

Electronic Supplementary Information

A Novel 2D Metal-Organic Framework Probe: Highly Sensitive and Visual Fluorescent Sensor for Al^{3+} , Cr^{3+} , Fe^{3+} Ions

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Table S1. X-ray crystallographic data for **Tb-DBA**

Formula	C ₃₃ H ₂₁ N _{2.5} O ₈ Tb
Fw	1507.11
Crystal system	Triclinic
Space group	<i>p</i> -1
<i>a</i> (Å)	8.3473 (3)
<i>b</i> (Å)	17.4598 (7)
<i>c</i> (Å)	18.3029 (7)
α (°)	76.871(2)
β (°)	89.666 (2)
γ (°)	81.002(2)
<i>V</i> (Å ³)	2564.59(17)
<i>Z</i>	1
<i>D_c</i> (g/cm ³)	0.976
<i>F</i> (000)	731.0
Reflns collected	10503
Completeness(%)	99.5
GOF on <i>F</i> ²	1.040
<i>R</i> ₁ /w <i>R</i> ₂ [I>2σ(I)]	0.0405/0.1169

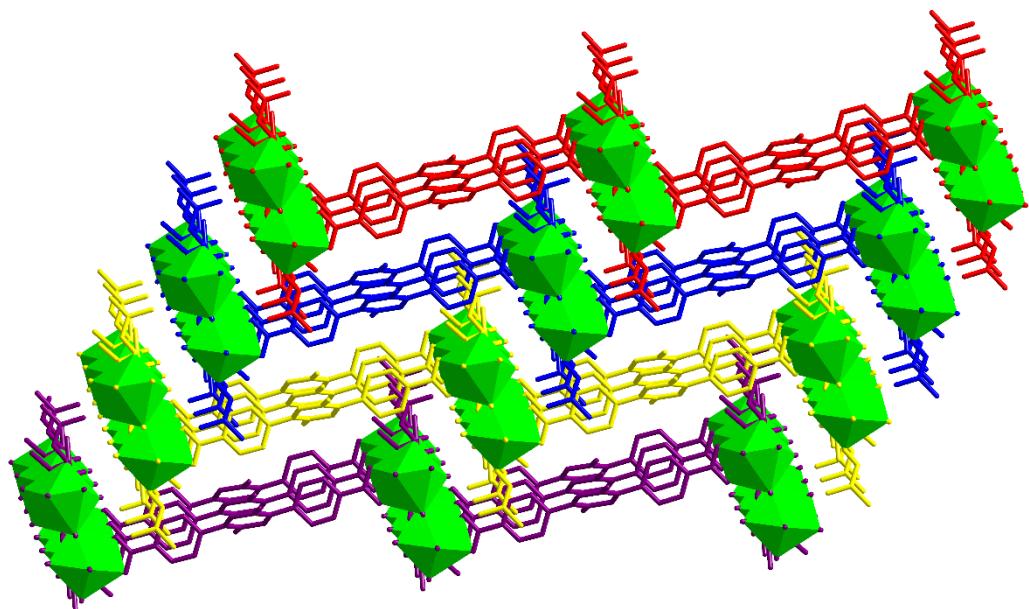


Fig. S1 The stacking structure of Tb-DBA.

