The efficient removal of diclofenac sodium and bromocresol green from aqueous solution by sea urchin-like Ni/Co-BTC bimetallic organic framework: adsorption isotherms, kinetics and mechanism

Jia Wang,^a Yue Su,^a Shi-Wen Lv $^{\rm b}$ and Li-Hui Sun $^{*\rm a}$

^a School of Ocean Science and Technology, Dalian University of Technology, Panjin,

Liaoning 124221, PR China

^b School of Environmental Science and Engineering, Zhejiang Gongshang University,

Hangzhou, Zhejiang 310018, PR China

^{*}Corresponding author.

Mailing address: School of Ocean Science and Technology, Dalian University of Technology, Panjin, Liaoning 124221, PR China E-mail: sunlihui@dlut.edu.cn

Tel: +86-427-2631780

Fax: +86-427-2631798



Fig. S1 EDX spectra of Ni/Co-BTC MOF.



Fig. S2 Effect of Ni/Co molar ratio on adsorption of DCF and BCG on Ni/Co-BTC MOF. Reaction conditions: DCF and BCG concentration = 100 mg/L, respectively; adsorbent loading = 0.5 g/L, temperature = 25°C, initial pH = 7.



Fig. S3 Effect of adsorbent dosage on adsorption of DCF and BCG on Ni/Co-BTC MOF. Reaction conditions: DCF and BCG concentration = 100 mg/L, respectively; temperature = 25°C,

initial pH = 7.



Fig. S4 The zeta potential of Ni/Co-BTC MOF at different pH.



Fig. S5 XPS spectra of Ni/Co-BTC MOF before and after DCF and BCG adsorption. Reaction conditions: DCF and BCG concentration = 100mg/L, respectively; adsorbent loading = 0.5 g/L, temperature = 25° C, initial pH = 7.



Fig. S6 Reusability of Ni/Co-BTC MOF for DCF and BCG adsorption. Reaction conditions: DCF and BCG concentration = 100mg/L, respectively; adsorbent loading = 0.5 g/L, temperature = 25° C, initial pH = 7.

Pollutants	Adsorbent	Maximum adsorption capabilities (mg/g)	Ref.
DCF	Glutaraldehyde-modified	237.8	1
	Plasma-based surface	31.08	2
	functionalization of palm fibers		
	Fe3O4@MOF-100 (Fe)	377.36	3
	$Ni_{0.5}Zn_{0.5}Fe_2O_4$	52.91	4
	Sycamore ball activated carbon	178.89	5
	Ni/Co-BTC MOF	343.05	This study
BCG	Cotton stalks	222.22	6
	activated carbon		
	Phragmites karka	392.3	7
	ZnO-NRs-AC	57.80	8
	Cd(OH) ₂ -NW-AC	108.7	9
	ZIF-11	150	10
	Poly (1,2-epoxy-3-phenoxy) propane	52.63	11
	Ni/Co-BTC MOF	569.29	This study

Table.S1 Maximum adsorption capacities of DCF and BCG on other adsorbents

References:

- T. S. Machado, L. Crestani, G. Marchezi, F. Melara, J. R. de Mello, G. L. Dotto and J. S. Piccin, *CARBOHYD POLYM*, 2022, 277, 118868.
- 2 C. M. Darryle, E. Acayanka, B. Takam, L. N. Line, G. Y. Kamgang, S. Laminsi,L. Sellaoui and A. Bonilla-Petriciolet, *Journal of Water Process Engineering*,

2021, **43**, 102254.

- 3 X. Zheng, J. Wang, X. Xue, W. Liu, Y. Kong, R. Cheng and D. Yuan, *ENVIRON SCI POLLUT R*, 2018, **25**, 31705-31717.
- 4 Z. Mohammadi, A. R. Kelishami and A. Ashrafi, *WATER SCI TECHNOL*, 2021,
 83, 1265-1277.
- 5 T. Avcu, O. Üner and Ü. Geçgel, *Surfaces and Interfaces*, 2021, 24, 101097.
- M. Ozdemir, O. Durmus, O. Sahin and C. Saka, *DESALIN WATER TREAT*, 2016, 57, 18038-18048.
- 7 B. M. Murmu, S. S. Behera, S. Das, R. K. Mohapatra, B. K. Bindhani and P. K.
 Parhi, *INDIAN J CHEM TECHN*, 2018, 25, 409-420.
- 8 F. Ansari, M. Ghaedi, M. Taghdiri and A. Asfaram, ULTRASON SONOCHEM, 2016, 33, 197-209.
- 9 M. Ghaedi, H. Khajesharifi, A. Hemmati Yadkuri, M. Roosta, R. Sahraei and A. Daneshfar, Spectrochimica Acta Part A: Molecular and Biomolecular Spectroscopy, 2012, 86, 62-68.
- 10 R. Lamari, B. Benotmane and S. Mezali, *DESALIN WATER TREAT*, 2021, 224, 407-420.
- 11 G. Torğut and K. Demirelli, *ARAB J SCI ENG*, 2018, **43**, 3503-3514.