

## Supplementary Material

### Synthesis, structure, and corrosion inhibiting properties of RE<sup>III</sup> 3-thiophenecarboxylate complexes

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**Table S1.** Crystal data and structural refinement for lanthanoid 3-thiophenecarboxylate complexes

	<b>1a</b>	<b>1b</b>	<b>1c</b>	<b>1d</b>	<b>1e</b>	<b>1f</b>
	[La(3TPC)(H <sub>2</sub> O) <sub>3</sub> ] <sub>n</sub>	[Ce(3TPC)(H <sub>2</sub> O) <sub>3</sub> ] <sub>n</sub>	[Pr(3TPC)(H <sub>2</sub> O) <sub>3</sub> ] <sub>n</sub>	[Nd(3TPC)(H <sub>2</sub> O) <sub>3</sub> ] <sub>n</sub>	[Sm(3TPC)(H <sub>2</sub> O) <sub>3</sub> ] <sub>n</sub>	[Gd(3TPC)(H <sub>2</sub> O) <sub>3</sub> ] <sub>n</sub>
Formula	C <sub>15</sub> H <sub>15</sub> LaO <sub>9</sub> S <sub>3</sub>	C <sub>15</sub> H <sub>15</sub> CeO <sub>9</sub> S <sub>3</sub>	C <sub>15</sub> H <sub>15</sub> O <sub>9</sub> PrS <sub>3</sub>	C <sub>15</sub> H <sub>15</sub> NdO <sub>9</sub> S <sub>3</sub>	C <sub>15</sub> H <sub>15</sub> O <sub>9</sub> S <sub>3</sub> Sm	C <sub>15</sub> H <sub>15</sub> GdO <sub>9</sub> S <sub>3</sub>
<i>M<sub>r</sub></i>	574.36	575.57	576.36	579.69	585.80	592.70
Space group	P2 <sub>1</sub> /c	P2 <sub>1</sub> /c	P2 <sub>1</sub> /c	P2 <sub>1</sub> /c	P2 <sub>1</sub> /c	P2 <sub>1</sub> /c
<i>a</i> (Å)	15.070(3)	15.0999(6)	15.059(3)	15.020(3)	15.042(3)	14.960(3)
<i>b</i> (Å)	12.120(2)	12.1722(4)	12.157(2)	12.070(2)	12.055(2)	12.000(2)
<i>c</i> (Å)	10.840(2)	10.8103(3)	10.775(2)	10.720(2)	10.666(2)	10.600(2)
<i>α</i> (°)	90	90	90	90	90	90
<i>β</i> (°)	100.41(3)	100.769(3)	100.81(3)	100.78(3)	101.00(3)	101.07(3)
<i>γ</i> (°)	90	90	90	90	90	90
<i>V</i> (Å <sup>3</sup> )	1947.3(7)	1951.93(12)	1937.5(7)	1909.1(7)	1898.6(7)	1867.5(7)
<i>Z</i>	4	4	4	4	4	4
<i>ρ</i> <sub>calc</sub> , g cm <sup>-3</sup>	1.959	1.959	1.976	2.017	2.049	2.108
<i>μ</i> , mm <sup>-1</sup>	2.562	2.699	22.751	3.095	3.470	3.935
<i>N<sub>τ</sub></i>	41288	11030	3776	3344	3819	39196
<i>N</i> ( <i>R</i> <sub>int</sub> )	3420 (0.0646)	4592(0.0457)	3776	3344	3819	3298(0.0491)
<i>R</i> <sub>1</sub> ( <i>I</i> > 2σ( <i>I</i> ))	0.0345	0.0408	0.0522	0.0342	0.0615	0.0312
<i>wR</i> <sub>2</sub> (all data)	0.0968	0.0851	0.1412	0.1060	0.1791	0.0896
GOF	1.099	1.034	1.050	1.080	1.047	1.037