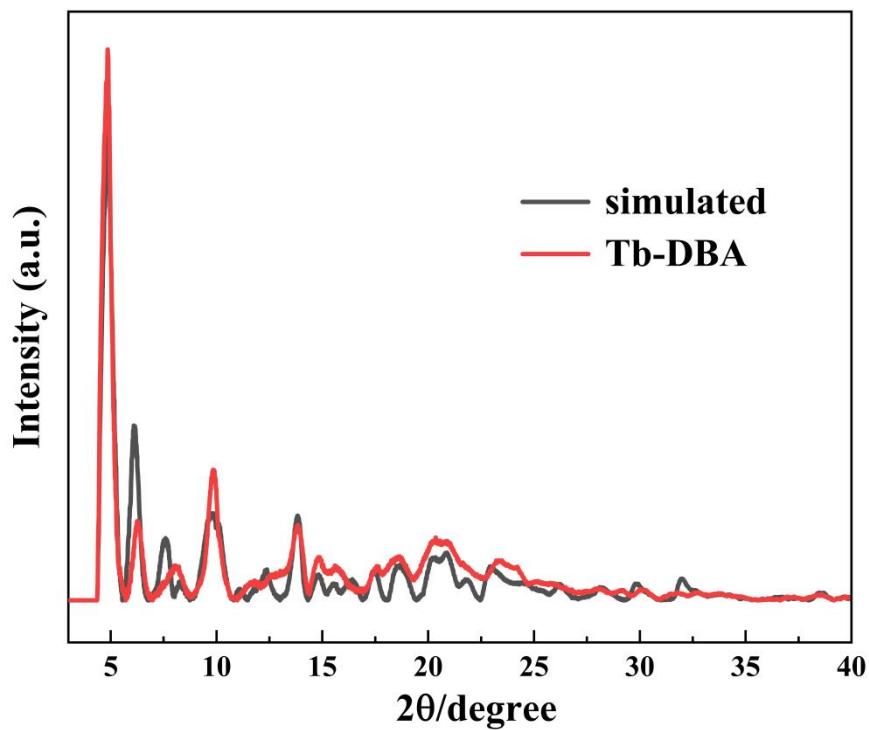


Fig. S2 As-synthesized and simulated PXRD patterns of Tb-DBA.

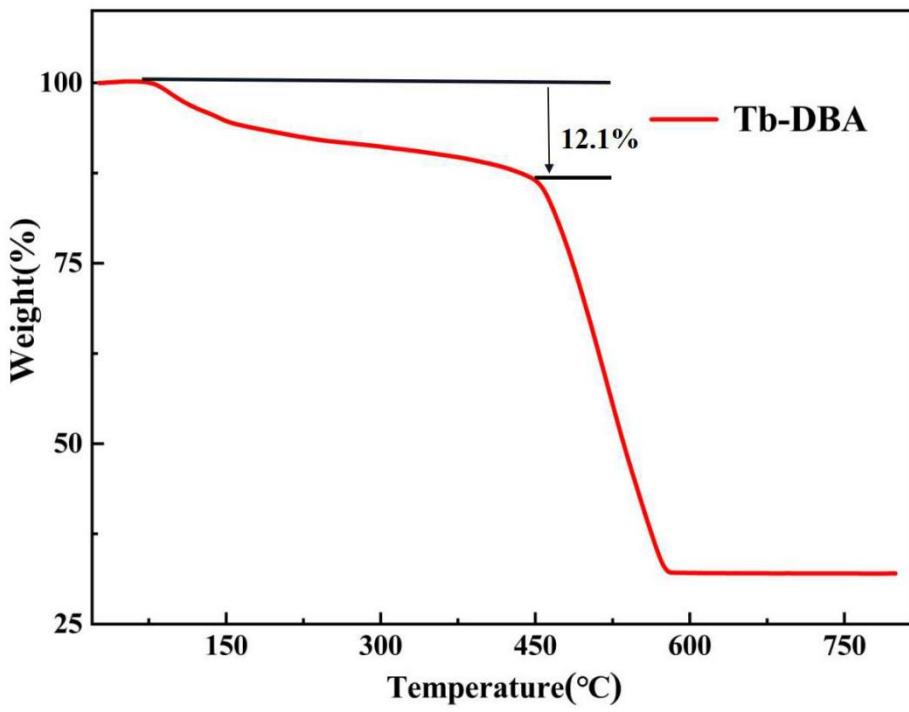


Fig. S3 TGA curves of Tb-DBA measured in N₂ atmosphere at a temperature ramp of 10°C min⁻¹.

