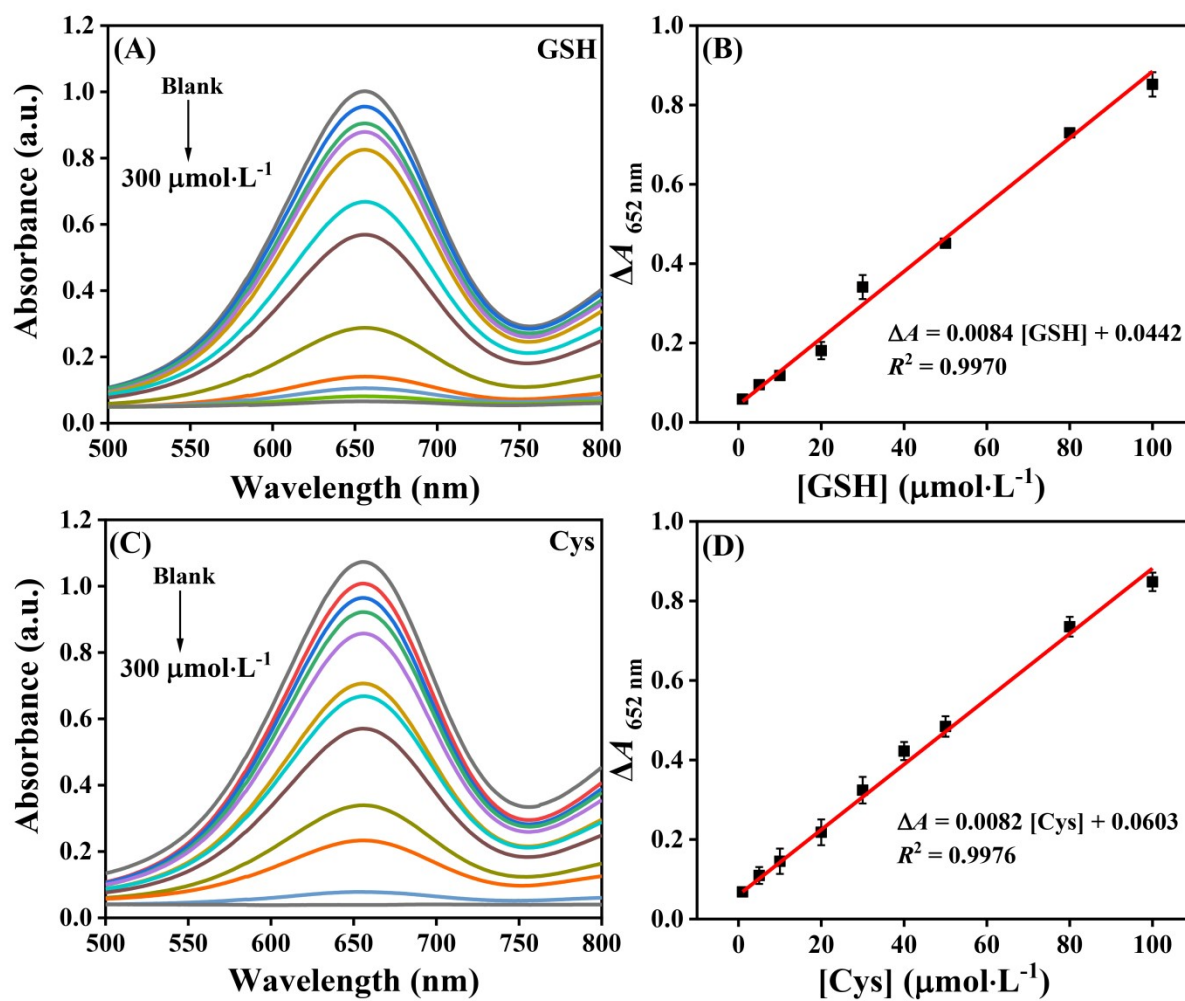


Fig.S6 UV-vis absorption spectra in the presence of various (A) GSH, (C) Cys concentrations.

The linear relationship between $\Delta A_{652 \text{ nm}}$ and (B) GSH, (D) Cys concentration

Table S1 Comparison of the kinetic parameters of AuNPs@MoS₂ NSs, HRP and MoS₂ NSs.

Materials	K_m (mmol·L ⁻¹)		V_{max} (10 ⁻⁸ mol·L ⁻¹ ·s ⁻¹)		Ref.
	TMB	H ₂ O ₂	TMB	H ₂ O ₂	
VS ₂ NSs	0.28	3.49	0.416	0.557	1
GQDs	8	0.01	0.117	7.3	2
PPy NPs	0.293	0.184	2.99	3.65	3
M-CQDs	0.219	0.431	0.882	0.046	4
HRP	0.434	3.702	10.0	8.71	5
MoS ₂ NSs	0.082	2.30	7.86	9.86	6
AuNPs@MoS ₂ NSs	0.043	0.537	49.2	49.8	This work

Table S2 Comparison of performance of colorimetric method for Hcy, GSH and Cys detection.

Target	Nanomaterials	Linear range ($\mu\text{mol}\cdot\text{L}^{-1}$)	LOD ($\mu\text{mol}\cdot\text{L}^{-1}$)	Ref.
Hcy	MIL-53(Fe)	0.3 ~ 20	0.1	7
	IrO ₂ /rGO	0.1 ~ 50	0.083	8
	Silver halides	10 ~ 100	9.0	9
	Cu ₂ (OH) ₃ NO ₃	5 ~ 40	0.099	10
	Ce-MOF	0 ~ 40	0.143	11
	AuNPs@MoS ₂ NSs	1 ~ 100	0.93	This work
GSH	NiO NFs	20 ~ 100	1.1	12
	TiO ₂ /MoS ₂	0.05 ~ 1	0.05	13
	Por-ZnFe ₂ O ₄ /rGO	2 ~ 40	0.76	14
	CDs@ZIF-8-a	0 ~ 100	1.04	15
	Pt ₁₀ -LP NCs	4 ~ 140	0.37	16
	AuNPs@MoS ₂ NSs	1 ~ 100	0.71	This work
Cys	VS ₄	5 ~ 100	2.5	17
	rGO-GP	2 ~ 30	0.1	18
	0.10CeO ₂ /CoO NC	5 ~ 10	3.71	19
	CB-CQDs	0.5 ~ 20	0.4	20
	MOF Eu-pydc	0 ~ 4	0.28	21
	AuNPs@MoS ₂ NSs	1 ~ 100	0.73	This work

Table S3 Determination of Hcy in healthy volunteers' serum samples

Samples	Added ($\mu\text{mol}\cdot\text{L}^{-1}$)	Total found ($\mu\text{mol}\cdot\text{L}^{-1}$)	Recovery (%)	RSD ($n = 3$, %)
1	10.0	10.03	100.3	4.1
	50.0	50.22	100.4	2.3
2	10.0	9.825	98.25	3.3
	50.0	50.58	101.2	3.9
3	10.0	9.773	97.73	4.0
	50.0	50.17	100.3	2.4

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