	<b>2a</b>	<b>2b</b>	<b>2c</b>	<b>2d</b>	<b>2e</b>
	<b>[Dy<sub>2</sub>(3TPC)<sub>3</sub>(H<sub>2</sub>O)<sub>4</sub>]·H<sub>2</sub>O</b>	<b>[Ho<sub>2</sub>(3TPC)<sub>3</sub>(H<sub>2</sub>O)<sub>4</sub>]·H<sub>2</sub>O</b>	<b>[Y<sub>2</sub>(3TPC)<sub>3</sub>(H<sub>2</sub>O)<sub>4</sub>]·H<sub>2</sub>O</b>	<b>[Er<sub>2</sub>(3TPC)<sub>3</sub>(H<sub>2</sub>O)<sub>4</sub>]·H<sub>2</sub>O</b>	<b>[Lu<sub>2</sub>(3TPC)<sub>3</sub>(H<sub>2</sub>O)<sub>4</sub>]·H<sub>2</sub>O</b>
Formula	C <sub>30</sub> H <sub>28</sub> Dy <sub>2</sub> O <sub>17</sub> S <sub>6</sub>	C <sub>30</sub> H <sub>28</sub> Ho <sub>2</sub> O <sub>17</sub> S <sub>6</sub>	C <sub>30</sub> H <sub>28</sub> O <sub>17</sub> S <sub>6</sub> Y <sub>2</sub>	C <sub>30</sub> H <sub>28</sub> Er <sub>2</sub> O <sub>17</sub> S <sub>6</sub>	C <sub>30</sub> H <sub>28</sub> Lu <sub>2</sub> O <sub>17</sub> S <sub>6</sub>
<i>M<sub>r</sub></i>	1177.88	1182.74	1030.70	1187.40	1202.82
Space group	P2 <sub>1</sub> /c	P2 <sub>1</sub> /c	P2 <sub>1</sub> /c	P2 <sub>1</sub> /c	P2 <sub>1</sub> /n
<i>a</i> (Å)	9.9091(7)	9.9388(6)	9.920(2)	9.9187(7)	9.8729(3)
<i>b</i> (Å)	19.1194(7)	19.2401(11)	19.170(4)	19.1800(7)	19.2032(5)
<i>c</i> (Å)	11.0210(7)	11.0949(8)	11.080(2)	11.0843(7)	11.0534(4)
<i>α</i> (°)	90	90	90	90	90
<i>β</i> (°)	115.932(8)	116.176(8)	116.41(3)	116.458(8)	116.308(4)
<i>γ</i> (°)	90	90	90	90	90
<i>V</i> (Å <sup>3</sup> )	1877.8(2)	1904.0(2)	1887.1(8)	1887.8(2)	1878.57(12)
<i>Z</i>	2	2	2	2	2
<i>ρ</i> <sub>calc</sub> , g cm <sup>-3</sup>	2.083	2.063	1.814	2.089	2.126
<i>μ</i> , mm <sup>-1</sup>	4.357	4.528	3.466	4.821	5.633
<i>N<sub>τ</sub></i>	12382	8447	30379	36686	26278
<i>N</i> ( <i>R</i> <sub>int</sub> )	3996(0.0339)	4364(0.0345)	3313(0.0407)	4315(0.0846)	5615(0.0460)
<i>R</i> <sub>1</sub> ( <i>I</i> > 2σ( <i>I</i> ))	0.0256	0.0484	0.0357	0.0235	0.0303
<i>wR</i> <sub>2</sub> (all data)	0.0581	0.0905	0.0978	0.0546	0.0620
GOF	1.051	1.136	1.073	1.036	1.051



**Table S2.** Selected bond lengths and distances (Å) for the isostructural [Ln(3TPC)<sub>3</sub>(H<sub>2</sub>O)<sub>3</sub>]<sub>n</sub> series.