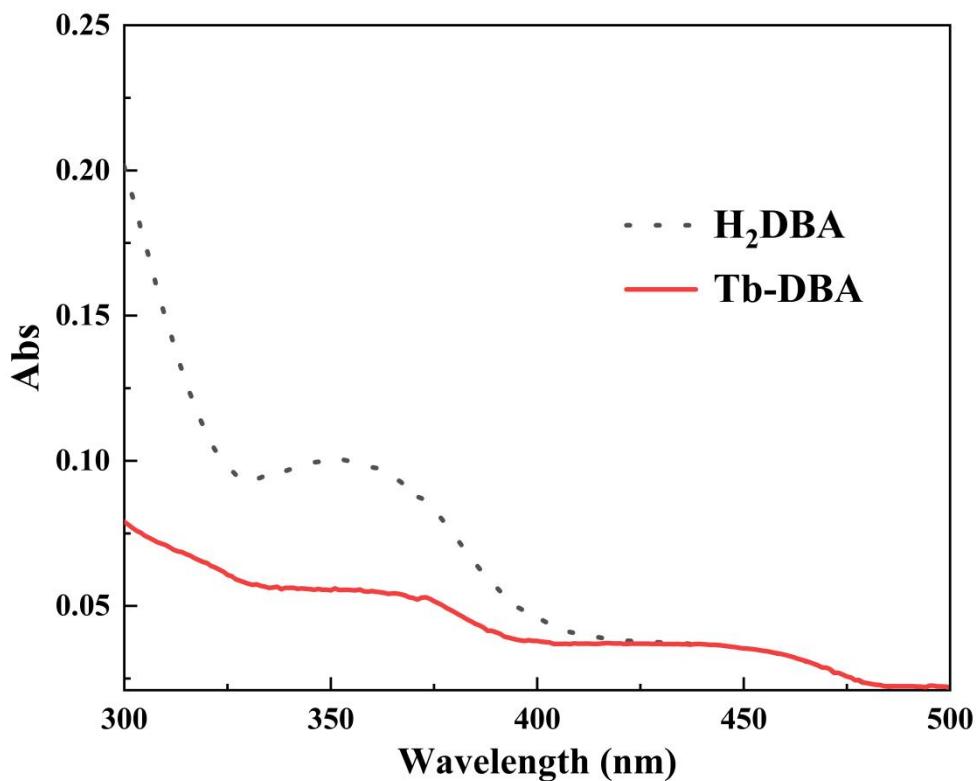


Fig. S4 UV-Vis spectra of H_2DBA and Tb-DBA .

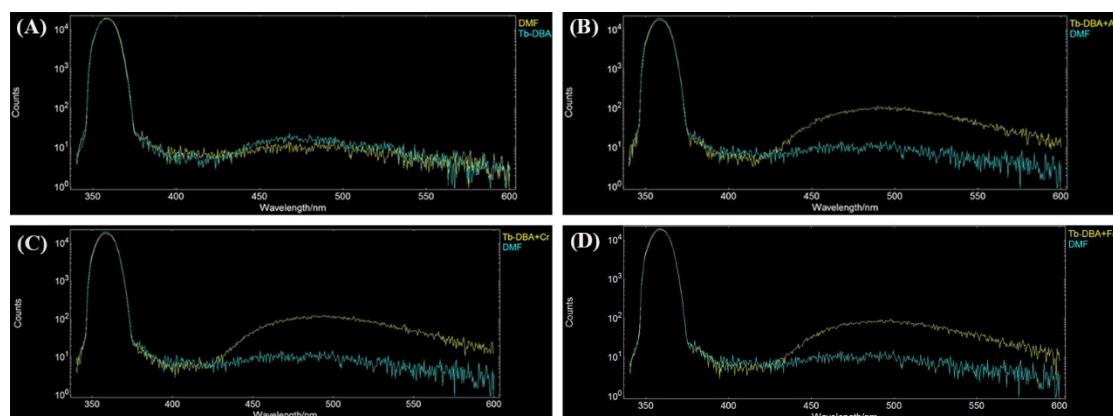


Figure S5. The quantum yields of Tb-DBA (A), $\text{Tb-DBA} + \text{Al}^{3+}$ (B), $\text{Tb-DBA} + \text{Cr}^{3+}$ (C) and $\text{Tb-DBA} + \text{Fe}^{3+}$ (D) .

