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## **Supporting Information**

## ZnS and Ag<sub>2</sub>Mo<sub>2</sub>O<sub>7</sub> regulate interface engineering by constructing S-scheme heterojunctions to facilitate photocatalytic hydrogen production

Zhenlu Liu<sup>1</sup>, Jing Xu<sup>\*1,3,4</sup>, Jiandong Wu<sup>2</sup>, Ye Liu<sup>1</sup>, Lijun Ma<sup>1</sup>, Linying Hu<sup>1</sup>

1 School of Chemistry and Chemical Engineering, North Minzu University, Yinchuan 750021, PR China

2 School of Materials Science and Engineering, North Minzu University, Yinchuan 750021, PR China

3 Ningxia Key Laboratory of Solar Chemical Conversion Technology, North Minzu University, Yinchuan 750021, PR China

4 Key Laboratory for Chemical Engineering and Technology, State Ethnic Affairs Commission, North Minzu University, Yinchuan 750021, PR China Email: wgyxj2000@163.com



Figure S1 XPS comparison of Zn and Ag before and after reaction.

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Material	Sacrificial agent	Hydrogen evolution	Light	
CuGaS <sub>2</sub> /ZnS	Na <sub>2</sub> S/Na <sub>2</sub> SO <sub>3</sub>	$137 \ \mu mol \cdot g^{-1} \ h^{-1}$	300 W Xe	[1]
ZnO/ZnS	/	247 $\mu$ mol $\cdot$ g <sup>-1</sup> h <sup>-1</sup>	300 W Xe	[2]
CeO <sub>2</sub> /ZnS	Na <sub>2</sub> S/Na <sub>2</sub> SO <sub>3</sub>	417 $\mu$ mol $\cdot$ g <sup>-1</sup> h <sup>-1</sup>	300 W Xe	[3]
ZIF-8/ZnS	TEOA	741.8 $\mu$ mol $\cdot$ g <sup>-1</sup> h <sup>-1</sup>	300 W Xe	[4]
ZnO/ZnS	Methanol	757 $\mu$ mol $\cdot$ g <sup>-1</sup> h <sup>-1</sup>	/	[5]
NiCo <sub>2</sub> O <sub>4</sub> /ZnS	Na <sub>2</sub> S/Na <sub>2</sub> SO <sub>3</sub>	880 $\mu$ mol $\cdot$ g <sup>-1</sup> h <sup>-1</sup>	300 W Xe	[6]
ZnO/ZnS	Na <sub>2</sub> S/Na <sub>2</sub> SO <sub>3</sub>	2.4 mmol $g^{-1} h^{-1}$	300 W Xe	[7]
Pt/SnO <sub>2</sub> /ZnS	TEOA	$2.17 \text{ mmol } g^{-1} h^{-1}$	300 W Xe	[8]
ZrO <sub>2</sub> /SnS <sub>2</sub> /ZnS	Na <sub>2</sub> S/Na <sub>2</sub> SO <sub>3</sub>	928.1 $\mu$ mol $\cdot$ g <sup>-1</sup> h <sup>-1</sup>	300 W Xe	[9]
Bi <sub>2</sub> S <sub>3</sub> /ZnS	Na <sub>2</sub> S/Na <sub>2</sub> SO <sub>3</sub>	176.2 $\mu$ mol·h <sup>-1</sup>	300 W Xe	[10]
g-C <sub>3</sub> N <sub>4</sub> /ZnS	Na <sub>2</sub> S/Na <sub>2</sub> SO <sub>3</sub>	713.7 $\mu$ mol $\cdot$ g <sup>-1</sup> h <sup>-1</sup>	300 W Xe	[11]
ZnO/ZnS	$Na_2S$	494.8 $\mu$ mol $\cdot$ g <sup>-1</sup> h <sup>-1</sup>	150 W Xe	[12]
Cu <sub>3</sub> P/ZnS	Na <sub>2</sub> S/Na <sub>2</sub> SO <sub>3</sub>	792 $\mu$ mol $\cdot$ g <sup>-1</sup> h <sup>-1</sup>	300 W Xe	[13]
Pt/ZnO/ZnS	H <sub>2</sub> O	87.6 $\mu$ mol $\cdot$ g <sup>-1</sup> h <sup>-1</sup>	300 W Xe	[14]
ZnO/ZnS	Na <sub>2</sub> S/Na <sub>2</sub> SO <sub>3</sub>	$15.7 \ \mu mol \cdot g^{-1} \ h^{-1}$	350 W Xe	[15]
Cu <sub>3</sub> P/ZnS	Na <sub>2</sub> S/Na <sub>2</sub> SO <sub>3</sub>	$14937 \ \mu mol \cdot g^{-1} \ h^{-1}$	250 W QTH	[16]
AgIn <sub>5</sub> S <sub>8</sub> /ZnS	Na <sub>2</sub> S/Na <sub>2</sub> SO <sub>3</sub>	932.8 $\mu$ mol $\cdot$ g <sup>-1</sup> h <sup>-1</sup>	300 W Xe	[17]
Ti <sub>3</sub> C <sub>2</sub> /ZnS	Lactic acid	502.6 $\mu$ mol $\cdot$ g <sup>-1</sup> h <sup>-1</sup>	300 W Xe	[18]
$MoS_2/ZnS$	Methanol	606 $\mu$ mol $\cdot$ g <sup>-1</sup> h <sup>-1</sup>	Hg-pen	[19]
NiS <sub>2</sub> /g-C <sub>3</sub> N <sub>4</sub> /ZnS	TEOA	348.5 $\mu$ mol $\cdot$ g <sup>-1</sup> h <sup>-1</sup>	300 W Xe	[20]
rGO/ZnS	Methanol	$108.8 \ \mu mol \cdot g^{-1} \ h^{-1}$	500 W Xe	[21]
Ag <sub>2</sub> Mo <sub>2</sub> O <sub>7</sub> /ZnS	TEOA	871.9 μmol·g <sup>-1</sup> h <sup>-1</sup>	5 W Xe	This work

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