	La1 ( <b>1a</b> )	Ce1 ( <b>1b</b> )	Pr1 ( <b>1c</b> )	Nd1 ( <b>1d</b> )	Sm1 ( <b>1e</b> )	Gd1 ( <b>1f</b> )
RE1#1 =RE1#2	6.3117(9)	6.2961(3)	6.2763(9)	6.2360(9)	6.2060(9)	6.1655(9)
O1	2.610(3)	2.598(3)	2.570(5)	2.529(4)	2.525(5)	2.478(4)
O2	2.583(3)	2.552(3)	2.531(5)	2.553(3)	2.464(5)	2.494(3)
O3	2.568(3)	2.602(3)	2.577(5)	2.555(4)	2.518(4)	2.451(3)
O4	2.620(3)	2.550(3)	2.523(5)	2.502(3)	2.499(5)	2.514(3)
O5	2.501(3)	2.484(3)	2.468(5)	2.464(3)	2.430(4)	2.413(3)
O6#1	2.510(3)	2.504(3)	2.489(5)	2.456(3)	2.441(4)	2.421(3)
O7	2.577(3)	2.518(3)	2.538(5)	2.461(3)	2.434(5)	2.410(3)
O8	2.524(3)	2.499(3)	2.485(5)	2.469(3)	2.452(4)	2.418(3)
O9	2.518(3)	2.557(3)	2.495(5)	2.523(3)	2.493(4)	2.471(3)

**Table S3.** Hydrogen bonds for [Pr(3TPC)<sub>3</sub>(H<sub>2</sub>O)<sub>3</sub>]<sub>n</sub> (**1c**) [d/Å and </°].

D-H...A	d(D-H)	d(H...A)	d(D...A)	<(DHA)
O7#1-H7B#1...O5	0.8699	2.2476	2.9587	138.882
C10-H10...O2	0.9502	2.1510	3.0093	149.599

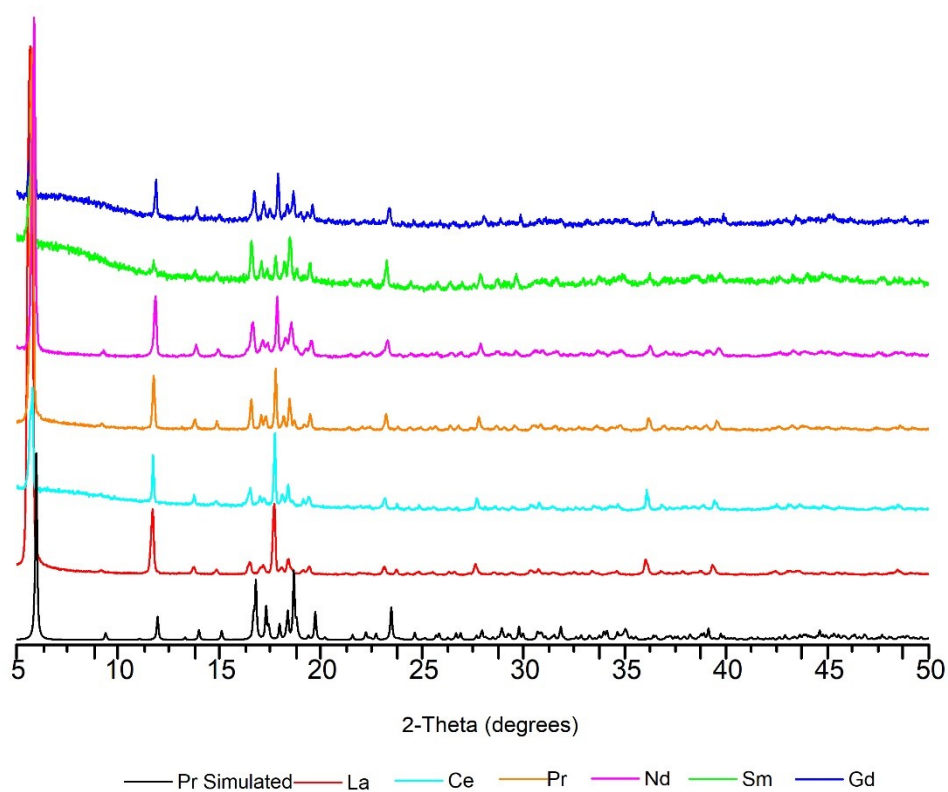
**Table S4.** Selected bond lengths and distances (Å) for the isostructural [Ln<sub>2</sub>(3TPC)<sub>6</sub>(H<sub>2</sub>O)<sub>4</sub>].H<sub>2</sub>O series.