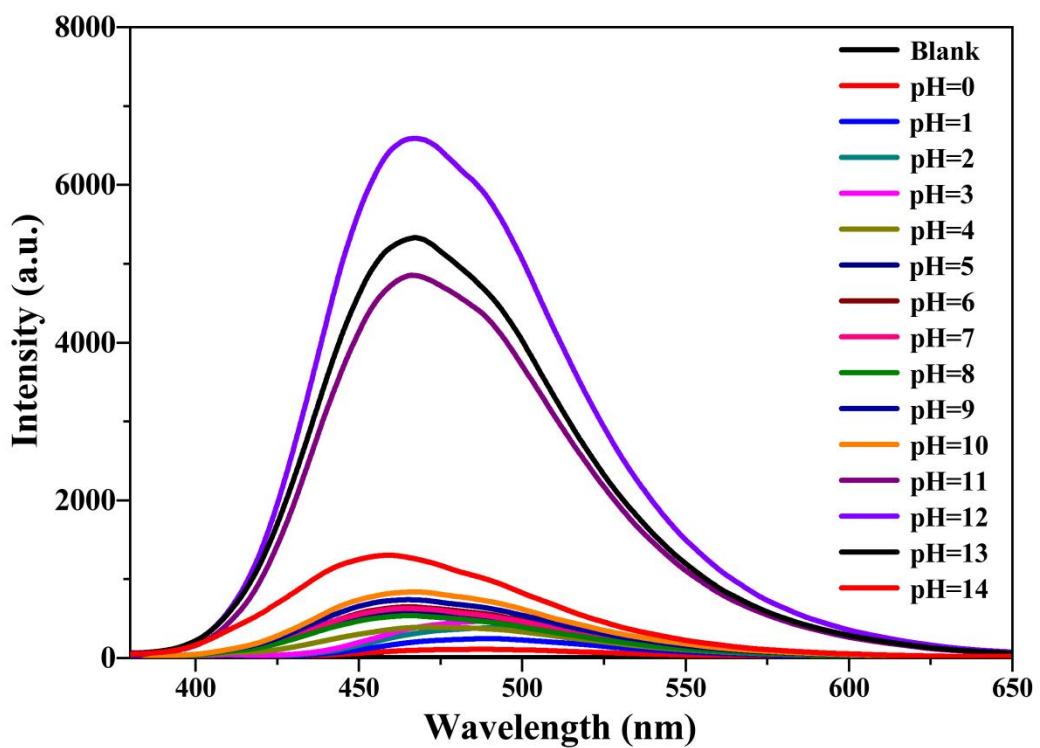


Fig. S6 Fluorescence of Tb-DBA in pH = 0-14 solutions.

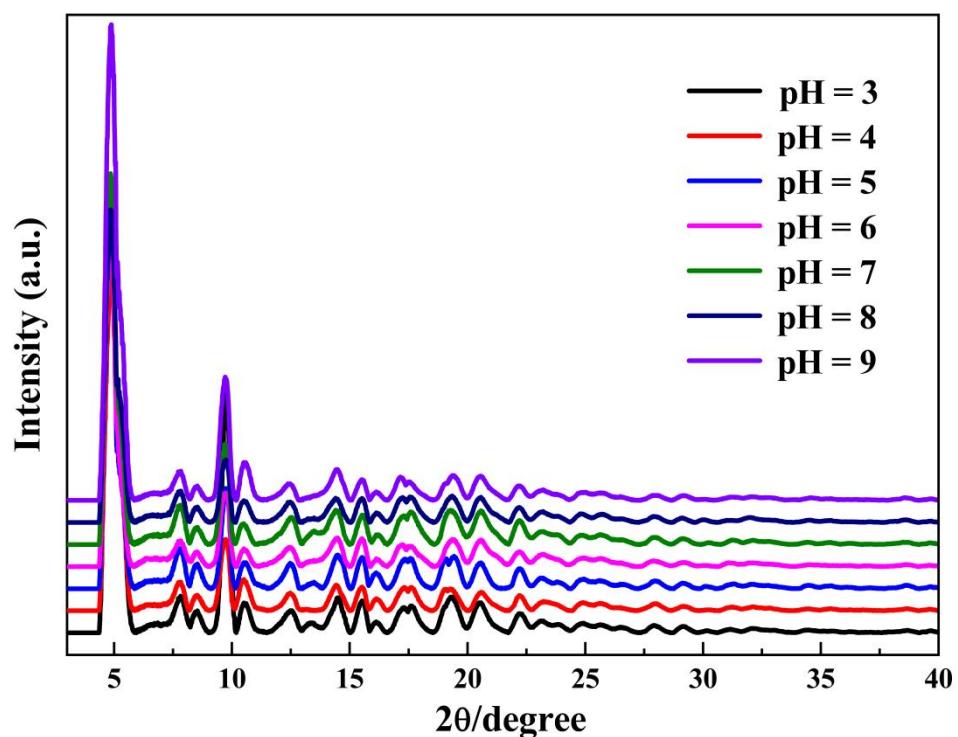


Fig. S7 PXRD patterns of Tb-DBA after soaked in different pH solutions.

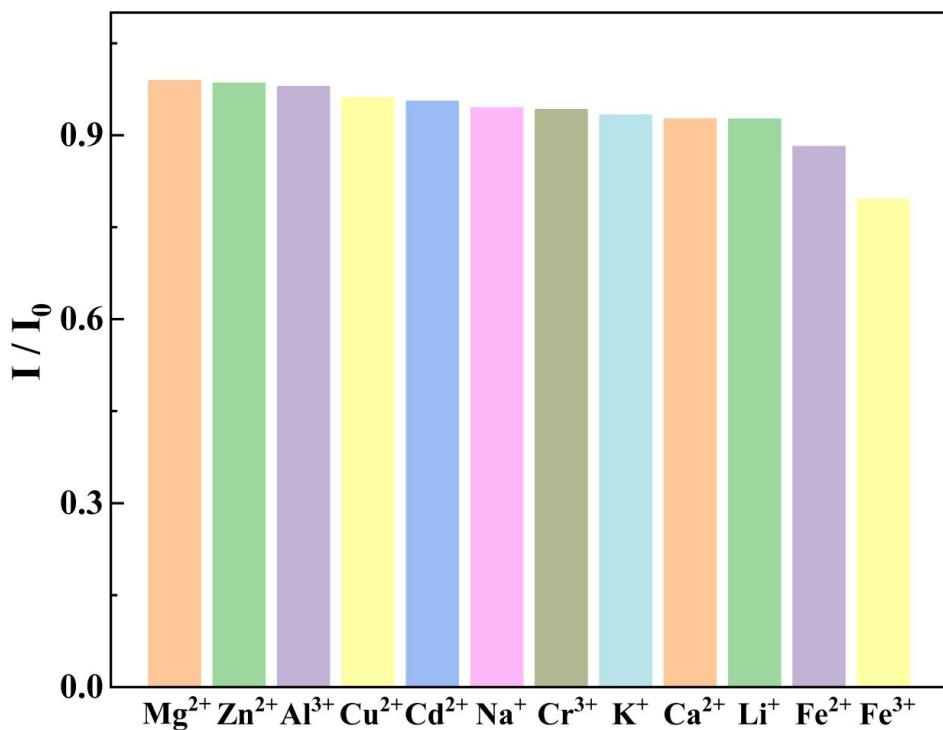


Fig. S8 Relative luminescent intensity of H₂DBA introduced into different cations at 485 nm.