	Dy1 ( <b>2a</b> )	Ho1 ( <b>2b</b> )	Y1 ( <b>2c</b> )	Er1 ( <b>2d</b> )	Lu1 ( <b>2e</b> )
RE1#1 =RE1#2	4.3346(5)	4.4325(6)	4.4708(10)	4.4809(4)	4.5153(2)
O1	2.446(2)	2.435(4)	2.405(19)	2.426(2)	2.367(2)
O2	2.423(2)	2.410(4)	2.435(2)	2.398(2)	2.409(2)
O3	2.459(2)	2.455(4)	2.437(2)	2.433(2)	2.368(2)
O4	2.400(2)	2.401(4)	2.402(19)	2.394(2)	2.396(2)
O5	2.285(3)	2.249(4)	2.235(2)	2.229(2)	2.187(2)
O6#1	2.325(2)	2.294(5)	2.290(2)	2.283(2)	2.240(3)
O7	2.325(2)	2.395(4)	2.392(2)	2.385(2)	2.299(2)
O8	2.361(2)	2.342(4)	2.341(18)	2.330(2)	2.355(2)

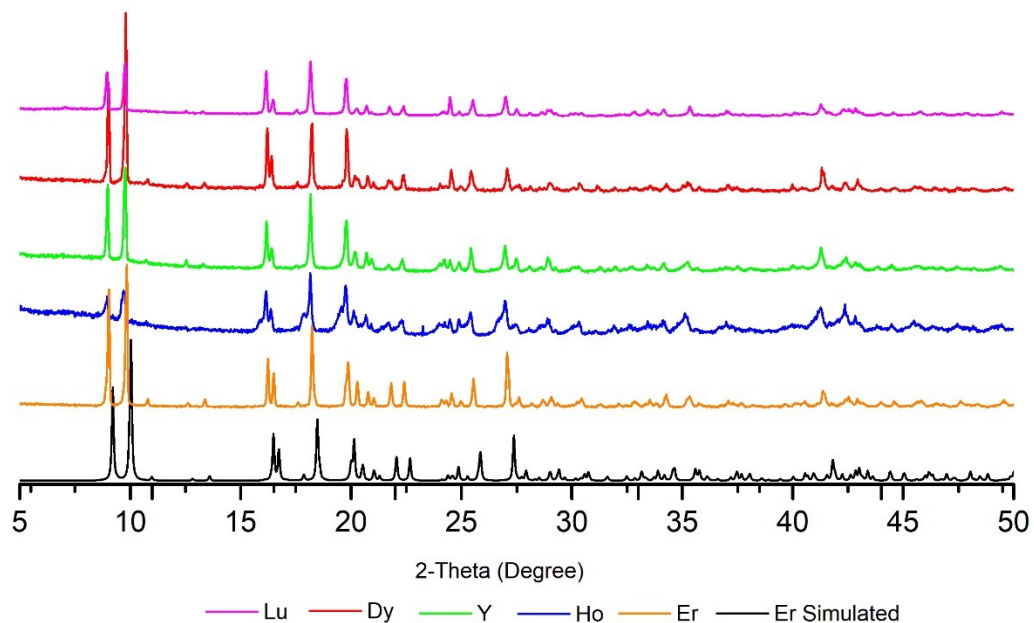
**Table S5.** Hydrogen bonds for  $[\text{Er}_2(3\text{TPC})_6(\text{H}_2\text{O})_4]\cdot\text{H}_2\text{O}$  (**2d**) [d/Å and </math>°].