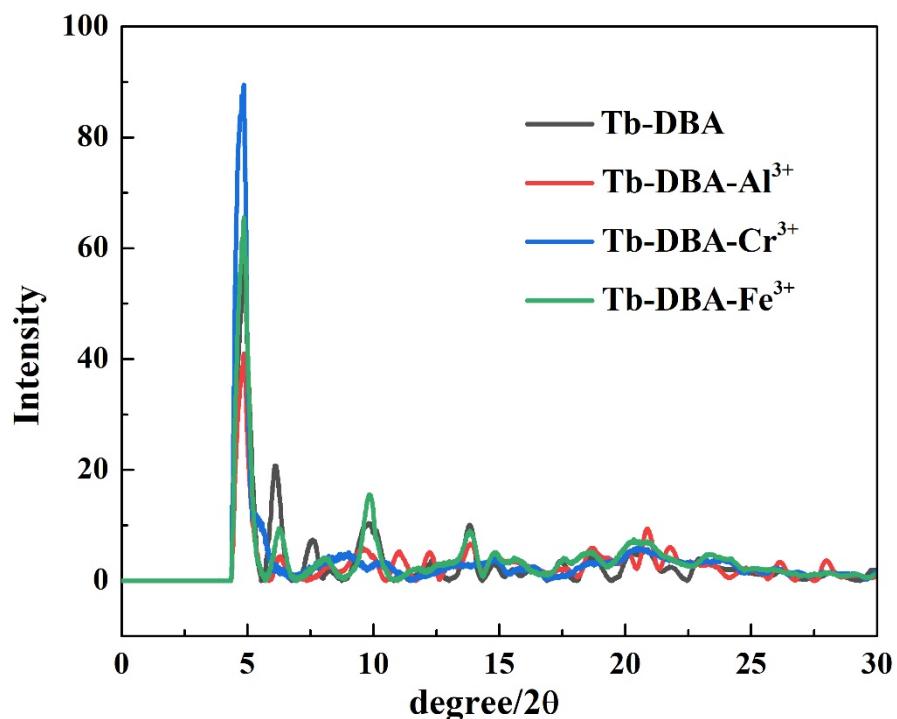


Fig. S9 The PXRD patterns of Tb-DBA, Tb-DBA-Al³⁺, Tb-DBA-Cr³⁺, and Tb-DBA-Fe³⁺.

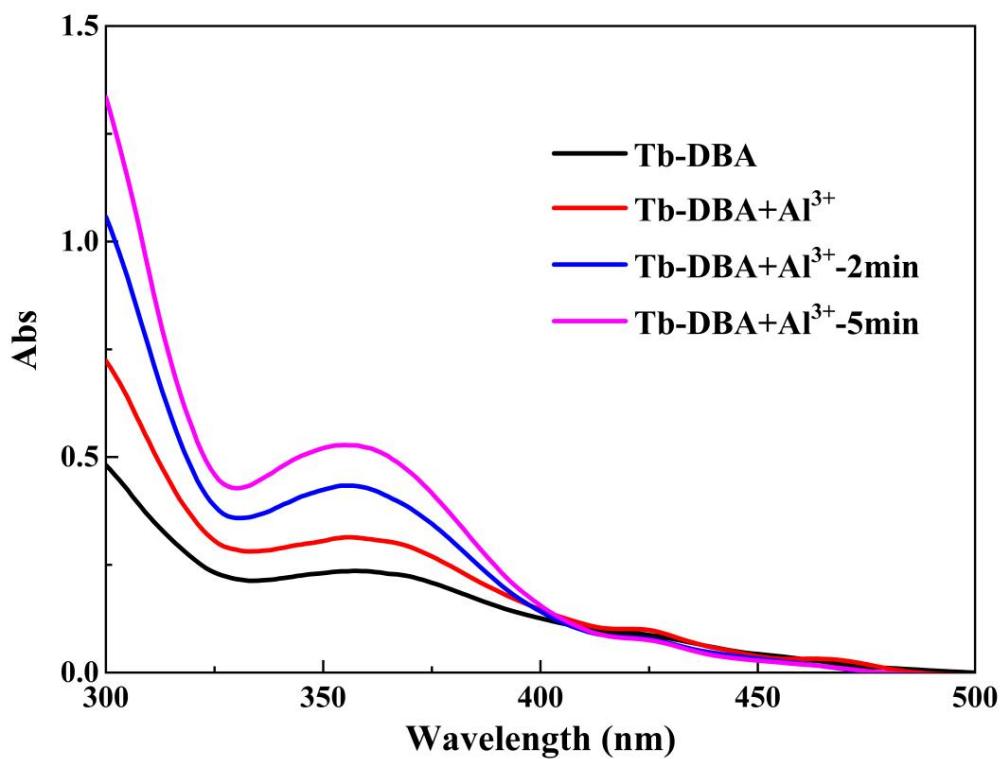


Fig. S10 UV-Vis spectra of Tb-DBA with Al³⁺ ions.