D-H...A	d(D-H)	d(H...A)	d(D...A)	<(DHA)
O7-H7B ...O9	0.8628(2)	1.9557(6)	2.7291(6)	148.551(2)
O7-H7A ...O4#1	0.8615(2)	1.8623(2)	2.7107(3)	167.887(16)

### Powder X-Ray Diffraction

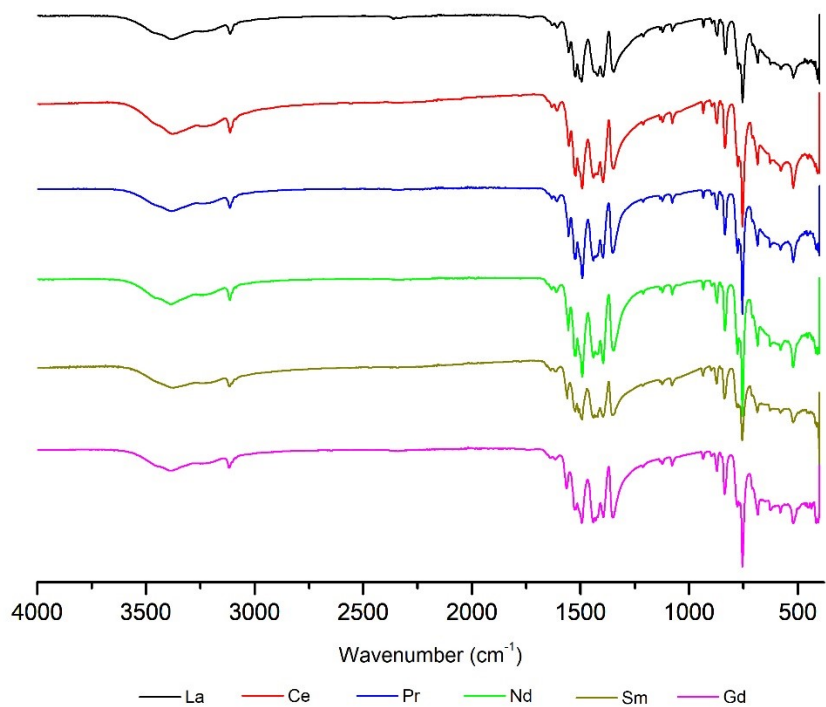


**Figure S1.** PXRD traces of the  $[\text{Ln}(3\text{TPC})_3(\text{H}_2\text{O})_3]$  (**1a-f**) series at room temperature compared to the simulated pattern of  $[\text{Pr}(3\text{TPC})_3(\text{H}_2\text{O})_3]_n$  (**1c**)

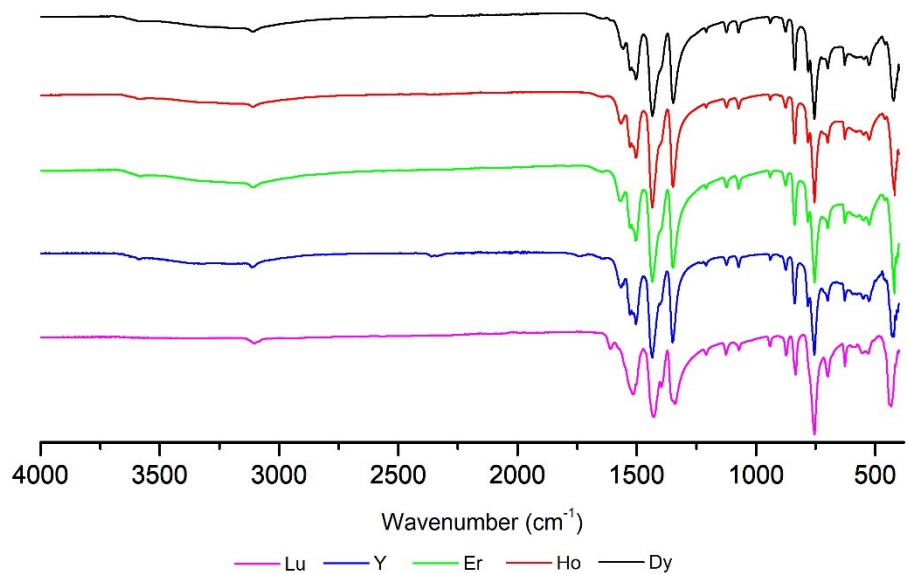


**Figure S2.** PXRD traces of the  $[\text{Ln}_2(3\text{TPC})_6(\text{H}_2\text{O})_4]\cdot\text{H}_2\text{O}$  (**2a-e**) series at room temperature compared to the simulated pattern of  $[\text{Er}_2(3\text{TPC})_6(\text{H}_2\text{O})_4]\cdot\text{H}_2\text{O}$  (**2d**)