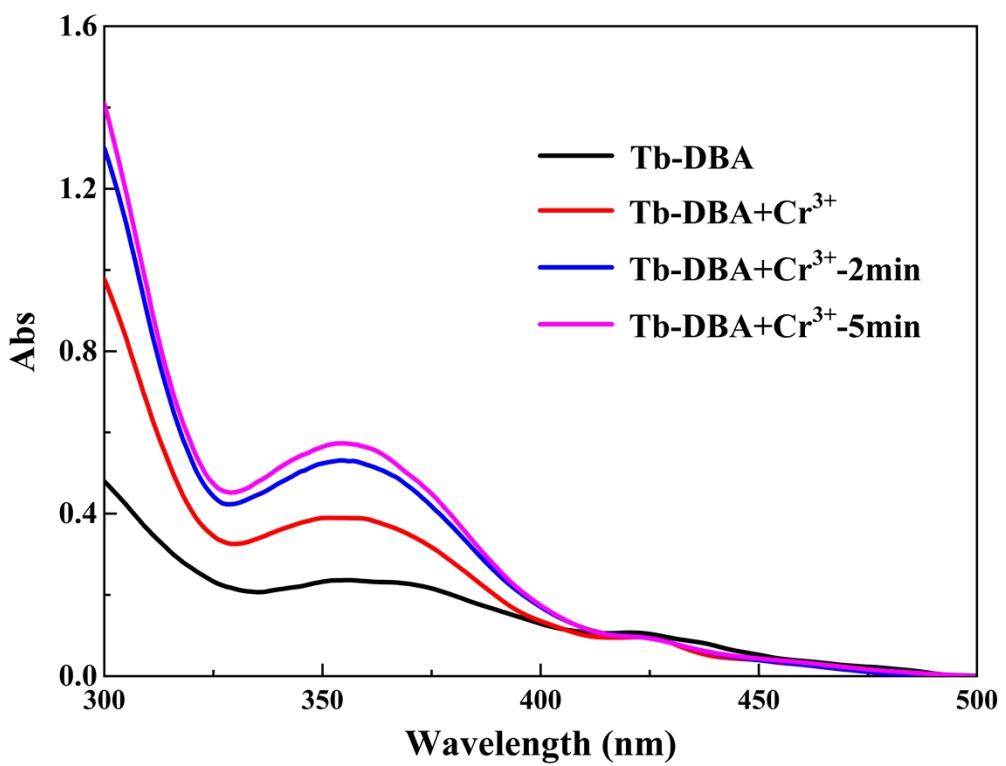


Fig. S11 UV-Vis spectra of Tb-DBA with Cr³⁺ ions.

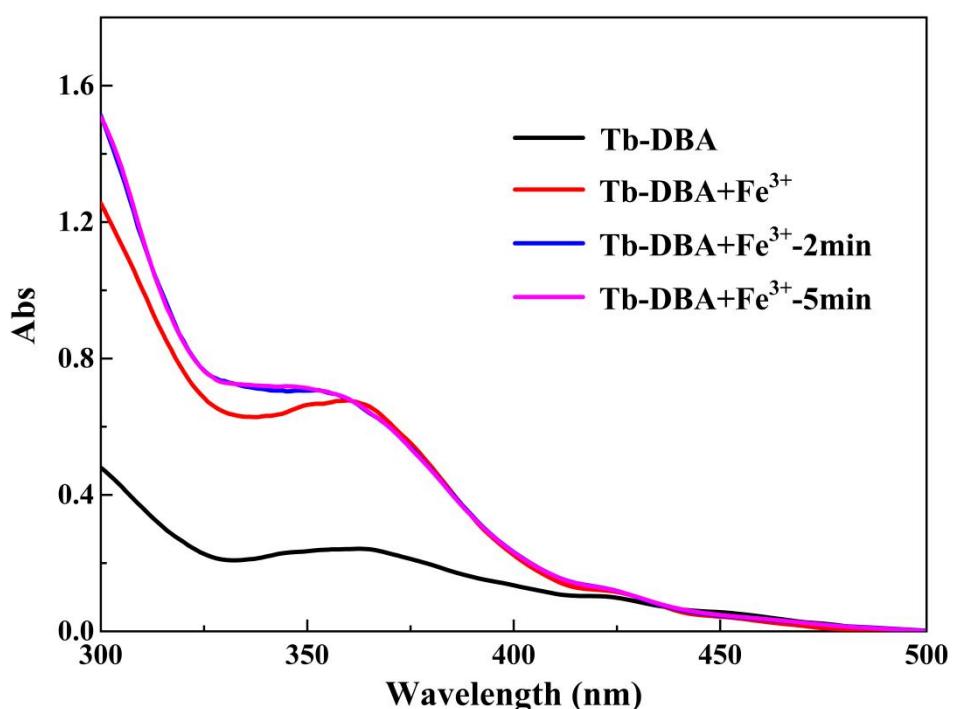


Fig. S12 UV-Vis spectra of Tb-DBA with Fe^{3+} ions.

Table S2. ICP analysis for Tb^{3+} contents.

Samples	Tb^{3+} (mmol/L)
Tb-DBA	0.00
Tb-DBA+Al ³⁺	2.0
Tb-DBA+Cr ³⁺	1.7
Tb-DBA+Fe ³⁺	1.2

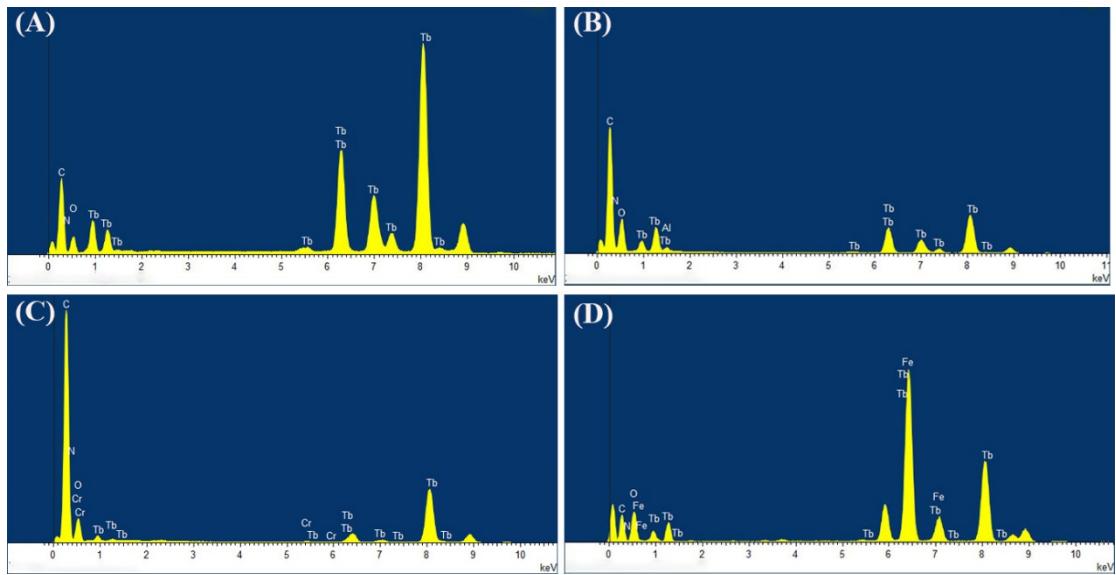


Figure S13. EDS measurement of Tb-DBA(A), Tb-DBA + Al³⁺(B), Tb-DBA + Cr³⁺(C) and Tb-DBA+Fe³⁺(D).