### Infrared Spectroscopy

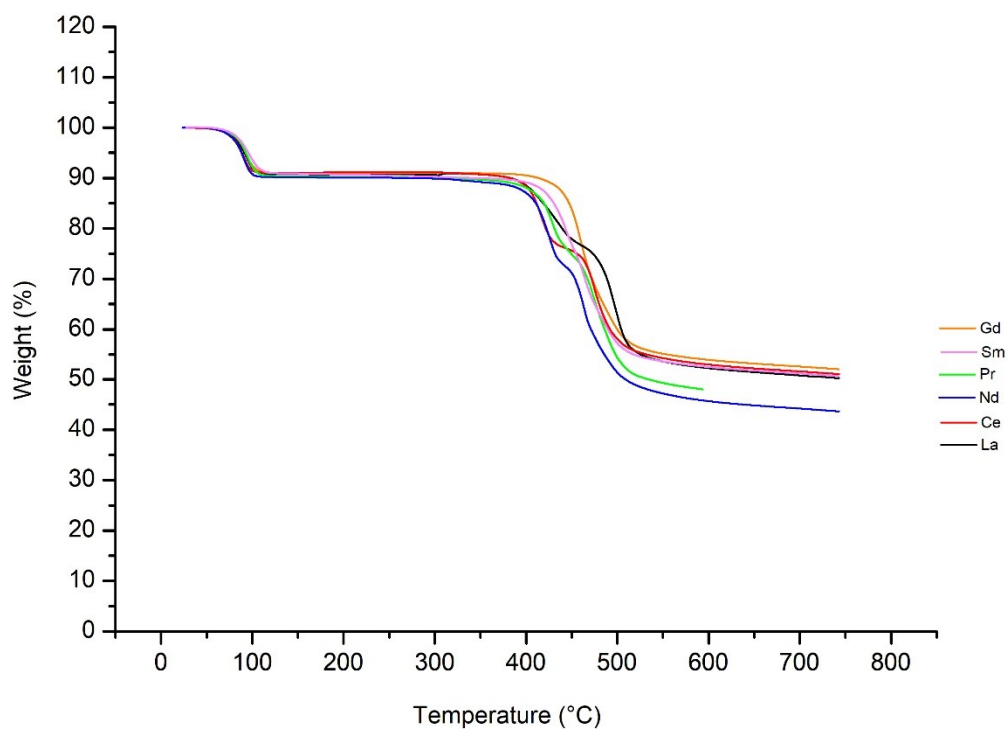


**Figure S3.** IR spectra of  $[\text{Ln}(3\text{TPC})_3(\text{H}_2\text{O})_3]_n$  (**1a-f**) series



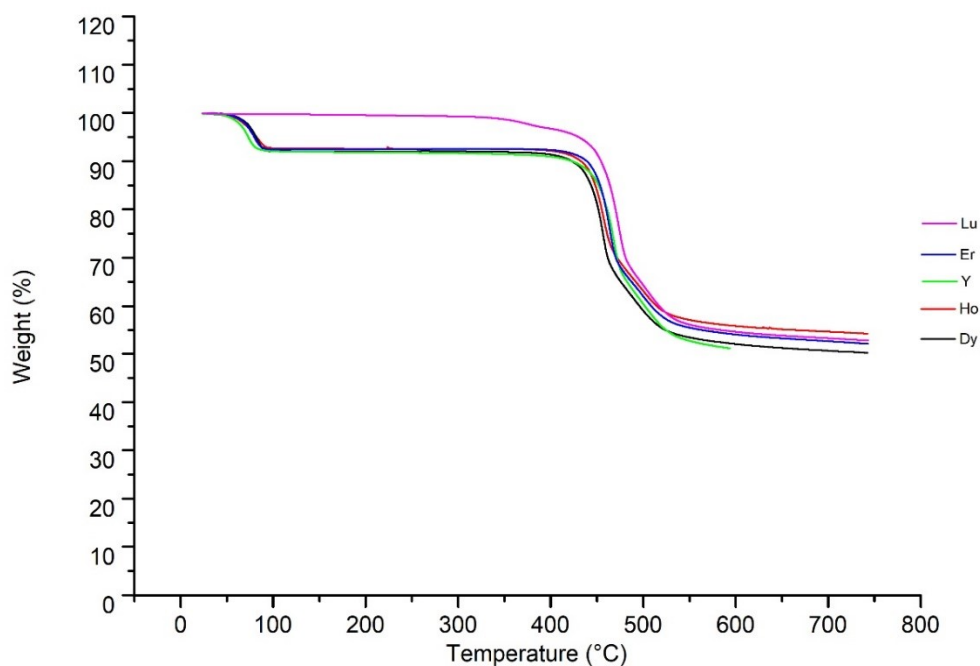
**Figure S4.** IR spectra of  $[\text{Ln}_2(3\text{TPC})_6(\text{H}_2\text{O})_4]\cdot\text{H}_2\text{O}$  (**2a-e**) series

### Thermogravimetric Analysis

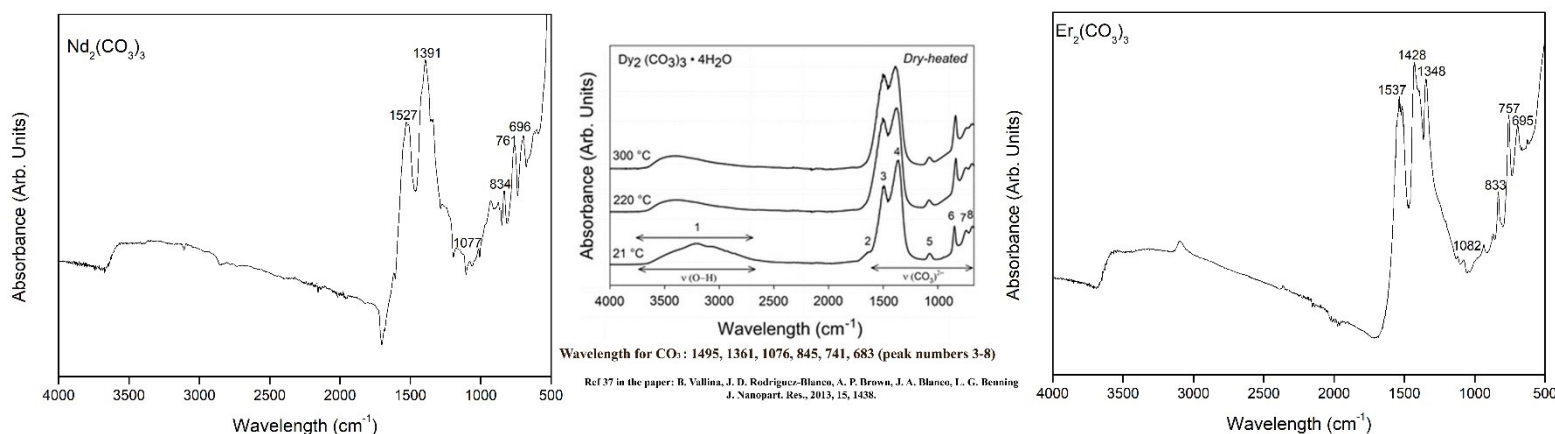


**Figure S5.** TGA plots of the  $[\text{Ln}(3\text{TPC})_3(\text{H}_2\text{O})_3]_n$  (**1a-f**) series





**Figure S6.** TGA plots of the  $[\text{Ln}_2(3\text{TPC})_6(\text{H}_2\text{O})_4]\cdot\text{H}_2\text{O}$  (**2a-e**) series



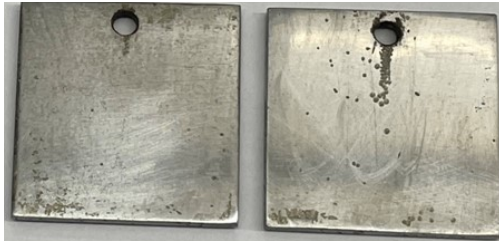
**Figure S7:** IR spectra of  $\text{Nd}_2(\text{CO}_3)_3$  (Left) from thermal decomposition of **1d**, reported  $\text{Dy}_2(\text{CO}_3)_3\cdot 4\text{H}_2\text{O}$  (middle) and  $\text{Er}_2(\text{CO}_3)_3$  (right) from thermal decomposition of **2d**

### Weight loss experiments

#### *NaCl 0.01 M control*



**$[Y_2(3TPC)_6(H_2O)_4] \cdot H_2O$  (2c) - 500 ppm**



**$[Er_2(3TPC)_6(H_2O)_4] \cdot H_2O$  (2d) - 500 ppm**



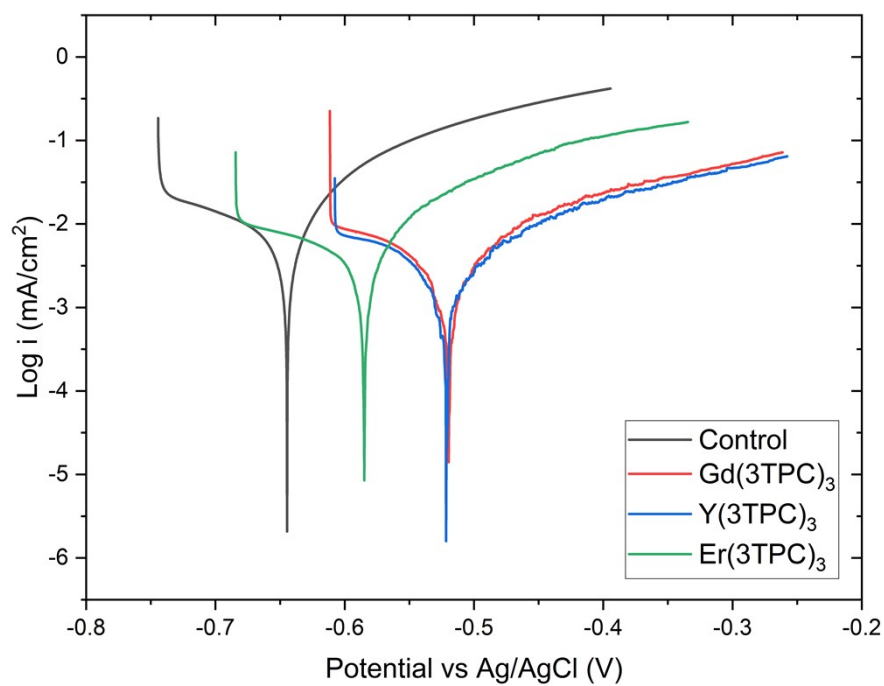
**$[Gd(3TPC)_3(H_2O)_3]_n$  (1f) - 500 ppm**



**$[La(3TPC)_3(H_2O)_3]_n$  (1a) - 500 ppm**



**Figure S8:** Visual comparisons pictures of few sample coupons after 7 days immersion in inhibitor solutions in NaCl 0.01M. (Left – Trial 1; right – Trial 2.)



**Figure S9:** Representative potentiodynamic polarisation scans after 24 hours for 0.01 M NaCl control, and solutions with 0.01 M NaCl and 500ppm of  $[\text{Gd}(\text{3TPC})_3(\text{H}_2\text{O})_3]_n$  (**1f**),  $[\text{Y}_2(\text{3TPC})_6(\text{H}_2\text{O})_4] \cdot \text{H}_2\text{O}$  (**2c**) or  $[\text{Er}_2(\text{3TPC})_6(\text{H}_2\text{O})_4] \cdot \text{H}_2\text{O}$  (**2d**)